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Reduction of soil carbon formation by tropospheric ozone under increased carbon dioxide levels

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In the Northern Hemisphere, ozone levels in the troposphere have increased by 35 per cent over the past century¹, with detrimental impacts on forest^{2,3} and agricultural⁴ productivity, even when forest productivity has been stimulated by increased carbon dioxide levels⁵. In addition to reducing productivity, increased tropospheric ozone levels could alter terrestrial carbon cycling by lowering the quantity and quality of carbon inputs to soils. However, the influence of elevated ozone levels on soil carbon formation and decomposition are unknown. Here we examine the effects of elevated ozone levels on the formation rates of total and decay-resistant acid-insoluble soil carbon under conditions of elevated carbon dioxide levels in experimental aspen (*Populus tremuloides*) stands and mixed aspen–birch (*Betula papyrifera*) stands. With ambient concentrations of ozone and carbon dioxide both raised by 50 per cent, we find that the formation rates of total and acid-insoluble soil carbon are reduced by 50 per cent relative to the amounts entering the soil when the forests were exposed to increased carbon dioxide alone. Our results suggest that, in a world with elevated atmospheric carbon dioxide concentrations, global-scale reductions in plant productivity due to elevated ozone levels will also lower soil carbon formation rates significantly.

Table 1 Total carbon and $\delta^{13}\text{C}$ of soils and the acid-insoluble fraction

	Total soil carbon		Acid-insoluble fraction	
	Total C (g m^{-2})	$\delta^{13}\text{C}$ (‰)	Total C (g m^{-2})	$\delta^{13}\text{C}$ (‰)
Control	5,385 ± 241	−26.7 ± 0.11	2,287 ± 185	−27.7 ± 0.06
Elevated O_3	5,237 ± 72	−26.9 ± 0.06	2,671 ± 224	−27.7 ± 0.11
Elevated CO_2	5,683 ± 480	−28.3 ± 0.10	2,594 ± 230	−29.0 ± 0.12
Elevated $\text{O}_3 + \text{CO}_2$	5,114 ± 369	−27.6 ± 0.10	2,473 ± 97	−28.5 ± 0.10

Determined for mineral soil from 0 to 20 cm. Values are mean ± s.e.

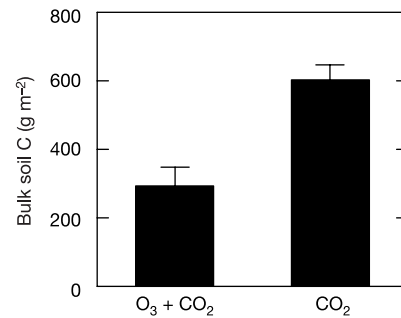


Figure 1 Total carbon incorporated into soils during 4 yr of exposure to elevated $\text{O}_3 + \text{CO}_2$ and elevated CO_2 . Values are mean ± 1 s.e.; $P < 0.01$.

Large areas of the Earth are exposed to concentrations of tropospheric ozone (O_3) that exceed levels known to be toxic to plants^{4,6}. In addition to reducing plant growth, exposure to elevated O_3 can also alter plant tissue chemistry⁷ and reduce allocation of carbon to roots and root exudates^{8–10}. Whereas the effects of O_3 on these aspects of plant biology have been widely investigated in chamber studies, examination of these effects on below-ground carbon cycling in intact forests only became possible in 1997 with the establishment of the FACTS-II (forest–atmosphere carbon transfer and storage) FACE (free-air carbon dioxide enrichment) experiment in Rhinelander, Wisconsin, USA. The long-term FACE experiment in Rhinelander examines how plant–plant and plant–microbe interactions may alter ecosystem responses to elevated O_3 and carbon dioxide (CO_2) through four treatments: control, elevated CO_2 , elevated O_3 and elevated $\text{O}_3 + \text{CO}_2$. In plots where O_3 and CO_2 are elevated, concentrations were maintained at ~150% of ambient levels¹¹. To examine the effects of atmospheric trace gases on both ecological interactions and on whole-ecosystem carbon cycling, each plot is split to include a pure aspen forest, a mixed aspen–birch forest and a mixed aspen–maple forest. These species were chosen because they are among the most widely distributed trees in northern temperate forests.

Here we compare soil carbon formation in aspen and aspen–birch subplots under elevated CO_2 and elevated $\text{O}_3 + \text{CO}_2$ (three plots each) to understand how exposure to O_3 under elevated CO_2 alters soil carbon formation. We used CO_2 derived from fossil fuel with its highly depleted ^{13}C signature to fumigate plant canopies in the elevated CO_2 and elevated $\text{O}_3 + \text{CO}_2$ plots. Leaf and root carbon inputs in the elevated CO_2 and elevated $\text{O}_3 + \text{CO}_2$ plots had a $\delta^{13}\text{C}$ signature of $-41.6 \pm 0.4\text{‰}$ (mean ± s.e.) in contrast to leaf and root inputs of $-27.6 \pm 0.3\text{‰}$ in the control plots. The $\delta^{13}\text{C}$ signature of soil carbon was $-26.7 \pm 0.2\text{‰}$ before fumigation and

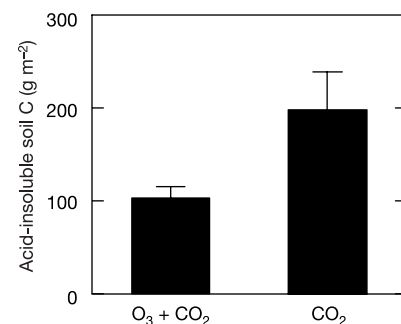


Figure 2 Carbon incorporated into the stable acid-insoluble fraction of soils during 4 yr of exposure to elevated $\text{O}_3 + \text{CO}_2$ and elevated CO_2 . Values are mean ± 1 s.e.; $P < 0.01$.

Table 2 Quantity and relative proportion of total soil carbon and soil carbon pools lost via microbial respiration in laboratory incubations

	Total carbon		Fumigation carbon		Pre-fumigation carbon	
	Cumulative loss	Relative mass loss	Cumulative loss	Relative mass loss	Cumulative loss	Relative mass loss
Control	1,372 ± 62	0.10 ± 0.01				
Elevated O ₃	1,392 ± 36	0.11 ± 0.00				
Elevated CO ₂	1,673 ± 85	0.12 ± 0.01	438 ± 23	0.27 ± 0.02	1235 ± 71	0.10 ± 0.00
Elevated O ₃ + CO ₂	1,342 ± 42	0.11 ± 0.01	267 ± 26	0.42 ± 0.06	1075 ± 36	0.09 ± 0.00
P	0.09	0.22	0.04	0.10	0.31	0.17

Cumulative loss (mg CO₂-C per kg soil). Relative mass loss (g CO₂-C per g soil-C) determined for day 281 of incubation. Values are mean ± s.e.

in control plots. Because plant carbon inputs to the soils fumigated with elevated CO₂ and elevated O₃ + CO₂ were depleted in ¹³C relative to soil carbon that existed before the experiment began, incorporation of new plant detritus into soil organic matter through time should decrease the δ¹³C of soil carbon. Using standard mixing models, we were able to determine the fraction of total soil carbon and acid-insoluble soil carbon derived from atmospheric CO₂ fixed by the trees over the 4 yr of experimental fumigation.

Elevated O₃ profoundly altered the ¹³C composition of soil carbon after 4 yr of fumigation. The less depleted δ¹³C signature of total soil carbon from the elevated O₃ + CO₂ treatment compared to the elevated CO₂ treatment (Table 1; *P* < 0.01) indicates that less carbon entered the soil when aspen and mixed aspen–birch forests were exposed to both elevated O₃ and CO₂. Carbon inputs to soil over 4 yr accounted for 10.7 ± 0.6% of the total soil carbon under elevated CO₂, but only 5.7 ± 0.9% under elevated O₃ + CO₂. Thus, elevated O₃ reduced total soil carbon formation by approximately 300 g C m⁻² compared to the amount formed under elevated CO₂ alone (Fig. 1).

Soils store carbon with both long and short turnover times, and to understand the carbon storage capacity of soils, it is necessary to partition soil carbon changes into decay-resistant and labile components. We used acid hydrolysis to isolate the most decay-resistant carbon in the soil¹². The acid-insoluble soil carbon fraction contains the longest-lived compounds, and is composed primarily of aromatic humic acids and residues of phenols, lignin and lignin-associated cellulose from newer plant residues¹³. However, after careful removal of plant residues before hydrolysis, new carbon inputs measured in this fraction are restricted to the most decomposition-resistant compounds. Compared to elevated CO₂ alone, simultaneous O₃ and CO₂ fumigation greatly reduced the quantity of carbon entering the acid-insoluble fraction in both aspen and aspen–birch soils, as indicated by the less depleted δ¹³C signature (Table 1; *P* < 0.04). Carbon incorporated into the soil since the initiation of the experiment accounted for approximately 9.1 ± 0.8% of the acid-insoluble carbon fraction in the elevated CO₂ plots, but only 4.2 ± 0.6% in the elevated O₃ + CO₂ plots. Thus, the elevated O₃ treatment reduced formation of acid-insoluble soil carbon by approximately 100 g C m⁻² (or 48%) compared with elevated CO₂ alone (Fig. 2).

Soil carbon formation is a balance between detrital inputs and decomposition losses. Observed reductions in soil carbon formation in the elevated O₃ + CO₂ treatment could result from reduced inputs, increased losses, or both. Previous research at this site suggests that plant detrital inputs are reduced under elevated O₃ + CO₂ relative to elevated CO₂ alone^{5,10}. To elucidate the effects of decomposition losses on soil carbon, we conducted a long-term laboratory incubation experiment. Despite greater microbial activity in soils from elevated CO₂ plots (Table 2), there was a strong trend of greater relative mass loss of carbon derived from fumigation in the elevated O₃ + CO₂ plots compared to the elevated CO₂ plots (Table 2; *P* < 0.1). This result suggests that the soil microbes responsible for carbon decomposition under elevated O₃ + CO₂ used a greater proportion (35%) of carbon formed since

fumigation began, leaving less of this carbon in the soil and in the acid-insoluble fraction. Possible reasons for greater consumption of recent carbon inputs include changes in microbial community composition¹⁴, microbial activity¹⁵, or carbon availability⁹. In turn, these below-ground changes could be driven by altered quantity and quality of detrital inputs^{5,10}.

The atmospheric trace gases CO₂ and O₃ are known to influence photosynthesis and ecosystem productivity in opposite ways. It is now clear that their interaction has the potential to cascade through ecosystems, altering soil carbon cycling. In response to reduced detrital carbon supply^{5,10} and increased microbial utilization, forest ecosystems exposed to both elevated O₃ and CO₂ accrued 51% less total, and 48% less acid-insoluble, soil carbon compared to ecosystems exposed only to elevated CO₂. Our results suggest that changes in the formation of soil carbon induced by tropospheric O₃ should be explicitly considered in models of soil carbon cycling under scenarios of elevated CO₂. □

Methods

Research area and field experiment

The FACTS-II FACE experiment in Rhinelander, Wisconsin, USA (49° 40.5' N, 89° 37.5' E, 490 m elevation) includes three experimental treatments (elevated CO₂, elevated O₃, and elevated O₃ + CO₂) and control plots arranged in a randomized complete block design with three replicates¹¹. FACE technology combines a trace gas monitoring system with a delivery system composed of blowers and vertical pipes placed around the perimeter of the forest stand to elevate ambient concentrations of O₃ and CO₂ (ref. 11). Ambient levels of CO₂ and O₃ are approximately 347 μl l⁻¹ and 37 nl l⁻¹, respectively. Elevated CO₂ and elevated O₃ treatments averaged 560 μl l⁻¹ and 52 nl l⁻¹, respectively, since the fumigation began. The experiment was initiated in the summer of 1997 with the planting of greenhouse-propagated tree seedlings at 1 × 1 m spacing, in the following combinations within three sections of each 30-m ring: (1) aspen (*Populus tremuloides* Michx.); (2) aspen and birch (*Betula papyrifera* Marsh.); and (3) aspen and maple (*Acer saccharum* Marsh.). Fumigation began in May 1998. In this investigation we focus only on soil carbon in the aspen and aspen–birch stands. Soils were sampled in 1997 before the start of fumigation with elevated CO₂ and O₃, and again in October 2001 when trees had grown to approximately 5 m height, forming a closed-canopy system in all but the elevated O₃ treatment where trees were smaller with less leaf production. Soils at the site are mixed, frigid, coarse loamy Alfic Haplorthods. The texture of the soil is sandy loam, underlain by a clay plough-layer, and then grading back into a sandy loam.

Soil carbon fractionation

Soil cores (10 cm diameter × 20 cm deep) were collected at 10 random locations within each community type in each plot, composited by community type within each plot, sieved to remove roots and other coarse materials, and oven-dried before analysis for total soil carbon and δ¹³C. Subsamples (40 g) of each soil were extracted with 150 ml of deionized water overnight to remove water-soluble carbohydrates. We then removed any remaining fine organic matter on a portion of the residual soil by flotation with 1.0 M NaCl, rinsing with deionized water, and then manually removing any remaining particulate organic matter (POM) under a compound microscope at 20 × magnification. We then performed acid hydrolysis with 6 M HCl on a 2-g subsample of POM-free soil under reflux for 8 h to isolate the most decay-resistant carbon for %C and δ¹³C. Analyses of %C and δ¹³C were performed on a Costech Elemental Combustion System 4010 connected to a ThermoFinnigan ConFloIII Interface and DeltaPlus Continuous Flow-Stable Isotope Ratio Mass Spectrometer (IRMS).

Soil incubation and respiration measurements

A separate subsample of sieved root-free soil was used to determine potential carbon availability by aerobic incubation. We placed 100 g of field moist soil into microlysimeters constructed from funnels, brought soils to approximately 60% field capacity, then placed soils into 1-litre incubation jars which were placed in an incubator at 25.0 ± 0.5 °C. At intervals that increased in length over time, we measured CO₂ accumulation in the headspace by withdrawing 50 μl from sealed jars fitted with septa and measured the

concentration using a Varian CP-3800 gas chromatograph with a thermal conductivity detector. Rates of CO₂ production were calculated by dividing concentrations by the time since jars had been sealed. On the same date, we collected samples of gas from the headspace of incubation jars for determining the $\delta^{13}\text{C}$ value of the accumulated CO₂. We injected 7 ml of headspace gas into He-flushed gas-tight LabCo exetainers, which were then placed into an autosampler and analysed on a FIN-MAT Gas Bench II connected to the IRMS. Soil respiration $\delta^{13}\text{C}$ values for control and elevated O₃ treatments were $-25.0 \pm 0.2\text{‰}$ across the incubation. For soils from elevated CO₂, soil respiration values initially were $-30.9 \pm 0.3\text{‰}$ and increased to $-28.8 \pm 0.2\text{‰}$ by day 281 of incubation. For the elevated O₃ + CO₂ treatment, values initially were $-31.1 \pm 0.4\text{‰}$ and -28.2 ± 0.2 at day 281 of incubation.

Stable isotope tracer, calculations and statistical analysis

To investigate the contribution of new carbon inputs to soil and microbial respiration in the elevated CO₂ and elevated O₃ + CO₂ plots, we used the $\delta^{13}\text{C}$ signature derived from the fossil-fuel fumigation gas mixed with ambient air ($-18.7 \pm 1.0\text{‰}$ compared to $-8.6 \pm 0.1\text{‰}$ for control and elevated O₃ plots). Subsequent fractionation by the plants produced leaf and root tissue with $\delta^{13}\text{C}$ values of approximately $-41.6 \pm 0.4\text{‰}$ in elevated CO₂ and elevated O₃ + CO₂ plots. Composite soil samples collected from all 12 plots before the experiment began in 1997 and from the three control plots in 2001 had similar ($P = 0.47$) $\delta^{13}\text{C}$ values, averaging $-26.7 \pm 0.2\text{‰}$. The proportional contribution of carbon derived from the fumigation gas (f) was calculated using the equation $f = (\delta_i - \delta_o)/(\delta_i - \delta_o)$, where δ_i refers to the isotopic composition ($\delta^{13}\text{C}$) of the soil organic carbon or soil respiration from the fumigated (elevated CO₂ or elevated O₃ + CO₂) plots, δ_o is the $\delta^{13}\text{C}$ of soil organic carbon or soil respiration from control plots, and δ_i is the $\delta^{13}\text{C}$ signature of the plant leaves and roots collected in 2001 from the fumigated plots, weighted equally and averaged across species as there were no significant differences found among the aspen and aspen–birch plots. We do not expect that the isotopic signature of the plants varied appreciably over the life of this experiment because fumigation was initiated when the trees were still seedlings. Because there were differences among treatments in the amount of carbon entering soils derived from fumigation (Fig. 1), we calculated the relative mass loss as the cumulative amount of carbon respired from each of these pools divided by the amount of soil carbon in that pool. Data were analysed using analysis of variance with general linear models for a split-plot randomized complete design using SAS 8.02 (Cary, North Carolina).

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Flying and swimming animals cruise at a Strouhal number tuned for high-power efficiency

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Dimensionless numbers are important in biomechanics because their constancy can imply dynamic similarity between systems, despite possible differences in medium or scale¹. A dimensionless parameter that describes the tail or wing kinematics of swimming and flying animals is the Strouhal number¹, $St = fA/U$, which divides stroke frequency (f) and amplitude (A) by forward speed (U)^{2–8}. It is known to govern a well-defined series of vortex growth and shedding regimes for airfoils undergoing pitching and heaving motions^{6,8}. Propulsive efficiency is high over a narrow range of St and usually peaks within the interval $0.2 < St < 0.4$ (refs 3–8). Because natural selection is likely to tune animals for optimal propulsive efficiency, we expect it to constrain the range of St that animals use. This seems to be true for dolphins^{2–5}, sharks^{3–5} and bony fish^{3–5}, which swim at $0.2 < St < 0.4$. Here we show that birds, bats and insects also converge on the same narrow range of St , but only when cruising. Tuning cruise kinematics to optimize St therefore seems to be a general principle of oscillatory lift-based propulsion.

Experiments with isolated pitching or heaving foils have measured extremely high peak propulsive efficiencies within the interval $0.2 < St < 0.4$ (modal peak at $St \approx 0.3$)^{3–7}. In this range, the propulsive efficiency (defined as the ratio of aerodynamic power output to mechanical power input) can be as high as 70% (ref. 7) or even 80% (ref. 6). Optimal St depends subtly on kinematic parameters including geometric angle of attack, amplitude-to-chord ratio, airfoil section and phase of motion^{6–8} but, for any given motion, efficiency is usually high ($>60\%$) over a range narrower than $0.2 < St < 0.4$ (refs 3–7). For example, measured efficiency can plummet from 80% at $St = 0.27$ to 10% at $St = 0.09$ (ref. 6) and also drops off at higher St , albeit more gently^{7,8}. Measured propulsive efficiency usually peaks when the kinematics result in maximum amplification of the shed vortices in the wake and an average velocity profile equivalent to a jet^{3,4,6}.

Theoretical treatments of flapping wings^{3,4,6,8} further confirm the empirical result that St tightly constrains propulsive efficiency. In fact, St is bound to affect aerodynamic force coefficients and propulsive efficiency, because it defines the maximum aerodynamic angle of attack and the timescales associated with the growth and shedding of vortices, which are the source of aerodynamic force production^{8,9}. Natural selection is expected to favour wing kinematics that combine high propulsive efficiency with a high aerodynamic force coefficient. These need not peak at identical St , but can do for certain motions^{7,8}: if not, selection should optimize the trade-off. Propulsive efficiency may be the more important selection pressure in cruising, whereas high aerodynamic force coefficients may be more important in accelerations, slow locomotion or hovering. In cruising flight or swimming, we therefore predict that St will be tuned for high propulsive efficiency.

This suggestion has already been made for cruising fish and dolphins^{2–5}, which operate within the range $0.2 < St < 0.4$, and the principle is considered so general for swimming animals that it has even been used to predict the speeds of extinct ichthyosaurs¹⁰. Whereas the fluid dynamic results described above^{3–8} refer to