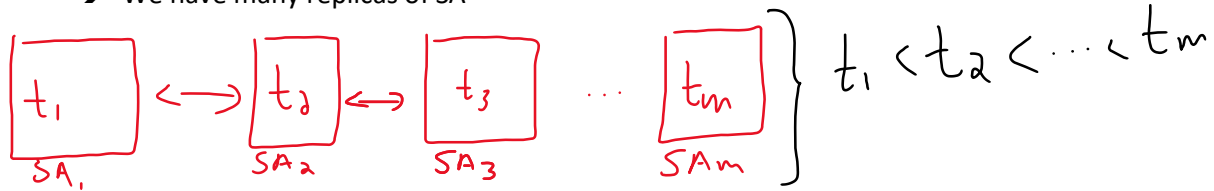


28- The parallel tempering.

- Parallel version of SA
- We have many replicas of SA



- As opposed to standard SA, here the temperature of each S_{ai} is constant (might change just a bit), all these S_{ai} run in parallel. They will interact by exchanging configurations (current solution)
- Neighbouring systems can exchange their current solution according to a probability law
- Let's consider 2 systems, i and $j=i+1$ (neighbours) with temperatures T_i and T_j , and current solution C_i and C_j , with energy E_i and E_j

$$p_{ij} = \min(1, e^{-\Delta E_{ij}}) \rightarrow \Delta E_{ij} = (E_i - E_j) \cdot \left(\frac{1}{T_j} - \frac{1}{T_i}\right)$$

$$p_{ij} = p_{ji}$$

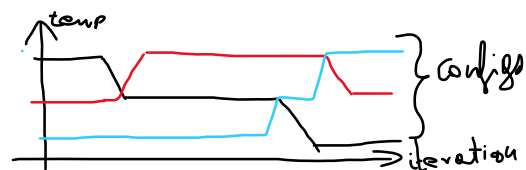
- If $T_i < T_j$ and $E_i > E_j \rightarrow \Delta E_{ij}$ is negative, and $p_{ij} = 1$
- What this means, is that if we have a good solution at high temperature, it will always exchange with a less good solution at lower temp
- The opposite is also possible, a good solution in a high temp can also go to a higher temp, with $p < 1$

! in parallel tempering the temperature schedule is distributed across the replicas of SA, and not over iterations !

- At low temperature we exploit the solution
- At high temp we explore other solutions and other regions of S
- Exchanging configuration is a way to combine diversification and intensification

GUIDING PARAMETERS

It obviously depends on how we define our parameters



- How many SA? M is often: $M = \sqrt{N} \rightarrow N$ is the problem size
- At what frequency should we consider exchange of configuration? When both systems reach an equilibrium state, according to the SA definition
- What is the range of temperature between T_1 and T_M ? T_1 should be small enough to allow the convergence, T_M should be large enough to allow exploration ($T_M = T_0$ of standard exploration)
- We can fine tune the temperature of each system over time, if the exchange rate is too large $> 2\%$ we increase the temperature difference between all systems
- If the exchange rate is too low $< 0.5\%$ all temperature difference between the systems is decreased