

### Offline #3

#### Topic: Integration

In a methanol-based fuel cell, DC electricity is generated from the combustion of methanol and oxygen. In such a fuel cell, the consumption of oxygen over time can be expressed using the following formula:

$$T = -\int_{x_1}^{x_2} \left( \frac{6.73x + 6.725 \times 10^{-8} + 7.26 \times 10^{-4} C_{me}}{3.62 \times 10^{-12} x + 3.908 \times 10^{-8} x C_{me}} \right) dx$$

Where

$T$  = time in seconds

$x$  = concentration of oxygen in  $\text{moles/cm}^3$

$C_{me}$  = concentration of methanol in  $\text{moles/cm}^3 = 5 \times 10^{-4} \text{ moles/cm}^3$

The initial concentration of oxygen is:  $x(t = 0) = 1.22 \times 10^{-4} \text{ moles/cm}^3$

Programming tasks:

1. Write a python program to evaluate the time required for 50% of the initial oxygen concentration to be consumed in the fuel cell in the presence of methanol. Your program should accept the number of sub-intervals  $n$  as a parameter from the user and use the trapezoid rule by partitioning the given interval into  $n$  equally spaced sub-intervals. **Print the calculated values for  $n = 1$  to 5 and show the absolute approximate relative errors.** (10 Marks)
2. Solve the same problem given in (1) **using Simpsons' 1/3 rule.** For this case, partition the given interval into  $2n$  number of equally spaced sub-intervals and use the single application of a Simpson's rule  $n$  times. Print the calculated values for  $n = 1$  to 5 and show the absolute approximate relative errors. (10 Marks)
3. Plot time vs. oxygen concentration for the following concentrations of oxygen  $x = [1.22, 1.20, 1.0, 0.8, 0.6, 0.4, 0.2] \times 10^{-4} \text{ moles/cm}^3$ . Use Simpsons' 1/3 rule with 10 sub-intervals. (5 Marks)