

Biology (130 MARKS)
Answer each of the questions 1, 2 and 3.

Question 1. (52 Marks) All Items, (a), (b), (c), etc. (7 × 6 + 1 × 10marks)

- | | | |
|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|-------------------|---------------------|
| (a) any two from: water/ salt/ urea... | (2 × 3) | [6] |
| (b) motor function, any one from: carry messages from the brain(spinal cord) (CNS)/ to muscles (effectors) (glands)
sensory function, any one from: carry messages to the brain(spinal cord) (CNS)/ from sensory organs (sensors) (ear) (eye) (nose) (skin) tongue) | (3)

(3) |

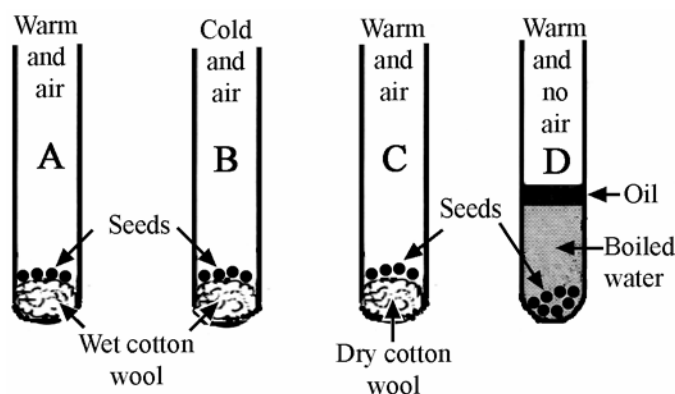
[6] |
| (c) fused/ fixed/ immovable/
no movement | (3)
(3) |
[6] |
| (d) A: lens
any one from: controls (changes) (alters) the shape (thickness) of the lens/ focuses | (3)
(3) |
[6] |
| (e) any one from: nucleus/ chromosomes/ genes/ mitochondria/ plasitds
protein | (3)
(3) |
[6] |
| (f) put tube B over the small animal
suck tube A | (3)
(3) |
[6] |
| (g) any two from production of: bread/ beer (lager) (stout)/ miso/ vinegar/ soy sauce/ wine/yoghurt/ antibiotics (penicillin)/ hormones (insulin)/ vaccines/ interferon/ biological detergents/ pesticides/ GM (genetically modified organisms)/ bio-fuels (methane).../ spirits (vodka)... | (2 × 3) | [6] |
| (h) Fat
test: rub food onto paper
translucent (greasy) spot
note: if a wrong food type is given by the candidate e.g. ‘starch’ then they lose the first 4 marks. If they then give the correct test for the food that they have named e.g. ‘iodine’ and ‘blue-black’ award the (2 × 3) | (4)
(3)
(3) |

[10] |

Question 2. (39 marks) All items, (a), (b) and (c).

- (a) (i) List **any three from:** water (moisture)/ oxygen (air)/ suitable temperature (warm) (heat)/ period of dormancy over/ light (3 × 3) [9]

- (ii) Describe assume that seed dormancy is over, the investigation deals with environmental factors (conditions) only
A has water, oxygen (air) and warm (15-20°C)
B has water, oxygen (air) and cold (4°C)
C has **no water**, oxygen (air) and warm (15-20°C)
D has water, **no oxygen (air)** and warm (15-20°C)
note 'cold' alone gets no marks, refrigerate/ 4°C is needed



show or state

A is required in all answers
 only seeds in A germinate

any two from B, C, or D can act as a **suitable control**
 (matched) for the **two conditions** selected by the candidate.

note only two conditions are investigated. In their answer the candidate may **not list** all three e.g. if water and air are selected and A, C and D are given and 'warm' is not given award the marks.

if 'light' is a selected condition then a **way** of keeping seeds light-free, from the start, is required in the answer
[no diagram deduct 3 marks]

- (b) (i) Give **any one from:** produces bile / helps digest (breakdown) (emulsify) fats or a named fat (3) [3]

- (ii) Identify **enzyme:** amylase (3)
substrate: starch (3) [6]

- (iii) Give **any one from:** absorption/ food (nutrients) into the blood/ transports (3) [3]

- (iv) Describe burn a peanut (crisp) (cracker) (3)
 temperature of water rises/ thermometer shows increase (3) [6]
accept equivalent experiments

Question 3. (39 marks) All items, (a) and (b).

- (a) (i) Complete **missing reactant:** oxygen (O₂) (3)
 missing product: carbon dioxide (CO₂) (3) **[6]**
- (ii) State **test for CO₂:** CO₂ (carbon dioxide) (3)
 lime water (3)
 goes milky (3)
 or **or**
 test for water: cobalt chloride/ copper sulphate (3)
 add water (3)
 turns pink/ turns blue (3) **[9]**
 note: colour must be matched with reagent
- note:** if the candidate gives oxygen, O₂, as a **product** in (i) they get no marks for it in (i), **allow** in (ii) (3) for 'relights' and (3) for 'glowing splint' i.e. in this case test for O₂ can get (2 × 3)
- (b) Name oxygen (O₂) (3) **[3]**
- Name carbon dioxide (CO₂) (3) **[3]**
- How? brighter (more) light/ increase (more) CO₂ (3) **[3]**
- (c) (i) Explain burning releases carbon dioxide (CO₂) (3)
 or **or**
 any one from: deforestation results in less carbon dioxide (CO₂) being used/ less photosynthesis/ burning forests releases CO₂ (3) **[3]**
- (ii) Suggest **any one from:** acidification of rivers (lakes) (seas) / melting of polar ice (glaciers)/ rise in sea levels / drought/ greenhouse effect/ disruption of aquatic food chains/ climate change (hotter) (colder) (more or less rain) (more or less wind)/ changes in ocean currents/ extinction of species/ global warming... (3) **[3]**
 note: 'damages the ozone layer' gets no marks
- (iii) Suggest **any one from:** carbon dioxide (CO₂) is taken in by plants/ used in photosynthesis/ more leaves/ less fuel is burnt for heating... (3) **[3]**
- (iv) How? **Allow any one from:** plant more trees (reforestation)/ by increasing photosynthesis/ by increasing photosynthesis/ burn less fuel/ turn down thermostats/ better heat insulation of houses (correct example)/ heat pump/ bio-fuel/ solar panels... (6) **[6]**
 note allow 6 marks for 'burn less fuel only it has not appeared in (iii) above

Chemistry (130 MARKS)
Answer each of the questions 4, 5 and 6.

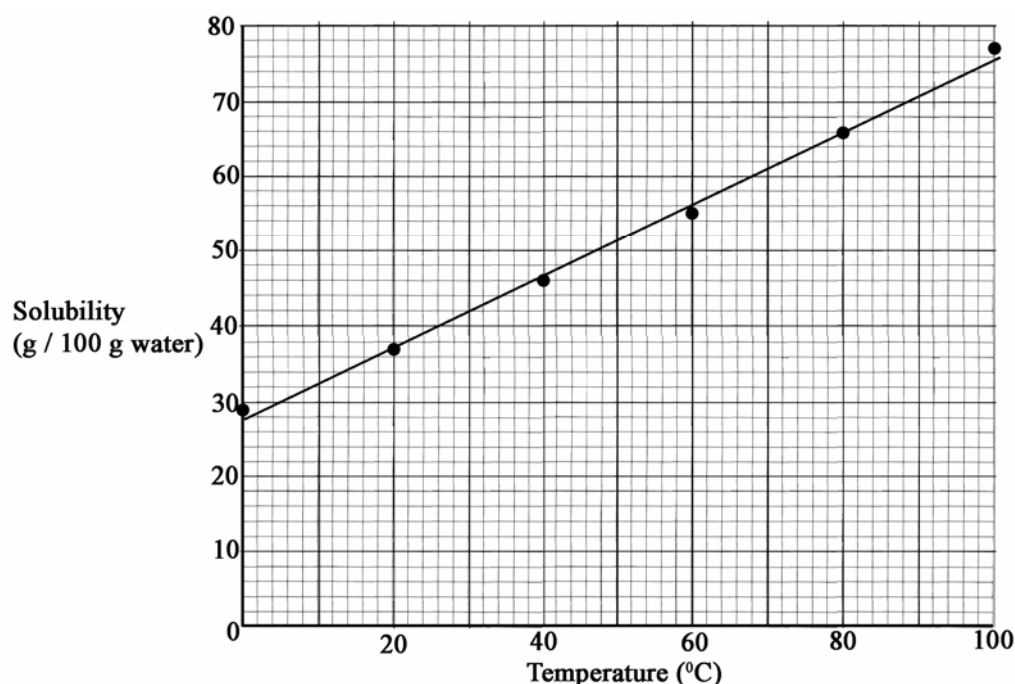
Question 4. (52 marks) All items, (a), (b), (c), etc. (7 × 6 + 1 × 10marks)

- (a) two dots (**Xs**) in inner circle and eight dots (**Xs**) in ‘middle’ circle (3)
 one dot (**X**) in outer circle (3) **[6]**
- (b) oil (3)
any one from: will not rot (decay)/ bacteria (fungi) (microbes) (micro-organisms) cannot break them down (3) **[6]**
accept: can not be decomposed (broken down) for 3 marks
- (c) graduated (measuring) cylinder (3)
any one from: burette/ pipette/ gas (graduated) syringe (3) **[6]**
- (d) **any two from:** soft (can be cut with a knife)/ low density(float on water)/ shiny (lustrous) when cut/ tarnish(form oxide) (reacts) with air/ burn in air/ coloured flames/ react with water/ hydrogen produced with water/ hydroxides formed with water/ silver’ (white metals)/ very reactive/ one electron in outer orbit... (2 × 3) **[6]**
- note** general properties of most metals like electrical and thermal conductivity, ductility, malleability, etc get no marks, specific properties of **alkali** metals are required.
- (e) shared electrons (3)
 (3) **[6]**
- (f) **any one from:** alum/ copper sulphate/ salt/ sugar/ silicon/ iodine/ diamond... (3)
any one from: crystalline solids have definite (geometric) shapes/ particles in crystals are in regular(geometric) order/ texture e.g. salt and sugar ‘feel’ different to flour/ crystals affect light/ shiny / sparkle... (3) **[6]**
- (g) **any one from:** Ca(OH)₂/ CaCO₃/ NaOH/ Na₂CO₃/ NaHCO₃/ NH₃... (3)
any one from: sodium hydroxide (caustic soda)/ sodium carbonate (washing soda) sodium hydrogen (bi) carbonate (bread soda)/ ammonia... (3)
accept any one from: the named household substances: toothpaste/ oven cleaner/ antacid (named antacid)/ baking powder... (3) **[6]**
do not accept: bleach/ shampoo
note: accept names **only** for the second 3 marks
- (h) **A** condenser (3)
Y (3)
 Water/ H₂O (2)
 salt/ sodium chloride/ NaCl... (2) **[10]**

Question 5. (39 Marks) All items, (a), (b), (c), etc.

- (a) Distinguish **any *one* from: concentrated** a lot in a small amount of solvent (water)/
dilute a small amount in a lot of solvent (water) (3) [3]
allow 3 marks for concentrated can be made dilute by adding solvent (water)

Plot



suitable scales drawn on both axes	(3)	
points plotted (a minimum of four points)	(3)	
line drawn (it is actually a curve, so accept a curve or a straight line)	(3)	[9]

Use 61 **accept** 58 to 64 i.e. +/- 3 (3) [3]

What? **accept** solubility is proportional to temperature for 3 marks solubility increases with temperature (3) [3]

- | | | | | |
|--------------------|-----------------------------------------------------------------------------------------|---------------------------------------------------------------|-----|-------------|
| (b) <u>Explain</u> | element: allow (3) for ‘only one type of atom’ or ‘can’t be made simpler’ | element: cannot be broken down into simpler substances | (3) | |
| | | | (3) | |
| | | compound: two or more elements chemically combined | (3) | |
| | allow (6) for all atoms have the same atomic number (nuclear charge) | | (3) | [12] |

- (c) Describe **any one from:** pH (universal) indicator/ pH meter (probe) (3)
 any one from: match colour/ read pH (3) [6]

Name **any *one* from:** battery acid/ sulphuric acid/ vinegar/ lemon (orange) juice/ sour milk/ carbonated drinks (named carbonated drink)... (3) [3]

Question 6. (39 marks) All items, (a), (b) and (c).

- (a) Give HCl, **accept** formulae(s) of other acids (3) [3]
- Give calcium carbonate
accept any named carbonate **or** bi(hydrogen) carbonate (3) [3]
- What? denser (heavier) than air (3) [3]
- What? both red (pink)/ blue to red (pink) (3) [3]
- Give **any two from:** fire extinguishers/ fizzy drinks/ photosynthesis/ 'dry ice'/ 'stage effects'... (2 × 3) [6]
- (b) (i) Describe **test:** add soap (3)
shake (3)
result: any one from: water in flask **A** (hard water) does not form a lather easily with soap / forms a scum with soap (3)
any one from: flask **B** contains water that forms a lather easily with soap/ does not form a scum with soap/ contains soft water (3) [12]
- (ii) What? **any one from:** compounds of calcium (magnesium)/ calcium **or** magnesium ions (salts)/ formulae(s) **or** names of compounds of calcium and magnesium, **except carbonates**, e.g. CaCl₂, Ca(HCO₃)₂, calcium chloride, calcium hydrogen carbonate.../ lime (3) [3]
- (c) (i) Name **any one from:** screening/ settling/ filtration/ UV 'light'/ chlorination/ fluoridation/ pH adjustment/ softening/ distillation... (3) [3]
- (ii) Give **any one from (matched):**
Screening: takes out large solids
settling: solids sink
filtration: solids are removed
UV (ultraviolet) irradiation: stops certain dangerous microbes multiplying
chlorination: to kill bacteria (microbes)
fluoridation: help prevent tooth decay
pH adjustment: stop damage to metal pipes
softening: removing dissolved calcium/ magnesium
distillation: removing dissolved solids (3) [3]

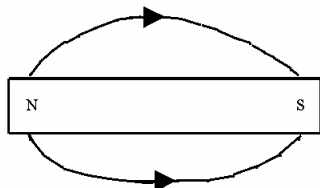
Physics (130 MARKS)
Answer each of the questions 7, 8 and 9.

Question 7. (52 marks) All items, (a), (b), (c), etc. (7 × 6 + 1 × 10marks)

- (a) (i) heat the ball, it does not pass through the ring (3)
(ii) let the ball cool, now passes through the ring (3) **[6]**
- (b) (i) ice is less dense than water (3)
(ii) ice is more dense than ethanol (3) **[6]**
- (c) bulb **A** lights (3)
the diode with **A** is in forward bias (allows current to flow)
(+ end of diode connected to + pole of battery) (3) **[6]**
note allow 3 marks for a correct reason for why **B** does not
light if a correct reason for why **A** does light is not given
- (d) **magnetic effect any one from:** electromagnets/ door bells/ remote opening of
doors/central locking in cars/ phones/ speakers/ electric motors/ any named
device incorporating an electric motor/ read (write) heads on audio (video)
recorders computer drives/ remote switching e.g. starter motor in car
electrical meters... (3)
chemical effect any one from: electroplating/ chromium plating/ silver
plating/cold galvanising/ refining of copper/ refining of aluminium/
production of sodium hydroxide/ production of hydrogen/
charging a battery... (3) **[6]**
accept 'electrolysis' for 3 marks
- (e) light moves faster than sound (6) **[6]**
- (f) **advantage any one from:** no 'greenhouse gas' emissions/ no carbon dioxide
(CO₂) emissions/ supply of nuclear fuel for many centuries/
can produce large amounts of energy/ cleaner supply of electricity... (3)
disadvantage any one from: devastation if a reactor erupts/
very long term storage of wastes/ wastes are radioactive(dangerous)/
mining for nuclear fuel damages the environment /
transport of nuclear fuel is hazardous/ terrorist threat
reprocessing nuclear fuel is hazardous (can damage the environment)/... (3) **[6]**
- (g) water is a poor conductor of heat (3)
any one from: hot water rises/ water is heated by convection(current) (3) **[6]**
note 'heat rises' alone gets no marks
- (h) **weather conditions:**
any two from: cloudy/ windy/ rain... (2 × 3)
explanation:
any one from: air rises/ water vapour condenses (cools)/
air moves in (4) **[10]**

Question 8. (39 marks) All items, (a), (b), (c), etc.

(a) Draw



two magnetic field lines shown, **one on each side** of the bar magnet.
[if direction is not given or given incorrectly for a magnetic field line award no marks for that 'line']

(2 × 3) [6]

What?

poles/ north and south

(3) [3]

(b) Calculate

note

(i) if no calculation is performed allow 3 marks for 'work = force × distance', award no marks for units in this case.

(ii) allow 2 marks for (8 × 20) or (8000 × 20), if the units are matched allow the 3 marks for the unit

(iii) unit alone gets no marks

160 000

J (Nm)

or

160

kJ (kNm)

(3)

(3)

or

(3)

(3)

[6]

Identify

any one from: kinetic to heat/ kinetic to sound/ kinetic to electrical/ kinetic to chemical/ kinetic to potential

note: the last three are included to allow for 'hybrid' cars

[allow 3 marks for a correctly named energy provided it is on the correct side of 'to' or of an arrow]

(2 × 3) [6]

(c) Define

accept turning effect or turning power for 3 marks

force (weight) multiplied by its distance from the fulcrum

(3)

(3)

[6]

Calculate

$30 \times F = 40 \times 3$

$F = 4 \text{ N}$ ('N' not required)

note (i) allow 5 marks for $(40 \times 3)/30$ if it is the only statement

(ii) '4' alone merits 6 marks

(3)

(3)

[6]

Give

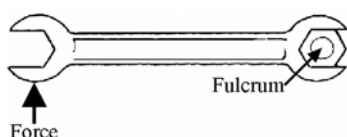
e.g. spanner and nut
accept any correct
everyday example of a lever

fulcrum shown correctly
one force shown correctly
[no diagram deduct 3 marks]

(3)

(3)

[6]



note: if fulcrum **and** force are not shown in the diagram allow (3) for a correct example of an application of a lever shown

note if no diagram is given allow 3 marks for **any one from:** correct named application e.g. 'door'/ correct fulcrum named e.g. 'hinge'/ correct position of a force e.g. 'handle'

Question 9. (39 marks) All items, (a) and (b).

(a) (i) What? refraction (3) [3]

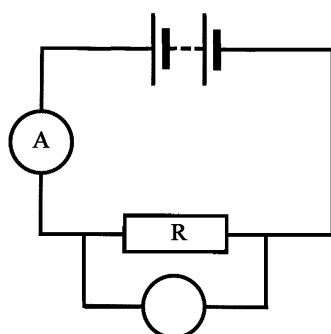
(ii) Pick **R** (3) [3]

(iii) Give **any one from:** lenses/ spectacles/ magnifying glass/
microscope/ binoculars/ telescopes/ camera lenses/
prisms/ projectors/ dispersion (rainbow) (spectrum)/ diascopes/
endoscope/ periscope/... (3) [3]

(iv) Name **any one from:** reflection (bouncing off) (mirror)/ gravity (3) [3]

(b) What? current/ Amperes (Amps) (3) [3]

Enter



A correctly shown in series with **R** and the battery

(3) [3]

Use

12, **accept** 11.5 to 12.5 i.e. +/- 0.5
Ω or Ohms

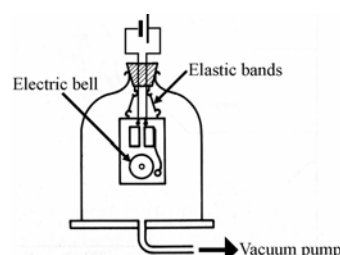
(6)

(3) [9]

allow 3 any correct ratio from the graph e.g. $\frac{1.2}{0.1}$ **or** $\frac{1.8}{0.15}$ etc

note if an incorrect ratio is given but it is used correctly in a calculation allow 3 marks e.g. $\frac{6}{5} = 1.2$

(c) Describe



show or state

battery/ cell/ source of electricity

(3)

bell in jar (container)

(3)

vacuum pump/ air removed

(3)

bell is less loud (can't be heard)

(3)

[no diagram deduct 3 marks]

[12]

acceptable sources of sound include: mobile phone, alarm clock, kitchen timer. Look for an equivalent point e.g. 'dial' for the phone for the first 3 marks

[if the candidate does not score 12 marks for the above but has 'elastic bands'/ sound insulation between bell and container award 3marks]

accept equivalent experiments, look four equivalent points