

Biology (130 MARKS)
Answer each of the questions 1, 2 and 3.

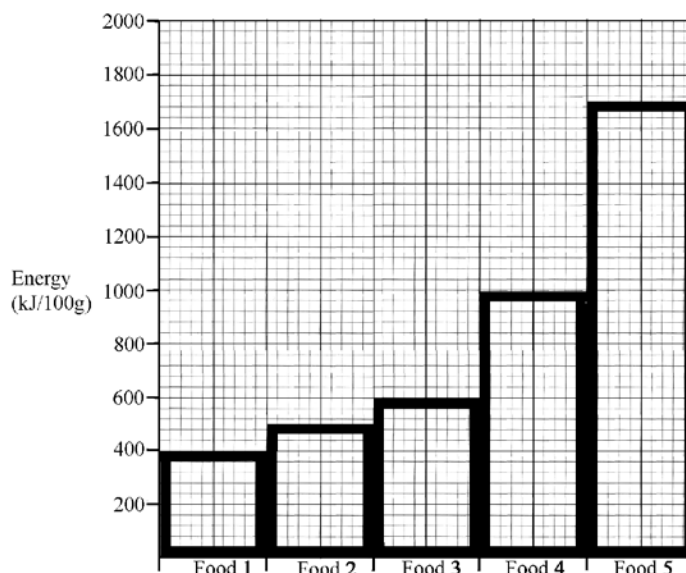
Question 1. (52 Marks) All Items, (a), (b), (c), etc. (7 × 6 + 1 × 10marks)

- (a) **any two from:** gills/ fins/ scales/ spines/ shape (streamlined)/ tail/ colour (camouflage) / lateral line... (2 × 3) [6]
- (b) **any two from:** urea/ water/ salts/... (2 × 3) [6]
accept urine for (3) if it is the *only* answer given
- (c) **any two from:** brain/ eyes/ ear/ semicircular canals (organ of balance) / tongue... (2 × 3) [6]
- (d) **A:** cell wall (3)
B: nucleus (3) [6]
- (e) **beneficial any one from:** decomposition/ decay/ food/ yoghurt/ vinegar/ cheese/ food supplements/ biotechnology/ insulin/ interferon/ healthy gut/ vaccination/ helps immune system/ antibiotics/ medicine / silage... (3)
harmful any one from: disease/ TB/ pneumonia/ meningitis/ tetanus/ cholera/ anthrax/ food poisoning/ tooth decay/ sore throat/ pimples... (3) [6]
- (f) DNA (3)
protein (3) [6]
- (g) (i) **what?:** phototropism (3)
(ii) **what?:** make more food/ more photosynthesis/ absorb more light... (3) [6]
- (h) **candidate must clearly state names/ formulas of gases and directions of movement.**
oxygen/ O₂ (2)
into bloodstream/ out of alveoli (3)
carbon dioxide/ CO₂ (2)
out of bloodstream/ into alveoli (3) [10]

Question 2. (39 marks) All items, (a), (b) and (c).

- (a) (i) What? breakdown of food (3) [3]
- (ii) Why? make food soluble/ food can enter bloodstream/ to obtain nutrients... (3) [3]
- (iii) Name **A:** liver (3)
B: pancreas (3) [6]
- (iv) Give **any one from:** kills bacteria/ digestion/ liquefies food/ mixes food/ produces HCl/ produces enzymes/ produces chyme... (3) [3]
- (v) Give **any one from:** absorb water/ form faeces/ store (transport) (expel) faeces... (3) [3]

- (b) (i) Draw



five bars correct

(9)

or

four bars correct

or

(6)

or

three bars correct

or

(3)

Tolerance $\frac{1}{2}$ ☐

[9]

- (ii) Which?

fat

(3)

What?

cheese has most fat/ more fat than other foods

(3)

[6]

- (iii) Describe

rub food on paper

(3)

translucent spot (translucent mark) (translucent stain)

(3)

[6]

Question 3. (39 marks) All items, (a) and (b).

(a) (i) Why? **any one from:** destarch leaves/ starch goes from leaves (3) [3]

(ii) Why? **any one from:** kill/ soften (3) [3]

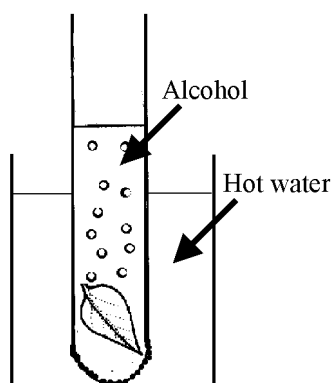
(iii) Draw

alcohol labelled correctly

(3)

hot water labelled correctly/
alcohol being heated

(3) [6]



[no diagram deduct 3 marks]

(iv) Name iodine (3) [3]

(v) Suggest **any one from:** no starch/ no photosynthesis/ no chlorophyll (green pigment) (3) [3]

(b) (i) Name ovary (3)
Role contains ovules/ egg(s)/ female gamete(s)/ embryo(s)/ seed(s) (3) [6]

(ii) Name anther (3)
Role produces pollen/ male gametes/ sperm (3) [6]

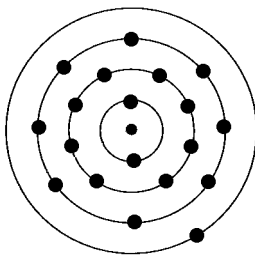
(iii) Give **any one from:** insects/ wind/ named insect (bee) (fly)/ water... (3) [3]

(iv) Name Zygote (3) [3]
accept: fertilised egg

(v) What? **any one from:** embryo/ seed/ plant (3) [3]
accept: fruit

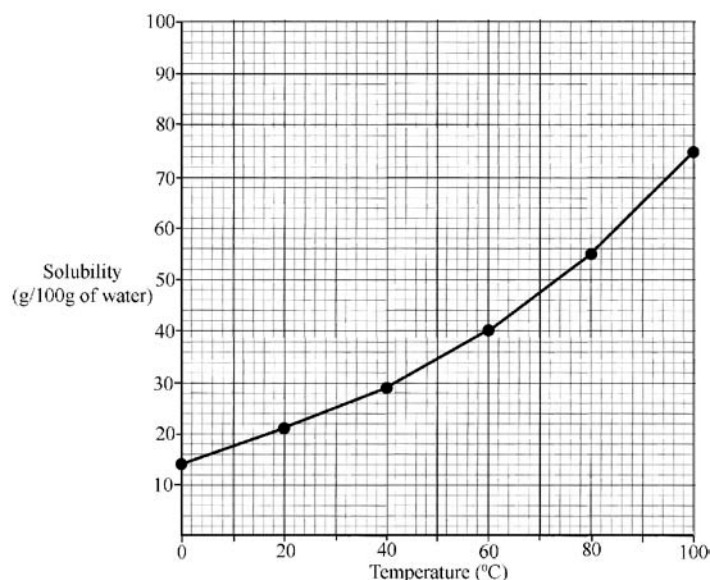
Chemistry (130 MARKS)
Answer each of the questions 4, 5 and 6.

Question 4. (52 marks) All items, (a), (b), (c), etc. (7 × 6 + 1 × 10marks)

- (a) **any two from either list or one from each list:** Element Mg: is 'silver' / metal / shines / bends / ductile / malleable / reacts with acids releasing hydrogen / conducts electricity / conducts heat / burns in air (Oxygen) ...
Compound Mg O: white / powder / base / does not burn / does not conduct electricity / does not conduct heat ... (2 × 3) [6]
note: two different properties must be given to merit (2 × 3),
assume that both answers refer to magnesium if the candidate does
not specify to which substance the properties given are assigned.
- (b) **any one from:** dissolves / erodes / corrodes / wears away (3)
any one from: limestone is calcium carbonate (CaCO₃) / chemical reaction (3) [6]
- (c) (i) **A** sand (3)
(ii) **B** water / salt (3) [6]
- (d) two electrons in first orbit and eight in second orbit shown (3)
eight electrons in third orbit and one in fourth orbit shown (3) [6]
2, 8, 8, 1 with electrons not shown in diagram (3) only
- 
- (e) soft (3)
only water in **B** / dissolved substances (solute) remains in **A** / hardness removed by distillation (3) [6]
- (f) **any two from:** fluoridation / chlorination / filtration / screening / settling / ultra violet (UV) / adjust pH / flocculation / ion exchange / boiling ... (2 × 3) [6]
- (g) **any two from:** electricity / heat / sound (2 × 3) [6]
- (h) (i) **A** (2)
(ii) to remove air (oxygen) (2)
(iii) to keep air (oxygen) out (2)
(iv) air (oxygen) is needed for rusting (4) [10]
accept air (oxygen) and water for (4) in (iv)
water alone zero in (iv)

Question 5. (39 Marks) All items, (a), (b), (c), etc.

(a) (i) Draw



(6)

six points plotted correctly
smooth curve (**accept** points joined by straight lines)
through all six points
allow (3) for four correct points
Tolerance $\frac{1}{2}$ ☐

(3) [9]

(ii) Use

15-18

(3) [3]

(iii) Describe

show or state: (Marks are awarded *only* for a diagram that is correct in context of the experiment described by the candidate.)

leave/cool

crystals form

filter/ evaporate

suitable diagram

or

crystal on string in solution

crystal grows

remove crystal (using string)

suitable diagram

or

heat solution

evaporate water

crystals form

suitable diagram

(3)

(3)

(3)

(3)

or

(3)

(3)

(3)

(3)

or

(3)

(3)

(3)

(3)

[12]

(b) (i) What?

0 – 14

(3)

any one qualification from: shows degree of acidity/ measures acidity/ shows degree of alkalinity (basicity)/ measures alkalinity (basicity)/ pH < 7 acid/ pH = 7 neutral/ pH > 7 alkali (base)

(3)

How?

any one from: pH paper/ pH meter/ pH probe / universal indicator

(3)

[9]

(ii) Name

gastric juice

(3)

blood

(3)

[6]

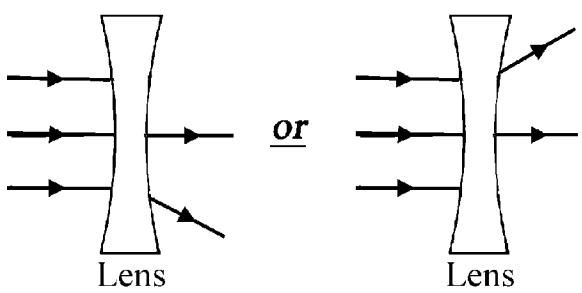
Question 6. (39 marks) All items, (a), (b) and (c).

(a)	(i) <u>Name</u>	electrolysis	(3)	[3]
	(ii) <u>Why?</u>	conductivity	(3)	[3]
	(iii) <u>Name</u>	A oxygen/ O ₂ (atomic symbol gets no marks)	(3)	
	<u>Give</u>	relights glowing splint	(3)	[6]
	(iv) <u>Name</u>	B hydrogen/ H ₂ (atomic symbol gets no marks)	(3)	
	<u>Give</u>	burns with a pop	(3)	[6]
		if names of gases <i>and</i> tests are <i>both</i> correct and <i>matched</i> but ‘reversed’ i.e. hydrogen for A (iii) and oxygen for B (iv) allow (2 × 3)		
	(v) <u>What?</u>	A:B = 1:2/ B:A = 2:1/ H₂O... accept: 1:2 alone for (3)	(3)	[3]
(b)	(i) <u>Name</u>	hydrogen/ H ₂ (atomic symbol gets no marks)	(3)	[3]
	(ii) <u>Name</u>	hydrochloric (HCl)/ sulphuric acid (H ₂ SO ₄)...	(3)	[3]
	(iii) <u>Name</u>	calcium (atomic symbol gets no marks, name only)	(3)	[3]
	(iv) <u>Name</u>	copper (atomic symbol gets no marks, name only)	(3)	[3]
	(v) <u>List</u>	Ca, Mg, Zn, Cu	(3)	[3]
		accept <i>names</i> of metals in correct order for (3)		
	(vi) <u>Give</u>	wear eye protection/ use small amounts/ view through side of test tube/ gloves...	(3)	[3]

Physics (130 MARKS)
Answer each of the questions 7, 8 and 9.

Question 7. (52 marks) All items, (a), (b), (c), etc. (7 × 6 + 1 × 10marks)

- (a) change of state/ liquid to solid/ latent heat (6) [6]
 (b) less dense/ lower density (6) [6]

- (c)  (3) [6]
 (3)

3 marks for each correct emergent ray from either diagram to a max. of (2 × 3)
allow marks if line is drawn correctly but the arrow is omitted

- (d) kiloWatt-hour/ kWh (3)
 450/ €4.5 (3) [6]
allow (2) for €450/ 3 × 10 × 15

- (e) $\frac{480000}{4}$ **or** 120 000 (3)
 Nm^{-2} (N/m²) (newtons per meter squared)/ Pa/ Pascal (3) [6]

- (f) **any two from either list or one from each list:**
Heat: form of energy/ Joules/ can not be measured at a point/ can be converted into other forms of energy/ depends on mass (substance) (temperature)...
Temperature: measure of how hot (cold) (degree of hotness)/ Celsius (centigrade)/ can be measured at a point/ differences can cause heat to flow/ independent of mass (substance)/ measured with thermometer... (2 × 3) [6]
note: two different points must be made to merit (2 × 3), it must be clear to which item the point is assigned in the candidate's answer.

- (g) $30 \times \mathbf{X} = 3 \times 40$ (3)
 $\mathbf{X} = 4$ (3) [6]
allow 6 marks if '4' alone appears.

- (h) (i) **B** (3)
 (ii) forward bias/ + end (anode) of LED to + of battery/
 - end (cathode) of LED to - of battery (3)
 (iii) control (limit)current/ without **R** the diode would burn out (4) [10]

Question 8. (39 marks) All items, (a), (b), (c), etc.

(a) (i) Calculate

Force (N)	Scale reading (cm)	Total extension (cm)
0	31.0	0
2	35.0	4
4	39.0	8
6	43.0	12
8	47.0	16
10	51.0	20

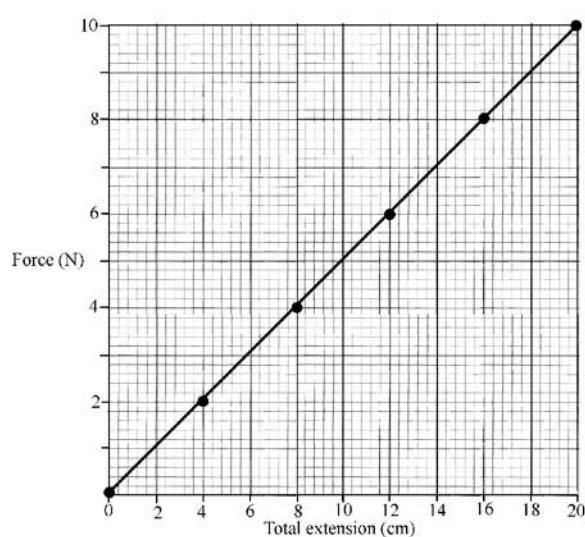
five extensions correctly calculated (6)

(6)

[6]

3-4 extensions correctly calculated (3) only

(ii) Draw



six points

(3)

plotted correctly

(3)

straight line through the six points

[6]

allow (3) for joining any six points

(iii) What?

extension directly proportional to applied force (6)

(6)

[6]

extension proportional to applied force (3) only

accept reverse order: 'applied force directly proportional to extension' for (6). If 'directly' is omitted from above (3) only.

(iv) Use

7 +/- 0.1

(3)

[3]

(b) (i) What?

bubbles of air/ water level falls

(3)

[3]

(ii) Explain

air in flask expanded

(3)

[3]

(iii) What?

water rises up glass tube/ bubbles stop

(3)

[3]

(iv) Explain

air in flask contracted/ air pressure in flask less than atmospheric/ partial vacuum

(3)

[3]

(c) Why?

any one from: light faster/ sound slower

(6)

[6]

Question 9. (39 marks) All items, (a) and (b).

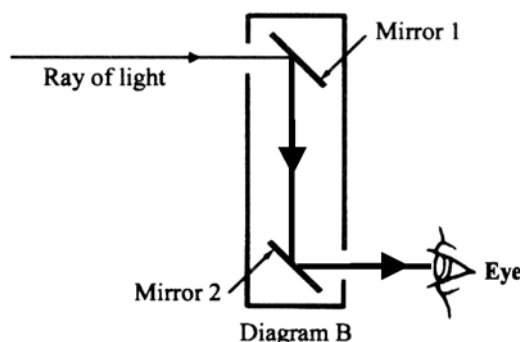
(a) (i) Name **any two from:** hydropower/ wind/ tidal/ geothermal/ biomass/ infra red (IR) from the sun/ wave (2 × 3) [6]

(ii) Give **any two from:** lower CO₂ emissions/ less carbon tax/ energy security/ lower energy costs/ sell surplus electricity/ sustainable/ cleaner/ kinder to the environment... (2 × 3) [6]

(b) (i) Complete

ray from mirror 1 to mirror 2 correctly drawn as shown (3)

ray from mirror 2 to eye correctly drawn as shown (3)

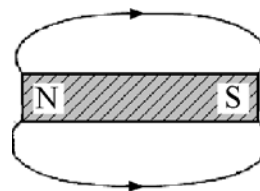


allow marks if line is drawn correctly but the arrow is omitted

(ii) Give **any one from:** see over objects/ see around corners/ submarine... (3) [3]

(c) (i) Label

north or south pole correctly labelled



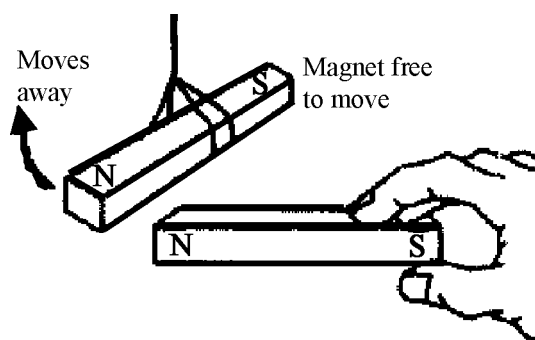
(ii) What? the direction in which a magnetic compass needle points/ the direction in which an isolated north pole would move if free to do so **accept: direction of magnetic force** (3) [3]

(iii) Describe **show or state**

bring a magnet towards a second magnet that is free to move

if two north poles or two south poles are brought close to each other they repel each other

[no diagram deduct 3 marks]



(iv) Name **any one from:** iron/ steel/ cobalt/ nickel (3) [3]

(v) How? the earths magnetism turns the needle of a magnetic compass/ use a compass/ freely suspended bar magnets north pole points north... (3) [3]