

Table 1
Crystallographic data of the ternary compounds in the ternary system Er–Ni–B

No.	Compound	Space group	Structure type	Lattice parameters				Reference
				<i>a</i> (nm)	<i>b</i> (nm)	<i>c</i> (nm)	β ($^\circ$)	
1	Er ₃ Ni ₁₃ B ₂	<i>P</i> 6/ <i>mmm</i>	Nd ₃ Ni ₁₃ B ₂	0.4938	—	1.090		[1]
a	Er _{0.917} Ni _{4.09} B	<i>P</i> 6/ <i>mmm</i>	Er _{0.917} Ni _{4.09} B	1.48399	—	0.6919		[13]
2	ErNi ₄ B	<i>P</i> 6/ <i>mmm</i>	CeCo ₄ B	0.4949	—	0.6931		[1]
3	Er ₃ Ni ₇ B ₂	<i>P</i> 6 ₃ / <i>mmc</i>	Dy ₃ Ni ₇ B ₂	0.5060	—	1.4276		[1]
a	~ErNi ₈ B ₂	—	Unknown	—	—	—		b
4	Er ₂ Ni ₂₁ B ₆	<i>F</i> m ₃ <i>m</i>	W ₂ Cr ₂₁ C ₆	1.0640	—	—		[1]
5	ErNi ₇ B ₃	<i>I</i> 4 ₁ / <i>amd</i>	ErNi ₇ B ₃	0.7665	—	1.5584		[13]
6	~ErNi _{6.5} B ₃	cubic	Unknown	0.7735 (3)	—	—		b
7	Er ₂ Ni ₁₀ B ₅	<i>P</i> bca	Ho ₂ Ni ₁₀ B ₅	1.7545	0.8977	0.9441		[6]
8	Er ₃ Ni ₁₉ B ₁₀	<i>C</i> 2/ <i>m</i>	Ho ₃ Ni ₁₉ B ₁₀	1.3101	0.8674	0.5761	91.05	[7]
9	Er ₂ Ni ₁₅ B ₉	<i>C</i> mca	Ho ₂ Ni ₁₅ B ₉	1.5874	1.1561	1.1235		[4]
10	ErNi ₂ B ₂	<i>C</i> 2/ <i>c</i>	HoNi ₂ B ₂	0.8345	0.5188	0.6887	126.60	[5]
11	Er ₂ Ni ₃ B ₆	<i>C</i> mmm	Lu ₂ Ni ₃ B ₆	0.7701	0.8632	0.3462		[3]
12	ErNiB ₄ ; LTM	<i>P</i> bam	YCrB ₄	0.5792	1.1544	0.3435		[14]
	ErNiB ₄ ; HTM	<i>I</i> 4/ <i>mmm</i>	ErNiB ₄	0.7505	—	0.8496		[13]
13	Er ₄ NiB ₁₃	<i>P</i> 4/ <i>mnc</i>	Er ₄ NiB ₁₃	0.7186	—	0.7446		[1]

^a Compounds were found only in the as-cast samples.

^b Data of our investigation.

The formation of solid solution regions for the binary compounds was not revealed, and practically no variation of the lattice parameters of the ternary borides in two- and three-phase samples was observed (Table 2). This fact indicates that no homogeneity ranges exist for the ternary compounds.

The ternary compounds Ln₂NiB₁₀ where Ln=Y, Gd, Tb, Dy, Ho with the Th₂NiB₁₀ structure type were reported in [12]. We prepared the sample of the composition Er₂NiB₁₀ to verify the formation of the boride with the same structure in the Er–Ni–B system. But the powder diffraction pattern of this sample contained the lines of three binary compounds (ErB₄ + ErB₁₂ + NiB), which indicates the ab-

sence of the ternary boride Er₂NiB₁₀ in the Er–Ni–B system (Table 2).

A compound of the composition ErNiB₄ was reported to crystallize with two different structures: ErNiB₄ [13] and YCrB₄ [14]. This can indicate the existence of polymorphism for this boride. Taking into account that the boride with the ErNiB₄ type structure was found by a single-crystal method, and that the boride with the YCrB₄ type structure was found by X-ray powder diffraction, we assumed the high-temperature modification to have the ErNiB₄ structure and the low-temperature modification to have the YCrB₄ structure. The results of a DTA investigation revealed a transition temperature of 1540 K.

A new compound of the approximate composition ErNi₈B₂ with unknown structure was found. The X-ray powder pattern of this compound differed from the powder patterns of the known ternary borides found in the earlier studied Ln–Ni–B systems. This compound exists only in the as-cast sample and decomposes into the Er₂Ni₂₁B₆ and ErNi₄B compounds after annealing. Since the single crystals were not available its structure was not determined.

A further new compound was found in the as-cast samples containing 16.6 at.% Er. The X-ray powder pattern of this compound was very similar to the powder pattern of the ErNi₄B boride. The crystal structure of the new Er_{0.917}Ni_{4.09}B boride was determined using an X-ray single-crystal method: space group *P*6/*mmm*; *a* = 1.48399 nm; *c* = 0.69194 nm [15]. This structure is very similar to the Y_{0.915}Ni_{4.12}B structure [16]. The Er_{0.917}Ni_{4.09}B boride decomposes after annealing, and the main phase is the ErNi₄B compound.

According to our data, the Er₄Ni₂₉B₁₀ boride has the correct composition ErNi₇B₃: space group *I*4₁/*amd*; *a* =

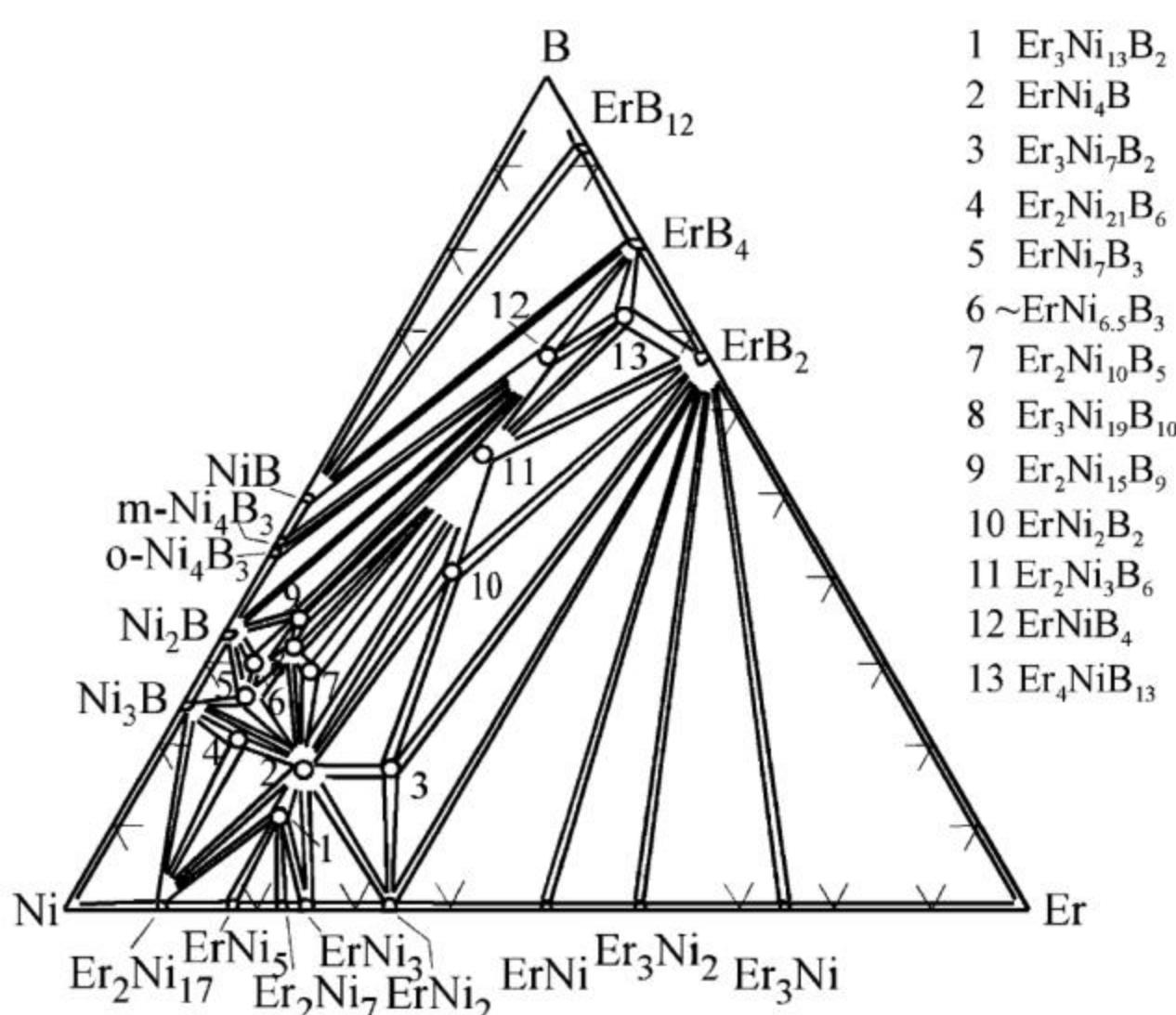


Fig. 1. The phase diagram of the ternary system Er–Ni–B at 1070 K.