BIOTECHNOLOGICAL PRODUCTS AND PROCESS ENGINEERING

Production of isopropanol by metabolically engineered Escherichia coli

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Abstract A genetically engineered strain of *Escherichia* coli JM109 harboring the isopropanol-producing pathway consisting of five genes encoding four enzymes, thiolase, coenzyme A (CoA) transferase, acetoacetate decarboxylase from Clostridium acetobutylicum ATCC 824, and primarysecondary alcohol dehydrogenase from C. beijerinckii NRRL B593, produced up to 227 mM of isopropanol from glucose under aerobic fed-batch culture conditions. Acetate production by the engineered strain was approximately one sixth that produced by a control E. coli strain bearing an expression vector without the clostridial genes. These results demonstrate a functional isopropanolproducing pathway in E. coli and consequently carbon flux from acetyl-CoA directed to isopropanol instead of acetate. This is the first report on isopropanol production by genetically engineered microorganism under aerobic culture conditions.

Keywords Isopropanol · Escherichia coli · Clostridium acetobutylicum · Clostridium beijerinckii · Metabolic engineering

Introduction

Production of fuel and valuable chemicals from biomassderived sugars has attracted much attention during last few decades due to the rising oil price and as an effort to reduce CO₂ emissions to tackle global warming. As a result,

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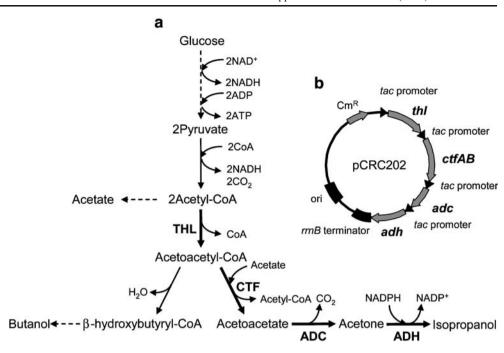
society's needs are gradually shifting from petroleum to renewable biomass for feedstock.

Solvent production from biomass is not a new concept. Large amounts of acetone and butanol were produced by fermentation (AB fermentation) early in the twentieth century, although a decline of the fermentation began as a result of economic competition from the petrochemical industry after the World War II (Dürre 1998). AB fermentation is achieved by microorganisms of the genus Clostridium under strictly anaerobic conditions. Biosynthetic pathways and relevant genes to AB fermentation in Clostridium acetobutylicum are well studied (Mitchell 1998). Figure 1a shows the metabolic pathways of acetone and butanol synthesis as well as related metabolic pathways. Another Clostridium bacterium, C. beijerinckii, produces isopropanol instead of acetone together with butanol (George et al. 1983). This microorganism possesses an additional primary-secondary alcohol dehydrogenase (ADH) that catalyzes the reduction of acetone (Ismaiel et al. 1993). Although several attempts have been made to overcome limitations of AB fermentation such as the loss of ability for solvent production (known as "degeneration") upon serial subculturing or in continuous culturing (Mitchell 1998) and growth inhibition by product solvents that cause low titer of products, acetone and butanol are currently produced from petrochemical processes.

Heterologous expression of clostridial genes for solvent production in industrial organisms, such as Escherichia coli, Corynebacterium glutamicum, and Saccharomyces cerevisiae, is a promising approach for industrial production of biosolvents competitive with petrosolvents because so much information on their physiology and well-established genetic tools are available. For example, an engineered strain of E. coli expressing four genes coding for three enzymes from C. acetobutylicum, thiolase (THL), coenzyme A (CoA)



Fig. 1 Biosynthetic pathway of isopropanol and related metabolic pathways in E. coli (a) and plasmid for introduction of isopropanol-producing pathway into E. coli (b). Enzymes introduced into engineered E. coli are indicated in bold letters in a: thiolase (THL), CoA-transferase (CTF), acetoacetate decarboxylase (ADC), primary-secondary alcohol dehydrogenase (ADH). Abbreviations in b: thl thiolase gene, ctfAB genes for two subunits of CoA-transferase, adc acetoacetate decarboxylase gene, adh primary-secondary alcohol dehydrogenase gene, Cm^R chloramphenicol resistance gene



transferase (CTF), and acetoacetate decarboxylase (ADC), produced acetone at titers comparable to those obtained with *C. acetobutylicum* (Bermejo et al. 1998).

This study demonstrates production of isopropanol from glucose by genetically engineered *E. coli*. Isopropanol is used as a solvent in many applications, as a cleaner for electric devices, as a fuel additive for dissolving water in fuel lines, and as a feedstock for production of chemicals such as propylene. Most studies for isopropanol–butanol fermentation by clostridia so far relate to improvement of fermentation processes to prevent product inhibition, for example, with a continuous fermentation using Ca-alginate-entrapped clostridial cells (Krouwel et al. 1983), pervaporation (Matsumura et al. 1992), and product adsorption by XAD resin (Groot and Luyben 1986). This is the first report describing isopropanol production by metabolic engineering.

Materials and methods

Microorganisms and cultivation

E. coli strain JM109 (Takara, Shiga, Japan) was grown in Luria–Bertani (LB) medium. Antibiotics were included as appropriate at the following concentrations: ampicillin, 50 μg/ml; chloramphenicol, 25 μg/ml. *C. beijerinckii* NRRL B593 (synonym: DSM6423) was grown in reinforced clostridial medium (BD Difco, NJ, USA) at 30°C in an anaerobic atmosphere (95% nitrogen and 5% hydrogen). After 24-h incubation, clostridial cell was harvested by centrifugation (5,000×g, 10 min) and was used for genome extraction.

DNA amplification and plasmid construction

General techniques for DNA manipulation were performed according to the reference (Sarmbrook and Russell 2001). Three DNA fragments containing the genes thl (Locus tag of C. acetobutylicum genome: CAC2873), ctfAB (CA P0163, CA P0164), and adc (CA P0165) were amplified by polymerase chain reaction (PCR) with primers 1 and 2, 3 and 4, 5 and 6, respectively (Table 1). Genomic DNA of C. acetobutylicum ATCC824 was purchased from American Type Culture Collection (ATCC 824D-5). PCR was performed using a GeneAmp PCR system 9700 (Applied Biosystems, CA, USA) and PrimeSTAR DNA polymerase (Takara). Thermal cycling consisted of 94°C for 3 min followed by 30 cycles of 95°C for 20 s and 58°C for 30 s and 72°C for 1 min. Genomic DNA of C. beijerinckii was extracted by the method (Marmur 1961). The gene (accession no. AF157307) for ADH was amplified from C. beijerinckii genome with primers 7 and 8. Primers were designed to fuse Shine-Dalgarno sequence into the 5' end of each gene. DNA fragment containing tac promoter was amplified from a plasmid pKK223-3 as a PCR template. Primer set for the tac promoter amplification was used as follows: 9 and 10 for ctfAB, 10 and 11 for adc, and 10 and 12 for adh. The tac promoter and each gene (ctfAB, adc, and adh) were joined by a crossover PCR method (Horton et al. 1989), and resulting amplicons were purified from an agarose gel and subsequently were cloned into a pGEM-T vector (Promega, WI, USA) after addition of deoxyadenosine to the 3' ends of the PCR products to ensure DNA sequences of the amplicons. The genes with the tac promoter were excised from the pGEM-T vectors



Table 1 Oligonucleotides used in this study	Primer	Sequence (5'-3')
	Primer 1	CGAATTCAAAGGAGGAGTGTGTTGATGAAAGAAGTTGTAATAGC
	Primer 2	GGATCCCTAGCACTTTTCTAGCAAT
	Primer 3	TTCACACAGGAAACAAAGGAGGAGTGTGTTGATGAACTCTAAAATAATTAGA
	Primer 4	GCATGCTAAACAGCCATGGGTCTAAG
	Primer 5	TTCACACAGGAAACAAAGGAGGAGTGTTGATGTTAAAGGATGAAGTAATTAA
	Primer 6	<u>CCCGGG</u> TTACTTAAGATAATCATATATAAC
	Primer 7	TTCACACAGGAAACAAAGGAGGAGTGTGTTGATGAAAGGTTTTGCAATGCTA
	Primer 8	<u>GTCGAC</u> TTATAATATAACTACTGCTTTA
	Primer 9	GGATCCCCATCGGAAGCTGTGG
	Primer 10	TGTTTCCTGTGTGAAATTG
The restriction site used in the	Primer 11	<u>GCATGC</u> CATCGGAAGCTGTGG
cloning procedure has been underlined.	Primer 12	CCCGGGCCATCGGAAGCTGTGG

and sequentially inserted into expression vector pCRC200 that has a multiple cloning site between *tac* promoter and *rrnB* terminator (Yasuda et al. 2007) to obtain pCRC202 (Fig. 1b). The plasmid pCRC202 was introduced into *E. coli* JM109 according to a standard method (Sarmbrook and Russell 2001).

Isopropanol production

E. coli was inoculated in LB medium (2.5 ml in a test tube) and grown overnight at 30°C. The resulting culture was inoculated into a 500-ml baffled flask containing 50 ml of a complex (SD8) medium (Luli and Strohl 1990). The culture was incubated for isopropanol production in an agitation incubator (Tasaki Scientific Instruments, Saitama, Japan) at 250 rpm and 37°C. Glucose was occasionally supplemented to maintain glucose concentration more than 30 mM. Data represent average values in triplicate experiments. Standard deviations were within 5% of the mean.

Analyses

Isopropanol and acetone were determined by using a gas chromatography (GC-14B, Shimadzu, Kyoto, Japan) equipped with Thermon-1000 Sunpak-A 50/80 (Shinwa Chemical Industries, Kyoto, Japan). Organic acids were quantified by high-performance liquid chromatography (model-8020, Tosoh, Tokyo, Japan) equipped with an electric conductivity detector and a TSKgel OApak-A column (Tosoh) operating at 40°C with a 0.75 mM H₂SO₄ mobile phase at a flow rate of 1.0 ml/min. The concentration of glucose was measured by an enzyme electrode glucose sensor according to manufacture's protocol (BF-5, Oji Scientific Instruments, Hyogo, Japan). Optical density was measured at 610 nm using a Novaspec II spectrophotometer (Amersham Pharmacia biotech, UK), and dry weights of cells were determined by drying at 80°C.

Results

Engineering isopropanol productivity in E. coli

Isopropanol-producing pathway was designed by combining endogenous glycolytic pathway in E. coli with four enzymes, THL, CTF, ADC from C. acetobutylicum ATCC824, and ADH from C. beijerinckii NRRL B593. Although C. acetobutylicum produces acetone and butanol but not isopropanol, its acetone-producing pathway shares the same three enzymes (THL, CTF, and ADC) with isopropanol-producing pathway of C. beijerinckii (Fig. 1a). Although the genes for the enzymes are identified from C. acetobutylicum (Cary et al. 1990; Petersen and Bennett 1990; Petersen and Bennett 1991), no report has been described about gene identification of the three enzymes from C. beijerinckii NRRL B593. Thus, we employed the genes from C. acetobutylicum for construction of a synthetic isopropanol pathway in E. coli. Plasmid pCRC202 was constructed to express the five genes (thl, ctfAB, adc, and adh) under control of the tac promoter in E. coli (Fig. 1b). Theoretical yield using this pathway was 1 mol of isopropanol formed from 1 mol of glucose consumed. Isopropanol production was conducted in aerobic culture conditions because oxidation of NADH without formation of fermentative metabolites, such as lactate, was preferable.

Effect of metabolic engineering on growth rate and glycolytic flux

Specific growth rates were calculated to be 1.0 and 1.4 (h⁻¹) for isopropanol-producing *E. coli* bearing plasmid pCRC202 (*E. coli*-pCRC202) and control *E. coli* bearing an expression vector pCRC200 (*E. coli*-pCRC200), respectively, indicating that engineered *E. coli* grew slightly slower than *E. coli*-pCRC200 under the conditions tested



(Fig. 2a). In contrast, dry cell mass of *E. coli*-pCRC202 and *E. coli*-pCRC200 after 48-h cultivation was 5.7 and 2.6 g/l, respectively. *E. coli*-pCRC202 consumed glucose continuously, whereas *E. coli*-pCRC200 almost ceased consuming glucose at the onset of stationary phase (Fig. 2b).

Production of isopropanol by genetically engineered E. coli

Fermentation was conducted in a glucose fed-batch culture in a baffled flask. Glucose was occasionally supplemented at concentrations ranging from 30 to 200 mM. After 36-h cultivation, 227±6 mM of isopropanol was detected in the culture broth, and further incubation did not increase isopropanol concentration (Fig. 3). Acetone formation was also observed. No isopropanol and acetone were detected in a broth of the *E. coli*-pCRC200 strain, indicating that the clostridial genes were actively expressed in *E. coli*-pCRC202. Isopropanol yield was 51% after 36-h cultivation, and total solvent yield (isopropanol plus acetone) was 61%. Maximum volumetric productivity of isopropanol was 10.3 mmol·1⁻¹·h⁻¹ under the conditions.

Production of acetic acid

Acetate production by the isopropanol-producing strain, *E. coli*-pCRC202, was expected to be lower than that by *E. coli*-pCRC200 because isopropanol-producing pathway competes with acetate-formation pathway at the acetyl-CoA node (Fig. 1a). After 48-h fed-batch culture, *E. coli*-

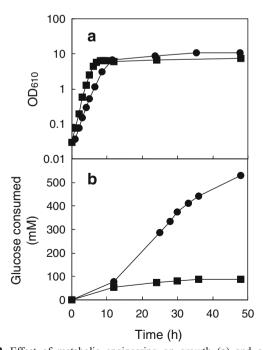


Fig. 2 Effect of metabolic engineering on growth (a) and glucose utilization (b) in glucose fed-batch cultures. *Circles* and *squares* indicate engineered *E coli*-pCRC202 and *E. coli*-pCRC200, respectively

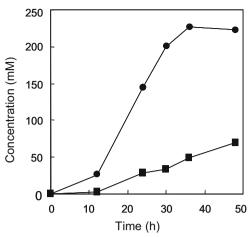


Fig. 3 Isopropanol production by engineered *E. coli* in glucose fedbatch culture. *Circles* and *squares* indicate concentrations of isopropanol and acetone in a culture broth, respectively

pCRC202 produced sixfold lower acetate than *E. coli*-pCRC200. The pH values of culture broths were 4.6 and 5.8 after 24 h in control and engineered *E. coli*, respectively. These results clearly indicated that carbon flux was redirected to isopropanol (and acetone) from acetate in the genetically engineered *E. coli*.

Discussion

This study demonstrated that *E. coli* with the five genes coding for four enzymes from two *Clostridium* species produced up to 227 mM of isopropanol under aerobic culture conditions (Fig. 3). Isopropanol titer and productivity of engineered *E. coli* were higher than those of clostridia in a batch culture. *C. isopropylicum* IAM 19239 immobilized with Ca-alginate has been demonstrated to produce

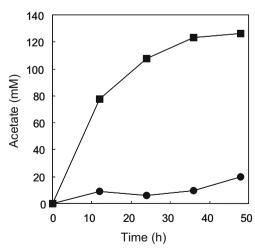


Fig. 4 Effect of metabolic engineering on acetate production in glucose fed-batch cultures. *Circles* and *squares* indicate acetate concentrations in culture broths of engineered *E coli*-pCRC202 and *E. coli*-pCRC200, respectively



77 mM of isopropanol for approximately 67 h of a batch fermentation (Matsumura et al. 1992). Employing XAD resin to adsorb solvent and repress fermentation inhibition, *C. beyerinckii* LMD 27.6 produces 67 mM for 100 h in a batch culture (Groot and Luyben 1986). Clostridia produce isopropanol together with butanol, which causes lower titer of isopropanol, while engineered *E. coli* developed in this study lacks butanol-producing pathway.

The biosynthetic pathway of isopropanol shares the three enzymes with the acetone-producing pathway (Fig. 1a). In particular, some strains of *C. beijerinckii* produce both acetone and isopropanol together with butanol (Chen and Hiu 1986). Acetone formation was also observed in the broth of isopropanol-producing *E. coli* in this study (Fig. 3). This result suggests that the reaction from acetone to isopropanol might be one of the rate-limiting steps for the biosynthetic pathway of isopropanol. Another possible explanation is reoxidation of isopropanol. Further studies on flux analyses and genetic and biochemical characterization of enzymes relevant to isopropanol metabolism are required to reduce acetone and increase isopropanol productivity.

Acetone-producing E. coli were constructed by introducing the genes encoding three enzymes, THL, CTF, and ADC from C. acetobutylicum (Bermejo et al. 1998). The E. coli produced 93 and 154 mM of acetone in shake flask and bioreactor cultivations, respectively, whereas the isopropanol titer was 227 mM in shake flask culture of E. colipCRC202 in this study. The yield of isopropanol and solvent (isopropanol plus acetone) was 51 and 61%, respectively, in the present study. These values were slightly higher than those of acetone-producing E. coli where acetone yield was below 50% (Bermejo et al. 1998). It is uncertain that isopropanol titer and yield were higher than those of acetone-producing E. coli. Possibly, differences in plasmid construction might affect productivity. In acetone-producing E. coli, one thiolase promoter from C. acetobutylicum is located at the 5' upstream of a synthetic acetone operon, while the tac promoter fused with every gene for THL, CTF, ADC, and ADH in this study (Fig. 1b).

Engineered *E. coli* continued glucose consumption for a longer time than *E.coli*-pCRC200 (Fig. 2b). This result is most probably explained by difference in acetate productivity between engineered *E. coli* and *E. coli*-pCRC200 (Fig. 4). The control *E. coli*-pCRC200 strain excreted much more acetate than engineered strain, resulting in lower broth pH of *E. coli*-pCRC200. The pH values were 4.6 and 5.8 in *E. coli*-pCRC200 and engineered *E. coli*, respectively. *E. coli* accumulates acetate when growing at a high rate of glucose consumption, which is known as overflow metabolism (El-Mansi and Holms 1989; Luli and Strohl 1990). Detrimental effect of acetate on growth (Luli and Strohl 1990) and production of recombinant protein has been reported (Jensen and Carlsen 1990). Isopropanol-producing

pathway competes with acetate-producing pathway in an acetyl-CoA node (Fig. 1a). Therefore, introduction of isopropanol-producing pathway led to low acetate productivity compared to *E. coli*-pCRC200 (Fig. 4). Consequently, inhibition of glucose consumption by acetate seems to be repressed in engineered *E. coli* in this study.

The results described above indicated that expression of the five genes from two *Clostridium* spp. led to production of isopropanol in *E. coli* under aerobic conditions. Further improvement of isopropanol titer and productivity can be achieved by combining existing technologies such as high-cell-density fermentation, continuous fermentation, and *in situ* product recovery.

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