MOFs NMR DFT

Calculation of nmr parameters in paramagnetic metal-organic materials

Jure Lapajne

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■ MOFs = Metal—organic frameworks

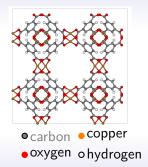
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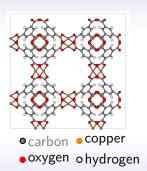
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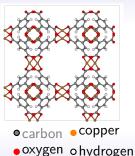


DFT

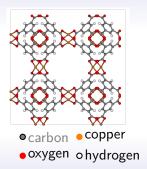
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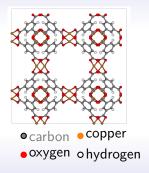
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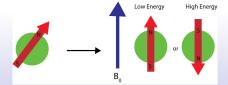


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- NMR spectra feature large shifts caused by unpaired electrons

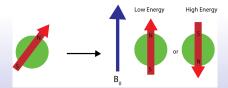


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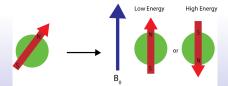
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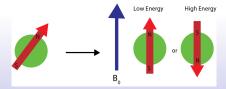
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- absorption peak at $\Delta E = \hbar \omega_{res}$
- ω_{res} depends on $B_{eff}(observed nucleus)$



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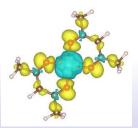
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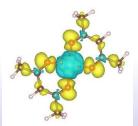
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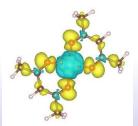
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Chemical and hyperfine shifts

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Total shift tensor $\underline{\sigma}$ is defined by:

$$ec{B}_{eff} = ec{B}_0 \left(\underline{\underline{I}} - \underline{\underline{\sigma}}
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Two sources of chemical shifts:

- electron density change caused by applied external magnetic field
- depends on electron density $n(\vec{r})$

- coupling between unpaired electron and observed nuclei
- depends on spin density $n_{\uparrow}(\vec{r}_{nuclei}) - n_{\downarrow}(\vec{r}_{nuclei})$

Calculation of NMR parameters: accurate electronic wave function needed!

$$\hat{H} = T_n + T_e + W_{n-n} + W_{e-n} + W_{e-e} + V_{ext}$$

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Born-Oppenheimer approximation

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- Born-Oppenheimer approximation
- direct solution of coupled pde not feasible
- at a given accuracy level: time needed grows exponentially as a function of number of particles



density functional theory — DFT

Approximations

DFT

- most widely used
- good tradeoff between accuracy and speed
- highly customizable suitable for various molecules

Alternatives

- * Hartree—Fock
- * Quantum Monte Carlo
- * Coupled-Cluster methods

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