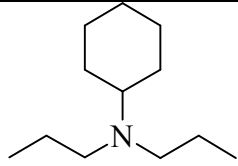
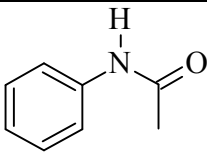
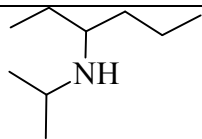
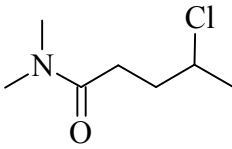
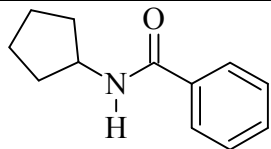
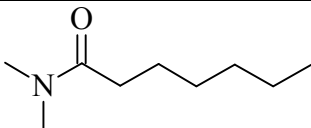
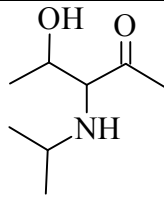
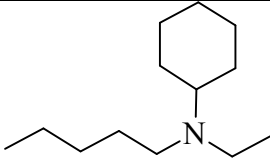
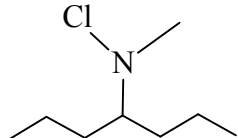
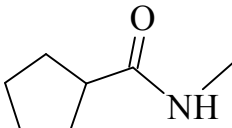
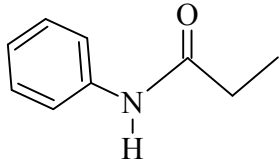
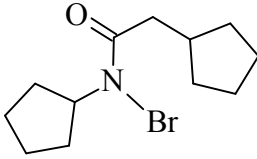


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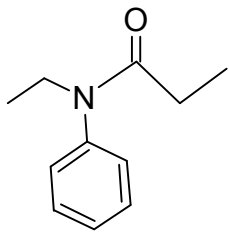
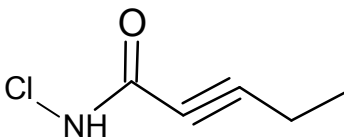
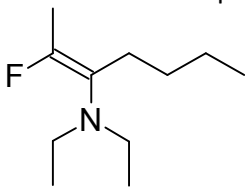
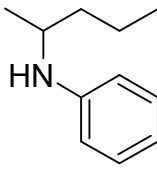
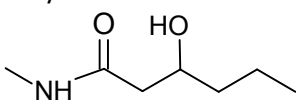
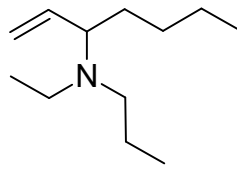
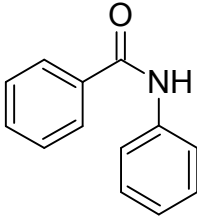
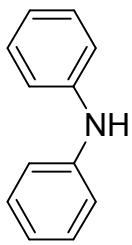
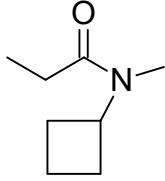
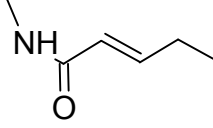
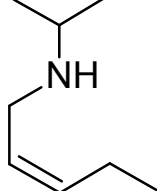
AMINES AND AMIDES

1. Name the following.

 <p>N,N-di-n-propylaminocyclohexane</p>	 <p>N-phenylethanamide</p>
 <p>N-isopropyl-3-aminoheptane</p>	 <p>N,N-dimethyl-4-chloropentanamide</p>
 <p>N-cyclopentylbenzamide</p>	 <p>N,N-dimethylheptanamide</p>
 <p>N-isopropylpentan-4-ol-2-one</p>	 <p>N-ethyl-N-n-pentylaminocyclohexane</p>
 <p>N-chloro-N-methyl-4-aminoheptane</p>	 <p>N-methylcyclopentylmethanamide</p>
 <p>N-phenylpropanamide</p>	 <p>N-bromo-N-cyclopentylcyclopentylethanamide</p> <p>(no number needed for cyclopentyl as it can only go on that carbon)</p>

Name: _____

2. Draw the following.

<p>N-ethyl-N-phenylpropanamide</p> 	<p>N-chloro-pent-2-ynamide</p> 
<p>cis-N-diethyl-3-amino-2-fluoro-2-heptene</p> 	<p>N-phenyl-2-aminopentane</p> 
<p>N-methyl-3-hydroxyhexanamide</p> 	<p>N-ethyl-N-n-propyl-3-amino-1-heptene</p> 
<p>N-phenylbenzamide</p> 	<p>diphenylamine</p> 
<p>N-cyclobutyl-N-methylpropanamide</p> 	<p>trans-N-methylpent-2-enamide</p> 
<p>cis-N-isopropyl-1-amino-2-pentene</p> 	<p>trans-N-chloro-N-ethylhex-2-enamide</p> 