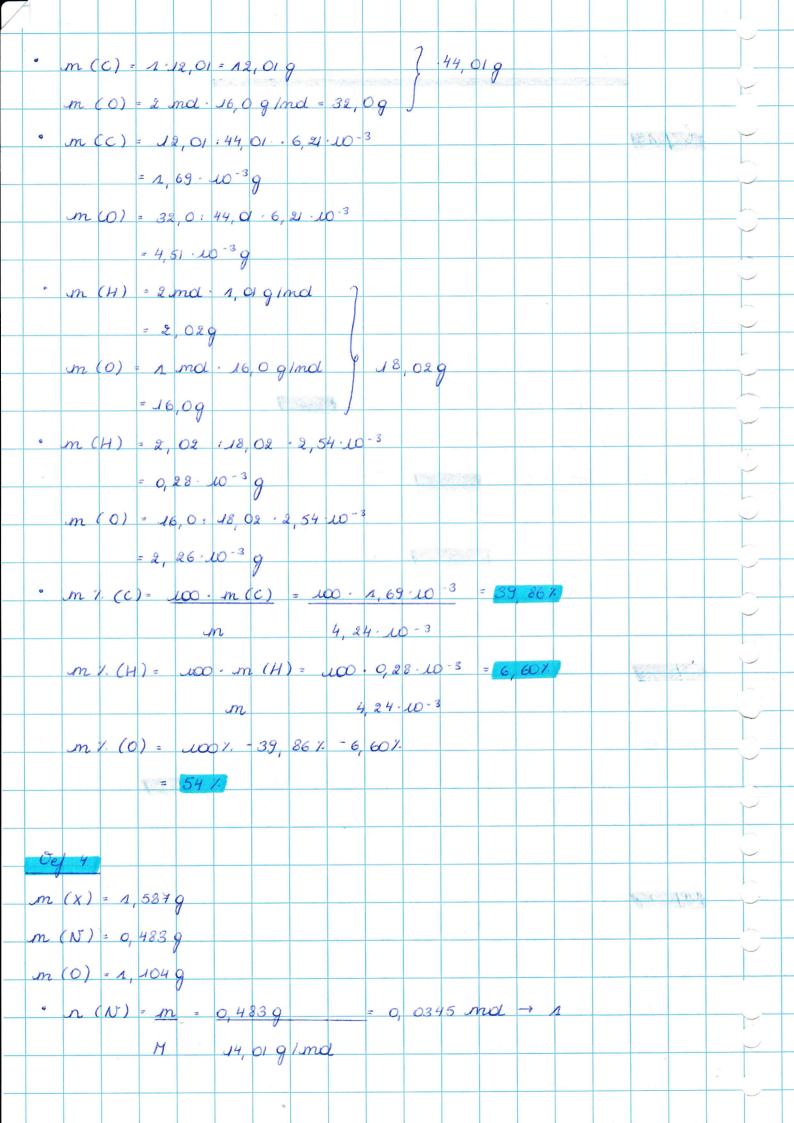
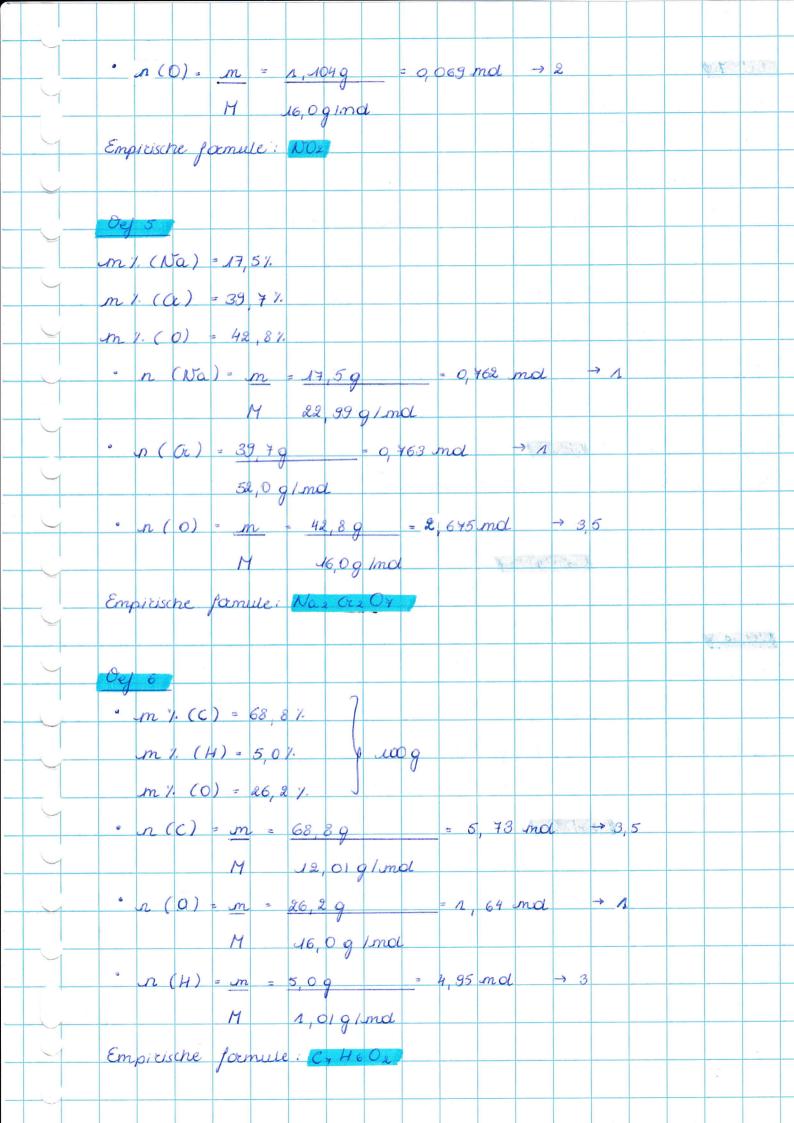
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		Che	miscl	r te	benen	: B	epoli	ing .	chemi	sche	John	ules					
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	Dej n													e			
	NH4 NO	3	→ .m	Ž.													
	N				O												
	å md					d					8						
<u> </u>	m (N)						1				<i>y</i> .		1 4			V	
	m (H)						/	m (NH4 N	JO3)	= 80	06 (7	10 12			
	m (0)													Я			
<u> </u>	m 1. (N				V		x · 2	8,02	= 3	5,0	7						
			· "m				80,0								14		
	m 1. (H)) =	· 000 ·	4,04	-								t				
4			80														
	m > (0					59 _, 9	6 %										
\perp				06			5										
4	Oej 2		Ę	7.30							181 181 0						
	m 1. (1																
	m (NH	4 NO	3)=	48	9												
	m (N)					(NH4	UO3) =	0,3	50·	48,5	9 =	16,	98 9			
			2	- U	∞	sui i				ıœ							
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	ā	m i	ı. (o.) = 5	3,4%			v												
L.	٠	n C	c)	= m	**	39,9	9		= 3,	32 4	mol		→ 1				12.	<u> </u>		0
	e	n ()				•			= 6,1	me	L	→ 2		<u>, , , , , , , , , , , , , , , , , , , </u>						
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	Oef 9	A.V.	
4	· C2H4 (etheen)		
	m(c) = 12 · 2 · = 24 g (28,04 g		
	$m(H) = 4 \cdot 100 = 4000$		
	m/(c) = 24g . 100 = 86 y.		
	28,049		
	° C3 47 OH (proponal)	Auricia Vicente	
	$m(c) = 42 \cdot 3 = 369$	600	
	m (H) = 8-12,01 = 8,08g 60,08g		
	m (0) = 16g		
	$m = (c) = 36 g \cdot 100 = 60\%$		
	60,089		
	° C7 H16 (heptoon)		
	m (c) = 49.7 = 849 7 100, 169		
	m (H) = 16 · 1,01 = 16, 16g		
	m 1. (c) = 84 g · 100 = 84%		
1	400,469		
	Etheen > heptoon > propancl		
	Oej 10		
	H (KBz) = 19,00 g/md		
	M (K2S) = 110,269 /mal		
	m (KBE) + m (K2S) = 8,029		
	$m(\kappa) = 4,50g$		
	n(K) = 0, 115 mod		
			-

