

Sodium sulfite

Sodium sulfite structure

Sodium sulfite is a [chemical compound](#). White odorless powder. Density 2.633 g / cm³. It is moderately toxic. ^[1]

It is made of [sodium](#) and [sulfite ions](#). It is one of the sulfites used to [preserve](#) dried food. It is used to remove [chlorine](#) from pools. It reacts with [acids](#) to produce [sulfur dioxide](#). It is a weak [reducing agent](#). When it is [oxidized](#), it is converted to [sodium sulfate](#). Its chemical formula is Na₂SO₃.

Related pages

[[change](#) | [change source](#)]

- [Sulfate](#)
- [Sulfuric acid](#)
- [Sulfurous acid](#)

References

[[change](#) | [change source](#)]

1. [↑] "[Sodium sulfite 7757-83-7 wiki](#)". *Guidechem*.

This [short article](#) about [chemistry](#) can be made longer. You can help Wikipedia by [adding to it](#).

Retrieved from "https://simple.wikipedia.org/w/index.php?title=Sodium_sulfite&oldid=9068192"