Copper(I) bromide

Oxidized copper(I) bromide. Pure copper(I) bromide is colorless.

Copper(I) bromide, also known as **cuprous bromide**, is a <u>chemical</u> <u>compound</u>. Its chemical formula is CuBr. It has <u>copper</u> and <u>bromide ions</u> in it. The copper is in its +1 <u>oxidation state</u>.

Properties

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Copper(I) bromide is a colorless solid. It turns green when in air. It does not dissolve in water. It is a <u>reducing agent</u>.

Preparation

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Copper(I) bromide is made by reacting <u>copper(II)</u> bromide with any <u>sulfite</u> and any <u>bromide</u> <u>salt</u>. This reaction makes <u>hydrogen bromide</u> gas.

Uses

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It is used in <u>organic chemistry</u>. It is used to make organic compounds that have copper in them.

Related pages

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- Copper(I) iodide
- Copper(I) chloride

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