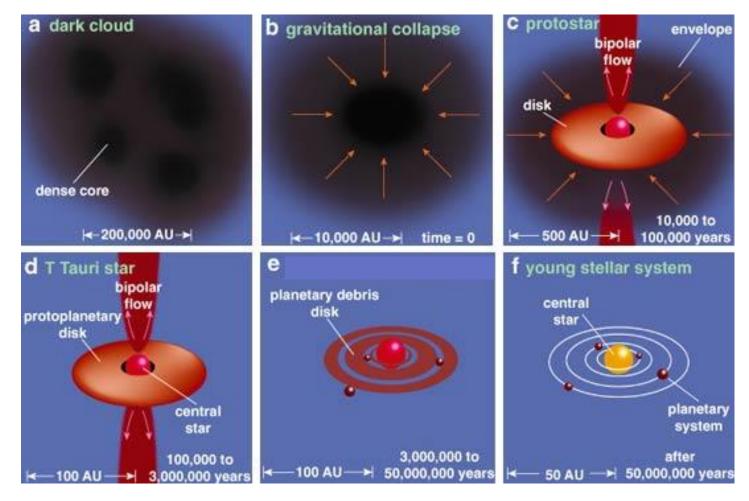
Water trail: from molecular clouds to protoplanetary disks

古家健次

筑波大・計算科学研究センター

## Follow the water trail

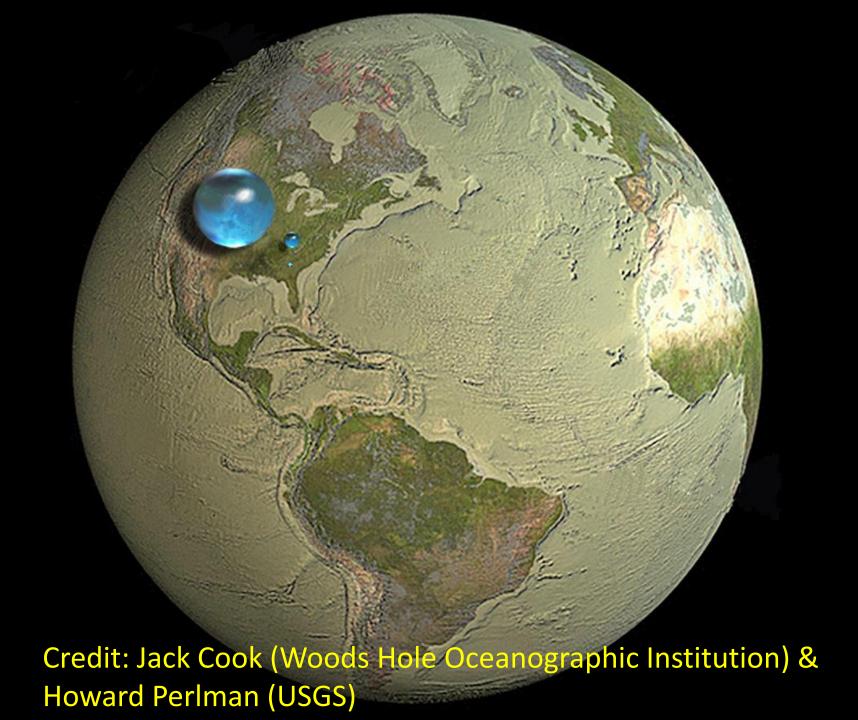


(Greene 2001)

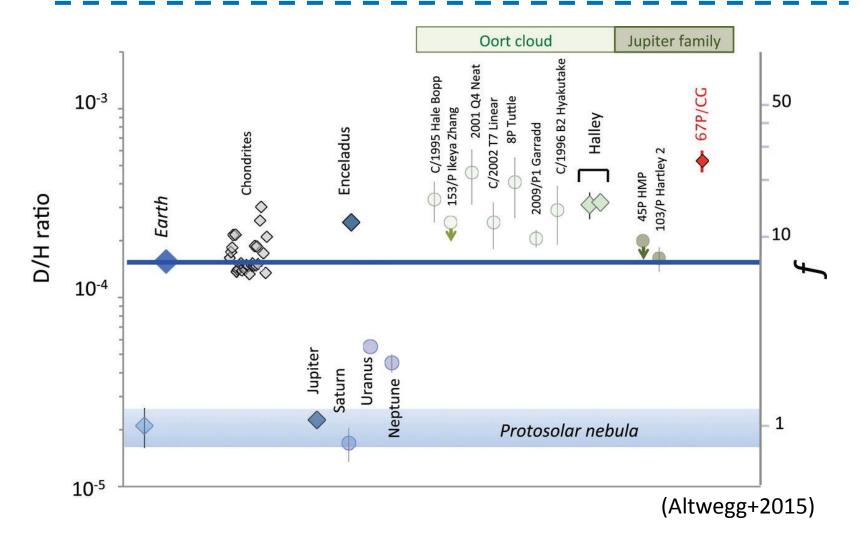
Where did our water come from?



Credit: Jack Cook (Woods Hole Oceanographic Institution) & Howard Perlman (USGS)

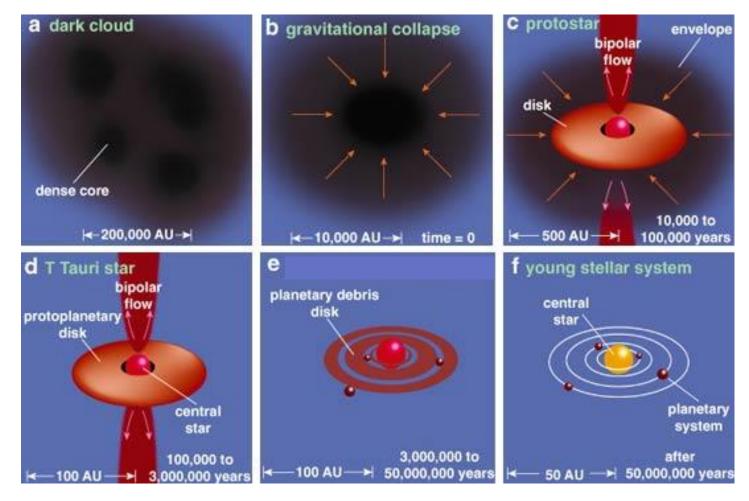


## Origin of Earth's ocean



Potential sources of Earth's ocean: asteroids and comets

## Follow the water trail



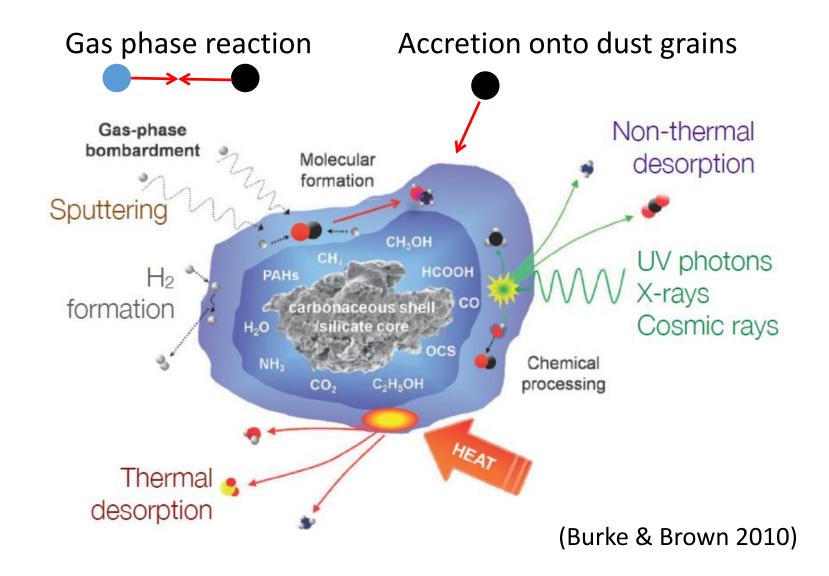
(Greene 2001)

Where did our water come from?

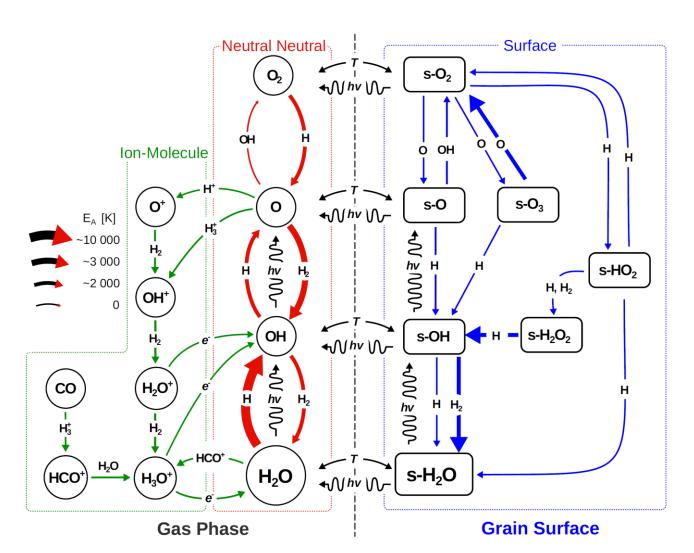
#### Contents

- Water chemistry in star-forming regions
- (Water) ice formation in molecular clouds and cores
- Water delivery from cores to disks
- Water deuteration as a probe to follow the water trail

## Chemical processes in star-forming regions



## Water chemistry: well studied



(van Dishoeck et al. 2013)

(Original ref. Jensen+2000; Miyauchi+2008; Ioppolo+2008; Oba+2012 and many others)

# The formation of water on interstellar dust particles

prof. Ewine F. van Dishoeck, PhD, A.L.M. "Thanja" Lamberts, MSc





Discover the world at Leiden University

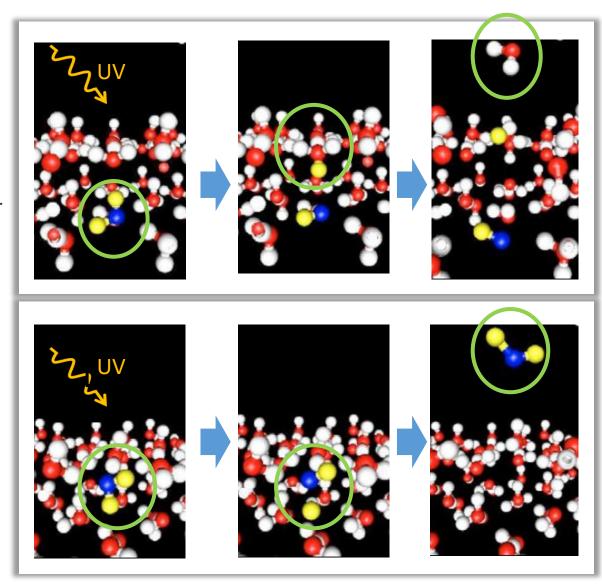
## 光脱離

#### Kick-out

$$H_2O(ice) + hv \rightarrow H^* + OH,$$
  
 $H^* + H_2O(ice) \rightarrow H + H_2O(gas).$ 

#### Recombination

$$H_2O(ice) + hv \rightarrow H^* + OH,$$
  
 $H + OH \rightarrow H_2O(gas)$ 



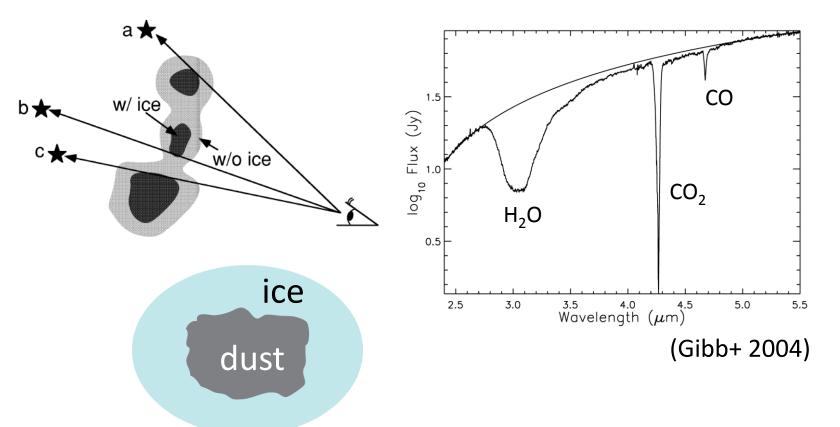
MD計算; Andersson & van Dishoeck (2008)

#### Contents

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## Water in prestellar stages

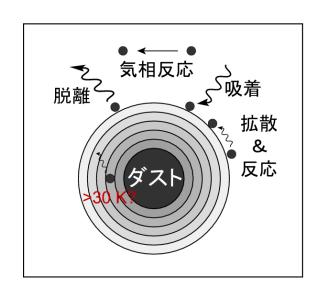
- 分子雲に既にダストを覆う氷として豊富に存在 (Whittet+1983) (1 M<sub>sun</sub>の分子雲コア → ~1000 M<sub>earth</sub>の水)
- ➤ 酸素の主要な形態 (~40 % of volatile oxygen, Whittet 2007)



## 反応ネットワークモデル

• 基礎方程式

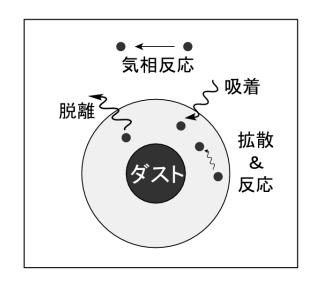
$$\dot{n}_i^{(x)} = \underline{P_i^{(x)} - L_i^{(x)}}$$
化学反応



- 多層モデル(e.g., Taquet+2012)
  - 気相 氷表面 <sup>相互作用</sup> 非一様な氷マントル

 $n_i$ :化学種 i の数密度

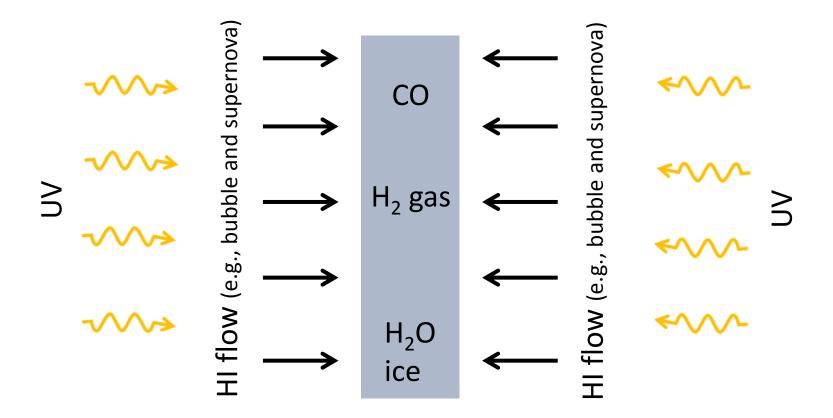
x: 気相、氷表面、氷マントル



- 単層モデル(Hasegawa & Herbst 1992)

  - 「 気相 ¬ 一様な氷マントル−

## Molecular cloud formation

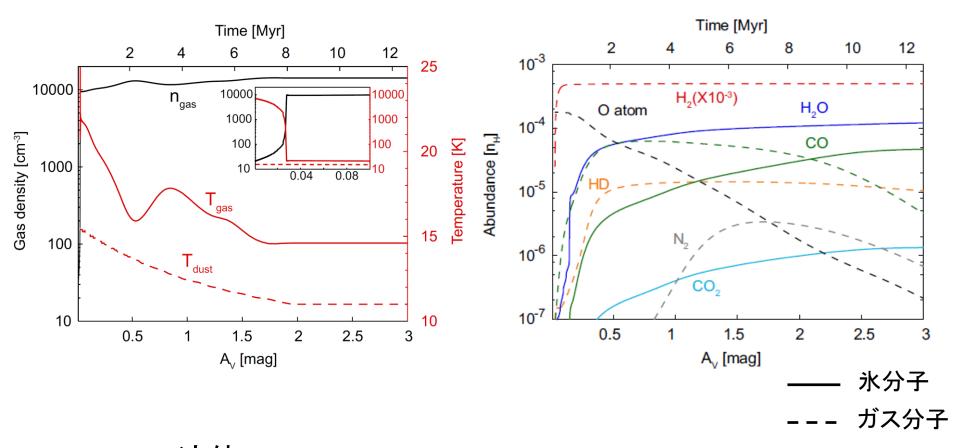


Accumulation of HI gas by accretion flows

→ molecular cloud formation

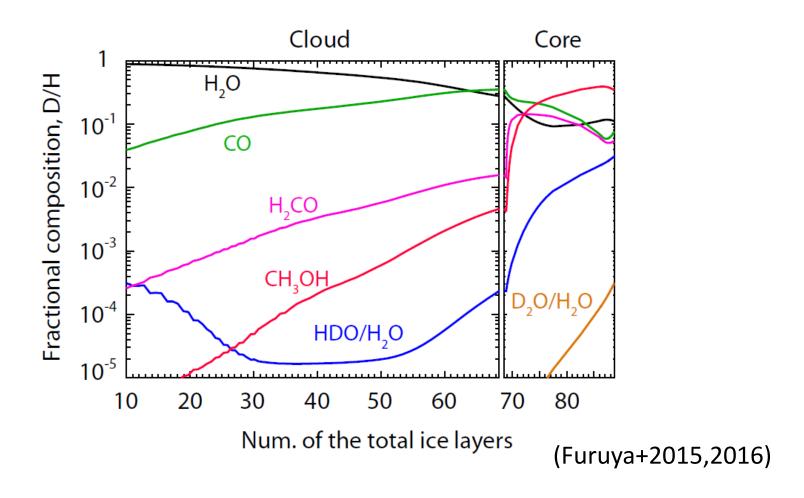
(e.g., Hartmann+2001; Inoue&inutsuka 2012)

## Atomic-to-molecular transitions

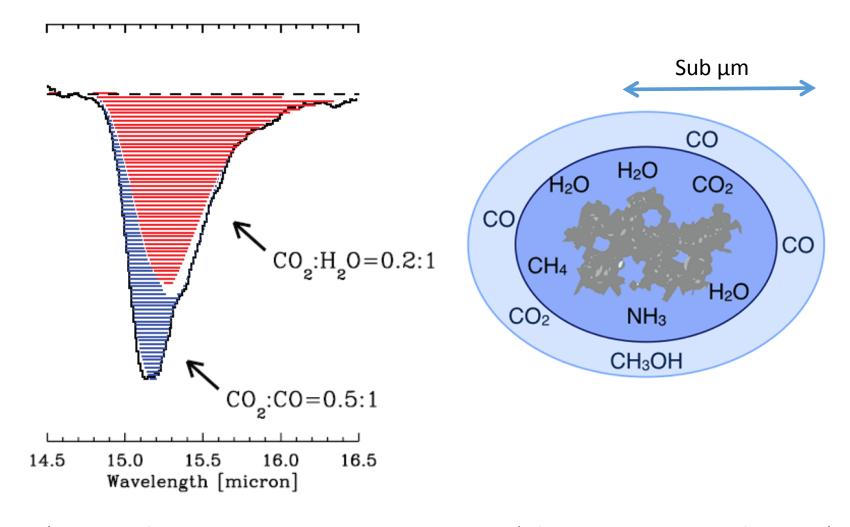


- 1D流体 + heating/cooling + gas-phase chemistry (Bergin+04)
- Post-processing gas-ice chemistry
- 柱密度増加→星間紫外線の遮蔽→原子から分子へ

## Ice layered structure



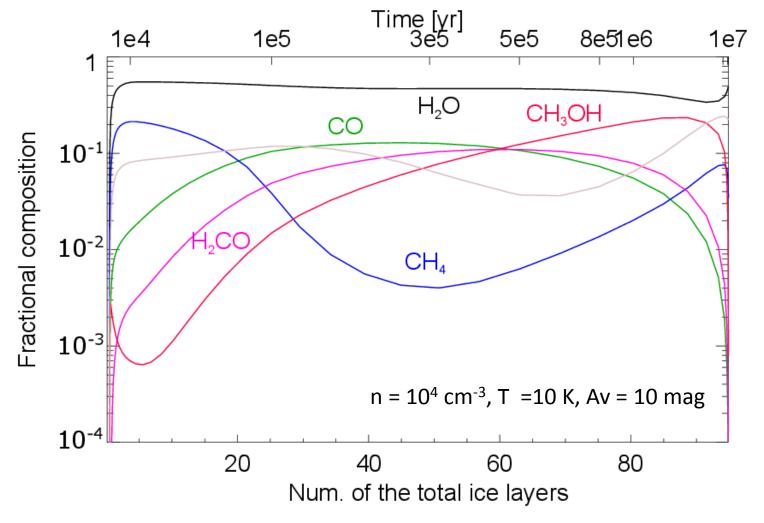
## Layered ice structure in the ISM ice



(Pontoppidan+ 2008; Ehrenfreund+1997)

(Oberg 2016; Pontoppidan 2006)

## Pseudo-time dependent model of dense clouds



(see e.g., Garrod & Pauly 2011)

## Intermediate summary

- Water is formed before the stellar birth as ice
- Chemical (and Isotopic) compositions in the ISM ice are inhomogeneous
  - preserves memory of the physical and chemical evolution

 $H_2O$ 

NH<sub>3</sub>

CH<sub>3</sub>OF

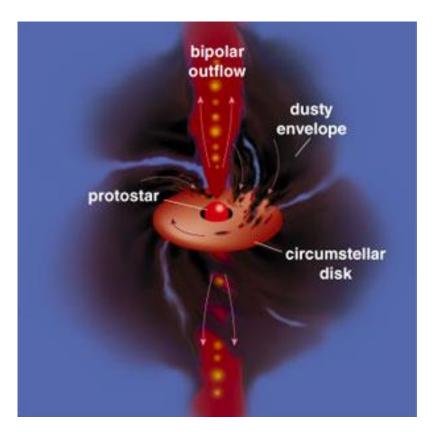
CO

CO

#### Contents

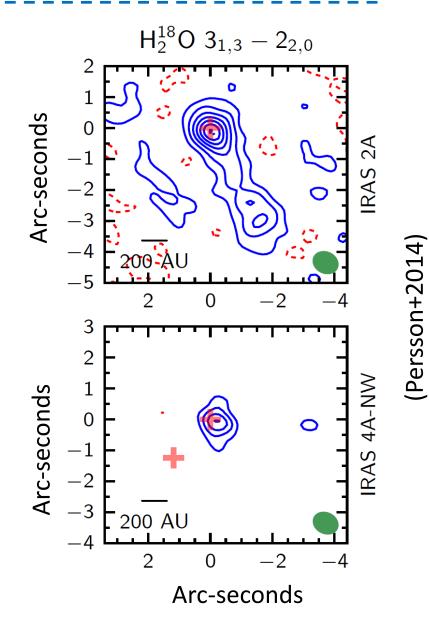
- Water chemistry in star-forming regions
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## Hot water near protostars

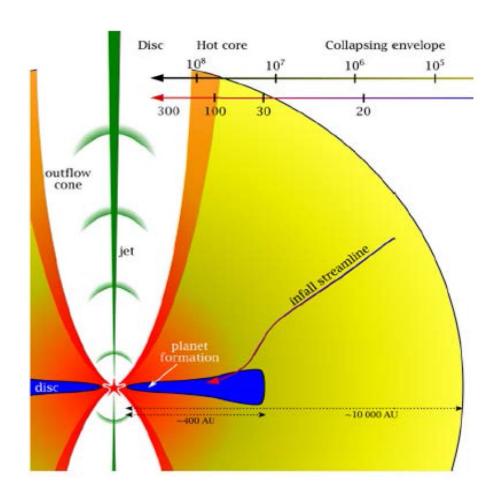


(Greene 2001)

Open question:
Where do the water emissions originate from?
Envelope, disk, or both?



## Ice delivery from cores to disks

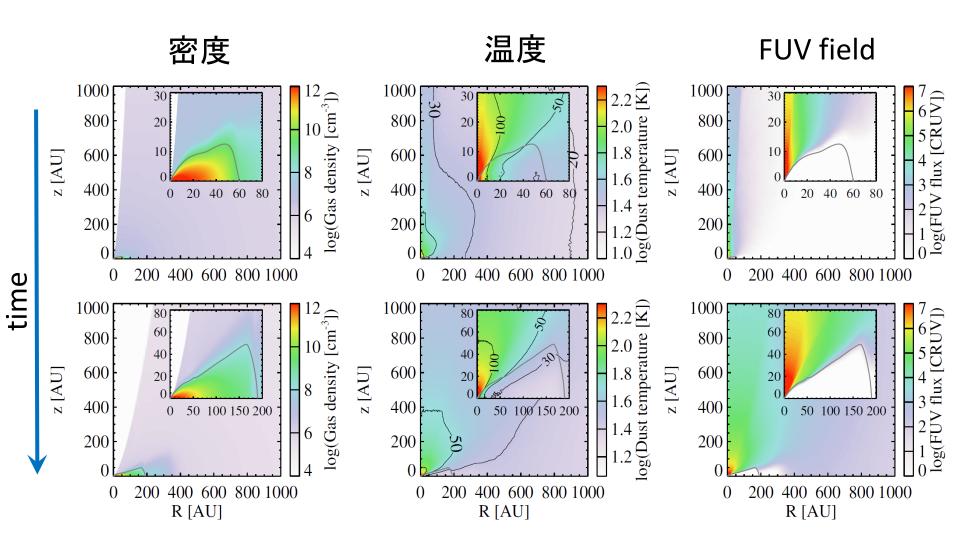


(Herbst & van Dishoeck 2009)

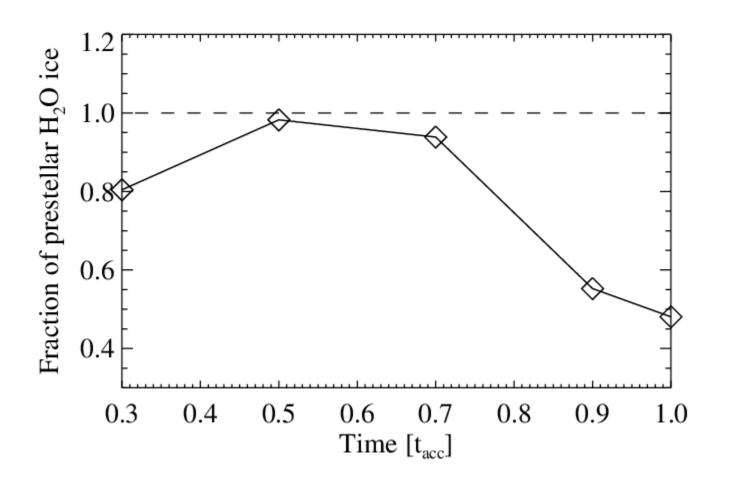
#### 2-D collapsing core model to the disk formation (Visser+2009, 2011)

- Density & velocity
   -semi analytical models
   (Terebey+1984; Cassen&Moosman
   1981; α-disk model)
- Temperature & UV field
   radiative transfer
   (RADMC-3D;
   Dullemond&Dominik 2004)
- Chemistry is solved along trajectories of in-falling materials

## Physical evolution



## Bulk-disk averaged H<sub>2</sub>O ice abundance



 $0.01~M_{sun}~disk \rightarrow several~M_{earth}~water~ice,~being~inherited$  from the prestellar stages

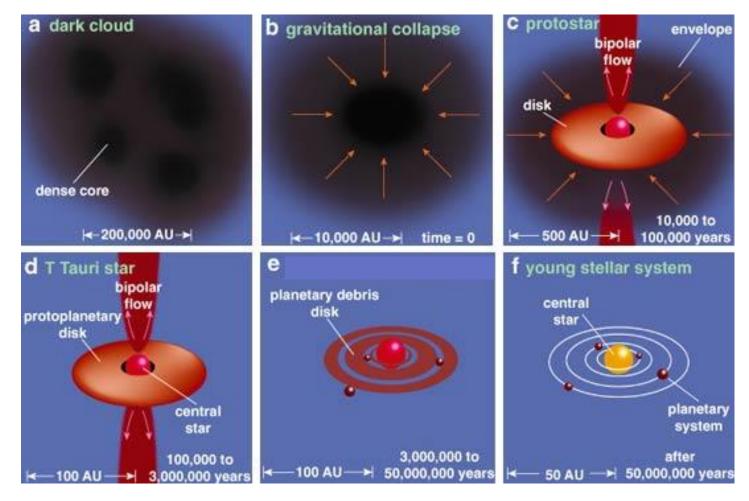
## Intermediate summary

- Forming disks contain lots of interstellar H<sub>2</sub>O ice
- → favorable for the formation of icy planetesimals (and planets)

#### Contents

- Water chemistry in star-forming regions
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## Follow the water trail



(Greene 2001)

Where did our water come from?

## Deuterium fractionation



Probe of the formation environments of molecules

- $\rightarrow$  The [D/H] elemental ratio in the local ISM  $\sim 10^{-5}$  (Linsky 2003)
- ➤ Molecules formed at low temperatures, XD/XH >> 10<sup>-5</sup>

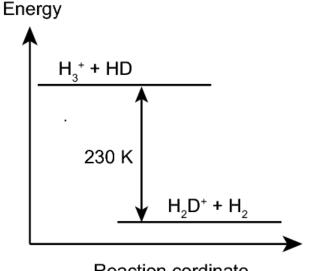
$$H_3^+ + HD \implies H_2D^+ + H_2 + 230 \text{ K}$$

$$\rightarrow$$
 H<sub>2</sub>D<sup>+</sup>/H<sub>3</sub><sup>+</sup>>> 10<sup>-5</sup>

$$H_2D^+ + e \rightarrow D + H_2$$

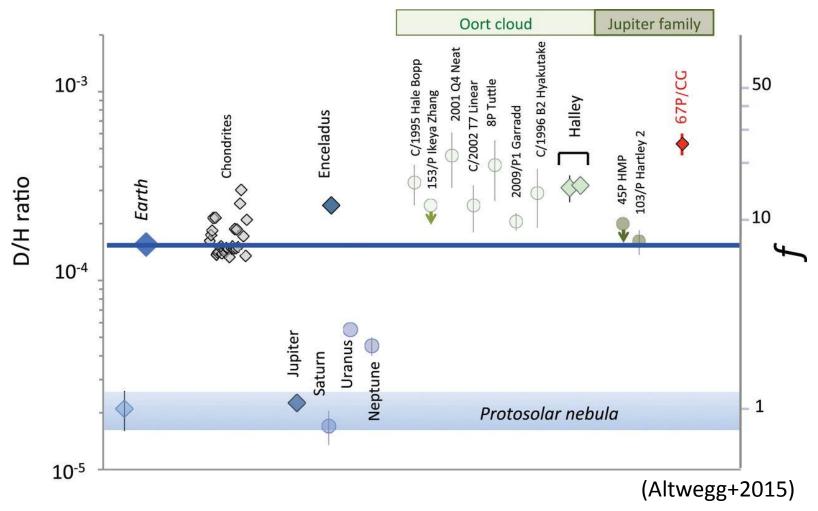
- High atomic D/H
- High D/H in Icy molecules

(e.g., Watson+1976; Tielens 1983)



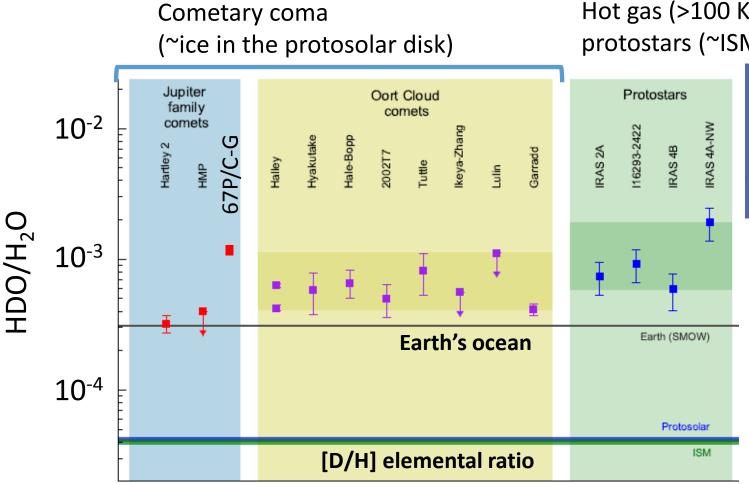
- Reaction cordinate
- > CO freeze-out, higher density, lower H<sub>2</sub> o/p
  - enhanced deuterium fractionation

## HDO/H<sub>2</sub>O measurements



Water is enriched in deuterium → formed at low temperatures → In molecular clouds? Or in disks?

## HDO/H<sub>2</sub>O measurements



Hot gas (>100 K) around protostars (~ISM ice)

(Mumma & Charnley 2011; Altwegg+2015; Persson+ 2014)

 The ISM and cometary water have the similar HDO/H<sub>2</sub>O > prestellar inheritance? BUT...

## Prestellar inheritance of water?

#### **BUT**

- Variations of HDO/H<sub>2</sub>O in pristine materials
   in the solar system -> how?
- In-situ formation in disks can also explain the cometary HDO/H<sub>2</sub>O ratios

(e.g., Furuya+2013; Albertsson+2014)

#### Yet inconclusive

(e.g., Geiss & Reeves 1981; Aikawa & Herbst 1999)

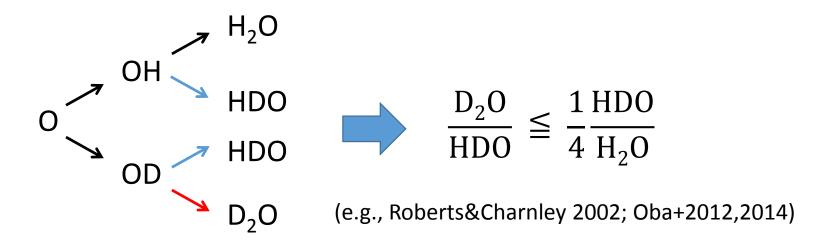
## Heavy water (D<sub>2</sub>O)

- In the hot inner regions around a protostar
   → D<sub>2</sub>O/HDO = 7\*HDO/H<sub>2</sub>O (Coutens+ 2014)
- The ratio of D<sub>2</sub>O/HDO to HDO/H<sub>2</sub>O reflects
   the ice layered structure
   (i.e., preserves memory of the physical and chemical evolution)
- The ratio is a better probe to distinguish between the two cases (inheritance vs. in-situ formation in disks)
  - → D<sub>2</sub>O observations toward comets are crucial

(Furuya+ in prep.)

## Constant atomic D/H case

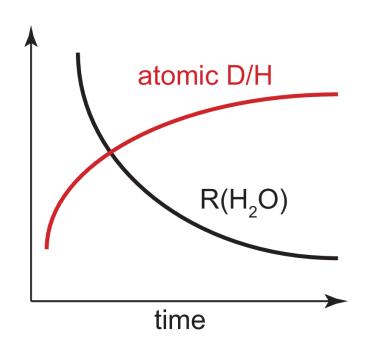
i.e., assumes quasi-steady state



contradicts with the observational relation  $\frac{D_2O}{HDO} \sim 7 \frac{HDO}{H_2O}$ 

→ time-dependency of the atomic D/H

## Consider chemical evolution



R(HDO)  $\propto$  R(H<sub>2</sub>O)\*(atomic D/H) R(D<sub>2</sub>O)  $\propto$  R(H<sub>2</sub>O)\*(atomic D/H)<sup>2</sup>

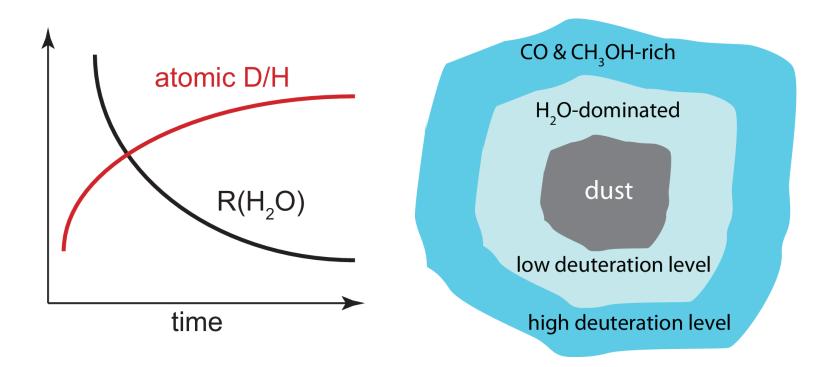
→ Production rates of HDO and D<sub>2</sub>O do not necessarily follow that of H<sub>2</sub>O

If the production of HDO and D<sub>2</sub>O are dominated in late times

 $D_2O/HDO$  in the whole ice ~ atomic D/H in late times  $HDO/H_2O$  in the whole ice << atomic D/H in late times

$$\frac{D_2O}{HDO} \gg \frac{HDO}{H_2O}$$

## Consider chemical evolution

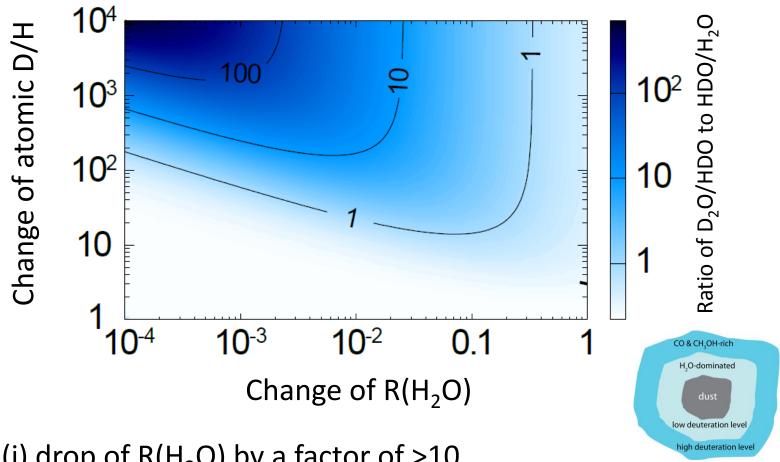


If the production of HDO and D<sub>2</sub>O are dominated in late times

 $D_2O/HDO$  in the whole ice ~ atomic D/H in late times  $HDO/H_2O$  in the whole ice << atomic D/H in late times

$$\frac{D_2O}{HDO} \gg \frac{HDO}{H_2O}$$

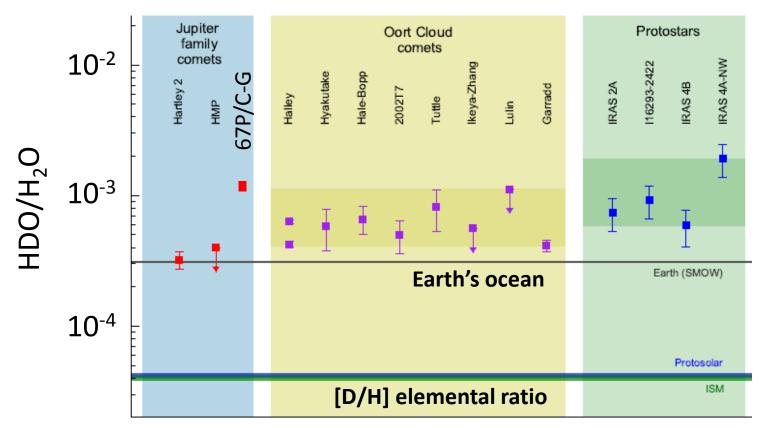
#### Required conditions for reproducing the observations



- (i) drop of  $R(H_2O)$  by a factor of >10
- (ii) enhancement of the atomic D/H by a factor of >100
  - → very inhomogeneous

## Ratio of D<sub>2</sub>O/HDO to HDO/H<sub>2</sub>O

- The high ratio likely characterizes the ISM ice
  - → D<sub>2</sub>O observations toward comets are crucial for studying the origin of cometary water



(Mumma & Charnley 2011; Altwegg+2015; Persson+ 2014)

## Summary

- Water is formed before the stellar birth
- Our model suggests that forming disks contain lots of interstellar H<sub>2</sub>O ice
  - → favorable for the formation of icy planetesimals (and planets)
- D<sub>2</sub>O observations toward comets are crucial for testing the model prediction