# PH 112: Quantum Physics and Applications

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Week 03, Lecture 2 : Schrodinger Equation Spring 2023

#### Matter Waves: Recap

- The wave character of light and the photon character of light are both but different manifestations of the same thing.
- de Broglie: If waves can mimic particles, then particles can mimic waves

$$\lambda = \frac{h}{m \, v} = \frac{h}{p}$$

- de Broglie expression is only qualitative and does not help us to make precise quantitative predictions!
- The idea of a perfectly predictable universe cannot be true!

$$\Delta x \Delta p \ge \frac{\hbar}{2}$$

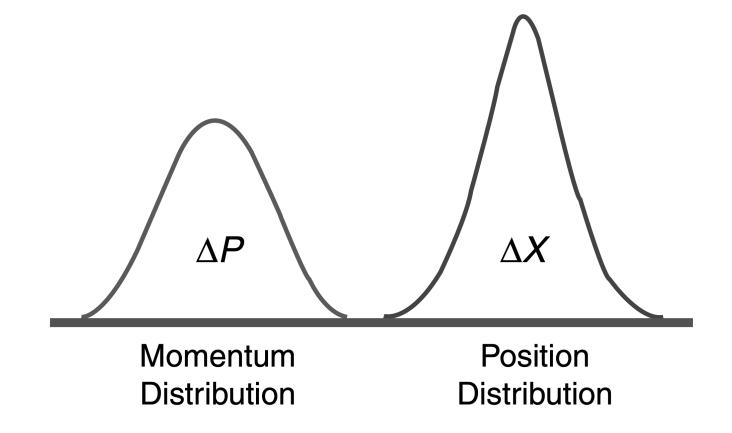
$$\Delta E \Delta t \geq \frac{\hbar}{2}$$

#### Matter Waves: Recap

• The idea of a perfectly predictable universe cannot be true!

$$\Delta x \Delta p \ge \frac{\hbar}{2}$$

$$\Delta E \Delta t \geq rac{\hbar}{2}$$



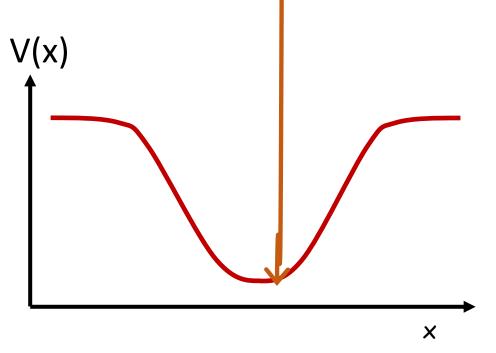
# What equation matter waves satisfy?

- Free particles:  $E = \hbar \omega$   $p = \hbar k$
- Consider a particle in a 1-D potential V(x)

$$E = \frac{p^2}{2m} + V(x)$$

- Questions we need answers:
  - 1. How do we describe waves for such a particle?
  - 2. What happens to the particle in lowest energy state?

Classically, a particle in the lowest energy state would sit right at the bottom of the well.

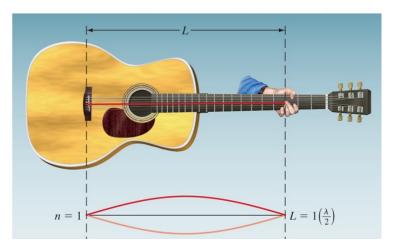


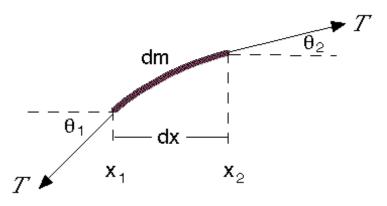
#### **Examples:**

- 1. Electron in hydrogen atom
- 2. Electron in a solid crystal
- 3. Electron in a nanostructure 'quantum dot'
- 4. Proton in the nuclear potential inside the nucleus

# Classical Wave equation

# Wave equation for a string





Short section of a stretched string stretched.

• Consider a string with uniform mass per length  $(\mu)$  that supports oscillations:

$$y(x,t) = A\sin(k[x-vt])$$

- We can derive the equation of motion of a string from Newton's laws of motion.
- String stretched by tension (T), Newton's law along y-axis

$$\mathrm{F_y} = \mathrm{ma_y}$$

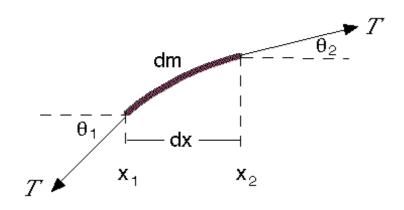
The sum of forces in the y-direction is

$$F_{\rm v} = T\sin\theta_2 - T\sin\theta_1$$

# Wave equation for a string

We now use small angle approximation,

$$\sin \theta \cong \tan \theta = \partial y / \partial x$$



• We have:

$$F_{y} = T \left( \frac{\partial y}{\partial x} \right)_{2} - T \left( \frac{\partial y}{\partial x} \right)_{1}$$

• Total force depends on the difference in slope between the two ends. For straight string, two forces add up to zero.

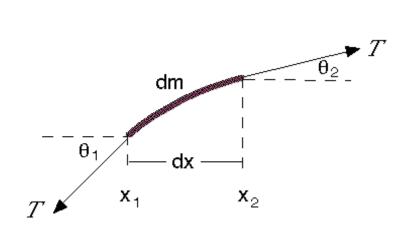
section of the stretched string

- For length dx,  $dm = \mu dx$
- Force along y-direction becomes

$$F_{y} = \mu dx \frac{\partial^{2} y}{\partial t^{2}}$$

# Wave equation for a string

• From the two equations, we get



$$T\left(\left(\frac{\partial y}{\partial x}\right)_2 - \left(\frac{\partial y}{\partial x}\right)_1\right) = \mu dx \frac{\partial^2 y}{\partial t^2}$$

• Rewriting we get,  $\frac{\partial^2 y}{\partial t^2} = \frac{T}{t} \frac{\left(\frac{\partial y}{\partial x}\right)}{t}$ 

• RHS is the rate of change of first derivative w.r.t x:

Short section of string stretched along x-axis.

$$\frac{\partial^2 y(x,t)}{\partial t^2} = v^2 \frac{\partial^2 y(x,t)}{\partial x^2} \qquad v^2 = \frac{T}{\mu}$$

## Solution to wave equation

- Wave equation is a partial differential equation y(x, t).
- One technique is to choose a likely function, test to see if it is a solution and modify it!
- We know that sine waves can propagate in a one dimensional medium like a string!

$$y(x,t) = Asin(kx - \omega t)$$

• In taking the partial derivative with respect to t, we hold x constant and vice versa.

$$\frac{\partial y}{\partial t} = -\omega A \cos(kx - \omega t) \quad \frac{\partial y}{\partial x} = kA \cos(kx - \omega t) 
\frac{\partial^2 y}{\partial t^2} = -\omega^2 A \sin(kx - \omega t) \quad \frac{\partial^2 y}{\partial x^2} = -k^2 A \sin(kx - \omega t)$$

# Solution to wave equation

•  $y(x,t) = Asin(kx-\omega t)$  is a solution of the wave equation if

$$\frac{T}{\mu} = \left(\frac{\omega}{k}\right)^2$$

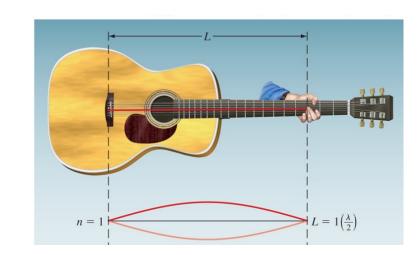
$$\frac{T}{\mu} = \left(\frac{\omega}{k}\right)^2 \qquad \omega = 2\pi f; k = \frac{2\pi}{\lambda} \Longrightarrow v = \sqrt{\frac{T}{\mu}} = f\lambda$$

Not all frequencies are possible, only that satisfy condition  $y_{\omega}(0)=0=y_{\omega}(L)$ 

$$y_{\omega}(0) = 0 = y_{\omega}(L)$$

- This leads to  $\ \omega = \omega_n = v k_n = v \frac{\pi}{\tau} n$
- General solution:

$$y(x,t) = \sum_{n=1}^{\infty} A_n \sin(k_n x) e^{-i\omega_n t}$$



### Relook at the classical wave equation and solution

Structure of the wave equation has ensured that the expected wave function is indeed a solution

$$y(x,t) = A\cos(kx - \omega t) \Longrightarrow \frac{\partial^2 y(x,t)}{\partial x^2} = \frac{1}{v^2} \frac{\partial^2 y(x,t)}{\partial t^2}$$

$$\frac{\partial^2 y(x,t)}{\partial x^2} \longrightarrow k^2$$

$$\frac{\partial^2 y(x,t)}{\partial t^2} \longrightarrow \omega^2$$

$$\Longrightarrow v^2 = \frac{\omega^2}{k^2} \Longrightarrow v = f\lambda$$

# Matter wave equation

### Matter wave equation

 Schroedinger was interested in obtaining relationship between wave function and wave equation.

• if de broglie was correct, particles behave like waves then these de Broglie wave functions should be solutions to a wave equation.

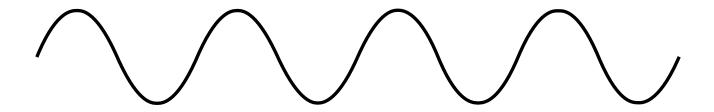


#### Questions:

- 1. What form should this wave equation take?
- 2. Will it have the same form as classical wave equation?
- 3. If not, why classical wave equation can not describe matter waves?

### Constructing a quantum wave equation

 Consider a freely moving particle is to be described by an associated wave then a reasonable first guess would be to



$$\Psi(x,t) = A\cos(kx - \omega t)$$

• According to de Broglie and Einstein relation:

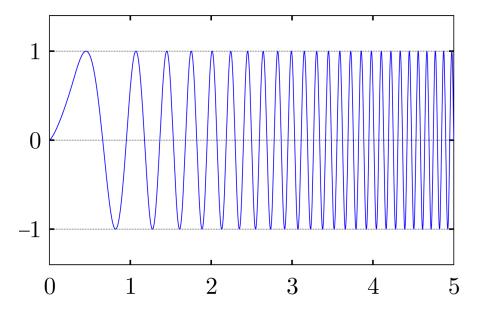
Constant 
$$p \longleftrightarrow Constant \lambda Constant E \longleftrightarrow Constant f$$

• However, if a force acts on a particle

Force 
$$\stackrel{F=\frac{dp}{dt}}{\Longrightarrow}$$
 Change in  $p \stackrel{\lambda=\frac{h}{p}}{\Longrightarrow}$  Change in  $\lambda$ 

#### Constructing a quantum wave equation

- If the wavelength changes rapidly, the concept of wavelength is not well defined.
- If matter wave changes rapidly, it's difficult to define variable wavelength, since separation between adjacent maxima is not equal to the separation between adjacent minima!



• If the linear momentum of a particle is not of constant magnitude, then more complicated functions than  $\Psi(x,t) = A\cos(kx - \omega t)$ 

We need a wave equation that determine  $\Psi(x,t)$  for any physical situation.

### Requirements of quantum wave equation

1. To be consistent with de Broglie-Einstein relations

$$E=\hbar\omega \qquad \lambda=rac{h}{p}$$

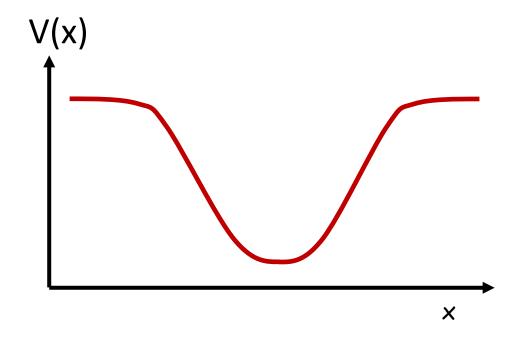
- 2.  $\Psi(x,t)$  provide information about the force acting on the associated particle. Specifying Potential energy corresponding to the force.
- 3. To be consistent with Energy conservation  $E = \frac{p^2}{2m} + V(x,t) \Longrightarrow F = -\frac{\partial V}{\partial x}$
- 4. Superposition principle of waves  $\Psi(x,t)$  must be satisfied
- 5. For free particle V(x)=0, solution should correspond to traveling wave  $V(x,t)=0 \Rightarrow F=-\frac{\partial V}{\partial x}=0 \Rightarrow p, E=constant \Rightarrow \lambda, f=constant$

### Matter Wave equation: Step I

Consider a particle in a 1-D potential V(x).

 Combining the first and third requirements, we have

$$E = \hbar\omega(k) = \frac{\hbar^2 k^2}{2m} + V(x)$$



 Fourth requirement implies quantum wave equation must be a linear differential equation (like classical wave equation)!

### Matter Wave equation: Step II

Fifth requirement, implies the solutions must be of the form

$$\hbar\omega(k) = \frac{\hbar^2 k^2}{2m} \implies \Psi(x,t) = \sin(kx - \omega t) \quad \text{or} \quad e^{i(kx - \omega t)}$$

Note that

 $rac{\partial^2}{\partial x^2}$  leads to  $k^2$  factor  $rac{\partial}{\partial t}$  leads to  $\omega$  factor

 Thus we might guess a wave equation to be

$$\alpha \frac{\partial \Psi(x,t)}{\partial t} = \beta \frac{\partial^2 \Psi(x,t)}{\partial x^2} + V(x,t)\Psi(x,t)$$

 $\alpha$ ,  $\beta$  are constants that are fixed such that wavefunction is consistent with de Broglie-Einstein relations.

## Time-dependent Schrodinger Equation

• Constants  $\alpha$  and  $\beta$  can be obtained using the exponential free particle solution.

$$-\frac{\hbar^2}{2m}\frac{\partial^2 \Psi}{\partial x^2} + V(x,t)\Psi = i\hbar\frac{\partial \Psi}{\partial t}$$



- Erwin Schrodinger proposed this equation in 1926 to describe the time and position dependence of the matter waves.
- This is referred to as time-dependent Schrodinger equation.

#### Time-independent Schrodinger equation

Suppose the potential is independent of time i.e. V(x, t) = V(x) then TDSE is:

$$-\frac{\hbar^2}{2m}\frac{\partial^2 \psi}{\partial x^2} + V(x)\psi = E\psi$$

#### In this course:

- 1. We will ONLY focus on time-independent Schrodinger equation.
- 2. We will restrict ourselves to 1-D problems.
- It is difficult to obtain solutions even for 1-D potentials!

#### Time-Dependent to time-independent Schrödinger equation

Suppose the potential is independent of time i.e. V(x, t) = V(x) then TDSE is:

LHS involves variation of  $\psi$  with t while RHS involves variation of  $\psi$  with x. Hence look for a separated solution:

Leading to

Now divide by  $\psi T$ :

LHS depends only upon x, RHS only on t. True for all x and t so both sides must equal a constant, E (E = separation constant). We then have

$$i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + V(x)\Psi$$

$$\Psi(x,t) = \psi(x)T(t)$$

$$-\frac{\hbar^2}{2m}T\frac{\partial^2\psi}{\partial x^2} + V(x)\psi T = i\hbar\psi\frac{\partial T}{\partial t}$$

$$-\frac{\hbar^2}{2m}\frac{1}{\psi}\frac{\partial^2\psi}{\partial x^2} + V(x) = i\hbar\frac{1}{T}\frac{\partial T}{\partial t}$$

$$i\hbar \frac{1}{T} \frac{\partial T}{\partial t} = E$$

$$-\frac{\hbar^2}{2m} \frac{1}{\psi} \frac{\partial^2 \psi}{\partial x^2} + V(x) = E$$

#### Time-independent Schrödinger equation

Solving T(t) equation, we have

$$i\hbar \frac{1}{T} \frac{dT}{dt} = E \implies \frac{dT}{T} = -\frac{iE}{\hbar} dt \implies T(t) = Ae^{-iEt/\hbar}$$

This is exactly like a wave  $e^{-i\omega t}$  with  $E = \hbar \omega$ . Therefore T(t) depends upon the energy E.

To find out what the energy actually is we must solve the space part of the problem....

The space equation becomes

$$-\frac{\hbar^2}{2m}\frac{\partial^2 \psi}{\partial x^2} + V(x)\psi = E\psi \qquad \text{or} \qquad \hat{H}\psi = E\psi$$

This is the time independent Schrödinger equation (TISE).

The TISE can often be very difficult to solve – it depends on V(x)!

# Classical wave solutions satisfy?

## Solutions of Schrodinger Equation

Question: Whether the solutions of classical wave equation are also solutions to Schrodinger equation?  $\Psi(x,t) = A\sin(kx-\omega t)$ 

**Answer: No** 

Proof:

$$\frac{\partial \Psi}{\partial t} = -\omega A \cos(kx - \omega t)$$

$$\frac{\partial \Psi}{\partial x} = kA\cos(kx - \omega t)$$

$$\frac{\partial^2 \Psi}{\partial x^2} = -k^2 A^2 \cos(kx - \omega t)$$

$$-i\hbar\omega A\cos(kx-\omega t) \neq \left(\frac{\hbar^2 k^2}{2m} + V\right)A\sin(kx-\omega t)$$

# Solutions of Schrodinger Equation

Question: What about 
$$\Psi(x,t) = A \exp\left(i[kx-\omega t]
ight)$$
 ? Answer: Yes

$$\frac{\partial \Psi}{\partial t} = -i\omega A e^{i(kx - \omega t)}$$

$$\frac{\partial \Psi}{\partial x} = ikAe^{i(kx - \omega t)}$$

$$\frac{\partial \Psi}{\partial t} = -i\omega A e^{i(kx - \omega t)}$$
$$\frac{\partial \Psi}{\partial x} = ikA e^{i(kx - \omega t)}$$
$$\frac{\partial^2 \Psi}{\partial x^2} = -k^2 A e^{i(kx - \omega t)}$$

$$-i\hbar(i\omega) = \left(\frac{\hbar^2 k^2}{2m} + V\right)$$

$$E = \frac{p^2}{2m} + V$$
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### The solution is complex!

 $\Rightarrow$ .  $\Psi(x,t)$  *cannot* be a physical wave (e.g. electromagnetic waves).

 $-i\hbar(i\omega) = \left(\frac{\hbar^2 k^2}{2m} + V\right)$  How to relate  $\Psi(x,t)$  to measurements on a system?

# Summary

Time-dependent Schrodinger equation

Time-independent Schrodinger equation

$$-\frac{\hbar^2}{2m}\frac{\partial^2 \Psi}{\partial x^2} + V(x,t)\Psi = i\hbar \frac{\partial \Psi}{\partial t}$$

$$-\frac{\hbar^2}{2m}\frac{\partial^2 \psi}{\partial x^2} + V(x)\psi = E\psi$$

#### Recommended Reading

Schrödinger equation sections 6.1,

6.2 and 6.3

