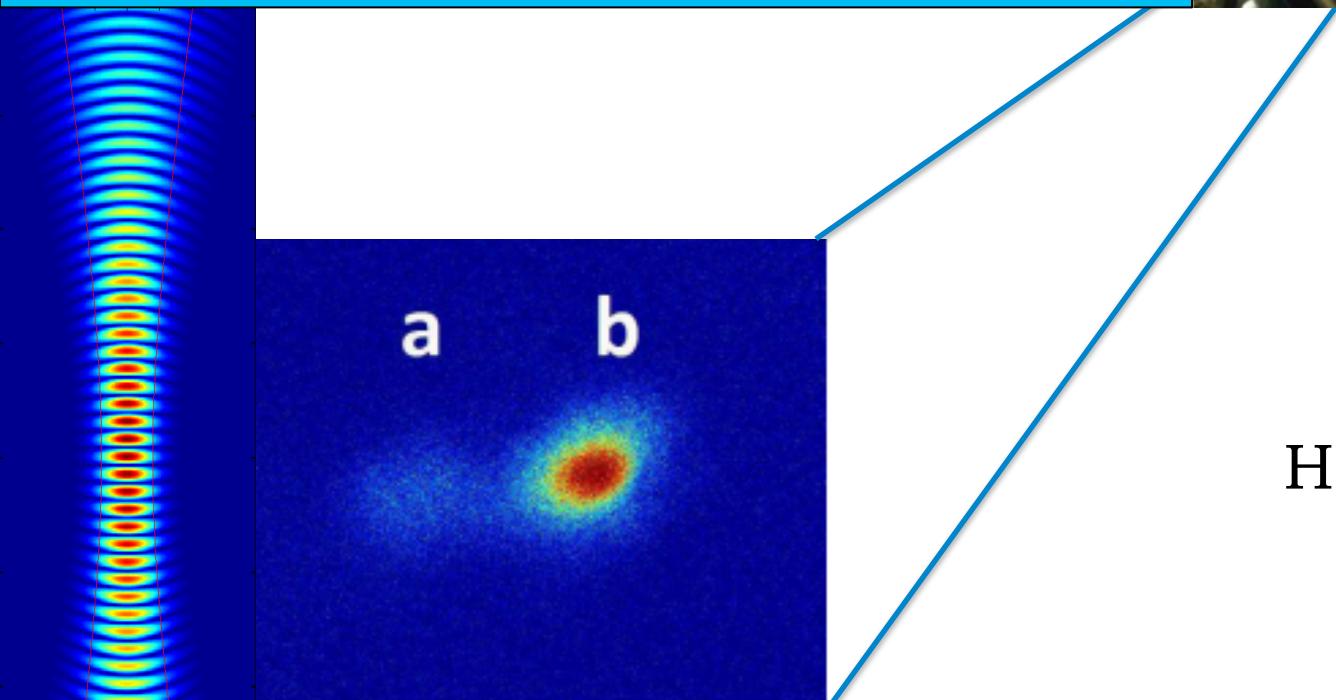
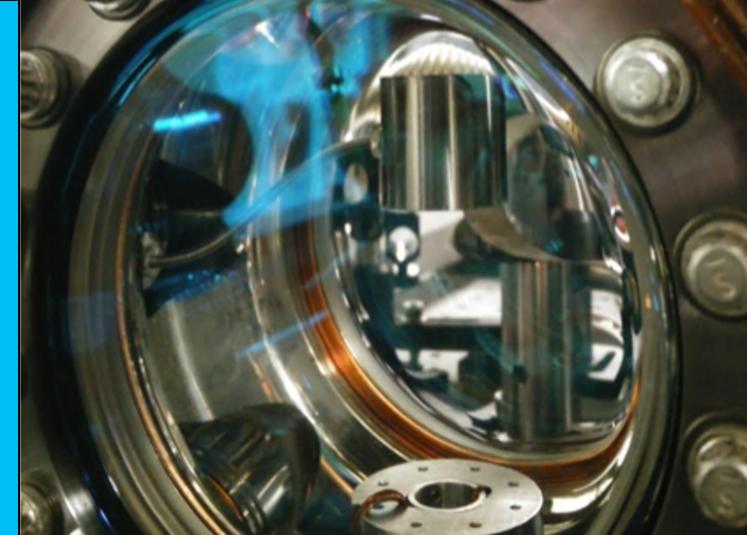


Blind analyses in precision measurements



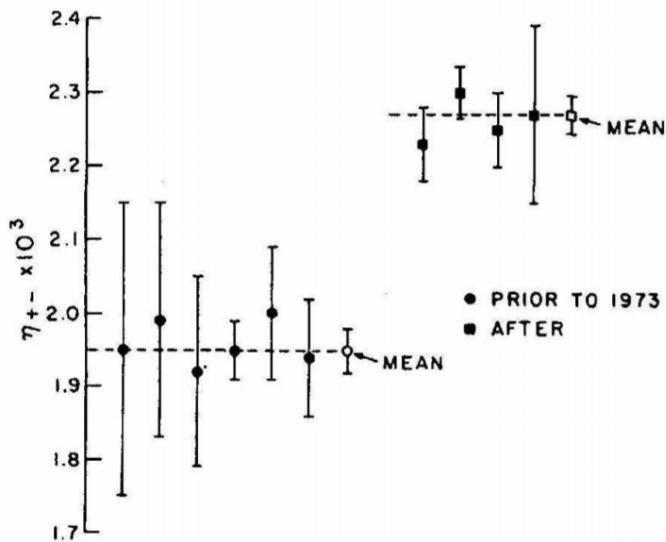
Holger Müller

Contents

1. Motivation & Introduction
2. Millikan
3. Dunnington
4. EDM Experiments
5. G
6. Fine structure constant

Motivation

CP violation



Speed of light

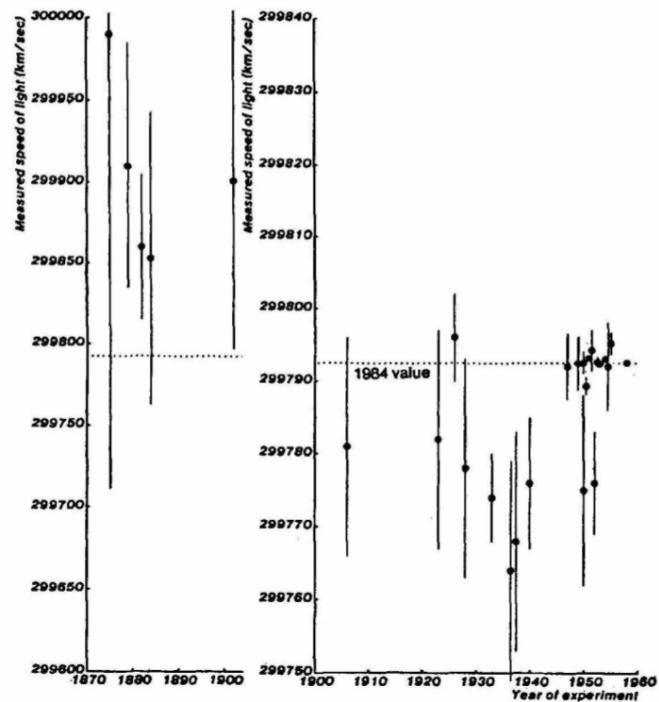
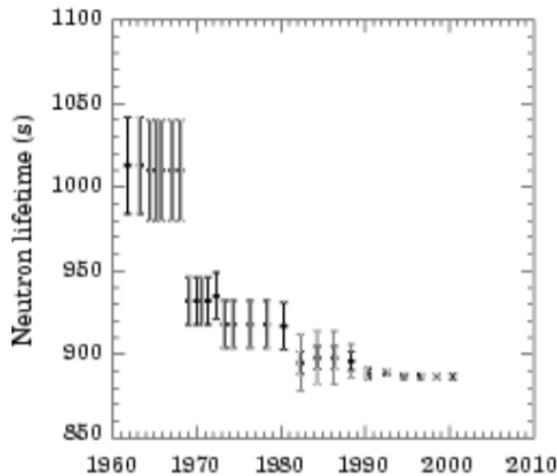


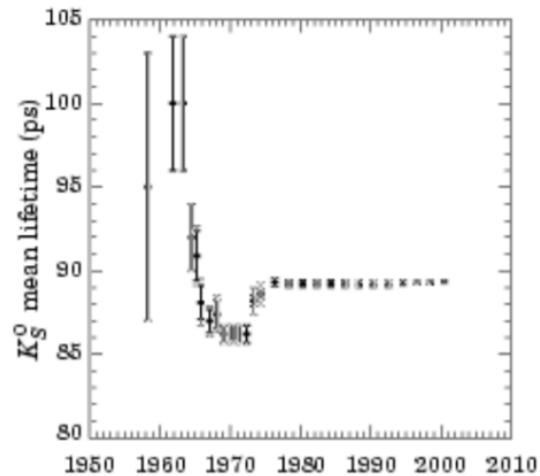
FIG. 1: Measurements of $|\eta_{+-}|$ in order of their year of publication. Reprinted with permission from A. Franklin, "Forging, cooking, trimming, and riding on the bandwagon," Am. J. Phys. **52**, 786-793 (1984), copyright 1984, American Association of Physics Teachers.

Motivation

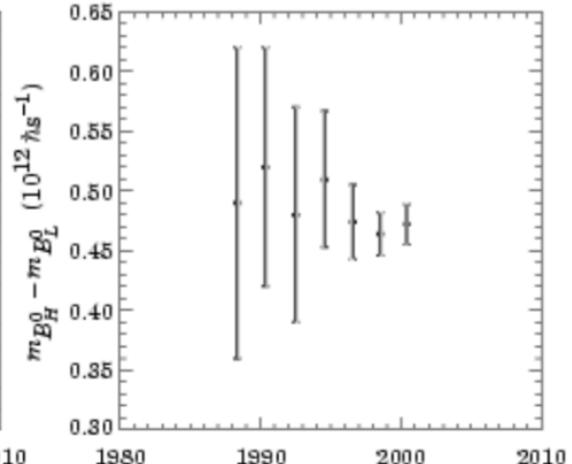
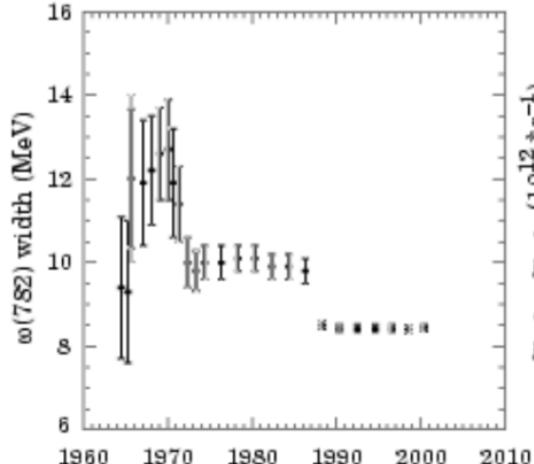
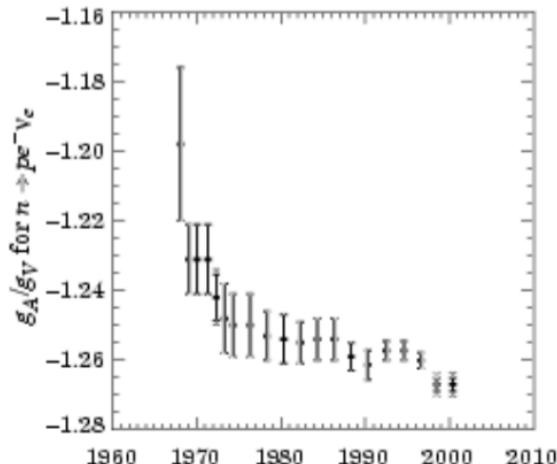
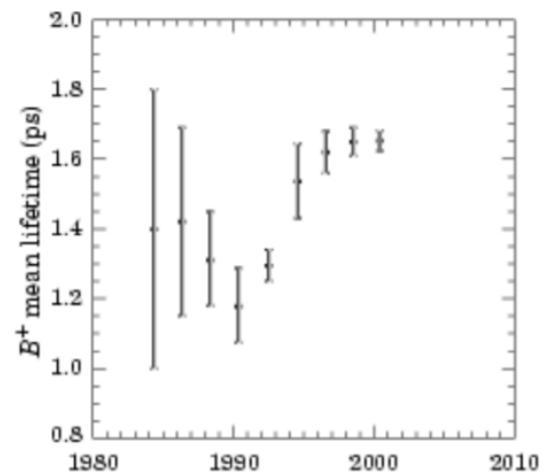
Neutron lifetime



Kaon lifetime



B+ lifetime



Accuracy versus precision



Low accuracy
Low precision



Low accuracy
High precision



High accuracy
High precision

Millikan oil drop experiment

Millikan measured the charge on an electron by an experiment with falling oil drops, and got an answer which we now know not to be quite right. It's a little bit off because he had the incorrect value for the viscosity of air. It's interesting to look at the history of measurements of the charge of an electron, after Millikan. If you plot them as a function of time, you find that one is a little bit bigger than Millikan's, and the next one's a little bit bigger than that, and the next one's a little bit bigger than that, until finally they settle down to a number which is higher.

Why didn't they discover the new number was higher right away? It's a thing that scientists are ashamed of—this history—because it's apparent that people did things like this: When they got a number that was too high above Millikan's, they thought something must be wrong—and they would look for and find a reason why something might be wrong. When they got a number close to Millikan's value they didn't look so hard. And so they eliminated the numbers that were too far off, and did other things like that ...[\[10\]\[11\]](#)

Die atomaren Konstanten e , e/m_0 und h^*).

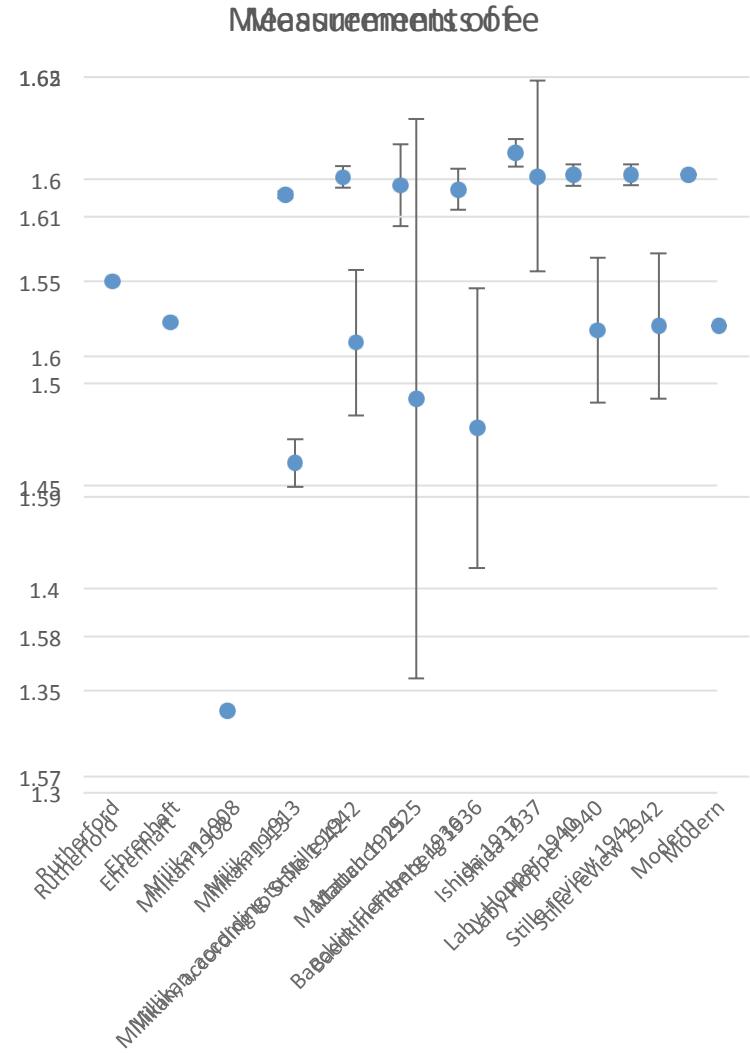
Von U. Stille in Braunschweig.

Mit 1 Abbildung. (Eingegangen am 1. Dezember 1942.)

Entsprechend den in einer früheren Veröffentlichung aufgestellten allgemeinen Gesichtspunkten werden die zur Bestimmung der Atomkonstanten verfügbaren experimentellen Daten neu zusammengestellt und diskutiert. In einem I. Teil sind die hierfür erforderlichen Hilfsgrößen zusammengefaßt. Die Zahlenwerte für die elektrischen Umrechnungsfaktoren werden dabei einer weiteren Veröffentlichung entnommen, die dieses Thema gesondert behandelt. Als zur Zeit experimentell gesicherte Werte sind für die Vakuumlichtgeschwindigkeit c_0 , die Induktionskonstante μ_0 , die Influenzkonstante ε_0 , die spezifische Molekülzahl N_L und die spezifische Ionenladung F folgende anzunehmen: $c_0 = (2,997\,77 \pm 0,000\,20) \cdot 10^8$ m/sec, $\mu_0 = (1,256\,02 \pm 0,000\,02) \cdot 10^{-6}$ Volt · sec/Amp · m, $\varepsilon_0 = (8,859\,4 \pm 0,001\,4) \cdot 10^{-12}$ Amp · sec/Volt · m, $N_L = (6,024\,3 \pm 0,004\,8) \cdot 10^{26}$ Kilomol⁻¹ und $F = (9,652\,0 \pm 0,000\,6) \cdot 10^7$ Amp · sec/Kiloäquivalent, wobei die beiden letzten Zahlenwerte auf die physikalische Atomgewichtsskala bezogen sind. Im Teil II werden auf Grund des vorliegenden experimentellen Materials Zahlenwerte für die drei atomaren Konstanten abgeleitet. Es ergeben sich für die Elementarladung e , die spezifische Elektronenladung e/m_0 und das Plancksche Wirkungsquantum h : $e = (1,602\,2 \pm 0,001\,4) \cdot 10^{-19}$ Amp · sec, $e/m_0 = (1,759 \pm 0,002) \cdot 10^{11}$ Amp · sec/kg und $h/e = (4,131\,2 \pm 0,002\,6) \cdot 10^{-15}$ Volt · sec bzw. hieraus $h = (6,619 \pm 0,010) \cdot 10^{-34}$ Watt · sec². Der aus den Messungen nach der Öltröpfchenmethode folgende e -Wert stimmt mit dem oben angegebenen, aus F und N_L folgenden innerhalb dessen Fehlergrenze

Millikan oil drop experiment

- There may never have been a long-lasting problem with the elementary charge.
- Controversy with Ehrenhaft (claimed “subelectrons”)
- 1942 review notes and corrects error
- Inexplicably, Millikan keeps defending his original value
- The problem isn't that the community keeps following false previous measurements, but that the authority figures don't like being corrected.
- Error 2% -> 0.5% [Franklin, A. (1997), *The Chemical Educator*. 2 (1): 1–14]



Frank Dunnington 1932 e/m measurement

- Two previous measurements disagree
- Electrons on circle must pass slits S during period $2\pi/\omega$ of rf drive
- Angle unknown to experimenter
- Phys. Rev 42, 734 (1932)

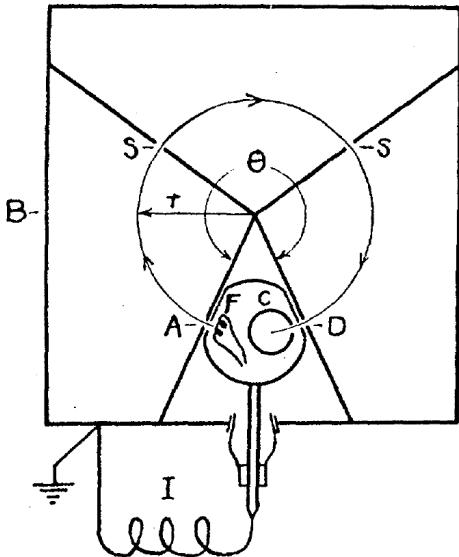


Fig. 1.

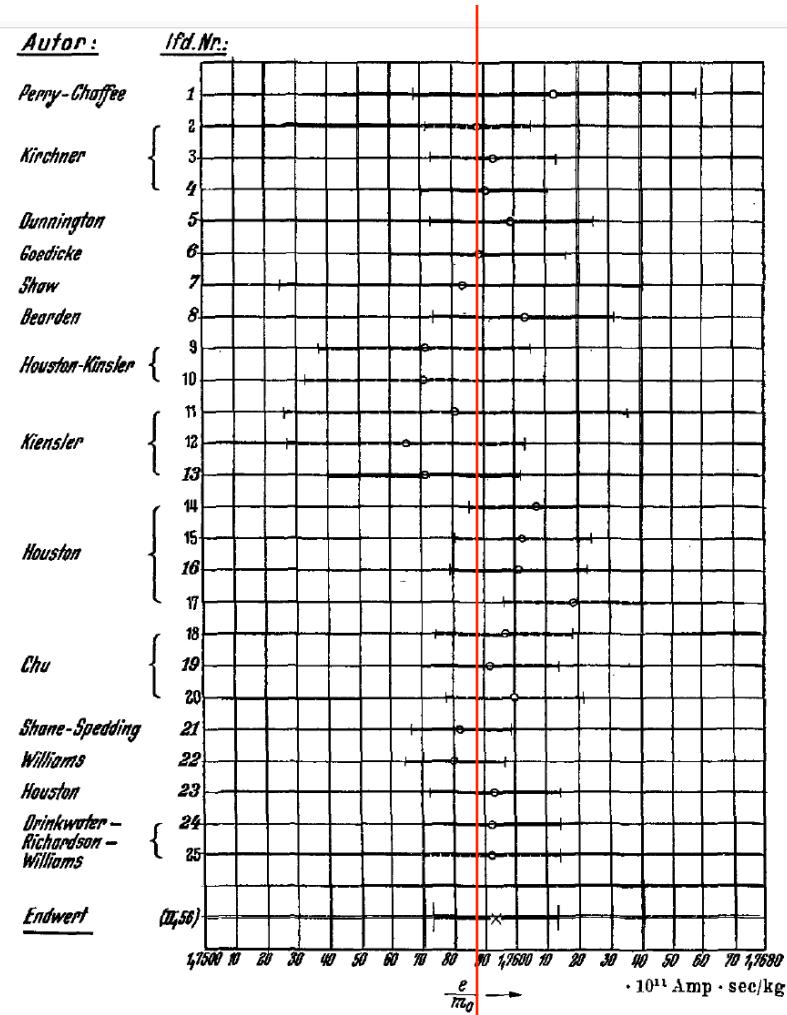
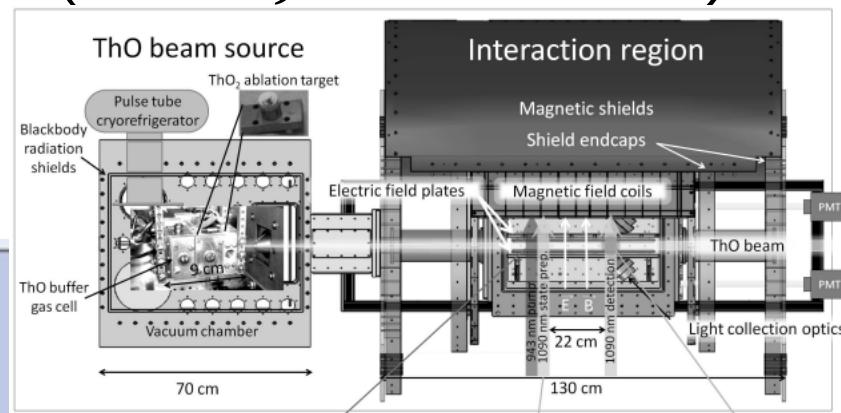
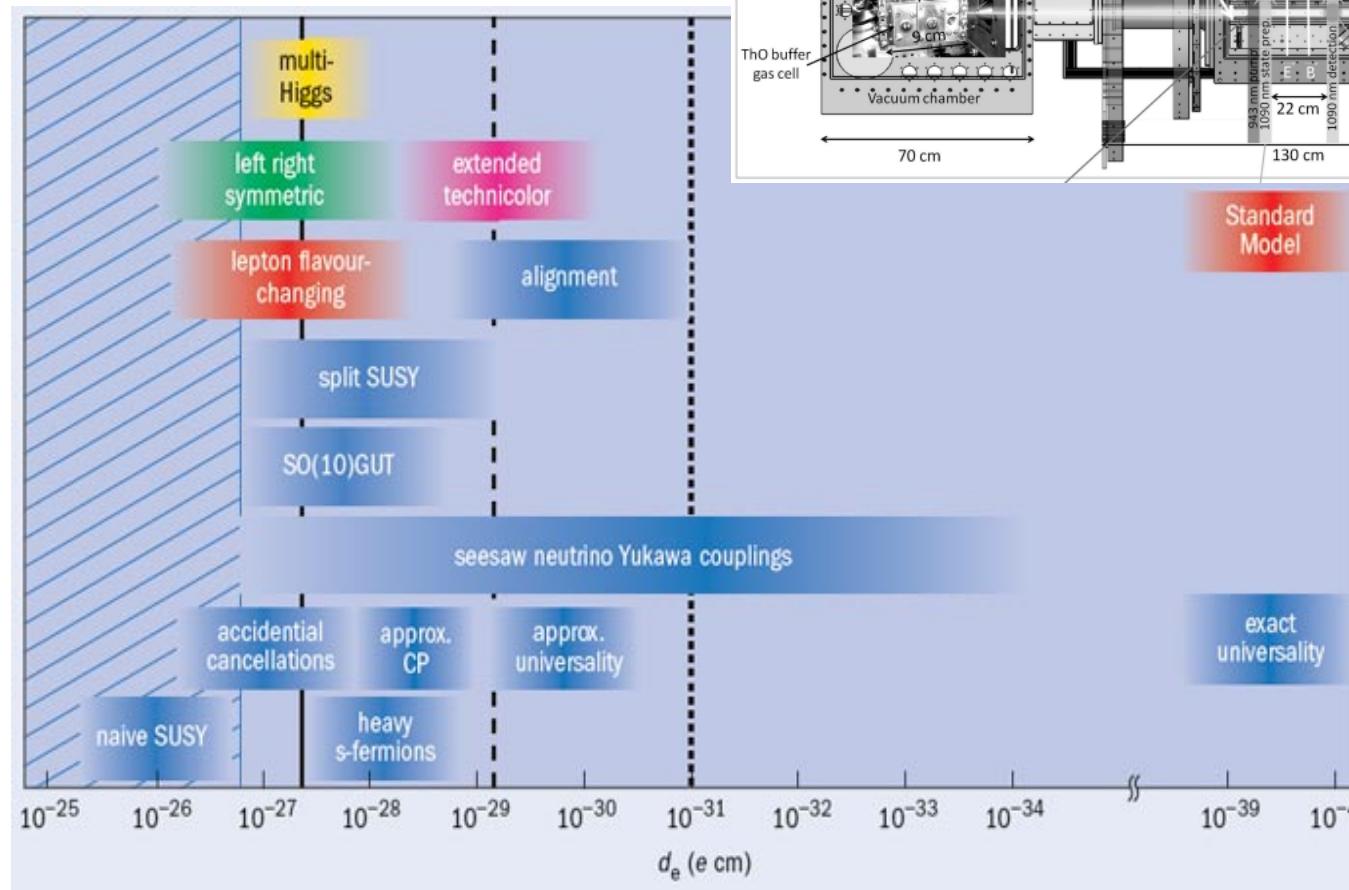


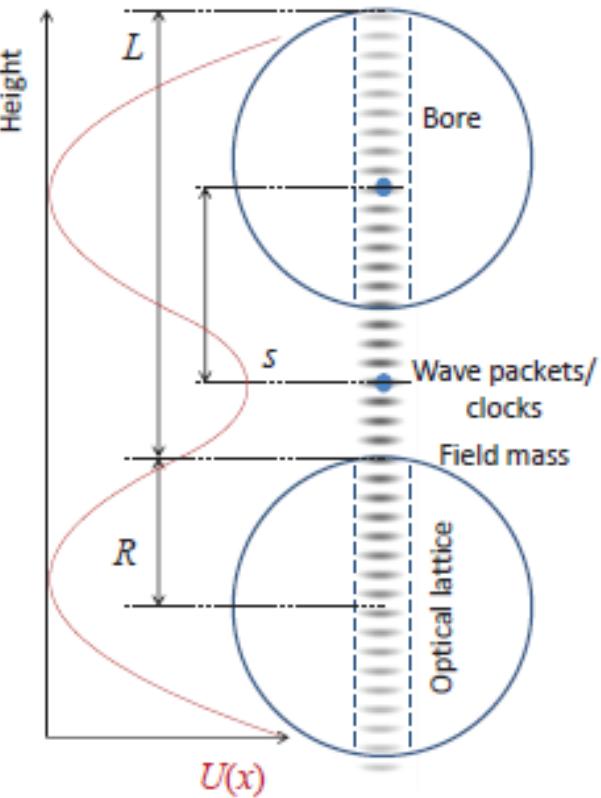
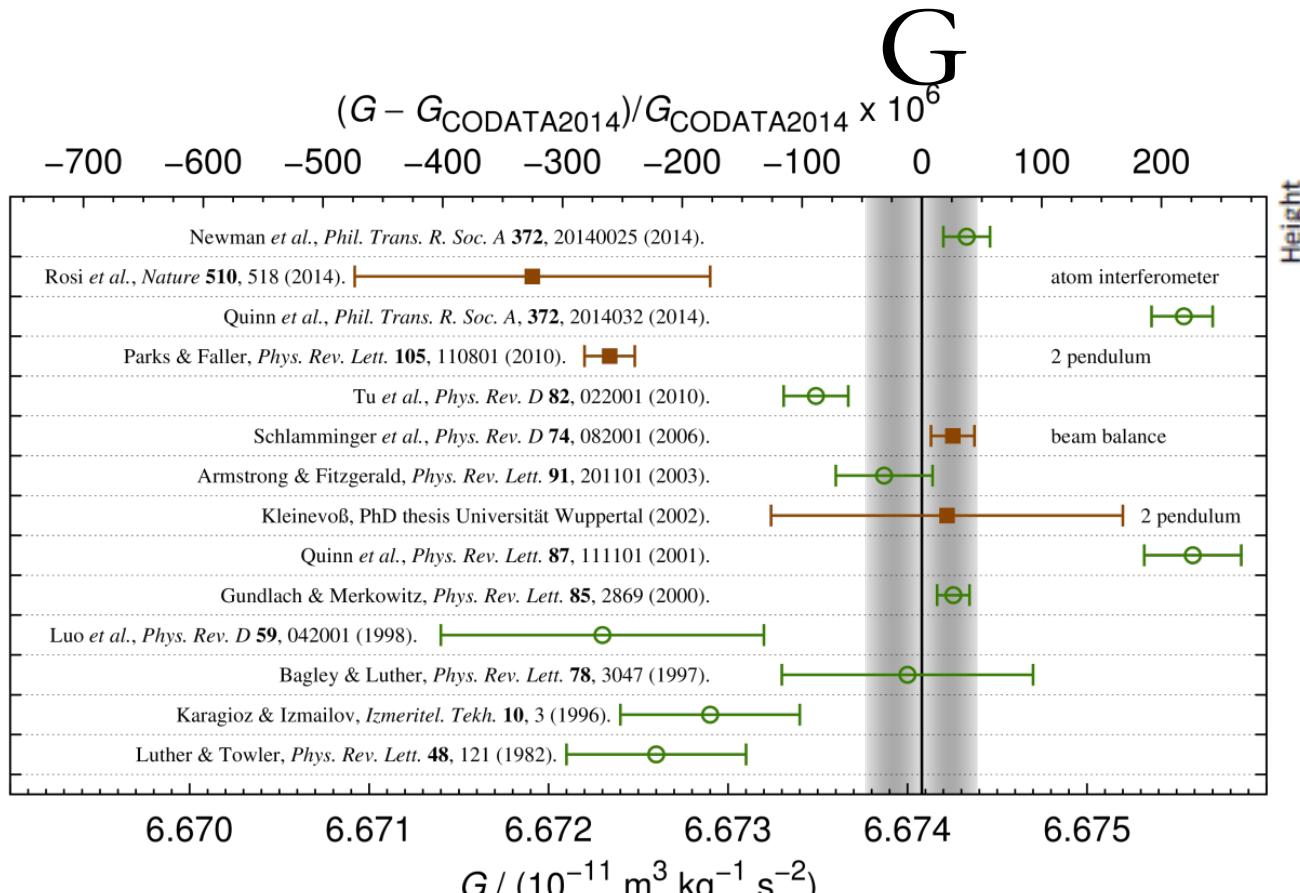
Fig. 1. Zusammenstellung der experimentell bestimmten e/m_0 -Werte.

Advanced Cold Molecule Edm experiment (Yale, Harvard)



Advanced cold molecule EDM

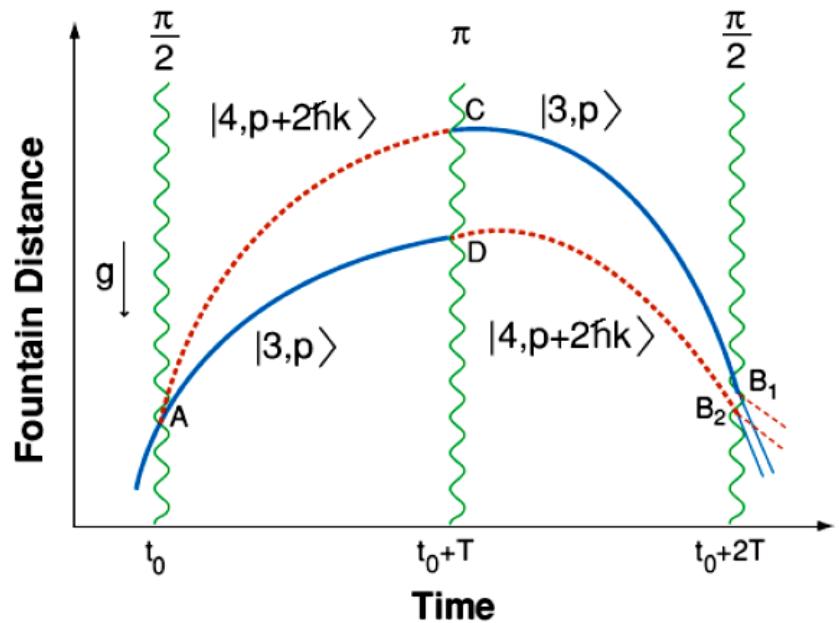
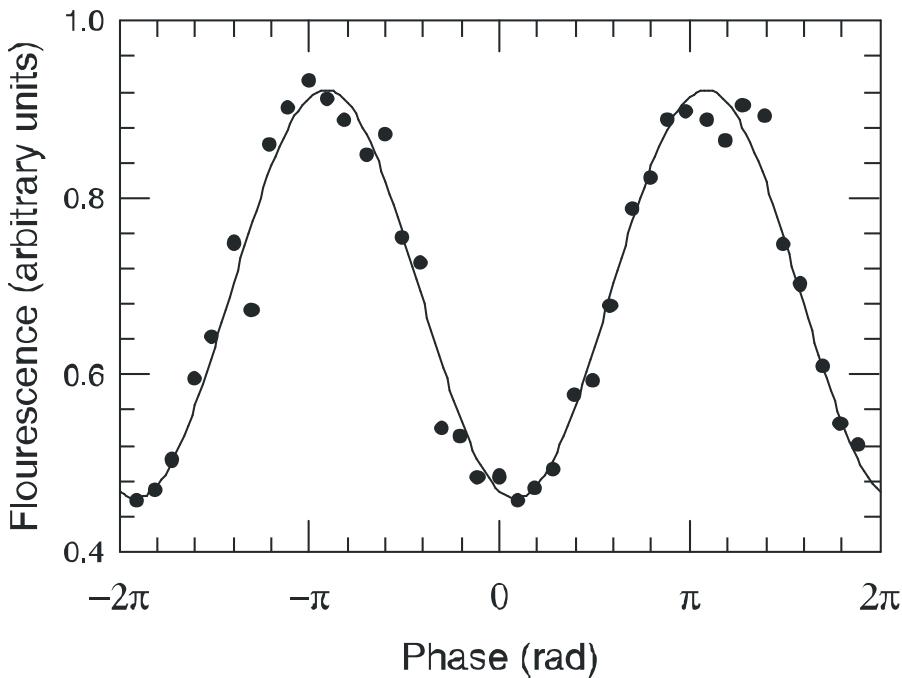
- Blind analysis has become the norm in EDM searches.
- ^{199}Hg EDM from Blayne Heckel & Norval Fortson's group. The Seattle group was the first EDM experiment to use blind analysis, to my knowledge.
- Hinds EDM blind
- Unknown offset added until after we had determined uncertainty and were resolved to publish the result



- Live unblinding

Light pulse atom interferometer

$$\begin{aligned}\Delta\varphi &= -\frac{1}{\hbar} \oint L dt + \Delta\varphi_{\text{laser}} \\ &= 2T^2 \vec{\Omega} \cdot [\vec{k} \times (\vec{v}_0 + \vec{a}T)] + \vec{k} \vec{a} T^2 \\ &\quad + O(1/c^4)\end{aligned}$$



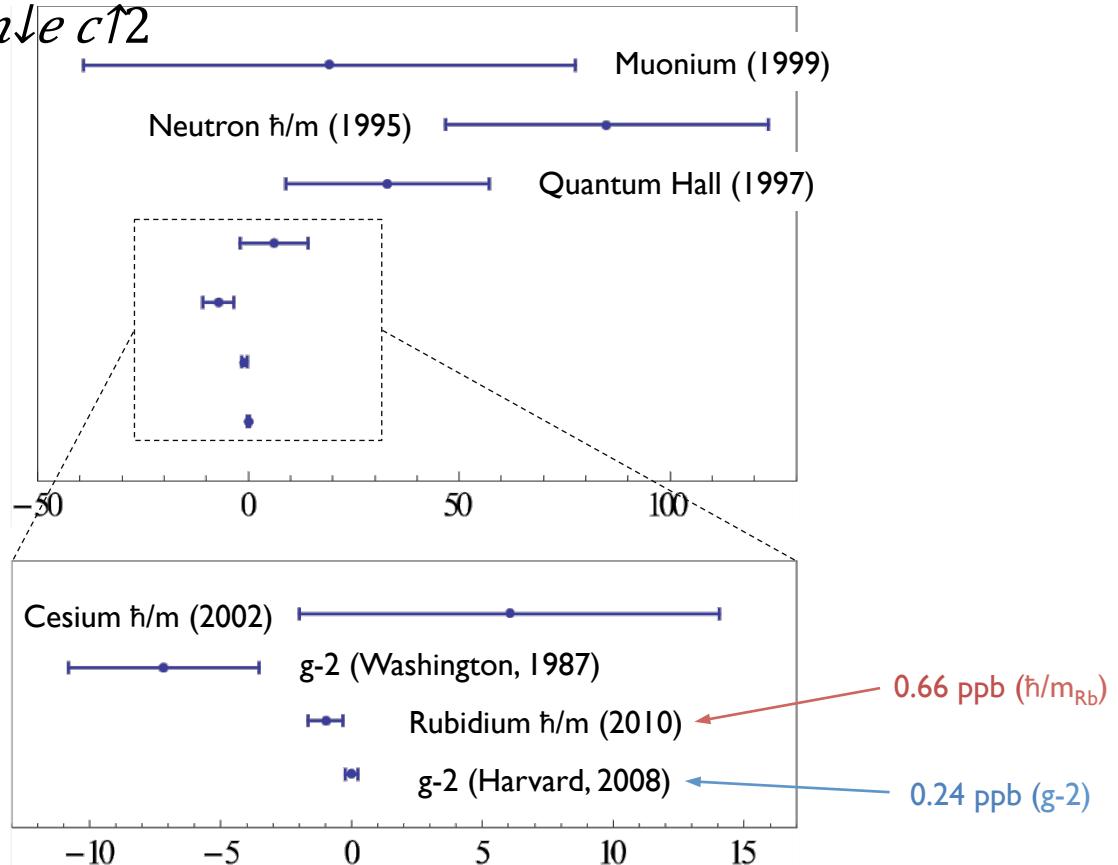
Each data point is from a single launch, determines g to 1.3ng
 $\Rightarrow 11\text{ng}/\sqrt{\text{Hz}}$

HM *et al.*, PRL 100, 031101
 (2008); PRD 80, 016002
 (2009)

Fine Structure Constant α

Select Alpha Measurements

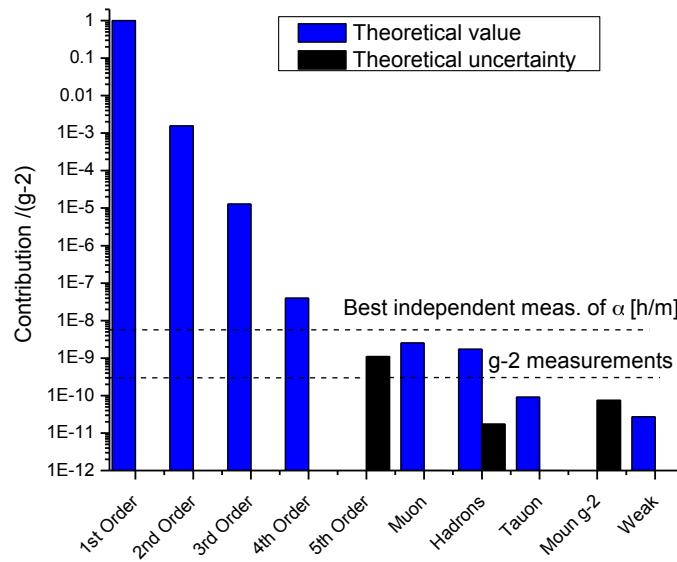
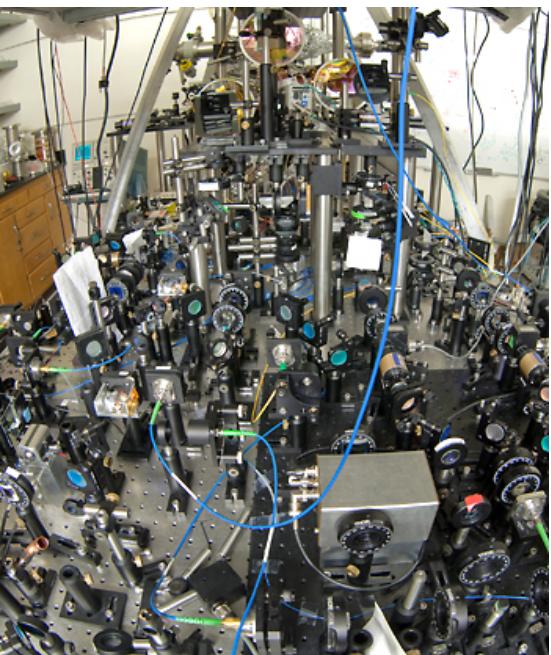
$$hcR\downarrow\infty = 1/2 \alpha^{1/2} m\downarrow e c^{1/2}$$



Relative Uncertainty $\delta\alpha/\alpha$ [ppb]

The most precise theory/experiment comparison in science

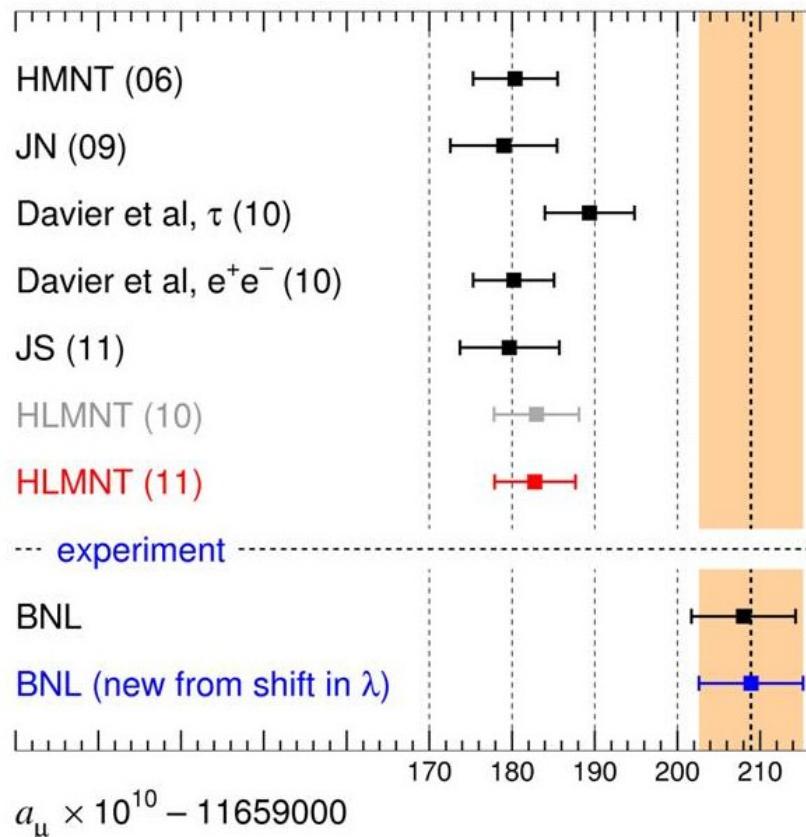
Fine structure constant



Electron gyromagnetic moment



$g-2$ and α



Muon $g-2$ campus at Fermilab

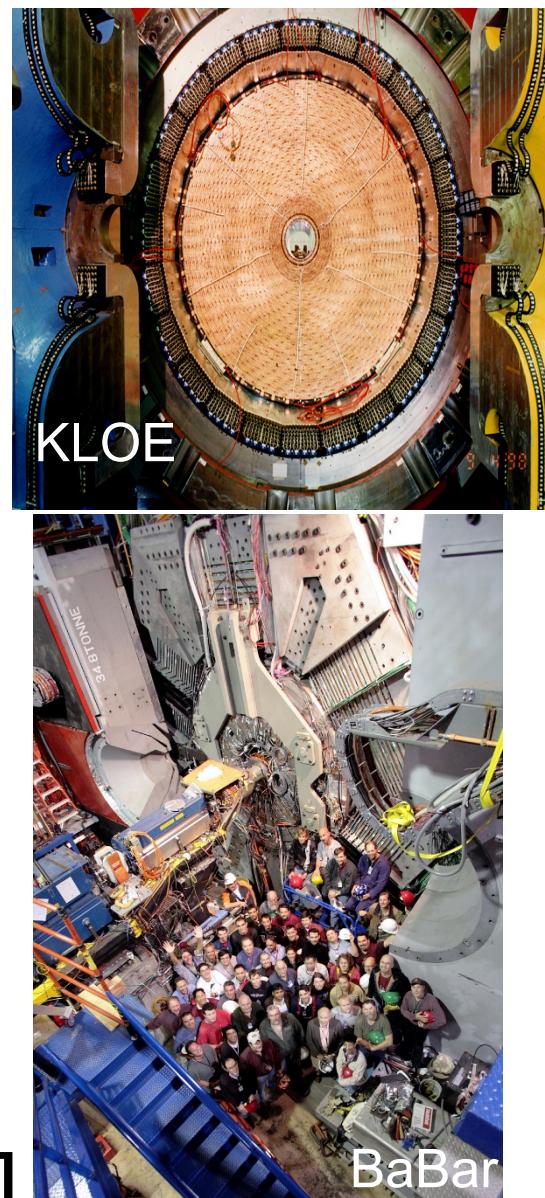
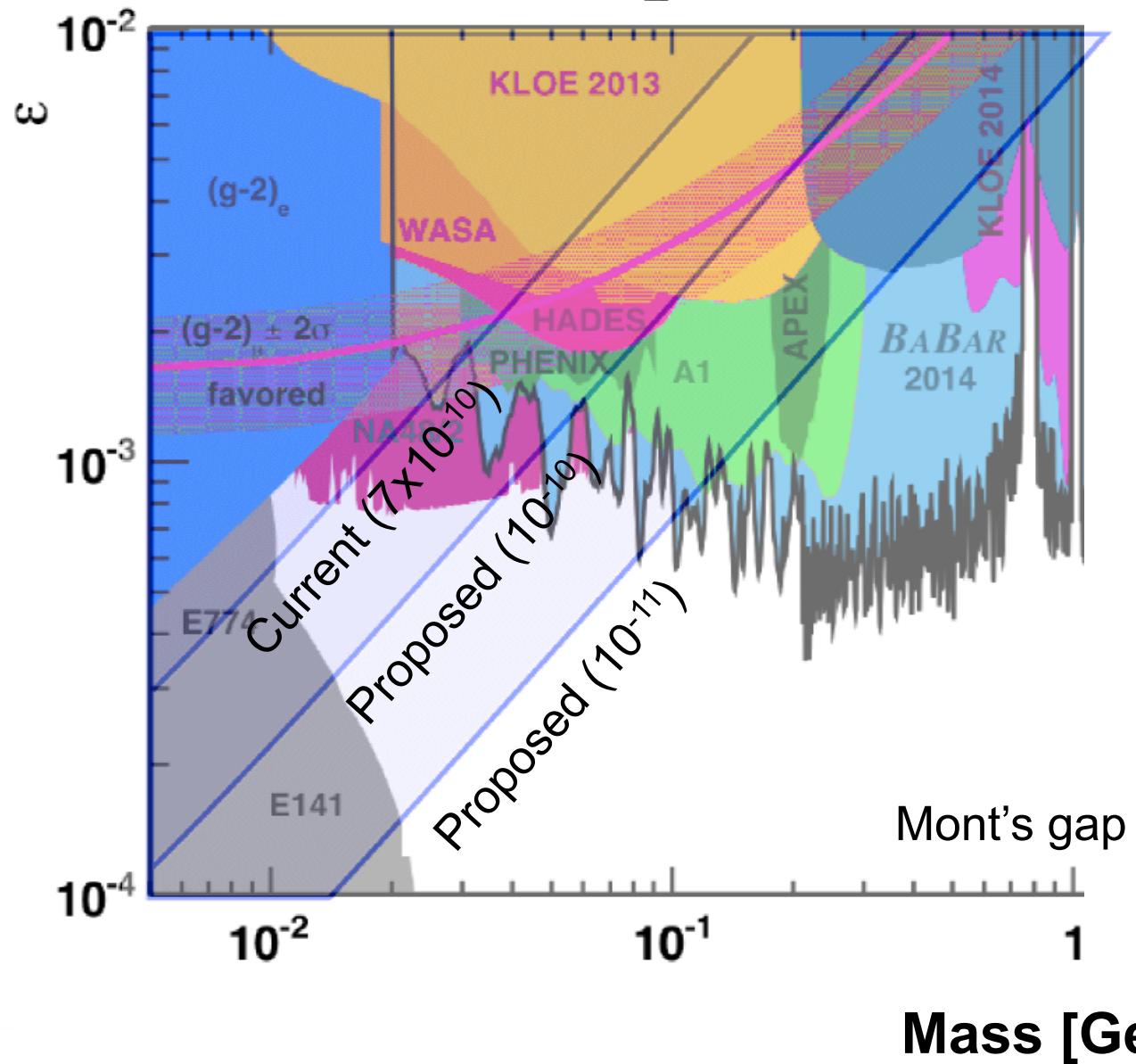
$$3 \times 10^{-6} \times \left(\frac{m_e}{m_\mu} \right)^2 = 0.07 \text{ ppb}$$

Can be tested in next-generation measurement of α and g_e-2

Dark matter candidates such as dark photons

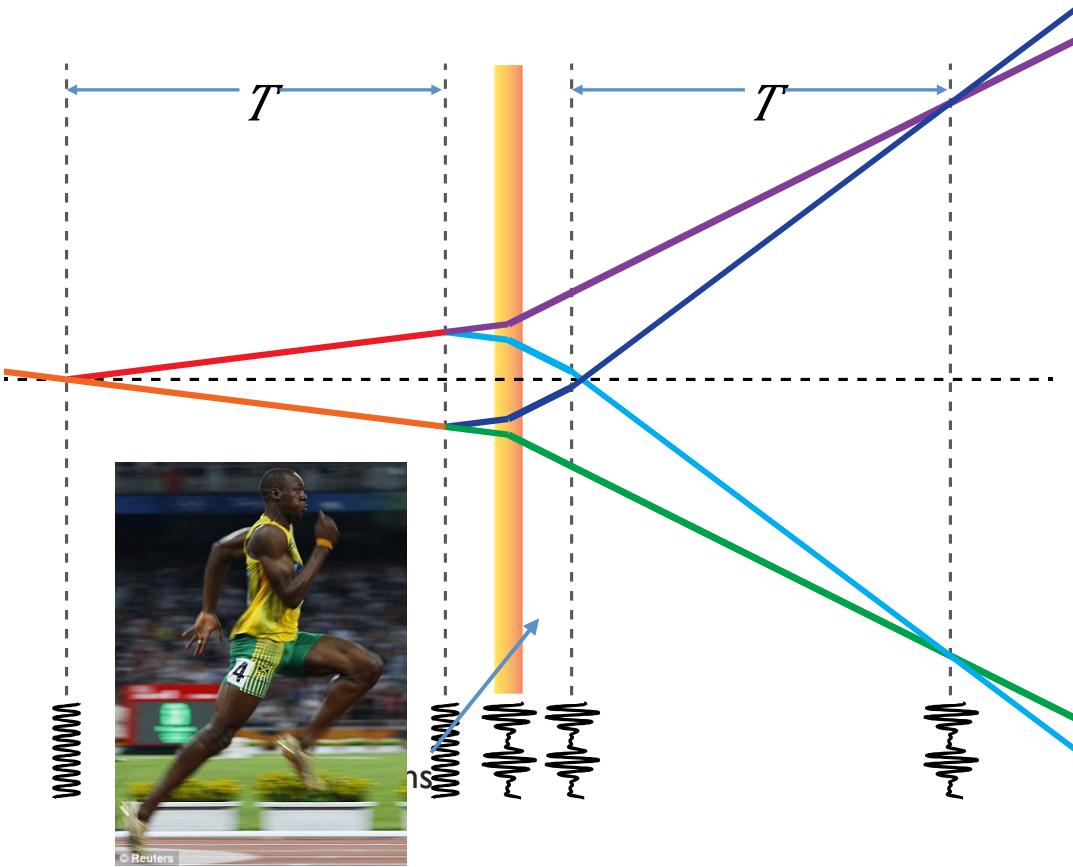
Sensitive up to energy scales $m \sim [m_e/(10^{-10})]^{1/2} \sim 100 \text{ GeV}$

Dark photon searches

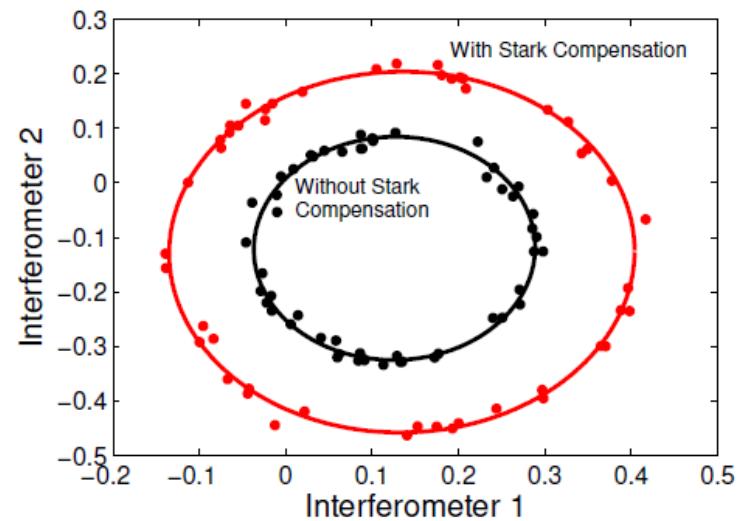


Atom-interferometer measurement of α

$$\Delta\phi = 16n(n+N)(\hbar k^2/2m)T$$



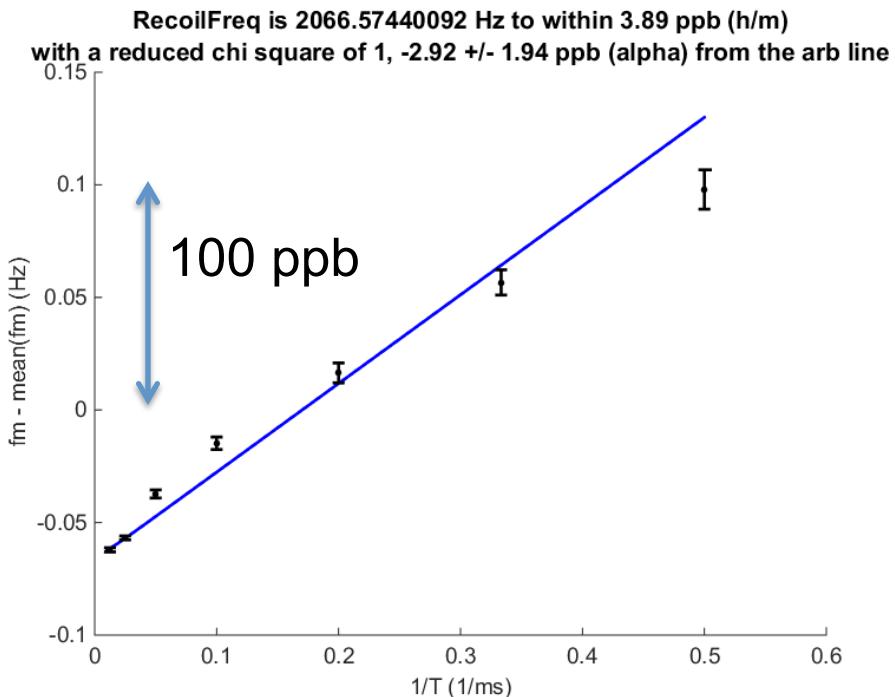
$$hcR/\infty = 1/2 \alpha^{1/2} m/e c^{1/2}$$



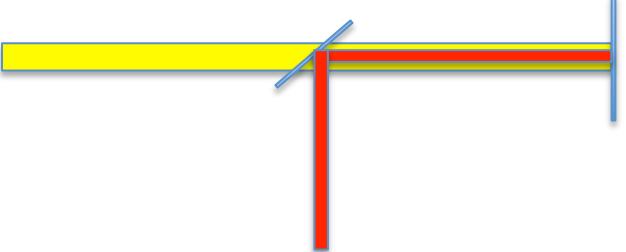
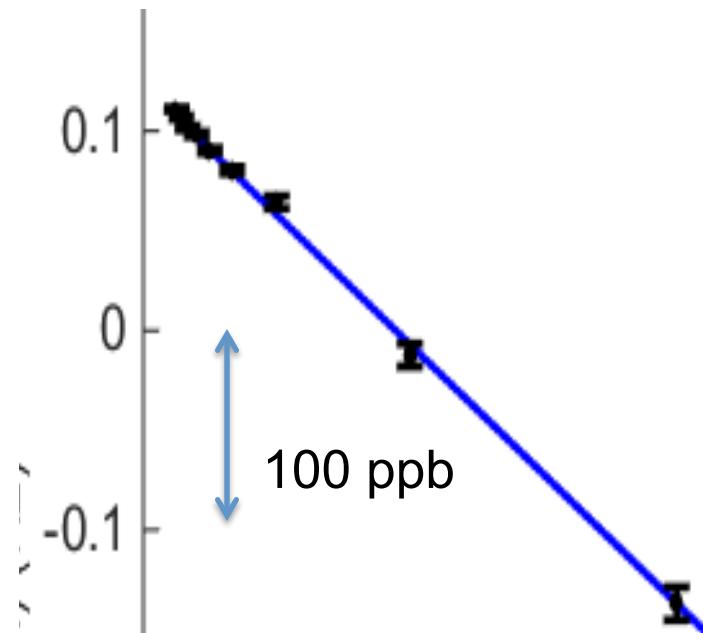
Now >800 photon kicks

Controlling systematics at 10^{-10}

Before filtering



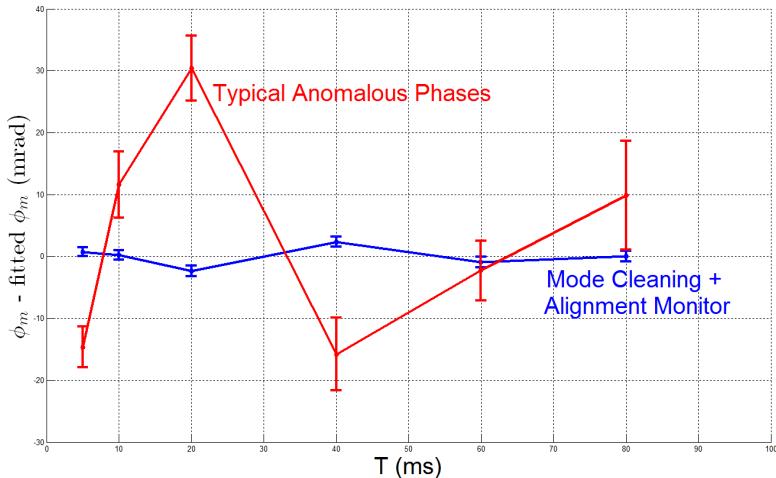
With spatial filtering



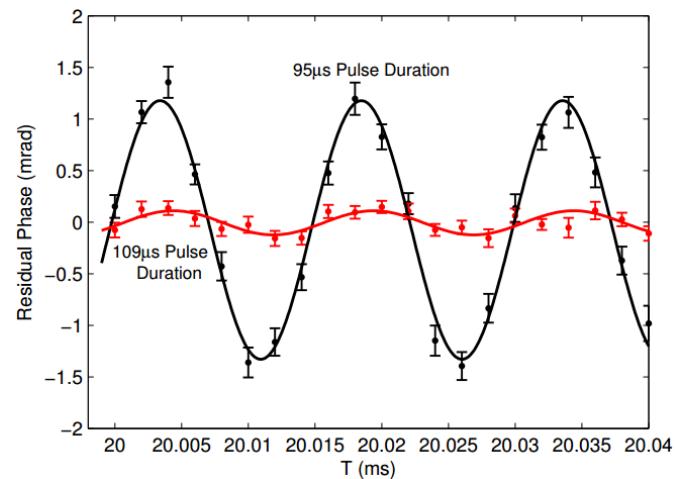
Spatial Filtering Beam

Estey et al, PRL (2015)

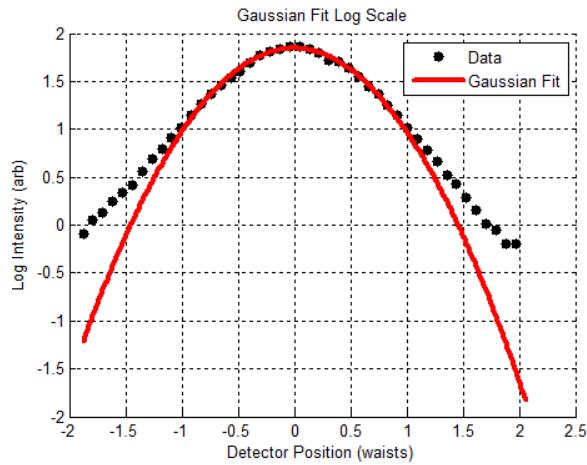
Control of systematic errors



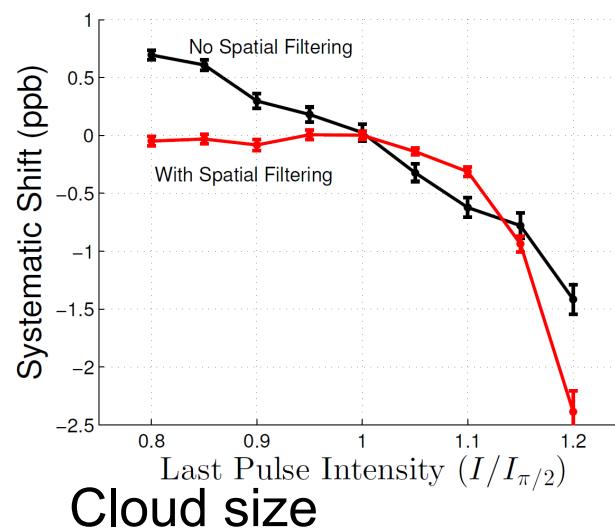
Shifts from beam distortions



Parasitic interferometers

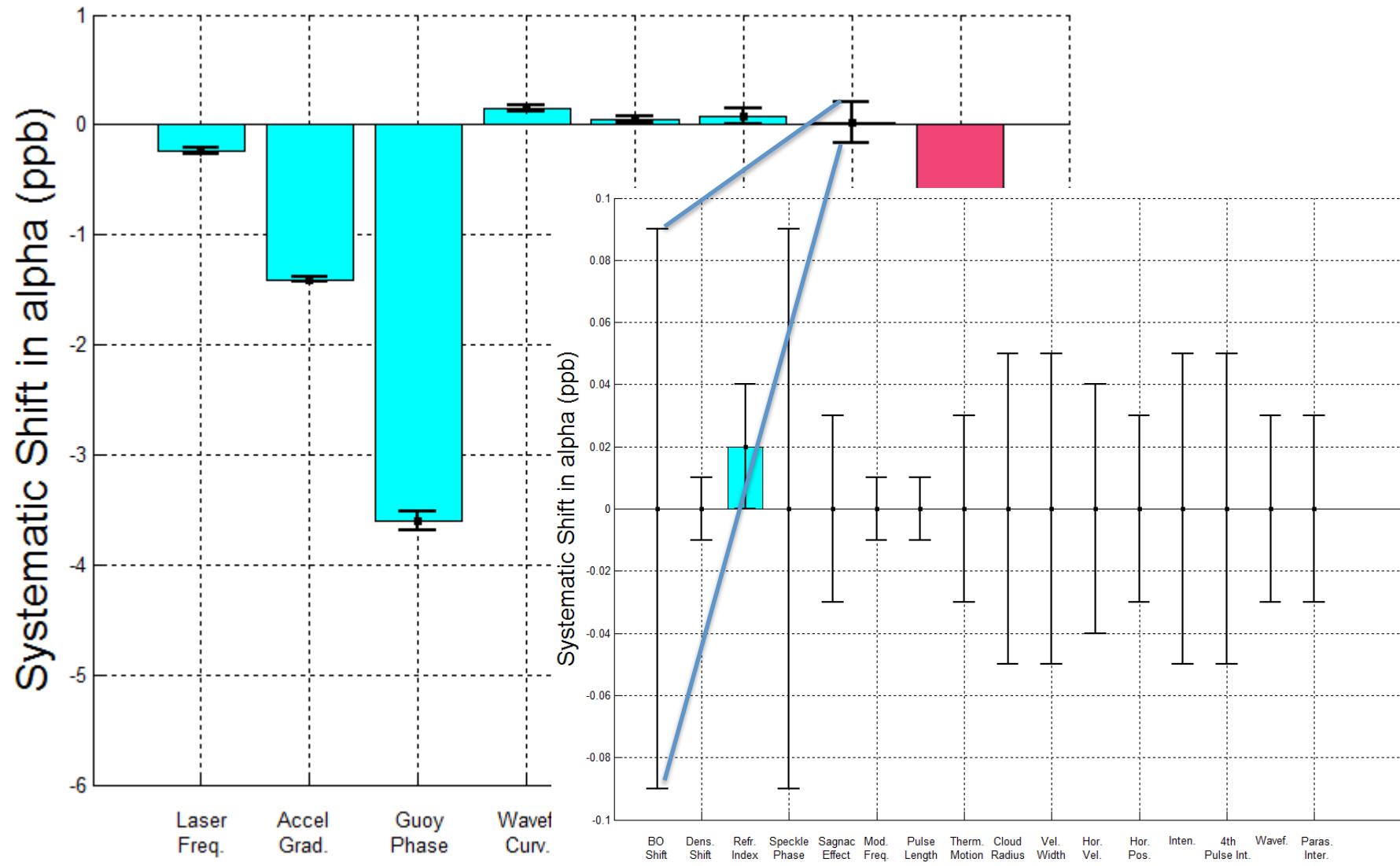


Beam distortions



Cloud size

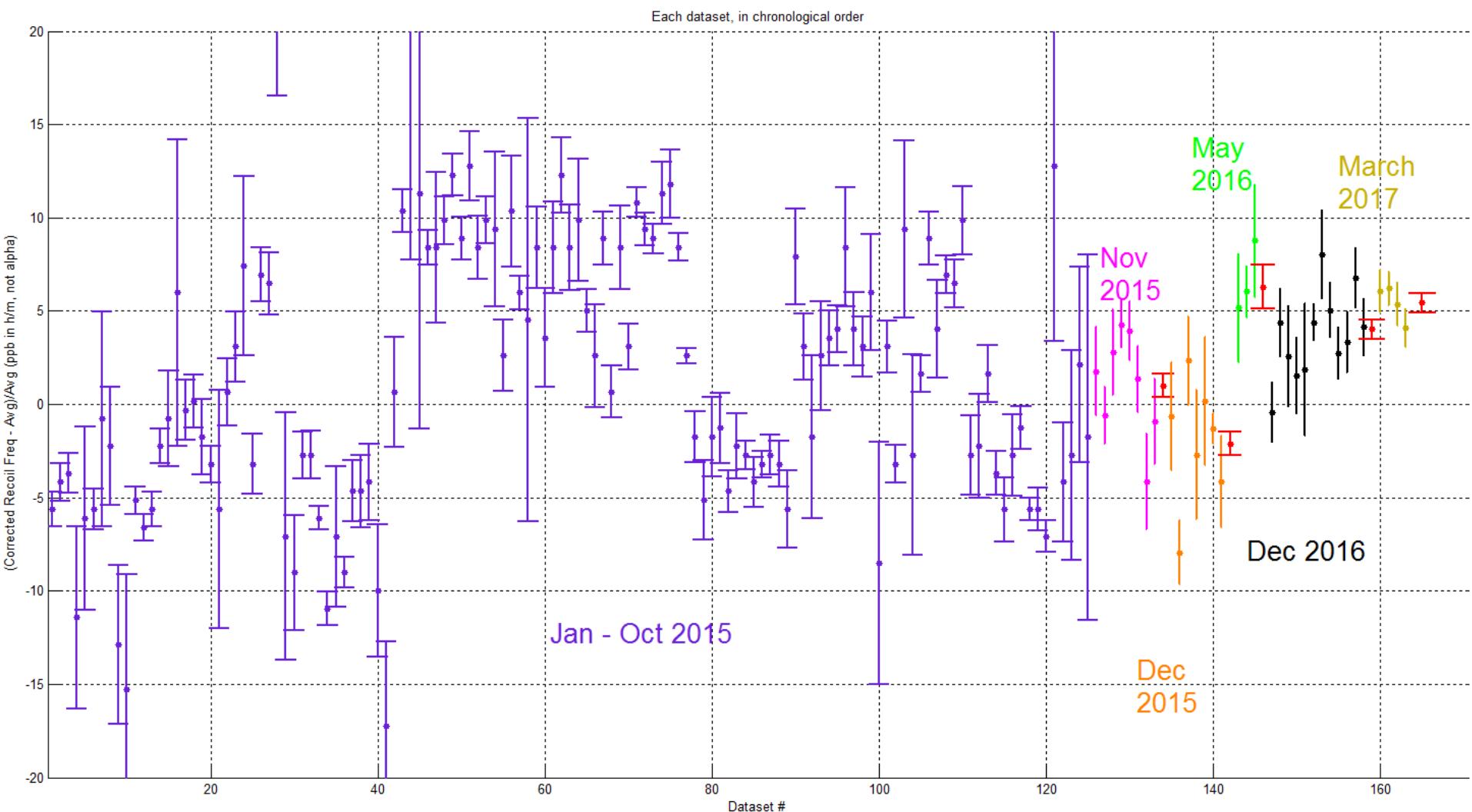
0.15 ppb error budget



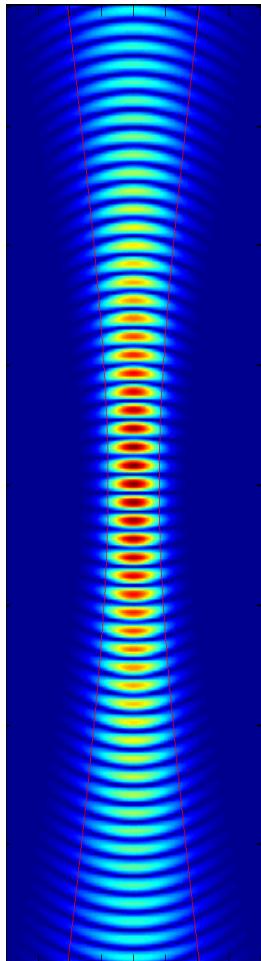
Blinding strategy

- Laser frequency: actual value hidden
- Default value used instead
- Three numerical codes will be created that should yield equal results

Overview of blinded data



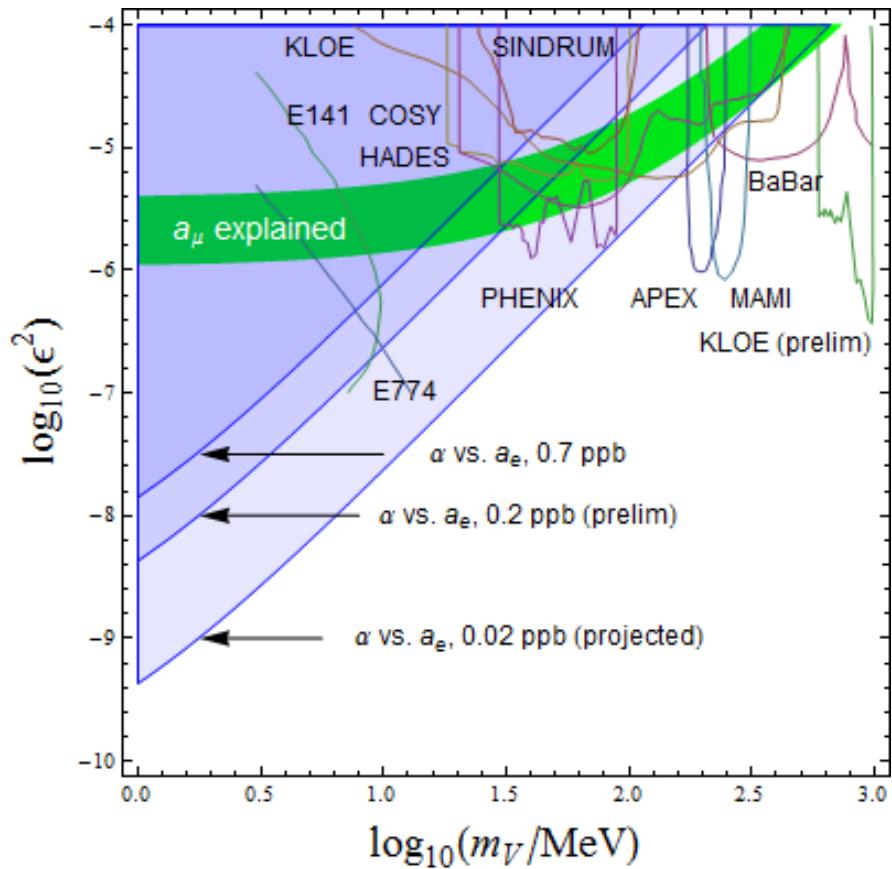
A more nearly perfect laser beam



- Infinite plane wave: $\lambda=c/v$, uniform intensity
- Wavelength errors $\sim(\lambda/\text{radius})^2$
- Beam splitter losses $\sim(\lambda/\text{radius})^4$
- 20x radius =>
 - 400-fold lower errors: **higher accuracy**
 - 160,000-fold lower losses: **higher momentum transfer, and thus sensitivity**

Summary

- Intellectual phase lock may be less widespread
- Other problems: politics, egos, vanity
- **Blind analysis forces you to take data seriously**
- Questions: how to unblind?
- Overarching goal should be to avoid errors
- Precision measurements are long-term projects.
- Culture of Care required



Thanks

- Patricia Burchardt
- Steve Chu
- Dave DeMille
- Gerald Gabrielse
- Richard Parker
- Steve Quake



A Clock Directly Linking Time to a Particle's Mass

Lan et al., Science 339, 554 (2013)



Key insight / innovation

- Feedback over atom interferometer using frequency comb
- Comb multiplies by N
- n-photon momentum transfer
- Clock frequency ~ 100 kHz = $(m^2/h)/(2nN^2)$, exactly

Technology impact

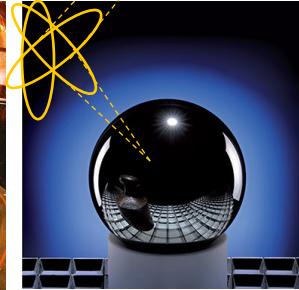
- A single particle is useful as a time/frequency reference
- Cesium atom used as approximation of a point mass
- 4 ppb accuracy demonstrated, < 1 ppb feasible

Application

- Mass standard in new SI where h and c are defined, kg measured
- With elementary (anti-) particles => equivalence principle with antimatter
- Light nanomechanical objects => mesoscopic mass standard

Watt balance

- 33 ppb for amu
- 33 ppb for kg
- Moving parts, gravity, standard resistors...
- Americans and Europeans disagree



Atom interferometer+counting

- 4 ppb for amu
- 30 ppb for kg
- No moving parts, gravity, standard resistors...
- Agrees with European versions
- <1 ppb in the future

Independent methods realize the same definition ☺

