

Advanced Electromagnetics: 21st Century Electromagnetics

### Lorentz Oscillator Model

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### Lecture Outline

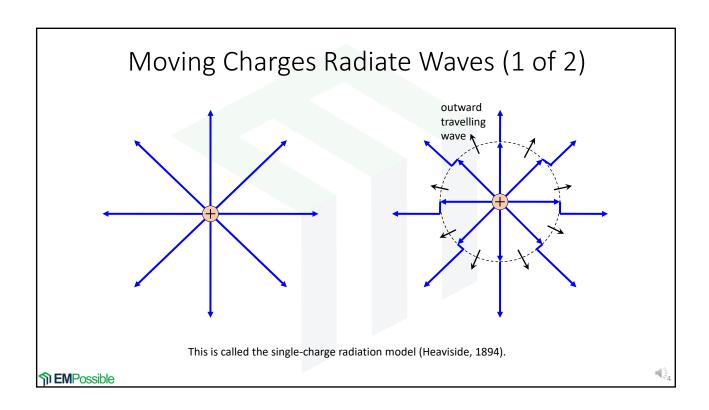
- High level picture of dielectric response
- Qualitative description of resonance
- Derivation of Lorentz oscillator model

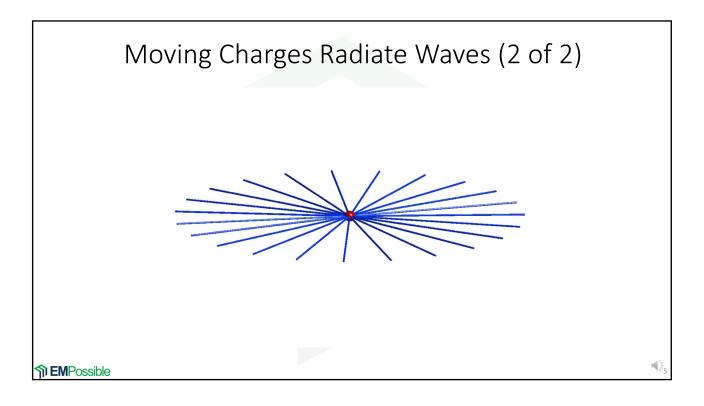
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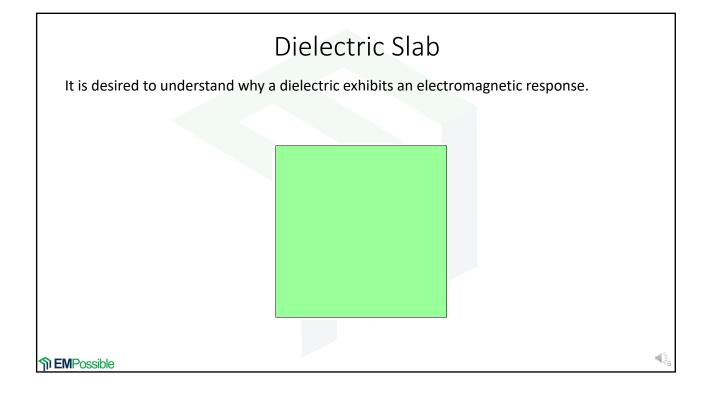
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### High Level Picture of Dielectric Response



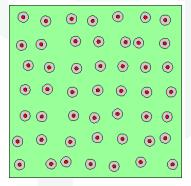






#### Atoms at Rest

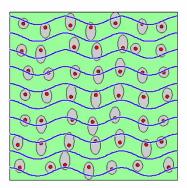
Without an applied electric field  $\vec{E}$ , the electron "clouds" around the nuclei are symmetric and at rest.



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### **Applied Wave**

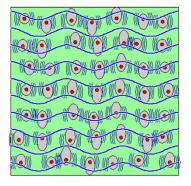
The electric field  $\vec{E}$  of a electromagnetic wave pushes the electrons away from the nuclei producing "clouds" that are offset.



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### Secondary Waves

The motion of the charges emits secondary waves that interfere with the applied wave to produce an overall slowing effect on the wave.



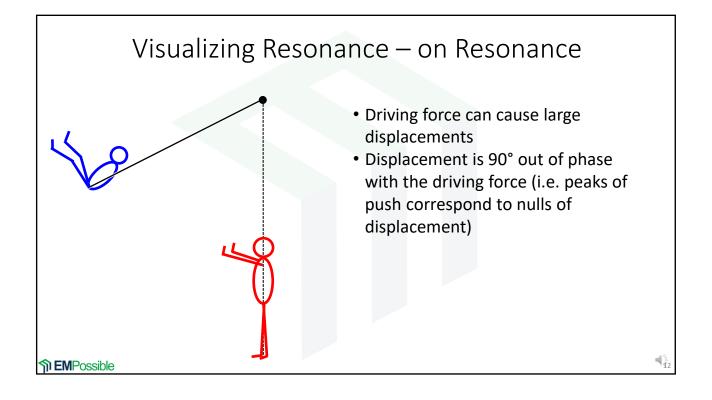
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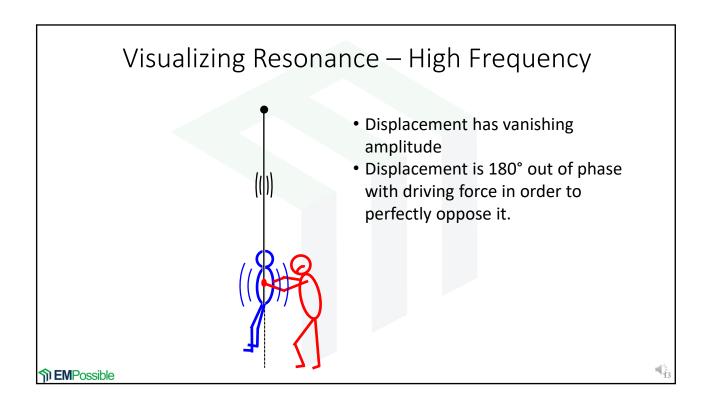
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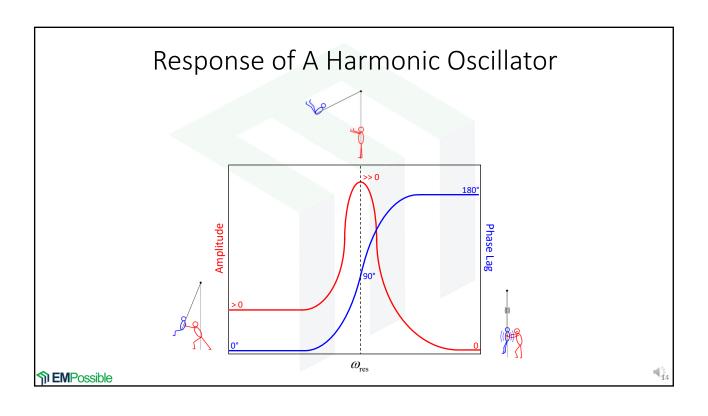
# Qualitative Description of Resonance

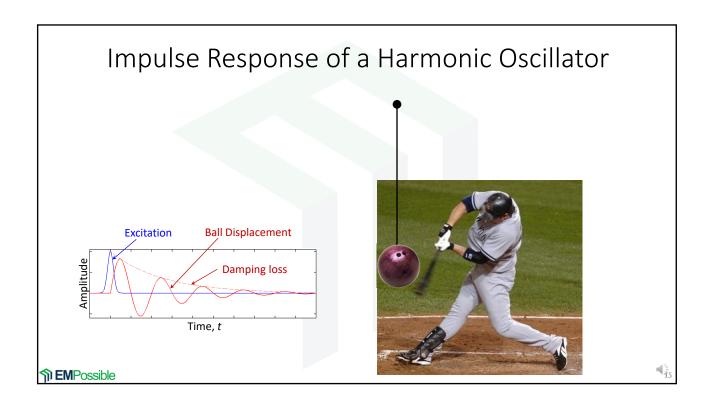
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# Visualizing Resonance — Low Frequency • Driving force is able to modulate amplitude • Displacement is in phase with driving force • There exists a DC offset



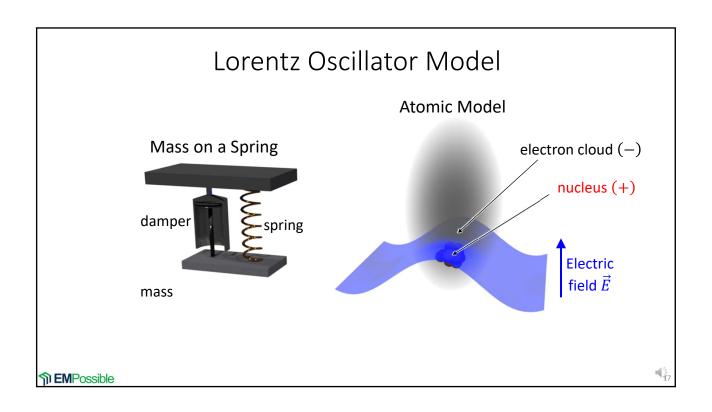


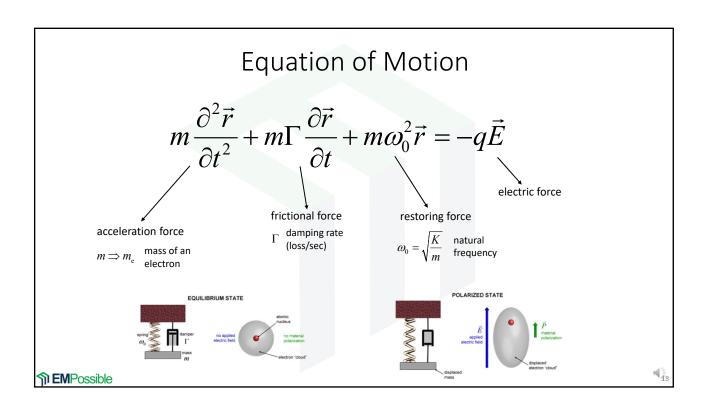




## Derivation of Lorentz Oscillator Model

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### Fourier Transform the Equation of Motion

$$m\frac{\partial^{2}\vec{r}}{\partial t^{2}} + m\Gamma\frac{\partial\vec{r}}{\partial t} + m\omega_{0}^{2}\vec{r} = -q\vec{E}$$

$$\downarrow^{\text{Fourier transform}}$$

$$m(-j\omega)^{2}\vec{r}(\omega) + m\Gamma(-j\omega)\vec{r}(\omega) + m\omega_{0}^{2}\vec{r}(\omega) = -q\vec{E}(\omega)$$

$$\downarrow^{\text{Simplify}}$$

$$\left(-m\omega^{2} - j\omega m\Gamma + m\omega_{0}^{2}\right)\vec{r}(\omega) = -q\vec{E}(\omega)$$

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### Charge Displacement $\vec{r}(\omega)$

$$(-m\omega^{2} - j\omega m\Gamma + m\omega_{0}^{2})\vec{r}(\omega) = -q\vec{E}(\omega)$$

$$\downarrow \text{Solve for } \vec{r}(\omega)$$

$$\vec{r}(\omega) = -\frac{q}{m_{e}} \frac{\vec{E}(\omega)}{\omega_{0}^{2} - \omega^{2} - j\omega\Gamma}$$



The displacement  $\vec{r}(\omega)$  describes how far charge is displaced from its equilibrium position.

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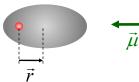
### Electric Dipole Moment $\vec{\mu}(\omega)$

Definition of Electric Dipole Moment: 
$$\vec{\mu}(\omega) = -q\vec{r}(\omega)$$

distance from center

\*\* Sorry for the confusing notation, but  $\mu$  here is NOT permeability.

$$\vec{\mu}(\omega) = \frac{q^2}{m_e} \frac{\vec{E}(\omega)}{\omega_0^2 - \omega^2 - j\omega\Gamma}$$



The electric dipole moment  $\vec{\mu}(\omega)$  is a measure of the strength and separation of positive and negative charges.

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### Lorentz Polarizability $\alpha(\omega)$

Definition of Lorentz Polarizability:  $\vec{\mu}(\omega) = [\alpha(\omega)]\vec{E}(\omega)$ 

\*\* Sorry for the confusing notation, but  $\alpha$  here is NOT absorption.

 $[\alpha(\omega)]$  is a tensor quantity for anisotropic materials. For simplicity, the scalar form will be adopted here. This is the Lorentz polarizability for a single atom.

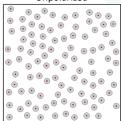
$$\alpha(\omega) = \frac{q^2}{m_e} \frac{1}{\omega_0^2 - \omega^2 - j\omega\Gamma}$$

The Lorentz polarizability  $[\alpha(\omega)]$  is a measure of how easily electrical charges are displaced. Charge may be more easily displaced in some directions that others.

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### Polarization Per Unit Volume $\vec{P}(\omega)$

Average dipole moment over all atoms in a material. Definition:  $\vec{P}(\omega) = \frac{1}{V} \sum_{v} \vec{\mu}_i(\omega)$ 



Equivalent uniform polarization

There is some randomness to the polarized atoms so a statistical approach is taken to compute the average.

$$\vec{P}(\omega) = N \langle \vec{\mu}(\omega) \rangle = \frac{Nq^2}{m_e} \frac{\vec{E}(\omega)}{\omega_0^2 - \omega^2 - j\omega\Gamma}$$

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 $N \equiv$  Number of atoms per unit volume

 $\langle \rangle \equiv$  Statistical volume average

### Electric Susceptibility $\chi_{\rm e}(\omega)$ (1 of 2)

A material becomes polarized  $\vec{P}$  in the presence of an electric field  $\vec{E}$  according to

$$\vec{P}(\omega) = \varepsilon_0 \chi_e(\omega) \vec{E}(\omega)$$

 $\chi_{\rm e}(\omega)$  is called the *electric susceptibility* and is a measure of how easily an electric field  $\vec{E}$  can polarize a material.

This leads to an expression for the electric susceptibility:

$$\chi_{\rm e}(\omega) = \frac{N\alpha(\omega)}{\varepsilon_0} = \left(\frac{Nq^2}{\varepsilon_0 m_{\rm e}}\right) \frac{1}{\omega_0^2 - \omega^2 - j\omega\Gamma}$$

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### Electric Susceptibility $\chi_{\rm e}(\omega)$ (2 of 2)

The electric susceptibility of a dielectric material is:

$$\chi_{\rm e}(\omega) = \frac{\omega_{\rm p}^2}{\omega_0^2 - \omega^2 - j\omega\Gamma} \qquad \omega_{\rm p}^2 = \frac{Nq^2}{\varepsilon_0 m_{\rm e}} _{\rm frequency}^{\rm plasma}$$

$$q = 1.60217646 \times 10^{-19} \text{ C}$$
  
 $\varepsilon_0 = 8.8541878176 \times 10^{-12} \text{ F/m}$   
 $m_c = 9.10938188 \times 10^{-31} \text{ kg}$ 

- Note this is the susceptibility of a dielectric which has only one resonance.
- The location of atoms is important because they can influence each other. This was ignored.
- Real materials have many sources of resonance and all of these must be added together.
- Electric susceptibility is the *transfer function* of the oscillator system.

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