Physics 1A Discussion (Week 2)

| Name | | UID | |
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Problem 1 (problem 1.53 on page 30 of the textbook)

BIO Estimate the number of atoms in your body. (Hint: Based on what you know about biology and chemistry, what are the most common types of atom in your body? What is the mass of each type of atom?



Group Period

| 1 | 1 H 1.008 | | | | | | | | | | | | | | | | | 2 He 4.003 |
|---|---------------------------|--------------------------------------|---------------------------|---------------------------|-------------------|---------------------------|--------------------|---------------------------|---------------------------|---------------------------|--------------------|---------------------------|---------------------------|-------------------------------|---------------------------|--------------------------|---------------------------|---------------------------|
| 2 | 3 Li 6.941 | 4 Be 9.012 | | | | | | | | | | | 5 B | 6 C 12.011 | 7 N 14.007 | 8 O 15.999 | 9 F 18.998 | 10 Ne 20.180 |
| 3 | 11 Na 22.990 | 12 Mg _{24.305} | | | | | | | | | | | 13 Al 26.982 | 14 Si _{28.086} | 15 P 30.974 | 16 S 32.065 | 17 Cl 35.453 | 18 Ar 39.948 |
| 4 | 19 K 39.098 | 20 Ca 40.078 | 21 Sc 44.956 | 22 Ti 47.867 | 23 V 50.942 | 24 Cr 51.996 | 25 Mn 54.938 | 26 Fe 55.845 | 27 Co 58.933 | 28 Ni 58.693 | 29 Cu 63.546 | 30 Zn 65.409 | 31 Ga 69.723 | 32 Ge 72.64 | 33 As 74.922 | 34 Se 78.96 | 35 Br 79.904 | 36 Kr 83.798 |

For each element, the average atomic mass of the mixture of isotopes occurring in nature is shown. For elements having no stable isotope, the approximate atomic mass of the longest-lived isotope is shown in parentheses. All atomic masses are expressed in atomic mass units $(1~\mathrm{u}=1.660539040~(20)\times10^{-27}~\mathrm{kg})$, equivalent to grams per mole $(\mathrm{g/mol})$.

| ${f Problem} {f 2}$ | (problem | 1.72 on | page 30 | of the | textbook) |) |
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| Ricardo and Jane are standing under a tree in the middle of a pasture. An argument ensues, and they walk away in different directions. Ricardo walks 26.0 m in direction 60.0° west of north. Jane walks 16.0 m in a direction 30° south of west. They then stop and turn to face each other. (a) What is the distance between them? (b) In what direction should Ricardo walk to go directly toward Jane? |
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| Problem 3 (problem 1.82 on page 31 of the textbook) Vector \overrightarrow{A} has magnitude 5.00 m and lies in the xy -plane in a direction 53.0° from the $+x$ -axis measured toward the $+y$ -axis. Vector \overrightarrow{B} has magnitude 8.00 m and a direction you can adjust. (a) You want the vector product $\overrightarrow{A} \times \overrightarrow{B}$ to have a positive z -component of the largest possible magnitude what direction should you select for vector \overrightarrow{B} ? (b) What is the direction of \overrightarrow{B} for which $\overrightarrow{A} \times \overrightarrow{B}$ has the most negative z -component? (c) What are the two directions of \overrightarrow{B} for which $\overrightarrow{A} \times \overrightarrow{B} = \overrightarrow{0}$? |
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