

A particle moving freely in an external potential $U(x)$ dissipates energy and settles to equilibrium at minimum value of U . ($\frac{d^2U}{dx^2} > 0$)

Minimize enthalpy: $H = U - Jx$ $\frac{d^2H}{dx^2} = \frac{d^2U}{dx^2} > 0 \Rightarrow$ new equilibrium point $x_{eq}(J) \rightarrow$ convex of $U(x)$ is accessible

More than one mechanical coordinates \rightarrow any δx_i results in an increase of energy

\Rightarrow more symmetric
|||

Equivalent eqn.

$$\sum_{i,j} \frac{\partial^2 U}{\partial x_i \partial x_j} \delta x_i \delta x_j > 0 \quad \text{or} \quad \sum_{i,j} \frac{\partial^2 H}{\partial x_i \partial x_j} \delta x_i \delta x_j > 0$$

$$\delta J_i = \delta \left(\frac{\partial U}{\partial x_i} \right) = \sum_j \frac{\partial^2 U}{\partial x_i \partial x_j} \delta x_j \quad (\delta: \text{change in non-equilibrium})$$

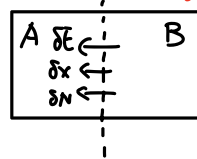
$$\sum_i \delta J_i \delta x_i > 0$$

\hookrightarrow generalized in chemistry

$$\delta T \delta S + \sum_i \delta J_i \delta x_i + \sum_\alpha \delta \mu_\alpha \delta N_\alpha > 0 \quad (\text{thermal \& chemical inputs})$$

uniformity of an extended thermodynamic body

A and B are equal (stable to unstable)



$E(T), x(J), \mu(N) \parallel$ at fixed E, x, N

\rightarrow first-order relation:

$$\delta T_A = -\delta T_B = \delta T, \quad \delta J_A = -\delta J_B = \delta J; \quad \delta \mu_A = -\delta \mu_B = \delta \mu$$

$$S = S_A + S_B = \left(\frac{E_A}{T_A} - \frac{J_A}{T_A} x_A - \frac{\mu_A}{T_A} N_A \right) + \left(\frac{E_B}{T_B} - \frac{J_B}{T_B} x_B - \frac{\mu_B}{T_B} N_B \right)$$

first order change $\frac{1}{T_A} \delta E_A$ & $\frac{1}{T_B} \delta E_B$ $E_A \delta(\frac{1}{T_A})$ & $E_B \delta(\frac{1}{T_B}) \dots$
are all eliminated

second order are the same

$$\begin{aligned} \delta S &= \delta S_A + \delta S_B = 2 \left[\delta E_A \delta \left(\frac{1}{T_A} \right) - \delta \left(\frac{J_A}{T_A} \right) \delta x_A - \delta \left(\frac{\mu_A}{T_A} \right) \delta N_A \right] \\ &= 2 \left[- \frac{\delta E_A \delta T_A}{T_A^2} - \frac{T_A \delta J_A - J_A}{T_A^2} \delta x_A - \frac{T_A \delta \mu_A - \mu_A}{T_A^2} \delta N_A \right] \\ &= - \frac{2}{T_A} \left[\delta T_A \left(\frac{\delta E_A - J_A \delta x_A - \mu_A \delta N_A}{T_A} \right) + \delta J_A \delta x_A + \delta \mu_A \delta N_A \right] \\ &= - \frac{2}{T_A} \left[\delta T_A \delta S_A + \delta J_A \delta x_A + \delta \mu_A \delta N_A \right] \end{aligned}$$

stable equilibrium: any change leads to a decrease of entropy

$$\delta T \delta S + \delta J \delta x + \delta \mu \delta N \geq 0$$

Example: δT and δx with $\delta N = 0$
 same result for any constraint

$$\begin{cases} \delta S = \left. \frac{\partial S}{\partial T} \right|_x \delta T + \left. \frac{\partial S}{\partial x_i} \right|_T \delta x_i \\ \delta J_i = \left. \frac{\partial J_i}{\partial T} \right|_x \delta T + \left. \frac{\partial J_i}{\partial x_j} \right|_T \delta x_j \end{cases} \quad \& \quad \delta T \delta S + \delta J \delta x + \delta \mu \delta N \geq 0$$

$$\left. \frac{\partial S}{\partial T} \right|_x (\delta T)^2 + \left. \frac{\partial J_i}{\partial x_j} \right|_T \delta x_j \delta x_i + \cancel{\left. \frac{\partial S}{\partial x_i} \right|_T \delta x_i \delta T} + \left. \frac{\partial J_i}{\partial T} \right|_x \delta T \delta x_i \geq 0$$

$$\text{as } dF = d(E - TS) = -SdT + Jdx$$

$$\text{Maxwell Relation: } -\left(\frac{\partial S}{\partial x} \right)_T = \left(\frac{\partial J}{\partial T} \right)_x$$

$\Rightarrow \left. \frac{\partial S}{\partial T} \right|_x > 0$
 \Rightarrow quadratic form, so it is positive for any δT or δx_i

Example: δT with $\delta x, \delta N = 0$

$$C_x = -\frac{dT}{dT} \Big|_x = T \left(\frac{\partial S}{\partial T} \right)_x \geq 0$$

δx_i with $\delta T, \delta x_{i \neq j}, \delta N = 0$

Matrix of coefficients $\left. \frac{\partial J_i}{\partial x_j} \right|_T$ must be positive definite

positive definite matrix: all eigenvalues are positive
 so trace and determinant are positive as well.

pressure (mechanical)
 & Chemical work for gas

$$\begin{bmatrix} -\left. \frac{\partial P}{\partial V} \right|_{T,N} & -\left. \frac{\partial P}{\partial N} \right|_{T,V} \\ \left. \frac{\partial \mu}{\partial V} \right|_{T,N} & \left. \frac{\partial \mu}{\partial N} \right|_{T,V} \end{bmatrix} \quad \begin{array}{l} \text{diagonals are all inverse response funcs.} \\ K_{T,N} = -\frac{1}{V} \left. \frac{\partial V}{\partial P} \right|_{T,N} \quad \& \quad \left. \frac{\partial N}{\partial \mu} \right|_{T,V} \\ \text{diagonals are positive.} \end{array}$$

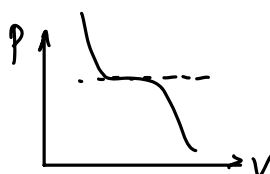
$$-\left. \frac{\partial P}{\partial V} \right|_{T,N} \left. \frac{\partial \mu}{\partial N} \right|_{T,V} + \left. \frac{\partial P}{\partial N} \right|_{T,V} \left. \frac{\partial \mu}{\partial V} \right|_{T,N} \geq 0$$

critical point of gas

$$\left. \frac{\partial P}{\partial V} \right|_{T,N} = 0 \quad (\text{Not justified but works for Taylor expansion})$$

$$\delta P(T=T_c) = \cancel{\left. \frac{\partial P}{\partial V} \right|_{T,N}} \delta V + \frac{1}{2} \left. \frac{\partial^2 P}{\partial V^2} \right|_{T,N} \delta V^2 + \frac{1}{3!} \left. \frac{\partial^3 P}{\partial V^3} \right|_{T,N} \delta V^3$$

$$\text{as } -\delta P \delta V \geq 0 \Rightarrow \delta P \delta V \leq 0 \Rightarrow \frac{1}{2} \left. \frac{\partial^2 P}{\partial V^2} \right|_{T,N} \delta V^2 + \frac{1}{3!} \left. \frac{\partial^3 P}{\partial V^3} \right|_{T,N} \delta V^3 \leq 0$$



$$\Rightarrow \left. \frac{\partial^2 P}{\partial V^2} \right|_{T,N} = 0 \quad \text{and} \quad \left. \frac{\partial^3 P}{\partial V^3} \right|_{T,N} < 0 \quad \text{as } \delta V^4 > 0$$