Exam 1 Equations

$$E_N = E_A + E_R$$

$$F_A = \frac{\left(1.602 \times 10^{-19}\right)^2}{4\pi \left(8.85 \times 10^{-12}\right) r^2} \left(\|Z_1\|\right) \left(\|Z_2\|\right) \tag{1.1}$$

- \bullet r Distance in m
- \bullet Z Number of Valence Electrons
- F_A Interatomic Force in N

Force
$$=\frac{dE}{dr}$$
 (1.2)

Elastic Modulus =
$$\frac{dF}{dr}$$
 (1.3)

Lattice Parameters 1.1

$$a_{\text{BCC}} = \frac{4r}{\sqrt{3}}$$
$$a_{\text{FCC}} = \frac{4r}{\sqrt{2}}$$

- $a_{\rm HCP} = \frac{c}{1.622}$
- a Lattice Parameter
- r Radius of atom

Volume of Hexagonal Prism 1.2

$$V_H = \frac{3\sqrt{3}}{2}a^2h {1.7}$$

Densities 1.3

$$\rho = \frac{nA}{V_C N_A} \tag{1.8}$$

- n Number of atoms/unit cell
- \bullet A Molar Mass of Material
- \bullet V_C Volume of Unit Cell in cm
- N_A Avogadro's Number (6.022×10^{23})

Planar Density =
$$\frac{\frac{\text{Atoms}}{2\text{D Unit Area}}}{\frac{\text{Area}}{2\text{D Repeat Unit}}}$$
 (1.9)

$$Linear\ Density = \frac{\#\ of\ Atoms\ in\ a\ Direction}{Magnitude\ of\ Linear\ Vector} \qquad (1.10)$$

• The repeat units/vector magnitude are in terms of atomic radii

$$APF = \frac{\frac{Atoms}{Unit \text{ Cell}} \left(\frac{4}{3}\pi \left(atom \text{ radius}\right)^{3}\right)}{Unit \text{ Cell Volume}}$$
(1.11)

$$\%$$
IC = $\left(1 - e^{\frac{(x_A - x_B)^2}{4}}\right) \times 100\%$ (1.12)

\bullet x - Electronegativities

Thermal Expansion

$$\frac{\Delta L}{L_0} = \alpha \left(T_2 - T_1 \right) \tag{1.13}$$

- $\begin{array}{ccc}
 \bullet & E \uparrow, T_m \uparrow \\
 \bullet & E \uparrow, \alpha \downarrow
 \end{array}$

Convert between Coordinates

$$a_{1} = \frac{1}{3} (2X - Y)$$

$$a_{2} = \frac{1}{3} (2Y - X)$$

$$a_{3} = -(a_{1} + a_{2})$$

$$c = Z$$

$$(1.14)$$

$a_1 + a_2 + a_3 = 0$

Planes 1.6

- 1. Given x, y, z as intersects
- (1.4)
- 2. Convert to $\frac{1}{x}$, $\frac{1}{y}$, $\frac{1}{z}$ 3. Reduce to smallest common denominator 4. Leave as $\left(\frac{1}{x}, \frac{1}{y}, \frac{1}{z}\right)$
- (1.5)

Light Refraction (1.6) **1.7**

$$D = \frac{n\lambda}{2\sin\theta} \tag{1.15}$$

- n = 1
- λ Wavelength in nm
- θ Angle of Incidence
- θ is usually given as 2θ . Be careful

Randoms

$$D_{HKL} = \frac{a}{\sqrt{h^2 + k^2 + l^2}} \tag{1.16}$$

• ONLY for cubic structures

2 Exam 2 Equations

2.1 Ch. 4 Imperfections

2.1.1 Vacancies

Number of Vacancies:

$$N_V = ext{Vacancy Frac.} imes rac{N_A
ho_{ ext{Ele}}}{A_{ ext{Ele}}}$$

- A_{Ele} Atomic Weight (g/mol)
- $\rho_{\rm Ele}$ Element Density (g/c³m)
- N_A Avogadro's Number (6.022×10^{23})

Number of Potential Vacancy Sites:

$$N = \frac{\rho_{\rm Ele} N_A}{A_{\rm Ele}}$$

- $A_{\rm Ele}$ Atomic Weight (g/mol)
- $\rho_{\rm Ele}$ Element Density (g/c³m)
- N_A Avogadro's Number (6.022×10^{23})

Grains per Area:

$$n_M = \left(\frac{100}{M}\right)^2 = 2^{G-1}$$

- n_M Grains/in²
- \bullet G Grain Size Number

$$-G = -6.6457 \log (\bar{\ell}) - 3.298$$
 for mm $-G = -6.6353 \log (\bar{\ell}) - 12.6$ for in.

2.1.2 Mean Intercept Length

$$\bar{\ell} = \frac{L_T}{PM}$$

- L_T Total Length of all Lines
- ullet P Number of Grain Intersections
- ullet M Magnification

$$M = \frac{\text{Scale Length}}{\# \text{ on Scale Bar}}$$
 (2.5)

2.1.3 Complete Substitution

To have a complete substitution, it must be:

- $\Delta r < 15\%$
- Electronegativity $\leq .4$
- SAME Crystal Structure
- SAME Valence

2.1.4 Edges

Burger's Vector:

- Burger's Vector
 - \perp for Edge Dislocations
 - − || for Screw Dislocations
- Twin Boundary Symmetric Around Fault
- Stacking Fault NOT Symmetric Around Fault
 - High Angle E↑
 - Low Angle $E \downarrow$

2.2 Ch. 5 Diffusion

2.2.1 Diffusion Coefficient

$$D = D_0 \times e^{\frac{-Q_d}{RT}} \tag{2.6}$$

- D Diffusion Coefficient (m²/s)
- Q_d Activation Energy (J/mol, eV/atom)
- (2.1) R 8.314 (J/molK)
 - T Temperature (K)

$$D_1 t_1 = D_2 t_2 (2.7)$$

- \bullet D Diffusion Coefficients
- \bullet t Time

(2.2) **2.2.2** Flux

(2.3)

$$J = -DA\frac{dC}{dx} \tag{2.8}$$

- For Steady State Diffusion
- ullet D Diffusion Coefficient
- dC Δ Concentration (Low-High)
- \bullet dx Distance to Cross
- \bullet A Area

2.2.3 Diffusion Concentration

$$\frac{C(x,t) - C_0}{C_s - C_0} = 1 - \operatorname{erf}\left(\frac{x}{2\sqrt{Dt}}\right)$$
 (2.9)

$$Z = \frac{x}{2\sqrt{Dt}} = \frac{z - \text{point below}}{\text{point above - point below}}$$
 (2.10)

- C(x,t) Concentration at a point AND time
- C_0 Initial Concentration
- (2.4) C_2 Surface Concentration of DIFFUSING species
 - \bullet x Position
 - ullet D Diffusion Coefficient
 - \bullet t Time

Trigonometry \mathbf{A}

A.1 Trigonometric Formulas

$$\sin\left(\alpha\right) + \sin\left(\beta\right) = 2\sin\left(\frac{\alpha+\beta}{2}\right)\cos\left(\frac{\alpha-\beta}{2}\right) \qquad (A.1) \qquad \cos\left(\alpha\right) + \cos\left(\alpha\right) = 2\cos\left(\frac{\alpha+\beta}{2}\right)\cos\left(\frac{\alpha-\beta}{2}\right)$$

$$\cos(\theta)\sin(\theta) = \frac{1}{2}\sin(2\theta) \tag{A.2}$$

$\sin(\alpha) \pm \sin(\beta) = 2\sin\left(\frac{\alpha \pm \beta}{2}\right)\cos\left(\frac{\alpha \mp \beta}{2}\right)$ (A.20)

$$\cos(\alpha) + \cos(\alpha) = 2\cos\left(\frac{\alpha+\beta}{2}\right)\cos\left(\frac{\alpha-\beta}{2}\right)$$
 (A.21)

(A.2)
$$\cos(\alpha) - \cos(\beta) = -2\sin(\frac{\alpha+\beta}{2})\sin(\frac{\alpha-\beta}{2})$$
 (A.22)

A.2Euler Equivalents of Trigonometric A.9 **Functions**

$$e^{\pm i\alpha} = \cos(\alpha) \pm i\sin(\alpha)$$
 (A.3)

$$\sin\left(x\right) = \frac{e^{ix} + e^{-ix}}{2} \tag{A.4}$$

$$\cos\left(x\right) = \frac{e^{\imath x} - e^{-\imath x}}{2\imath} \tag{A.5}$$

$$\sinh\left(x\right) = \frac{e^x - e^{-x}}{2} \tag{A.6}$$

$$\cosh\left(x\right) = \frac{e^x + e^{-x}}{2} \tag{A.7}$$

Pythagorean Theorem for Trig

Sum-to-Product Identities

$$\cos^2(\alpha) + \sin^2(\alpha) = 1^2 \tag{A.23}$$

Rectangular to Polar A.10

$$a + ib = \sqrt{a^2 + b^2}e^{i\theta} = re^{i\theta} \tag{A.24}$$

$$\theta = \begin{cases} \arctan\left(\frac{b}{a}\right) & a > 0\\ \pi - \arctan\left(\frac{b}{a}\right) & a < 0 \end{cases}$$
 (A.25)

Polar to Rectangular A.11

$$re^{i\theta} = r\cos(\theta) + ir\sin(\theta)$$
 (A.26)

Angle Sum and Difference Identities **A.3**

$$\sin(\alpha \pm \beta) = \sin(\alpha)\cos(\beta) \pm \cos(\alpha)\sin(\beta) \tag{A.8}$$

$$\cos(\alpha \pm \beta) = \cos(\alpha)\cos(\beta) \mp \sin(\alpha)\sin(\beta) \tag{A.9}$$

A.4 Double-Angle Formulae

$$\sin(2\alpha) = 2\sin(\alpha)\cos(\alpha) \tag{A.10}$$

$$\cos(2\alpha) = \cos^2(\alpha) - \sin^2(\alpha) \tag{A.11}$$

A.5Half-Angle Formulae

$$\sin\left(\frac{\alpha}{2}\right) = \sqrt{\frac{1 - \cos\left(\alpha\right)}{2}} \tag{A.12}$$

$$\cos\left(\frac{\alpha}{2}\right) = \sqrt{\frac{1 + \cos\left(\alpha\right)}{2}} \tag{A.13}$$

Exponent Reduction Formulae A.6

$$\sin^2(\alpha) = \frac{1 - \cos(2\alpha)}{2} \tag{A.14}$$

$$\cos^2(\alpha) = \frac{1 + \cos(2\alpha)}{2} \tag{A.15}$$

Product-to-Sum Identities A.7

$$2\cos(\alpha)\cos(\beta) = \cos(\alpha - \beta) + \cos(\alpha + \beta) \tag{A.16}$$

$$2\sin(\alpha)\sin(\beta) = \cos(\alpha - \beta) - \cos(\alpha + \beta) \tag{A.17}$$

$$2\sin(\alpha)\cos(\beta) = \sin(\alpha + \beta) + \sin(\alpha - \beta) \tag{A.18}$$

$$2\cos(\alpha)\sin(\beta) = \sin(\alpha + \beta) - \sin(\alpha - \beta) \tag{A.19}$$