# Chem 30324, Spring 2018, Homework 1

# Due on January 24, 2018

#### **Problem 1: Discrete, probably**

In five card stud, a poker player is dealt five cards from a standard deck of 52 cards.

- 1. How many different 5-card hands are there? (Remember, in poker the order in which the cards are received does *not* matter.)
- 2. What is the probability of being dealt four of a kind (a card of the same rank from each suit)?
- 3. What is the probability of being dealt a flush (five cards of the same suit)?

## **Problem 2: Continuous, probably**

The probability distribution function for a random variable x is given by  $P(x) = xe^{-2x}, 0 \le x < \infty$ .

- 1. Is P(x) normalized? If not, normalize it. Plot the normalized P(x).
- 2. What is the most probable value of x?
- 3. What is the expectation value of x?
- 4. What is the variance of x?

### **Problem 3: One rough night**

It's late on a Friday night and people are stumbling up Notre Dame Ave. to their dorms. You observe one particularly impaired individual who is taking steps of equal length \SI{1}{m} to the north or south (i.e., in one dimension), with equal probability.

- 1. What is the furthest distance the person could travel after 20 steps?
- 2. What is the probability that the person won't have traveled any net distance at all after 20 steps?
- 3. What is the probability that the person has traveled half the maximum distance after 20 steps?
- 4. Plot the probability of traveling a given distance vs distance. Does the probability distribution look familiar? You'll see it again when we talk about diffusion.

### Problem 4: Now this is what I call equilibrium

The Boltzmann distribution tells us that, at thermal equilibrium, the probability of a particle having an energy E is proportional to  $\exp(-E/k_{\rm B}T)$ , where  $k_{\rm B}$  is the Boltzmann constant. Suppose a bunch of gas particles of mass m are in thermal equilibrium at temperature T and are traveling back and forth in one dimension with various velocities v and kinetic energies  $K = mv^2/2$ .

- 1. What is the expectation value of the velocity  $\boldsymbol{v}$  of a particle?
- 2. What is the expectation value of the kinetic energy K of a particle? How does your answer depend on the particle mass? On temperature?