

## SV3 - Simulation de réactions chimiques

December 5, 2017

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In [6]: import numpy as np
import matplotlib.pyplot as plt
import time

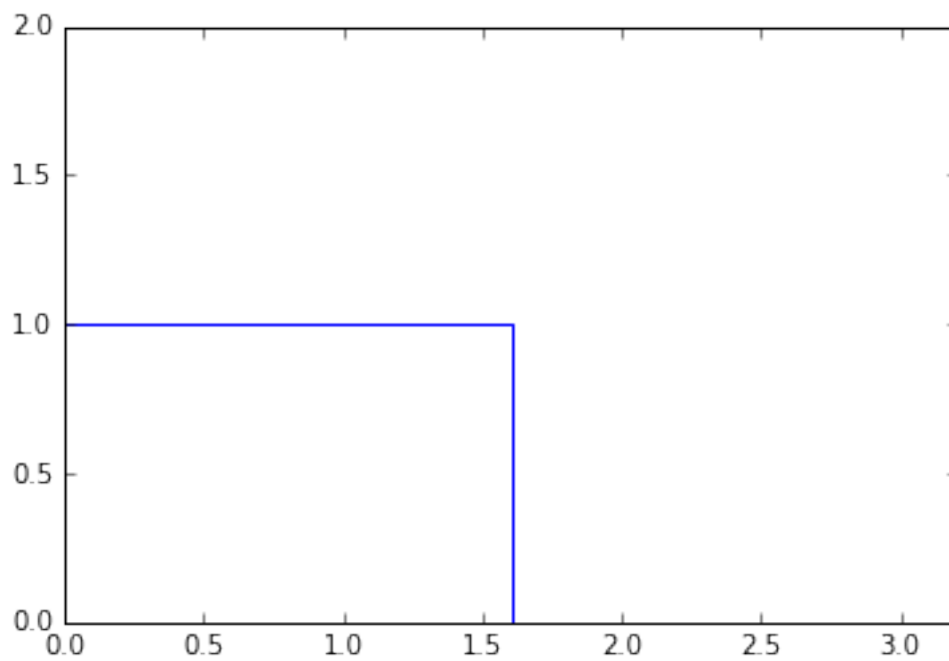
%matplotlib inline

def f_time(c, r):
    return (1/c) * np.log(1/r)

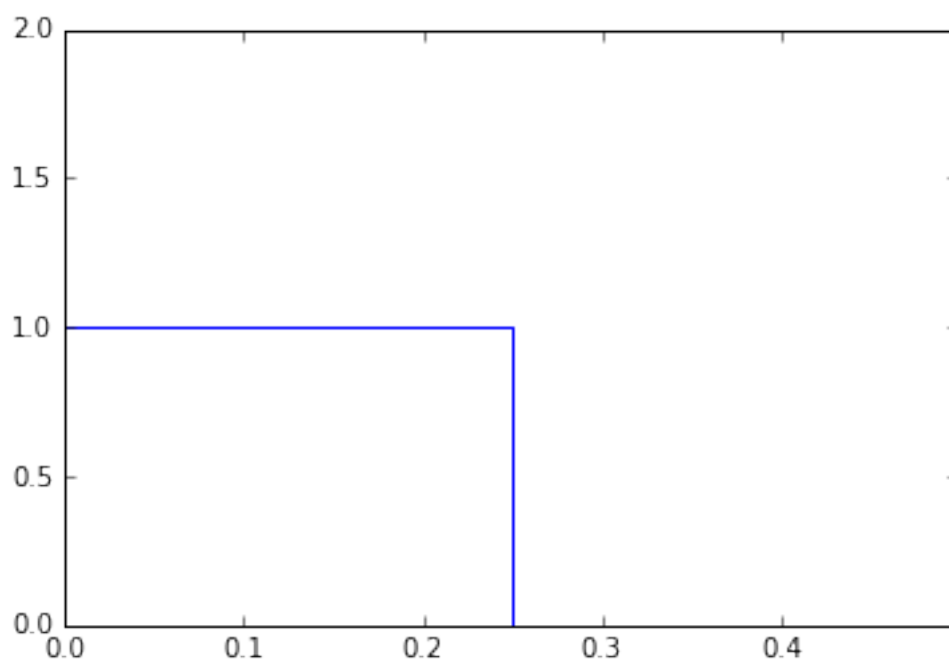
def create_graph(t):
    plt.xlim((0, t*2))
    plt.ylim((0, 2))
    plt.plot([0, t, t, 2], [1, 1, 0, 0])
    plt.show()

# 3 graphes
for i in range(3):
    r = np.random.random()
    print("r: {}".format(r))
    create_graph(f_time(1, r))
```

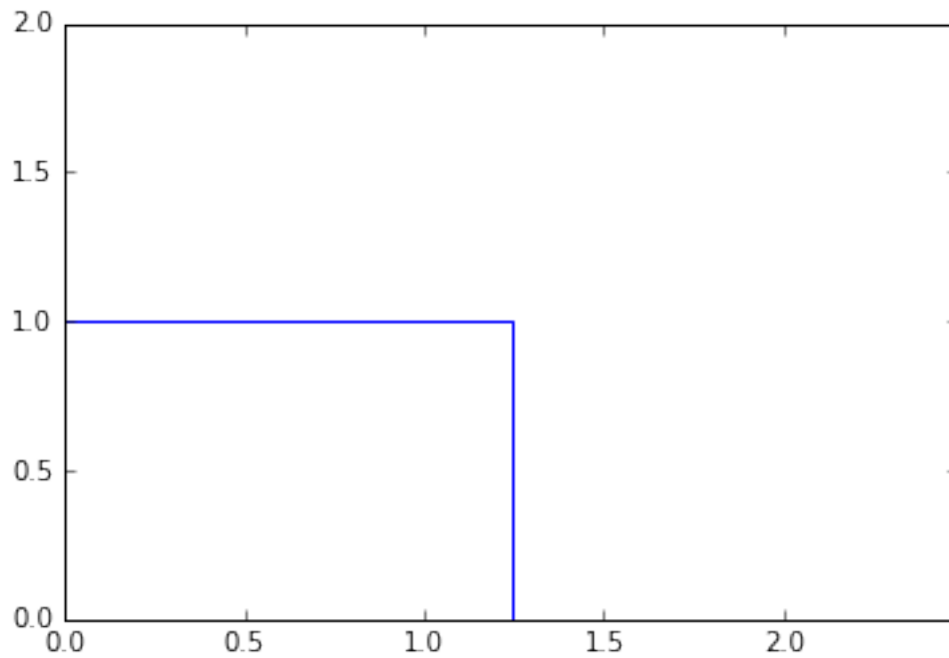
r: 0.2013692703144434



r: 0.7794901476290889



r: 0.2890000675271911



```
In [5]: def create_multi_graphs(g):
        x = g
        y = []
        for i in range(len(x)):
            y.append(1 - (i / x.size))

        plt.xlim((0, np.max(x)*1.1))
        plt.ylim((0, 2))

        # Ajout du premier point
        y = np.concatenate((np.array([1]), y))
        x = np.concatenate((np.array([0]), x))

        # Ajout du dernier point
        y = np.concatenate((y, (np.array([0]))))
        x = np.concatenate((x, (np.array([x[x.size-1]]))))

        plt.step(x, y)
        plt.show()

        # 12 graphes
        cs = [0.1, 1, 100]# 0.1, 1, 100
```

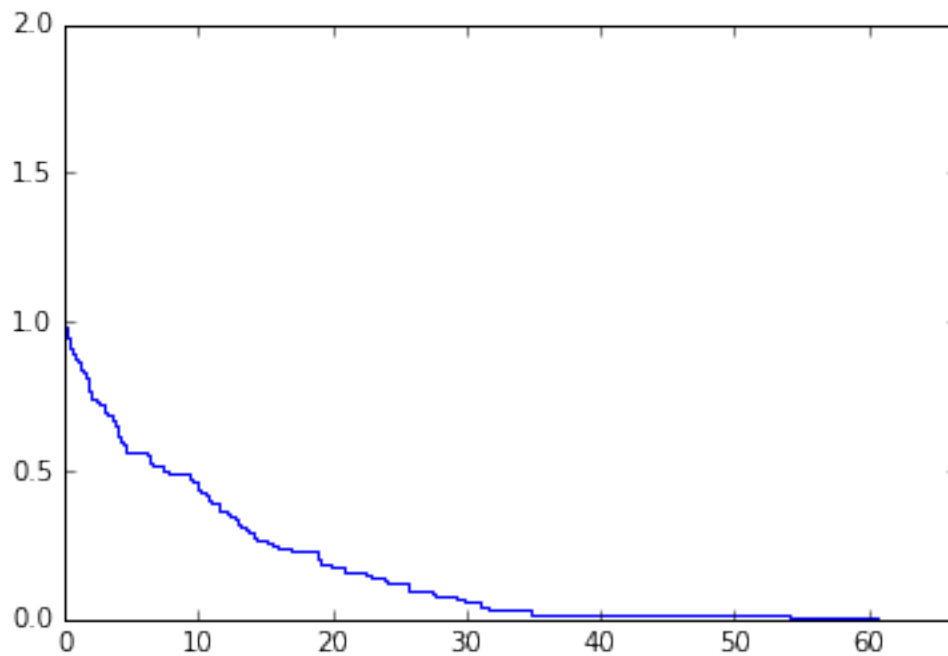
```

nb = [100, 10000, 1000000, 1]

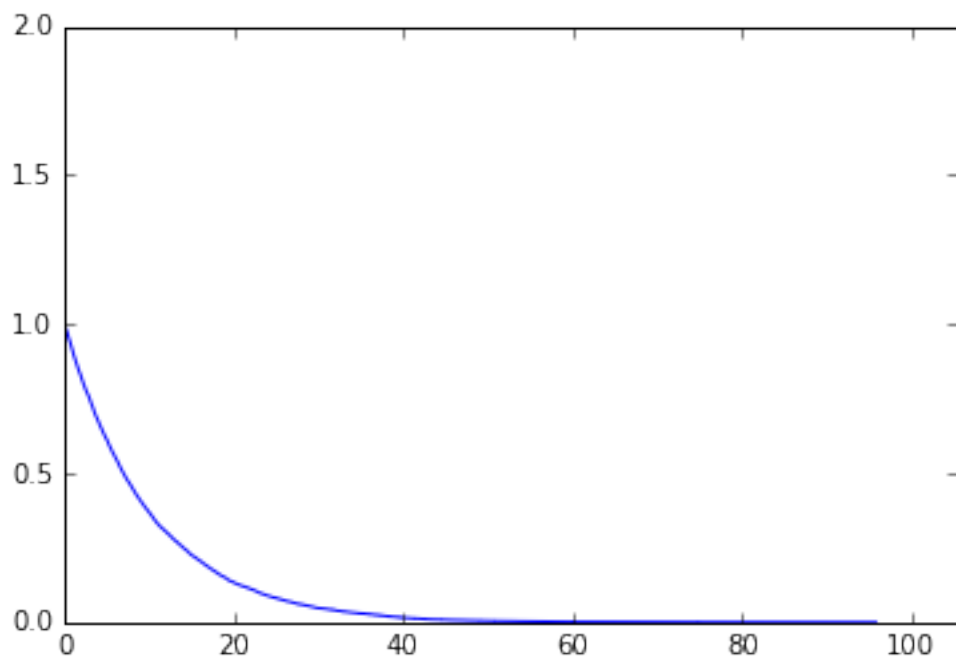
for c in cs:
    for n in nb:
        res = []
        if n == 1: # 1 minute
            t1 = time.time()
            while (time.time() - t1 < 60):
                r = np.random.random()
                res.append(f_time(c, r))
            print("c: {} and nb: {}".format(c, len(res)))
        else:
            for i in range(n):
                r = np.random.random()
                res.append(f_time(c, r))
            print("c: {} and nb: {}".format(c, n))
        res = np.sort(res)
        create_multi_graphs(res)

```

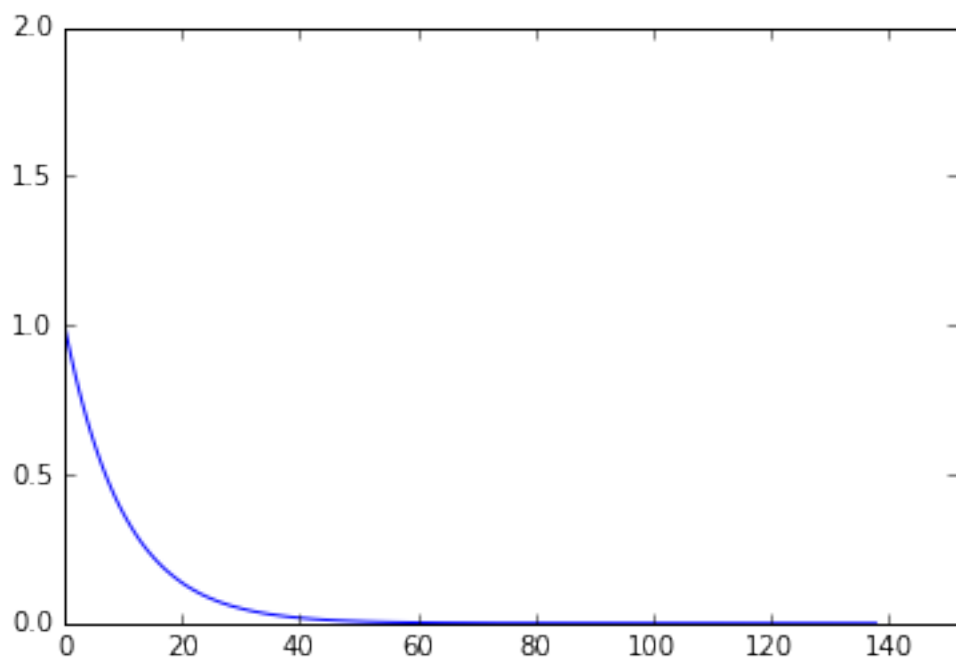
c: 0.1 and nb: 100



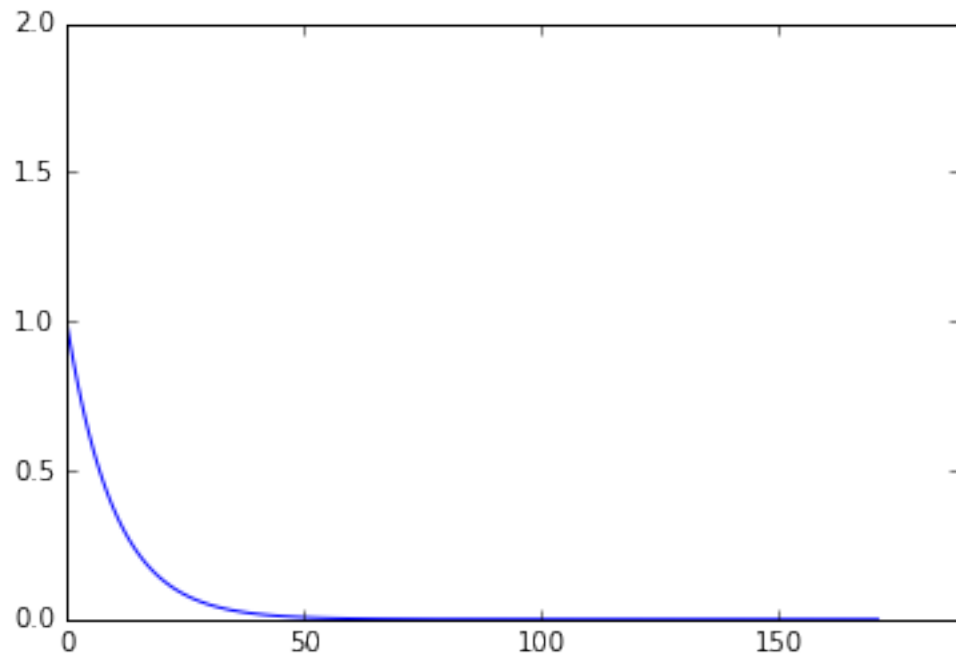
c: 0.1 and nb: 10000



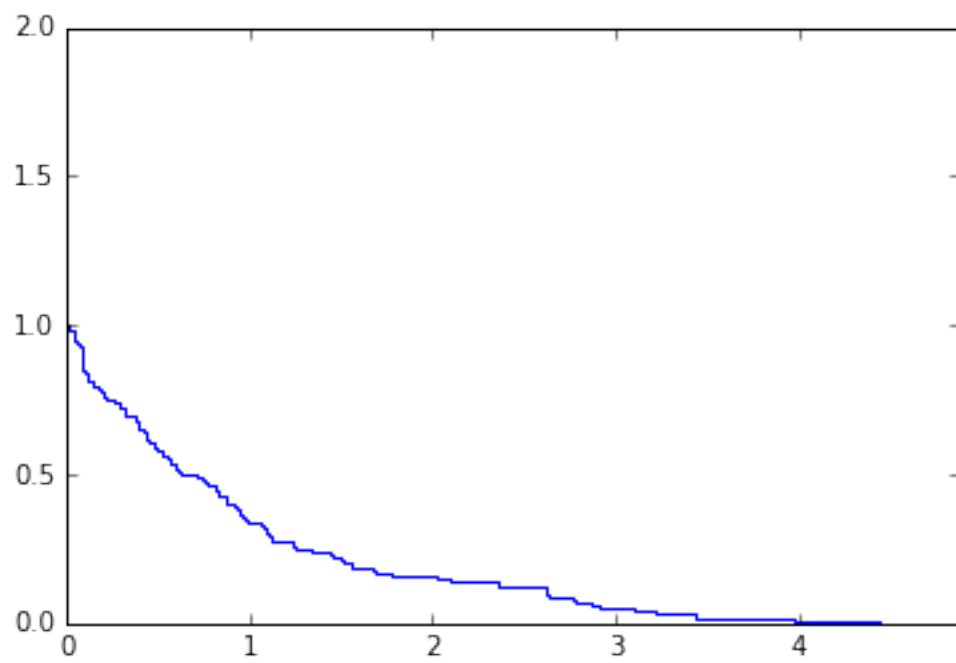
c: 0.1 and nb: 1000000



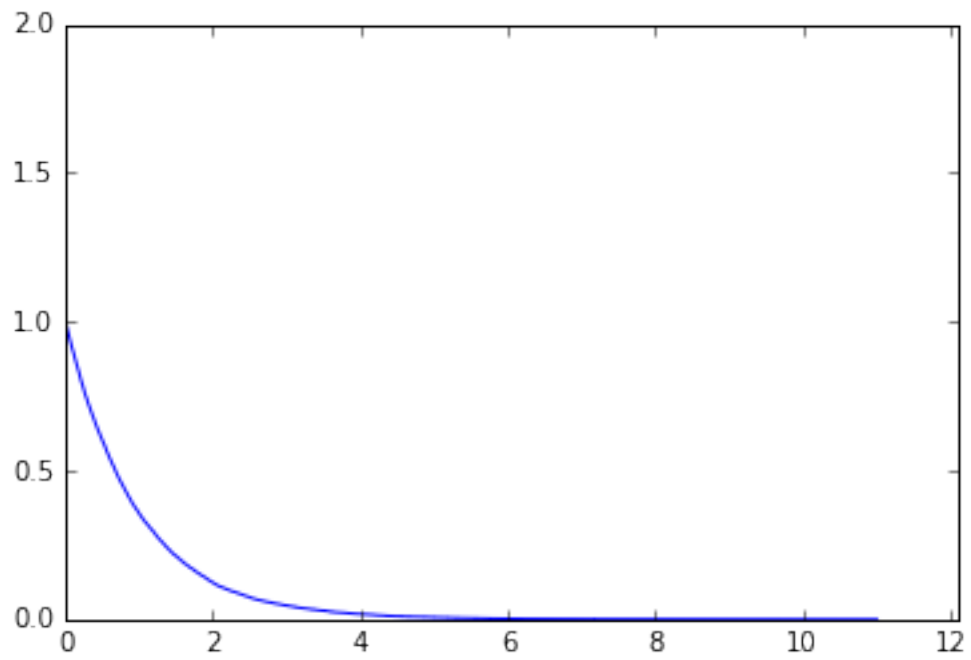
c: 0.1 and nb: 42394971



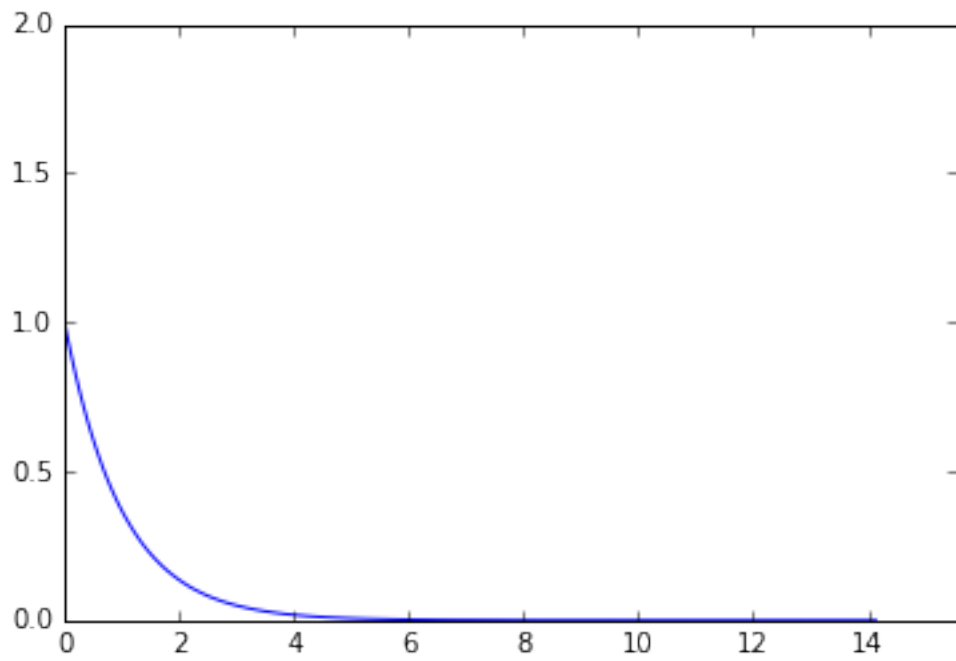
c: 1 and nb: 100



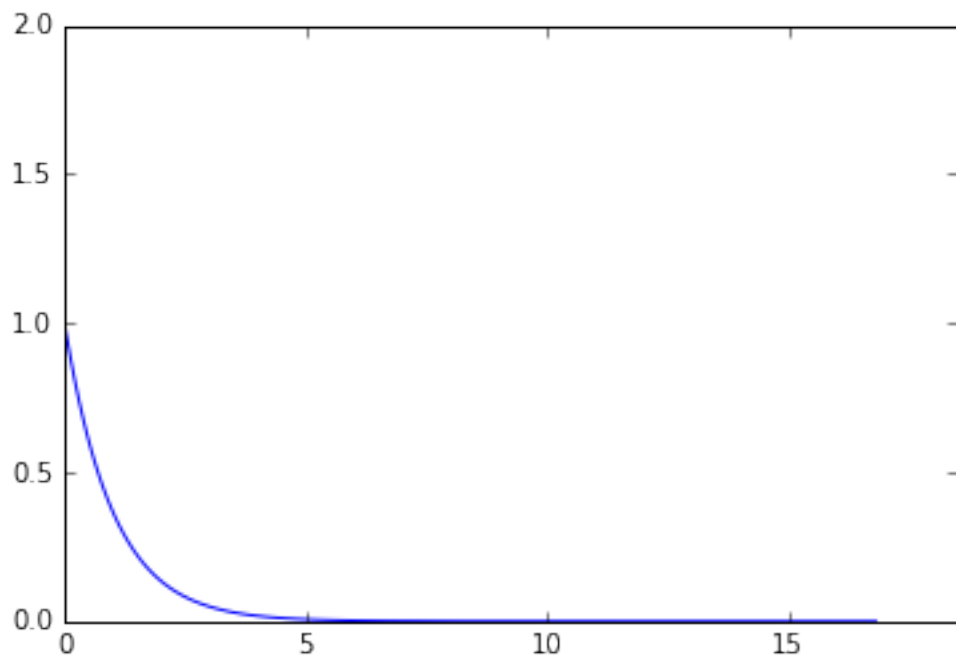
c: 1 and nb: 10000



c: 1 and nb: 1000000

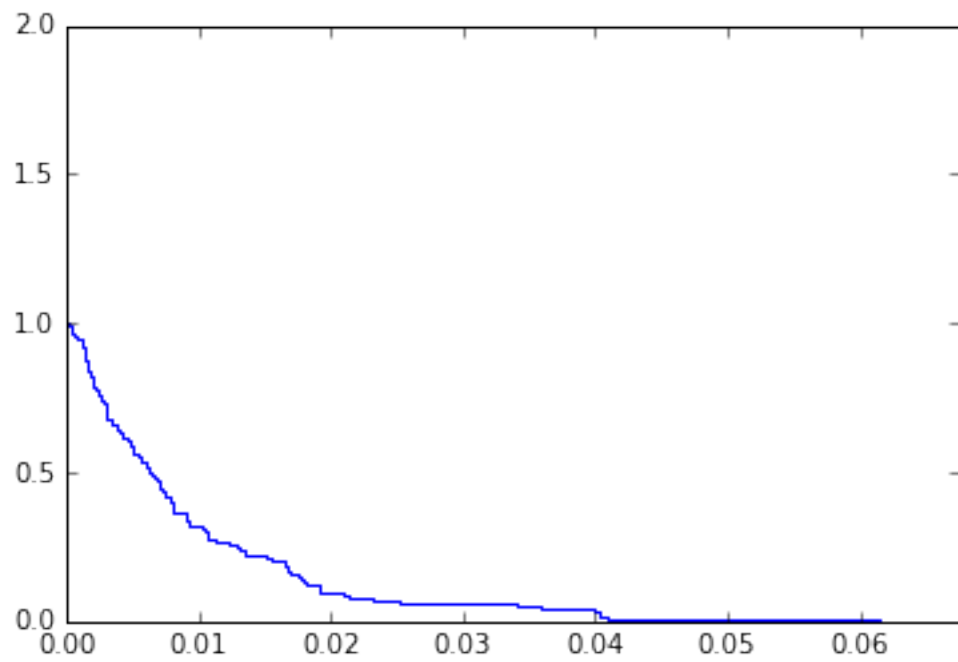


c: 1 and nb: 44894685

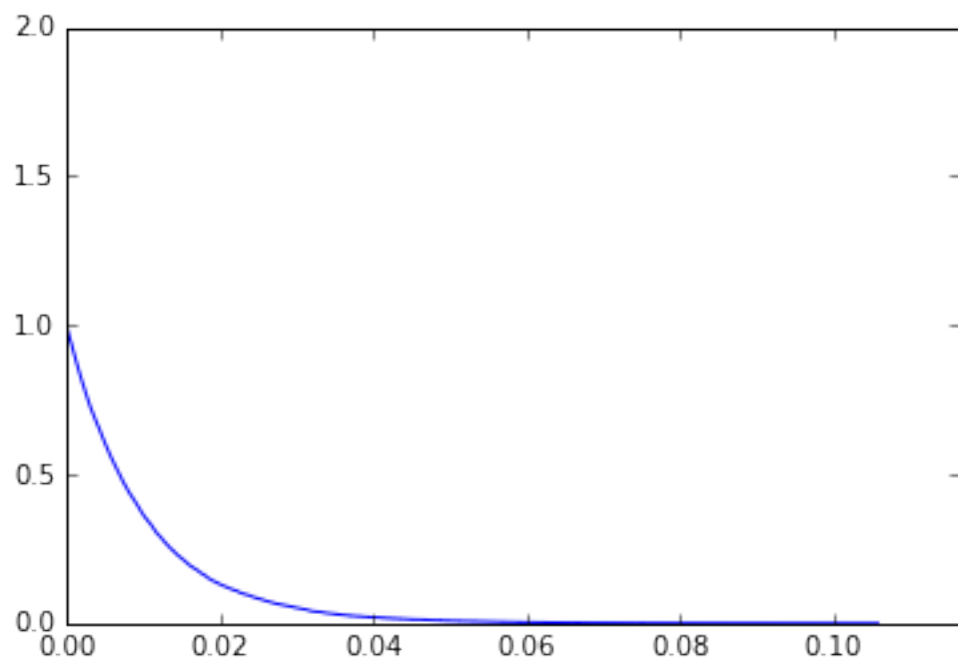




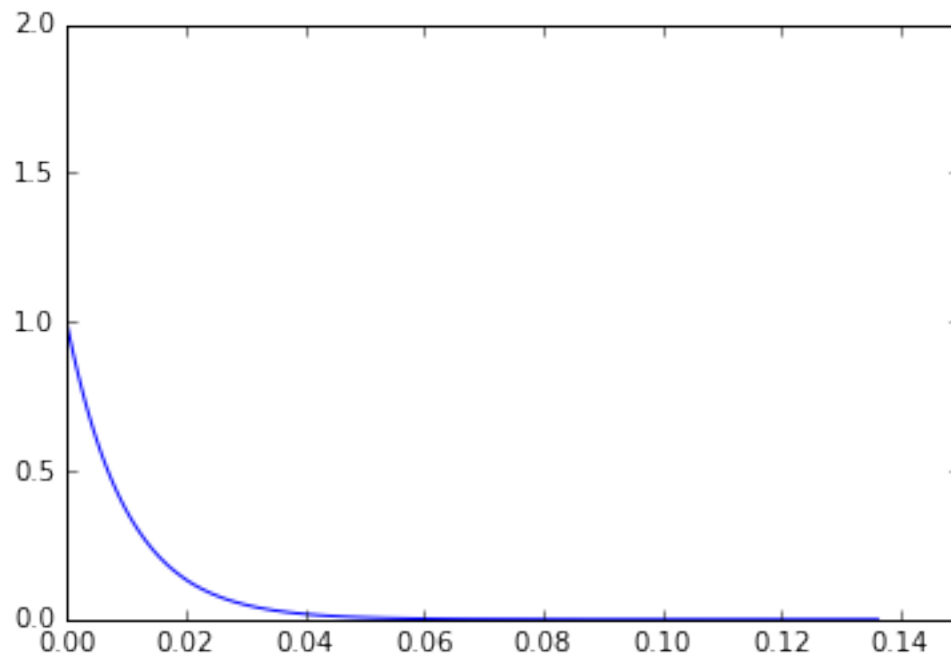
c: 100 and nb: 100



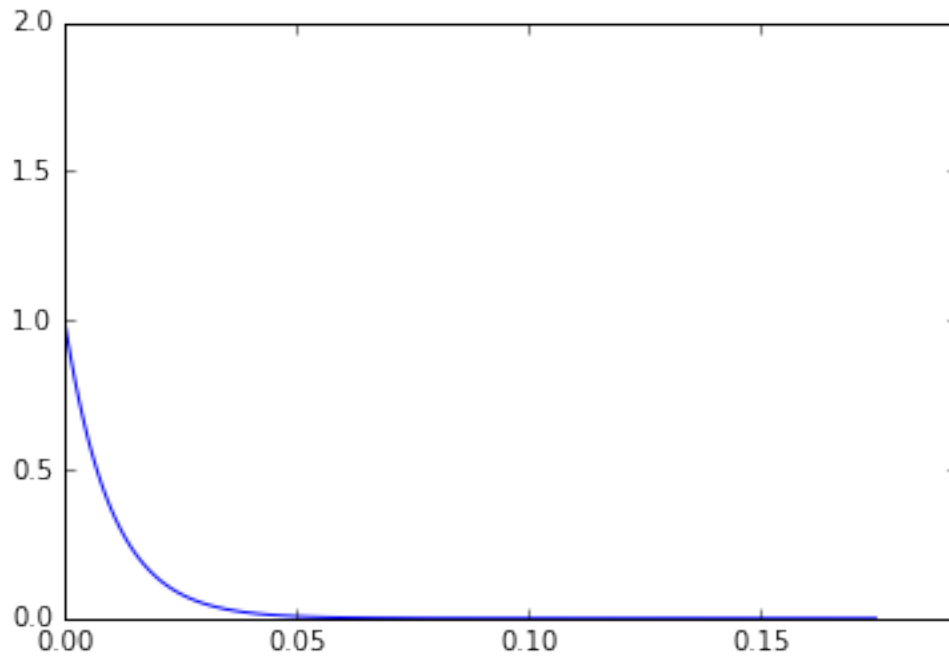
c: 100 and nb: 10000



c: 100 and nb: 1000000



c: 100 and nb: 44918019



Comme on peut le voir, plus le nombre d'essais est grand, plus on observe une courbe "lisse" qui converge vers 0.

En une minute, on arrive à faire environ 42 millions de calcul.

La variation du  $c$  fait varier simplement l'échelle de temps qu'on peut voir sur les graphes, avec un grand  $c$ , on voit que la réaction met beaucoup moins de temps à se déclencher en moyenne.