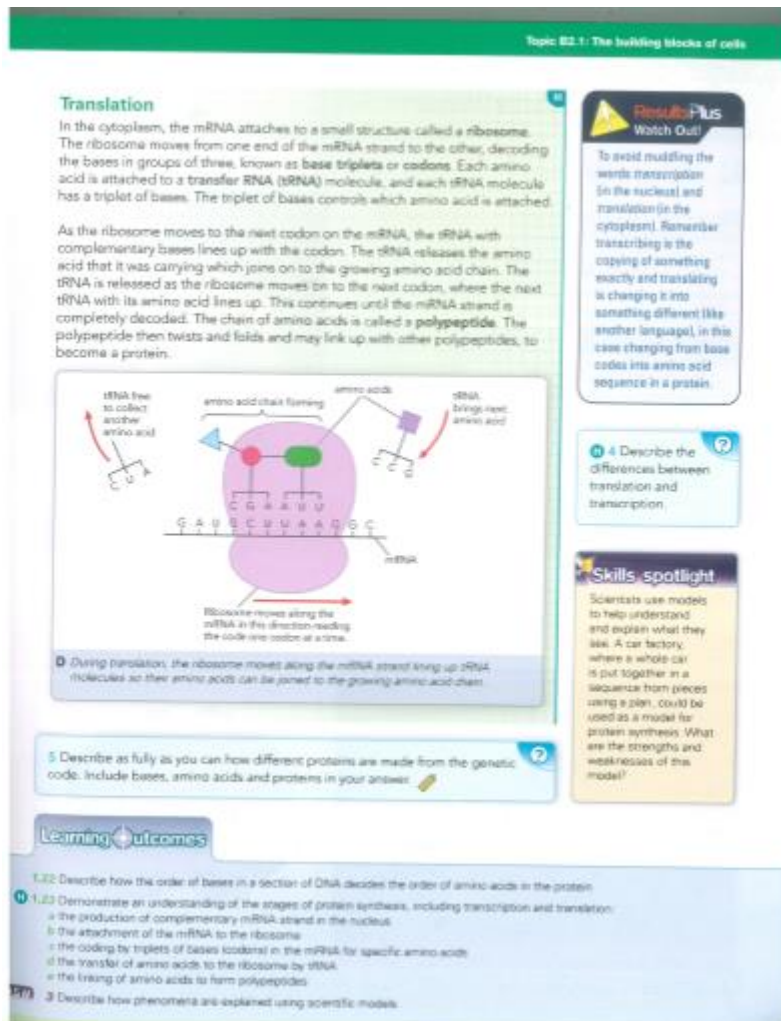


## Research the Problem: Edexcel GCSE Science

Unlike any other material I have covered, these are three official educational text books used by the school my client works at to aid in teaching GCSE students the content they will be tested on.



These are huge books full of information directly related to the content that the reader will be tested on. Notice how the page above has:

- Main text
- A model
- Exam tips on the side
- Questions to test your understanding of the text on the side
- Skills that make a good scientist on the side
- Questions at the end of each part to mimic the content you would find in an actual examination

If you have really good eyes, you would notice that unlike the other books that are split into named chapters, these books are separated into science > topic > section of that topic. This allows the

entirety of GCSE science to be numbered up according to the current content.

**Uses of the noble gases**

The noble gases are useful because they are unreactive.

- Xenon and argon were formerly used inside filament lamps, instead of air, to stop the hot filament reacting with oxygen and burning away.
- Argon and helium are used in welding, to form a blanket over the hot metal to stop it reacting with oxygen from the air.
- Argon is used in fire-extinguishing systems because it is non-flammable. Spaces such as a computer server room can be filled with argon if a fire breaks out.
- Helium has a low density and so is used for filling balloons and airships.
- Neon is often used in fluorescent lamps and advertising displays as it produces a red coloured light when electric current is passed through the tube under low pressure.

**E** Welding uses an inert gas to stop the hot metal reacting.

**F** Helium is used for filling airships.

**Skills spotlight**

Patterns in data can be used to find missing values by **interpolation**. Figure D shows the densities of some noble gases.

**a** Draw a graph to show the densities of the noble gases. Leave a space at the correct place for argon.

**b** Use your graph to estimate the density of argon.

**c** How could a chemist find out if this estimate is correct?

**Maths skills**

A line of best fit on a scatter graph can be used to estimate unknown values.

Estimating an unknown value within the range of known values is called **interpolation**.

Estimating an unknown value outside the range of known values is called **extrapolation**.

**ResultsPlus**  
Watch Out!

You could think of the noble gases as being like

3 Explain why the noble gases are non-flammable.

4 Hydrogen is the only gas with a lower density than helium. Suggest why helium is used in balloons rather than hydrogen.

5 Suggest why it was easier for early chemists to discover elements such as oxygen, rather than elements such as argon.

6 Explain why the ability to make very accurate measurements was important in the discovery of the noble gases.

Unlike

all the other books that I have researched, this book very clearly has a lot more text than images. This is not to say there are no images, they are used for an example of something just covered in the paragraph as a real life example of what they just explained, or it is either a model or a diagram.

While I included this into my research into other methods of solving my problem, this book is actually the problem. I have personally seen people stare at pages of this book for extended periods of time trying to understand concepts, their eyes glossed over before the information sinks in.

Interestingly enough, my levels will be designed using the same system as these books. I will also try to use the information contained in these books to design the levels. It is also interesting to compare these books to the BBC Bitesize digital learning resource. While the Bitesize website has the high ground by being digital, I find that these books deliver a higher quality of education. Of course, this only applies if you're doing the Exedcel GCSEs.