BME 4400 Fall 2022

Assignment #4

Due: October 20, 2022 9:00 PM

specific activity. A cyclotron provides ¹⁸F with specific activity of 90,000 mCi/µg. From this, [¹⁸F]-FDG will be produced and shipped to nearby PET imaging facilities. If a 440 MBq dose is to be given to a patient 8 hr after ¹⁸F production from the cyclotron, what mass of [18F]-FDG will be injected?

coincidence detections 2 positrons

2. Look at the figure at the bottom of the page, and look up appropriate material characteristics for cortical bone and soft tissue from the NIST site (https://www.nist.gov/pml/x-ray-mass- attenuation-coefficients, tables 2 and 4; use values tabulated at the nearest energy value). A positron emitter (assume ¹⁸F) has accumulated in the purple region and is producing an activity of 0.25 MBq along the horizontal path leading to the D+ and D- detectors. Compute the number of coincidence detections expected over a 2 min acquisition.

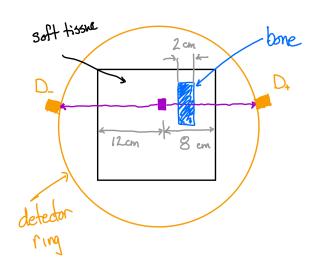
detector

- 3. When imaging 99mTc with a NaI(Tl) scintillation screen with PMT array,
 - a. what is a typical photon energy window used to accept detections?
 - b. if a scintillation event results in a lower energy deposition that is below this window, why should you not count this event?
 - c. if scintillation event results in a higher energy deposition that is above this window, why should you not count this event?

dose & adjacent Assume that 600/ of the second control of 100 MBq of 99mTc sulfur colloid. Assume that 60% of the activity is trapped in the liver, 30% is trapped in the spleen, and 10% in red bone marrow. Assume instantaneous uptake and no biological excretion.

resolution

5. In a ¹⁸FDG abdominal scan made with a 90 cm diameter detector ring, how high a resolution are you likely to be able to get and why?



Z	Nuclide	Half-life	Photon Energy (keV)					
24	Chromium-51	28d	320					
31	Gallium-67	79.2h	92, 184, 296					
34	Selenium-75	120d	265					
38	Strontium-87m	2.8h	388					
43	Technetium-99m	6h	140					
49	Indium-111	2.8d	173, 247					
"	Indium-113m	1.73h	393					
53	Iodine-123	13.3h	159					
00	Iodine-125	60d	35, 27					
	Iodine-131	8.04d	364					
54	Xenon-133	5.3d	81					
80	Mercury-197	2.7d	77					
81	Thallium-201	73h	135, 167					
	Po	sitron Emitters						
	Nuclide	Half-life	Positron Energy (keV					

326 20.3min Carbon-11 6 432 10.0min Nitrogen-13 696 2.1min 8 Oxygen-15 202 110min 9 Fluorine-18 656 12.7h 29 Copper-64 1900 68min Gallium-68 31 3340 Arsenic-72 26h 33 3600 16.1h Bromine-76 35 1.3min 3150 Rubidium-82 37 3100 3.5min Iodine-122 53

Source: Wolbarst, 1993.

hydrogen	-		257	276	150	6	153	6	1977	15	0.00	17	650	705	650	10	55	helium
Là																		He
п																		
1.0079 lithium	beryllium											1	boron	carbon	nitrogen	oxygen	fluorine	4.0026 neon
3	4												5	6	7	8	9	10
Li	Be												В	C	N	0	F	Ne
6.941	9.0122												10.811	12.011	14.007	15.999	18.998	20.180
sodium 11	magnesium 12												aluminium 13	silicon 14	phosphorus 15	sulfur 16	chlorine 17	argon 18
200																100	10.00	_300
Na	Mg												ΑI	Si	Р	S	CI	Ar
22.990 potassium	24.305 calcium		scandium	titanium	vanadium	chromium	manganese	iron	cobalt	nickel	copper	zinc	26.982 gallium	28.086 germanium	30.974 arsenic	32.065 selenium	35,453 bromine	39,948 krypton
19	20		21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36
K	Ca		Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
	550 V C C C C C C C C C C C C C C C C C C											100						
39.098 rubidium	40.078 strontium		44.956 yttrium	47.867 zirconium	50.942 niobium	51.996 molybdenum	54.938 technetium	55.845 ruthenium	58,933 rhodium	58.693 palladium	63.546 silver	65.39 cadmium	69.723 indium	72.61 tin	74.922 antimony	78.96 tellurium	79.904 iodine	83.80 xenon
37	38		39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54
Rb	Sr		Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	1	Xe
85.468	87.62		88.906	91.224	92.906	95.94	[98]	101.07	102.91	106.42	107.87	112.41	114.82	118.71	121.76	127.60	126.90	131.29
caesium	barium	F7 70	lutetium	hafnium	tantalum	tungsten	rhenium	osmium	iridium	platinum	gold	mercury	thallium	lead	bismuth	polonium	astatine	radon
55	56	57-70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86
Cs	Ba	*	Lu	Hf	Ta	W	Re	Os	lr	Pt	Au	Hg	TI	Pb	Bi	Po	At	Rn
132.91 francium	137.33 radium		174.97 lawrencium	178.49 rutherfordium	180.95 dubnium	183.84 seaborgium	186.21 bohrium	190.23 hassium	192.22 meitnerium	195.08 ununnilium	196.97 unununium	200.59 ununbium	204.38	207.2	208.98	[209]	[210]	[222]
87	88	89-102	103	104	105	106	107	108	109	110	111	112		ununquadium 114				
Fr		* *	200000				Bh	Hs	Mt	S3387	_ 500000							
[223]	Ra	* *	Lr	Rf	Db	Sg	[264]	12691	[268]	UU11	Uuu	UUD 12771		Uuq				

*Lanthanide series

* * Actinide series

	lanthanum 57	cerium 58	praseodymium 59	neodymium 60	promethium 61	samarium 62	europium 63	gadolinium 64	terbium 65	dysprosium 66	holmium 67	erbium 68	thulium 69	ytterbium 70
S	La	Сe	Pr	Nd	Pm	Sm	Ĕu	Gd	Τ̈́b	Ďν	Но	Ĕr	Tm	Yb
	138.91	140.12	140.91	144.24	[145]	150.36	151.96	157.25	158.93	162.50	164.93	167.26	168.93	173.04
	actinium 89	thorium 90	protactinium 91	uranium 92	neptunium 93	plutonium 94	americium 95	curium 96	berkelium 97	californium 98	einsteinium 99	fermium 100	mendelevium 101	nobelium 102
	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No
	[227]	232.04	231.04	238.03	[237]	[244]	[243]	[247]	[247]	[251]	[252]	[257]	[258]	[259]