# Kinetic Nucleation in Thermal Non-Equilibrium

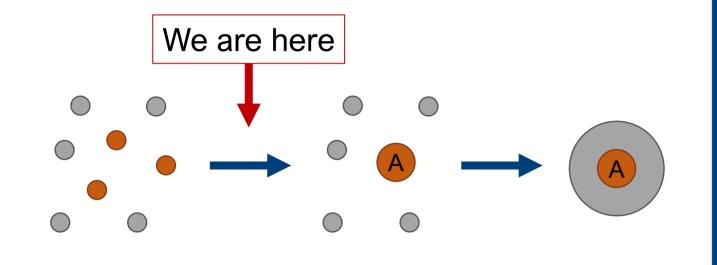
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# Take home

Kinetic nucleation is affected by internal and kinetic temperatures of clusters and temperature differences between cluster sizes. Nonetheless, the assumption of thermal equilibrium is generally justified for exoplanet atmospheres.

#### **Connection to Exoplanets**

Clouds form when materials (•) condense onto aerosols (•). In gaseous exoplanets, aerosols must form from the gas phase (•) via kinetic nucleation. With this work we look at the effect of thermal non-equilibrium.



# Background $T_{M}^{int} \qquad T_{M}^{kin}$ $T_{M+N}^{int} \qquad T_{M+N}^{kin}$ $T_{gas}$ $T_{N}^{int} \qquad \sigma_{1} (V_{r})$

#### Growth reaction rate *k*<sup>+</sup>:

- T<sup>int</sup> internal temperature
- Tkin kinetic temperature
- T<sub>gas</sub> gas temperature
- N, M, N+M cluster sizes
- *v<sub>r</sub>* relative velocity between colliding particles
- $\sigma_i(v_r)$  reaction cross section
- $f(v_r)$  velocity distribution

$$k_j^+ = \int_0^\infty \sigma_j(\nu_r) \, \nu_r \, f(\nu_r) \, d\nu_r$$

# Where it gets complicated

### Basis

$$k^{-} = k^{+} \alpha \frac{p^{\star}}{kT_{\text{gas}}} ABC$$

#### Reverse reaction rate k<sup>-</sup>:

- the principle of detailed balance
- chemical equilibrium as reference state [2]
- law of mass action including thermal nonequilibrium effects

# Thermal equilibrium

$$A = \exp\left(\frac{G_{(N+M)}^{\star}(T_{(N+M)}^{\mathrm{kin}}, p^{\star})}{RT_{(N+M)}^{\mathrm{kin}}} - \frac{G_{N}^{\star}(T_{N}^{\mathrm{kin}}, p^{\star})}{RT_{N}^{\mathrm{kin}}} - \frac{G_{M}^{\star}(T_{M}^{\mathrm{kin}}, p^{\star})}{RT_{M}^{\mathrm{kin}}}\right)$$

#### Kinetic-to-gas non-equilibrium

$$B = \exp\left(\frac{(T_{(N+M)}^{\rm kin} - T_{\rm gas})}{T_{(N+M)}^{\rm kin}} - \frac{(T_{N}^{\rm kin} - T_{\rm gas})}{T_{N}^{\rm kin}} - \frac{(T_{M}^{\rm kin} - T_{\rm gas})}{T_{M}^{\rm kin}}\right)$$

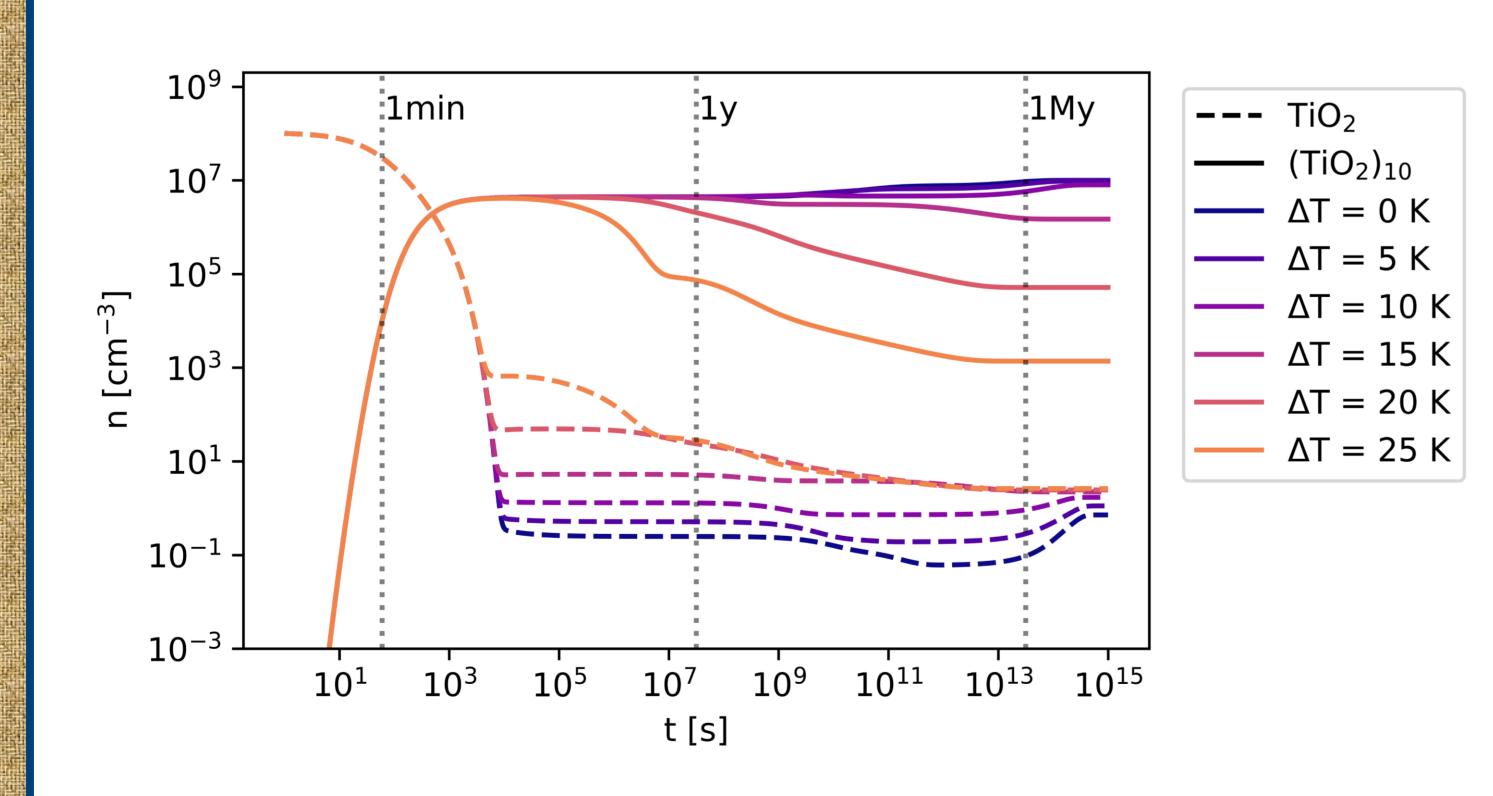
#### Internal-to-kinetic non-equilibrium

$$C = \exp\left(-\frac{\omega_{(N+M)}(T_{(N+M)}^{\mathrm{kin}}, T_{(N+M)}^{\mathrm{int}})}{kT_{(N+M)}^{\mathrm{kin}}} + \frac{\omega_{N}(T_{N}^{\mathrm{kin}}, T_{N}^{\mathrm{int}})}{kT_{N}^{\mathrm{kin}}} + \frac{\omega_{M}(T_{M}^{\mathrm{kin}}, T_{M}^{\mathrm{int}})}{kT_{M}^{\mathrm{kin}}}\right)$$

#### Parameters used:

- standard pressure  $p^* = 10^5 \text{ Pa}$
- cluster sizes
   N, M, N+M
- Gibbs free energy [4]  $G_N^*$   $(T_N^{kin}, p^*)$
- Internal change in Gibbs free energy  $\omega_N (T_N^{\rm kin}, T_N^{\rm int})$
- Gibbs free energy gauge α

## Results



#### **Assumptions for this example:**

- TiO<sub>2</sub> nucleation in a H<sub>2</sub> gas at  $T_{gas}$  = 1000 K
- Initial number density  $n_{TiO_2} = 10^8$  cm<sup>-3</sup>
- Internal-to-kinetic equilibrium  $T_N = T_N^{kin} = T_N^{int}$
- Temperature offset  $\Delta T = T_{(TiO_2)_{10}} T_{TiO_2}$

#### Conclusions

- Thermal non-equilibrium can enhance or reduce  $(TiO_2)_{10}$  formation.
- Kinetic nucleation in hot, low-density environments (like AGB stars [5]) can be affected by thermal non-equilibrium.
- Thermal equilibrium is a good assumption for exoplanet atmospheres.

#### Get in Touch!



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#### References

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