

Methods	Examples of mesoporous materials	Advantages	Disadvantages
Soft-templating method	<ul style="list-style-type: none"> <li>• Silica</li> <li>• Aluminosilicates</li> <li>• Organosilica</li> <li>• Carbon</li> <li>• Metal oxides</li> <li>• Metals</li> <li>• Polymers</li> <li>• Phosphates</li> </ul>	<ul style="list-style-type: none"> <li>• Controllable mesostructures and pore sizes</li> <li>• Tunable morphologies and easily processable</li> <li>• High quality product</li> <li>• large scale production</li> </ul>	<ul style="list-style-type: none"> <li>• Uses surfactants</li> <li>• Mesostructure formation is highly sensitive to the reaction conditions</li> <li>• Relatively low crystallinity</li> </ul>
Hard-templating method	<ul style="list-style-type: none"> <li>• Carbon</li> <li>• Metal oxides</li> <li>• Metals</li> <li>• Metal sulfides</li> <li>• Metal nitrides</li> <li>• Metal carbides</li> <li>• Zeolites</li> <li>• Non-oxide ceramics</li> </ul>	<ul style="list-style-type: none"> <li>• Low sensitivity to the reaction conditions</li> <li>• High quality product</li> <li>• Highly crystalline product</li> <li>• Ordered nanoarrays structure</li> </ul>	<ul style="list-style-type: none"> <li>• Uses preformed hard templates</li> <li>• High cost</li> <li>• Time consuming</li> </ul>
Multiple-templating method	<ul style="list-style-type: none"> <li>• Silica</li> <li>• Carbon</li> <li>• Metal oxides</li> <li>• Zeolites</li> </ul>	<ul style="list-style-type: none"> <li>• Hierarchically porous structure</li> </ul>	<ul style="list-style-type: none"> <li>• Requires multiple templates</li> <li>• High cost</li> <li>• Time consuming</li> </ul>
In-situ templating pathway	<ul style="list-style-type: none"> <li>• Silica</li> <li>• Carbon</li> <li>• Metal oxides</li> <li>• Metals</li> <li>• Polymers</li> </ul>	<ul style="list-style-type: none"> <li>• Simple method</li> <li>• No preformed templates or surfactants required</li> <li>• Low cost</li> </ul>	<ul style="list-style-type: none"> <li>• Low quality</li> <li>• Hard to obtain ordered structures</li> </ul>
Template-free packing method	<ul style="list-style-type: none"> <li>• Carbon</li> <li>• Metal oxides</li> <li>• Metal sulfides</li> </ul>	<ul style="list-style-type: none"> <li>• Simple method</li> <li>• Easily processable</li> <li>• No templates required</li> <li>• Highly crystalline product</li> </ul>	<ul style="list-style-type: none"> <li>• Hard to obtain ordered structures</li> </ul>
Reticular chemistry guiding approach	<ul style="list-style-type: none"> <li>• Metal-organic frameworks</li> <li>• Covalent organic frameworks</li> <li>• Zeolitic imidazolate frameworks</li> </ul>	<ul style="list-style-type: none"> <li>• Controllable mesostructures and pore sizes</li> <li>• No templates</li> <li>• Highly crystalline product</li> </ul>	<ul style="list-style-type: none"> <li>• Low stability</li> <li>• Mesostructure formation is highly sensitive to the reaction conditions</li> <li>• Not easily processable</li> </ul>