Homework No.5

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1 Part A: Vector Field

Matlab's quiver(X, Y, U, V) function plots arrows with directional components U and V at the Cartesian coordinates specified by X and Y. When implementing quiver, the first arrow originates from the point (X(1), Y(1)), extends horizontally according to U(1), and extends vertically according to V(1). This function scales the arrow lengths so that they do not overlap.

On the other hand, Matlab's contour(X, Y, Z) function creates a contour plot containing the isolines of a matrix Z, where Z contains height values on the x-y plane. contour automatically selects the contour lines to display. X and Y are the x and y coordinates in the plane, respectively.

Listing 1 implements functions *quiver* and *contour* to visualize the velocity field and pressure lines given a vector field. Figure 1 is the result.

```
1 %% HW05 part A - Velocity Field, adapted from (jose lopez salinas)'s solution
2 clear;
3 close all;
5% create points to visualize
6 \text{ xyLim} = 2.5;
7 xyStep = xyLim/10;
8 [x, y] = meshgrid(-xyLim : xyStep : xyLim);
10 VectorX = cos(y); % vector in the x direction
vectorY = sin(x); % vector in the y direction
13 V = sqrt(VectorX.^2 + VectorY.^2);
_{14} PHI = 6 + x.^3 / 3 - y.^2 .* x - y;
15 [Dx, Dy] = gradient(V, 0.2, 0.2);
17 % Display
18 figure;
19 quiver(x, y, VectorX, VectorY);
20 hold on;
21 contour(x, y, PHI);
22 colorbar;
23 hold off;
24 xlabel('x-axis');
25 ylabel('y-axis');
26 title('Velocity Field, and Pressure Lines');
```

Listing 1: Vector Field Visualization

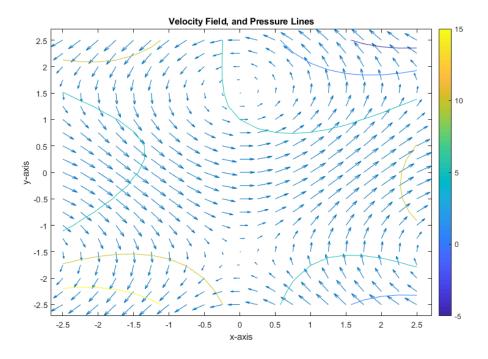


Figure 1: Visualization of a Velocity Field

2 Part B: 1-D PDE Tubular Chemical Reactor

The equation of conservation of chemical species under a chemical reaction of decomposition can be represented with the PDE given below.

$$\frac{\partial C}{\partial t} = \vec{\nabla} \cdot (D\vec{\nabla}C) - \vec{v} \cdot \vec{\nabla}C - kC^n$$

If a tubular catalytic chemical reactor initially filled with an inert solvent (C=0) is fed by a stream of component "A" with a concentration of $1kmol/m^3$ (C=1) and speed of 1m/s (v=1), calculate the distribution of "A" across the reactor and as a function of time C(x,t). The dispersion coefficient of the component "A" is $0.02m^2/s$ (D=0.001), the kinetic decomposition coefficient $0.05s^{-1}$ (k=1.5). The chemical decomposition kinetics is first order (n=1).

The molar balance in axial direction for a 1D flow can be written as:

$$\frac{\partial C}{\partial t} = D \frac{\partial^2 C}{\partial x^2} - v \frac{\partial C}{\partial x} - kC^n$$

The initial condition IC is:

$$C|_{t=0} = 0, 0 \le x \le 1$$

The boundary conditions BCs are:

$$C|_{x=0} = 1, t > 0$$

$$\left.\frac{\partial C}{\partial t}\right|_{x=L}=0,\,t\geq0$$

Where,

D is the diffusion coefficient C is the injection concentration v is the velocity of fluid injection k is the first order kinetic coefficient

L is the length of domain t is the simulation time x is the distance mesh

The PDE shall be transformed into a set of ordinary differential equations ODEs using central finite differences in space, as depicted in Equation 1.

$$\frac{\mathrm{d}C_i}{\mathrm{d}t} = D \frac{C_{i+1} - 2C_i + C_{i-1}}{(\Delta x)^2} - v \frac{C_{i+1} - C_{i-1}}{2\Delta x} - kC_i^n \tag{1}$$

```
1 %% Runge-Kutta
p(1) = 0.001; % Diffusion coefficient D
p(2) = 1.0; % Injection concentration c0
4p(3) = 1.5; % First order kinetic coefficient k
               % Velocity of fluid injection vo
_{5} p(4) = 1.0;
      = 2*640; % Number of nodes
_{7}p(5) = M;
8 Tspan = [0 1]; % Domain of time
      = linspace(0, 1, M);
11 % Initial conditions of the resulting set of ODEs
_{12} YO = zeros(M, 1);
_{13} YO(1) = 1.0;
15 % Solve differential equation (medium order method)
16 % use @reactub_2 for O(h^2) truncation error
17 % use @reactub_3 for O(h^3) truncation error
18 % use @reactub_4 for O(h^4) truncation error
19 OPTIONS
           = [];
20 [time_2, Y_2] = ode45(@reactub_2, Tspan, YO, OPTIONS, p);
21 [time_3, Y_3] = ode45(@reactub_3, Tspan, Y0, OPTIONS, p);
22 [time_4, Y_4] = ode45 (@reactub_4, Tspan, Y0, OPTIONS, p);
_{24}\,\% group all data / prepare to plot ...
25 time = {time_2, time_3, time_4};
          = \{ Y_2,
                        Y_3,
                                 Y_4};
_{27} \text{ Yprime} = \{ Y_2', Y_3', \}
                               Y_4'};
28 plotName = {'Oh2_truncationError', 'Oh3_truncationError', '
 Oh4_truncationError'};
```

Listing 2: Reactor: Runge-Kutta ODE solver

```
1 % Plot limits
2 noOf_curvesToPlot = 10;
                  = 0.02;
3 dlim
                   = [0 - dlim, 1 + dlim];
4 time_lim
                   = [0 - dlim, 1 + dlim];
5 Y_lim
                    = [0 - dlim, 1 + dlim];
6 xi_lim
7 Yprime_lim
                  = [0 - dlim, 1 + dlim];
9 for plotCount = 1:1:3
     % Display concentration vs. time
     totalNoOf_curves = size(Y{1, plotCount}, 2);
     noOf_curvesToSkip = fix(totalNoOf_curves/noOf_curvesToPlot);
figure;
```

```
subplot(1, 2, 1)
     for n = linspace(1, totalNoOf_curves, totalNoOf_curves)
          hold all
          if mod(n, noOf_curvesToSkip) == 0
              plot(time{1, plotCount}, Y{1, plotCount}(:, n));
          end
19
     end
20
     xlabel('time \tau');
21
     ylabel('Concentration mol/dm^3');
     axis([time_lim(1) time_lim(2) Y_lim(1) Y_lim(2)])
     \% Display concentration vs. distance
     totalNoOf_curves = size(Yprime{1, plotCount}, 2);
26
     noOf_curvesToSkip = fix(totalNoOf_curves/noOf_curvesToPlot);
     %figure;
     subplot(1, 2, 2)
     for n = linspace(1, totalNoOf_curves, totalNoOf_curves)
          hold all
          if mod(n, noOf_curvesToSkip) == 0
              plot(xi, Yprime{1, plotCount}(:, n));
          end
     end
35
     xlabel('distance x/L');
36
     ylabel('Concentration mol/dm^3');
     axis([xi_lim(1) xi_lim(2) Yprime_lim(1) Yprime_lim(2)])
```

Listing 3: Reactor: Plot the solutions

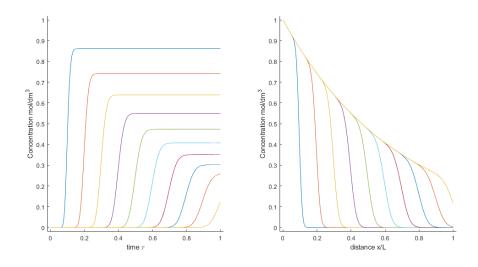


Figure 2: $O(h^2)$ Truncation Error

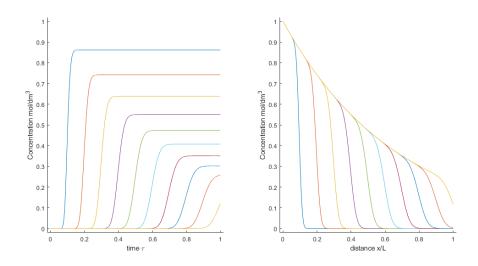


Figure 3: $O(h^3)$ Truncation Error

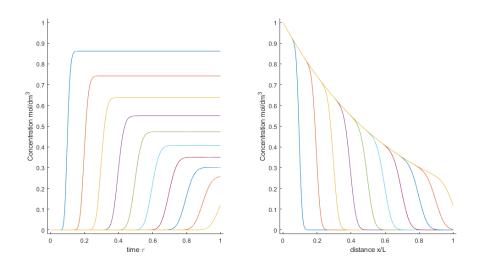


Figure 4: $O(h^4)$ Truncation Error

```
10 sprintf( ...

11 '(0(h^3) - 0(h^4))/0(h^3)*100 error: %f%%', ...

12 (Yprime{1, 2}(end) - Yprime{1, 3}(end))/Yprime{1, 2}(end)*100 ...

13)
```

Listing 4: Reactor : Compute percent error for each implementation

```
1 ans =
2          '(0(h^2) - 0(h^3))/0(h^2)*100 error: 1.908447%'
3 ans =
4          '(0(h^2) - 0(h^4))/0(h^2)*100 error: 1.919032%'
5 ans =
6          '(0(h^3) - 0(h^4))/0(h^3)*100 error: 0.010792%'
```

Listing 5: Reactor: Percent errors for each truncation

${\bf 3}\quad {\bf Part}\ {\bf C}: {\bf Growing\ Bubbles}$

```
1 % Runge-Kutta
```

```
_{2} % k = 1.4 adiabatic process, k = 1 isothermic
3 % alphaM = 0 inviscid
4 % betaM = 0 negligible surface tension
        = \{1.0, 1.4, 1.7\};
6 \text{ alphaM} = \{0.0, 0.5, 0.1\};
7 \text{ betaM} = \{0.0, 0.2, 0.7\};
9 % Solve and Plot
10 figure;
11 solveNplot_growingBubbles(k{1}, alphaM{1}, betaM{1}, sprintf('k = %1.1f', k
_{12} solveNplot_growingBubbles(k{2}, alphaM{1}, betaM{1}, sprintf('k = \%1.1f', k
     {2}))
13 solveNplot_growingBubbles(k{3}, alphaM{1}, betaM{1}, sprintf('k = %1.1f', k
14 suptitle(sprintf('Gas Molecule Shape and Size Effect : \\alpha = %1.1f and \\
     beta = %1.1f', alphaM{1}, betaM{1}))
16 figure;
17 solveNplot_growingBubbles(k{2}, alphaM{1}, betaM{1}, sprintf('\\alpha = %1.1f
     ', alphaM{1}))
18 solveNplot_growingBubbles(k{2}, alphaM{2}, betaM{1}, sprintf('\\alpha = %1.1f
     ', alphaM{2}))
{\tt 19} \ solveNplot\_growingBubbles(k\{2\}, \ alphaM\{3\}, \ betaM\{1\}, \ {\tt sprintf('}\alpha = \%1.1f)
     ', alphaM{3}))
20 suptitle(sprintf('Effect of the Viscosity : k = %1.1f and \beta = %1.1f', k
     {2}, betaM{1}))
22 figure;
23 solveNplot_growingBubbles(k{2}, alphaM{1}, betaM{1}, sprintf('\\beta = \%1.1f'
      , betaM{1}))
24 solveNplot_growingBubbles(k{2}, alphaM{1}, betaM{2}, sprintf('\\beta = %1.1f'
      , betaM{2}))
25 solveNplot_growingBubbles(k{2}, alphaM{1}, betaM{3}, sprintf('\\beta = %1.1f')
      , betaM{3}))
26 suptitle(sprintf('Effect of the Surface Tension : k = %1.1f and \\alpha =
     %1.1f', k{2}, alphaM{1}))
28 function[] = solveNplot_growingBubbles(k, alphaM, betaM, curveLabel)
      tspan = linspace(0, 35, 500);
           = 1;
      y 1
      у2
            = 0;
     yo = [y1, y2]';
      % Solve differential equation (medium order method)
      par(1) = k;
      par(2) = alphaM;
      par(3) = betaM;
      [t, Y] = ode45(@growingBubbles, tspan, yo, [], par);
      time = t;
      Yout = Y;
     %% Plot
42
     \% Display radius vs. time
43
     subplot(2,1,1);
```

```
hold all

plot(time, Yout(:,1), 'DisplayName', curveLabel);

xlabel('\tau');ylabel('R/Ro');

legend

Display d(radius) vs. time

subplot(2,1,2);

hold all

plot(time, Yout(:,2), 'DisplayName', curveLabel);

xlabel('\tau');ylabel('d(R/Ro) /d\tau');

legend

end
```

Listing 6: Growing Bubbles : Solve and Plot

Gas Molecule Shape and Size Effect : α = 0.0 and β = 0.0

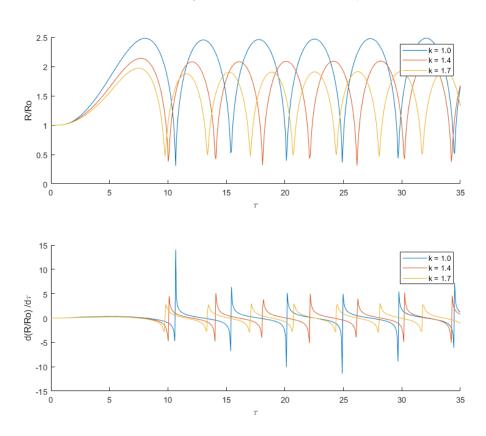


Figure 5: Gas molecule shape and size effect

4 Part C : Draining Tank

```
1 % Runge-Kutta
2 g = 9.81; % gravity
3 lambda = 10; % Area/Area0
4 L = 2; % length of the pipe
5
6 % Plot
7 figure;
8 solveNplot_drainingTank(g, lambda, L, '')
```

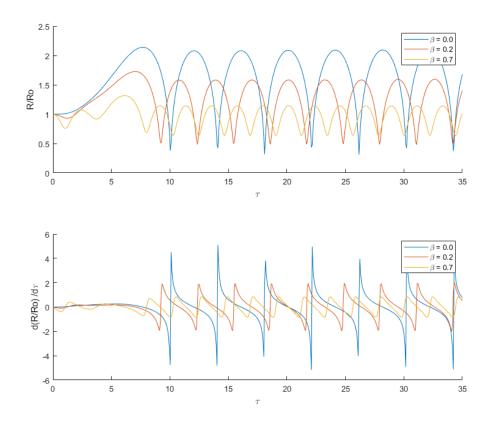


Figure 6: Effect of the surface tension

```
9 suptitle(sprintf('Draining Tank : g = %1.1f, A/A_{0} = %1.1f, and L = %1.1f',
      g, lambda, L))
10
11 function[] = solveNplot_drainingTank(g, lambda, L, curveLabel)
      tspan = linspace(0, 50, 500);
            = 1; % initial height (1m)
13
            = 0; % initial velocity (0m/s)
      у2
            = [y1, y2]';
     % Solve differential equation (medium order method)
     par(1) = g;
18
     par(2) = lambda;
19
      par(3) = L;
      [t, Y] = ode45(@drainingTank, tspan, yo, [], par);
21
22
      Yout
             = Y;
23
     %% Plot
     % Display radius vs. time
     hold all
      plot(time, Yout(:,1), 'DisplayName', curveLabel);
      xlabel('\tau');ylabel('z/z_o');
      %legend
30
31 end
```

Listing 7: Draining Tank: Solve and Plot

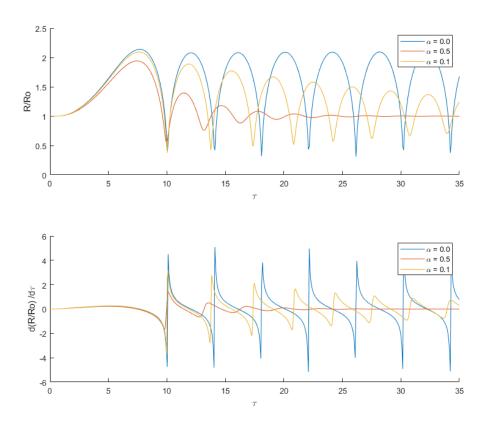


Figure 7: Effect of the viscosity

Draining Tank : g = 9.8, A/A $_0 = 10.0$, and L = 2.0

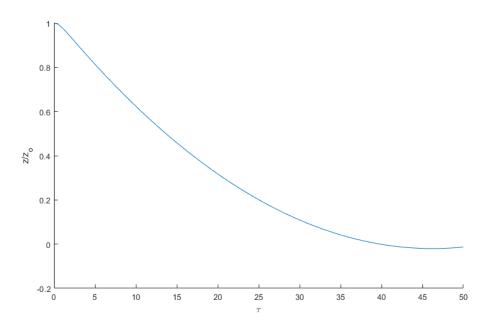


Figure 8: Draining Tank Height \boldsymbol{z}

5 Part D : Introduction to Fluid Kinematics (ANSYS-Fluent) [1]

References

[1] Model Wing - Simulation Example - ANSYS Innovation Courses. [Online]. Available: https://courses.ansys.com/index.php/courses/governing-equations-of-fluid-dynamics/(visited on 09/07/2020).