

# General Physics Midterm Review

Kongsak Tipakornrojanakit

1. A poorly designed electric device has two bolts attached to different parts of the device that almost touch each other in its interior, as shown. When they touch, a short circuit will develop and damage the device. If the initial gap between the ends of the bolts is  $5.0\text{ }\mu\text{m}$  at  $27^\circ\text{C}$ , at what temperature will the bolts touch?

- (a)  $\text{Steel}_{\text{length}} = 0.01\text{ m} = 1 \times 10^{-2}\text{ m}$   
(b)  $\text{Brass}_{\text{length}} = 0.03\text{ m} = 3 \times 10^{-2}\text{ m}$   
(c)  $\text{Gap}_{\text{length}} = 0.05\text{ }\mu\text{m} = 5 \times 10^{-6}\text{ m}$   
(d)  $\alpha_{\text{Steel}} = 11 \times 10^{-6}\text{ }^\circ\text{C}^{-1}$   
(e)  $\alpha_{\text{Brass}} = 19 \times 10^{-6}\text{ }^\circ\text{C}^{-1}$   
(f)  $\Delta L = \alpha L_0 \Delta T$   
(g)  $\Delta L = \text{Gap}_{\text{length}} = 5 \times 10^{-6}\text{ m}$

$$\begin{aligned}\Delta T &= \frac{\Delta L}{\alpha L_0} \\ &= \frac{\text{Gap}_{\text{length}}}{\alpha_{\text{Steel}} \text{Steel}_{\text{length}} + \alpha_{\text{Brass}} \text{Brass}_{\text{length}}} \text{ }^\circ\text{C} \\ &= \frac{5 \times 10^{-6}}{(11 \times 10^{-6} \times 1 \times 10^{-2}) + (19 \times 10^{-6} \times 3 \times 10^{-2})} \text{ }^\circ\text{C} \\ &= \frac{5 \times 10^{-6}}{(11 \times 10^{-8}) + (57 \times 10^{-8})} \text{ }^\circ\text{C} \\ &= \frac{5 \times 10^{-6}\text{ m}}{(68 \times 10^{-8})} \text{ }^\circ\text{C} \\ &= \frac{5}{(68 \times 10^{-2})} \text{ }^\circ\text{C} \\ &= \frac{5}{(68 \times 10^{-2})} \text{ }^\circ\text{C} \\ &= 7.3529 \text{ }^\circ\text{C}\end{aligned}$$

$$\begin{aligned}T_1 &= T_0 + \Delta T \text{ }^\circ\text{C} \\ &= 27 + 7.3529 \text{ }^\circ\text{C} \\ &= 34.3529 \text{ }^\circ\text{C}\end{aligned}$$

2. 500 g of ice is added to an insulated cup that contains 200 g of water at 50.0 °C. What is the final temperature.

(a)  $\text{Ice}_{\text{mass}} = 500 \text{ g} = 5 \times 10^{-1} \text{ kg}$

(b)  $\text{Water}_{\text{mass}} = 200 \text{ g} = 2 \times 10^{-1} \text{ kg}$

(c)  $\text{Ice}_{\text{temp}} = 0 \text{ }^{\circ}\text{C}$

(d)  $\text{Water}_{\text{temp}} = 50 \text{ }^{\circ}\text{C}$

(e)  $c_{\text{water}} = 4.186 \times 10^3 \frac{\text{J}}{\text{kg}} \text{ }^{\circ}\text{C}$

(f)  $L_{\text{ice}} = 3.33 \times 10^5 \frac{\text{J}}{\text{kg}}$

(g)  $Q_{\text{water}} = -Q_{\text{ice}}$

(h)  $Q = mc\Delta T$

(i)  $Q = mL$

$$Q_{\text{water}} = -Q_{\text{ice}}$$

$$\text{Water}_{\text{mass}} \times c_{\text{water}} \times (\text{Water}_{\text{temp}_1} - \text{Water}_{\text{temp}_0}) = -((L_{\text{ice}} \times \text{Ice}_{\text{mass}}) + (\text{Ice}_{\text{mass}} \times c_{\text{water}} \times (\text{Ice}_{\text{temp}_1} - \text{Ice}_{\text{temp}_0})))$$

$$2 \times 10^{-1} \times 4.186 \times 10^3 \times (T_1 - 50) = -(5 \times 10^{-1} \times 3.33 \times 10^5) - (5 \times 10^{-1} \times 4.186 \times 10^3 \times (T_1 - 0))$$

$$8.372 \times 10^2 \times (T_1 - 50) = -(16.5 \times 10^4) - (20.93 \times 10^2 \times (T_1 - 0))$$

$$8.372 \times (T_1 - 50) = -(16.5 \times 10^2) - (20.93 \times (T_1 - 0))$$

$$8.372T_1 - 418.6 = -(16.5 \times 10^2) - 20.93T_1$$

$$29.302T_1 - 418.6 = -(16.5 \times 10^2)$$

$$29.302T_1 = -16.5 \times 10^2 + 4.186 \times 10^2$$

$$29.302T_1 = -1.2314 \times 10^3$$