Evolutionary aspects of elemental hyperaccumulation

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Abstract Hyperaccumulation is the uptake of one or more metal/metalloids to concentrations greater than 50-100× those of the surrounding vegetation or 100-10,000 mg/kg dry weight depending on the element. Hyperaccumulation has been documented in at least 515 taxa of angiosperms. By mapping the occurrence of hyperaccumulators onto the angiosperm phylogeny, we show hyperaccumulation has had multiple origins across the angiosperms. Even within a given order, family or genus, there are typically multiple origins of hyperaccumulation, either for the same or different elements. We address which selective pressures may have led to the evolution of hyperaccumulation and whether there is evidence for co-evolution with ecological partners. Considerable evidence supports the elementaldefense hypothesis, which states that hyperaccumulated elements protect the plants from herbivores and pathogens. There is also evidence that hyperaccumulation can result in drought stress protection, allelopathic effects or physiological benefits. In many instances, ecological partners of hyperaccumulators have evolved resistance to the hyperaccumulated element, indicating co-evolution. Studies on the molecular evolution of hyperaccumulation have pinpointed gene duplication as a common cause of increased metal transporter abundance. Hypertolerance to the hyperaccumulated element often relies upon chelating agents, such as organic acids (e.g., malate, citrate) or peptide/protein chelators that can facilitate transport and sequestration. We

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conclude the review with a summary and suggested future directions for hyperaccumulator research.

Introduction

Over five hundred species of plants have been documented to hyperaccumulate one or more trace metal/metalloids, including arsenic (As), cadmium (Cd), chromium (Cr), cobalt (Co), copper (Cu), lead (Pb), manganese (Mn), nickel (Ni), selenium (Se) and zinc (Zn). Most hyperaccumulators are flowering plants, but a few fern species have been documented to hyperaccumulate As (Reeves and Baker 2000; Ma et al. 2001). In this review, we focus on angiosperm hyperaccumulators.

Plant hyperaccumulators are defined as having 50–100× higher concentrations of the element relative to that of the surrounding vegetation, or equivalent to 100-10,000 mg/kg dry weight (DW; Maestri et al. 2010; for a recent review on hyperaccumulation see also Van der Ent et al. 2013). Hyperaccumulators are predominantly found on soils that contain elevated levels of the hyperaccumulated element, which suggests that they have evolved in situ on these soils. Importantly, these soils are not just naturally occurring soils, but also anthropogenically contaminated soil, which may indicate that hyperaccumulation can evolve rapidly. Alternatively, the hyperaccumulators may just have colonized the mine tailings from an original outcropping of surface-facing ore. Given the immobility of plants it is understandable that taxa inhabiting soils with high levels of toxic elements have evolved tolerance mechanisms. Indeed, hyperaccumulators are tolerant to the extreme soils they grow on. However, it is intriguing that hyperaccumulators appear to actively concentrate the toxic element(s). Hyperaccumulation is not necessary for survival on soils



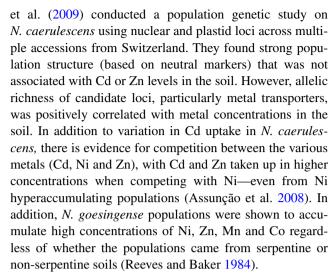
with elevated levels of the element, but rather requires extra evolutionary steps for hypertolerance to the high levels of the toxic metal/metalloid in their tissues. This review will address the following four evolutionary questions. First, how many times did hyperaccumulation evolve? Second, what selection pressures have driven the evolution of hyperaccumulation? Third, is there co-evolution with the ecological partners of hyperaccumulating plants? Fourth, what do we know about the molecular evolution of hyperaccumulation?

How many times did hyperaccumulation evolve?

To understand the evolutionary basis of hyperaccumulation we need to know how hyperaccumulation is distributed across the angiosperms. There have been reports of over 500 hyperaccumulators among the ~250,000 angiosperm species (Stevens 2001 onwards), equating to ~0.2 % of species (Suppl. Table 1). The majority of these 500 species are Ni hyperaccumulators. At the genus level, 181/13,000 genera = 1.4 % hyperaccumulate and 60/443families = 13.5 % contain hyperaccumulators. The following angiosperm families contain an unusually high percentage of the total hyperaccumulating taxa (Fig. 1): Brassicaceae, Euphorbiaceae and Asteraceae. Hyperaccumulation, even of a single element, appears to have evolved several times within vascular plants: it is distributed widely across the phylogeny. At multiple taxonomic levels (order, family, genus), we find several origins of hyperaccumulation. For instance, Krämer (2010) proposes that within the Brassicaceae there are at least six independent origins of Ni hyperaccumulation, three origins of Zn and Cd hyperaccumulation, two origins of As hyperaccumulation and one origin of Se hyperaccumulation. Alyssum (Brassicaceae) contains upward of 50 nickel hyperaccumulating species without an obvious pattern of origin either at the phylogenetic or geographic level (Cecchi et al. 2010). Given the high number of hyperaccumulating taxa and the general lack of phylogenetic or geographic correlation of Ni hyperaccumulators, there may be a predisposition within the Alysseae to adapt to serpentine soils high in Ni.

Natural variation within species

Probably, the two most studied hyperaccumulator species are *Arabidopsis halleri* and *Noccaea caerulescens* (formerly, *Thlaspi caerulescens*) (both in the Brassicaceae) which hyperaccumulate Zn and Cd, and Zn, Ni and Cd, respectively. There is evidence that Zn hyperaccumulation is a constitutive trait in *A. halleri*, but that Cd hyperaccumulation is variable (Bert et al. 2002). A parallel situation exists in *N. caerulescens* (Lombi et al. 2000). Besnard



In contrast to hyperaccumulation being a constitutive trait, some species show strong ecotypic variation. Populations of *Stanleya pinnata* have been shown to vary in Se concentration depending on the population tested; these differences appear to be fixed based on common garden experiments (Feist and Parker 2001). Substantial population-level variation was also found for Ni accumulation in the Ni hyperaccumulator *Alyssum bertolonii* (Galardi et al. 2007). The Cd hyperaccumulator *Sedum alfredii* also has fixed ecotypic differences including differences in Cd sequestration (Tian et al. 2011) and translocation (Lu et al. 2008).

Phylogenetic conclusions

In conclusion, hyperaccumulation is broadly distributed over the plant phylogeny and must have evolved independently many times. Even within a family or genus there is rarely unequivocal evidence for one origin of hyperaccumulation when more than one species is a hyperaccumulator. If we take an even closer look at the species level, populations differ in their ability to accumulate and tolerate certain elements, and within a population individuals commonly differ in their tolerance and accumulation and these differences can be dependent on the element accumulated, e.g., Zn vs. Cd. Thus, hyperaccumulation is a constantly evolving trait and under selection.

Which selection pressures favored the evolution of hyperaccumulation?

Several hypotheses have been proposed for the functional significance of metal and metalloid hyperaccumulation by plants (Boyd and Martens 1992). There is considerable evidence that accumulation of toxic elements can offer plants protection from herbivores and pathogens, which is



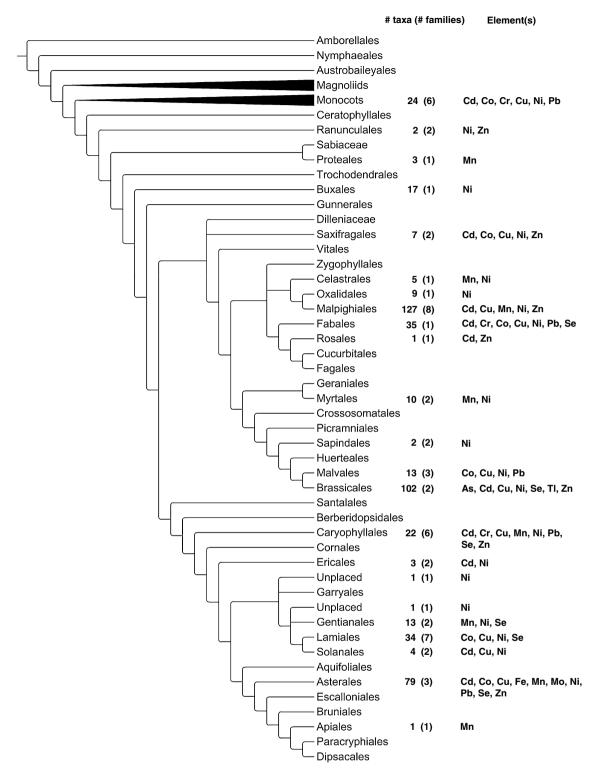


Fig. 1 Occurrence of elemental hyperaccumulation in angiosperms. Phylogenetic tree modified from angiosperm phylogeny group (APG) III (Stevens 2001). For more detailed information about which taxa hyperaccumulate each element, see Supplemental Table 1

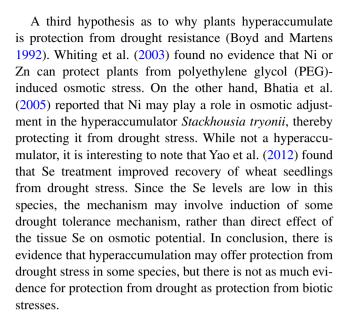
in agreement with the elemental-defense hypothesis (Boyd and Martens 1992). The protective effect against herbivores includes both deterrence and toxicity. High levels of Ni

in plants have been shown to protect them from a variety of herbivores and pathogens (Boyd et al. 1994; Boyd and Moar 1999; Martens and Boyd 2002). Zinc and Cd can



also protect plants from invertebrate herbivory (Pollard and Bakers 1997; Jhee et al. 1999). Similarly, Se can protect plants from a variety of herbivores and pathogens, including moth and butterfly larvae, aphids, thrips, spider mites, grasshoppers, prairie dogs, as well as fungal pathogens (Vickerman and Trumble 1999; Hanson et al. 2004; Freeman et al. 2006b, 2007, 2009; Quinn et al. 2008, 2010a, b). Arsenic has also been shown to protect plants from herbivores (Mathews et al. 2009). In further support of a defensive role, it appears that the hyperaccumulated element is often concentrated in organs and tissues that are most susceptible to herbivores and pathogens (Freeman et al. 2006b; Tappero 2008). The levels at which elements become protective against herbivores or pathogens depend on the element and chemical form the plant accumulates, as well as the sensitivity of the herbivore or pathogen in question. For example, Se in plants is effective against aphids below 10 mg kg⁻¹ DW (Hanson et al. 2004), whereas plants needed an order of magnitude more Se to be protected from Lepidoptera larvae or prairie dogs, and even higher levels to be protected from spider mites (Freeman et al. 2006b, 2009; Quinn et al. 2010a, b). Different elements may also show synergistic interactions in vivo with each other and with organic defense compounds, reducing the level at which they become effective. This phenomenon has been termed the joint-effects hypothesis by Boyd (2012). These examples provide strong evidence that herbivores and pathogens can act as selection pressure for the evolution of increasing elemental accumulation in plants. This selection pressure can be effective at tissue elemental concentrations well below hyperaccumulator level and is expected to be incremental: the higher the elemental tissue concentration, the more effective and the broader is the protection from a range of pests.

There is also some evidence that hyperaccumulators benefit from their accumulated element via allelopathy, i.e., negative effects on neighboring plants, in support of another hypothesis put forward by Boyd and Martens (1992). Soil around Se hyperaccumulators Astragalus bisulcatus and Stanleya pinnata was found to be significantly enriched with Se, and this soil significantly reduced germination and growth of the Se-sensitive plant species Arabidopsis thaliana and enhanced its Se levels compared to soil collected from non-hyperaccumulators (El Mehdawi et al. 2011a). No clear allelopathic effects were found for the Ni hyperaccumulator Alyssum murale (Zhang et al. 2007). While Ni-amended soil reduced seed germination of eight herbaceous species, leaf litter from A. murale did not, perhaps due to low Ni phytoavailability in leaf litter. More studies are needed, but it appears that for some hyperaccumulators allelopathic benefits may have contributed to the evolution of hyperaccumulation.



In some cases, enhanced accumulation of elements may have a positive physiological effect. This may be true for both essential and non-essential elements. For instance, the hyperaccumulator *Noccaea caerulescens* had a 75 % increase in biomass when supplied with Cd after 31 days (Roosens et al. 2003). The element Se is not known to be essential for plants, but is beneficial for the growth of many plant species, especially Se hyperaccumulators (Virupaksha and Shrift 1965; El Mehdawi et al. 2012).

It is also possible that hyperaccumulation resulted from inadvertent uptake if the accumulated elements are chemically similar to essential nutrients and plant populations evolve more efficient uptake and translocation systems for these nutrients. For instance, Boyd and Martens (1998) suggested that Ni hyperaccumulation in *N. montanum* may be an inadvertent consequence of an efficient nutrient uptake system for another nutrient, perhaps for Zn or Ca. Similarly, it is feasible that hyperaccumulation of As, Cd and Se evolved as inadvertent side effects of enhanced phosphate, zinc/calcium and sulfate uptake, respectively.

Thus, several biotic and abiotic environmental factors as well as physiological processes may contribute to the evolution of elemental hyperaccumulation in a lineage. These various selection pressures may act simultaneously or individually, depending on the habitat, element and taxon.

Is there co-evolution of hyperaccumulators with ecological partners?

Plants have many ecological partners, including microbes, pollinators, herbivores and other plants. Elemental hyperaccumulation has been shown to negatively affect these partners if they are sensitive to the element in question, as described above. Thus, the hyperaccumulator may exert



selection pressure on its ecological partners, leading to their partners having increasing resistance to the toxic element over time. Such resistant ecological partners can then exclusively occupy the new niches offered by hyperaccumulator plants. There is indeed evidence that ecological partners of hyperaccumulators have evolved resistance to the hyperaccumulated element. For example, Boyd (2009) reported insects that feed almost exclusively on Ni hyperaccumulators. He identified 15 insect species across three (sub)orders that can tolerate an internal Ni concentration of $>500 \text{ mg/kg}^{-1}$. A few of these species appeared to be specialists, since they chose to feed on a particular Ni hyperaccumulator even when given the choice of a different Ni hyperaccumulator or a non-hyperaccumulator. Similarly, several Se-resistant herbivorous moths, seed beetles and seed wasps were found to feed on Se hyperaccumulators A. bisulcatus and S. pinnata in their natural seleniferous habitat (Freeman et al. 2006b, 2012; Valdez-Barillas et al. 2012). These situations appear to be classical examples of co-evolution, where plant and herbivore are in an "arms race", leading to ever increasing accumulation in the plant and resistance in the herbivore. There is also evidence of Se-tolerant microbial and micro-athropod detrivores in seleniferous ecosystems that efficiently decompose hyperaccumulator litter (Quinn et al. 2011a, b), as well as evidence of Se-tolerant endophytes and leaf pathogens that successfully occupy the niche provided by Se hyperaccumulators (Valdez-Barillas et al. 2012). Also, some Se-tolerant plant species were shown to benefit from close proximity to Se hyperaccumulators: they contained up to tenfold higher Se levels and enjoyed less herbivory and enhanced growth (El Mehdawi et al. 2011b). As a result of their negative effects on sensitive ecological partners and positive effects on resistant ecological partners, hyperaccumulators may affect species composition in the community at multiple trophic levels. Hyperaccumulators may also affect cycling of the hyperaccumulated element through their local ecosystem, by affecting its soil distribution and chemical form, and forming a portal for its entry into higher trophic levels (El Mehdawi and Pilon-Smits 2012).

How did hyperaccumulation evolve, at the molecular level?

Hyperaccumulators have evolved both enhanced capacity to sequester toxic elements and enhanced tolerance to these elements. The underlying mechanisms may be hypothesized to include mutations in transporter protein genes that enhance expression levels or change the kinetic properties of the transporter. Hyperaccumulators may be expected to have higher expression levels of

various membrane transporters responsible for uptake into the root symplast, loading into the root xylem, uptake into the leaf symplast and sequestration into the vacuole and in specific tissues such as the epidermis. To facilitate transport of the toxic elements in xylem and phloem, hyperaccumulators may also be expected to have enhanced levels of chelating agents in either of these vascular tissues or in adjacent tissues. Such chelating agents may include organic acids (e.g., malate, citrate) or peptide/protein chelators (e.g., nicotianamine, glutathione, phytochelatins, histidine). Some of these same chelators may help the plants tolerate their extraordinarily high levels of these toxic elements. Hyperaccumulators may have enhanced levels of (other) antioxidant compounds and enzymes as additional tolerance mechanisms. In the case of Se, hyperaccumulators appear to have evolved mechanisms by which this element is converted to less toxic, organic forms (Neuhierl and Böck 1996).

From studies to date on different hyperaccumulator species, there appear to be several different cases by which hyperaccumulators have modified transporter abundance or activity. Some hyperaccumulators have increased the gene copy number of transporters via gene duplication events; examples include the HMA (heavy metal binding) transporter in A. halleri (Hanikenne et al. 2008) and N. caerulescens (Ueno et al. 2001; Craciun et al. 2012). In A. halleri, Hanikenne et al. (2008) sequenced bacterial artificial chromosomes and found three tandem copies of metal ATPase 4 (HMA4) relative to A. thaliana. In N. caerulescens, HMA4 was again shown to have a variable copy number in populations that differ in Cd accumulation and tolerance. The population with the least efficient Cd translocation had the lowest expression level and the lowest copy number (Craciun et al. 2012). There can be up to four tandem copies of *HMA4* in *N. caerulescens* (Ó Lochlainn et al. 2011). In addition to HMA4, heavy metal ATPase 3 (HMA3) was implicated as conferring tolerance by being a tonoplastlocated transporter sequestering Cd into the leaf vacuoles. When higher and lower Cd accumulating ecotypes of N. caerulescens populations were compared, the authors found significantly higher HMA3 expression in the more tolerant ecotype Ganges, which also had a greater number of gene copies (Ueno et al. 2001).

In hyperaccumulators of non-essential elements it appears that the hyperaccumulated element enters the plant via transporters for other, essential elements that are chemically similar to it. For instance, the non-essential elements Se, As and Cd are chemically similar to S, P and Zn, respectively, and enter the plant through the transporters for these elements. Sulfate transporter (Sultr)1;2 is apparently responsible for selenate import into the root in *Arabidopsis thaliana*, since lesions in *sultr1*;2 confer selenate resistance and increased tolerance in *A. thaliana* by reducing



symplastic accumulation of Se (Shibagaki et al. 2002; Ohno et al. 2012). Several sulfate transporters were found to be more highly expressed in hyperaccumulator Astragalus spp. at levels comparable to those in non-hyperaccumulator plants present under S starvation (Cabannes et al. 2011). Similarly, several genes encoding sulfate transporters are constitutively upregulated in the Se hyperaccumulator S. pinnata relative to the non-hyperaccumulator S. albescens (Freeman et al. 2010). Thus, constitutive expression of sulfate transporters may be one of the mechanisms of Se hyperaccumulation. In addition, there is evidence of at least one Se-specific transporter in the hyperaccumulator S. pinnata: in contrast to non-hyperaccumulator species, selenate uptake in S. pinnata is not significantly inhibited by high sulfate levels (Harris and Pilon-Smits, unpublished results). Similarly, it has been shown that arsenate most likely is taken up via a phosphate transporter; for a review of arsenic uptake and metabolism in plants see Zhao et al. (2009).

The ZIP family (ZRT, IRT-like proteins) transporters are known to be involved in the movement of Fe, Zn, Mn and Cd, with 15 paralogs identified in A. thaliana. In Zn/Cd hyperaccumulators A. halleri and N. caerulescens, ZIP transporters have been shown to have a constitutively higher expression level relative to A. thaliana and Thlaspi arvense, respectively (Assunção et al. 2001; Becher et al. 2004; Weber et al. 2004). Furthermore, expression of two copies of the ZIP transporter gene IRT3, AtIRT3 and AhIRT3 were able to rescue a Zn/Fe uptake-deficient Saccharomyces cerevisiae mutant (Lin et al. 2009). This work suggests that root transporters in the ZIP gene family are important factors in Zn hyperaccumulation. Beyond transporters, many metals/metalloids are complexed with chelators to allow for specific uptake, movement and sequestration within the different compartments of the plant body.

Many peptide/protein chelators have been implicated in providing a mechanism for (hyper)tolerance, including glutathione, histidine, nicotianamine and phytochelatins. Glutathione (GSH) may contribute to tolerance not only by binding toxic elements, but also by scavenging free radicals. Levels of GSH were found to be constitutively enhanced in Ni hyperaccumulator Noccaea goesingense (Freeman et al. 2004), as well as in Se hyperaccumulator Stanleya pinnata (Freeman et al. 2010). In Phytolacca americana, a Cd hyperaccumulator, the tissue concentration of GSH was shown to increase when supplied with Cd (Zhao et al. 2011). Histidine has been shown to be an important factor in Ni hyperaccumulators Alyssum lesbiacum (Krämer et al. 1996) and N. goesingense (Krämer et al. 2000). Histidine has also been shown to be important in root-to-shoot translocation in *N. caerulescens* by preventing the accumulation of histidine-bound Ni in the root vacuoles (Richau et al. 2009). Nicotianamine is present at elevated levels in the hyperaccumulator species, A. halleri and N. caerulescens relative to closely related non-hyperaccumulator species (Deinlein et al. 2012). Furthermore, in A. halleri an RNAi knockout of nicotianamine synthase resulted in decreased translocation of Zn from the root to and shoot, and the knockout plants did not reach hyperaccumulator levels of Zn in the shoots (Deinlein et al. 2012). Lastly, phytochelatin levels have been to shown to significantly increase in P. americana with increasing Cd supply, potentially to help sequester Cd ions in the leaf vacuoles (Gao et al. 2013). However, phytochelatin concentrations are typically lower in hyperaccumulator species than related non-hyperaccumulator species (Meyer et al. 2011) and therefore may not actually be an evolutionary mechanism in hyperaccumulator species. In addition to peptide/protein chelators, organic acids such as citrate and malate have also been implicated in tolerance and transport in hyperaccumulator species. It has been shown that citrate and malate bind to Zn and Ni to allow for loading and unloading into and out of xylem, phloem and across the tonoplast. In Sedum alfredii, Lu et al. (2013) showed increased levels of citrate in the xylem sap of plants from a hyperaccumulating population relative to a non-hyperaccumulating population and that the levels of citrate increased with increasing Zn concentration. Nickel hyperaccumulator Alyssum murale has increased citric acid and malic acid synthesis in the root mitochondria relative to non-hyperaccumulator A. montanum (Agrawal et al. 2013). In N. goesingense Ni has been shown to be bound to citrate (Krämer et al. 2000). In N. caerulescens Zn has been found to be stored chelated by malate and citrate in the vacuoles of epidermal cells. Elevated levels of these organic acids may help in Zn sequestration and tolerance of Zn in *N. caerulescens* (Schneider et al. 2013).

In addition to chelation, hyperaccumulators have in some cases evolved mechanisms by which elements are metabolized to less toxic forms. This is the case for Se. All plants can assimilate selenate to selenocysteine (SeCys) via the sulfate assimilation pathway. This SeCys may be non-specifically incorporated into protein, which is toxic. SeCys may also be further metabolized via selenocystathionine (SeCyst) to selenomethionine (SeMet) and other analogs of reduced S compounds. Plants can also form volatile dimethylselenide (DMSe) from SeMet (Terry et al. 2000). A transcriptome study showed that in the hyperaccumulator Stanleya pinnata many genes from the S assimilation pathway were upregulated, as compared to non-hyperaccumulator S. albescens, perhaps due to higher levels of the plant-growth regulators, jasmonate, salicylate and ethylene (Freeman et al. 2010). This may explain the hyperaccumulator's higher levels of Se and S. The forms of Se in hyperaccumulators and non-hyperaccumulators are also different: while many non-hyperaccumulators accumulate predominantly selenate when treated with selenate, hyperaccumulators tend to accumulate organic Se. In



S. pinnata, for instance, a large fraction of the Se is present as SeCyst (Shrift and Virupaksha 1965; Freeman et al. 2006b). These findings suggest that certain enzymes within the S assimilation pathway may be differentially expressed in hyperaccumulators and non-hyperaccumulators. In addition, Se hyperaccumulators often contain methyl-SeCys, which is produced by an Se-specific SeCys methyltransferase (Neuhierl and Böck 1996; Sors et al. 2005; Freeman et al. 2006b). The accumulation of these non-protein amino acids, SeCyst and methyl-SeCys, prevent incorporation into protein, and thereby prevent toxicity to the plant (Brown and Shrift 1981; Neuhierl and Böck 1996). Hyperaccumulators may also volatilize Se at very high rates in the form of dimethyldiselenide (DMDSe), which is synthesized from methyl-SeCys (Freeman and Bañuelos 2011). Therefore, Se hyperaccumulators in the genera Stanleya and Astragalus appear to have evolved a unique pathway that plays an important role in their Se tolerance.

Summary and future directions

Many different hyperaccumulators have been studied and from these studies several mechanisms responsible for hyperaccumulation have been identified. Often, hyperaccumulators have enhanced levels of transporters (as a result of gene duplication) for uptake into the root and translocation within the plant. Hypertolerance mechanisms that have been identified include enhanced levels of chelators or of enzymes that convert the element to less toxic forms. Some interesting questions that may still be addressed include: Is there a "master switch" gene (e.g., some transcription factor) that upregulates all genes involved in hyperaccumulation and hypertolerance? Did hypertolerance evolve simultaneously with hyperaccumulation or did it evolve sequentially with hyperaccumulation? If the latter case is true, then which came first? Does one commonly precede the other, and is one trait more difficult to evolve?

Studies on elemental hyperaccumulation so far generally have used relatively few approaches: molecular, physiological, genetic or ecological. The next challenge will be to integrate these diverse approaches within a selected taxon. Such a comprehensive study may start with a complete documentation of the variation in tolerance and accumulation in a group of species/populations, in concert with phylogenetics/population genetics of the taxa of interest. This may be followed up by a comparative study of selected taxa to obtain insight into underlying biochemical and genetic mechanisms, hopefully leading to identification of candidate alleles that contribute to hyperaccumulation. In parallel, the ecological effects of hyperaccumulation may be studied within natural ecosystems to obtain insight into selection pressures. Finally, candidate alleles may be

studied in natural systems for evidence of selection, and at the molecular level for evidence of positive selection.

The results from such studies have intrinsic scientific value. Adaptation to selective pressures on populations from biotic or abiotic factors is a key process to study in evolutionary biology, because it helps explain how interspecific differences arose. Rigorously studying local adaptation requires researchers to address questions at multiple organization levels of biology, because only by synthesizing information from disciplines such as molecular biology, cell biology, physiology, ecology and phylogenetics can we thoroughly document local adaptation by linking cellular responses to the relationships of species over time and correlate those with the ecological landscape. Hyperaccumulation offers a very promising model trait in this respect.

Beyond basic scientific value, results from such studies also may have broad applications in biofortification and phytoremediation. Identification of hyperaccumulation mechanisms and specific alleles allows for development of plants with enhanced tolerance to, and accumulation of, both nutrients and environmental pollutants.

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