



Friedrich-Alexander-Universität
Faculty of Engineering

Department Materials Science

WW8: Materials Simulation

Practical: Phase-Field Method

Basics and Application in Materials Science

Leon Pyka (22030137)

Supervision:

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1 Introduction

Phase-field simulation is a versatile application in the toolbox of materials simulation. It is often used for simulations of phase transitions, dislocation evolution, fracture simulations etc. The following practicals aim is to get a practical introduction into the subject. In two separate tasks a 1D single-component solidification simulation, calculation of a bulk energy density coefficient and gradient energy density coefficient are going to be conducted.

2 Task 1: 1D Single-Component Solidification

To simulate a 1D single-component solidification we utilize the following energy density:

$$F = \int (f_0 \phi^2 (1 - \phi)^2 + \frac{K_\phi}{2} |\nabla \phi|^2) d\vec{r} \quad (1)$$

As solidification is a non-conservative process, the kinetics are governed by the Allen-Cahn equation (for the homogeneous, isotropic case).

$$\frac{\partial \phi}{\partial t} = -L \frac{\delta F}{\delta \phi} \quad (2)$$

Here the functional derivative $\frac{\delta F}{\delta \phi}$ of an energy functional eq. 3a can be evaluated as eq. 3b.

$$F = \int f(\vec{r}, \phi, \nabla \phi) d\vec{r} \quad (3a)$$

$$\frac{\delta F}{\delta \phi} = \frac{\partial f}{\partial \phi} - \nabla \cdot \frac{\partial f}{\partial (\nabla \phi)} \quad (3b)$$

Solving the functional derivative from eq. 2 yields:

$$\frac{\delta F}{\delta \phi} = 2f_0 \phi (1 - \phi)^2 + 2f_0 \phi^2 (\phi - 1) - K_\phi \nabla^2 \phi \Leftrightarrow \quad (4a)$$

$$\Leftrightarrow 2\phi^3 - 4\phi^2 + 2\phi \quad (4b)$$

Inserting the result into eq. 2 results in:

$$\frac{\partial \phi}{\partial t} = -L[f_0(2\phi^3 - 4\phi^2 + 2\phi) - K_\phi \nabla^2 \phi] \quad (5)$$

As the governing equation for solving the PDE we apply Dirichlet boundary conditions $\phi_1 = 0; \phi_n = 1$ and discretize the partial derivative $\frac{\partial \phi}{\partial t}$ as:

$$\frac{\partial \phi}{\partial t} \approx \frac{\phi_{next} - \phi}{\Delta t} = -L[f_0(2\phi^3 - 4\phi^2 + 2\phi) - K_\phi \nabla^2 \phi] \Leftrightarrow \quad (6a)$$

$$\Leftrightarrow \phi_{next} = -\Delta t L[f_0(2\phi^3 - 4\phi^2 + 2\phi) - K_\phi \nabla^2 \phi] \quad (6b)$$

and $\nabla^2 \phi(x)$ is discretized according to the (forward) finite differences scheme:

$$\nabla^2 \phi(x_i) \approx \frac{\phi(x_{i+2}) - 2\phi(x_{i+1}) + \phi(x_i)}{\Delta x} \quad (7)$$

where x_i is the node at which the value is computed and Δx is the node distance. The initial result is plotted in fig. 2. As can be seen in 2 f_0 seems to be reciprocally proportional energy functional over time. With decreasing value the transition region (where ϕ changes from 0 to 1) becomes broader. The opposite effect can be observed in fig. 2 with variation of K_ϕ . Additional the energy density functional seems to become broader but only changes its peak value for increased K-values. L does not seem to have any influence on the ϕ , the energy or energy density - see fig. 2 .

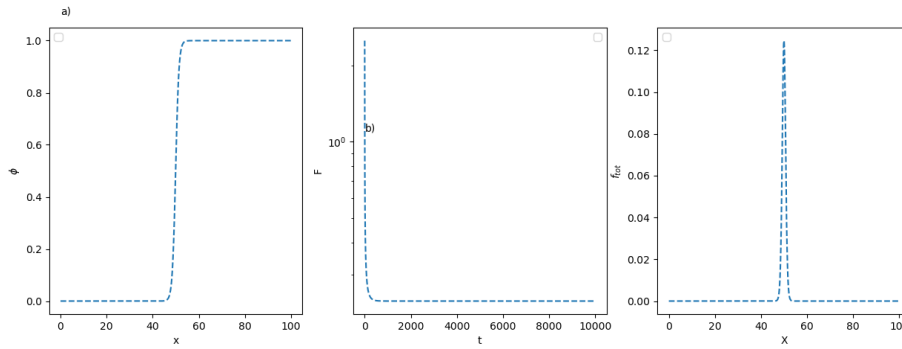


Figure 1: Initial results of the 1D single-component solidification simulation.

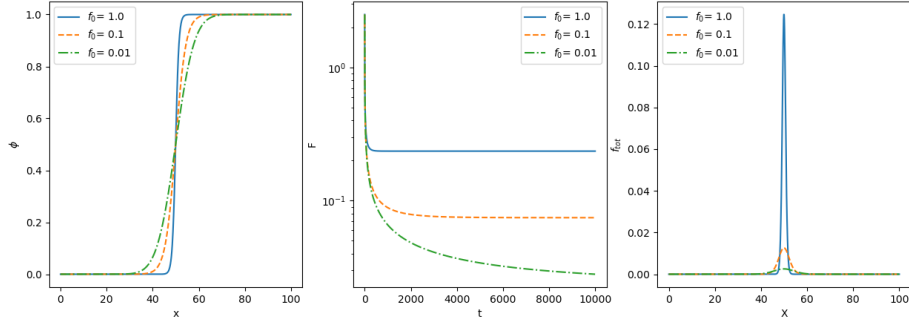


Figure 2: Results of the 1D single-component solidification simulation with various f_0 values.

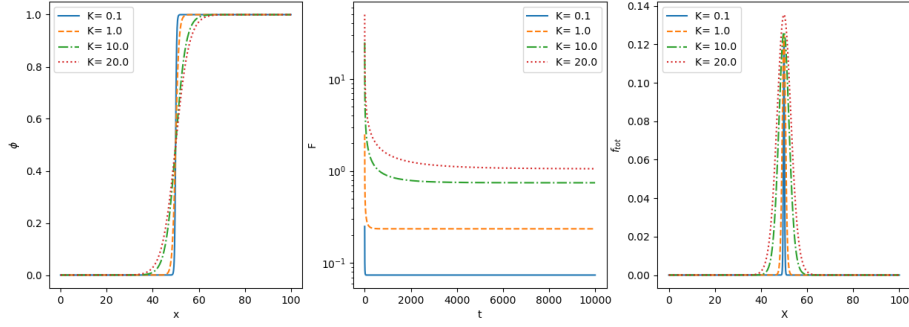


Figure 3: Results of the 1D single-component solidification simulation with various K values.

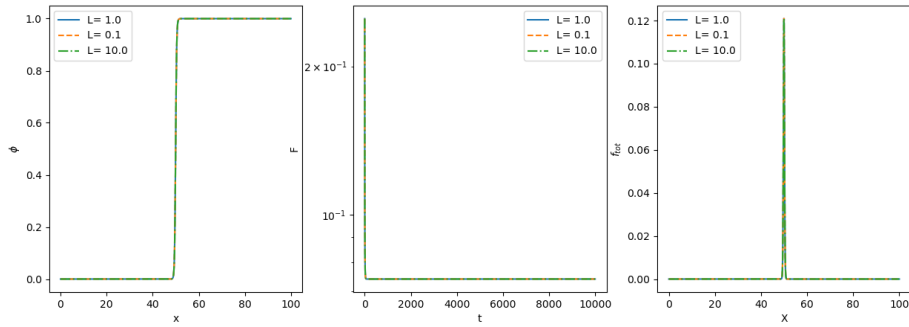


Figure 4: Results of the 1D single-component solidification simulation with various L values.

3 Task 2

The equilibrium Al concentration of the disordered γ -Phase has been determined to be 0.165 and the Al concentration of the ordered γ' -Phase has been determined to be 0.230 [1].

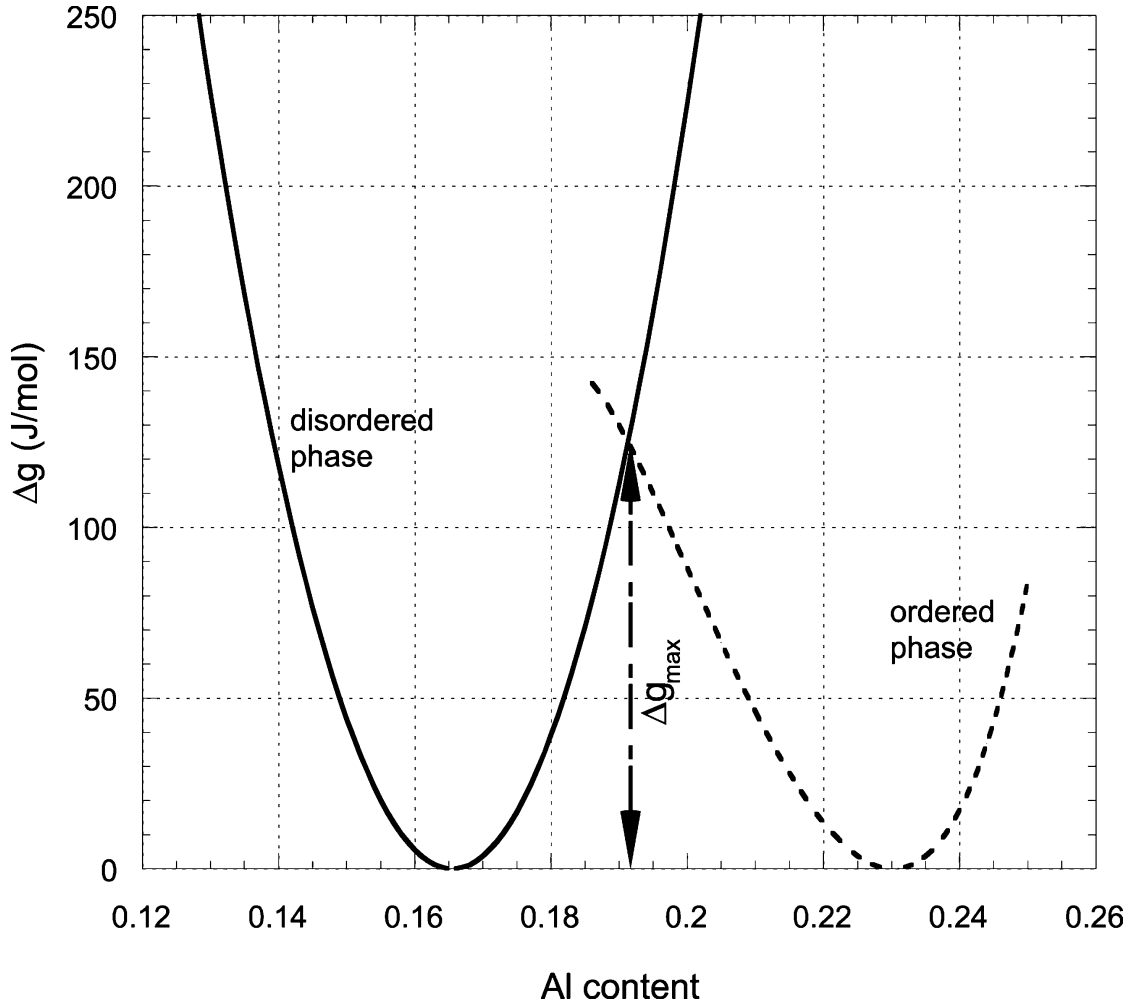


Figure 5: Calculated chemical free energy as a function of composition for both the disordered and ordered phase at 1300 K reproduced from [1].

References

1. Zhu, J. Z., Liu, Z. K., Vaithyanathan, V. & Chen, L. Q. Linking Phase-Field Model to CALPHAD: Application to Precipitate Shape Evolution in Ni-base Alloys. *Scripta Materialia* **46**, 401–406. ISSN: 1359-6462. <https://www.sciencedirect.com/science/article/pii/S1359646202000131> (2024) (Mar. 11, 2002).