	A COMPANY TO SELECT THE SELECT THE PROPERTY OF
2)	CO + C/2 -> COC/2 Pp(mol/min) = 8.75 Cco Cc12
	(1+58. 6 Cy2 +34, 3 Croces)
	let A = CO A + B > C A4 7 = 303.8 2 W/ 19 charcoal
	B = C/2
	$C = COCl_2$
The second second	a) V= 3L 40%, mul C/2
	T= 303.8K 60/. NOI CO
	P= latm
VANTOR	
	PV=NP=
->	(1)(3) = n(0.0821)(303.8 2) (n)(0.4) = 0.0723 mol (12 (B)
	N = 0.12043 mol (total) (n) (0.61 = 0.0487 mol (0) (A)
Christian and Christian Annual Con-	
	PA = XAP Paz = 0.4 atm
	Pco = 0.5 atm
All the state of t	
\rightarrow	Intal conditions / concentration
	AV-URT
	012 -> (0.4)(1) = (10/20) (0.0821) (303.8) 7 initial
	[CU20 = 0,01605 moly Concentrations
	(U + (0.6) (1) = (0. (0.0821) (303.8)
	1CO, = 0.02407 no1/c
14 W 21	
777	(Cco = 0.02407 - Cp(6-)
	Caz = 0.01605 - cp(6)
2 2 3	
60 de la companya del companya de la companya de la companya del companya de la companya del la companya de la	

6)	P4 = 8.75 Cio Ce/2	ply in value
	(1+58,60 q2 +34.3 (cour)	from Fix a)
1et V=3	d Ceogy V	The second secon
April 1995	3 dtoc12 = 8.75 (0.02007 - Cp) (0.01605 - Cp)	
	(1 + 58.6 (0,01605 - Cp) +343 Cp)2	
	= 2.92 (0.02007 - Cp) (0.01605 - Cp)	
a distance	Joep 1 11 + 0,941 - 58.60p +34.3 (p)2	
	1 0 = 2,92 (0,02407-cp)(0.01605-cp)	
	(1.941 - 24.3 (p)2	
any and a second	(000 =0) (4=00)	
C	75% of limiting reactant	
	(1947-24,36 Cp) dep = ds	
	2.92 (0.02907 -Cp) (0.01507 +cp)	
The second	(10,012)	
	11,941-24,364)	de
	7.92 (0.01903 -Cp) (0.01605 (7Cp)	V. 37. 13. 14. 14.
		And the second s
	0 . 1 . 1	The second secon
9)	Real Answer from	
	Calculater: 89, 6698	
The second secon		
	P/	
10.7		