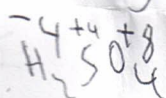


$$+2 \text{ Mg } s \quad \text{Mg}_2(\text{O}_2)_2$$
$$5.5 \times 10^7 \text{ H}^+ / \text{r} \quad 26. \text{K} \times 10^7 \text{ H}^+ / \text{r} \quad 4 \text{ } \text{Al}(\text{H}_2\text{SO}_4) \text{ mo}$$

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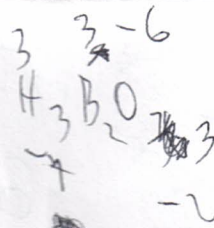
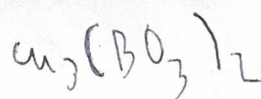
Sulfato de magnesio (II)

①, ②, ③, ④, ⑤
 1. hipo
 2. pro
 3. neo
 4. Arino

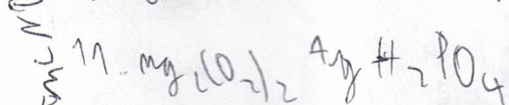
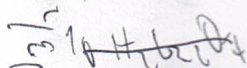
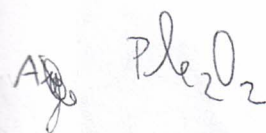
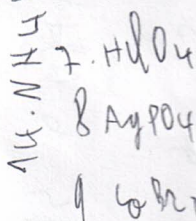


2. 14, 6

3. Mo 4. Brno
Hidrogeno sulfido de plumbum (PbS)


$$o \rightarrow i$$

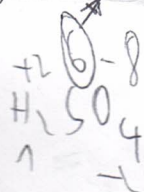
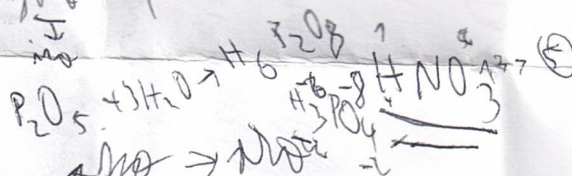
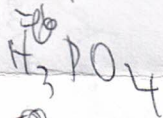
Borito de colere ($K\pm$)



Ag 9,35 dióxido $\text{Zn}(\text{NO}_3)_2$
Folha de prata

substantive and
also (1)

Pensado de
maquino



22. $M_2 + 2H_2$



⑤ Am_2SO_4
 NH_4F
 The compound is amorphous

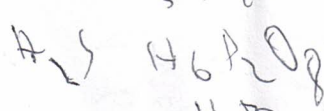
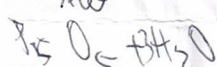
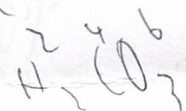
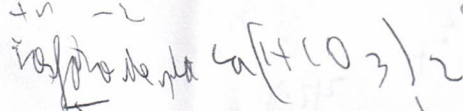
F
D
M

perforado de plom
4^a inv
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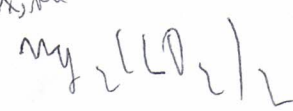
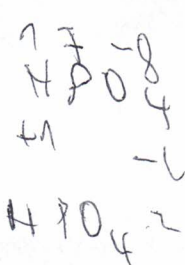
Nº 3
A¹ A⁰ 3

⑤ → no + no

Silenea delilis



Sulfuro de hidrogeno



Hidroxiide fersa



colaboração de mag

