




pyscscs: a PYthon Space-Charge Site-Explicit Solver

Georgina L. Wellock¹ and Benjamin J. Morgan^{1, 2}

¹ Department of Chemistry, University of Bath, Claverton Down, UK, BA2 7AY ² The Faraday Institution, Quad One, Harwell Science and Innovation Campus, Didcot, UK

DOI:

Software

- [Review](#) 
- [Repository](#) 
- [Archive](#) 

Submitted:

Published:

License

Authors of papers retain copyright and release the work under a Creative Commons Attribution 4.0 International License ([CC-BY](#)).

Summary

Solid electrolytes are often crystalline materials with high ionic conductivities, often used in solid state electrochemical devices. Most crystalline materials are characterised by defects, imperfections in the crystal that violate the perfect symmetries. Many solid electrolytes are polycrystalline with grain boundaries separating individual locally ordered regions with different orientations. In ionic crystals, thermodynamic arguments show that charged point defects will segregate to, or away from, interfaces and grain boundaries, and that there will be an associated redistribution of point defects in the regions adjacent. This redistribution leads to the regions having a net charge - known as space charge regions (Chiang, Lavik, Kosacki, Tuller, & Ying, 1996; Kim & Maier, 2002; Maier, 1984, 1985, 2003). The segregation of defects can produce local defect concentrations that strongly deviate from the average “bulk” values in a material (Fleig & J., 1996), which can result in large variations in ionic conductivities (Jamnik & Maier, 1999). It is therefore important to have a good understanding of defect behaviour and space charge formation to improve ionic conductivities in solid electrolytes for use in electrochemical devices. **pyscscs** is an open-source Python package for modelling ionic space-charge regions in crystalline solids. While the theory of space charge formation is well documented, there is currently no resources available for calculating space charge properties in solid electrolytes using explicitly defined defect positions and defect segregation energies.

The numerical model

pyscscs considers simple one-dimensional models of crystallographic interfaces, and calculates equilibrium defect distributions by solving a modified Poisson-Boltzmann equation (De Souza, Munir, Kim, & Martin, 2011; Maier, 1995). The driving force for point-defect segregation to or from the interface is described by sets of defect segregation energies, $\Delta E_{\text{seg}}^{i,x}$, defined as

$$\Delta E_{\text{seg}}^{i,x} = E_{\text{f}}^{i,x} - E_{\text{f}}^{i,\infty}$$

i.e. the difference in defect formation energy for defect species i at site x compared to a reference site in the “bulk” of the crystal.

The default model implemented in **pyscscs** calculates the equilibrium distribution of point-charge defect species on a 1D grid (De Souza et al., 2011; Helgee, Lindman, & Wahnström, 2013; Nyman, Helgee, & Wahnström, 2012; Polfus, Norby, & Bredesen, 2016). Finding the equilibrium distribution of defects is described by solving a modified Poisson-Boltzmann equation, which can be derived by requiring the electrochemical potentials at each site to be equal to a set of reference bulk electrochemical potentials (Maier, 2005).

$$\mu_{i,x}^o + RT \ln \left(\frac{c_{i,x}}{1 - c_{i,x}} \right) + z_i F \Phi_x = \mu_{i,\infty}^o + RT \ln \left(\frac{c_{i,\infty}}{1 - c_{i,\infty}} \right) + z_i F \Phi_{\infty}.$$

Within the general framework of solving this modified 1D Poisson-Boltzmann equation, `pyscses` can follow a number of different modelling approaches:

- Continuum (regular grid) and site-explicit (irregular grid) models.
- Periodic and Dirichlet boundary conditions.
- “Mott-Schottky” and “Gouy-Champman” conditions. These are implemented by setting the mobilities of different defect species. In the case of Mott-Schottky conditions, all but one defect species have a mobility of zero.
- Inclusion of “lattice” site charges to account for non-defective species in the crystal structure.

To calculate a number of space charge properties:

- Defect mole fractions.
- Charge density.
- Electrostatic potential.
- Parallel and perpendicular grain boundary resistivities (Hwang, McLachlan, & Mason, 1999).
- Grain boundary activation energies (Kim, 2016).

Approximations and limitations

- `pyscses` implements a modified Poisson-Boltzmann equation, which assumes that defects are non-interacting except for point-charge electrostatics and site exclusion.
- The resistivity and activation energy calculations implemented in `pyscses` assume mobilities of each defect species are independent of the local crystal structure.

Acknowledgements

B. J. M. acknowledges support from the Royal Society (Grant No. UF130329).

References

- Chiang, Y. M., Lavik, E. B., Kosacki, I., Tuller, H. L., & Ying, J. Y. (1996). Defect and transport properties of nanocrystalline CeO_{2-x} . *Appl. Phys. Lett.*, 69(2), 185–187. doi:[10.1063/1.117366](https://doi.org/10.1063/1.117366)
- De Souza, R. A., Munir, Z. A., Kim, S., & Martin, M. (2011). Defect chemistry of grain boundaries in proton-conducting solid oxides. *Solid State Ionics*, 196(1), 1–8. doi:[10.1016/j.ssi.2011.07.001](https://doi.org/10.1016/j.ssi.2011.07.001)
- Fleig, J., & J., M. (1996). Point contacts in solid state ionics: Finite element calculations and local conductivity measurements. *Solid State Ionics*, 86-88, 1351–1356. doi:[10.1016/0167-2738\(96\)00316-5](https://doi.org/10.1016/0167-2738(96)00316-5)
- Helgee, E. E., Lindman, A., & Wahnström, G. (2013). Origin of space charge in grain boundaries of proton-conducting BaZrO_3 . *Fuel Cells*, 13(1), 19–28. doi:[10.1002/fuce.201200071](https://doi.org/10.1002/fuce.201200071)
- Hwang, J.-H., McLachlan, D. S., & Mason, T. O. (1999). Brick layer model analysis of nanoscale-to-microscale cerium dioxide. *J. Electroceram.*, 3(1), 7–16. doi:[10.1023/A:1009998114205](https://doi.org/10.1023/A:1009998114205)

- Jamnik, J., & Maier, J. (1999). Defect chemistry and chemical transport involving interfaces. *Solid State Ionics*, 119(1), 191–198. doi:[10.1016/S0167-2738\(98\)00502-5](https://doi.org/10.1016/S0167-2738(98)00502-5)
- Kim, S. (2016). Isn't the space-charge potential in ceria-based solid electrolytes largely overestimated? *Phys. Chem. Chem. Phys.*, 18, 197870–19791. doi:[10.1039/C6CP02177H](https://doi.org/10.1039/C6CP02177H)
- Kim, S., & Maier, J. (2002). On the conductivity mechanism of nanocrystalline ceria. *J. Electrochem. Soc.*, 149(10), J73–J83. doi:[10.1149/1.1507597](https://doi.org/10.1149/1.1507597)
- Maier, J. (1984). Enhancement of the ionic conductivity in solid-solid-dispersions by surface induced defects. *Berichte der Bunsengesellschaft für physikalische Chemie*, 88(11), 1057–1062. doi:[10.1002/bbpc.198400007](https://doi.org/10.1002/bbpc.198400007)
- Maier, J. (1985). Space charge regions in solid two-phase systems and their conduction contribution I. Conductance enhancement in the system ionic conductor-inert phase and application on AgCl:Al₂O₃ and AgCl:SiO₂. *J. Phys. Chem. Sol.*, 46(3), 309–320. doi:[10.1016/0022-3697\(85\)90172-6](https://doi.org/10.1016/0022-3697(85)90172-6)
- Maier, J. (1995). Ionic conduction in space charge regions. *Prog. Sol. Stat. Chem.*, 23(3), 171–263. doi:[10.1016/0079-6786\(95\)00004-E](https://doi.org/10.1016/0079-6786(95)00004-E)
- Maier, J. (2003). Defect chemistry and ion transport in nanostructured materials: Part ii. Aspects of nanoionics. *Solid State Ionics*, 157(1), 327–334. doi:[10.1016/S0167-2738\(02\)00229-1](https://doi.org/10.1016/S0167-2738(02)00229-1)
- Maier, J. (2005). Physical chemistry of ionic materials: Ions and electrons in solid. doi:[10.1002/0470020229.ch1](https://doi.org/10.1002/0470020229.ch1)
- Nyman, B. J., Helgee, E. E., & Wahnström, G. (2012). Oxygen vacancy segregation and space-charge effects in grain boundaries of dry and hydrated BaZrO₃. *Appl. Phys. Lett.*, 100(6). doi:[10.1063/1.3681169](https://doi.org/10.1063/1.3681169)
- Polfus, J. M., Norby, T., & Bredesen, R. (2016). Proton segregation and space-charge at the BaZrO₃ (001)/MgO (001) heterointerface. *Sol. Stat. Ionics*, 297, 77–81. doi:[10.1016/j.ssi.2016.10.010](https://doi.org/10.1016/j.ssi.2016.10.010)