

## 0.1 Idea and Assumptions

The VGM links three basic quantities to describe the growth trajectory of single cells: volume, osmotic and turgor pressure. Based on a formalism established by Kedem-Katchalsky water flux across the cell membrane is proportional to the difference between pressures driving water out of the cell (outer osmotic pressure and turgor pressure) and pressures driving water into the cell (inner osmotic pressure). Using Van t'Hoff's law inner and outer osmotic pressure are proportional to inner and outer concentration of osmotically active molecules (osmolytes). The concentration of osmolytes outside of the cell is set constant. The concentration of osmolytes inside the cell increases with an uptake rate proportional to the cell's surface and decreases due to dilution of the growing cell and osmolyte consumption proportional to the cell's volume.

The turgor pressure acts on the cell wall and is increased by water influx- the cell wall gets elastically expanded. When a critical turgor pressure is reached, the cell wall expands plastically (permanently) and turgor pressure is released. This circuit results in a stepwise increase in cell volume, until the final size is reached: When the inner osmotic pressure is greater than the sum of outer osmotic pressure and turgor pressure water flows in and the cell grows. The turgor pressure increases, ultimately leading to a growth stop, but upon reaching the critical value for plastic expansion drops again, allowing some more growth.

To simulate the growth of a bud during S/G2/M phase, after letting the mother cell grow in G1, a second cell is initialized with small starting volume. Both cells are coupled via exchange terms for water and osmolytes, depending on turgor pressure and osmolyte concentrations. Both mother and bud are approximated as spheres.

## 0.2 Equations and parameters

The tables below contain all species, parameters and equations used in the VGM.

Species	description
$V_{os}$	Osmolotically active volume, increases with water influx
$V_b$	Volume of solid components, proportional to reference volume
$V$	Total cell volume
$V_{ref}$	Reference volume, volume of relaxed cell without elastic expansion (grows by plastic expansion)
$\Pi_t$	Turgor pressure
$c_i$	Inner osmolyte concentration

parameter	decription	value	unit
$V_{os}^0$	Initial volume of solid components	10	$\mu m^3$
$V_b^0$	Initial volume of solid components	3	$\mu m^3$
$c_i^0$	Initial inner osmolyte concentration	322.2	$mM$
$\Pi_t^0$	Initial turgor pressure	$2.0 \cdot 10^5$	$Pa$
$c_e$	Outer osmolyte concentration	240.0	$mM$
$R$	Ideal gass constant	8.314	$\frac{J}{mol \cdot K}$
$T$	Temperature	293.0	$K$
$L_p$	Membrane water permeability	$1.19 \cdot 10^{-6}$	$\frac{\mu m}{s \cdot Pa}$
$\Pi_{tc}$	Critical turgor pressure	$2.0 \cdot 10^5$	$Pa$
$d$	Cell wall thickness	0.115	$\mu m$
$\Phi$	Cell wall extensibility	$1.0 \cdot 10^{-7}$	$\frac{1}{s \cdot Pa}$
$E$	Young's modulus	$2.58 \cdot 10^6$	$Pa$
$k_{uptake}$	Osmolyte uptake rate constant	$2.5 \cdot 10^{-16}$	$\frac{mM}{s \cdot \mu m^2}$
$k_{cons}$	Osmolyte consumption rate constant	$3.0 \cdot 10^{-16}$	$arb.unit$

<b>ODE</b>
$\frac{d}{dt} V_{os} = -L_p \cdot A \cdot (\Pi_t + \Pi_e - \Pi_i)$
$\frac{d}{dt} V_b = 0.2 \cdot V_{ref}$
$\frac{d}{dt} V = \dot{V}_{os} + \dot{V}_b$
$\frac{d}{dt} V_{ref} = \frac{\Phi \cdot r}{d \cdot f(\Pi_t)} \cdot V_{ref}$
$\frac{d}{dt} \Pi_t = \frac{E \cdot 2d}{r} \cdot \frac{\dot{V}}{V_{ref}} - E \cdot \Phi \cdot f(\Pi_t) - \frac{\Pi_t}{V} \cdot \dot{V}$
$\frac{d}{dt} c_i = k_{uptake} \cdot \frac{A}{V} - k_{cons} - \frac{c_i}{V} \cdot \dot{V}$
$f(\Pi_t) = \max(\Pi_{tc} - \Pi_t, 0)$
$A = (4\pi)^{\frac{1}{3}} \cdot (3V)^{\frac{2}{3}}$
$r = \left(\frac{3}{4\pi} V\right)^{\frac{1}{3}}$

### 0.3 Main results