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1 Alcanes

1.1 Butane [A.1]

1.2 Cyclohexane [A.2]



1.3 Hexane [A.3]



1.4 Isobutane [A.4]

$$\begin{array}{c} \operatorname{CH_3} \\ | \\ \operatorname{C} \\ \vdots \\ \operatorname{CH_3} \end{array}$$

2 Carboxylic Acids

2.1 Acetic Acid [B.1]

$$\begin{array}{c} H - C - C \\ \downarrow \\ O \\ \end{array} O H$$

2.2 Acrylic Acid [B.2]

2.3 Benzoic Acid [B.3]

2.4 Fumaric Acid [B.4]

2.5 Maleic Acid [B.5]

2.6 Oleic Acid [B.6]

3 Polymers

3.1 Poly(Bisphenol A Carbonate) [C.1]

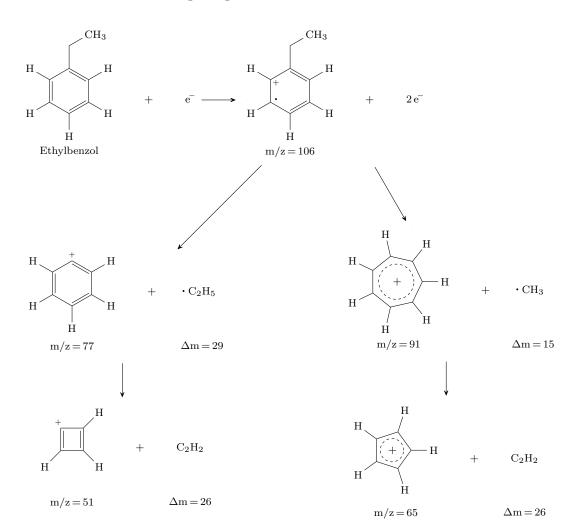
3.2 Polyurethane [C.2]

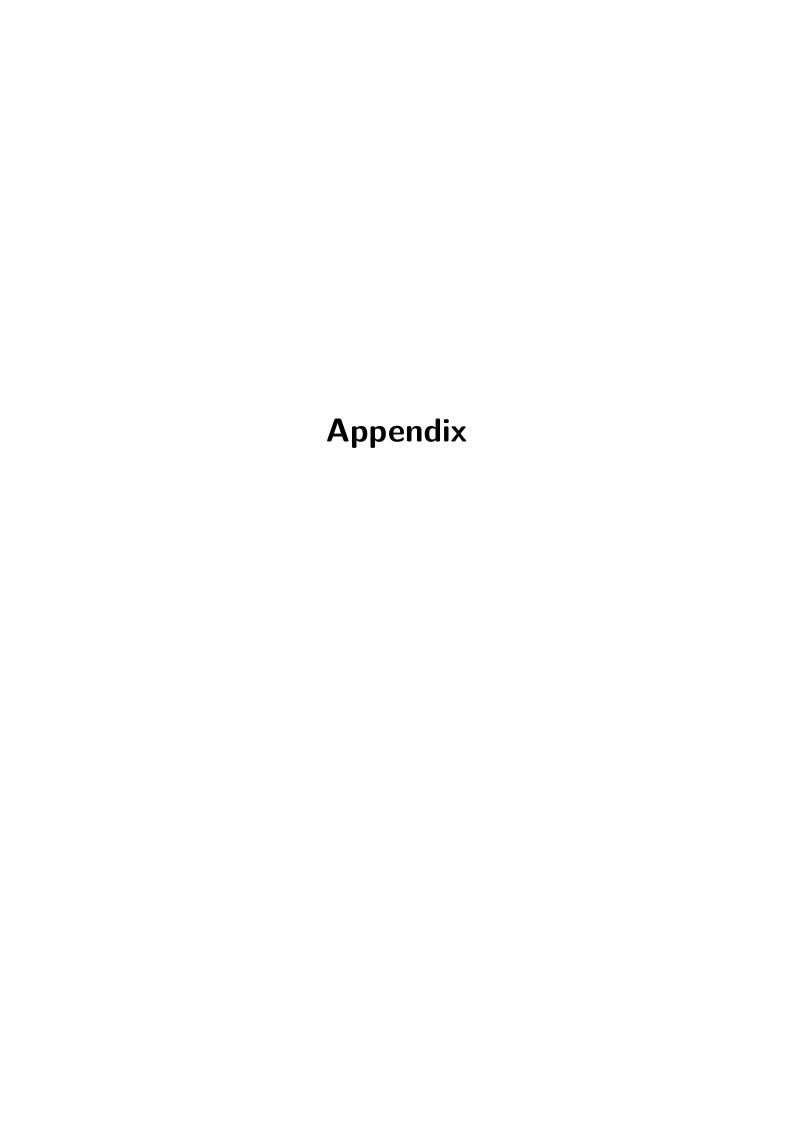
$$\begin{bmatrix} \mathbf{O} & \mathbf{H} & \mathbf{H} \\ \vdots \\ \mathbf{C} - \mathbf{N} & & \vdots \\ \mathbf{H} & & \mathbf{H} & \mathbf{H} \end{bmatrix}_n \\ \mathbf{O} & \mathbf{H} & \mathbf{H} \\ \vdots & \vdots & \vdots \\ \mathbf{N} - \mathbf{C} - \mathbf{O} - \mathbf{C} - \mathbf{C} - \mathbf{O} \\ \vdots & \vdots & \vdots \\ \mathbf{H} & & \mathbf{H} & \mathbf{H} \end{bmatrix}_n$$

4 Mass Spectrometry

4.1 1-Propanol [D.1]

4.2 Ethylbenzene [D.2]





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A Alcanes

A.1 Butane

```
\label{eq:chemfig} $$ \left( H-C(-[2]H)(-[-2]H)-C(-[2]H)(-[-2]H)-C(-[2]H)(-[-2]H)-C(-[2]H) \right) = C(-[2]H)(-[-2]H)-H $$
```

A.2 Cyclohexane

```
\left( \text{chemfig} \left\{ *6(-----) \right\} \right)
```

A.3 Hexane

A.4 Isobutane

```
\label{localization} $$ \left[ angle increment = 30 \right] \left\{ H_3C - [1]C(<: [-2.75]H)(-[3]CH_3) \right. $$ < [-1]CH_3 \right\} $$
```

B Carboxylic Acids

B.1 Acetic Acid

```
\left[ \left( -[2]H \right) - \left( -[-1]OH \right) = [1]O \right]
```

B.2 Acrylic Acid

```
\ensuremath{$\backslash$} chemfig[angle increment=30]{=^[-1]-[1](-[-1]0H)=[3,0.8]0}
```

B.3 Benzoic Acid

```
\left\{ *6(-=-=(-(=[3.333]0)(-[0.666]0H))-=) \right\}
```

B.4 Fumaric Acid

```
\label{localization} $$ \left( -[-2](=_[:-30](-[-2](=[:-30]0)(-[4.666]H0)) \right) = [:30]0 $$
```

B.5 Maleic Acid

```
\chemfig[baseline=(b.base)]{HO-[-0.66](=[0.666]0)(*6(-0{b})=-(-[2]0H)(=[-0.666]0))}
```

B.6 Oleic Acid

```
\chemfig[]
{HO-[0.666](=[2,0.8]0)
-[-0.666]-[0.666]-[-0.666]-[-0.666]-[-0.666]
-[0.666]=_
-[0.666]=_
-[-0.666]-[0.666]-[-0.666]-[-0.666]-[-0.666]
-[0.666]
```

C polymers

C.1 Poly(Bisphenol A Carbonate)

C.2 Polyurethane

```
\chemfig[]
{-[@{op,.5}]C(=[2]0)-N(-[-2]H)-*6(-=-(-C(-*6(=-=(-N(-[-2]H)-C(=[2]0)-0-C(-[2]H)(-[-2]H)-C(-[2]H)(-[-2]H)-0-[@{cl,0.5}])
-=-))(-[
2]H)
(-[-2]H))=-=)}
\polymerdelim[height = 20pt, depth = 20pt, delimiters
={[]}, indice = \!\!n]{op}{cl}
```

D Mass Spectrometry

D.1 1-Propanol

```
\schemestart
charge {90:1pt=\|,-90:1pt=\|}{Propan-1-ol}
\chemfig{\charge{45=$\criptscriptstyle{-}$}{e}}
\arrow(.mid east--.mid west)
\chemname
{\left\{ chemfig\{H-C(-[2]H)(-[-2]H)-C(-[2]H)(-[-2]H)-C(-[2]H)(-[-2]H)-charge \right\} }
   \{115:1pt=\,,-90:1pt=\|,60:1pt=\\scriptscriptstyle\{+\}\}\}\{0\}-H\}
\{m/z\,=\,60\}
\+
\left\{2\right, \left\{45=\right\}\right\}
\arrow(@c2--n1)[-70,1.5]
\chemname{\chemfig{\charge{180:1pt=$\criptscriptstyle{+}}}{C}(-[2]H)}
   (-[-2]H) - charge {90:1pt=\|,-90:1pt=\|}{0}-H}}{\|
\arrow(.mid east--.mid west)[0,0.7]
\chemnameinit{}
\chemname{\chemfig{C(-[3]H)(-[-3]H)=\charge{90:2pt=$\scriptscriptstyle{+}}}
  ,-90:1pt=\{0\}-H\}\{m/z\},=\{31\}
\chemnameinit{}
\arrow(@c2--n2)[-100,5]
\chemname{\chemfig{H-C(-[2]H)(-[-2]H)-\charge{90:2pt=$\criptscriptstyle}}
   \{+\}$\{C\}(-[-2]H)-\charge\{0:1pt=\.\}\{C\}(-[2]H)(-[-2]H)}
\{m/z\,=\,42\}\qquad
\+\qquad
\chemname{\chemfig{H_20}}{\$\Delta\$m\,=\,18}
\chemnameinit{}
\arrow(@c2--n4)[225,3]
\chemname
{\left\{ chemfig\{H-C(-[2]H)(-[-2]H)-C(-[2]H)(-[-2]H)-C(-[2]H)(-[-2]H)-charge \right\} }
   \{m/z\,=\,59\}\qquad
\chemname{\chemfig{\charge{0:1pt=\.}{H}}}{\shortengtash}, =\,1}
\arrow(@n1--nn1)[-90,0.45,white]
}{{\bf m},=\,29}
\schemestop
```

D.2 Ethylbenzene

```
\schemestart
          \chemname {\chemfig {*6((-H)-(-H)=(-H)-(-H)=(-[2](-[0.666])}
                 CH 3))-(-H)=(-H))}{Ethylbenzol} \qquad
          \+ \qquad
          \left(\frac{45=\$}{criptscriptstyle}, \frac{45=\$}{e}\right)
          \arrow(.mid east--.mid west)
          \chemname{\chemfig} *6(\charge{30:3pt=\.}{}(-H)-(-H)=(-H)
                 -(-H)=(-[2](-[0.666]CH 3))-\charge{-30:3pt=$}
                 scriptscriptstyle {+}$}{}(-H
)-(-H))}{m/z\,=\,106} \qquad
          \+ \qquad
          \left(2\right, \left(45=\$\right) = \left(45=\$\right)
          \arrow(@c2--n1)[-60,1.5]
          \chemname{\chemfig{**[0,360,dash pattern=on 2pt off 2pt
                 ]7(\charge{25:18.5pt=\+}{}(-H)-(-H)-(-H)-(-H)-(-H)-(-H)
                 -(-H)-(-H))}{m/z\,=\,91}
          \chemname{\chemfig{\charge{180:2pt=\.}{C}H 3}}{\sharpe{180:2pt=\.}{C}H 3}}
                 \ \ ,=\ ,15\ 
          \arrow(@n1--m2)[-90,1]
          \chemname{\chemfig{**[0,360,dash pattern=on 2pt off 2pt
                 ]5(\langle H) - (-H) - (-H)
                 -(-H)-)}{m/z\,=\,65}
          \qquad \+ \qquad
          \chemname{\chemfig{C_2H_2}}{$\Delta$m\,=\,26}
          \arrow(@c2--n4)[225,2.5]
          \chemname{\chemfig} {*6((-H)-(-H)=(-H)-(-H)=\charge{90:3pt=$
                 \scriptscriptstyle\{+\}$\}\{\}-(-H)=(-H))\}\{m/z\,=\,77\}
          \qquad \+ \qquad
          \chemname{\chemfig{\charge{180:2pt=\.}{C} 2H 5}}{\shape{180:2pt=\.}{C}}
                 \ \ ,=\ ,29
          \arrow(@n4--m2)[-90,1]
          scriptscriptstyle\{+\}$\}{\}=)}\{m/z\setminus,=\setminus,51\}
          \qquad \+ \qquad
          \chemname{\chemfig{C_2H_2}}{$\Delta$m\,=\,26}
\schemestop
```