# SEMESTER EXAMINATION DECEMBER-2023

Course Name: -

B.Tech

Semester:- 1st

Paper Name: -

**Engineering Chemistry** 

Paper Code:- TBS 103

Time - 3 Hrs + 20 minutes per hour extra time for V.I. & examinees with writer.

Max Marks-70

समय- 3 घण्टे + 20 मिनट प्रति घंटे अतिरिक्त-दृष्टिबाधित एवं सह लेखक परीक्षार्थियों के लिए।

अधिकतम अंक-70

#### Instructions:

- The question paper consists of three sections namely A, B, C. All sections are compulsory.
- Section A- Each question carries 3 mark. All questions are compulsory.
- Section B- Answer any 5 out of 7 given questions in maximum one hundred fifty (150) words. Each question carries 7 marks.
- Section C- Answer any 2 out of 3 given questions in maximum three hundred (300) words. Each question carries 10 marks.

#### निर्देशः

- प्रश्न पत्र में तीन खण्ड अ, ब, व स हैं। सभी खण्ड अनिवार्य हैं।
- खण्ड—अ में प्रत्येक प्रश्न तीन अंक का है। सभी प्रश्न अनिवार्य हैं।
- खण्ड-ब में सात प्रश्नों में से किन्हीं पाँच प्रश्नों के उत्तर अधिकतम 150 शब्दों में दें। प्रत्येक प्रश्न सात अंक का है।
- खण्ड-स में तीन प्रश्नों में से किन्हीं दो प्रश्नों के उत्तर अधिकतम (250-300) शब्दों में दें। प्रत्येक प्रश्न 10 अंक का है।

#### Section - A (खण्ड-अ)

### Objective Questions(वस्तुनिष्ठ प्रशन )

1. Answer all the following questions.

निम्नलिखित सभी प्रश्न अनिवार्य हैं।

5x3 = 15

- (i) Correct Bond length increasing order is,
  - a)  $O_2^- < F_2 < N_2 < C_2$
  - b)  $N_2 < F_2 < O_2 < C_2$
  - c)  $N_2 < C_2 < O_2 < F_2$ 
    - d)  $F_2 < C_2 < O_2 < N_2$
- ii) No of Pi electrons, magnetic character, unpaired electrons and magnetic moment in  $F_2$ 
  - a) 2, Paramagnetic, 2 and 2.5
  - b) 1, Diamagnetic, 0 and  $\sqrt{8}$
  - c) 2, Paramagnetic, 2 and 2.8
  - d) 2, Diamagnetic, 1 and  $\sqrt{3}$
- iii) Degree of hardness caused by 50 ppm CaCl2 is
  - a) 45.0 mg/L permanent
  - b) 43.2 mg/L permanent
  - c) 50.7 mg/L temporary
  - d) 12.3 mg/L permanent
- iv) Correct relation is,
  - a) Animal lubricant ↔ castor oil
  - b) Graphite ↔ 450 degree centigrade
  - c) Soda lubricant ↔ water repellent
  - d) MoS<sub>2</sub> and BN ↔ semisolid lubricant

- v) Semiconductors used as electrically supportable materials, most likely because they have
  - a) whole
  - b) electrons
  - c) p-n junction
  - d) strength

## Section - B (खण्ड-ब) Short Answer Questions (लघुउत्तरीय प्रश्न)

2. Answer any five of the following questions in maximum 150 words.

5x7 = 35

निम्नलिखित में से किन्हीं पाँच प्रश्नों के उत्तर अधिकतम 150 शब्दों में दें।

- i. Write the percentage composition of Portland cement.
- ii. Explain electrochemical theory of corrosion and different types of corrosion.
- iii. Give the structure of monomers of all listed polymers.

  Nylon 6, Nylon 66, Buna-S, Buna-N, PAN, Bakelite, Polyethene.
- iv. Give five difference between Addition and Condensation Polymers/ Polymerisation.
- v. What are lubricants? How its uses can increase industry efficiency? What the properties of a good lubricant.
- vi. Draw MO diagram and calculate bond order of  $O_2^+$ ,  $F_2^+$ ,  $N_2^{--}$
- vii. Calculate total hardness of a water sample containing 15 mg CaCO<sub>3</sub>, 25 mg MgCl<sub>2</sub>, 30 mg MgSO<sub>4</sub> and 50 mg CaHCO<sub>3</sub>. Express hardness value in degree Clark and degree French.

# Section - C (खण्ड-स) Descriptive Questions (विवरणात्मक प्रश्न)

3. Answer any two of the following question in maximum 300 words.

2x10=20

निम्नलिखित में से किन्हीं दो प्रश्नों के उत्तर अधिकतम 300 शब्दों में दें।

- i) Draw phase diagram of water system as according to phase rule based on curve and area.
- ii) What are liquid crystals? Classify them on the basis of properties owned by respective mesogens.
- iii) Explain Zeolite and Ion exchange method for softening hard water with detail of softening and regeneration

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