

GILES CHEMICAL COMPANY PROCEDURE

Loss On Ignition (LOI) – Magnesite (MgO)

Page : 1 of 4 Revision Date

06/29/09

Carl Mooney Job Specific Author:

Safety: Wear the appropriate lab PPE and follow area safety rules.

Purpose: LOSS ON IGNITION (LOI) -- Magnesite (MgO)

Procedure:

Introduction

Magnesite (MgO) is the raw material source of Magnesium used for the production of Magnesium Sulfate (MgSO4) at this location. It is the result product of roasting Magnesium Carbonate (MgCO₃) ore to MgO and CO₂ by suppliers. For yield and operating purposes it is important to know the extent of removal of CO2 as delivered, as well as the moisture content.

Procedure

A small amount of sample is placed in a porcelain crucible, with appropriate weightings, and heated at 105°C, during which free moisture is driven off. Following this the crucible and contents are heated at 1000°C, during which any residual MgCO3 is reduced to MgO and CO2. Using the weight determinations, percentage figures can be calculated for H2O and LOI, individually and in total, and these compared with the C of A submitted by the supplier. Any substantial difference is reported to the POC

Equipment.

Porcelain Crucible -- #C6450-4. 10 mL. Capacity

Weighing balance—B440 Satorius

Drying Oven (Low temperature) -- S/P TempCon Gravity convection N86205 Muffle Furnace (High temperature) -- Thermodyne Type 1300 120V

Dessicator—Jencon's 10" Dry Seal Tongs—T5010 9" Steel Small Spatula

Method

Results Recorded in Computer Log

- 1. Place the porcelain crucible in the 100°C oven for 10 minutes to assure dryness.
- 2. Tare the weighing scale to Zero.
- 3. Place the dry crucible on the scale and record the weight.
- 4. Add approximately 5 grams of sample to the crucible and record the combined weight.
- 5. Subtract the weight of the crucible from the combined weight. This is the weight of the sample (A).
- 6. Place the crucible in the 105°C oven for 30 minutes.
- 7 Remove the crucible from the oven and place it upon the scale. Record the weight.
- 8. Subtract the weight of the crucible and contents from the combined weight at the start. This is the weight of the moisture H2O removed – (B)
- 9. Increase the setting for the muffle furnace to 1000°C
- 10. Place the crucible in the furnace, using forceps, for two hours.



GILES CHEMICAL							
COMPANY PROCEDURE							
Loss On Igni	tion (LOI) – Magnesite (MgO)	Page	:	2 of 4	Revision Date	:	01 06/29/09
Author:	Carl Mooney	Job	Spe	ecific	Dute		00/2//0/

- 11. Transfer the crucible, using the forceps, to the dessicator to cool.
- 12. Return the crucible to the furnace for 15 minutes
- 13. Repeat items 11 and 12 until constant weight is achieved.
- 14. Record the weight and subtract that weight from the weight at item 7. This is the weight of the CO2 driven off (C).

B/A = % moisture in the sample

 $C/A = \% CO_2$ driven off.

(B+C)/A = %LOI

- 15. Return the muffle furnace temperature setting to 500°C
- 16. Compare the results with the C of A for the carload from which the sample came, and report any substantial discrepancy to the POC.



GILES CHEMICAL COMPANY PROCEDURE

Loss On Ignition (LOI) – Magnesite (MgO)

Page : 3 of 4

Revision : Date : 06

01 06/29/09

Author: Carl Mooney

Job Specific

TRAINING DOCUMENTATION

	EMPLOYEE	TITLE	SIGNATURE	DATE
1				
2				
3				
4				
5				
6				
7				
8				
9				
10				
11				
12				
13				
14				
15				
16				
17				
18				
19				
20				
21				
22				
23				
24				
25				
26				
27				
28	-			
29				
30				



GILES CHEMICAL COMPANY PROCEDURE

Loss On Ignition (LOI) – Magnesite (MgO)

Page : 4 of 4

Revision : 01 Date : 06/29/09

Author: Carl Mooney Job Specific

Revision Number	Revision Date	Revision Author	Revision Description
00	03/28/06	СМ	New Document
01	06/29/09	SL	-Placed Procedure on New Form