IMPERIAL COLLEGE LONDON

BSc and MSci DEGREES – JANUARY 2013, for Internal Students of the Imperial College of Science, Technology and Medicine

This paper is also taken for the relevant examination for the Associateship

ADVANCED CHEMISTRY THEORY IIA

Organic Chemistry

Thursday 10th January 2013, 14:00-16:00

PLEASE NOTE THAT IT IS DEPARTMENTAL POLICY THAT THESE EXAM QUESTIONS MAY REQUIRE UNDERSTANDING OF ANY PRIOR CORE COURSE.

USE A SEPARATE ANSWER BOOK FOR EACH QUESTION. WRITE YOUR CANDIDATE NUMBER ON EACH ANSWER BOOK.

Year 2/0113 Turn Over

2.01 – Organic Synthesis Part 1

Q1. Answer ALL parts of this question.

a) Explain the relative reactivity of the three carbonyl containing compounds below to nucleophiles (e.g. MeMgCl) and place in order of most reactive to least reactive.

b) Give the organic product of **THREE** out of the FIVE transformations i) to v) below after work-up. In each case the reaction is worked up with a source of aqueous H⁺ to isolate the desired product. Provide a mechanism for the formation of each product and identify any selectivity features.

(5 marks each)

QUESTION CONTINUED OVERLEAF

c) Answer **BOTH** parts of this question.

In the reaction scheme shown below, the product shown ${\bf C}$ was formed and not the desired target ${\bf X}$ in the box.

i) Explain the selectivity for product ${\bf C}$ over product ${\bf X}$ under the conditions shown and provide a mechanism for formation of ${\bf C}$.

(4 marks)

ii) Suggest an approach to form the desired product \boldsymbol{X} from \boldsymbol{A} and $\boldsymbol{B}.$

(2 marks)

2.O1 – Organic Synthesis Part 1

Q2. Answer **BOTH** parts of this question.

a) Suggest reagents to carry out **TWO** of the following transformations. In each case, explain the key mechanistic features that lead to any selectivity that is required.

(6 marks each)

(ii)
$$CO_2H$$

OH

(iii) MeO_2C

OH

 CO_2H
 CO_2H

OH

 CO_2H
 CO_2H

- b) For the synthetic sequence below:
 - i) Suggest a structure for **B**, including stereochemistry.

(3 marks)

ii) Give a mechanism for the formation of ${\bf B}$ from ${\bf A}$ and explain any issues of selectivity.

(4 marks)

iii) Suggest a structure for C.

(1 mark)

iv) Suggest a structure for **D**.

(1 mark)

v) Give a mechanism for the formation of ${\bf D}$ from ${\bf C}$ and explain any issues of selectivity.

(4 marks)

2IS.2 – Introduction to NMR and EPR Spectroscopy

Answer part a) and **EITHER** part b) **OR** part c) of this question.

- a) Answer **ALL** parts of this question.
 - i) Give a brief definition of the following terms:
 - Boltzmann distribution
 - Spin angular momentum
 - Zeeman interaction

(6 marks)

ii) Explain two ways to increase the sensitivity in NMR.

(2 marks)

iii) Explain why the ¹H NMR chemical shift of the methyl groups are -4.25 ppm in the structure below.

(1 mark)

iv) For the following molecules rearrange in order of increasing ¹H NMR chemical shift.

(1 mark)

QUESTION CONTINUED OVERLEAF

v) Describe the splitting for H_A and H_B . With the aid of a Newman projection give an explanation for these splitting patterns.

(3 marks)

$$HO_2C$$
 H^{WW}
 H_2N
 H_A
 H_B
 H_B
 H_B
 H_B

vi) Propose the structure of the molecule, $C_8H_{12}O_2$, which has the following 1H NMR spectrum

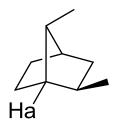
Chemical Shift (ppm)	Multiplicity	Integration	J Coupling (Hz)
0.92	d	6H	7.0
1.84	m	1H	7.0
1.91	t	1H	1.5
2.42	d	2H	1.5
3.81	d	2H	7.0

and infra-red stretching peaks at 1742 and 2150 cm⁻¹. Give a brief explanation for your proposed structure. (s=singlet, t=triplet, q=quartet, m= multiplet)

(6 marks)

- b) Answer **BOTH** parts of this question.
 - i) Draw the splitting pattern for Ha. Include all couplings for H_a up to and including 3J .

(3 marks)



QUESTION CONTINUED OVERELAF

ii) What is the chemical shift and splitting pattern for H_a up to and including 4J ? (3 marks)

- c) Answer **BOTH** parts of this question.
 - i) What is the approximate chemical shift and splitting pattern for H_a up to and including 4J ?

(2 marks)

$$H_b$$
 H_c
 H_c
 H_c

ii) What is the approximate chemical shift for all the protons in the molecule shown below? Draw the splitting pattern for H_a . Include all couplings for H_a up to and including 3J .

(4 marks)