

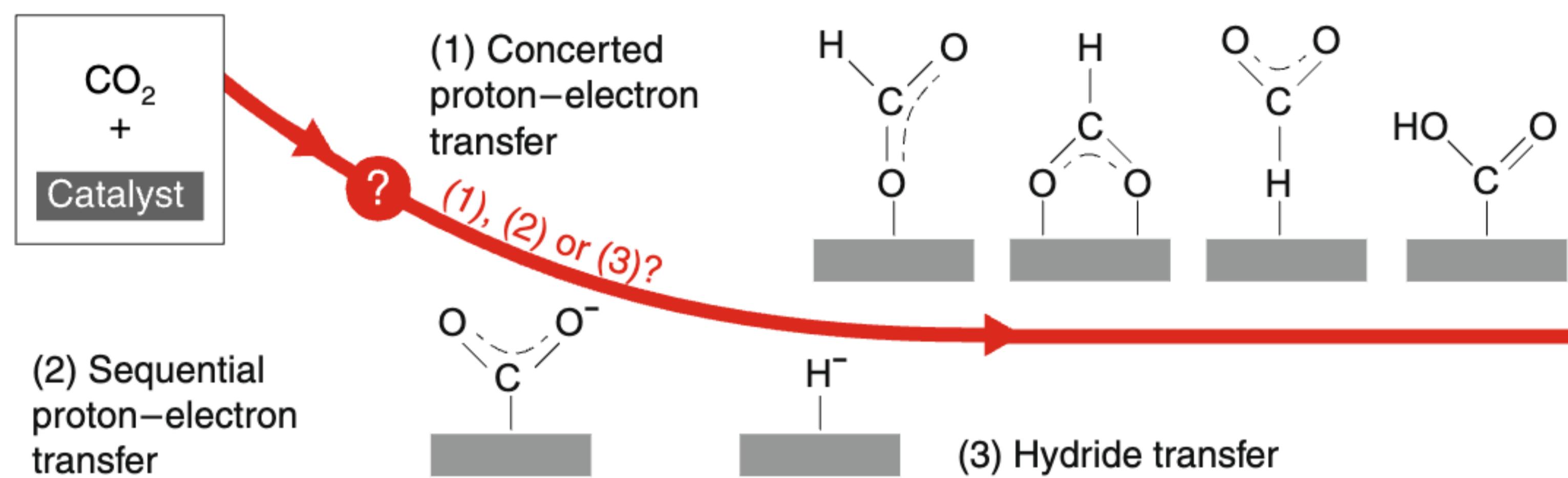
# Inverted Region in Electrochemical Reduction of CO<sub>2</sub> Induced by Potential-dependent Pauli Repulsion

Leyu Liu, Hai Xiao\*

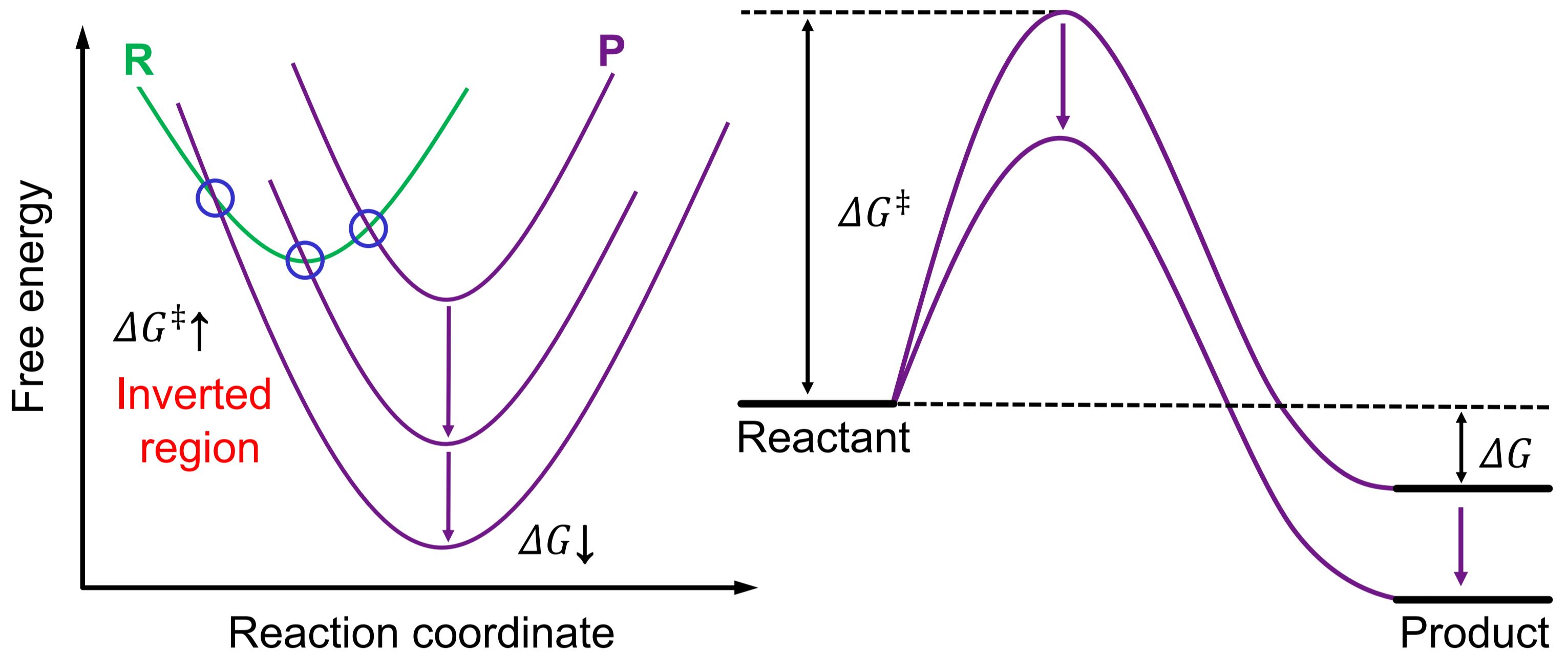
Department of Chemistry, Tsinghua University, Beijing 100084 Email:liu-ly19@mails.tsinghua.edu.cn

## Introduction

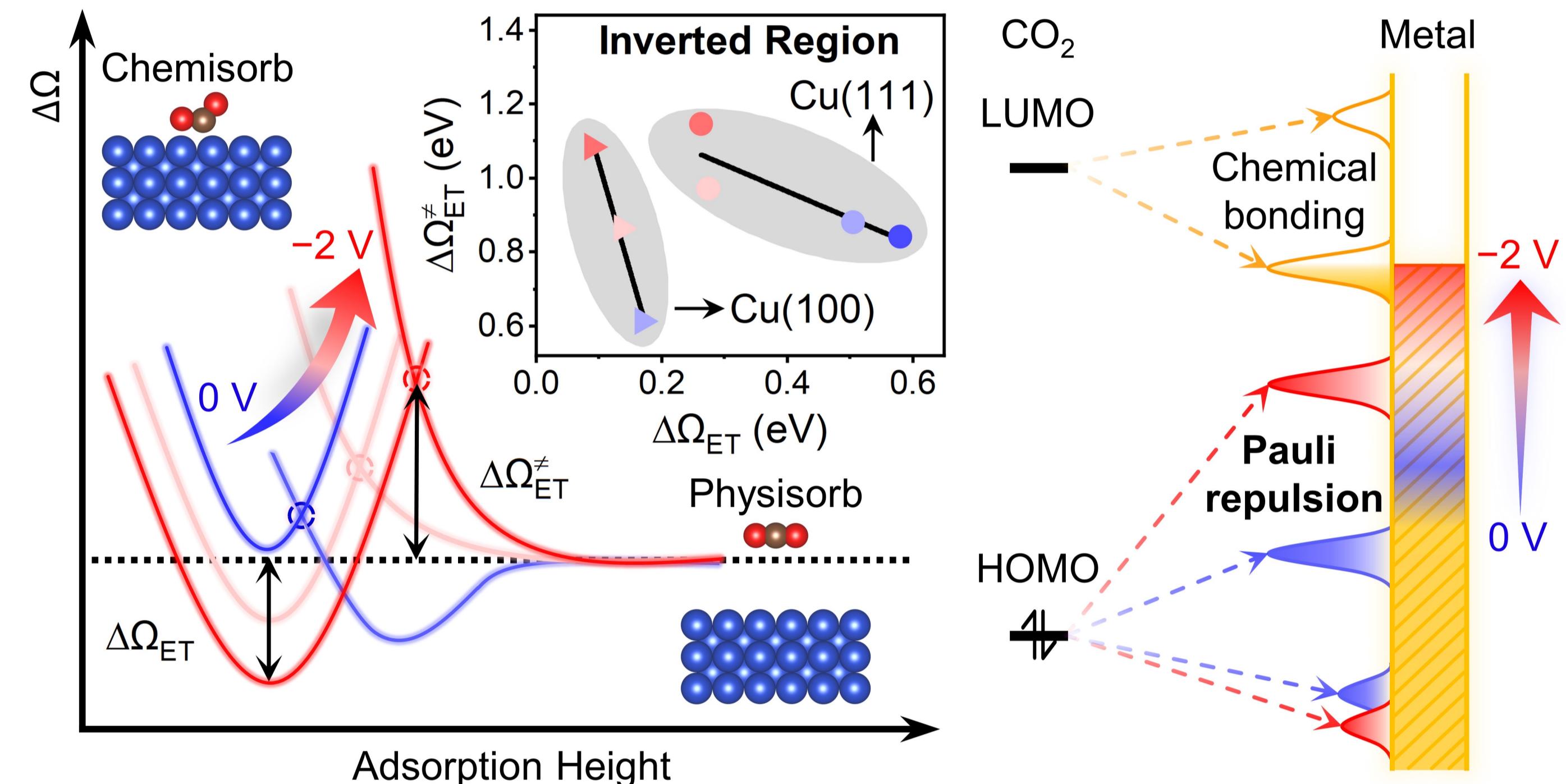
### CO<sub>2</sub> activation: SEPT vs. CPET



### Electron transfer kinetics: Marcus theory vs. BEP relation



## Abstract

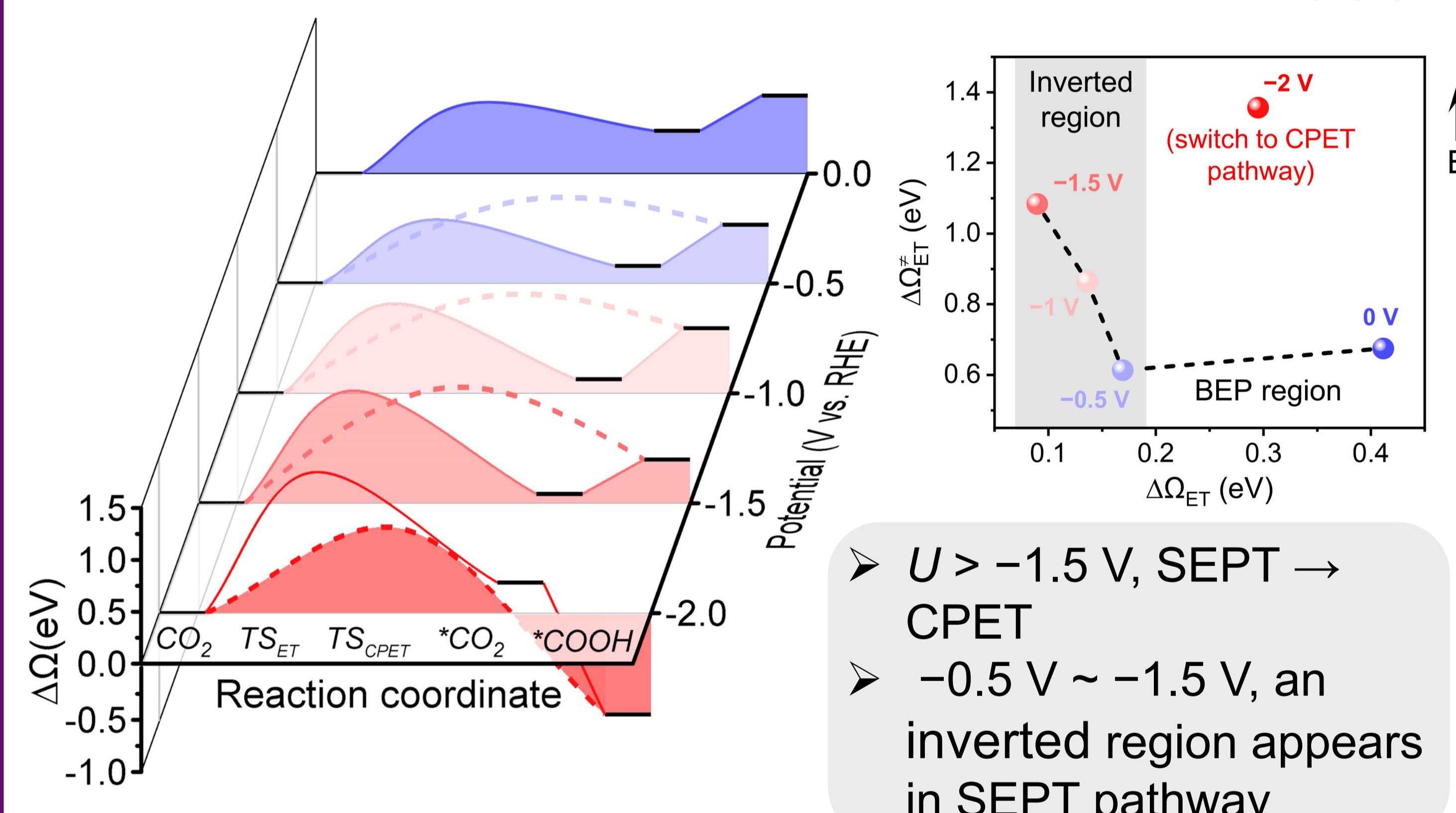


- The CO<sub>2</sub> activation mechanism in eCO<sub>2</sub>RR varies with  $U$
- The barrier of the electron-transfer step in the SEPT mechanism exhibits an inverted region as  $U$  decrease

## Methods

- Grand canonical density functional theory (GC-DFT)
- ASE - Atomic Simulation Environment
- JDFTx - software for joint density-functional theory

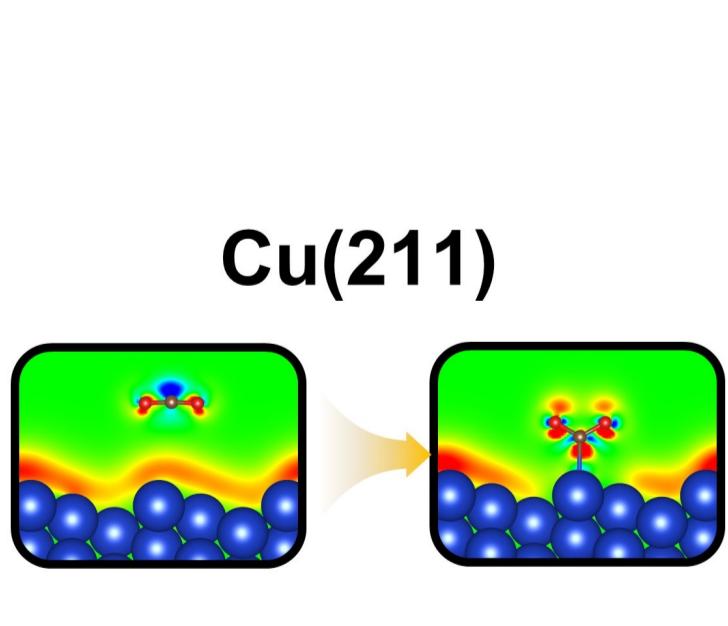
## Results & Discussion



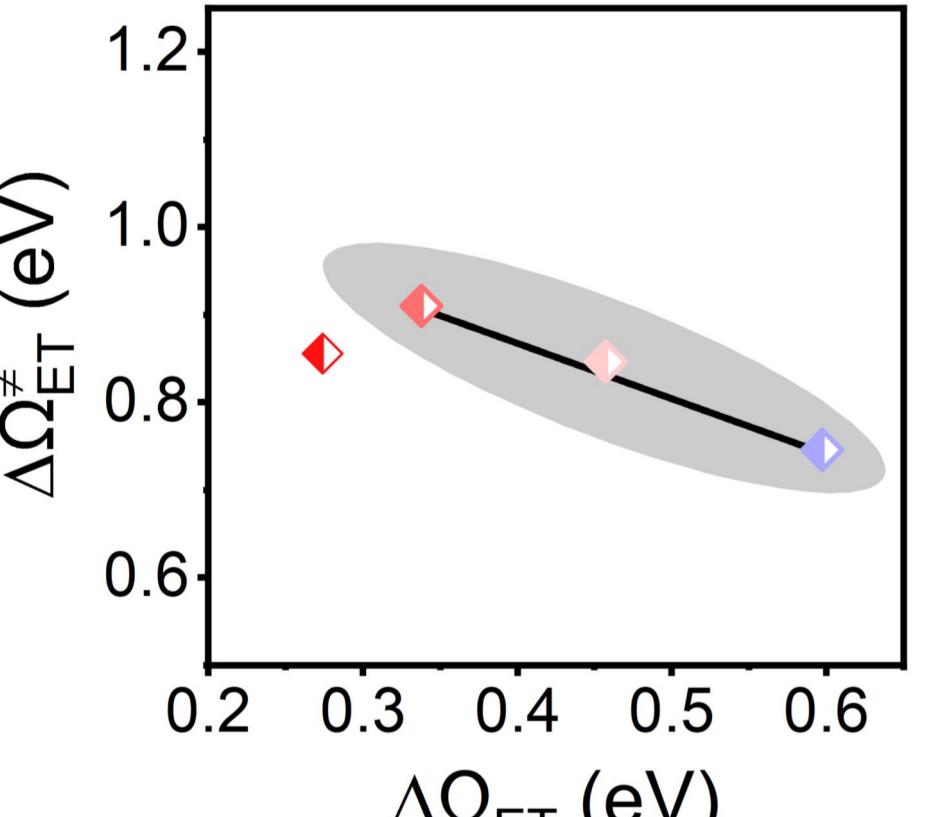
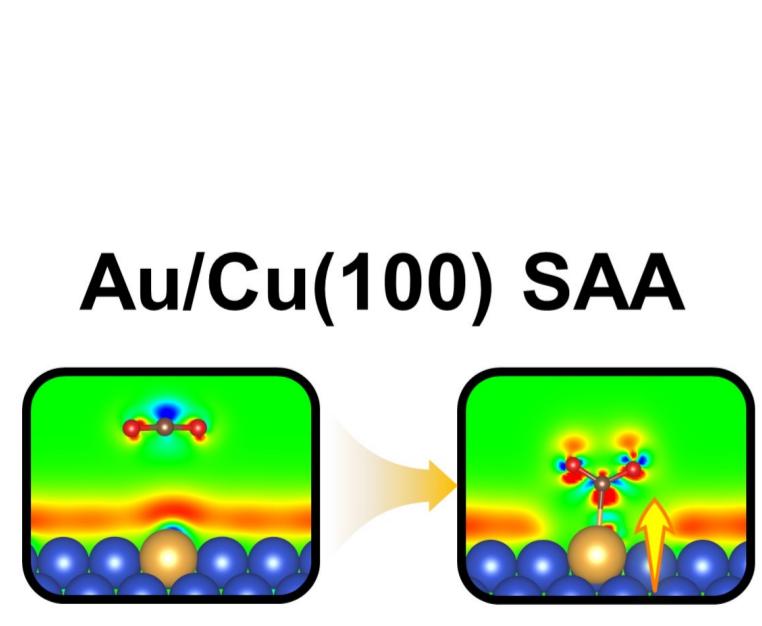
- $U > -1.5$  V, SEPT → CPET
- $-0.5$  V ~  $-1.5$  V, an inverted region appears in SEPT pathway

### Catalyst design

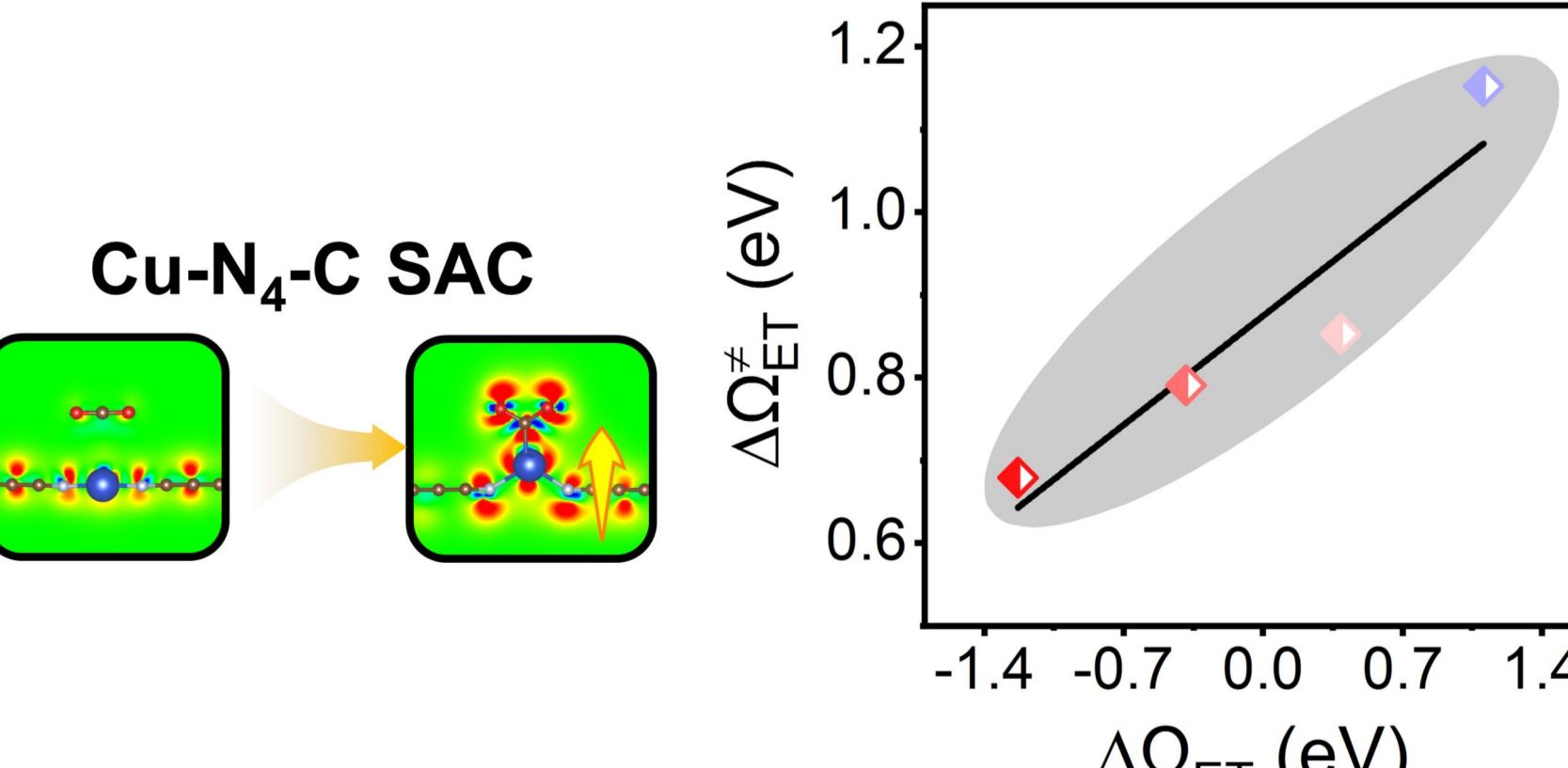
#### 1. Increasing surface roughness



#### 2. Increasing the coordination flexibility of the active sites



#### 3. a substrate with less charging



## Conclusion

- CO<sub>2</sub> activation mechanism in eCO<sub>2</sub>RR: SEPT mechanism at the common working  $U$  → CPET mechanism at the highly negative  $U$
- The inverted region in the SEPT mechanism originates from the rapid rising of Pauli repulsion in the physisorption PEC as  $U$  decrease
- Effective designs of electrocatalysts can suppress the adverse effect of Pauli repulsion on the kinetics of CO<sub>2</sub> activation in eCO<sub>2</sub>RR

1. L. Liu, H. Xiao, Inverted Region in Electrochemical Reduction of CO<sub>2</sub> Induced by Potential-Dependent Pauli Repulsion. *J. Am. Chem. Soc.* **145**, 14267–14275 (2023).

2. M. T. M. Koper et al. Advances and challenges in understanding the electrocatalytic conversion of carbon dioxide to fuels. *Nat. Energy* **4**, 732–745 (2019)