

Part 1: Waves as Particles - The Photon

Aims of Part 1

- To take a brief historical tour of the competing **wave** and **particle** pictures of light.
- Consider evidence from **interference experiments** that light is a **wave**.
- Consider evidence that indicate that light is **not a wave**.
- To introduce the **photon**,
 - and by studying **interference experiments** learn that light can be neither completely **wave-** or **particle-like**.
- And thus see the need for a **new physical theory** - quantum mechanics.

Light: A wave or a particle?

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- *Rømer: 1676*
 - Light has finite speed - but what is moving?

Ole Rømer

Light: A wave or a particle?



Ole Rømer

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 - Light has finite speed - but what is moving?
- *Huygens: 1678*
 - Light is a wave



Christiaan Huygens

Light: A wave or a particle?

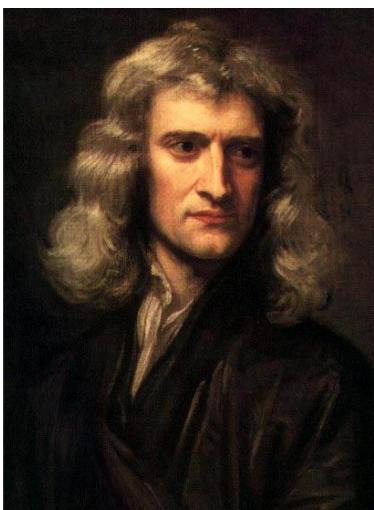


Ole Rømer

- *Rømer: 1676*
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- *Huygens: 1678*
 - Light is a wave
- *Newton: 1704*
 - Light is a particle



Christiaan Huygens



Isaac Newton

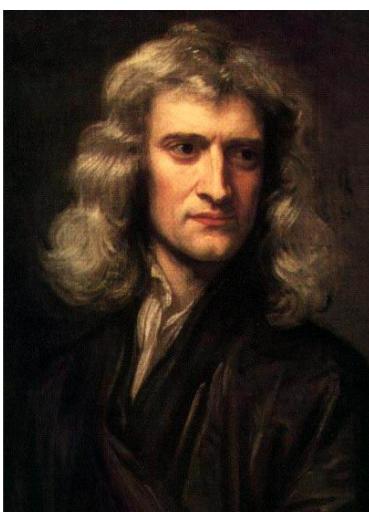
Light: A wave or a particle?



Ole Rømer



Christiaan Huygens



Isaac Newton

- *Rømer: 1676*
 - Light has finite speed - but what is moving?
- *Huygens: 1678*
 - Light is a wave
- *Newton: 1704*
 - Light is a particle
- *Both theories could explain:*
 - Finite light speed
 - Shadows
 - Reflection
 - Refraction

The difference between particles and waves?



The difference between particles and waves?



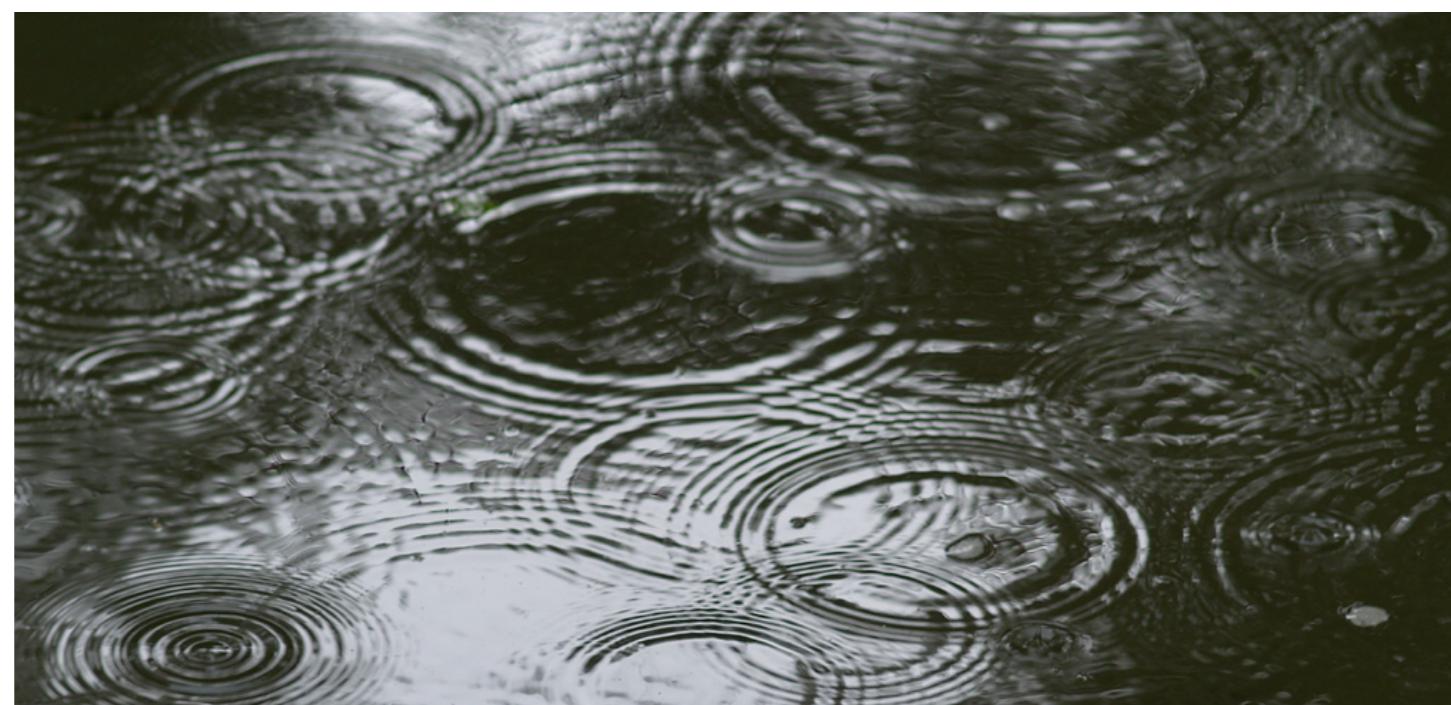
Interference!

The clincher? Young's Double-slit experiment



- 1800s - If light is a wave, **interference effects** will be measurable.
- 1803 - Proposed **Double Slit Experiment**.

Thomas Young



Poll

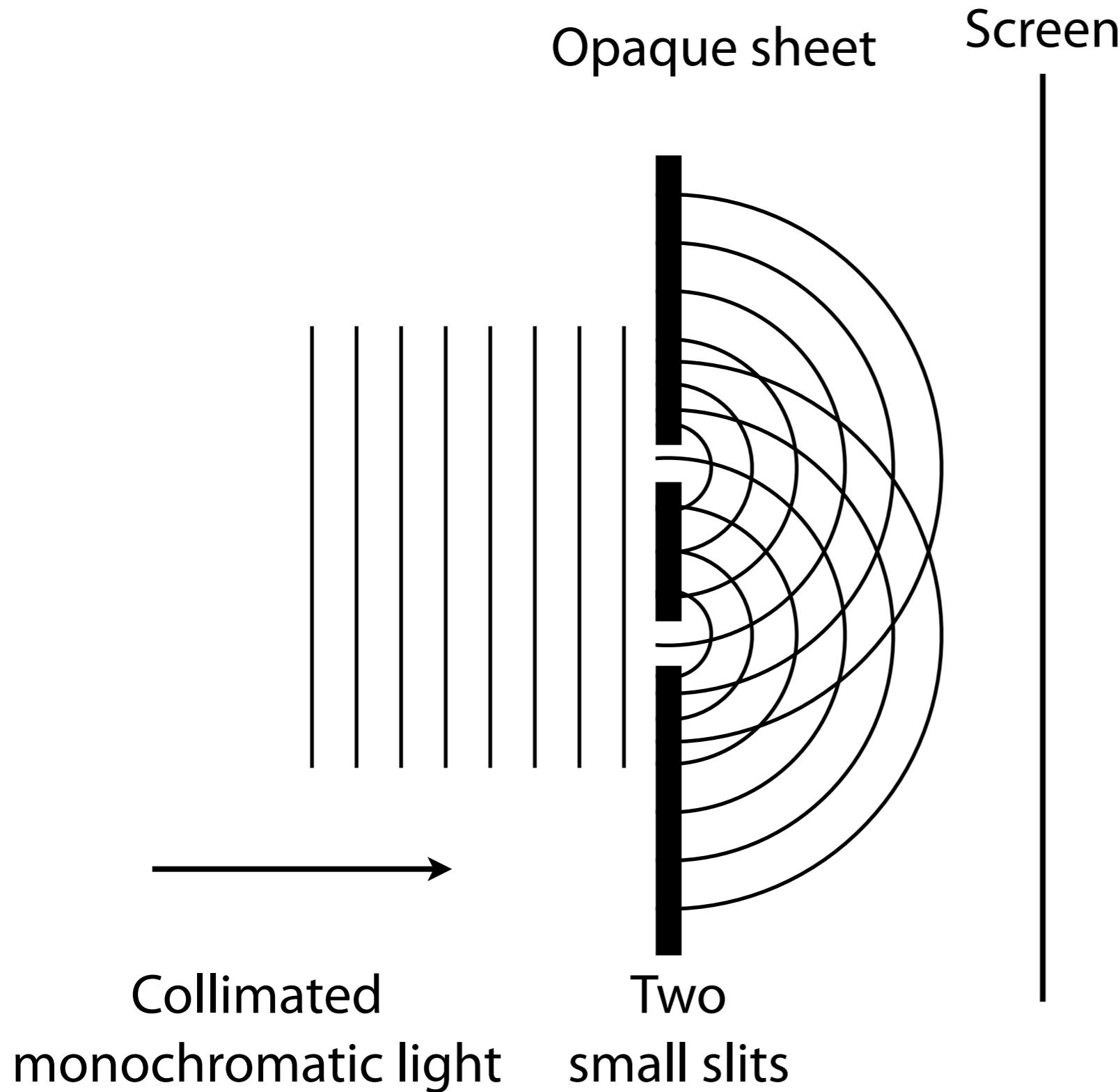
- Have you studied Young's Double-slit experiment before?

Young's Double-slit experiment

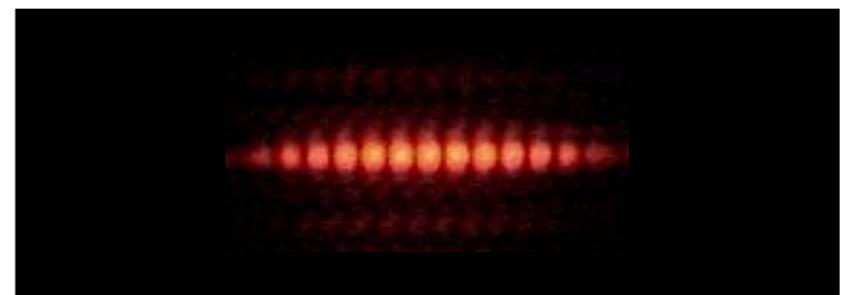
Details: Jewett and Serway, Chapter 37



T. Young



What we observe
on the screen

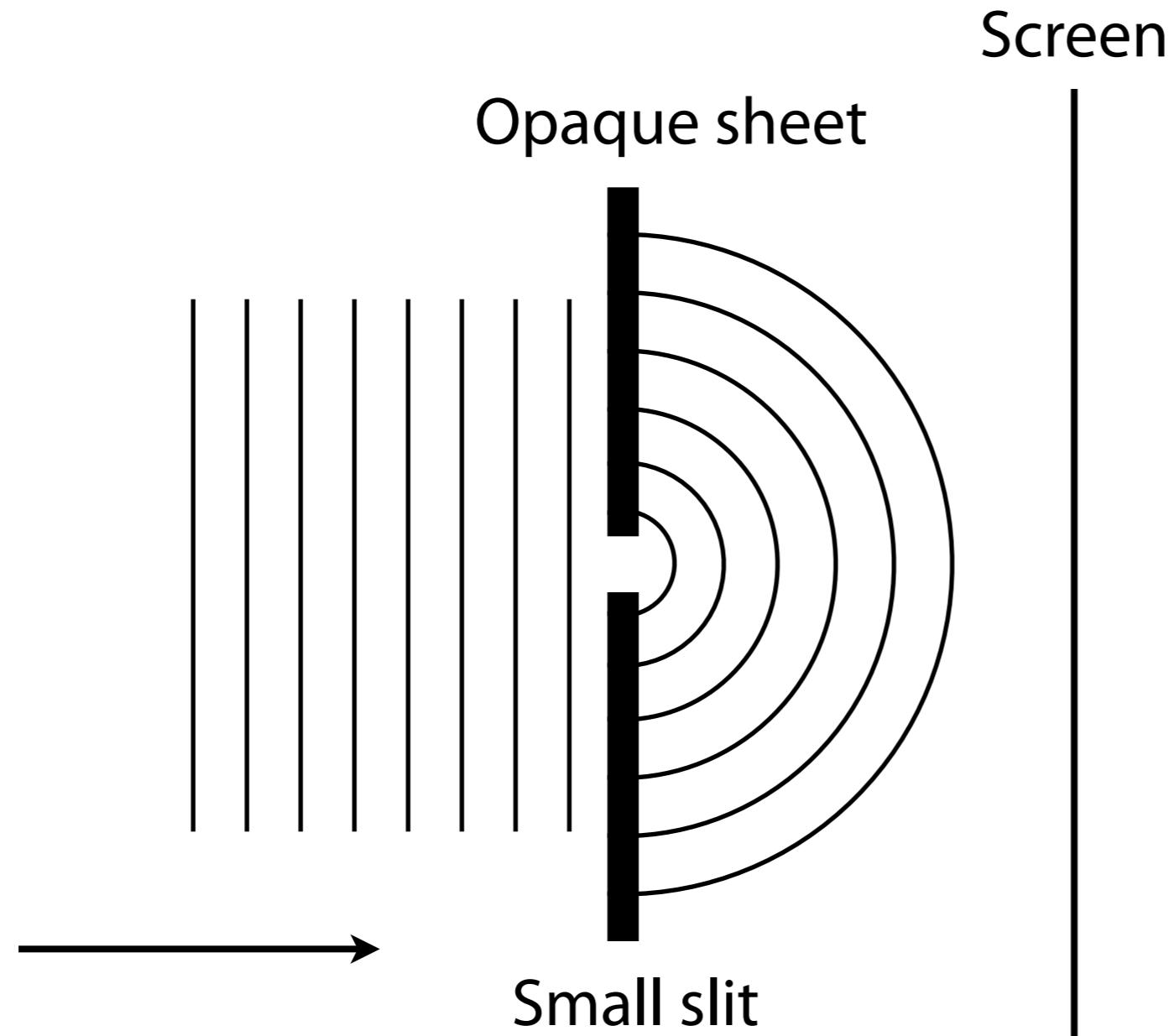


Young's Double-slit experiment

- If we block one slit:



T. Young



What we observe
on the screen



Collimated
monochromatic light

Young's Double-slit experiment



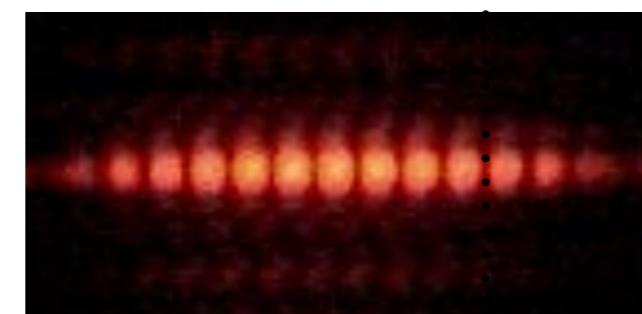
T. Young

- Consider a spot which is **dark** in the two slit pattern.

One slit

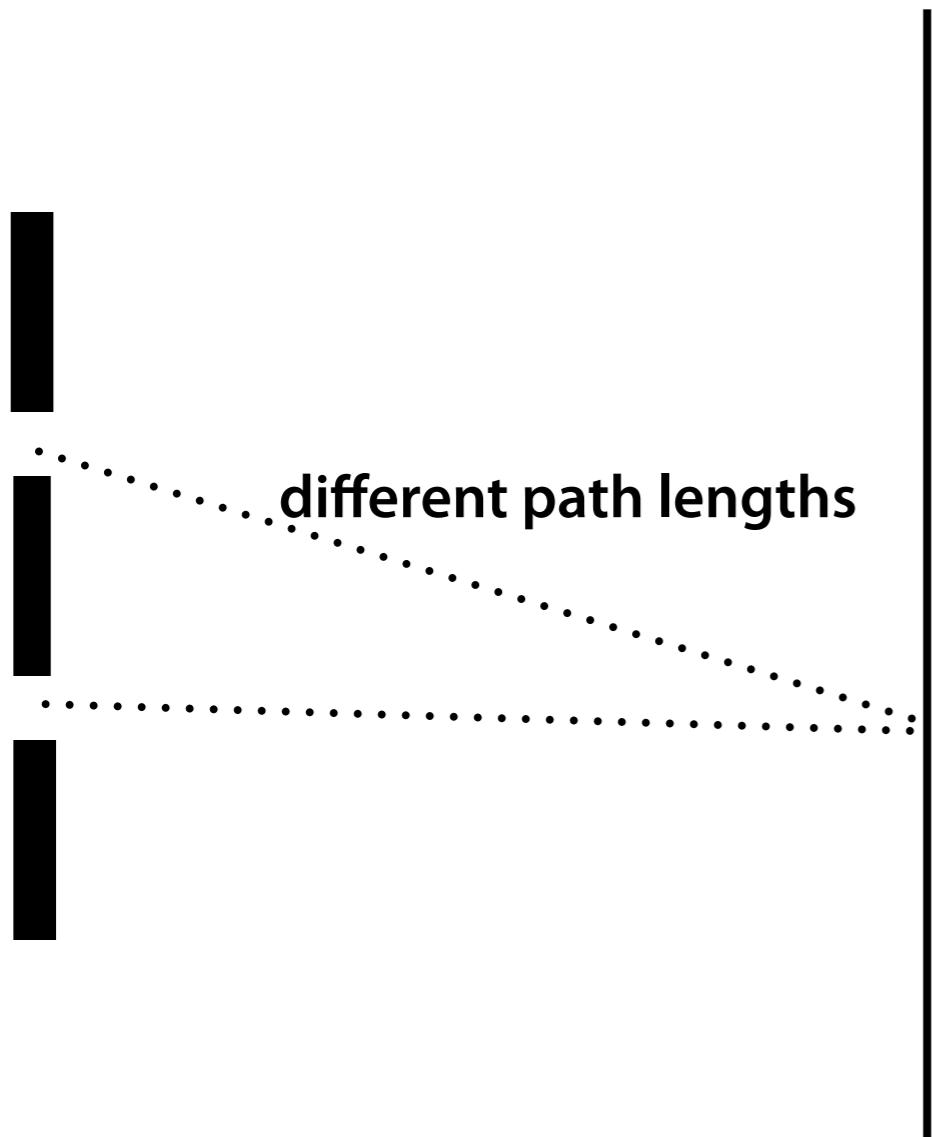


Two slits

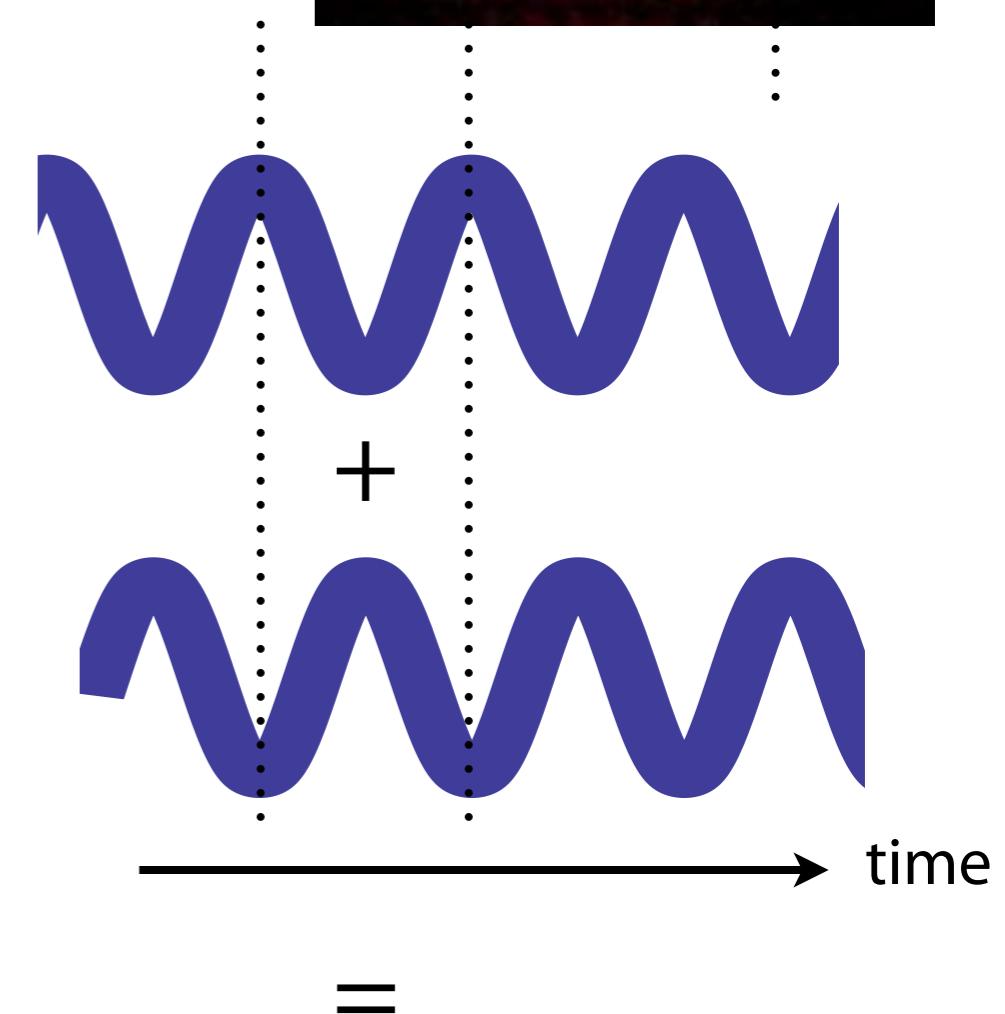


- Note that opening the 2nd slit (providing extra light) has made this point **darker**!

Young's Double-slit experiment



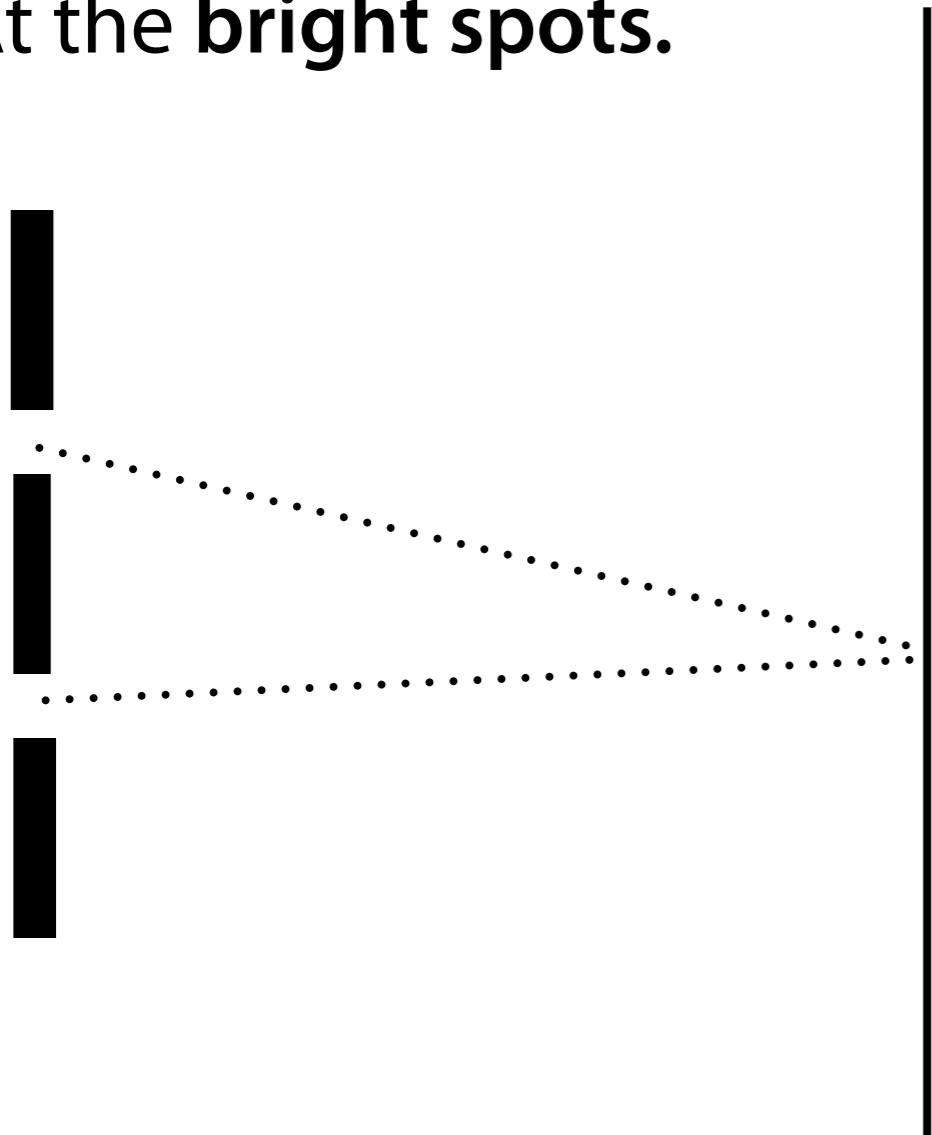
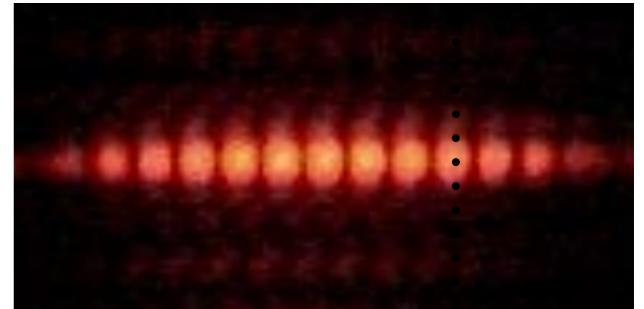
"in-antiphase"



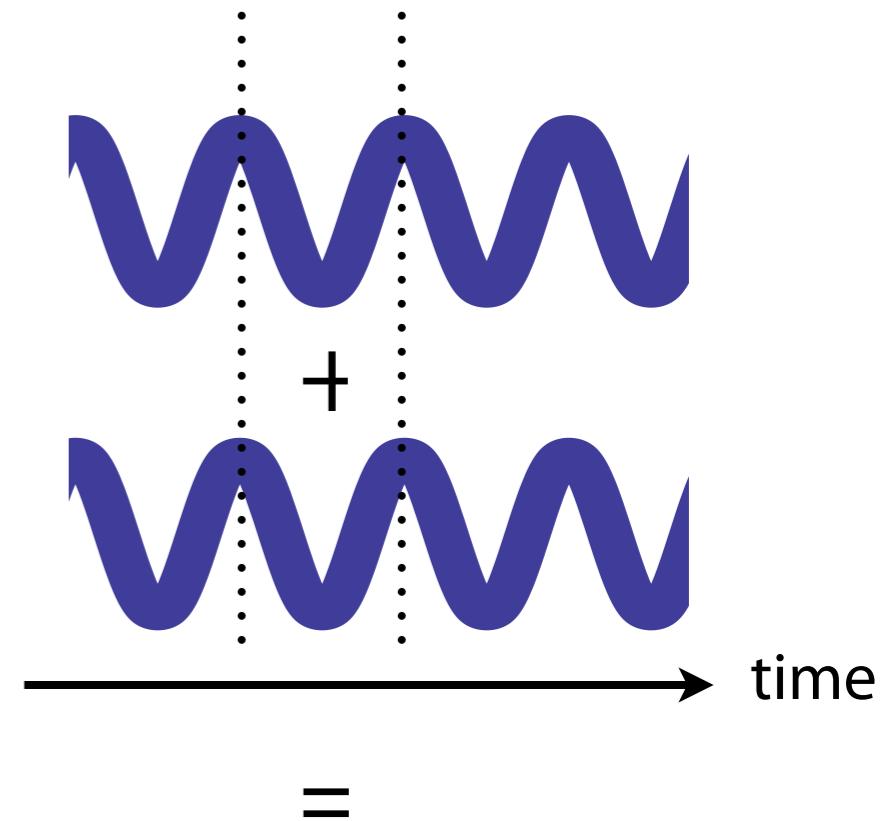
- When the light from the **top slit** has a **peak**, the light from the **bottom slit** has a **trough**.
- The waves **cancel out**.
- We call this **destructive interference**.

Young's Double-slit experiment

At the bright spots.



"in-phase"



- When the light from the **top slit** has a **peak**, the light from the **bottom slit** has a **peak**.
- The waves **add up**.
- We call this **constructive interference**.

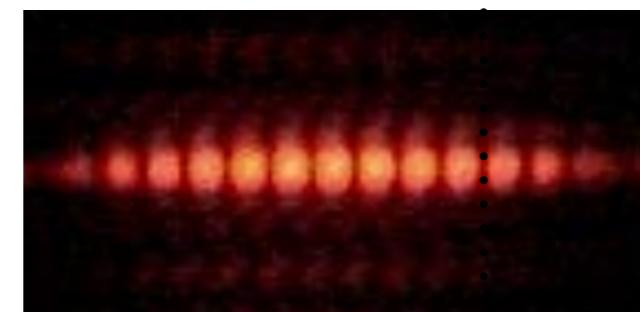
Young's Double-slit experiment

- Strong evidence against Newton's particle model.

One slit



Two slits



- The dark spot has to “know” about both slits.
- Therefore a **particle picture is ruled out**.
- (Since - *surely* particles can *only go through one slit...*)

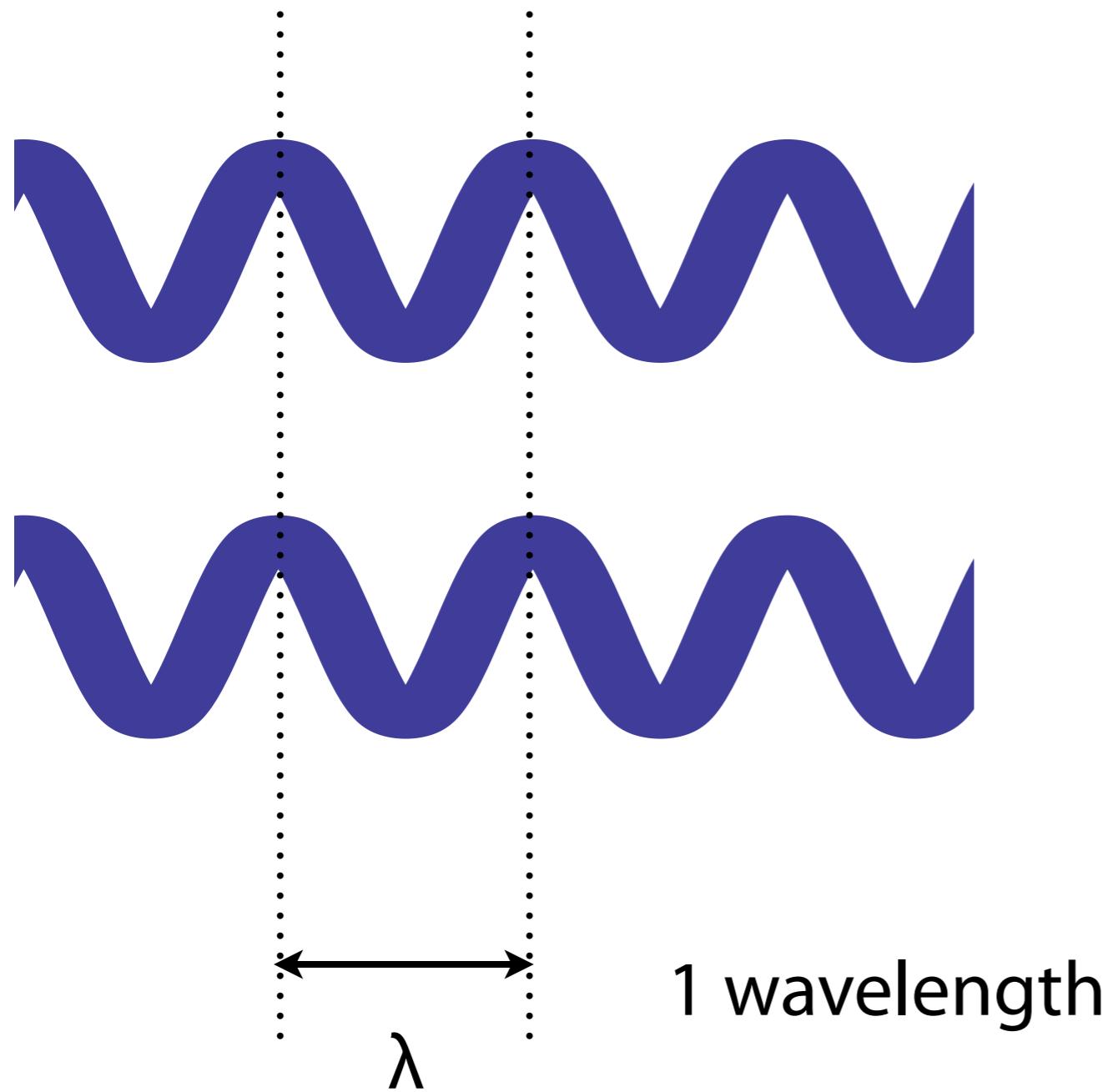
Phase shift

- When a phase difference arises between two waves we say a **phase shift** has occurred.

Phase shift

- Examples

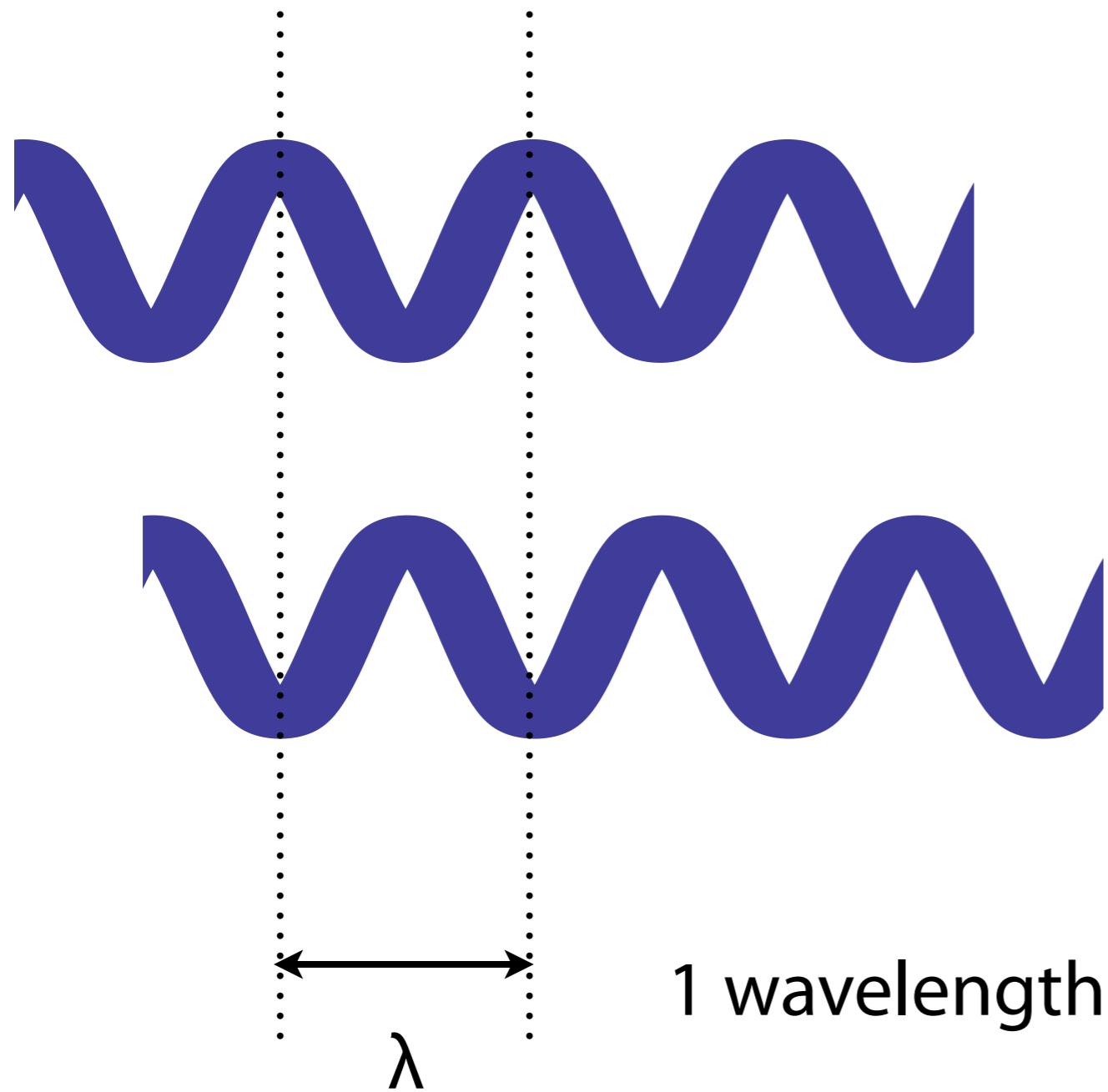
1/2 wavelength phase shift



Phase shift

- Examples

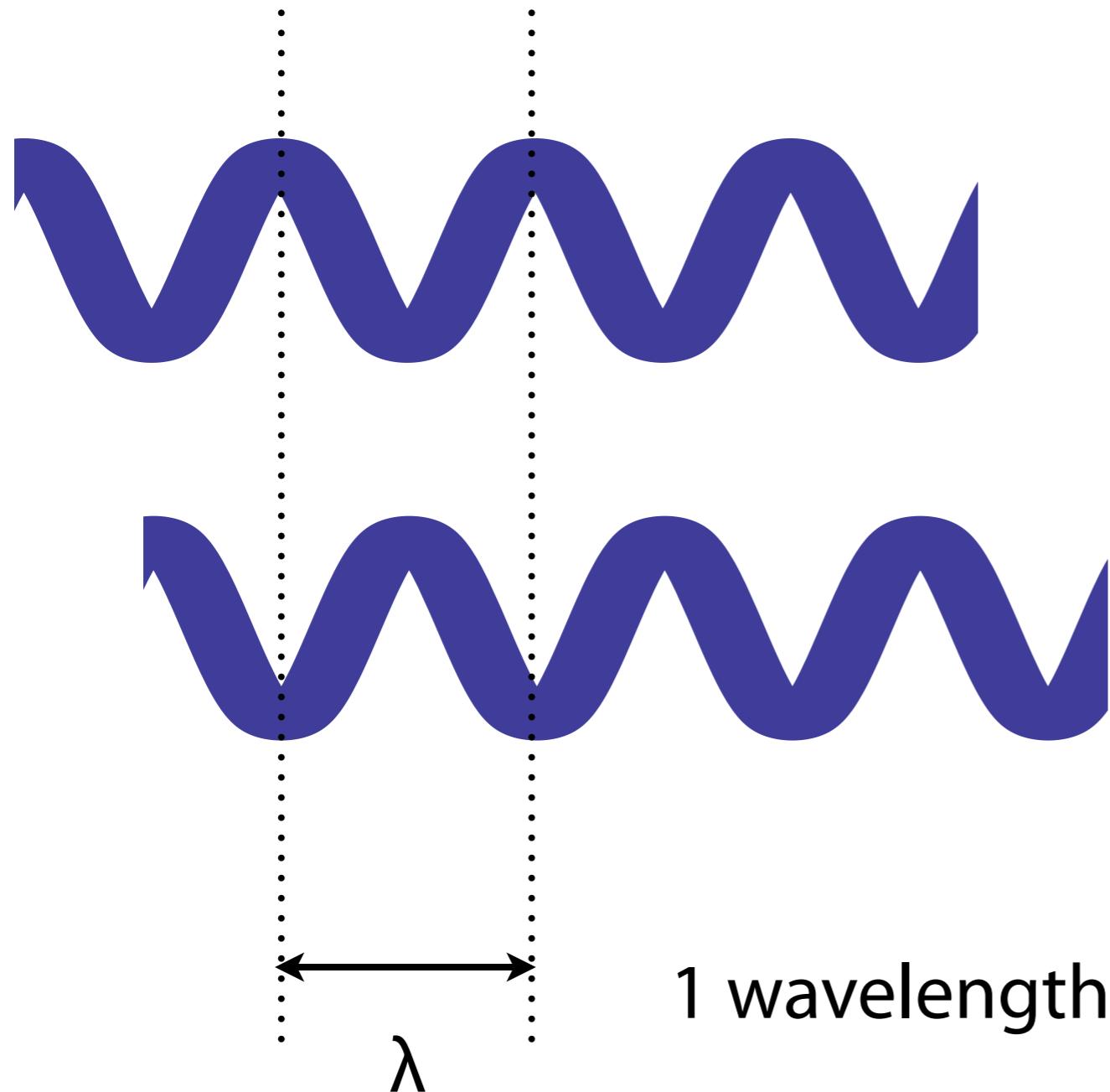
1/2 wavelength phase shift



Phase shift

- Examples

$1/2$ wavelength phase shift

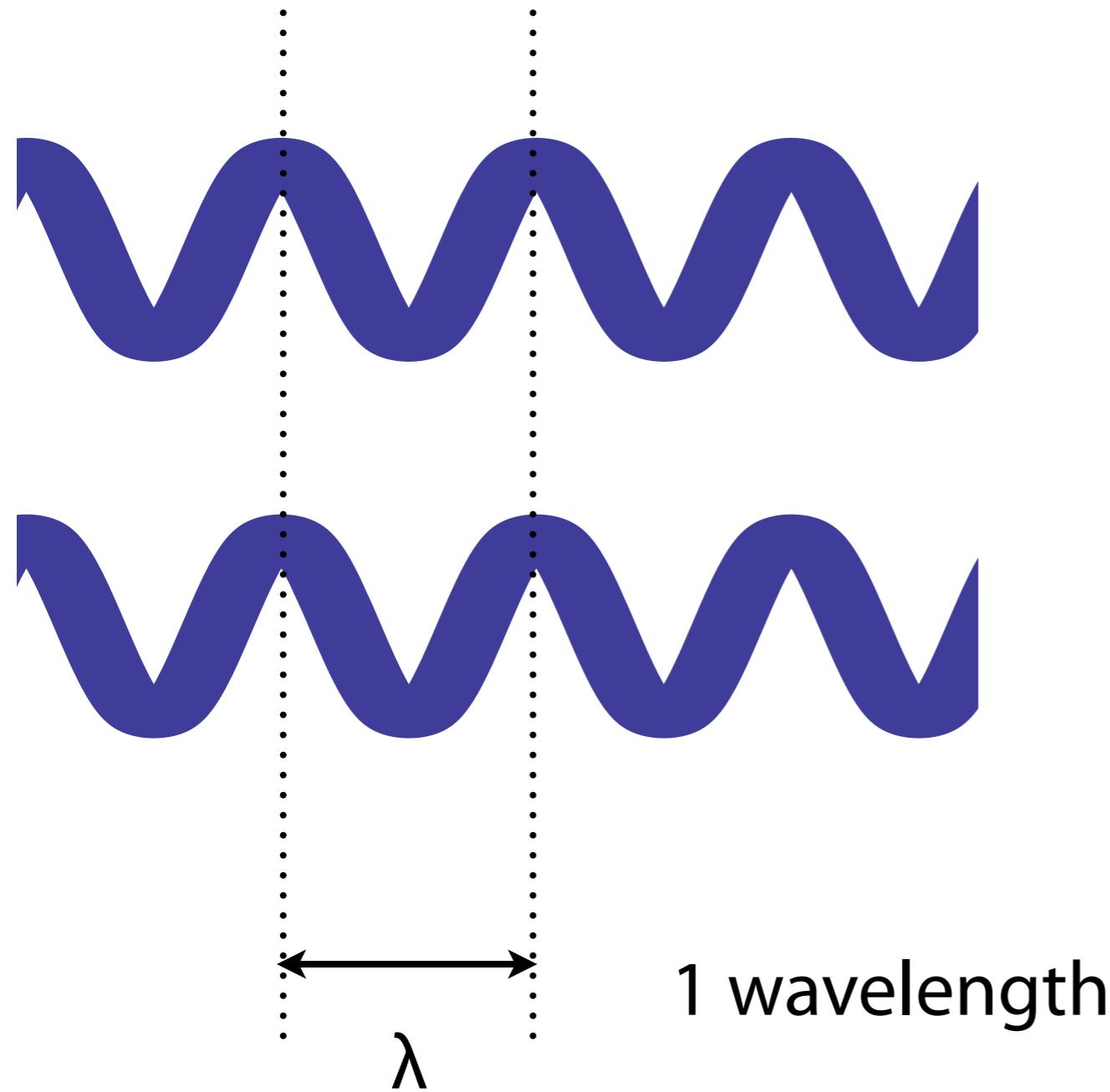


The waves are now **in antiphase** and will destructively interfere.

Phase shift

- Examples

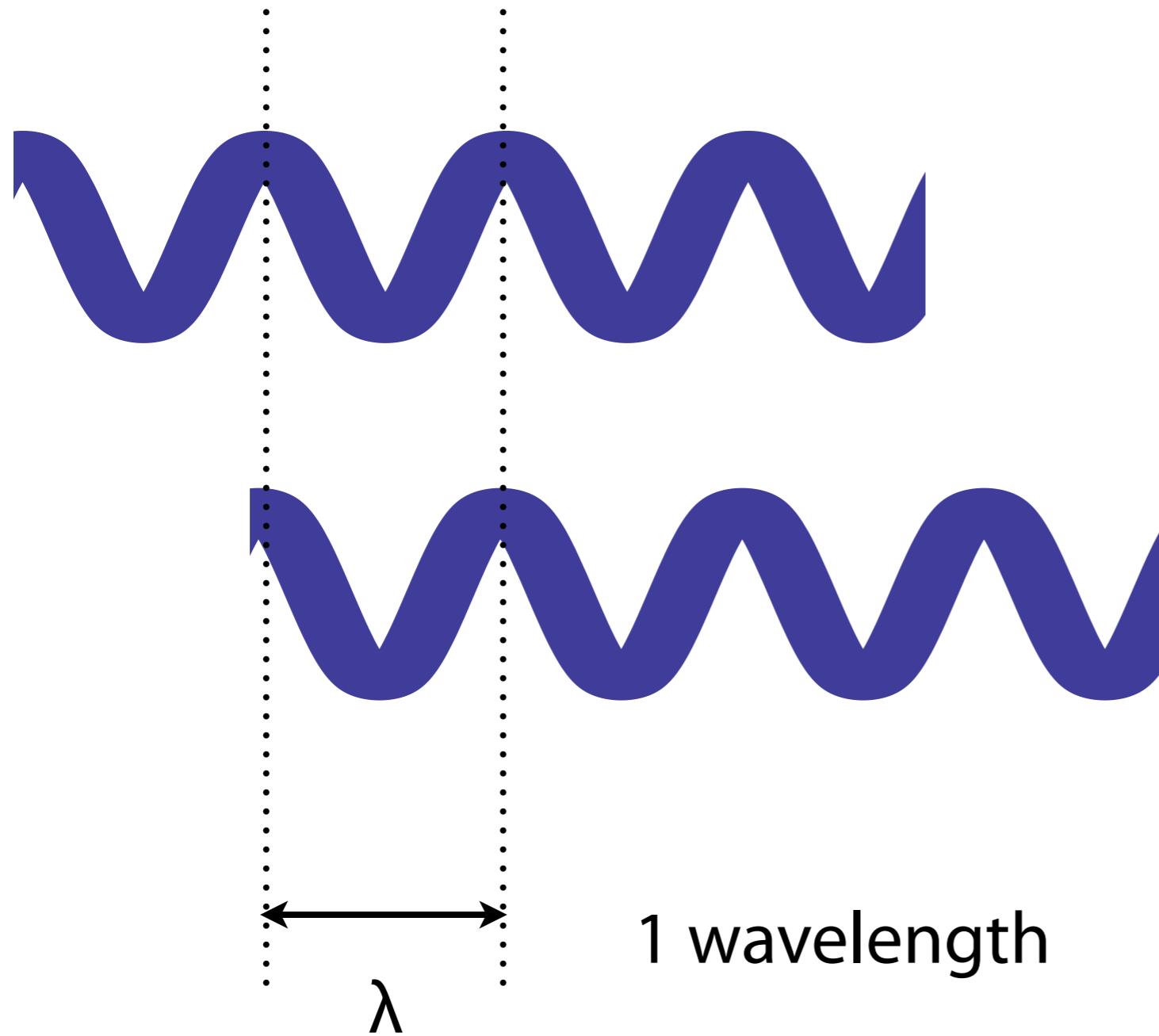
Whole wavelength phase shift



Phase shift

- Examples

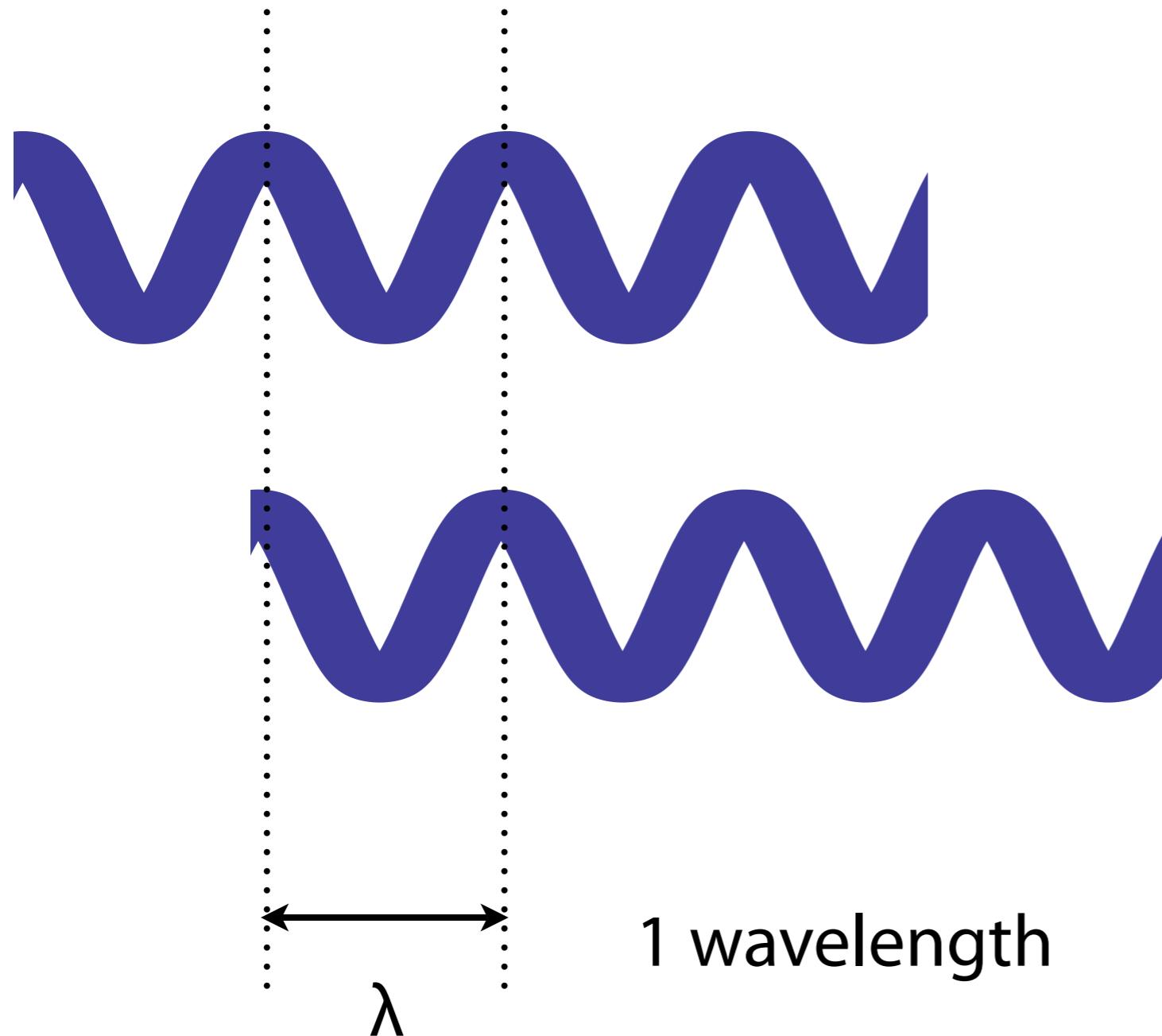
Whole wavelength phase shift



Phase shift

- Examples

Whole wavelength phase shift

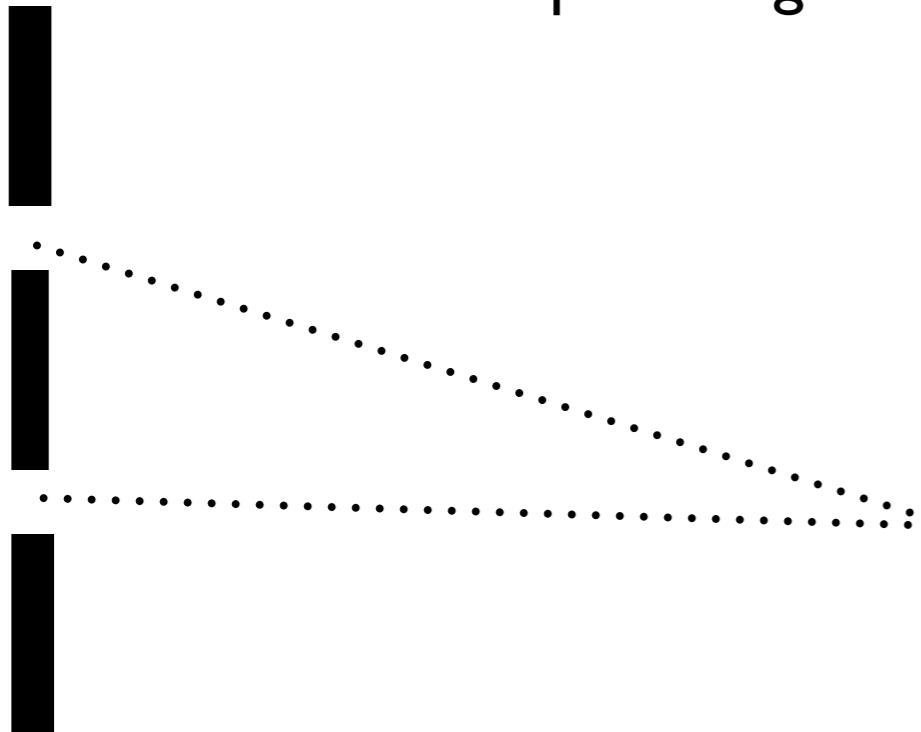


The waves are now **in phase** and will constructively interfere.

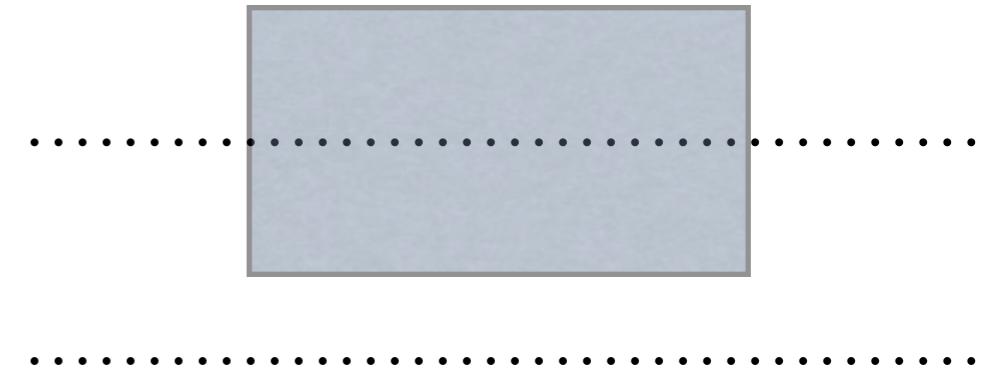
Phase shift

- In optics phase shifts can be caused by different things:

Difference in path length



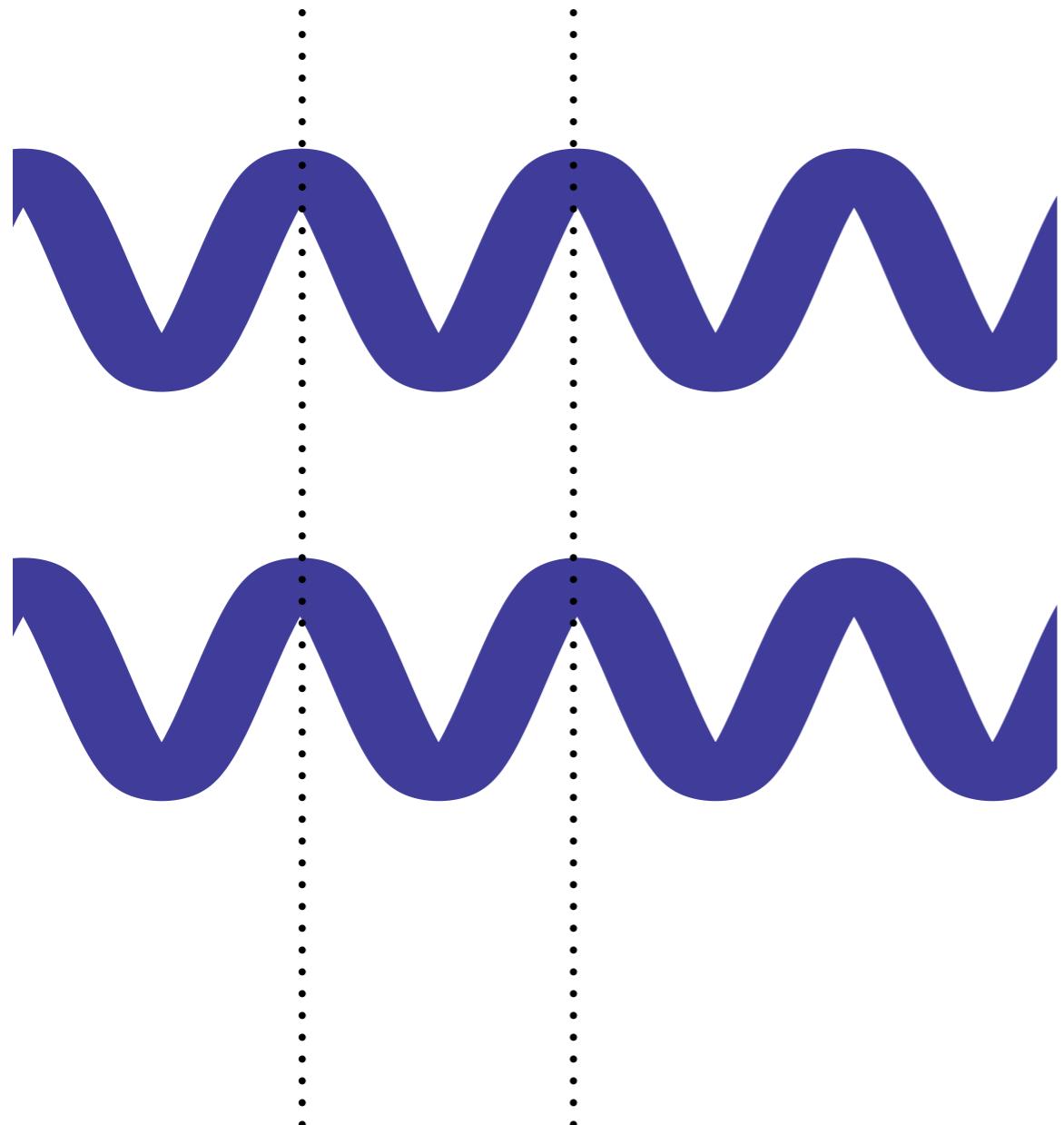
Passing through a crystal



- NB A **phase shift** is a relative property, one beam shifts phase relative to the other.

Quiz

- If I have two beams of light which are initially **in phase** and I then give the second beam a $1/4$ wavelength phase shift, will the beams be:



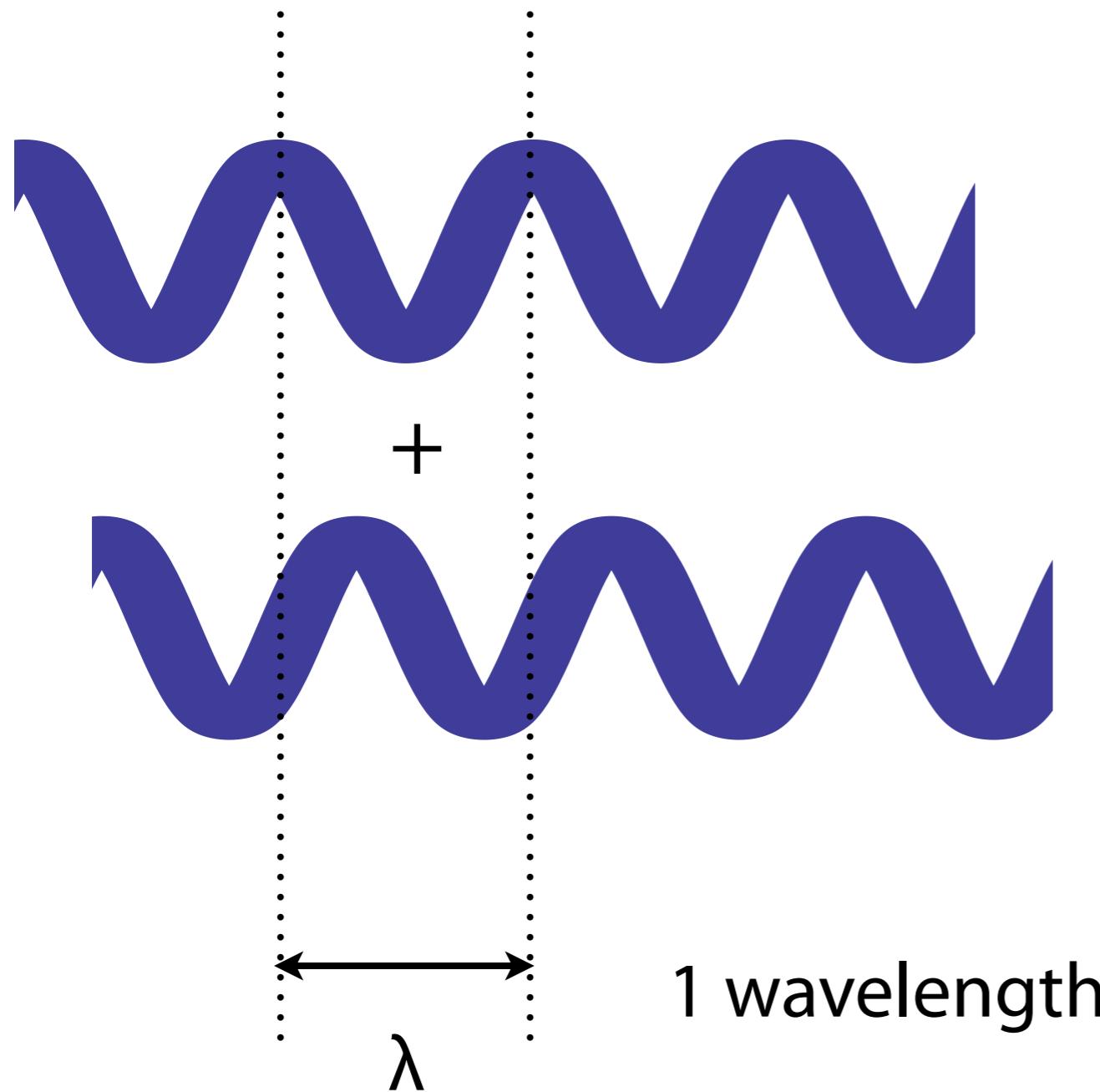
- 1. In phase
- 2. In anti-phase
- 3. Neither in phase or anti-phase.

Phase shift

- Examples

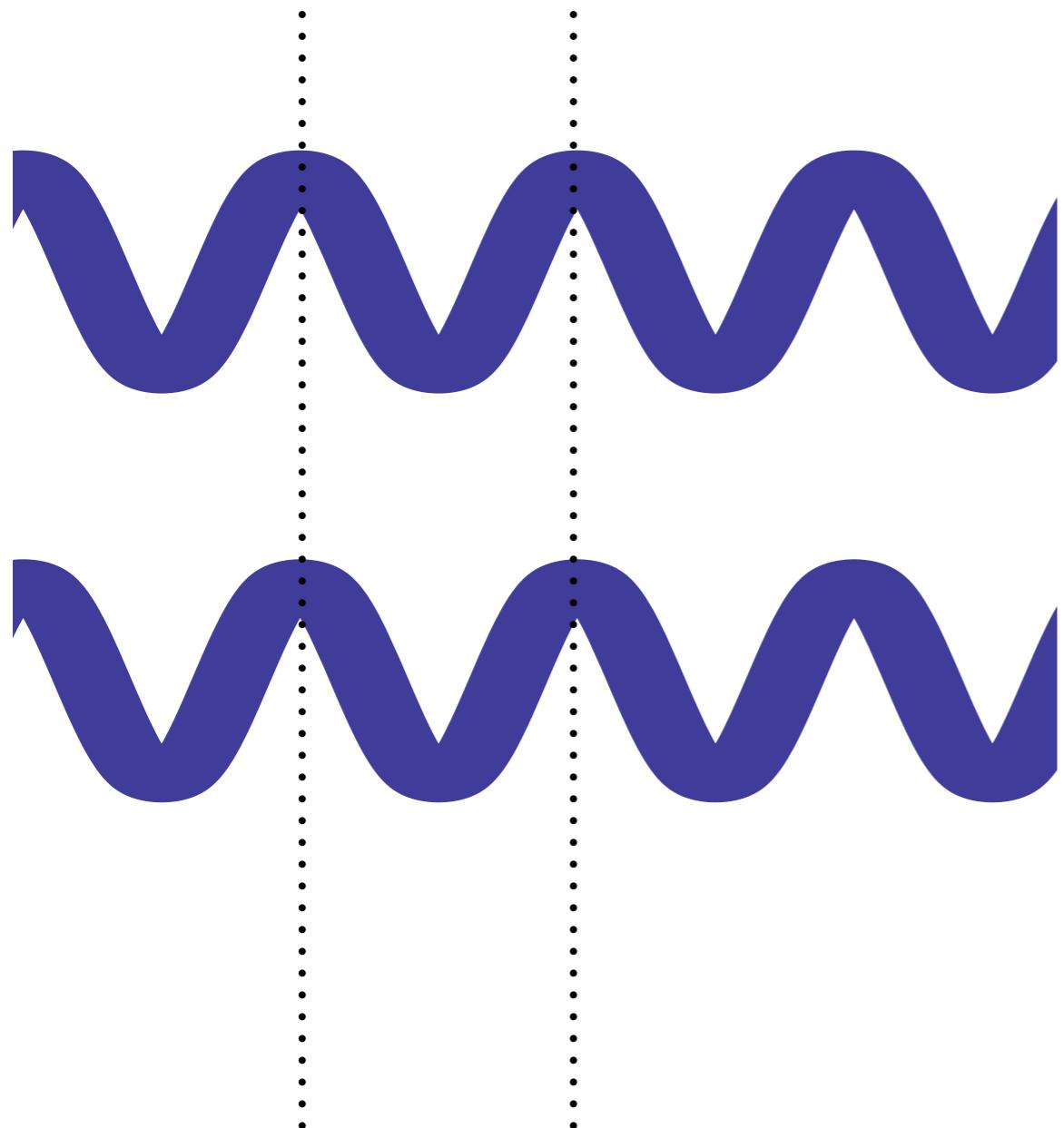
1/4 wavelength phase shift

The waves are neither
in phase or in
antiphase but half-
way in between



Quiz

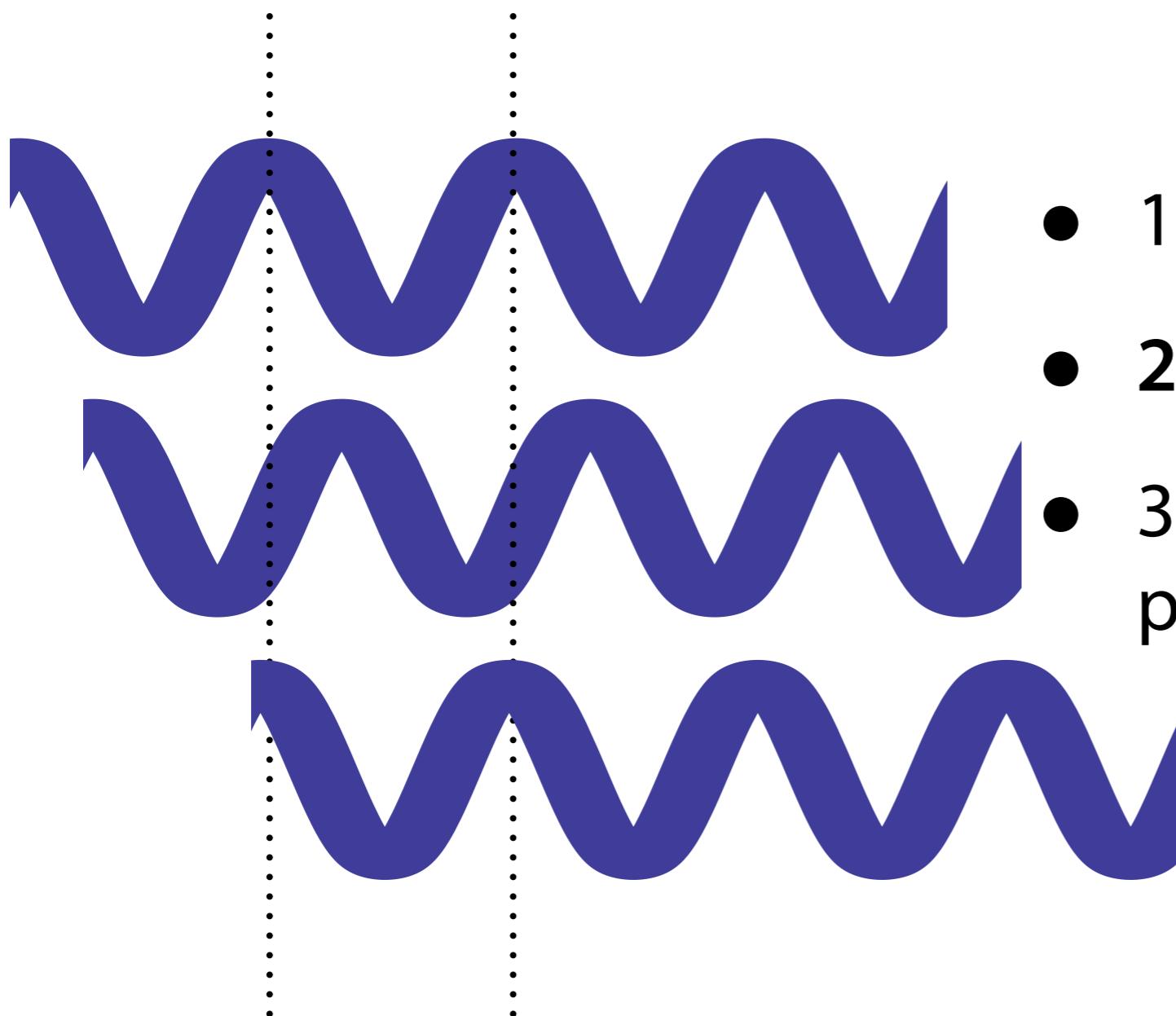
- If I have two beams of light which are initially **in phase** and I then give the second beam a **1/4 wavelength phase** and then give it another **1/4 wavelength phase-shift**, will the beams be:



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Quiz

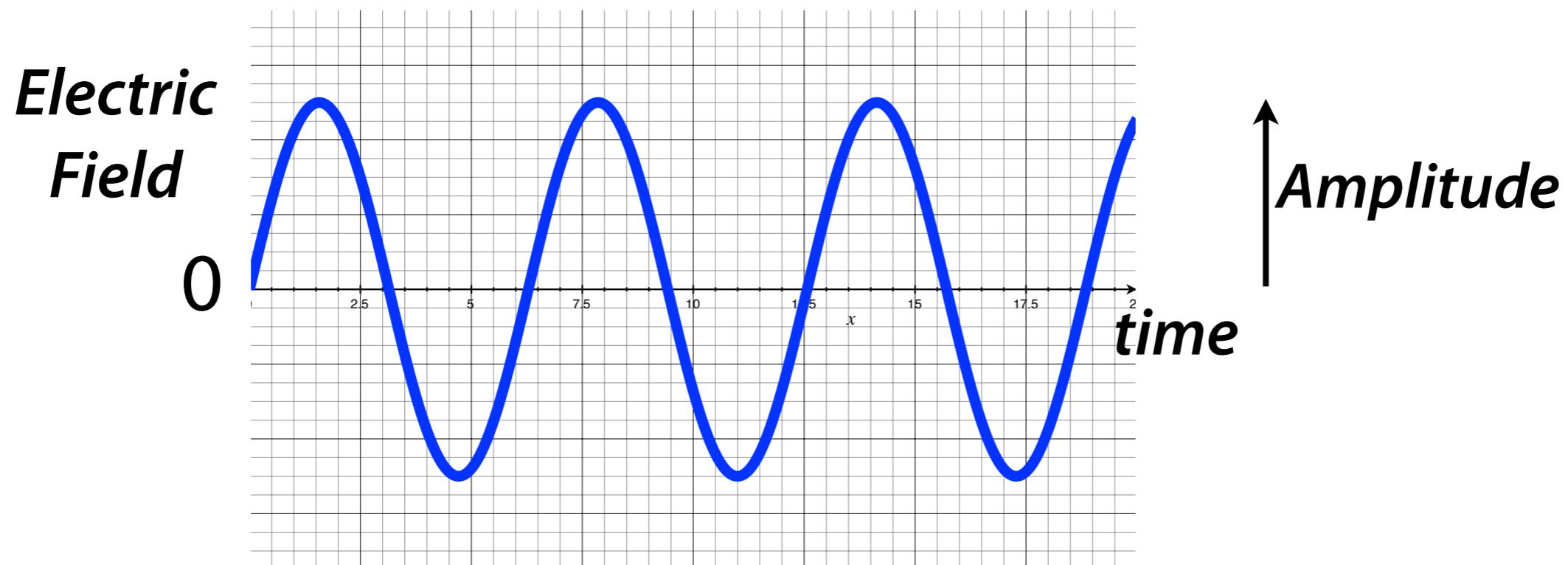
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Aside: Intensity vs Amplitude

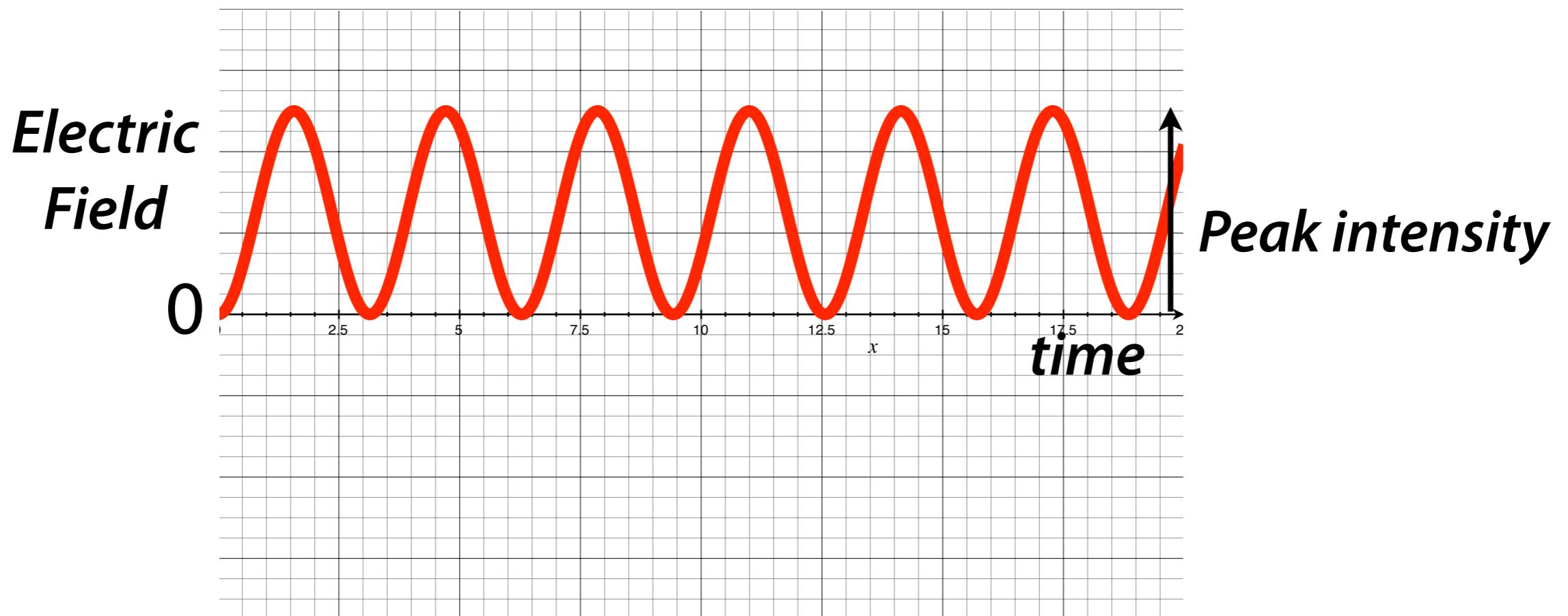
- *(Classical theory of) light*
 - A wave in electric and magnetic fields.



- NB Electric Field is **positive** and **negative**
- The **height** of the wave from 0 is called the **amplitude**.

Aside: Intensity vs Amplitude

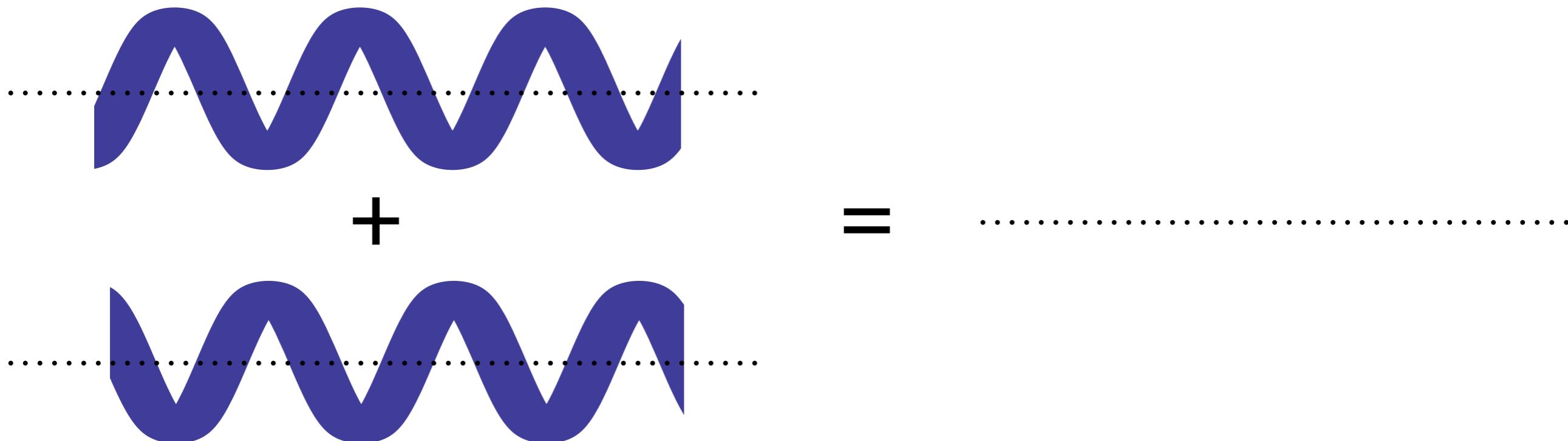
- *(Classical theory of) light*
 - Intensity is the amplitude squared.



- It is always **never negative than zero**.
- The intensity (power) of light we see is the average over time.

Aside: Intensity vs Amplitude

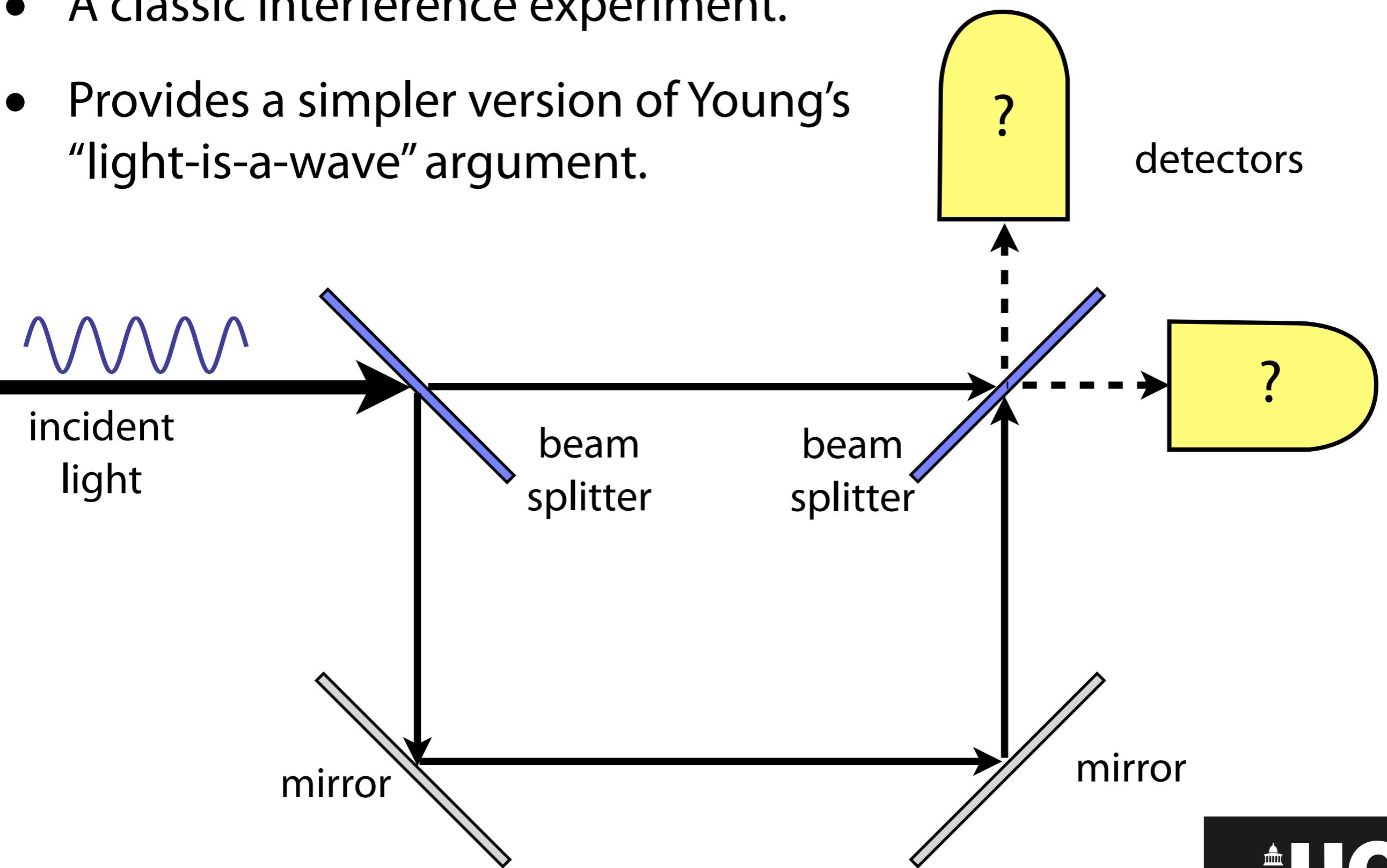
- *(Classical theory of) light*
 - Interference happens on the level of amplitudes.



- The brightness we observe is the resulting **intensity**.

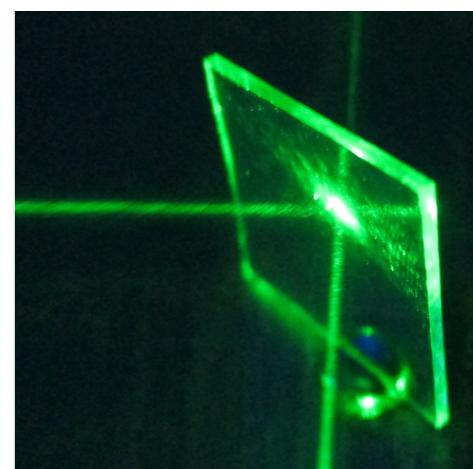
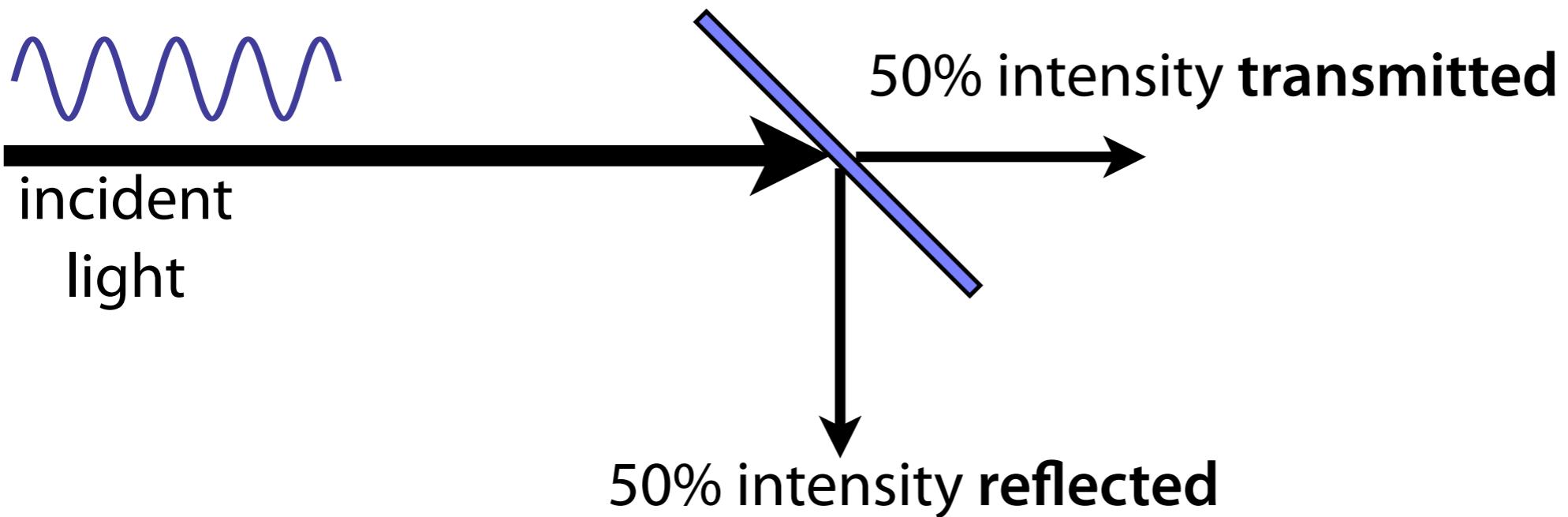
The Mach-Zehnder interferometer

- A classic interference experiment.
- Provides a simpler version of Young's "light-is-a-wave" argument.



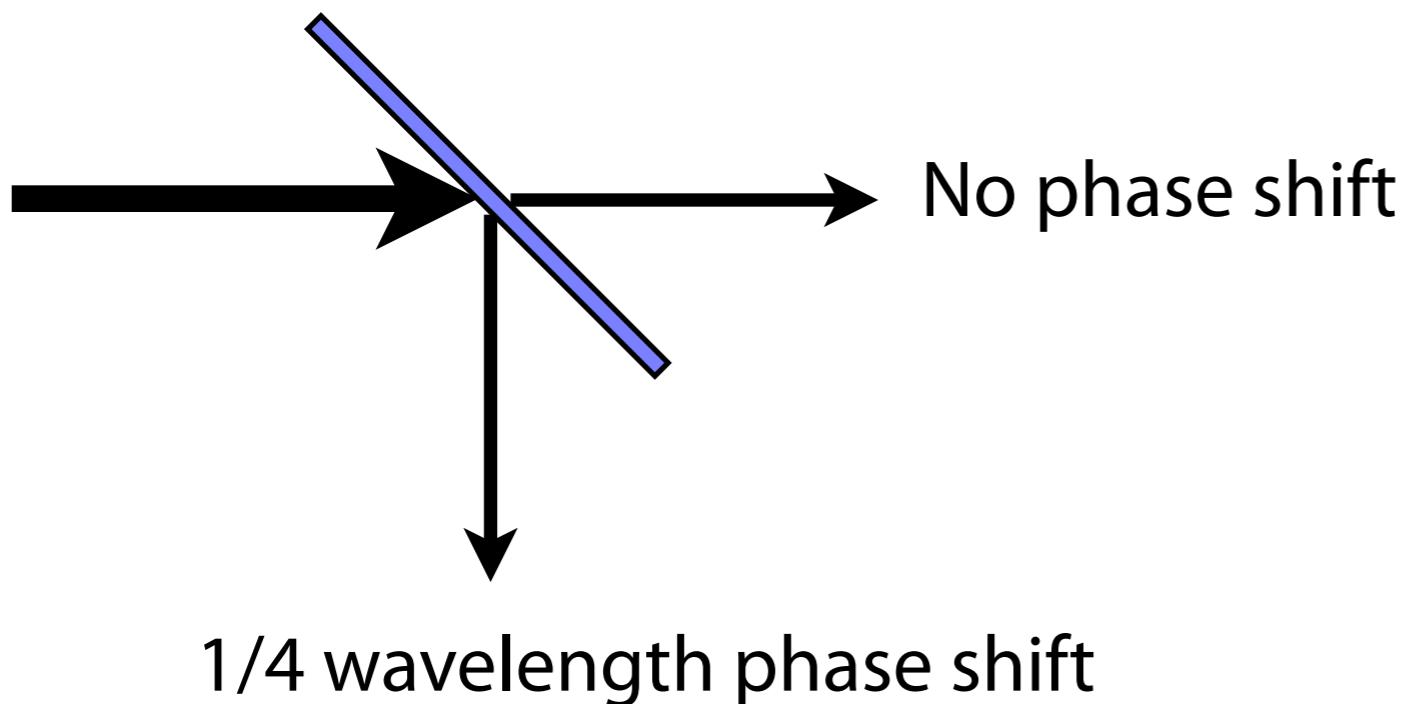
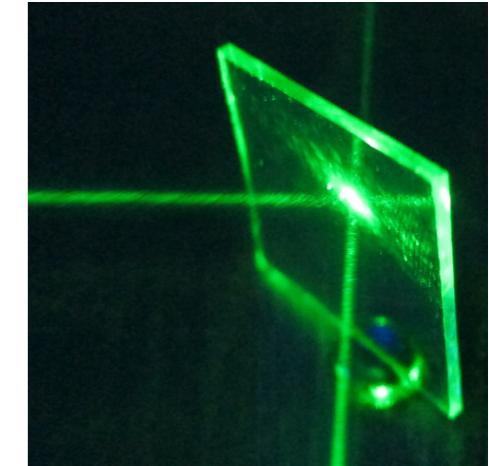
Mach-Zehnder Interferometer

- Beam splitter - A “semi-silvered mirror”
 - Most reflecting surfaces reflect and transmit.
 - A beam-splitter is specially tuned mirror which transmits 50% of light (**intensity**) and reflects 50% of light (**intensity**).



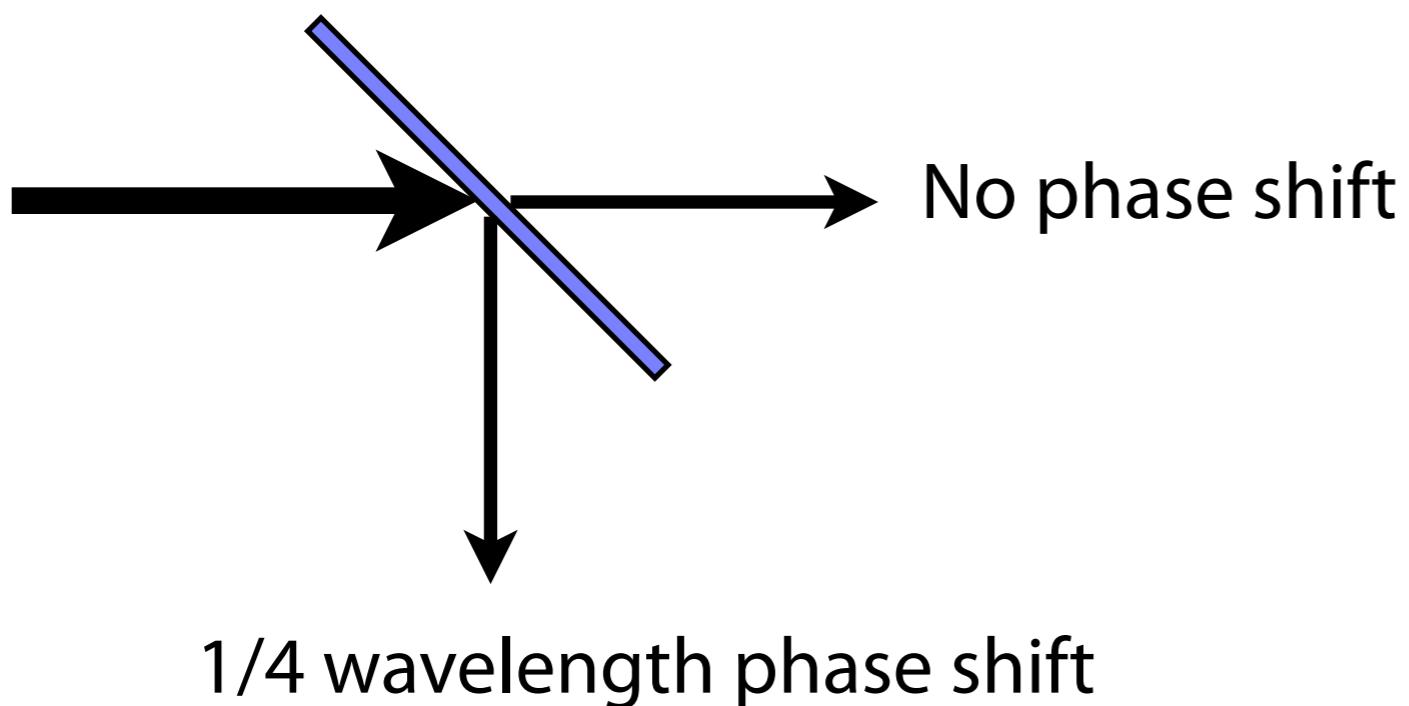
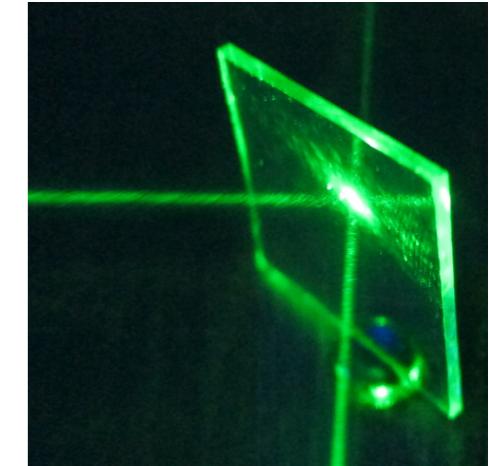
Mach-Zehnder Interferometer

- Beam splitter phase shift
 - Reflection on a beam splitter can cause a **phase shift** (necessary for energy conservation).
- Symmetric beam splitter
 - Transmission causes **no phase shift**
 - Reflection causes a **1/4 wavelength phase shift**
 - In this course we will **always** consider symmetric beam splitters.



Mach-Zehnder Interferometer

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Mach-Zehnder Interferometer

- *Other components of the M-Z interferometer*

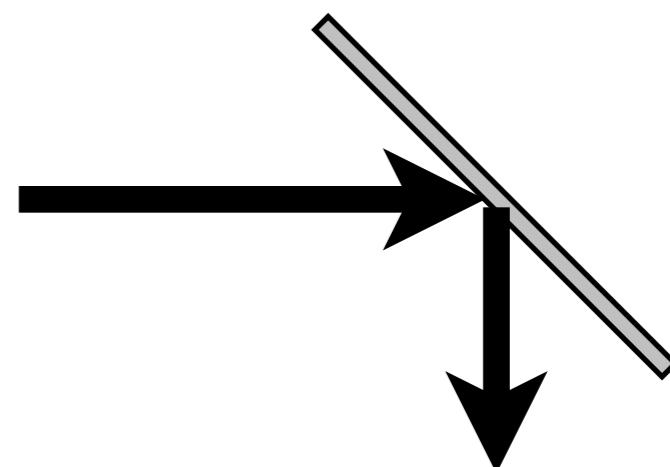
- **Detector**

- Indicates the intensity (power) of incident light.
- Units: Watts = Joules / second



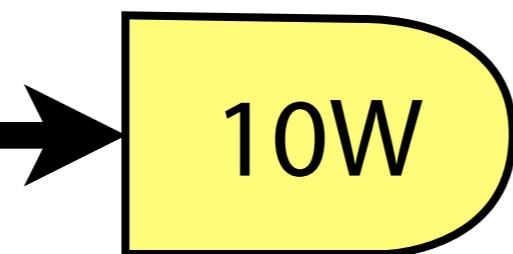
- **Full Mirror**

- No transmission
- No phase shift on reflection.

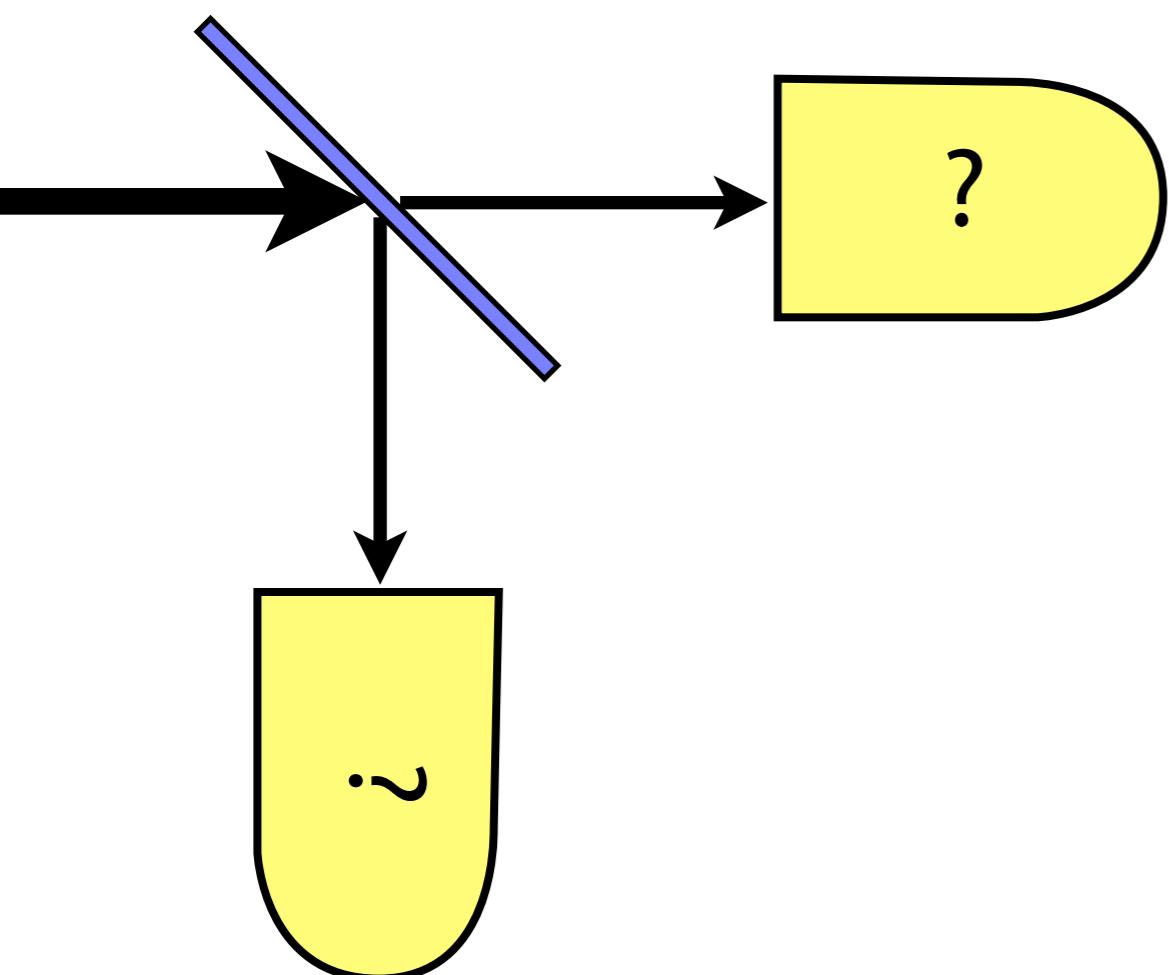


Quiz

- Consider a laser beam with power 10W.



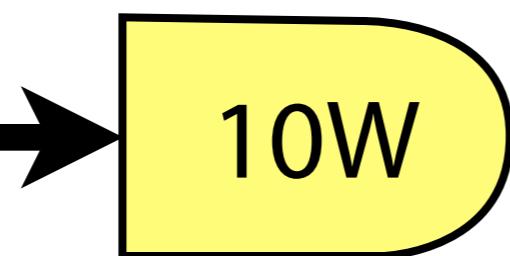
- We split this beam at a beam splitter. What will the detectors say?



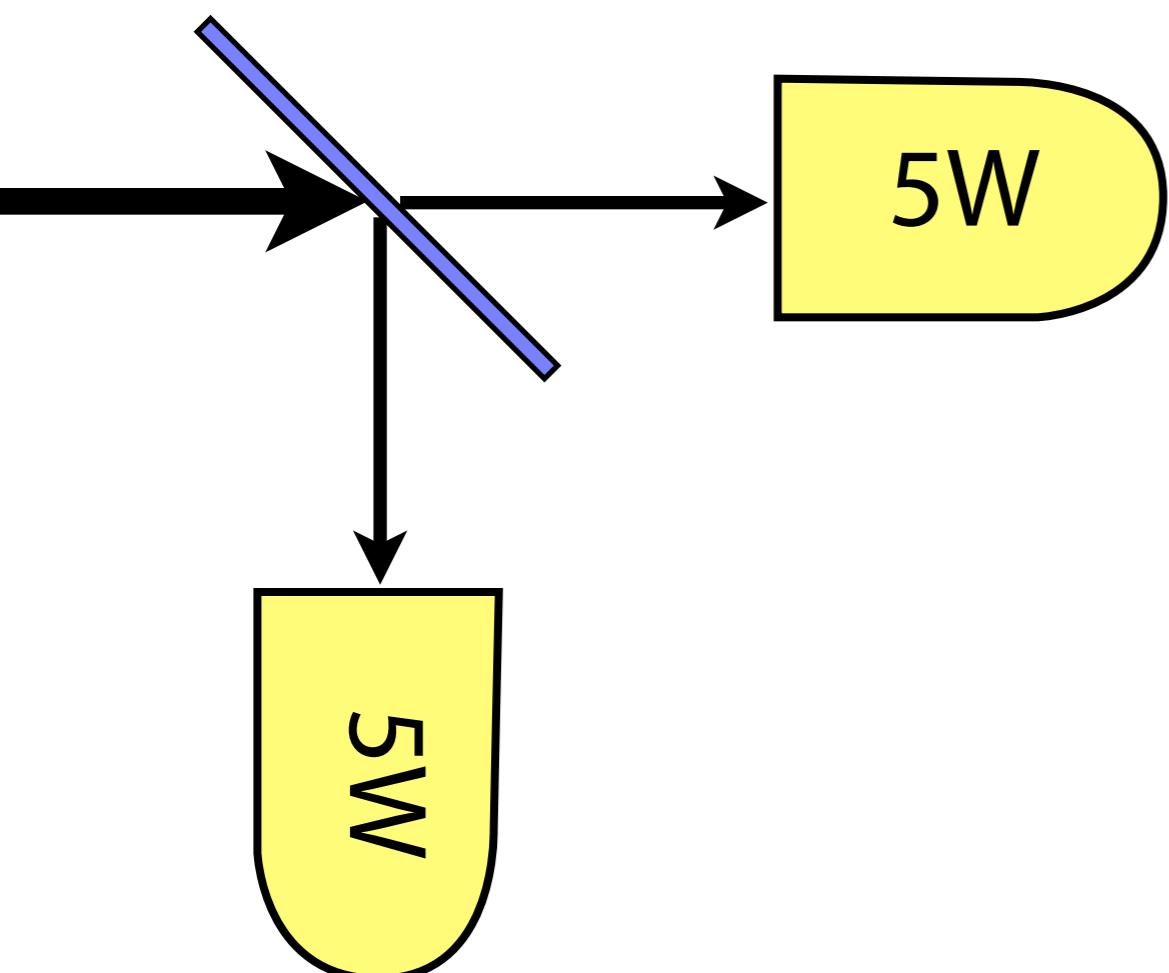
- 1. 10W and 10W
- 2. 10W and 0 W
- 3. 5 W and 5 W

Quiz

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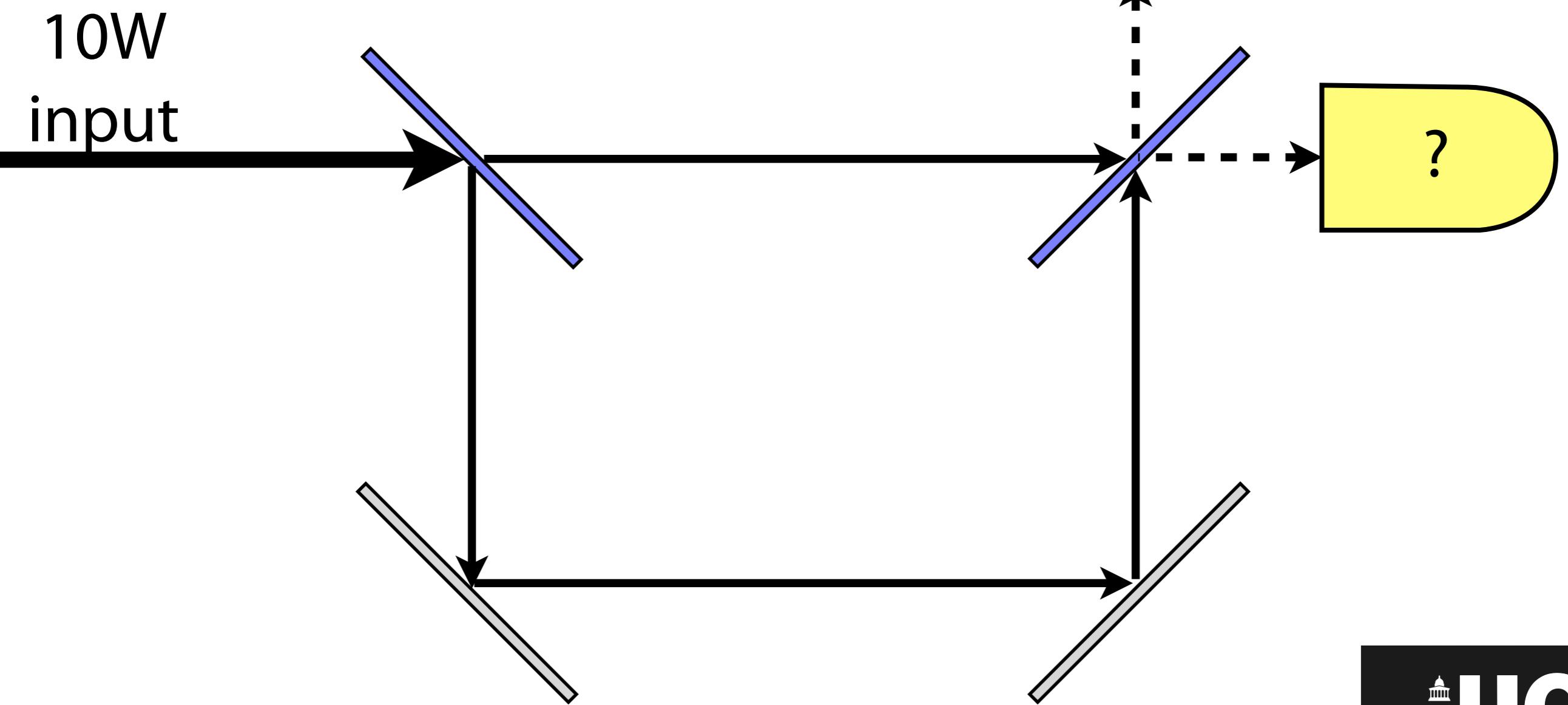
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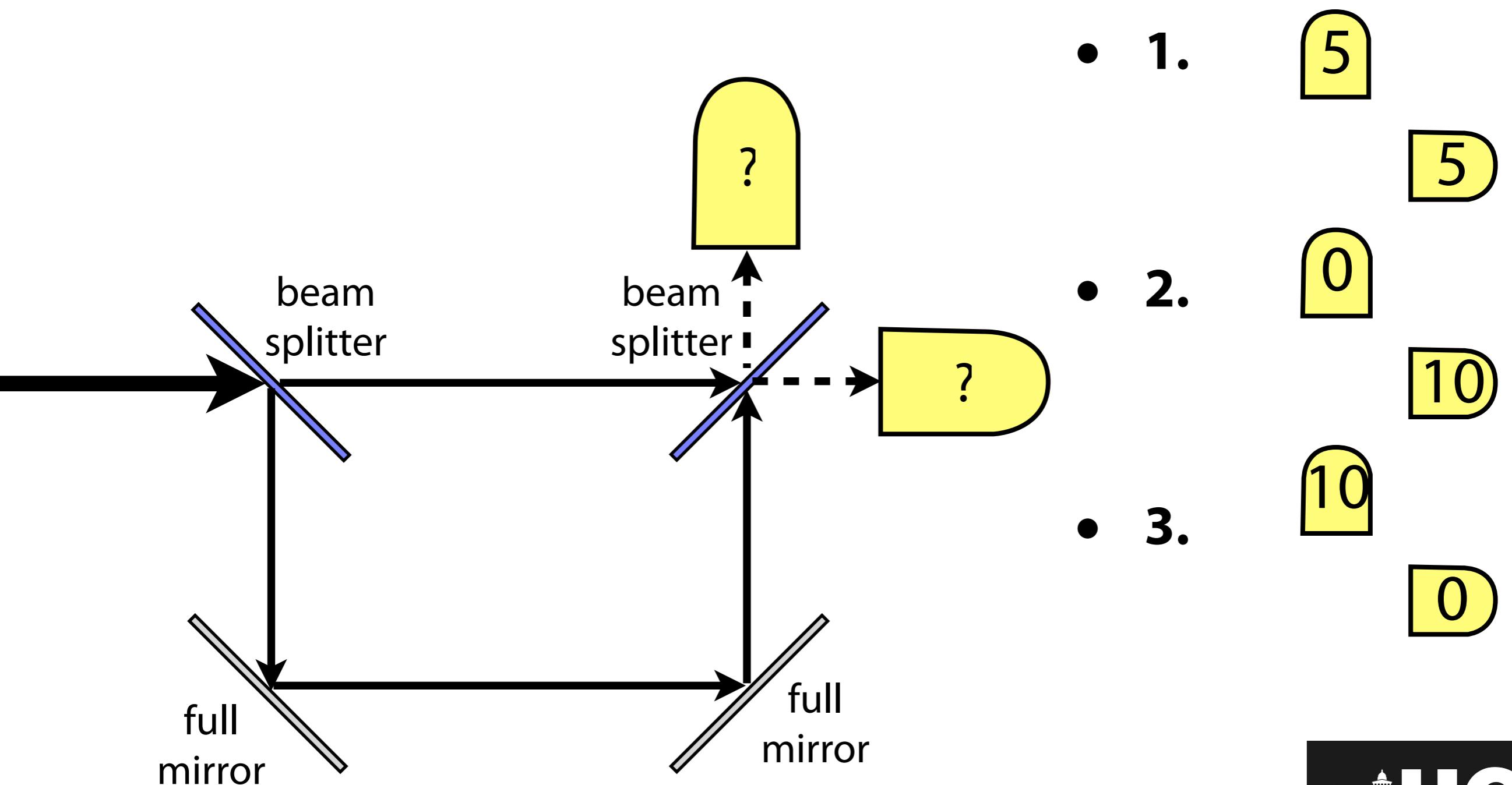
Mach-Zehnder Interferometer

- The full Mach-Zehnder Interferometer:
 - two beam splitters, two full mirrors and two detectors:



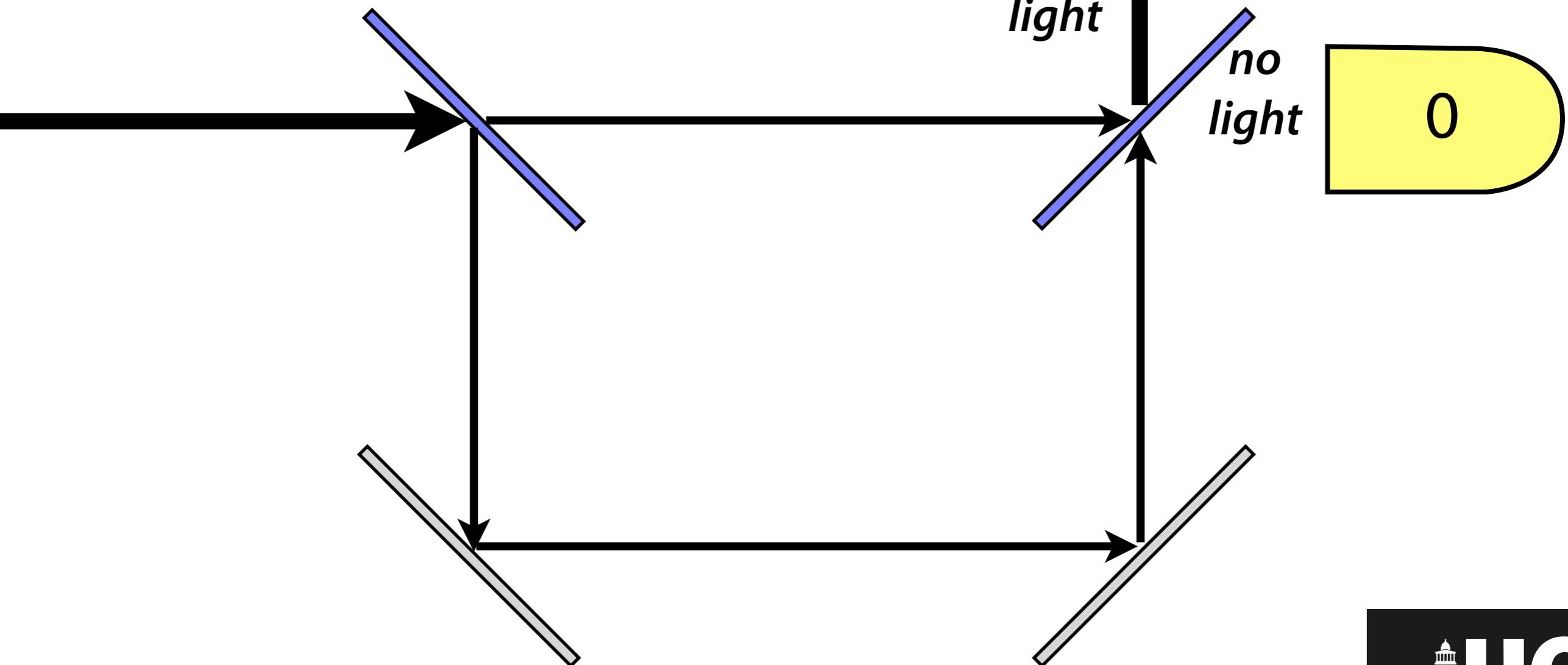
Quiz

- With 10W input what do you expect to see on the detectors?



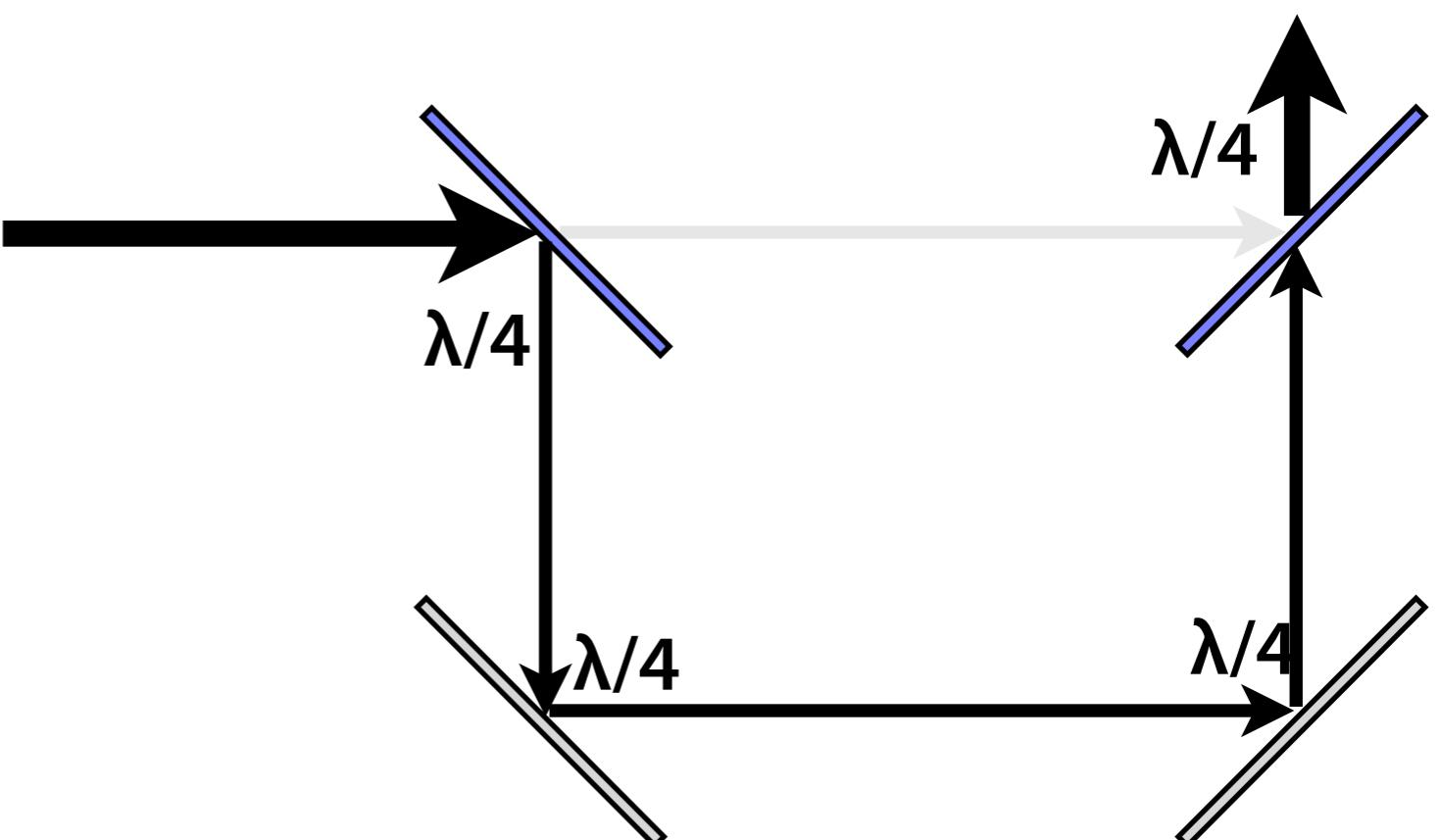
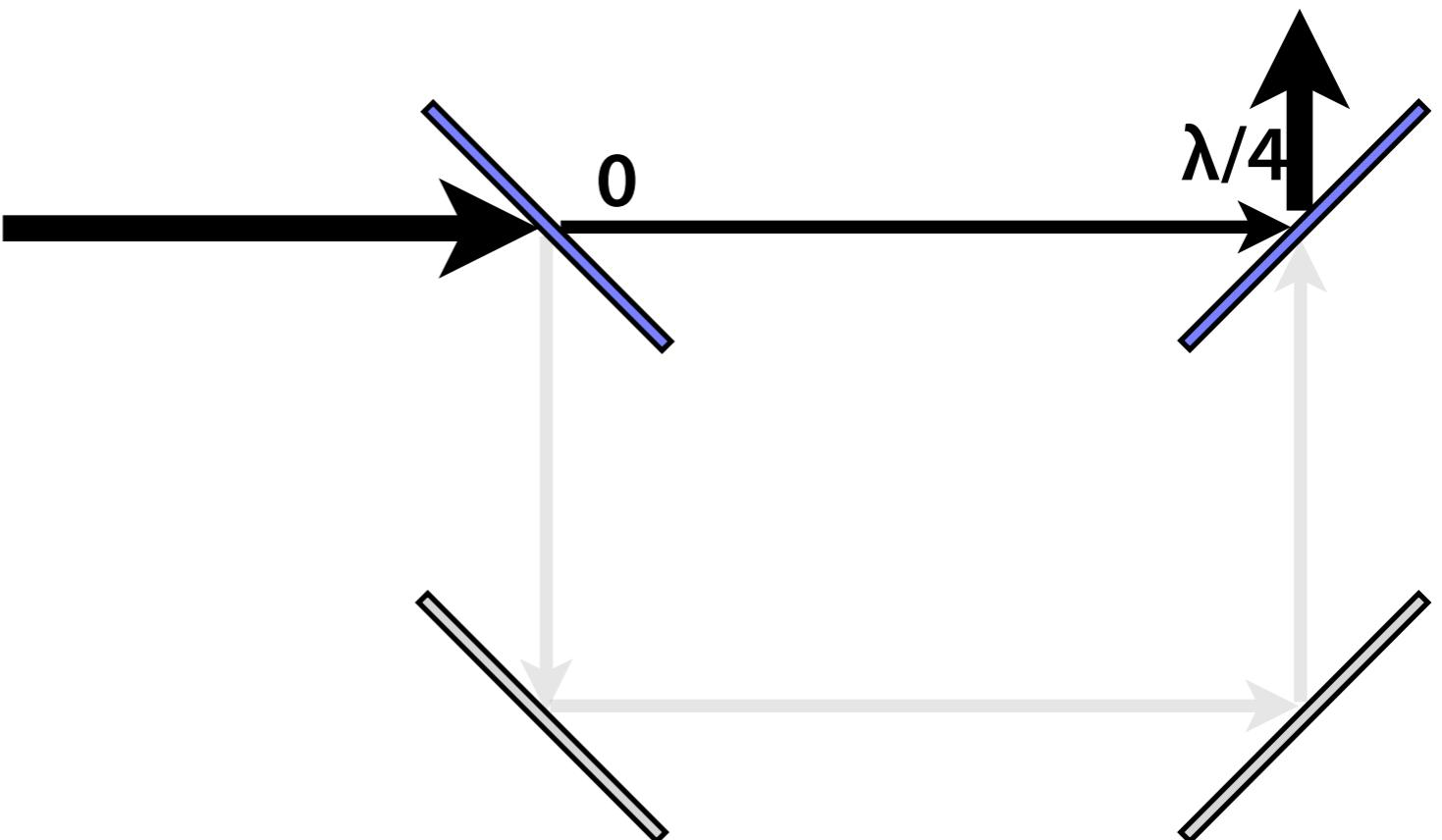
Mach-Zehnder Interferometer

- Option **3** is correct.
- **All light** enters the top detector.
None enters the right detector.
- This effect is due to **interference**.



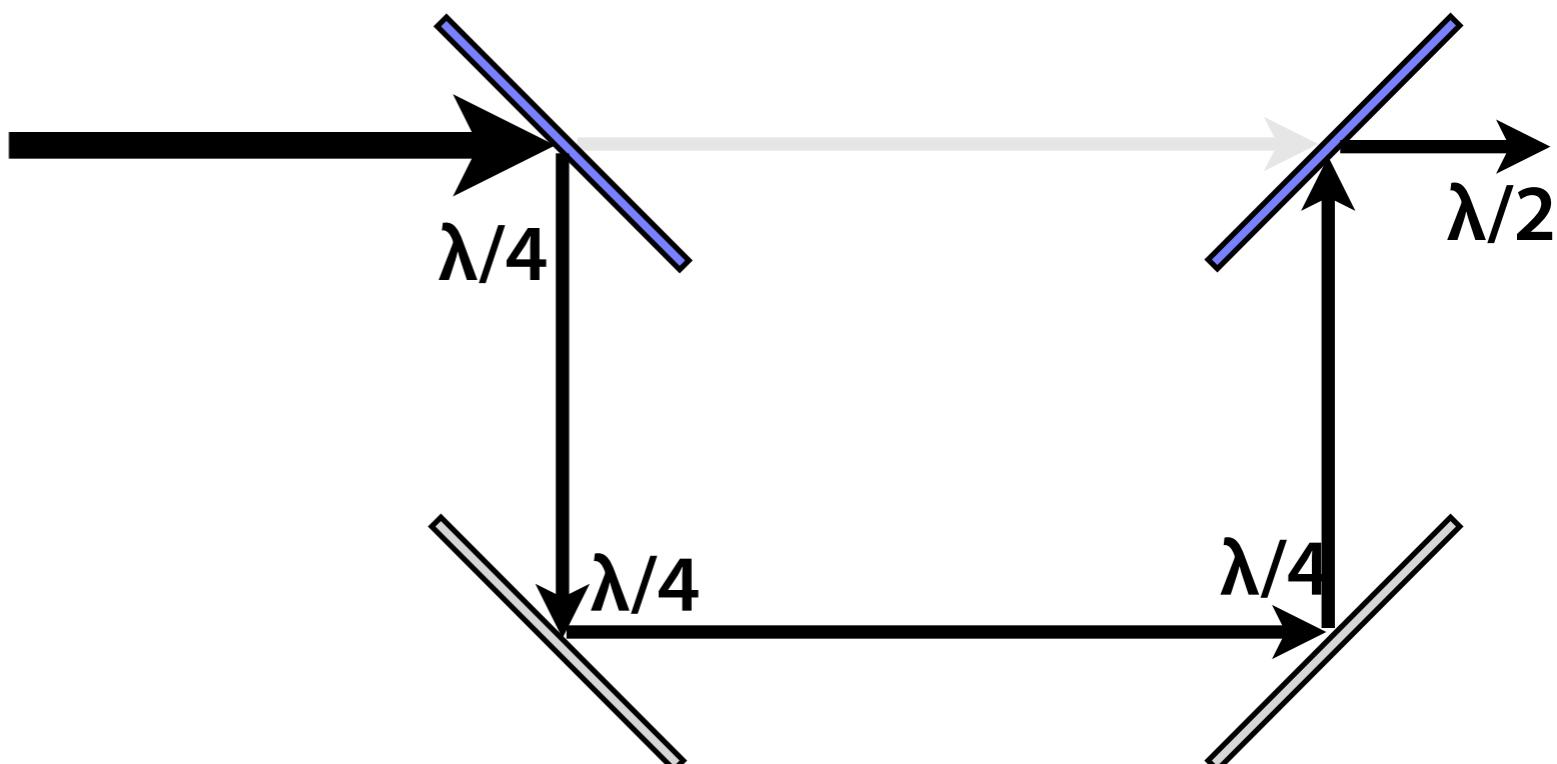
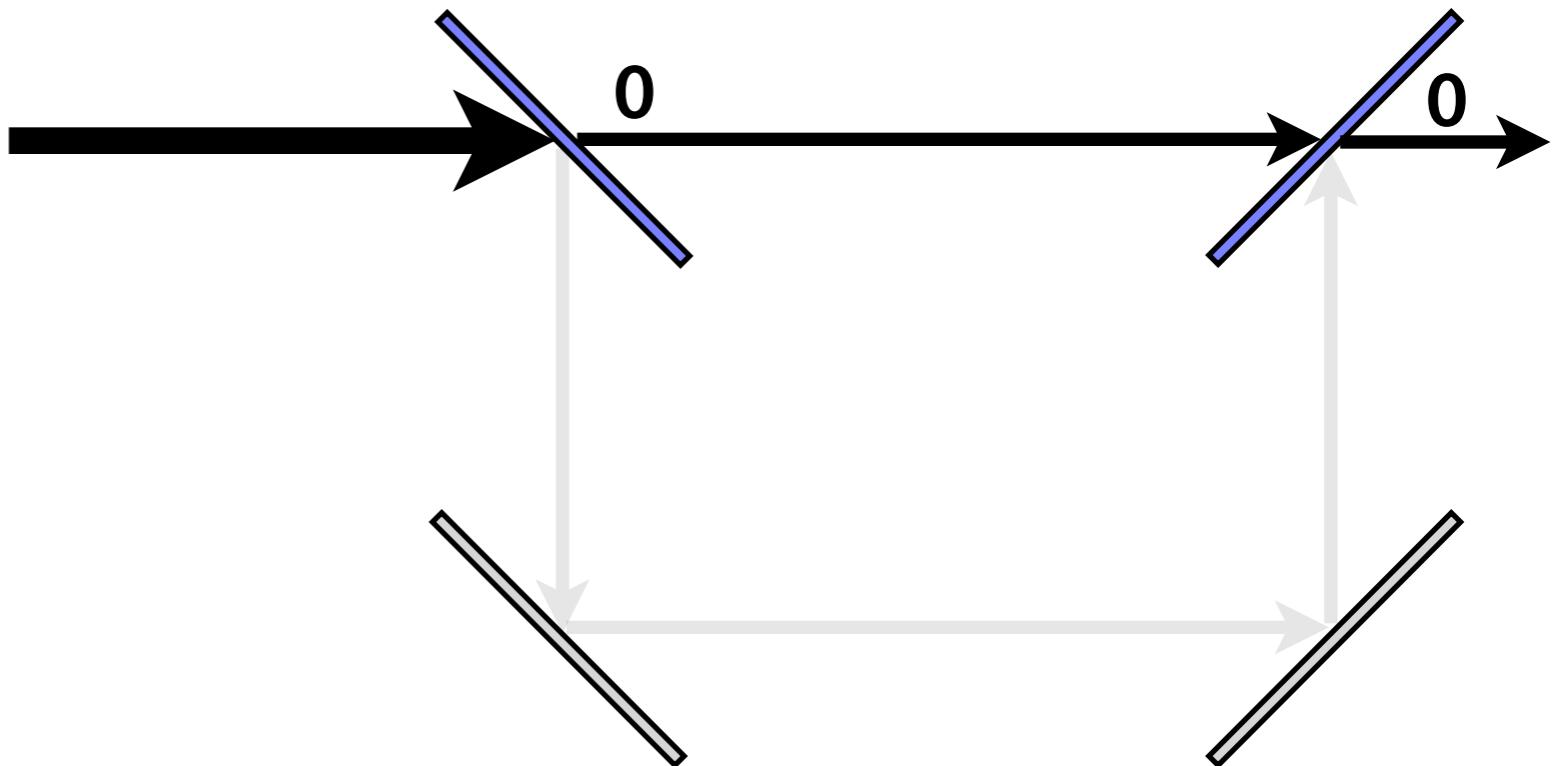
Mach-Zehnder Interferometer

- We can see why by considering the phase shifts on each path.
- First to the **top detector**.
- In both paths, there is just **one reflection in** a beam splitter.
- The paths have **same** phase shift.
- **In-phase** waves **constructively** interfere.



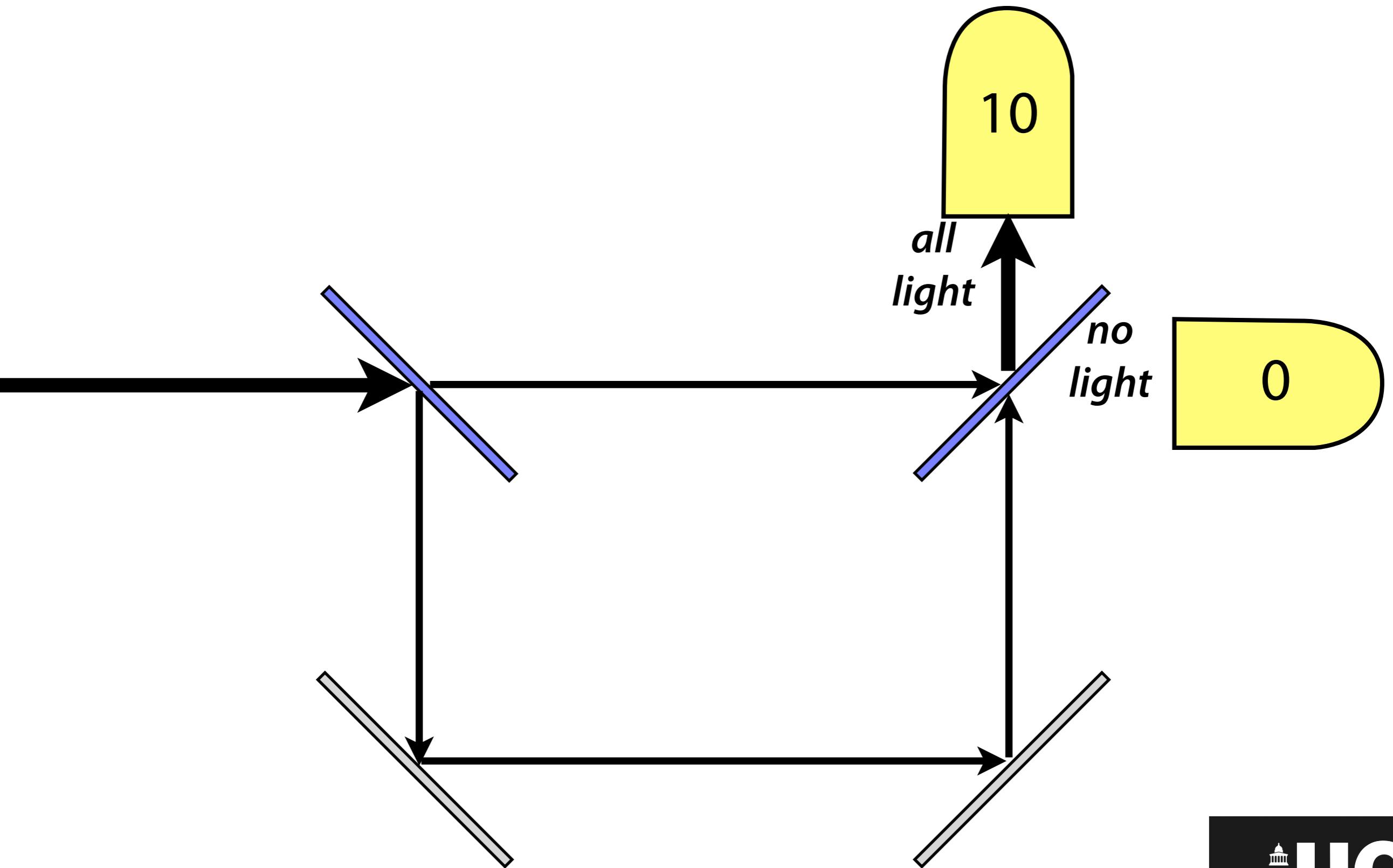
Mach-Zehnder Interferometer

- Now we consider the **to the right detector.**
- In the upper path, there are **no reflections.**
- In the lower path there are **two reflections.**
- The lower path has a phase shift of $\lambda/2$, while the upper path has **no phase** shift.
- The waves from the two paths are **in antiphase**, and hence we see **destructive** interference.



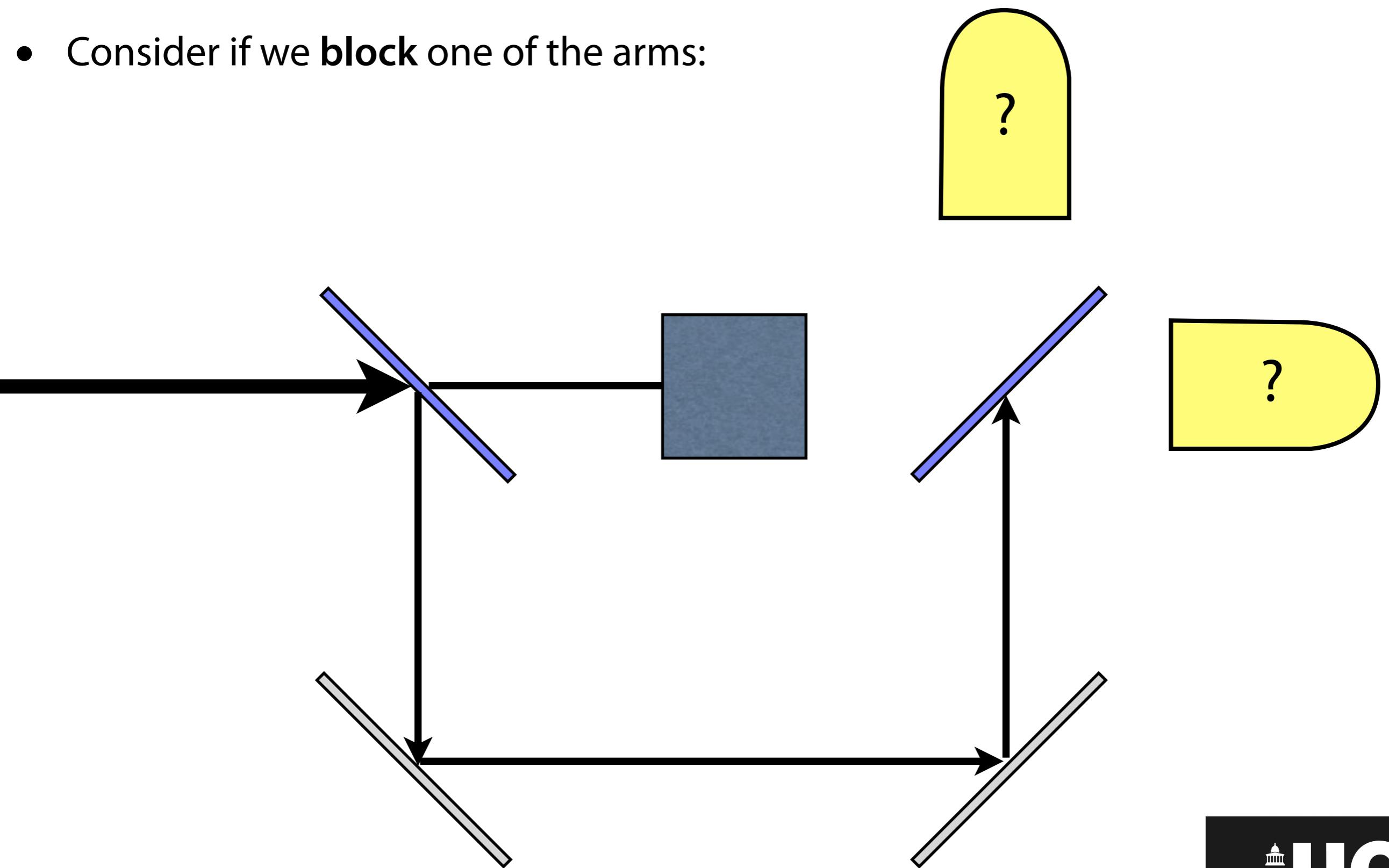
Mach-Zehnder Interferometer

- A particle picture cannot explain this.



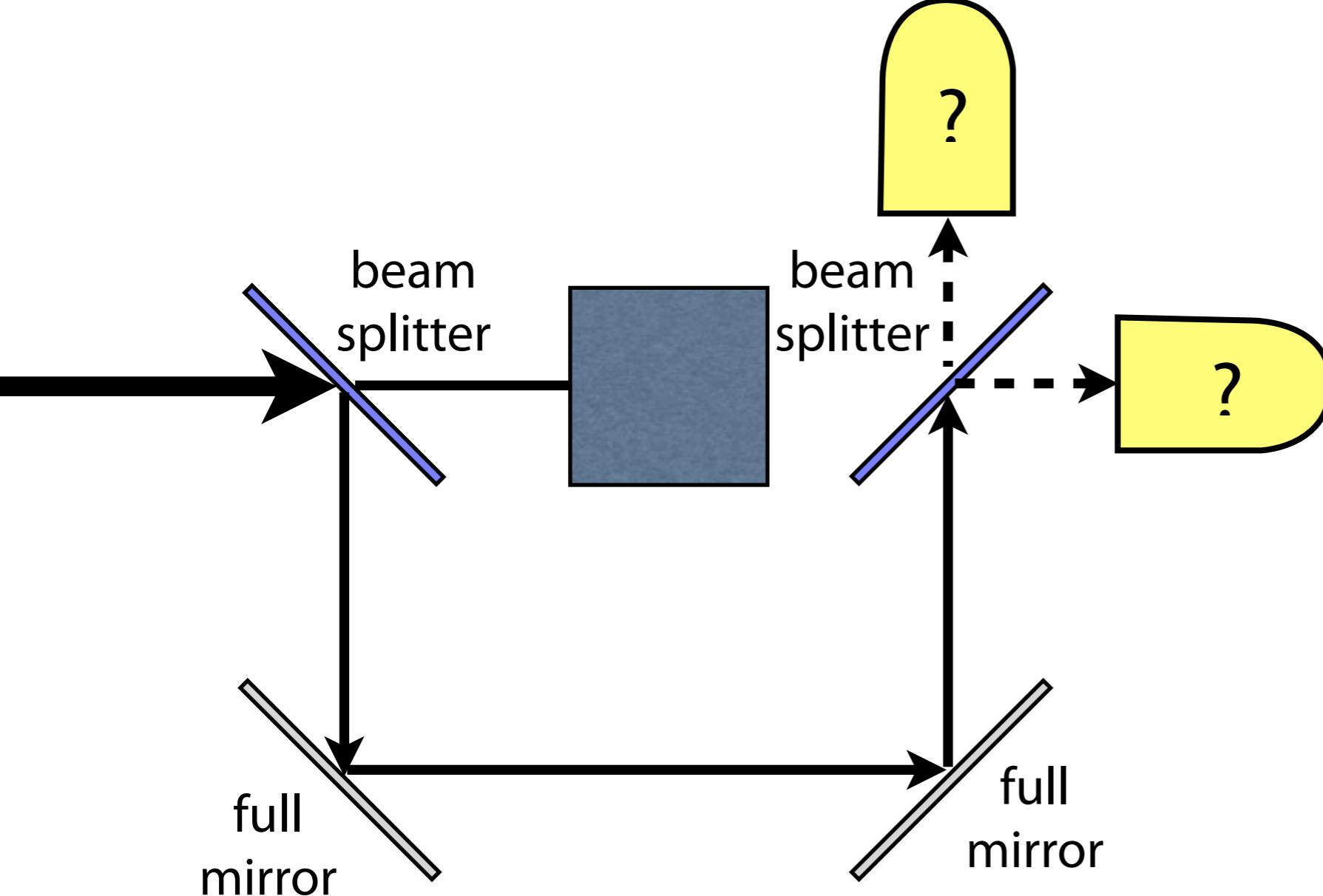
Mach-Zehnder Interferometer

- A particle picture cannot explain this.
- Consider if we **block** one of the arms:



Quiz

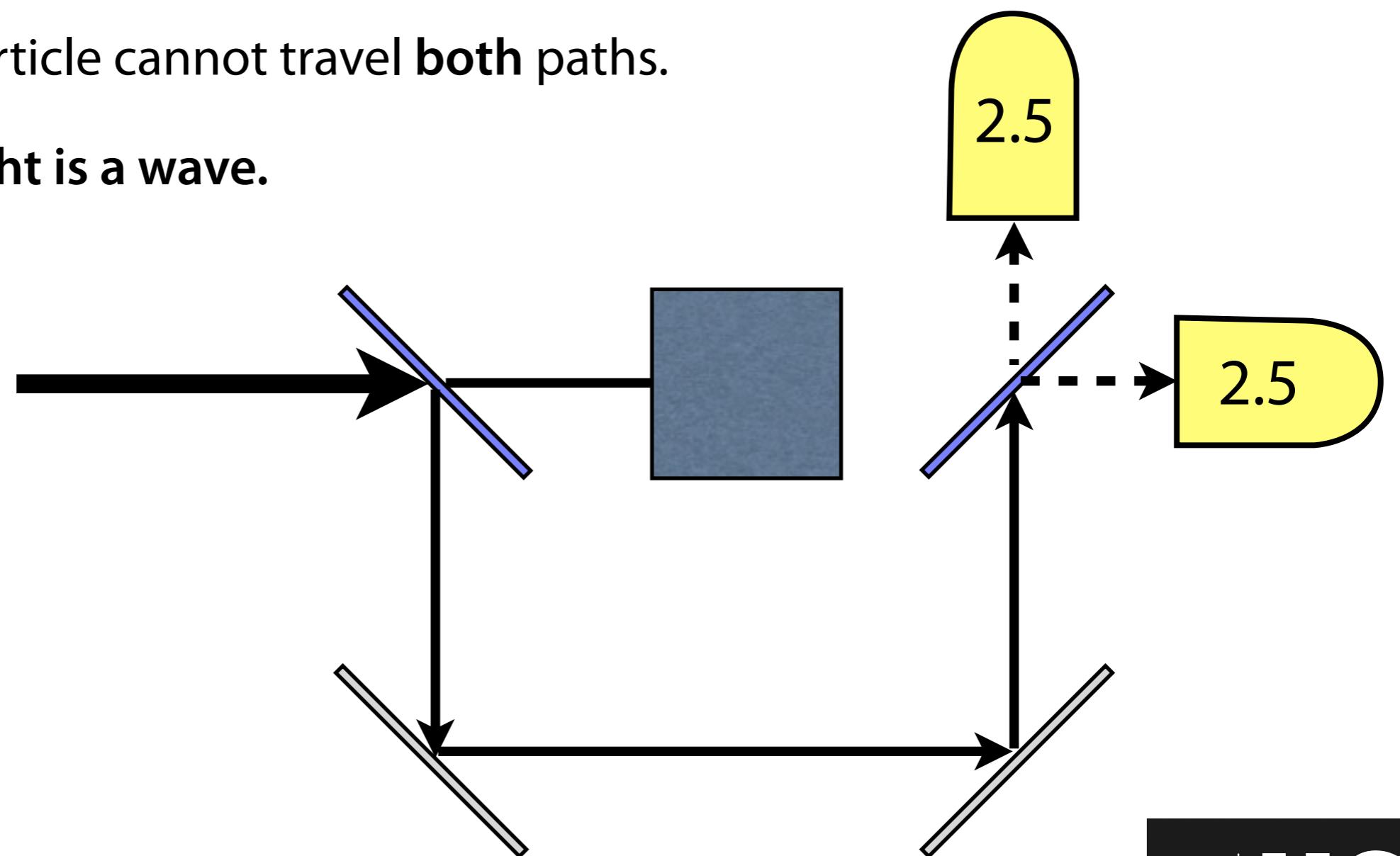
- With 10W input and an obstruction in the top path what do you expect to see on the detectors?



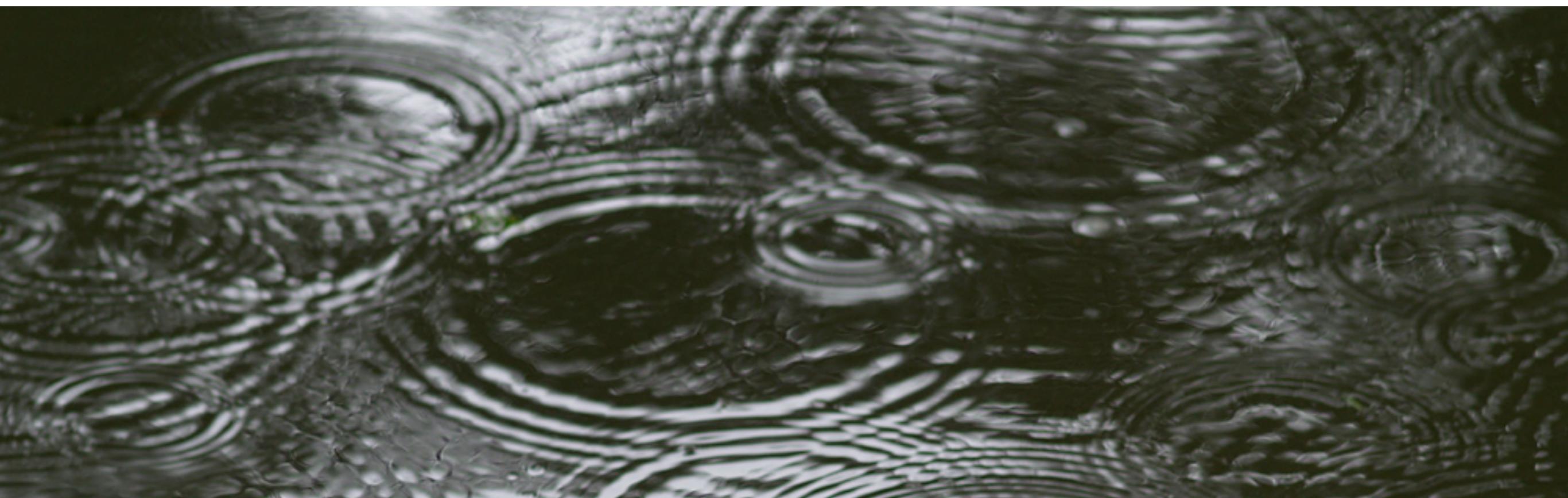
- 1. 2.5
 - 2. 5
 - 3. 10
- 1. 0
 - 2. 0

Mach-Zehnder Interferometer

- If we obstruct **one** of the paths **both** detectors change their value.
- Same is true if other path blocked instead.
- Light at each detector “knows” about **both** paths.
- A classical particle cannot travel **both** paths.
- **Therefore light is a wave.**

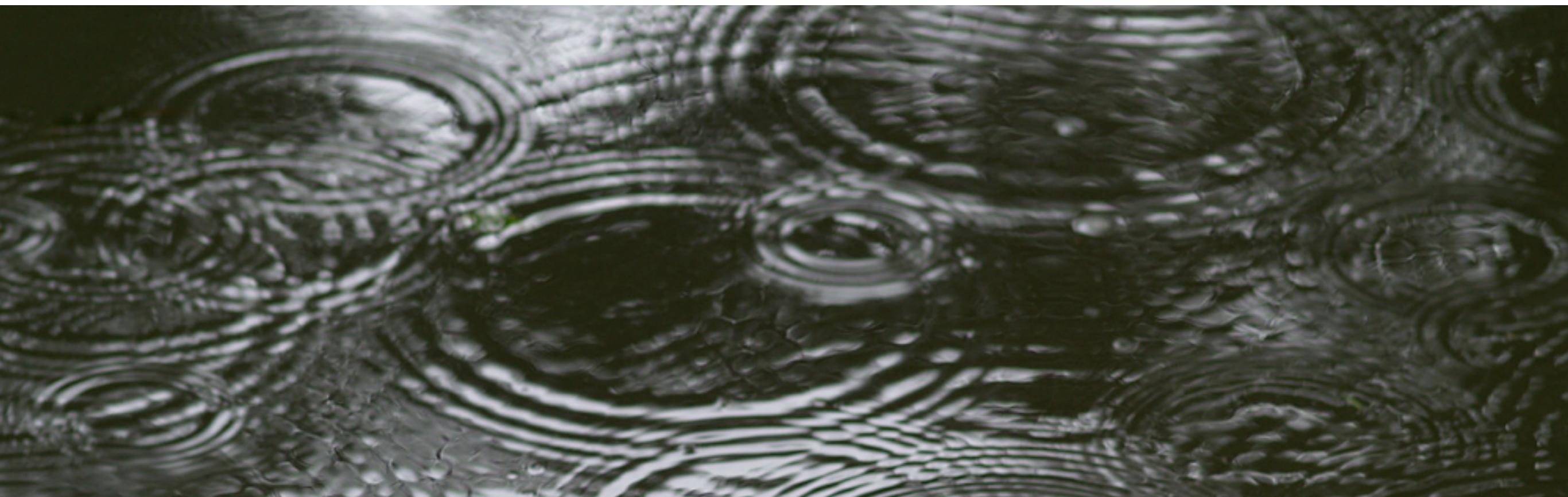


Problems with the wave picture



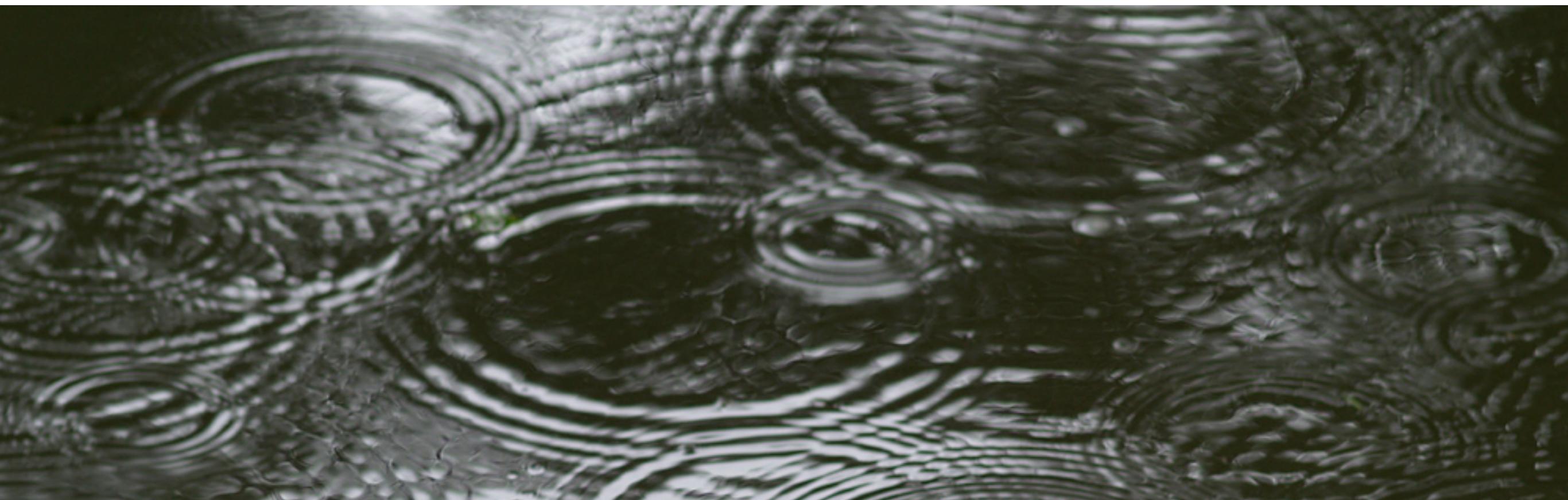
Problems with the wave picture

- The Young double slit experiment and the Mach-Zehnder both seem **irrefutable evidence that light is a wave.**



Problems with the wave picture

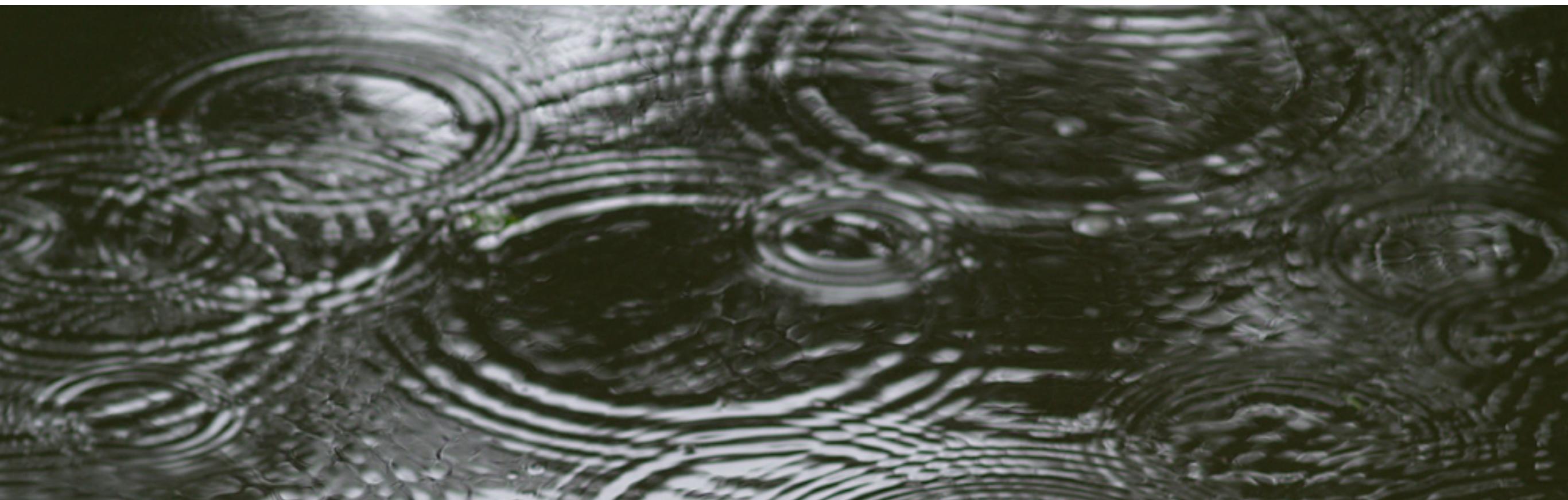
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- Fortunately Physics is **more interesting than that!**

Problems with the wave picture

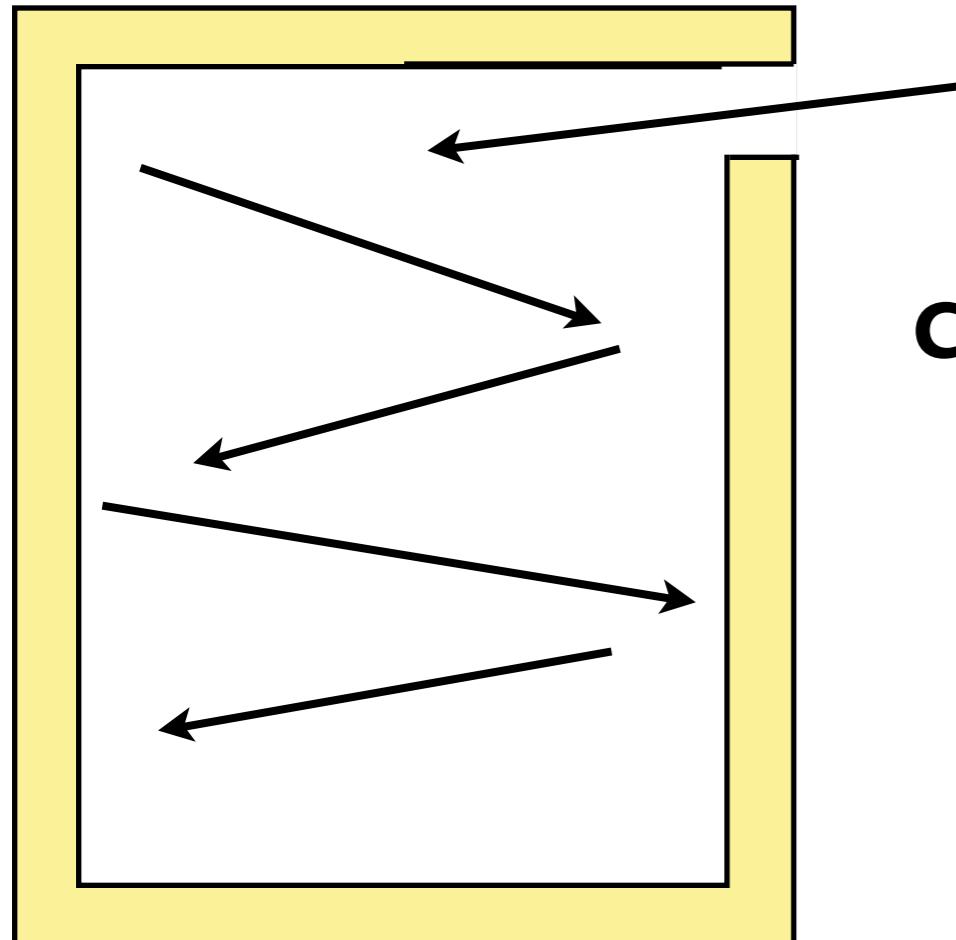
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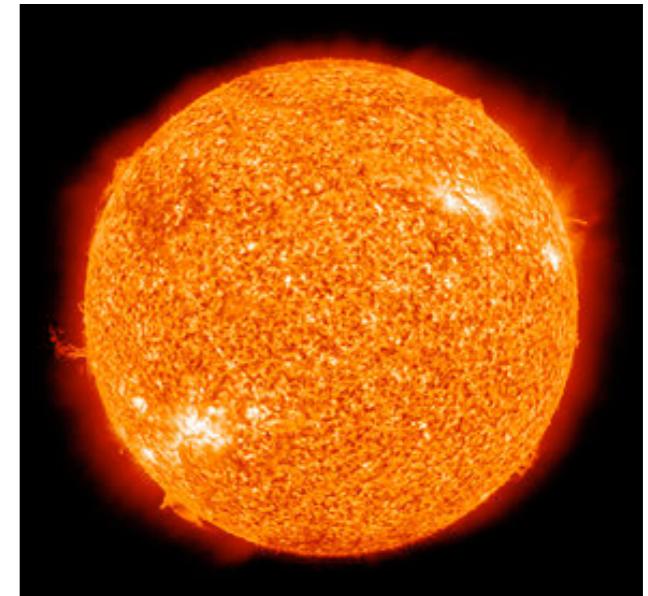
- Fortunately Physics is **more interesting than that!**
- In the first decade of the 20th century, **strong evidence** started to build that the wave picture is **not the whole story.**

Black-body radiation

- A black body is an idealised object which absorbs **all light** incident on it.
- E.g. a box with a **very small hole**.



Light
entering
cannot leave



- Although an idealisation, many physical objects behave approximately like a **black body**.

Black-body radiation

- The black body was the simplest idealised system for 19th century physicists struggling to understand **absorption** and **radiation** of light.
- Kirchoff's Challenge: (1862)
 - **experimentally measure** and
 - **theoretically understand** black body radiation.
- *To solve this became one of the key goals of 19th c. Physics!*



Gustav Kirchoff

Black-body radiation

- **Experimental Data**

- By measuring the emission of light from black-bodies at different temperatures they established:

- **Stefan's law**

$$P = \sigma AT^4$$

- **P:** Total power (Joules per second) radiated.
- **A:** Surface area of black body (m^2)
- **T:** Temperature (Kelvin)
- **σ :** Stefan's constant, $\sigma = 5.7 \times 10^{-8} \text{ W/m}^2$
- Still used by astronomers (will feature in second half of this course).

Black-body radiation

- **Experimental Data**

- The full spectrum of emitted light could be measured, as a function of temperature.

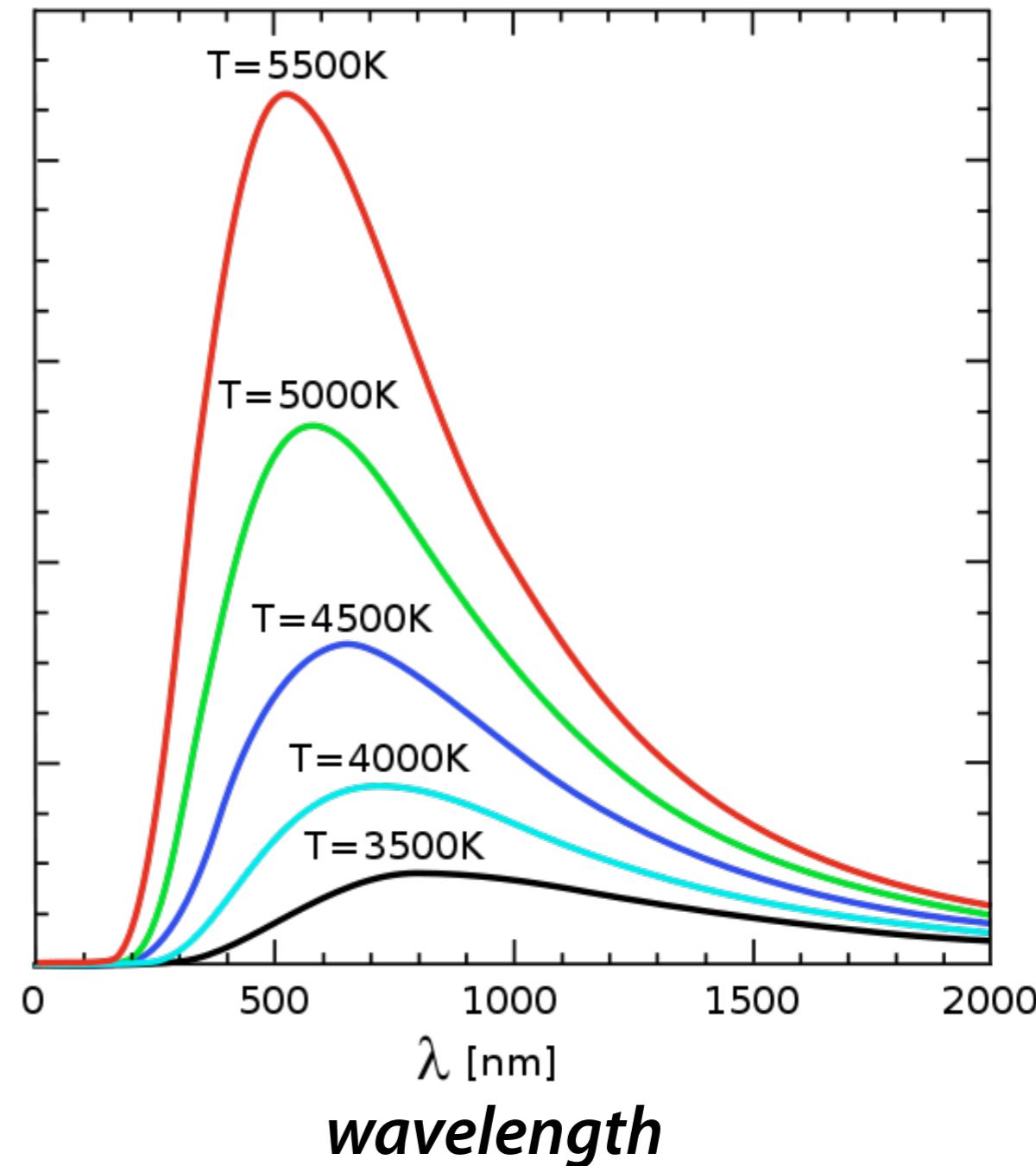
- **Wien's law**

- The **peak wavelength λ_{\max}** of the radiation is **inversely proportional to temperature T**.

$$\lambda_{\max} = \frac{2.9 \times 10^{-3} \text{ m K}}{T}$$

- “The hotter, the bluer”.
- Also used by astronomers - star colour indicates temperature.

Black-body emission spectra
intensity

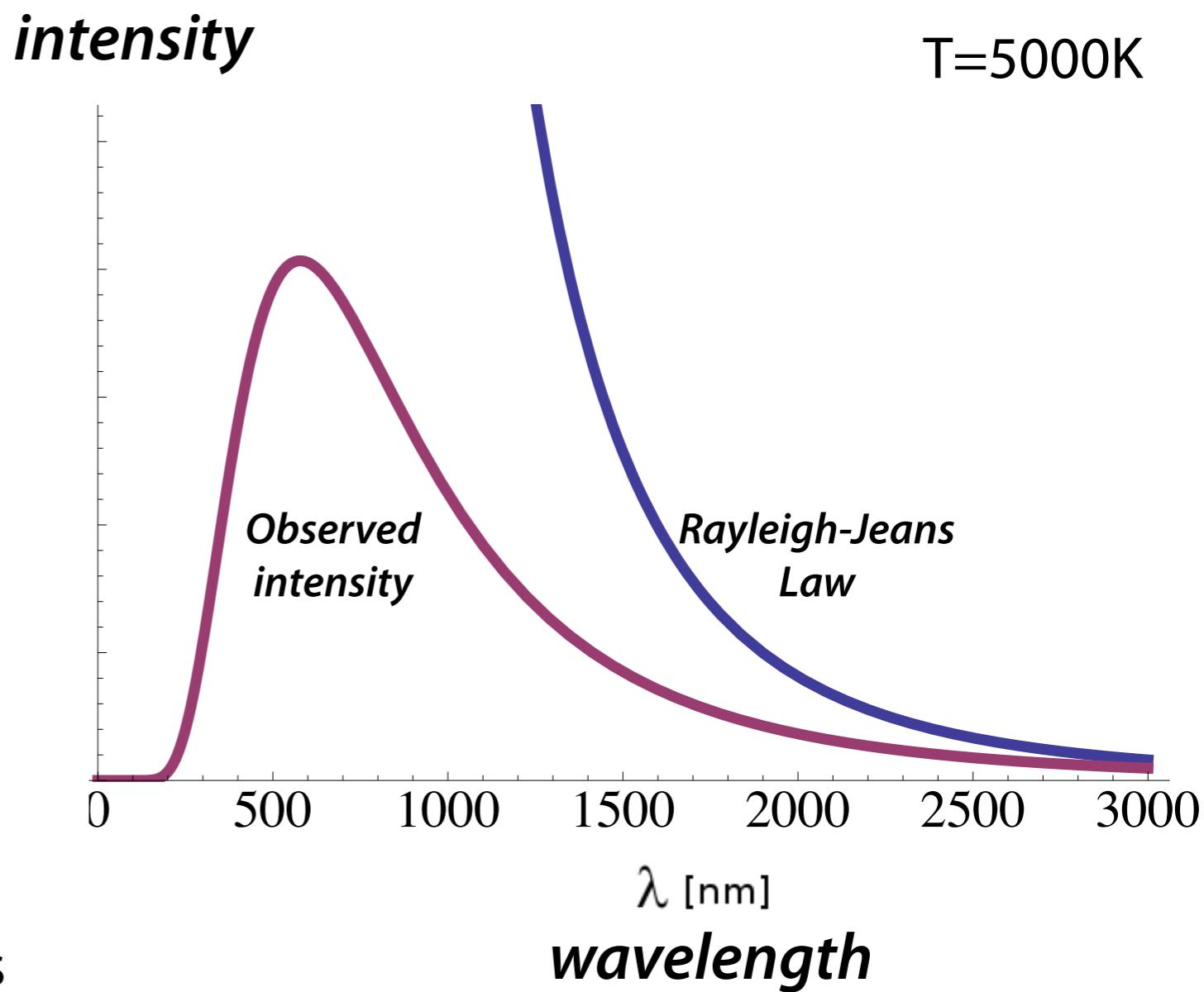


Black-body radiation

- Aim: A theoretical derivation of these curves.
- Best attempt: Rayleigh-Jeans law

$$I(\lambda, T) = \frac{2\pi c k T}{\lambda^4}$$

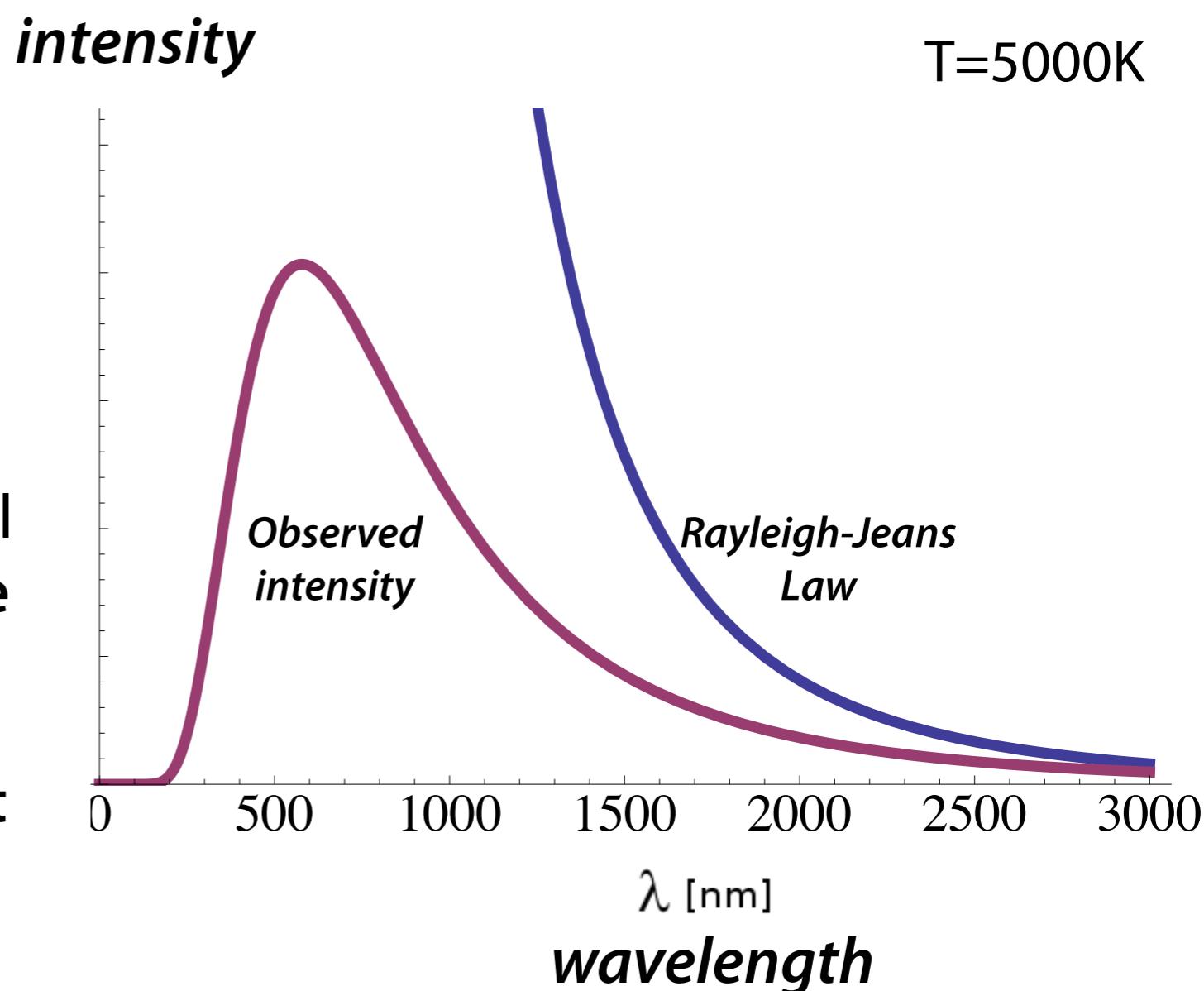
- c is the speed of light,
 k is Boltzmann's constant,
 T is temperature, and
 λ is wavelength
- Derivation used thermodynamics and wave model of light. (*See 2nd year Statistical Thermodynamics course*).
- Key assumption - at all wavelengths light energy is absorbed and emitted **continuously**.



Black-body radiation

$$I(\lambda, T) = \frac{2\pi c k T}{\lambda^4}$$

- Rayleigh-Jeans law was unsatisfactory.
 - First - it didn't fit the data.
 - Even worse, it diverged for small wavelengths, predicting **infinite radiation** for very small λ .
 - This is known as the **ultra-violet catastrophe!**



Black-body radiation

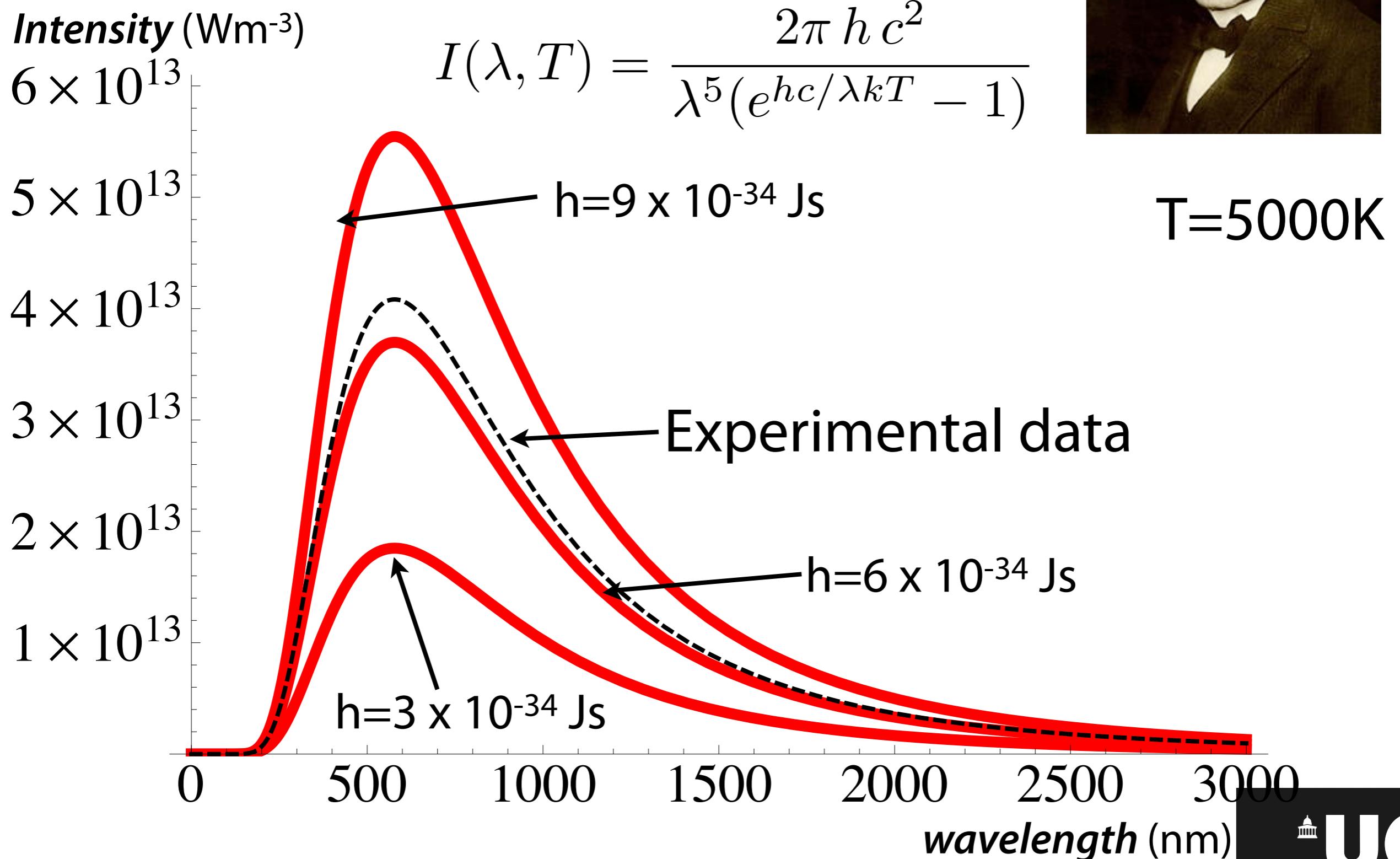


- **Planck (1900)**
 - Initial Idea: A mathematical trick:
 - Assume light energy is absorbed and emitted in **discrete units**.
 - Each “unit” has energy proportional to its frequency.
 - where **f** is **frequency** and **h** is a constant **to be determined**
 - The result: **Planck's law** (derived in 2nd year Thermal Physics)

$$I(\lambda, T) = \frac{2\pi h c^2}{\lambda^5 (e^{hc/\lambda kT} - 1)}$$

Black-body radiation

- Planck's law - for different h-values



Black-body radiation

- *Planck's constant*
 - Modern experiments have measured **h - Planck's constant** - to high accuracy:
$$h = 6.626 \times 10^{-34} \text{ Js}$$
 - In research-level Physics, we often use “hbar”:
$$\hbar = \frac{h}{2\pi} = 1.055 \times 10^{-34} \text{ Js}$$
- The success of Planck's law seemed mysterious. It was known that light was a wave.
- Yet in Planck's derivation, light behaved as if in discrete “**packets**” of energy.
$$E = hf = \hbar\omega$$
- where ω is **angular frequency** $\omega = 2\pi f$

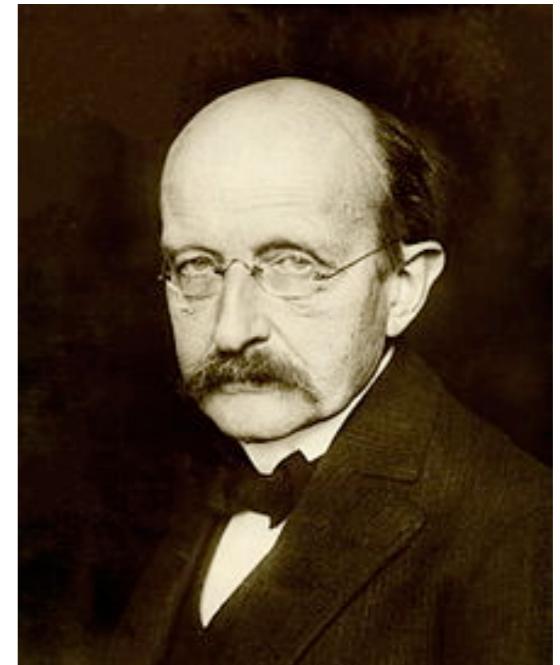
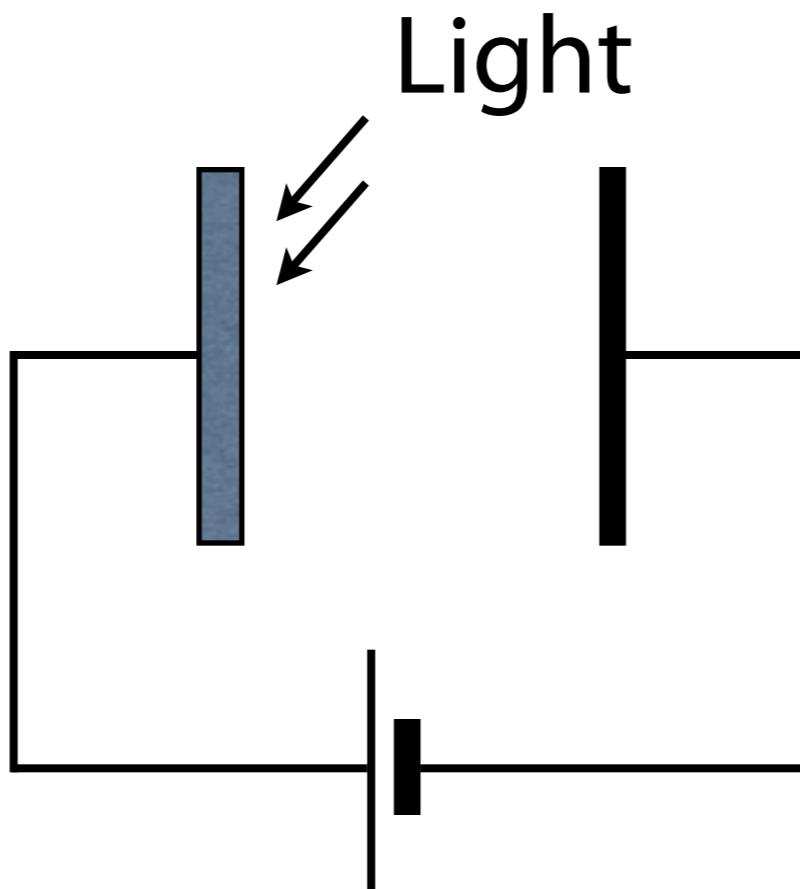
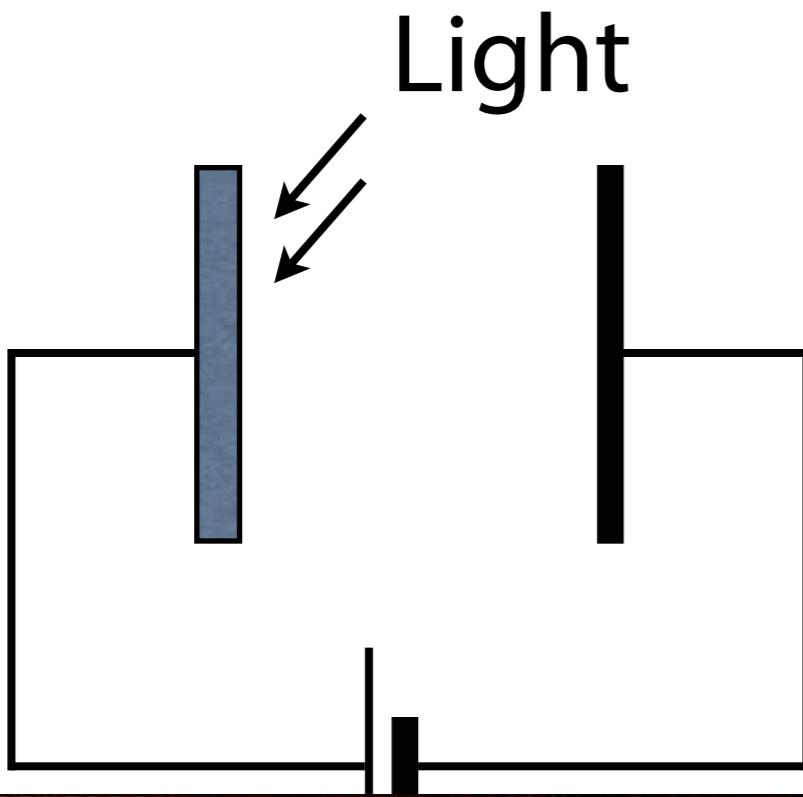


Photo-electric effect



- **Experimental Observations**
 - When incident light is above a **threshold frequency**, current flows, proportional to intensity.
 - When incident light is below the **threshold frequency**, no current flows.

Photo-electric effect



- Einstein's idea
 - Take Planck's model seriously.
 - Assume light consists of **particles** of energy $hf = \hbar\omega$
 - Electrons can only absorb energy from one photon at a time.
 - Each electron has a minimum **binding energy** ("work function") which it needs to receive to escape the material and travel round the circuit.

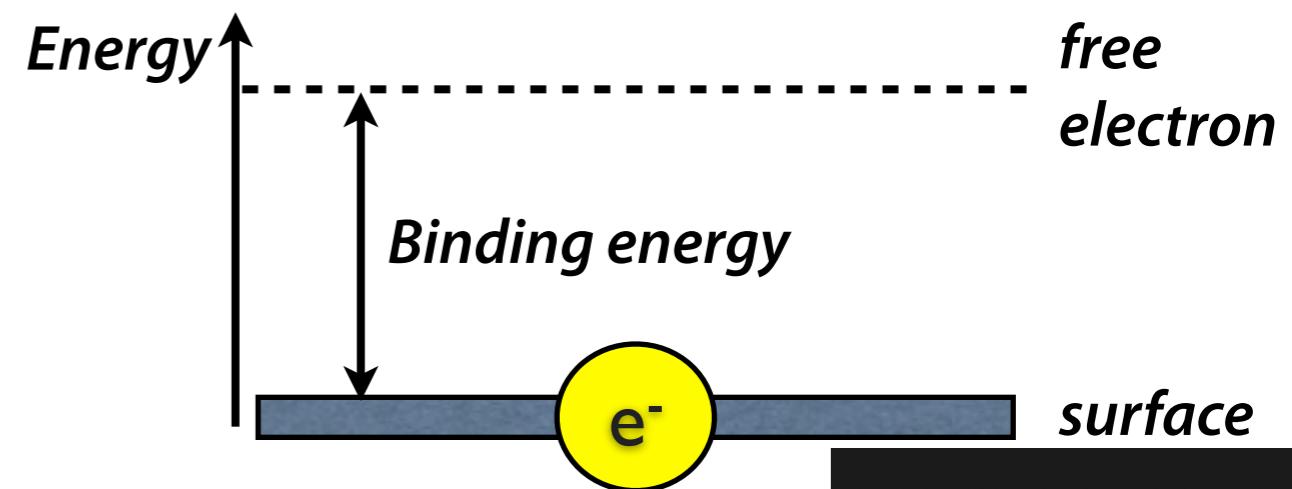
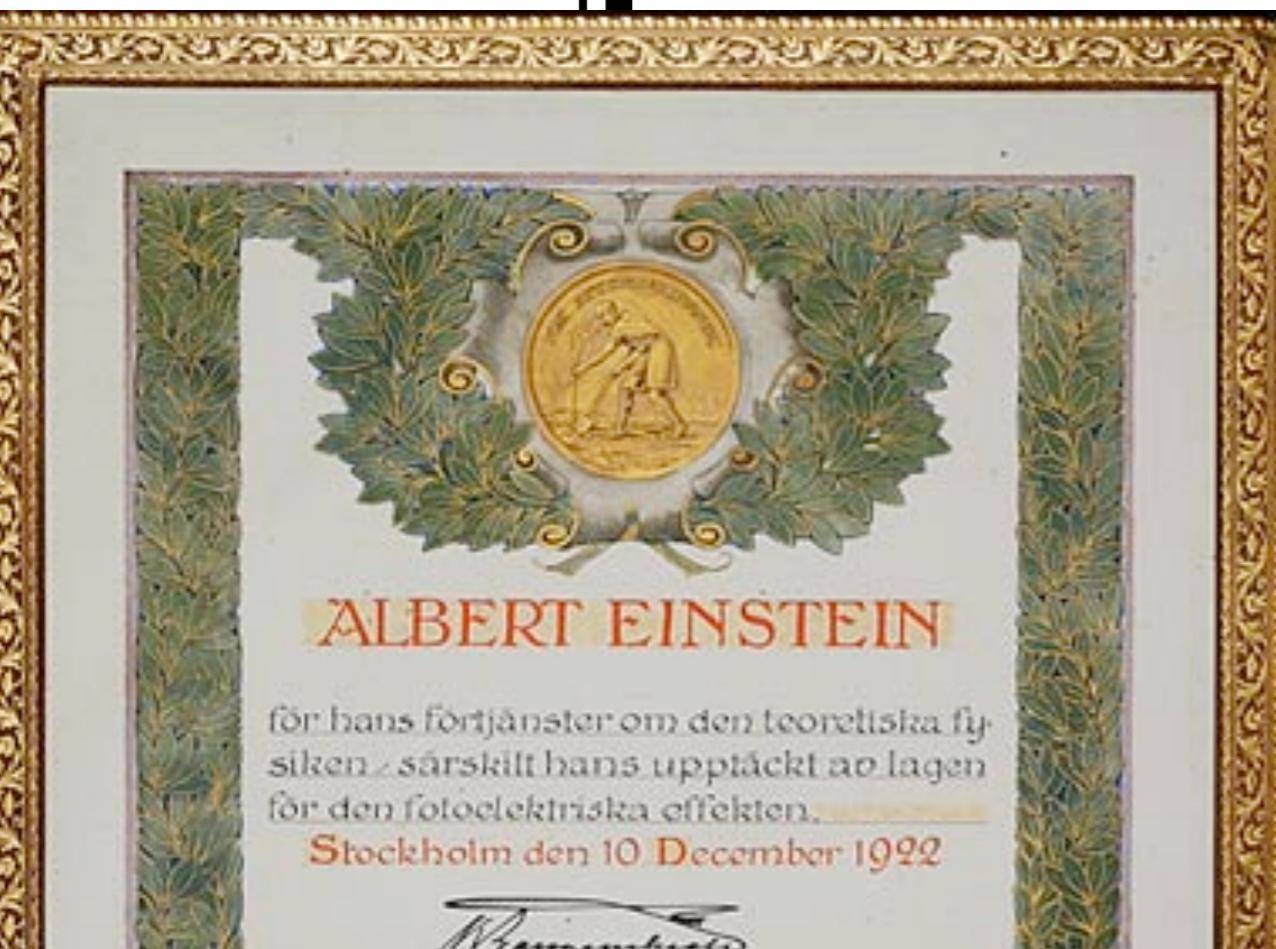
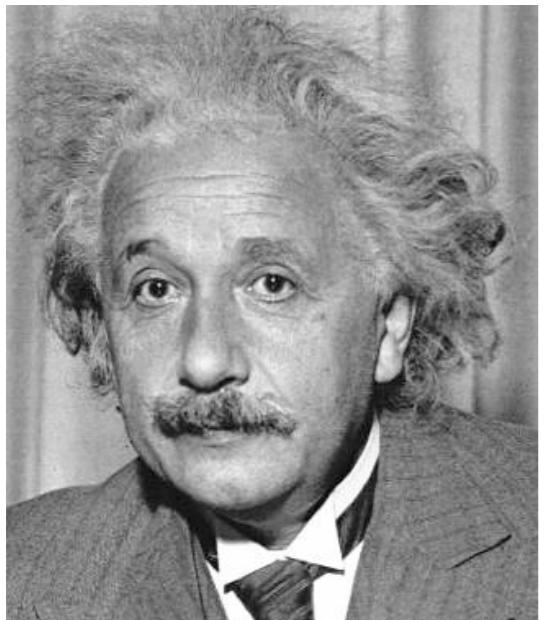


Photo-electric effect



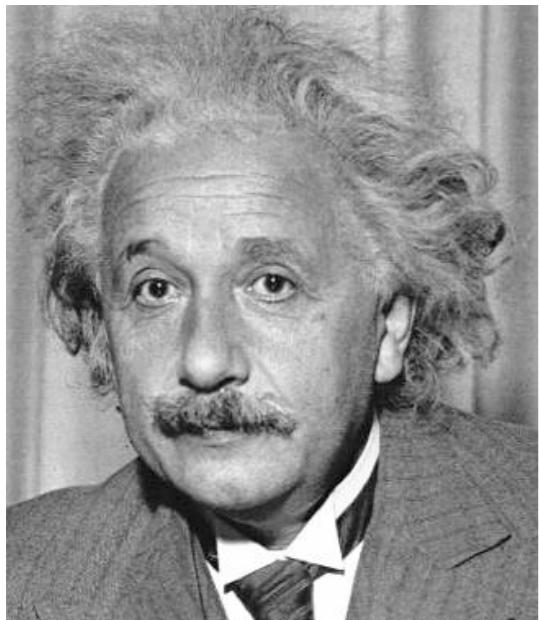
- Einstein's model explained **all** features of the photo-electric effect.
- But could the **wave picture** of light be given up so readily?
- But Young's **wave evidence was very strong**. Perhaps the photo-electric effect had **another explanation**?
- To convince the bulk of physicists more **persuasive evidence** was needed.

Photon Momentum



- The answer lay in photon **momentum**
 - (In part due to another contribution of Einstein!)
- Classical mechanics:
 - **Waves** have momentum
 - **Light** has momentum
 - (consequence of electromagnetic theory - 2nd year EM course)
 - Therefore **photons** must have momentum too.

Photon Momentum



- Special relativity (1905)
 - **Massless particles travel at the speed of light.**
 - Massless particles carry **momentum** - given by:

$$p = \frac{E}{c}$$

- (Related to $E=mc^2$ which holds for particles with mass)
- The photon is a massless particle with **energy**

$$E = hf$$

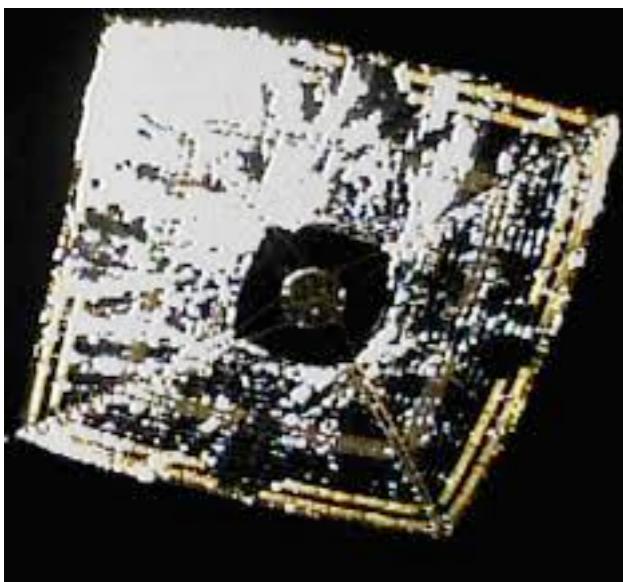
- and therefore **momentum**

$$p = h\frac{f}{c} = \frac{h}{\lambda}$$

Photon Momentum

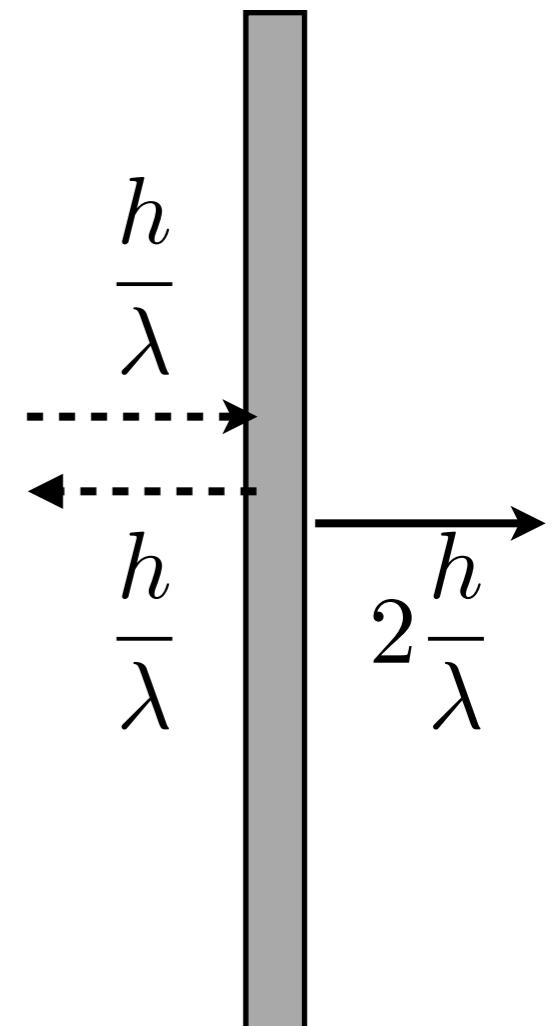
- Every photon carries momentum:

$$p = \frac{h}{\lambda}$$



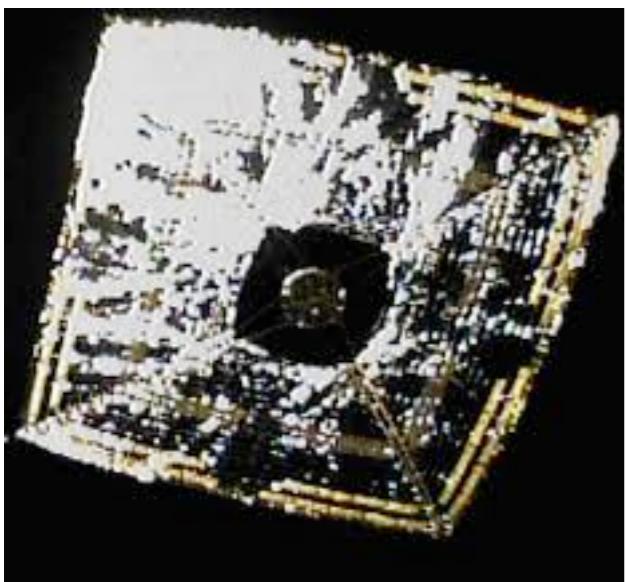
IKORAS
Venus
probe

- Thus reflecting light exerts **pressure**.
- The physics behind **solar sails**.
- Each photon reflected imparts a **momentum kick** to the sail, accelerating the probe.



Quiz

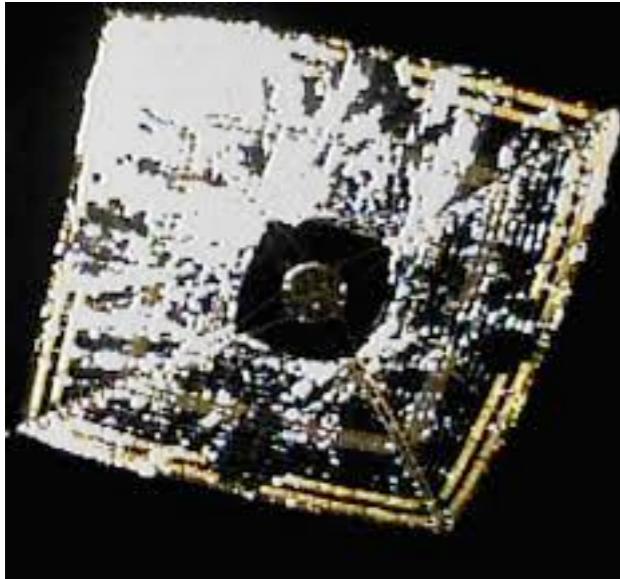
- Is radiation pressure evidence for the particle nature of light?



- 1. Yes
- 2. No

Quiz

- Is radiation pressure evidence for the particle nature of light?



- 1. Yes
- 2. No

- Classical electromagnetism predicts the **same** radiation pressure when light reflects from **macroscopic objects**.



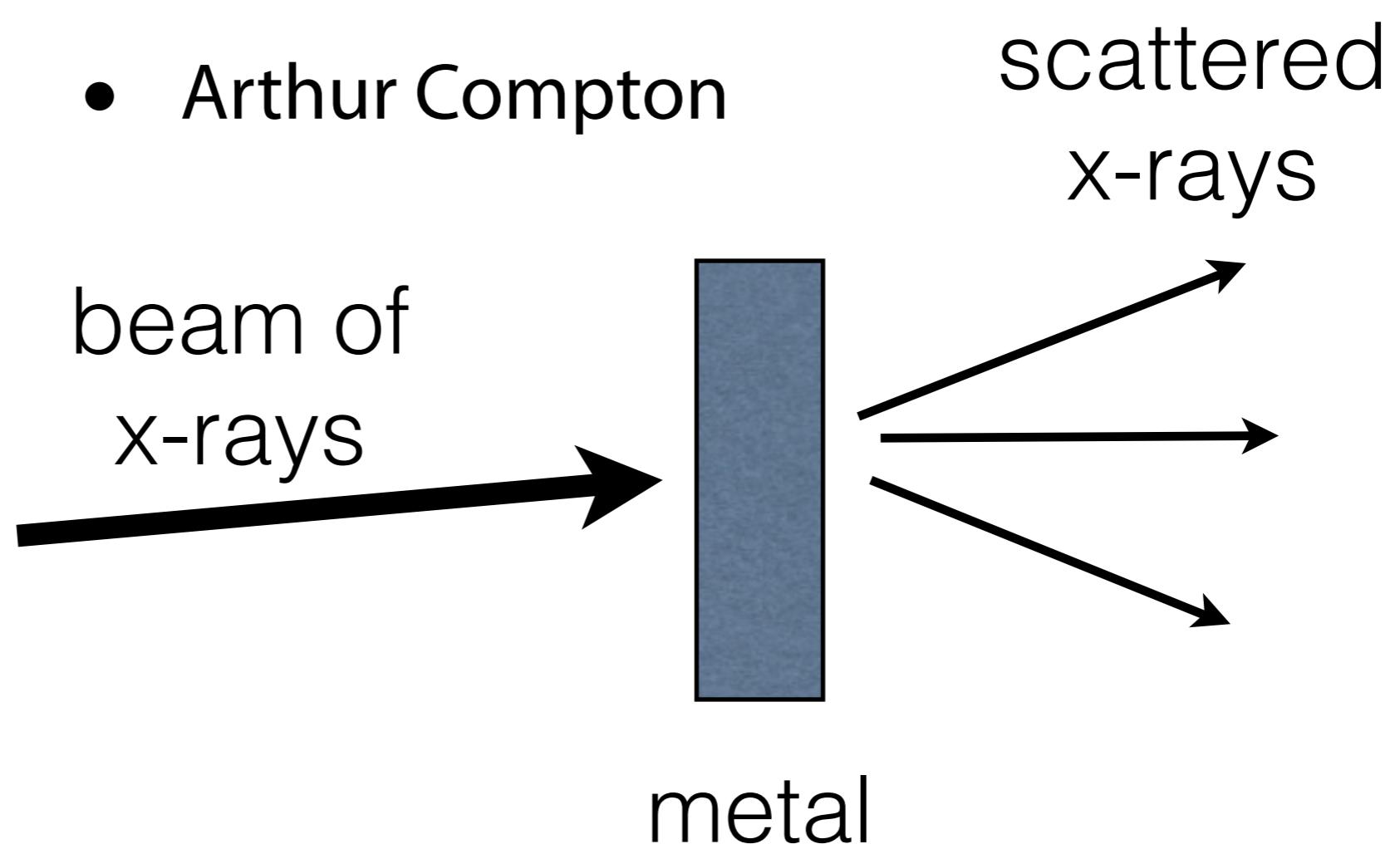


Arthur Compton

- This is **not true** for microscopic charged particles.
 - In interaction with **single charged particles** (e.g. electrons).
 - **Classical electromagnetism** and **photon picture** give **very different** predictions.
- This is called the **Compton Effect**.

The Compton Effect

- 1922 - *Compton Scattering*



Arthur Compton

metal

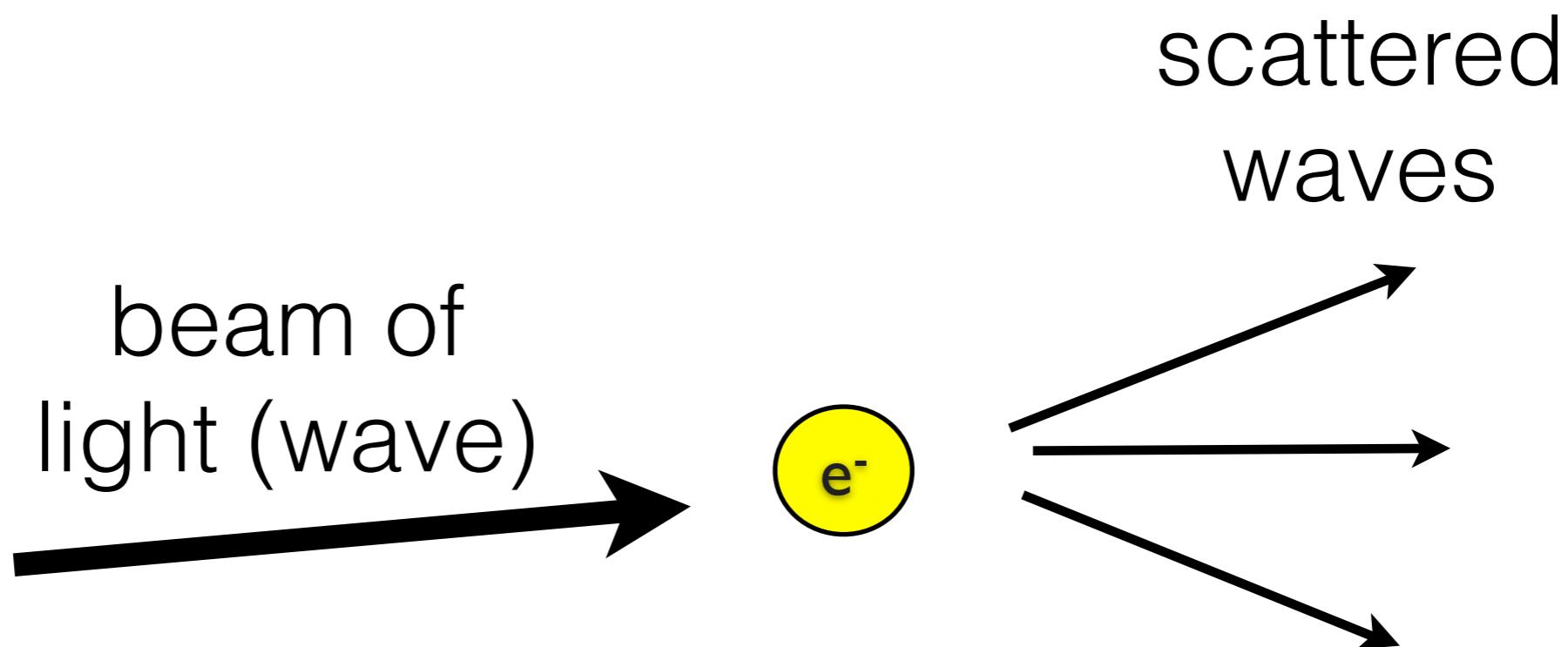
- When **X-rays** (nm light) are incident on a metal, they **scatter** from the free electrons in the metal.

The Compton Effect

- *Thomson scattering (1900s)*
 - Thompson developed a theory for how **electromagnetic waves** should scatter from **charged particles**.



J.J. Thomson

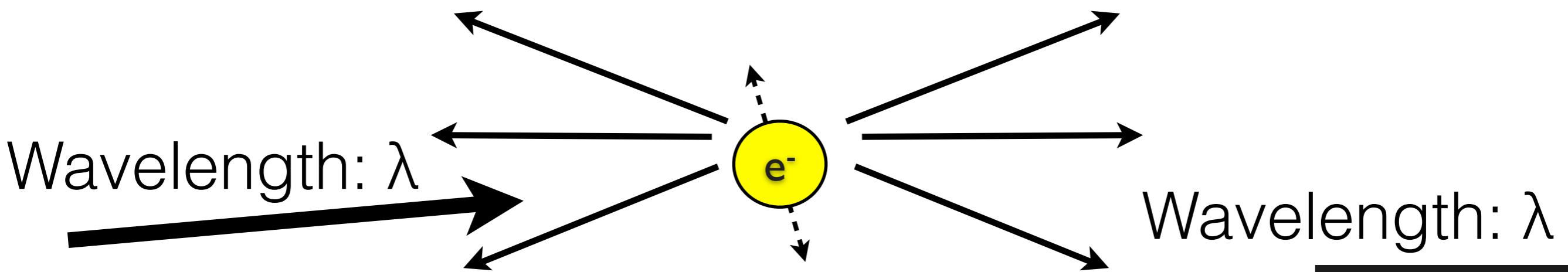


The Compton Effect

- *Thomson scattering (1900s)*
- Key features:
 - The incoming wave (frequency $f=c/\lambda$) induces charged particle to oscillate at **same frequency** f .
 - Oscillating particle emits light at the **same frequency** f in all directions.
- An important effect in Plasma physics.



J.J. Thomson

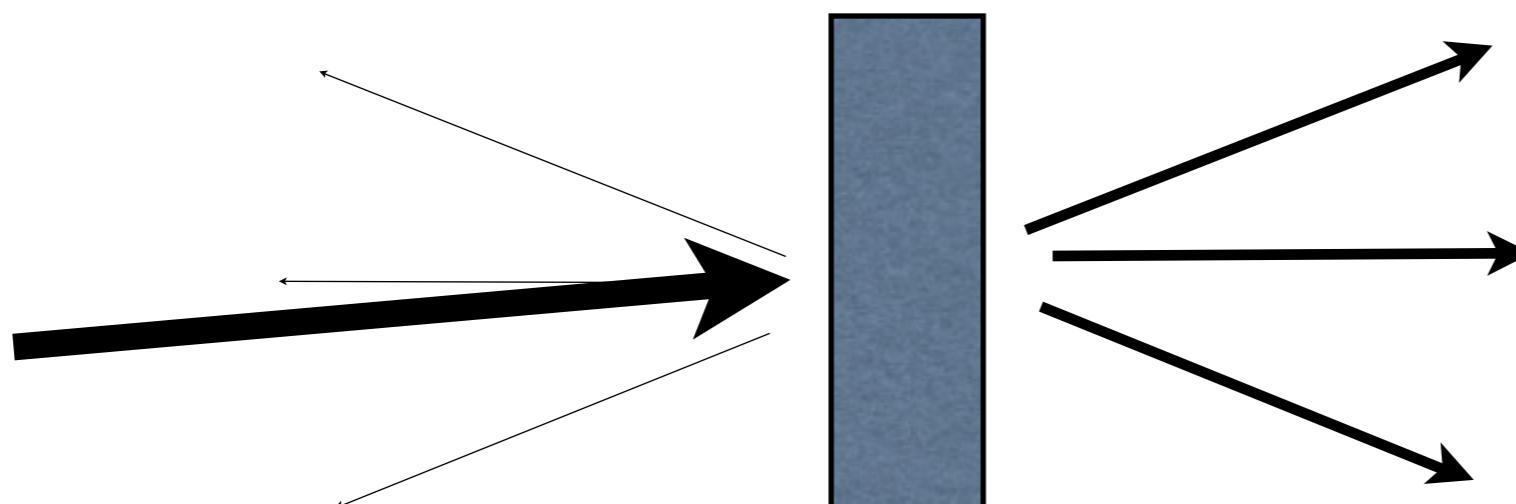


The Compton Effect



Arthur Compton

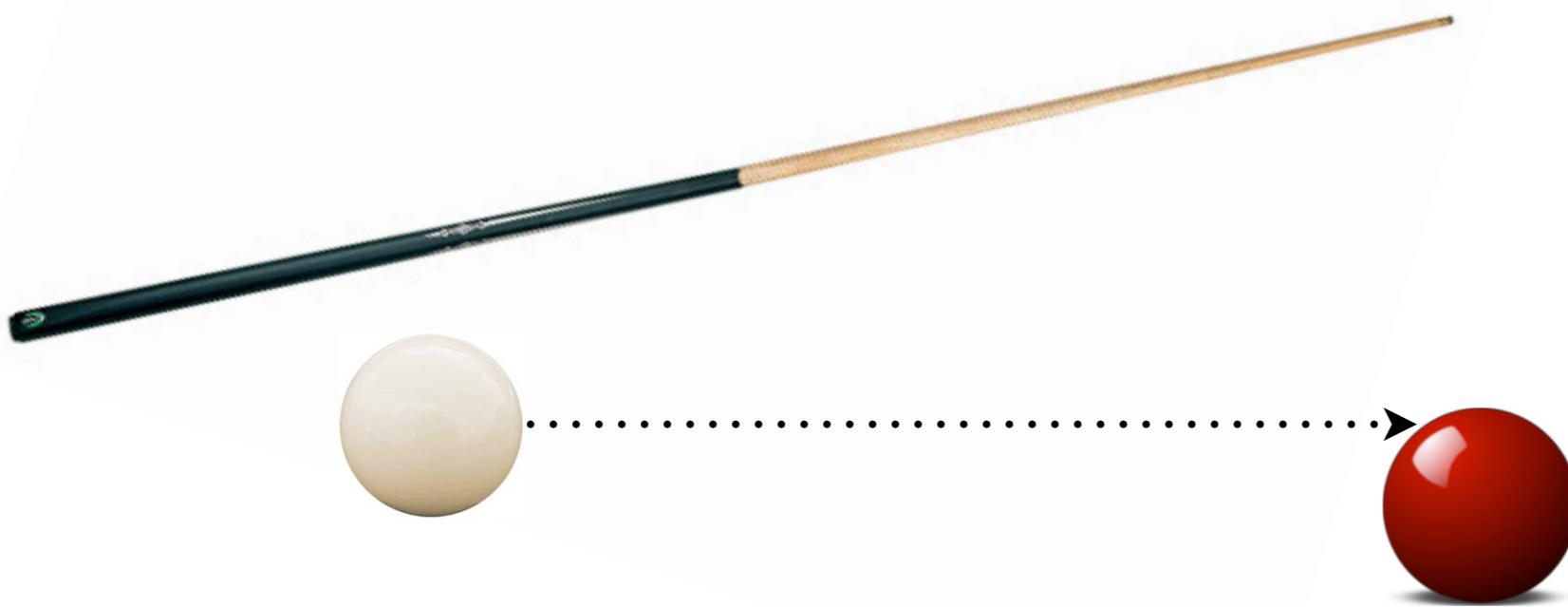
- **1922 - Compton Scattering**
 - Thomson scattering was the classical theory of X-ray scattering.
 - But **high energy X-ray** scattering experiments did **not match** Thomson's predictions in several ways.
 - E.g. back-scattered light was **weaker** than Thomson predicted.
 - Scattered light experienced an **angle-dependent frequency shift**.



The Compton Effect



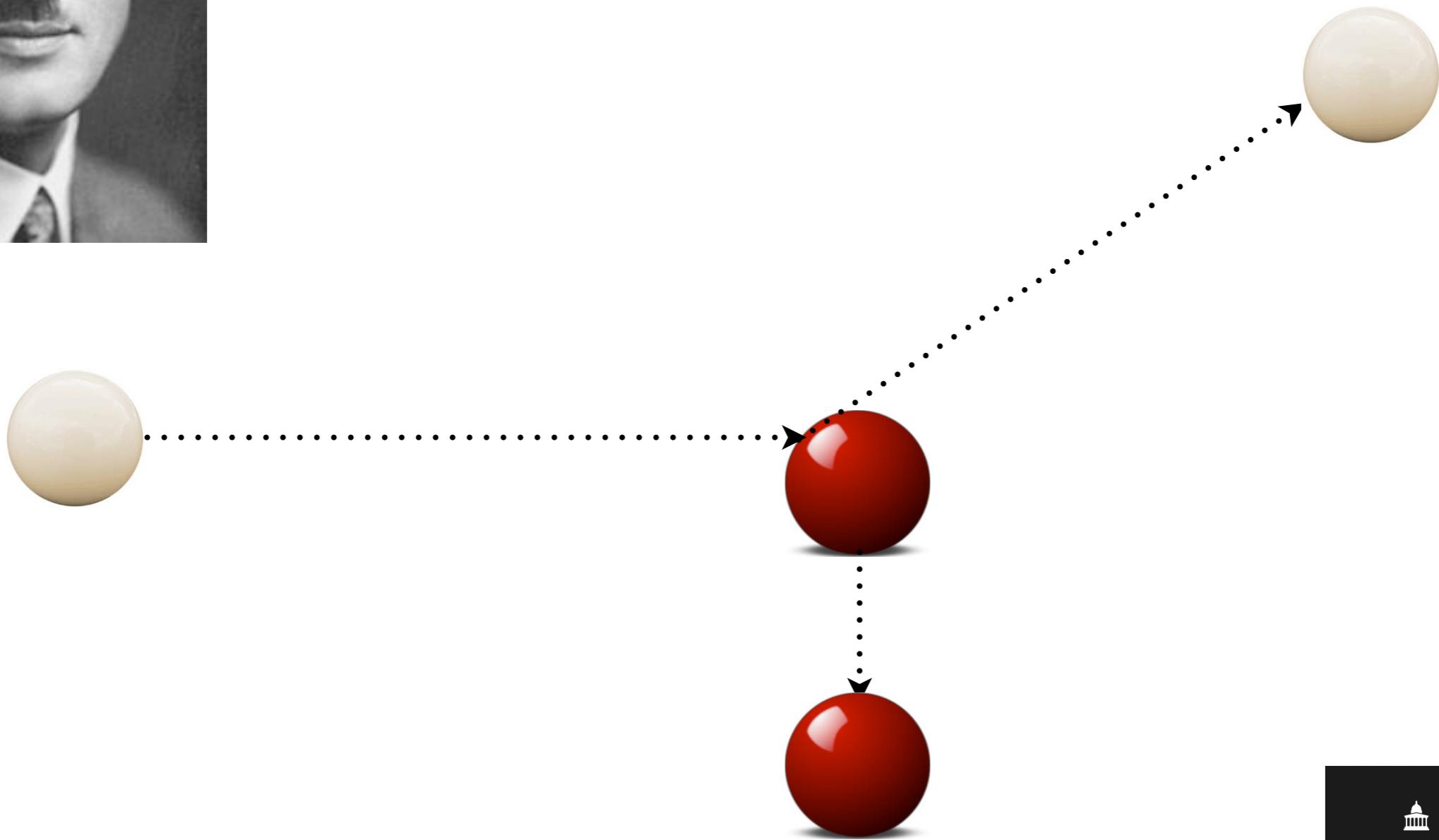
- 1922 - *Compton Scattering*
 - Compton's proposal:
 - Take Planck and Einstein seriously.
 - Treat incoming x-ray as a stream of particles.
 - Model scattering as an elastic collision.



The Compton Effect



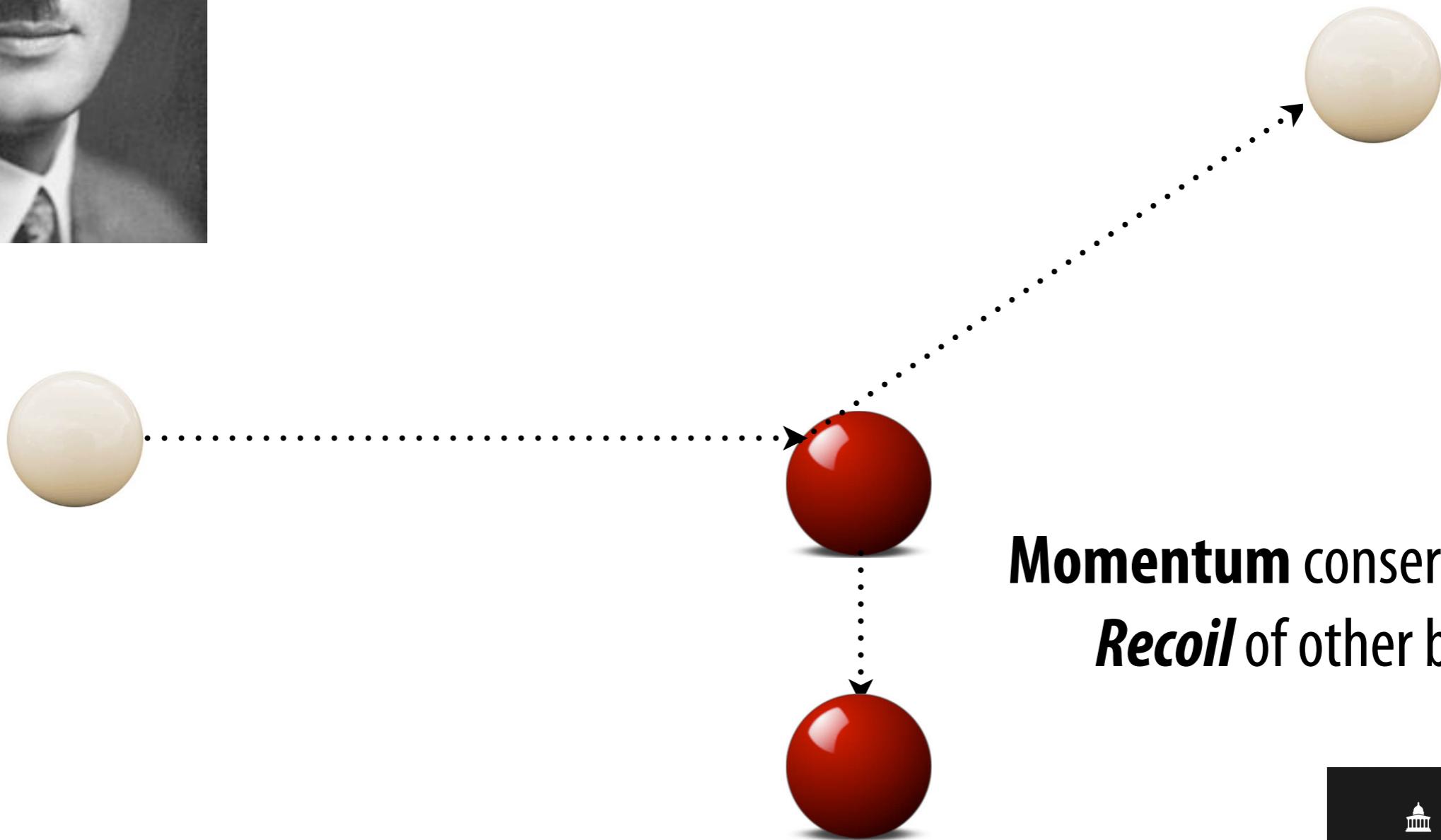
- 1922 - *Compton Scattering*
 - Elastic collision:
 - Momentum and energy conserved.



The Compton Effect



- 1922 - *Compton Scattering*
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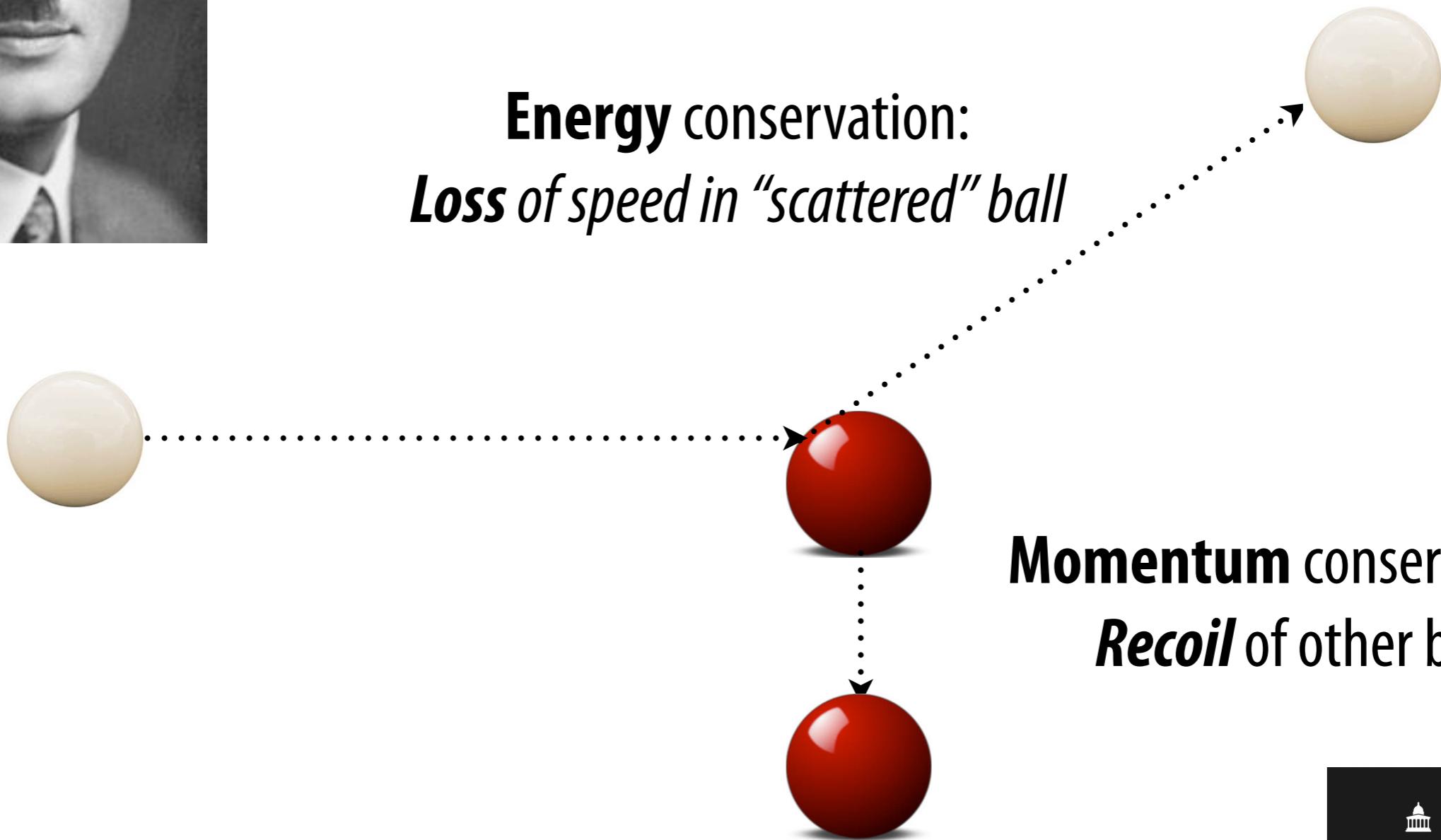


The Compton Effect



- 1922 - *Compton Scattering*
 - Elastic collision:
 - Momentum and energy conserved.

Energy conservation:
Loss of speed in “scattered” ball



Momentum conservation:
Recoil of other ball

The Compton Effect



- **1922 - Compton Scattering**
 - Compton's model:
 - Assume X-ray **photons** with energy and momentum following the **Planck-Einstein** formula:

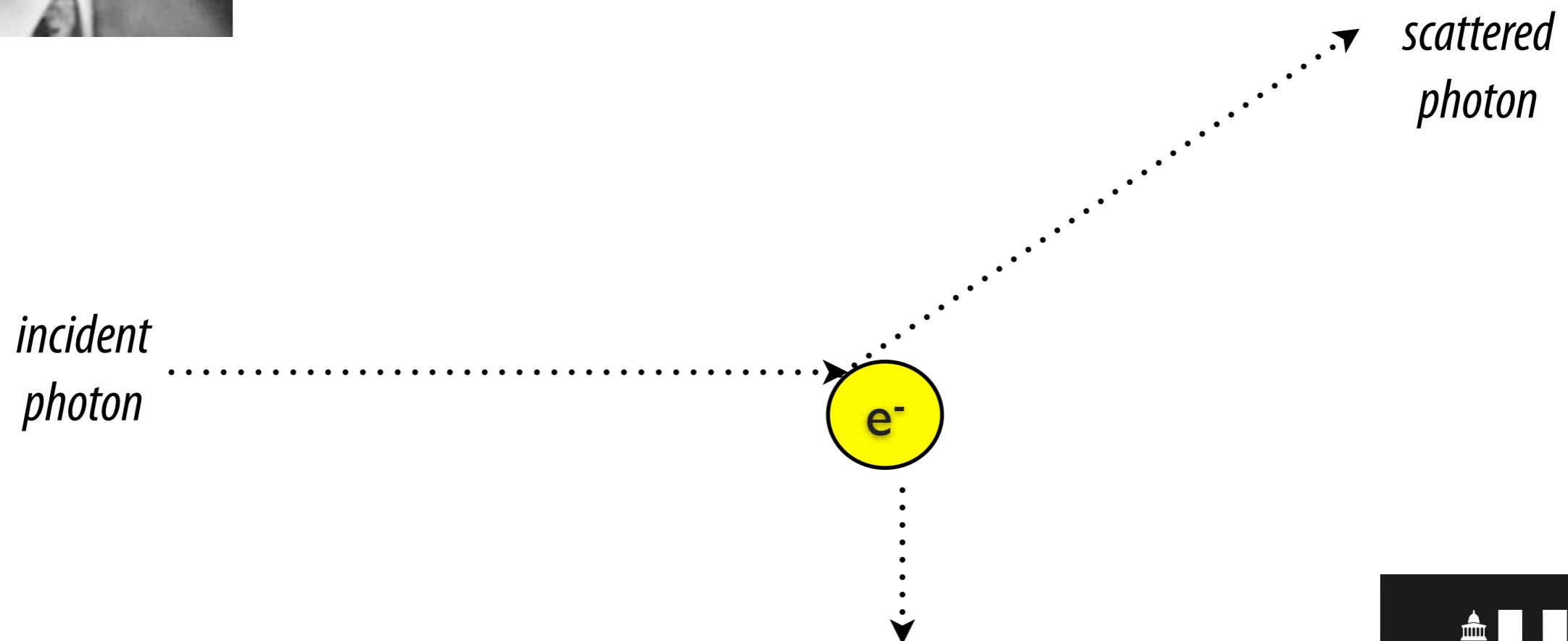
$$E = hf \qquad p = \frac{h}{\lambda}$$

- Treat collision elastically
 - **Momentum and energy conserved.**
 - and treat electron motion using **special relativity.**

The Compton Effect



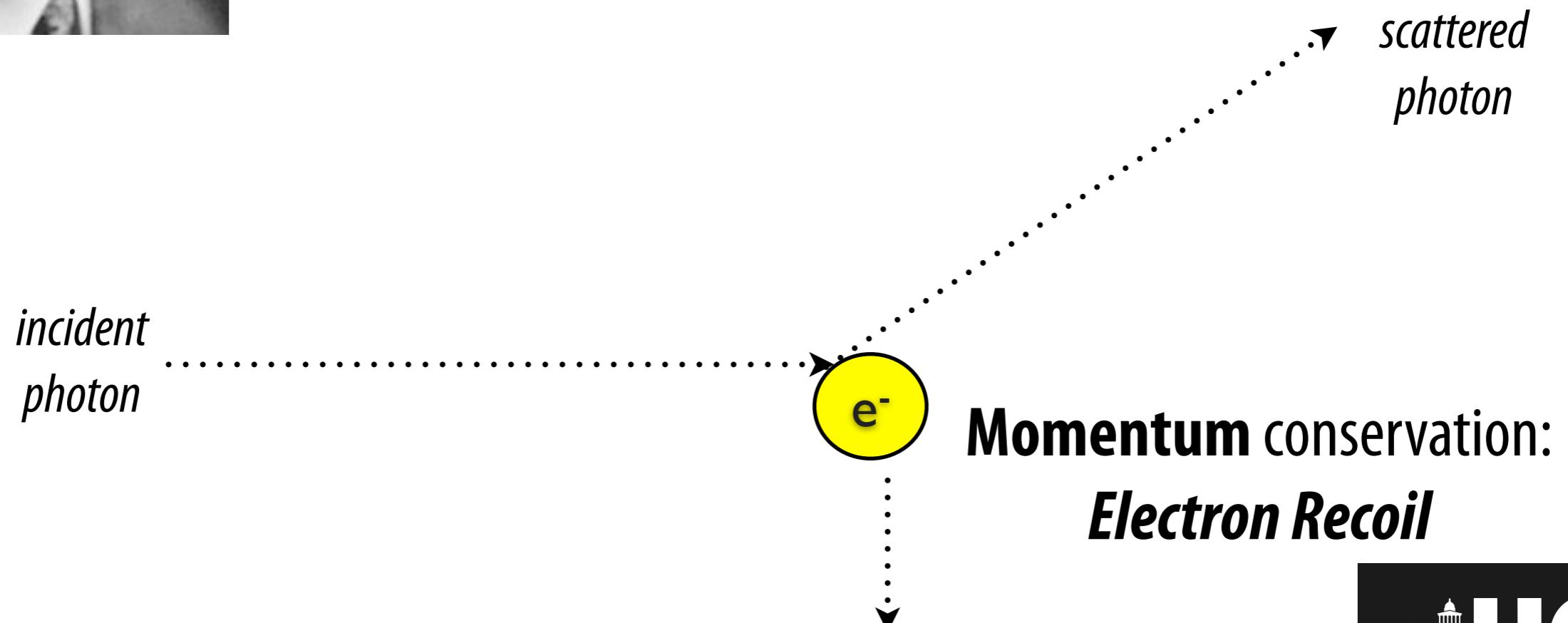
- 1922 - *Compton Scattering*
 - Same basic physics as non-relativistic billiard balls.
 - (Relativistic calculation is actually easier! See Serway and Jewett, Chap 41)



The Compton Effect



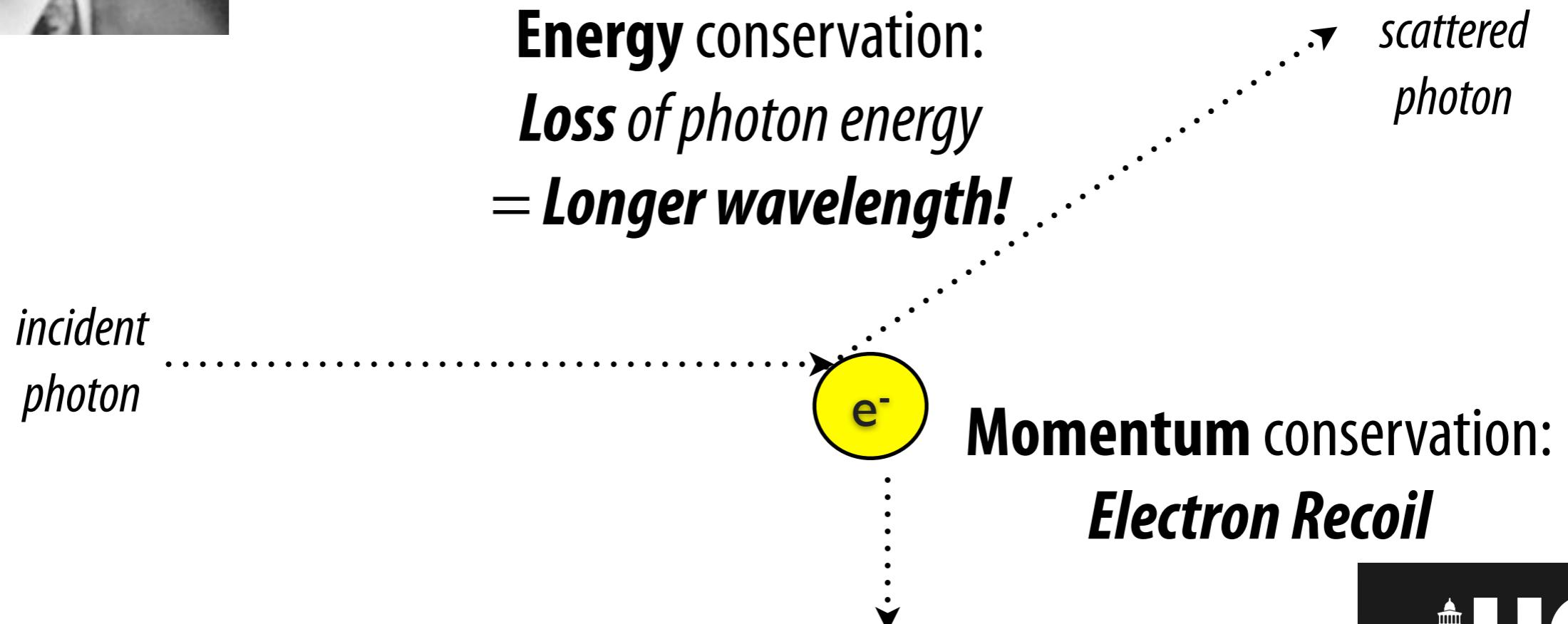
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The Compton Effect

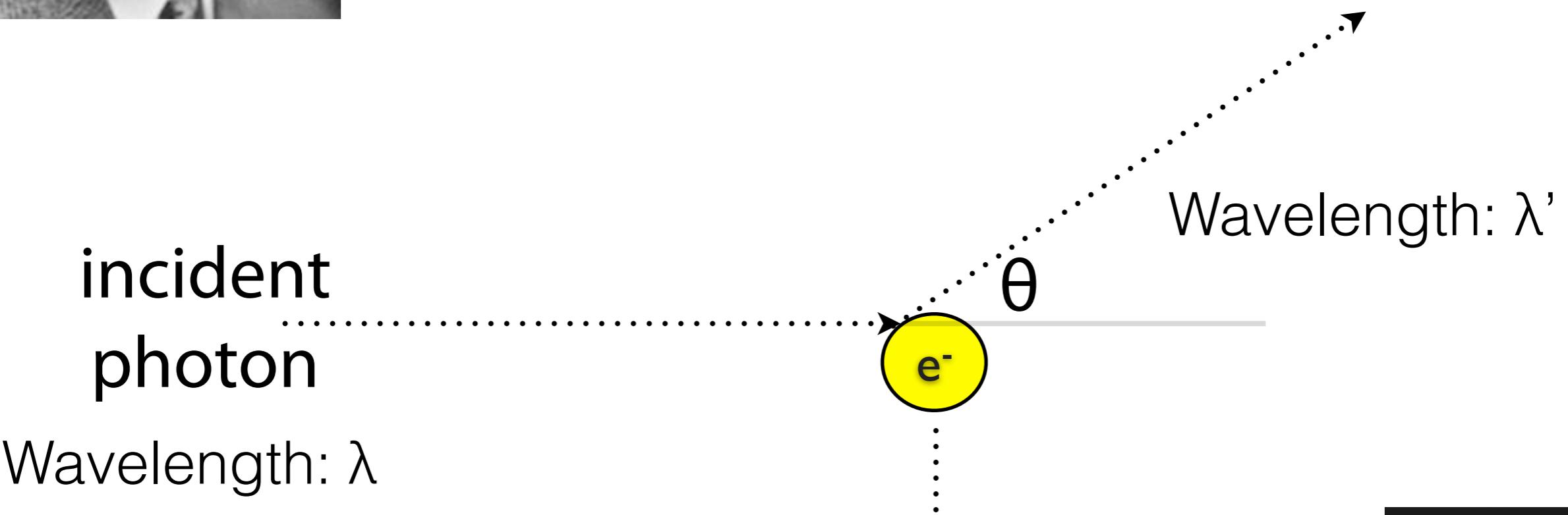


- 1922 - *Compton Scattering*

- Compton scattering formula:

$$\lambda' - \lambda = \frac{h}{m_e c} (1 - \cos \theta)$$

- where m_e is electron mass.
- **Wavelength shift depends on angle only.**



The Compton Effect



- 1922 - *Compton Scattering*

- Compton scattering formula:

$$\lambda' - \lambda = \frac{h}{m_e c} (1 - \cos \theta)$$

- where m_e is electron mass.
 - **Wavelength shift depends on angle only.**
 - For an electron mass:

$$\frac{h}{m_e c} \approx 2 \times 10^{-12} \text{m} = 2 \text{pm}$$

- Shift is only **measurable** when photon wavelength a few orders of magnitude above a picometre - e.g. x-rays (nanometre wavelength).

The Compton Effect



- **1922 - Compton Scattering**
 - Compton verified this wavelength shift **experimentally** scattering x-rays from a variety of different conductors.
 - In all cases, experiment closely **matched his theory** and disagreed with wave scattering!

From A.H. Compton, Physical Review 1923:

$$\lambda_\theta - \lambda_0 = 0.022 \text{ \AA} \quad (\text{experiment}).$$

But according to the present theory (Eq. 5),

$$\lambda_\theta - \lambda_0 = 0.0484 \sin^2 45^\circ = 0.024 \text{ \AA} \quad (\text{theory}),$$

which is a very satisfactory agreement.

The Compton Effect



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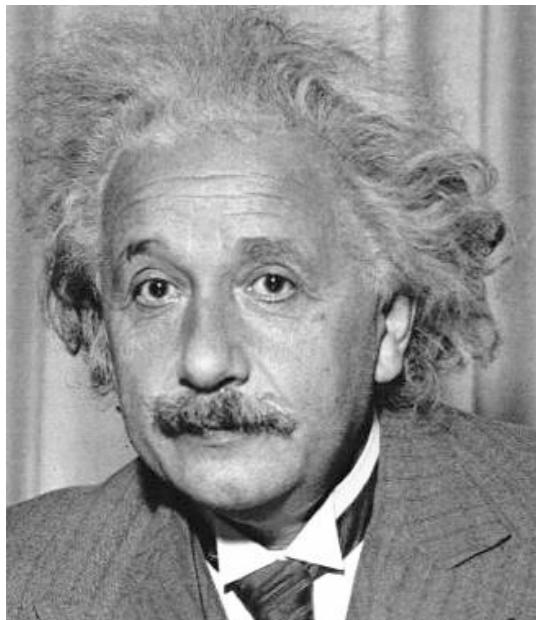
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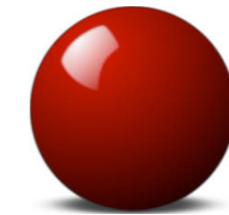
A particle and a wave?



- Between them, Einstein's **photoelectric effect** argument and Compton's scattering experiments were unambiguous!
 - Light as a **particle** had to be taken seriously!
 - But **interferometry** experiments "proved" that light a **wave**.
- How could light be **both a particle and a wave?**
 - Answer: **Quantum Mechanics**

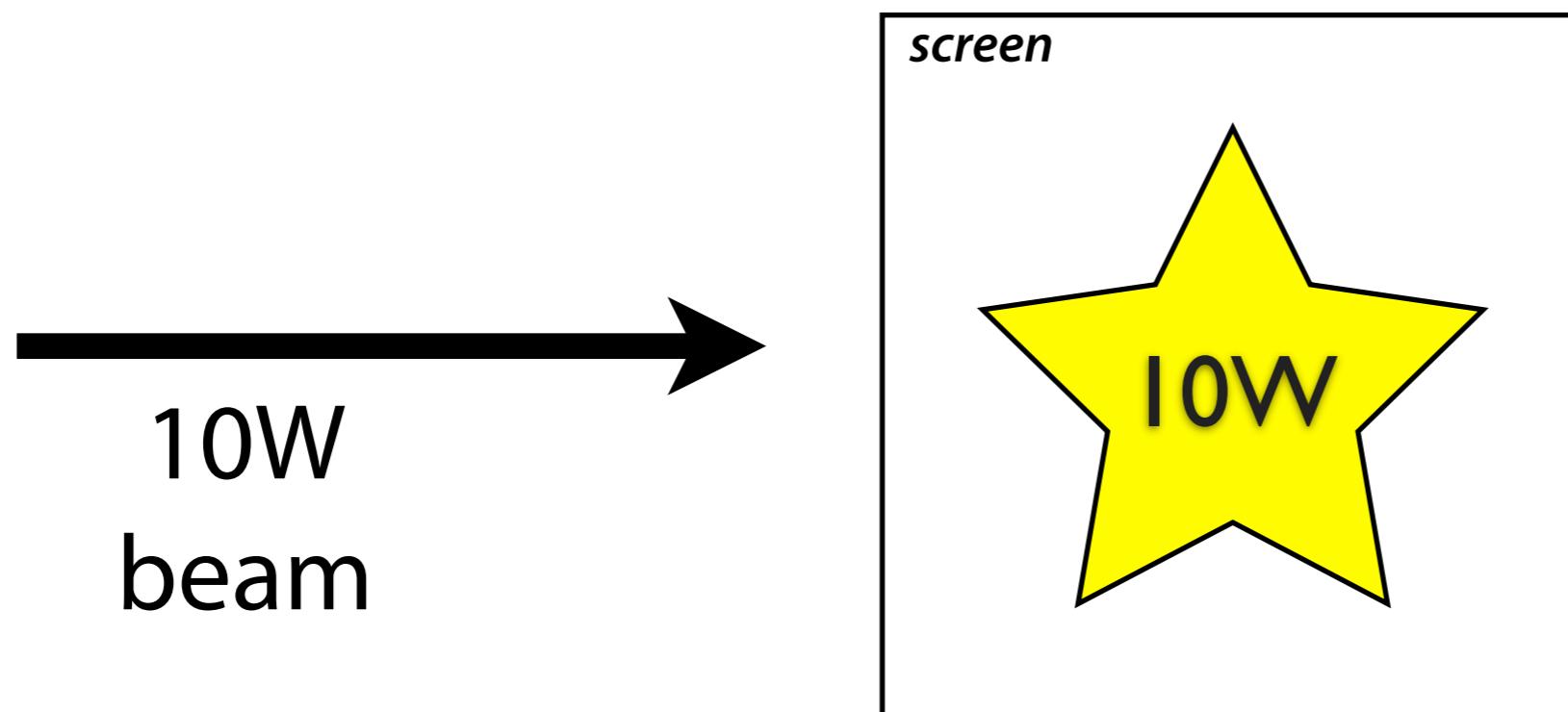


Wave-like and particle-like behaviour - the photon



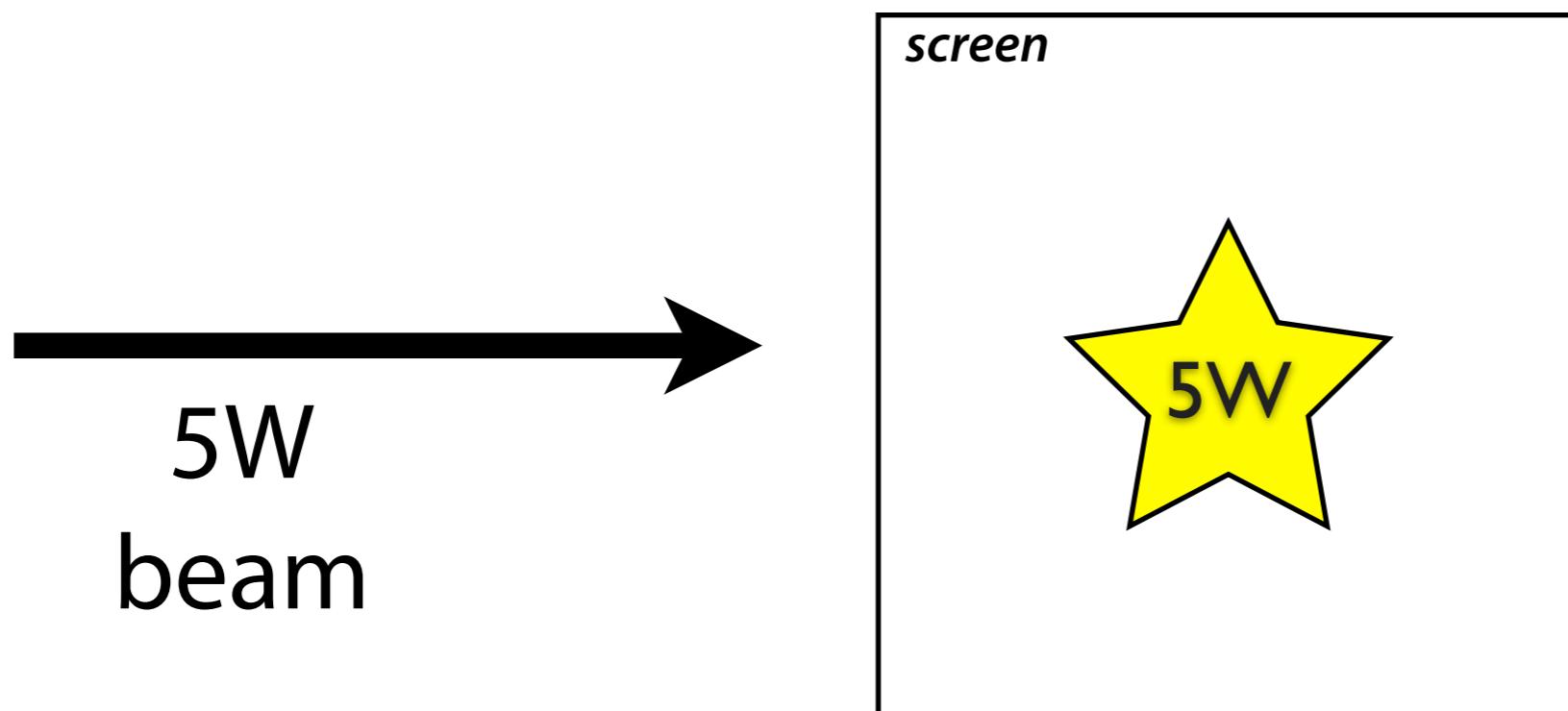
Towards single photons

- What happens if we turn down the power of a laser?
- And keep on turning it down?



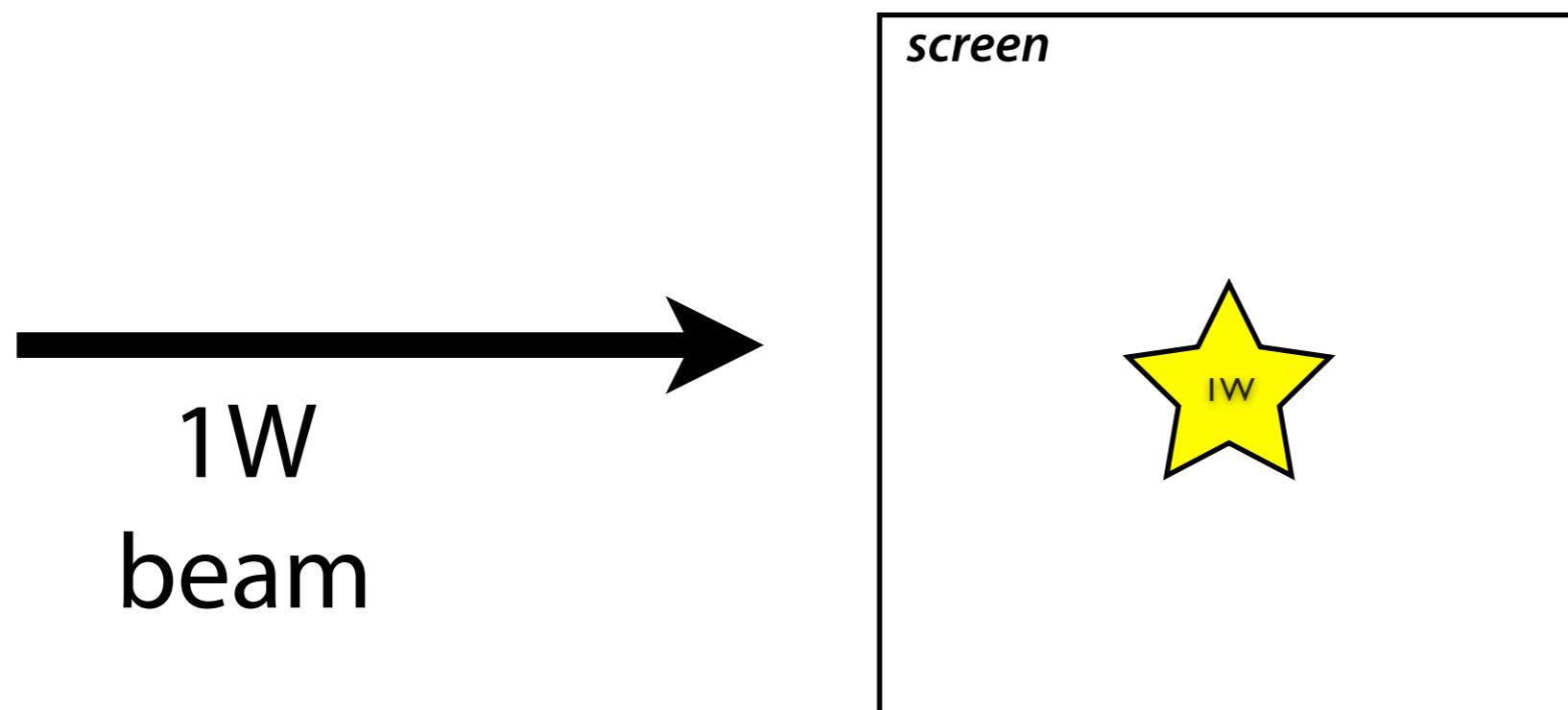
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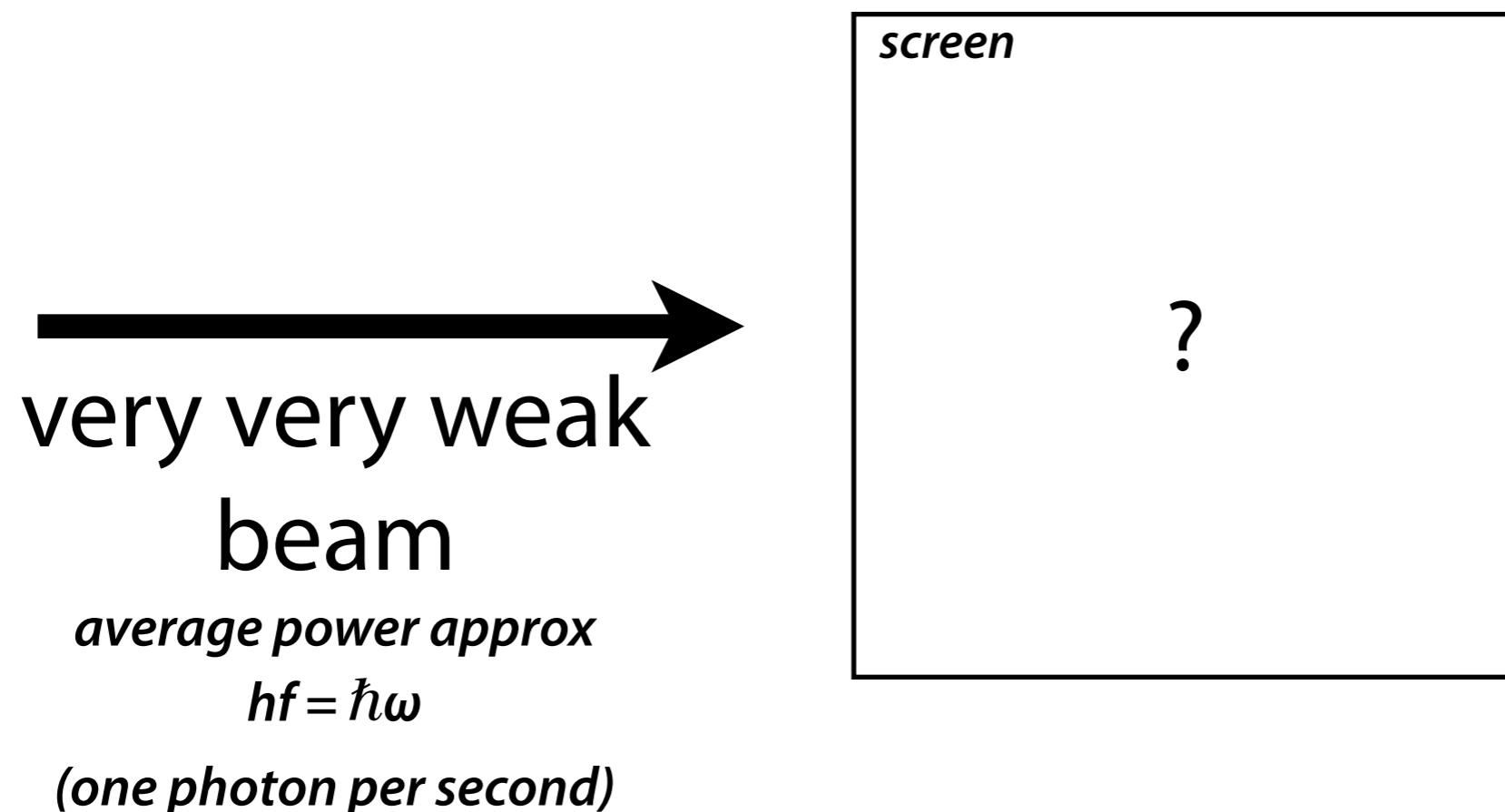
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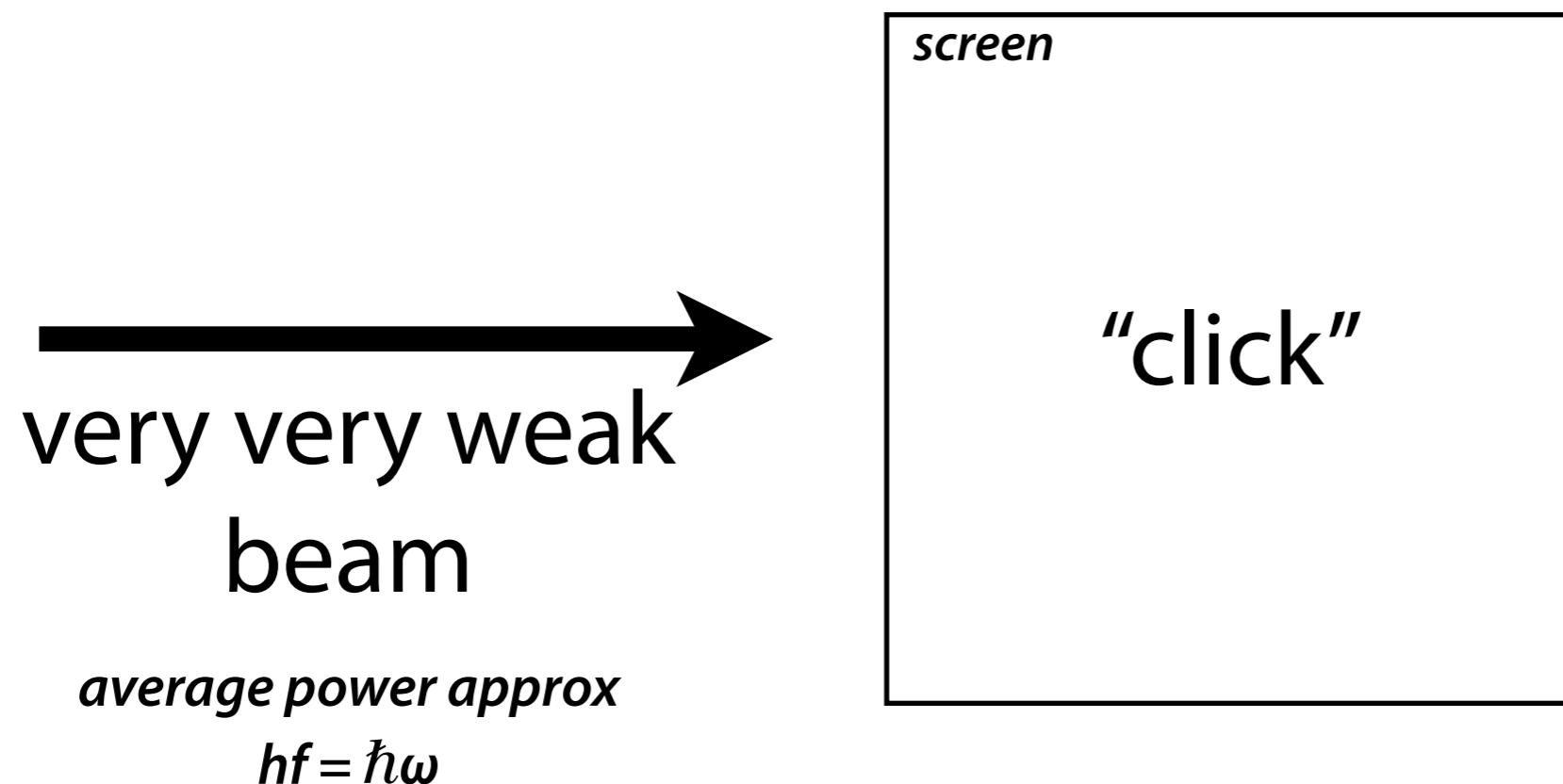


Freq. of a yellow photon: $f \approx 5 \times 10^{14} \text{ s}^{-1}$

Energy of a yellow photon: $hf \approx 3 \times 10^{-19} \text{ J}$

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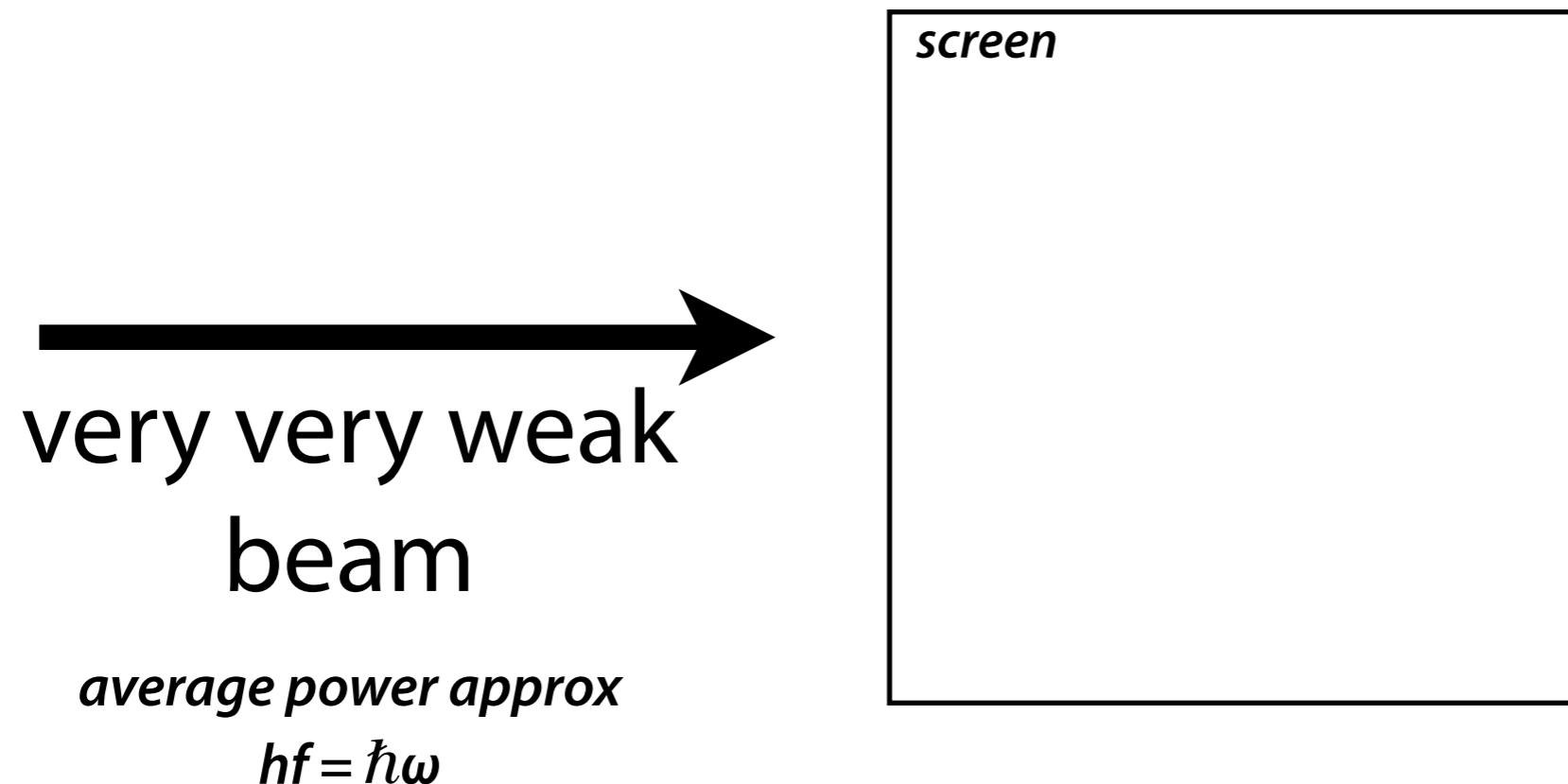


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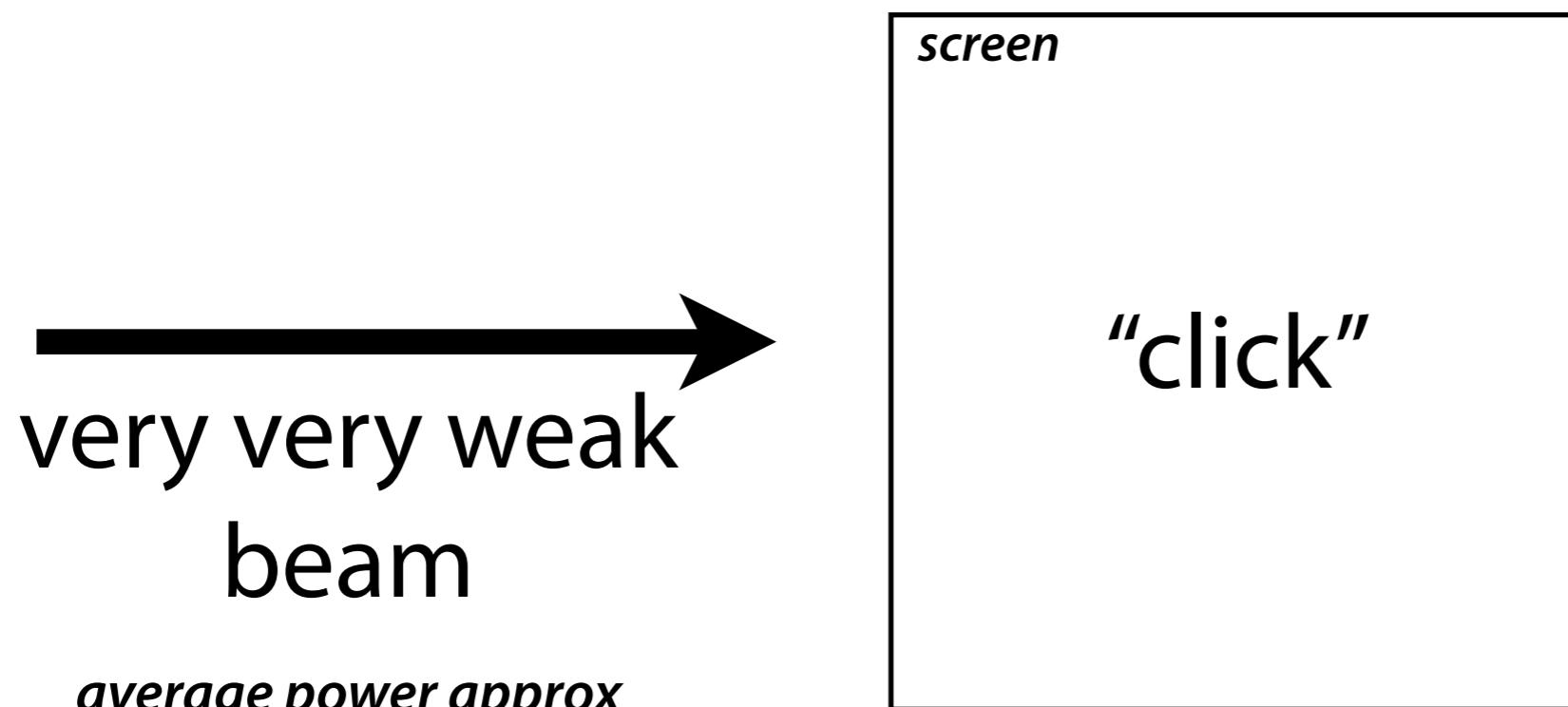


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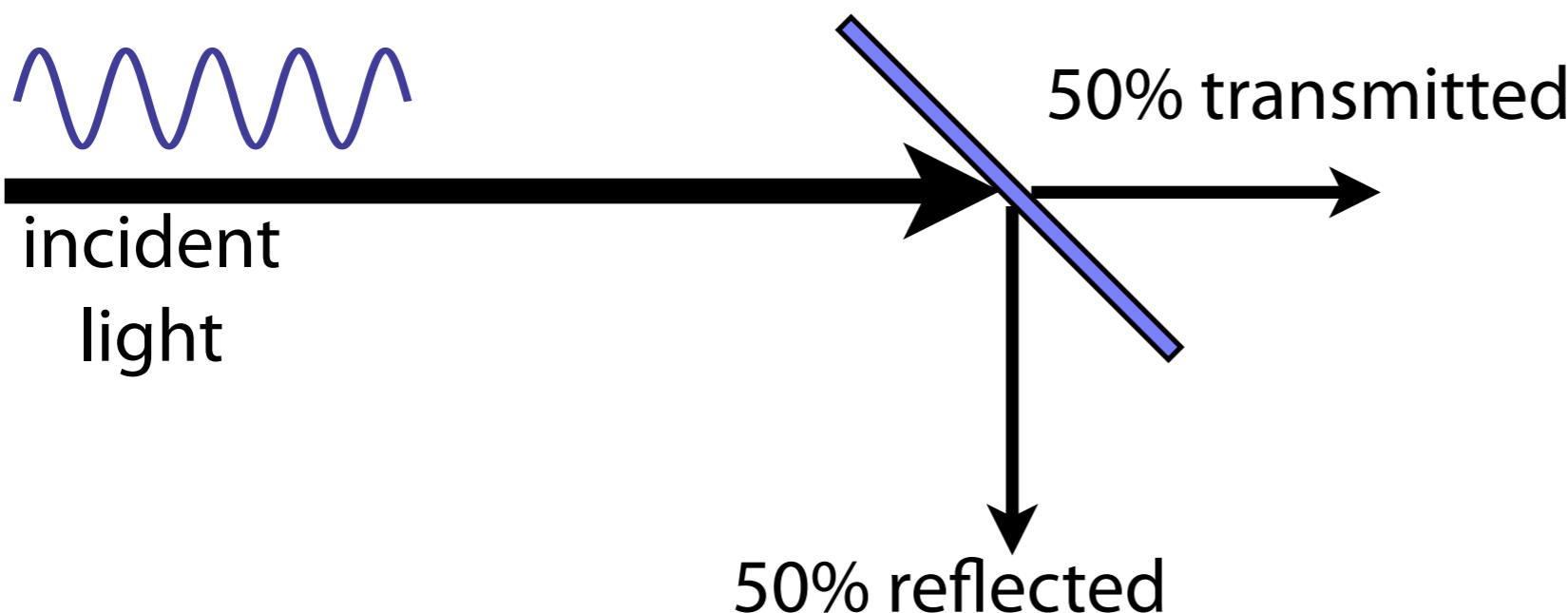


Freq. of a yellow photon: $f \approx 5 \times 10^{14} \text{ s}^{-1}$
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- Detection no longer continuous.
- Detector “clicks” at random intervals, on average once per second.

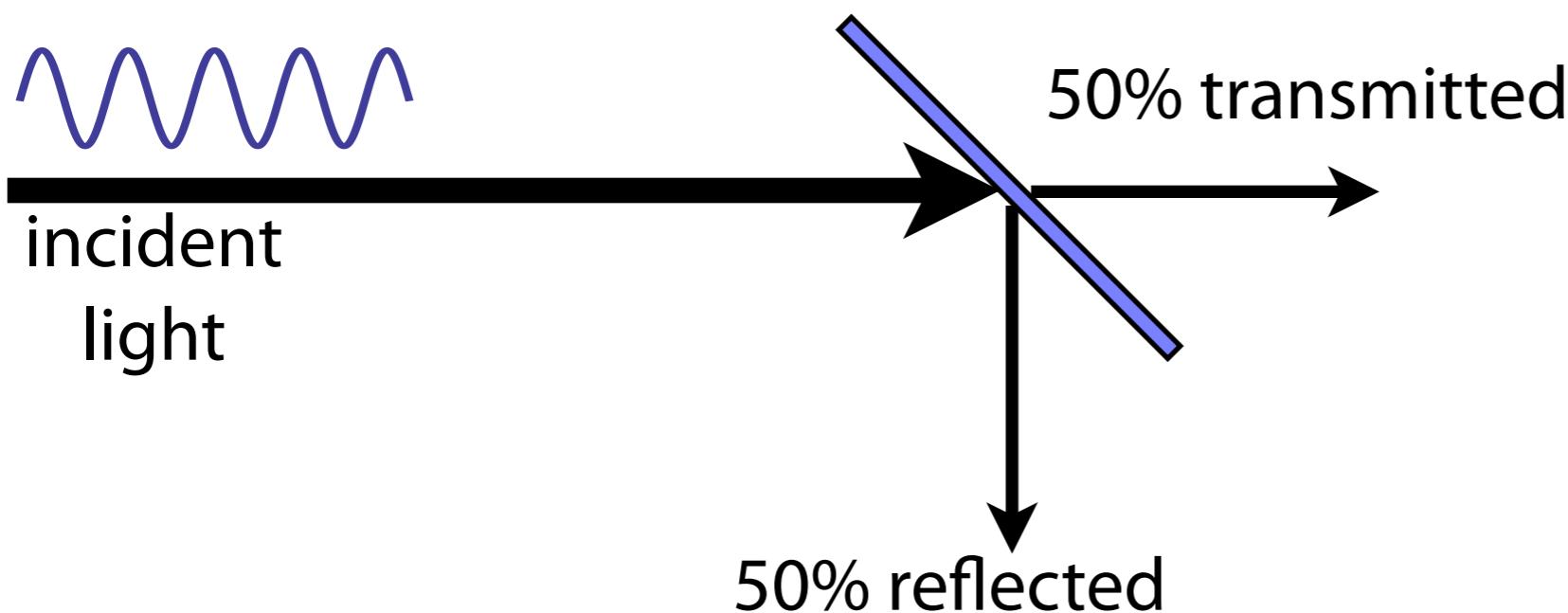
Photons on a beam splitter

- Recall the beam splitter
 - *50% light transmitted*
 - *50% light reflected*



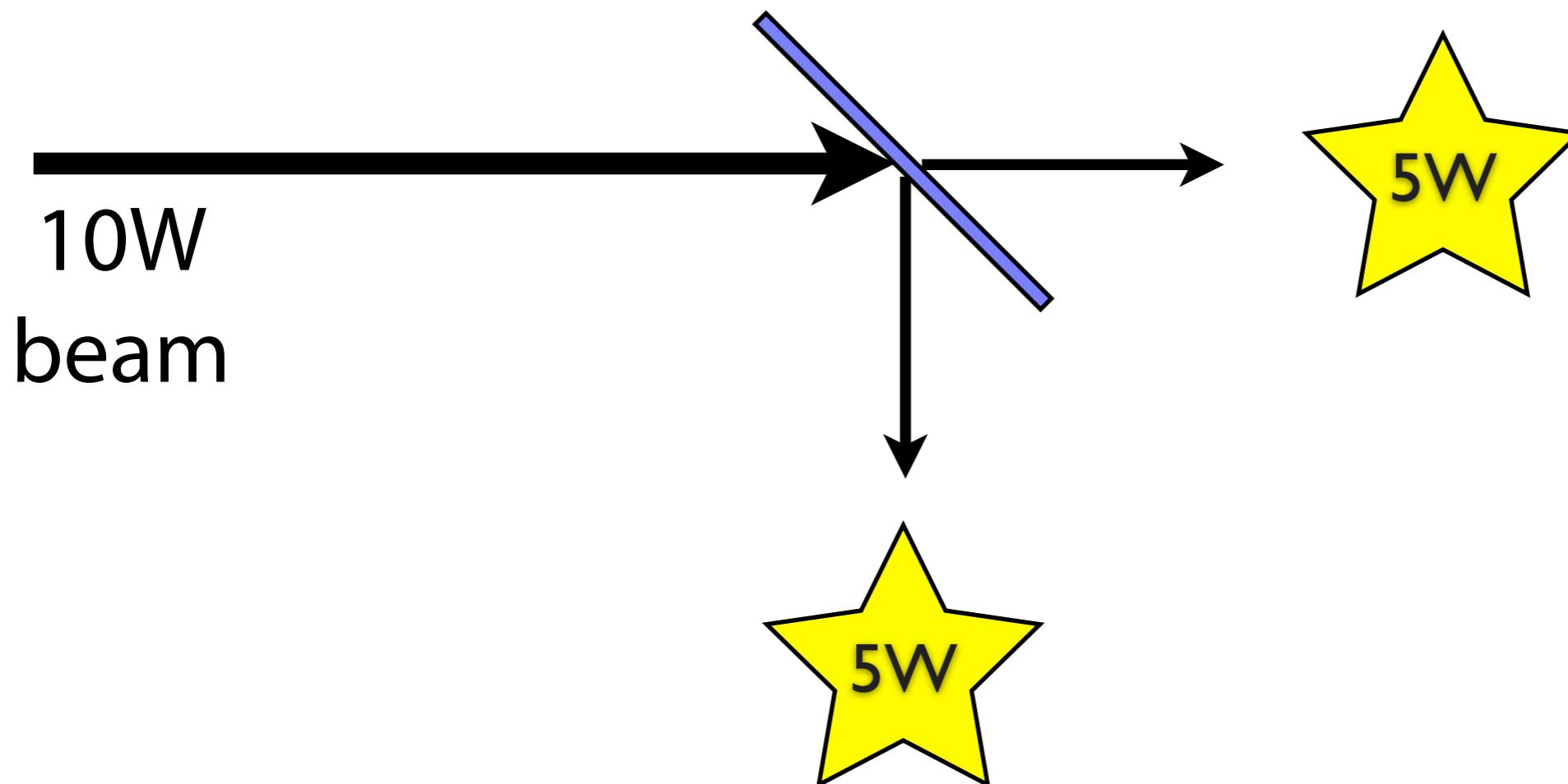
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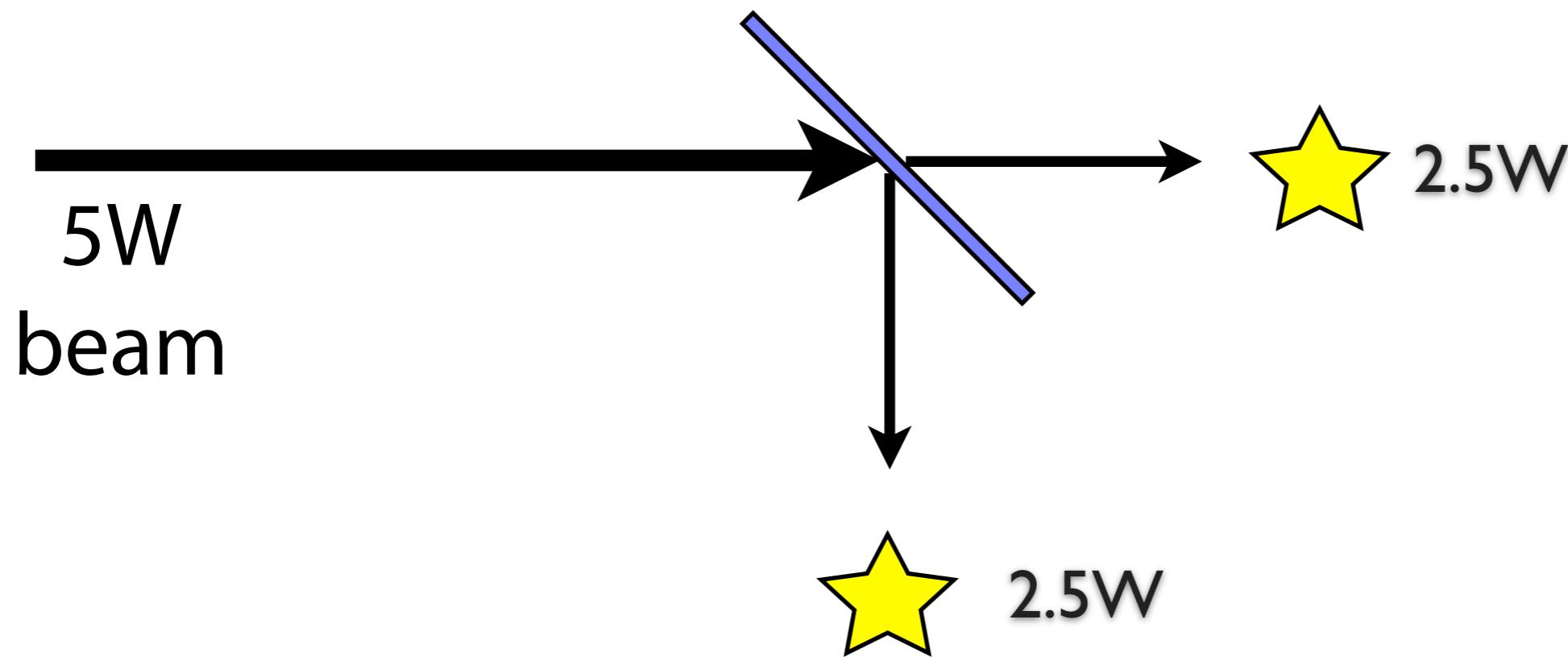
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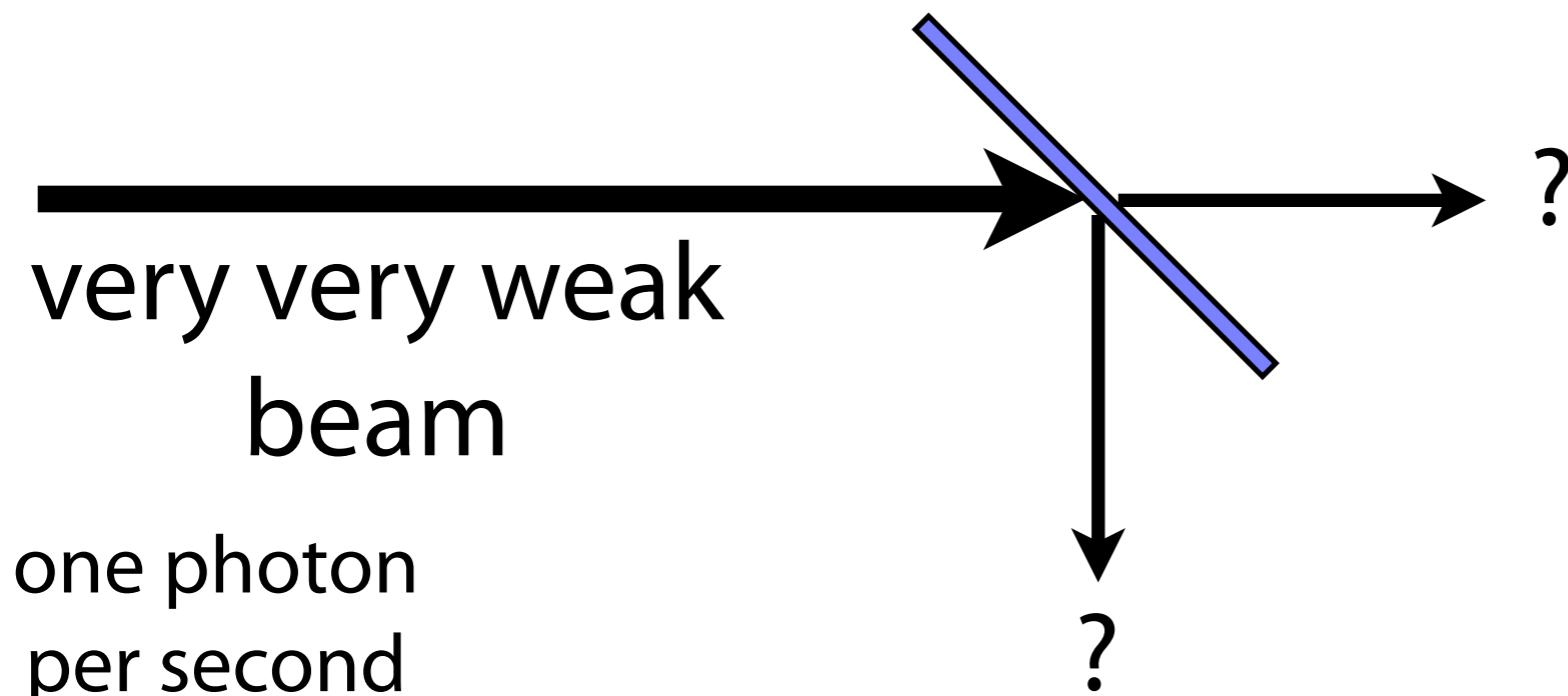
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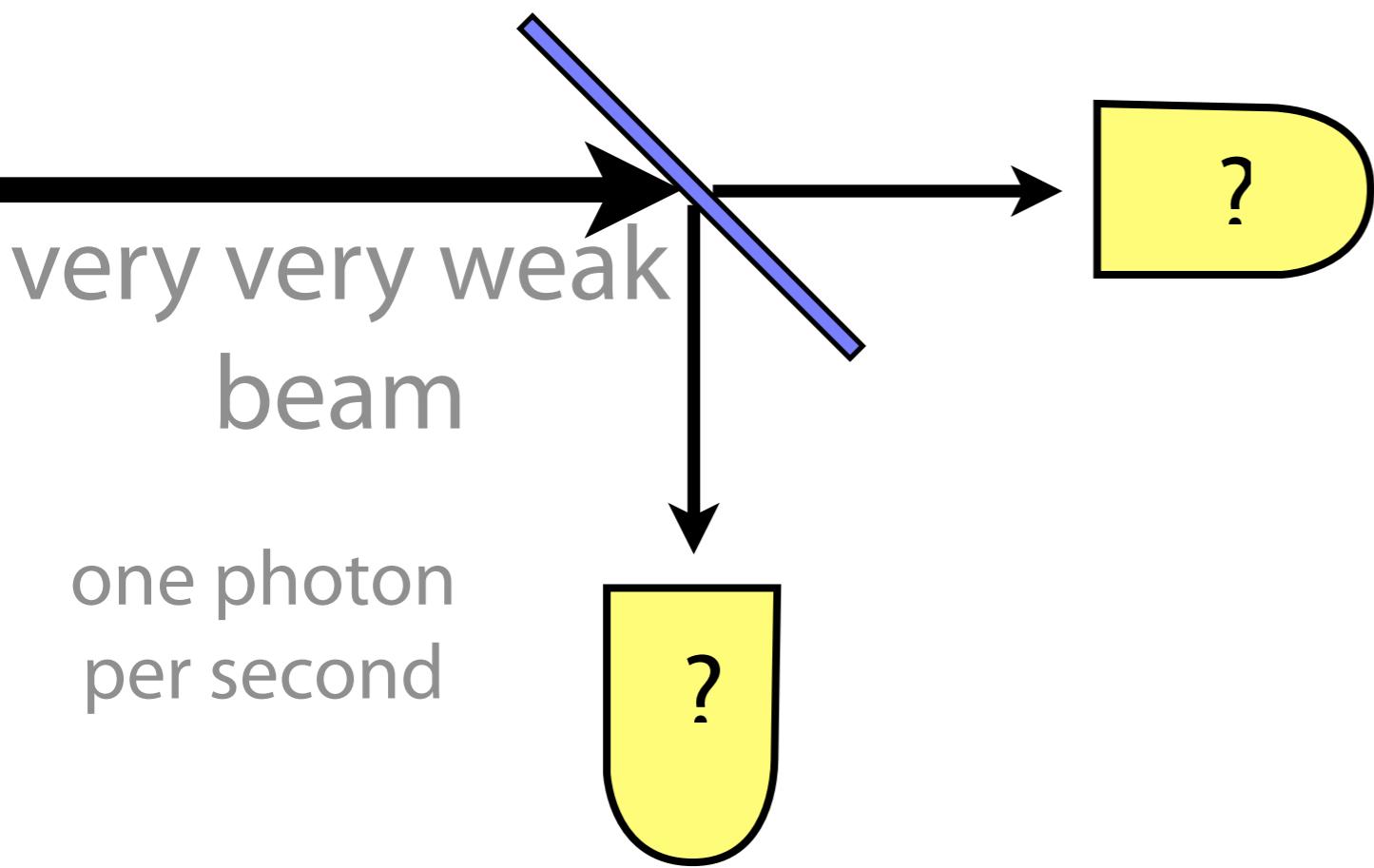


Photons on a beam splitter

- Recall the beam splitter
 - *50% light transmitted*
 - *50% light reflected*



Quiz



- ***What will we see?***
- 1. Clicks always come in **pairs**.
- 2. **Only one** detector clicks, and always the **same** detector.
- 3. Only one detector clicks, and which detector clicks is **random**.

Photons on a beam splitter

- Single photons **cannot be divided**.
- **Half** of the photons are detected in one beam and **half** in the other.
- The probabilities of each case (50:50) are reminiscent of the intensities of a strong classical beam.

With 50% probability we see:

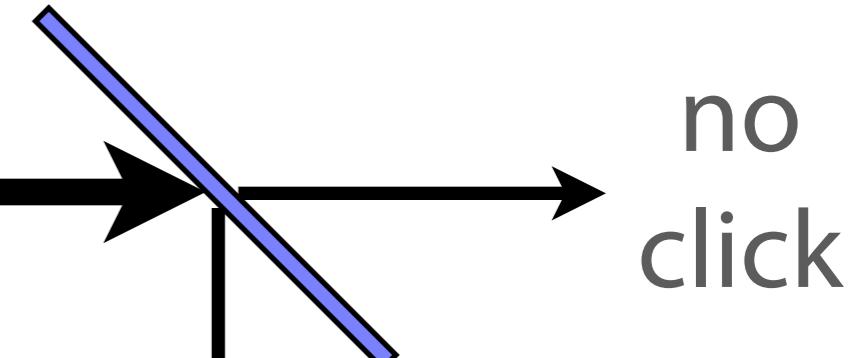
very very weak
beam

one photon

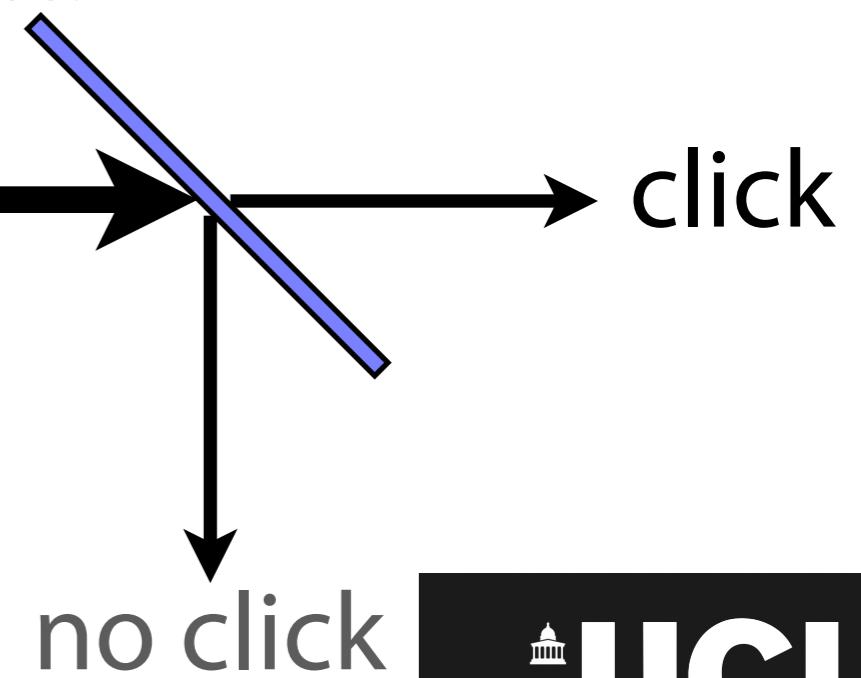
With 50% probability we see:

very very weak
beam

one photon



click



Classical limit and Quantum limit

1
photon

 10^{15}
photons
/ sec



“click”

- **Quantum Limit**
 - Small numbers of photons
 - Quantum behaviour dominates
 - Quantum mechanics **essential**
- **Classical Limit**
 - Very large numbers of photons
 - Quantum behaviour “averages away”
 - Classical physics a **very good approximation**.

Classical limit and Quantum limit

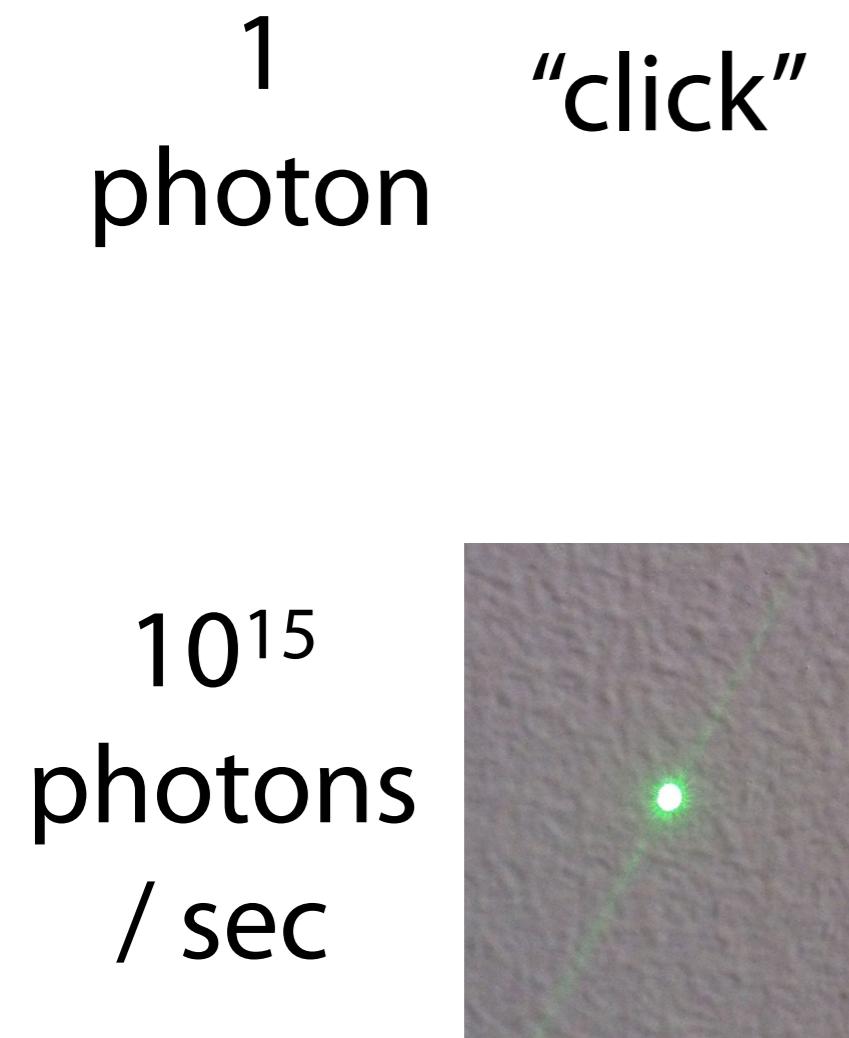
1 “click”
photon

10¹⁵
photons / sec



- Quantum and classical physics should (often but not always) be consistent.
- Classical physics works very well for most physical situations at “large” scales.
- So often if we consider average behaviour of large numbers of quantum particles, we should recover classical behaviour.
- *But not always: Exceptions will be* phenomena not well described by classical physics
 - E.g. Black-body radiation
 - photo-electric effect.

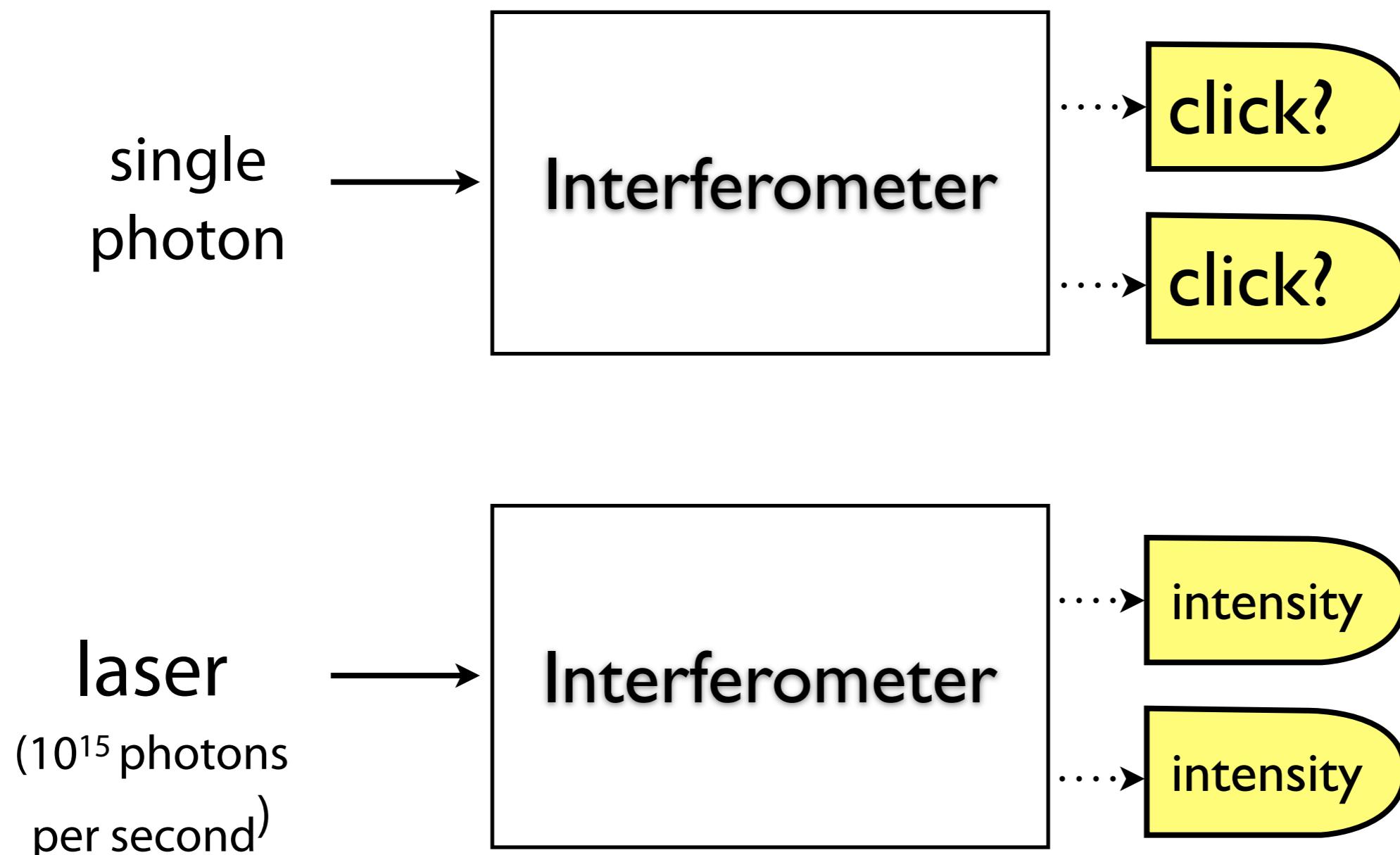
Classical limit and Quantum limit



- But single photon behaviour is a good example of this.
- If we take **large numbers** of photons, we'll no longer see the "granularity" of individual photons.
 - E.g. can you tell the difference between **10^{15} photons** and **$10^{15} - 1$ photons**?
- Classical physics reproduces the **average behaviour** of single photons in this case.

Single photon vs laser

- Consider two almost identical interference experiments.



Single photon vs laser

- Consider two almost identical interference experiments.

Quantum Limit



Classical Limit



Single photon vs laser

- The classical experiment, can be thought of as 10^{15} repetitions of the quantum experiment.

Quantum Limit



Classical Limit



Quiz

- If, after repeating the single photon experiment we observe the following probabilities of clicks in the detectors:

Quantum Limit



- What will be the intensities observed in the same experiment with a 10W laser?
 - 1. Top: 5 W Bottom: 5 W
 - 2. Top: 9.5 W, Bottom: 0.5W
 - 3. Something else

Quiz

- If, after repeating the single photon experiment we observe the following probabilities of clicks in the detectors:

Quantum Limit



- What will be the intensities observed in the same experiment with a 10W laser?
 - 1. Top: 5 W Bottom: 5 W
 - 2. **Top: 9.5 W, Bottom: 0.5W**
 - 3. Something else

Single photon probability rule

- In general, in the intensities in the classical experiment are proportionate to probabilities in the equivalent single photon experiment.

Quantum Limit

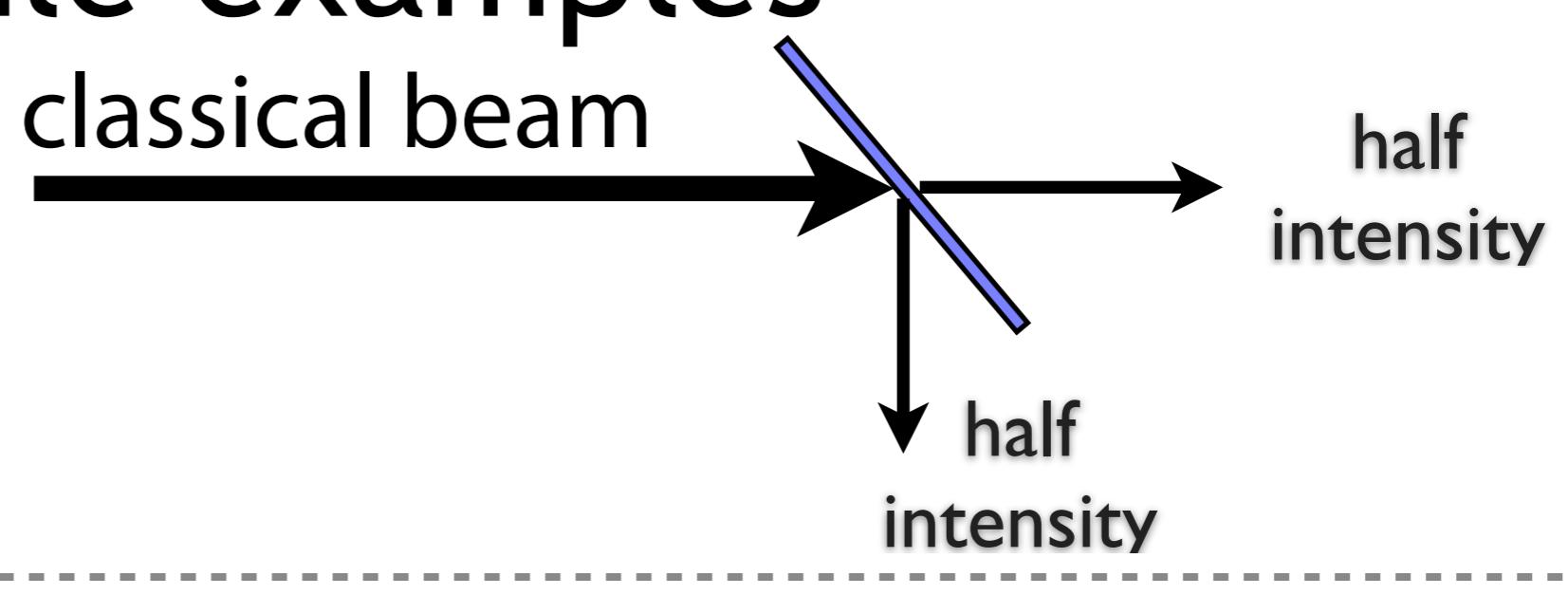


Classical Limit

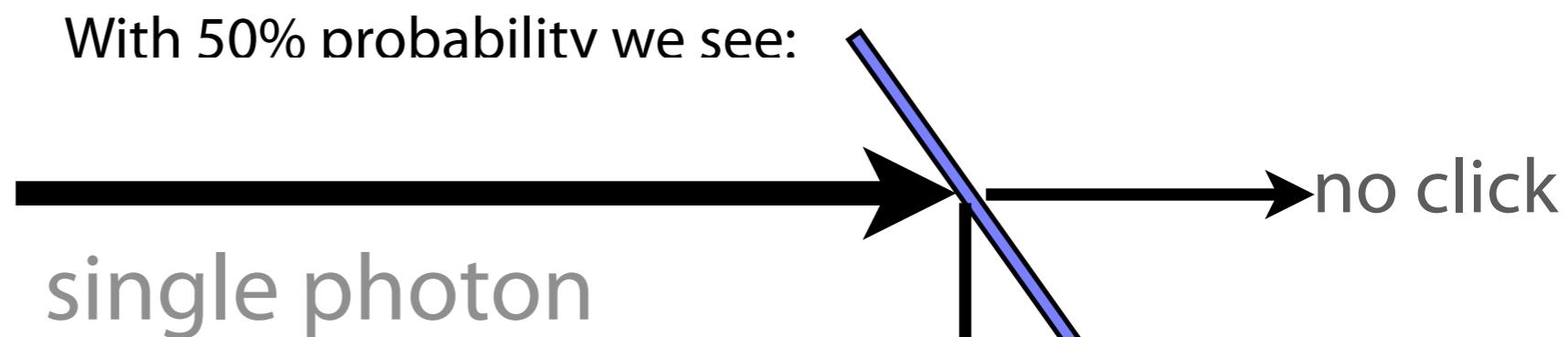


Probability rule examples

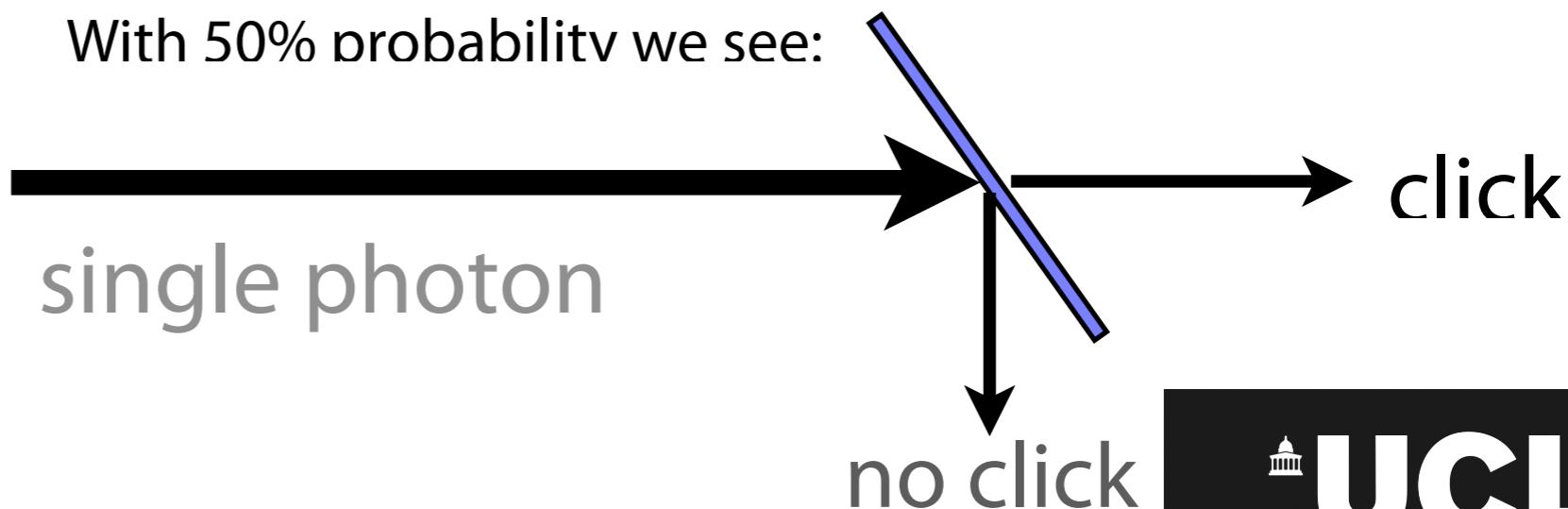
- Using this probability rule, we can look again at a single photon on a beam splitter.
- Classical intensity is split 50:50.
- Photon can only click in 1 detector.
- Hence there must be a 50% probability for a click in each detector.



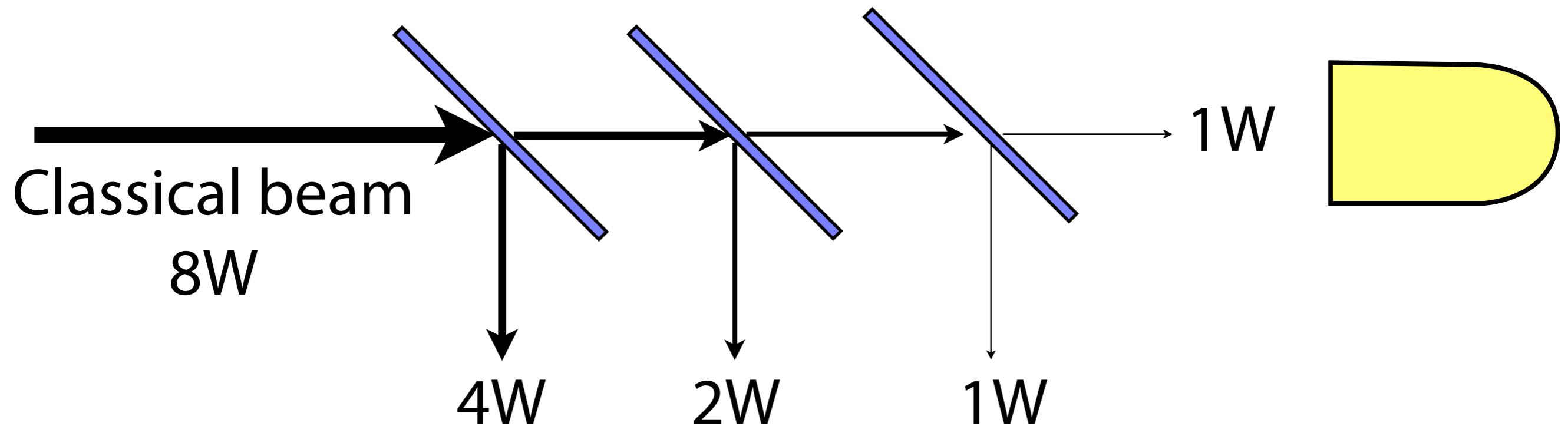
With 50% probability we see:



With 50% probability we see:



Quiz



- If we repeat this experiment with a single photon, what is the **probability** that the photon **clicks** in the “yellow detector”.

- 1. 100%
- 2. 50%
- 3. 25%
- 4. 12.5%

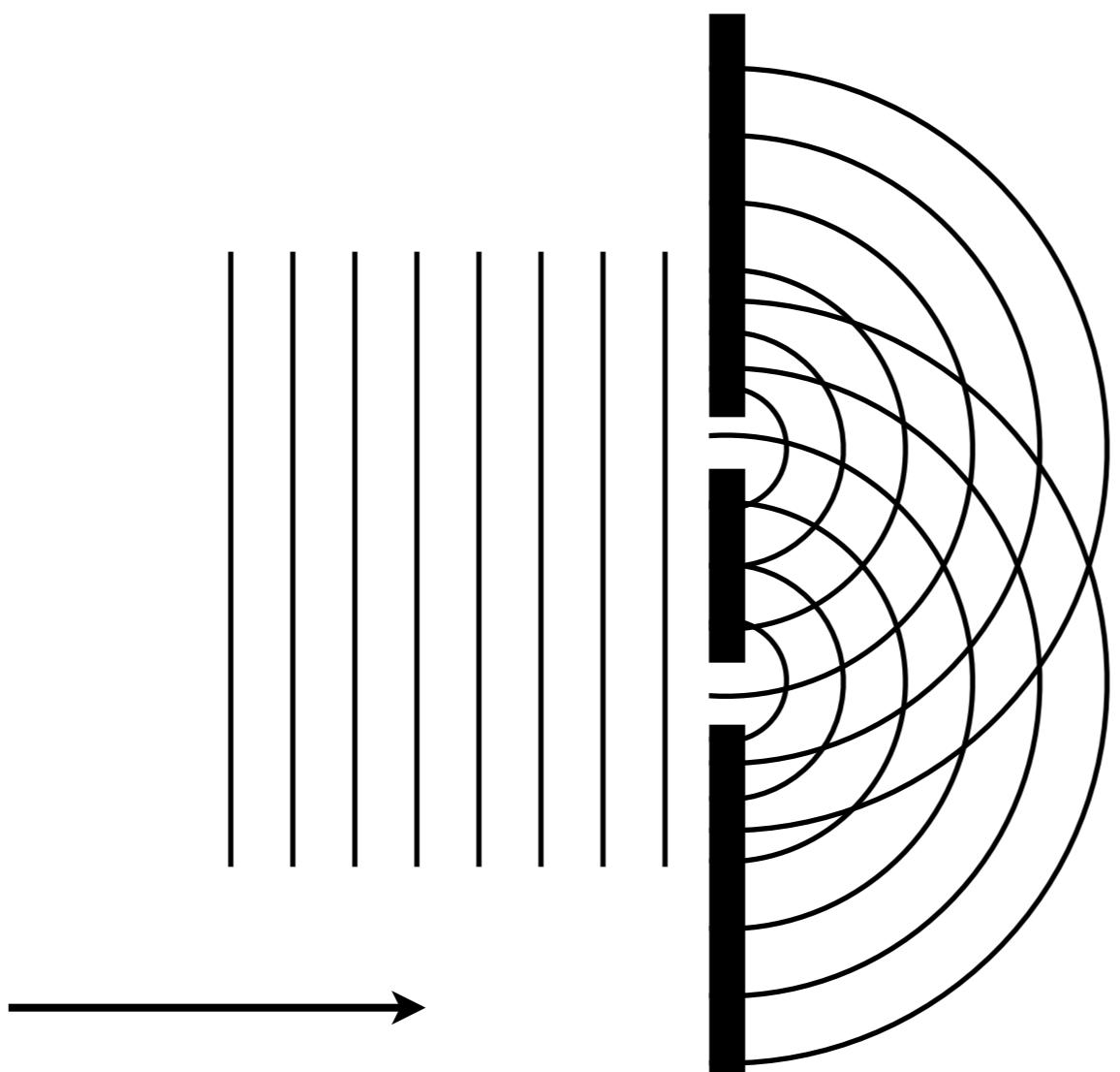
The Young Two-slit experiment

With classical light:



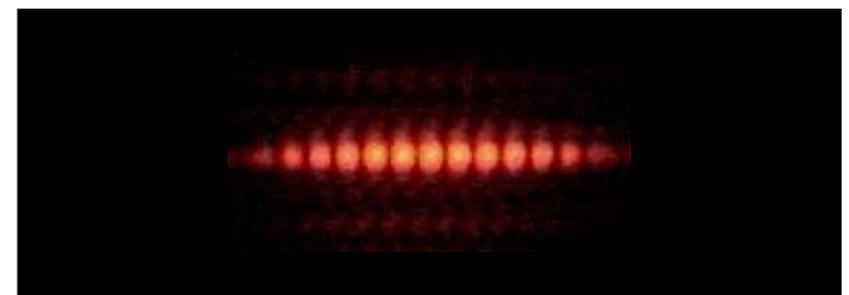
T. Young

Opaque sheet Screen



Collimated
monochromatic light Two
small slits

What we observe
on the screen

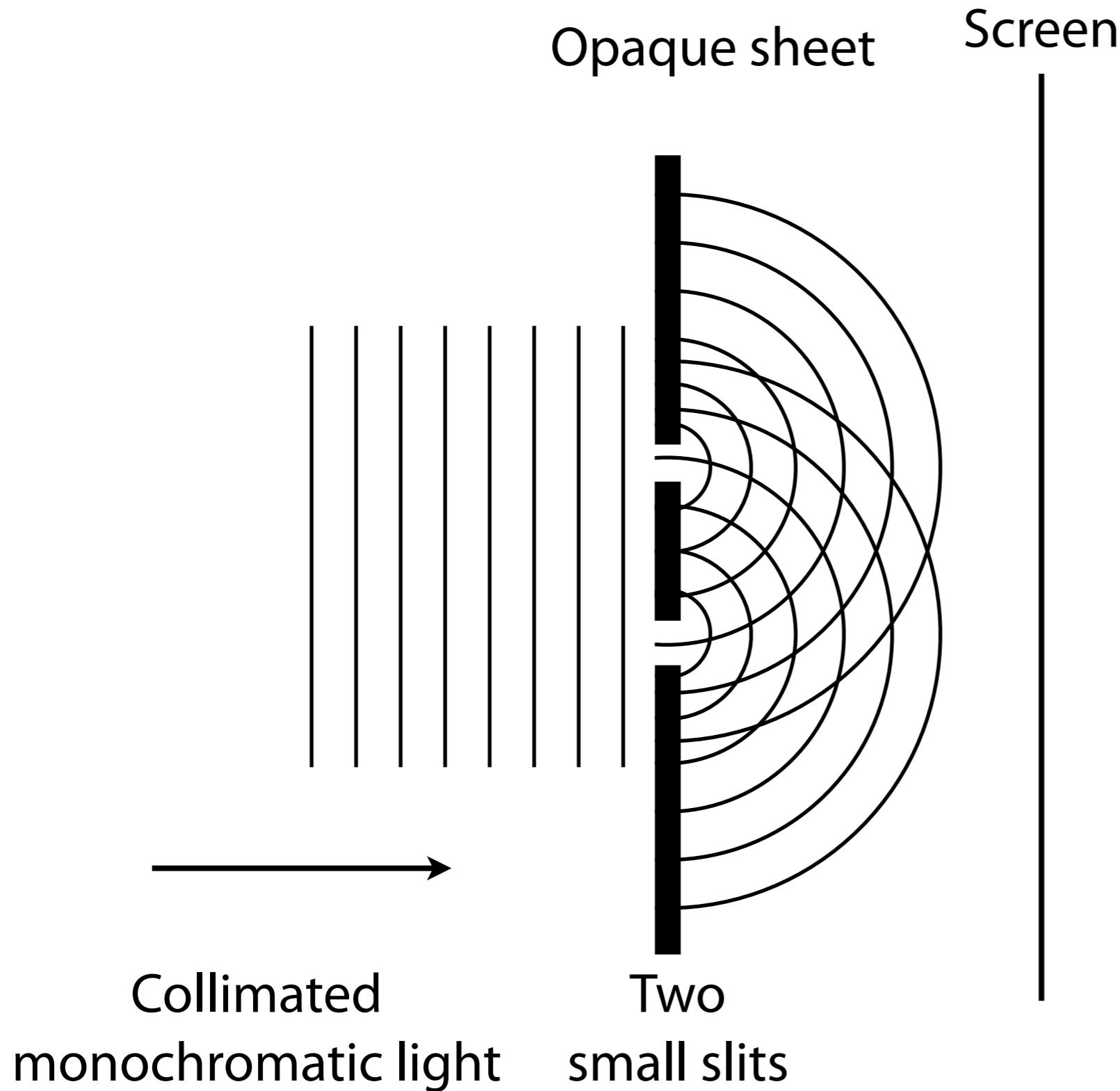


The Young Two-slit experiment

With single photons?



T. Young



What do we observe?

The Young Two-slit experiment

with single photons.



T. Young

This is a video of a two-slit experiment with single photons.



The Young Two-slit experiment

with single photons.



T. Young

This is a video of a two-slit experiment with single photons.





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0,025 msec



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0,025 msec

The Young Two-slit experiment

with single photons.



T. Young

The experiment satisfies the probability rule: **single photon probability proportional to classical light intensity.**



The Young Two-slit experiment

with single photons.



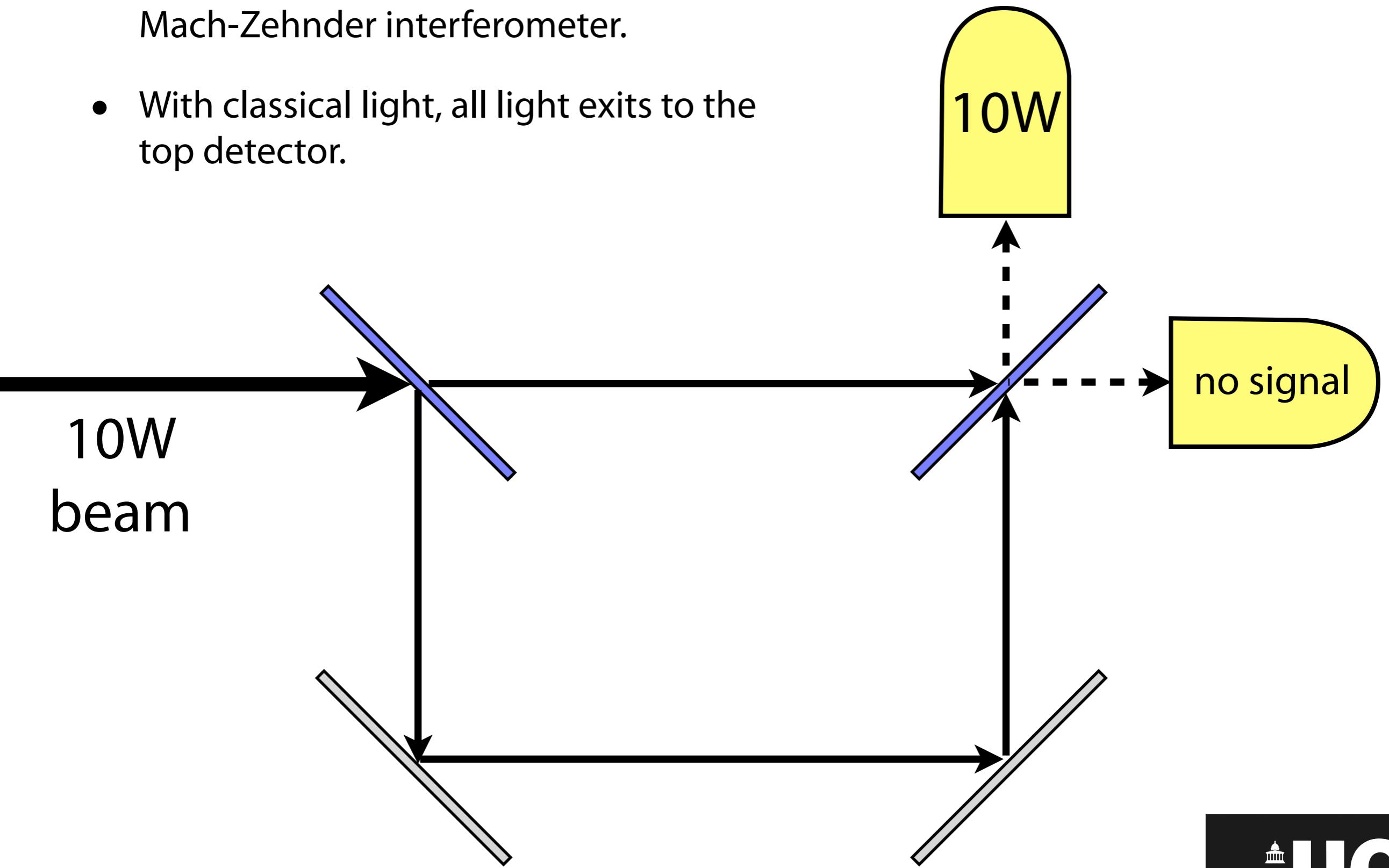
T. Young

The experiment satisfies the probability rule: **single photon probability proportional to classical light intensity.**



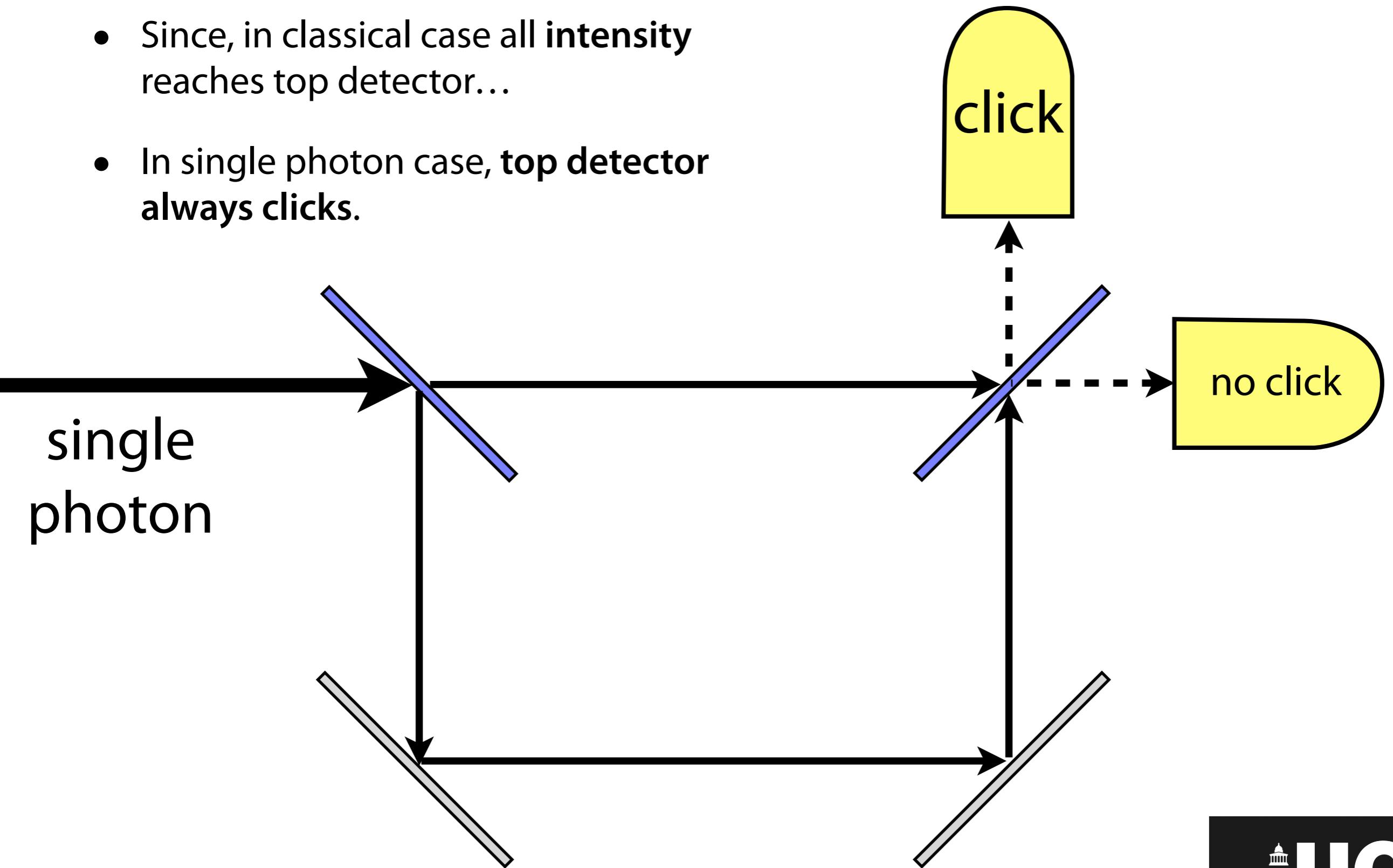
Mach-Zehnder interferometer

- Now let's consider single photons in the Mach-Zehnder interferometer.
- With classical light, all light exits to the top detector.



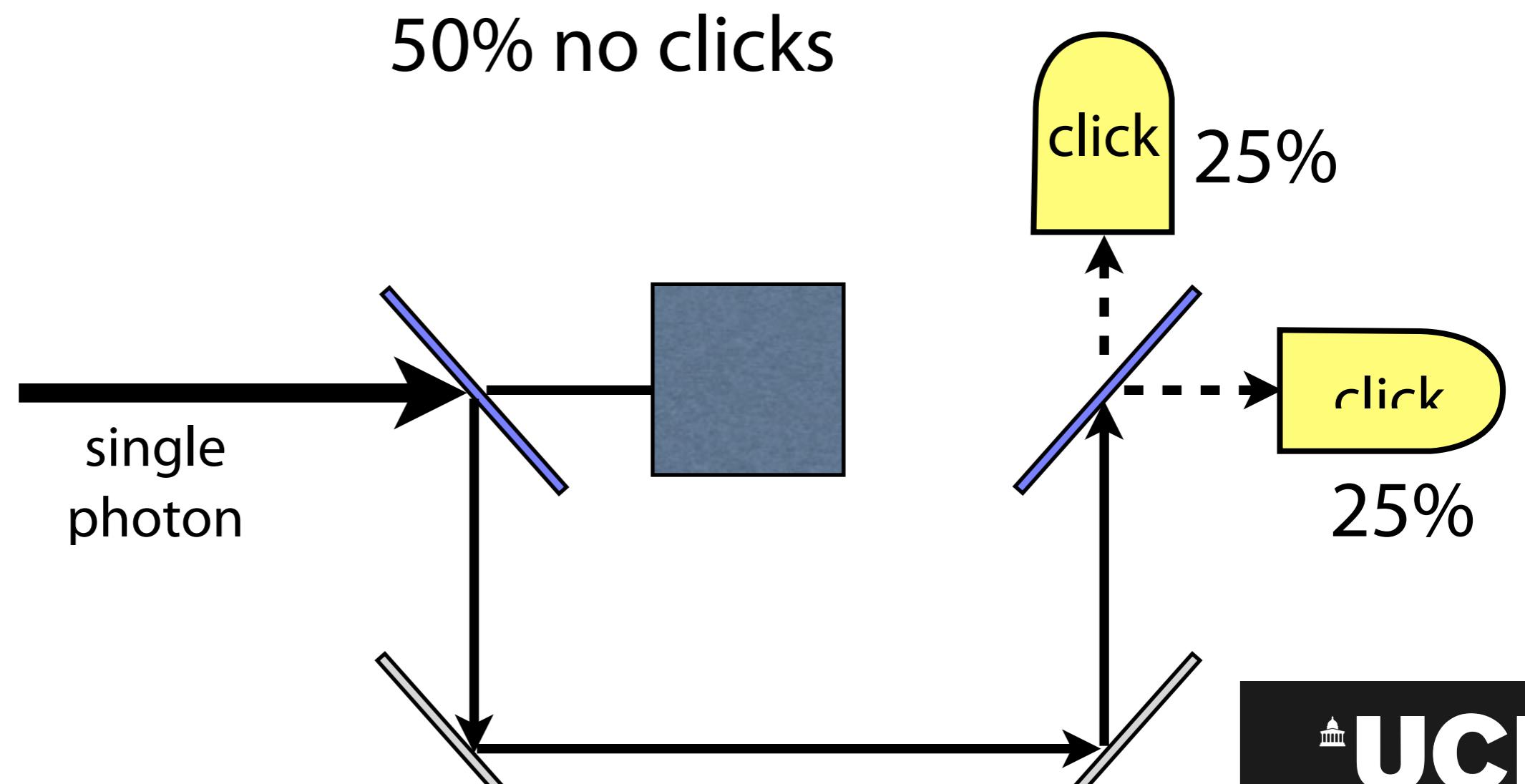
Mach-Zehnder interferometer

- We apply the probability rule:
- Since, in classical case all **intensity** reaches top detector...
- In single photon case, **top detector always clicks**.



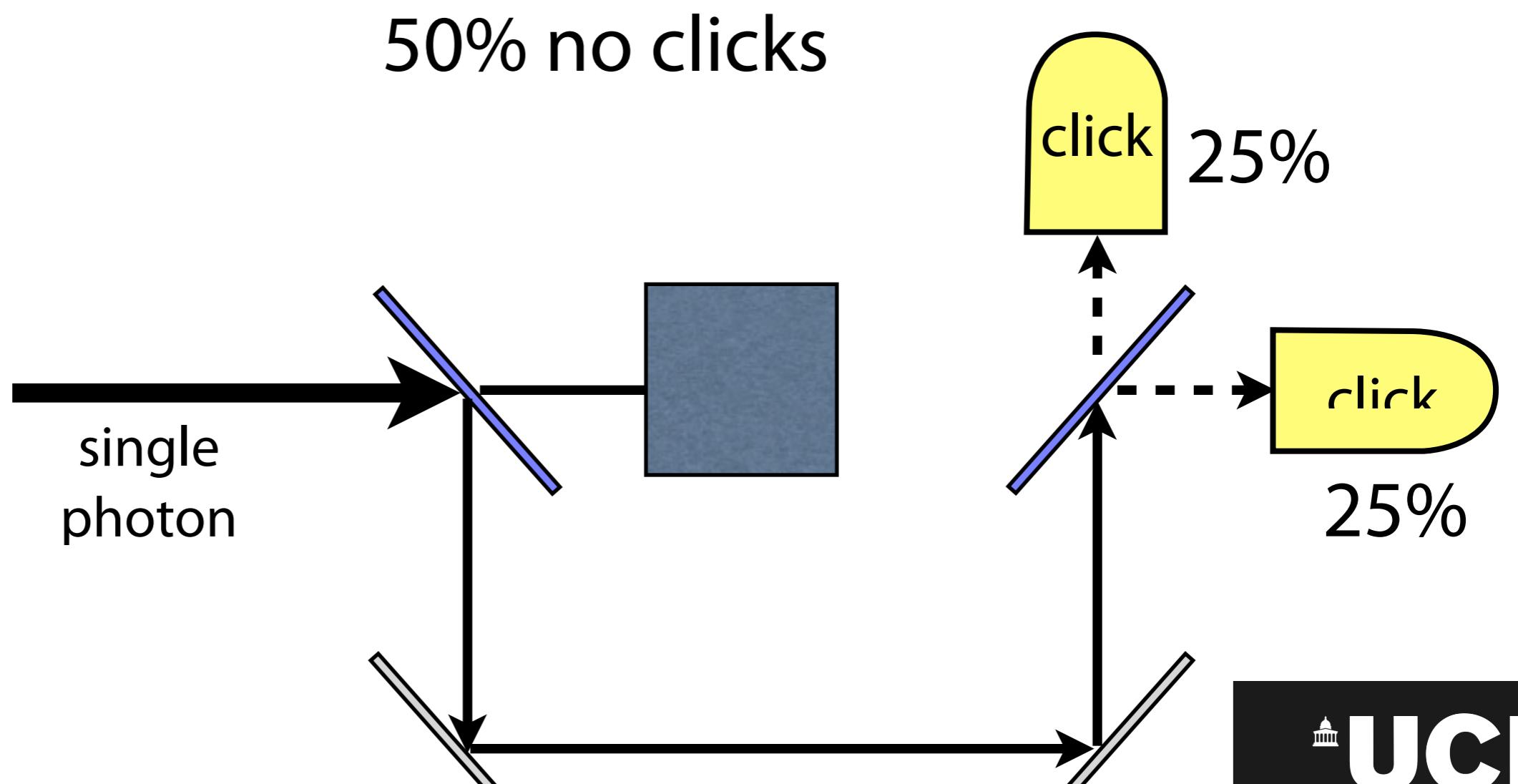
Wave-particle duality of the photon

- What happens when we block one of the paths?
 - The **probabilities** match the intensities of the classical light.
 - **50% chance** - photon is absorbed by the barrier - **no click** at either detector.
 - **25% equal chance** - either one of the **two detectors**.



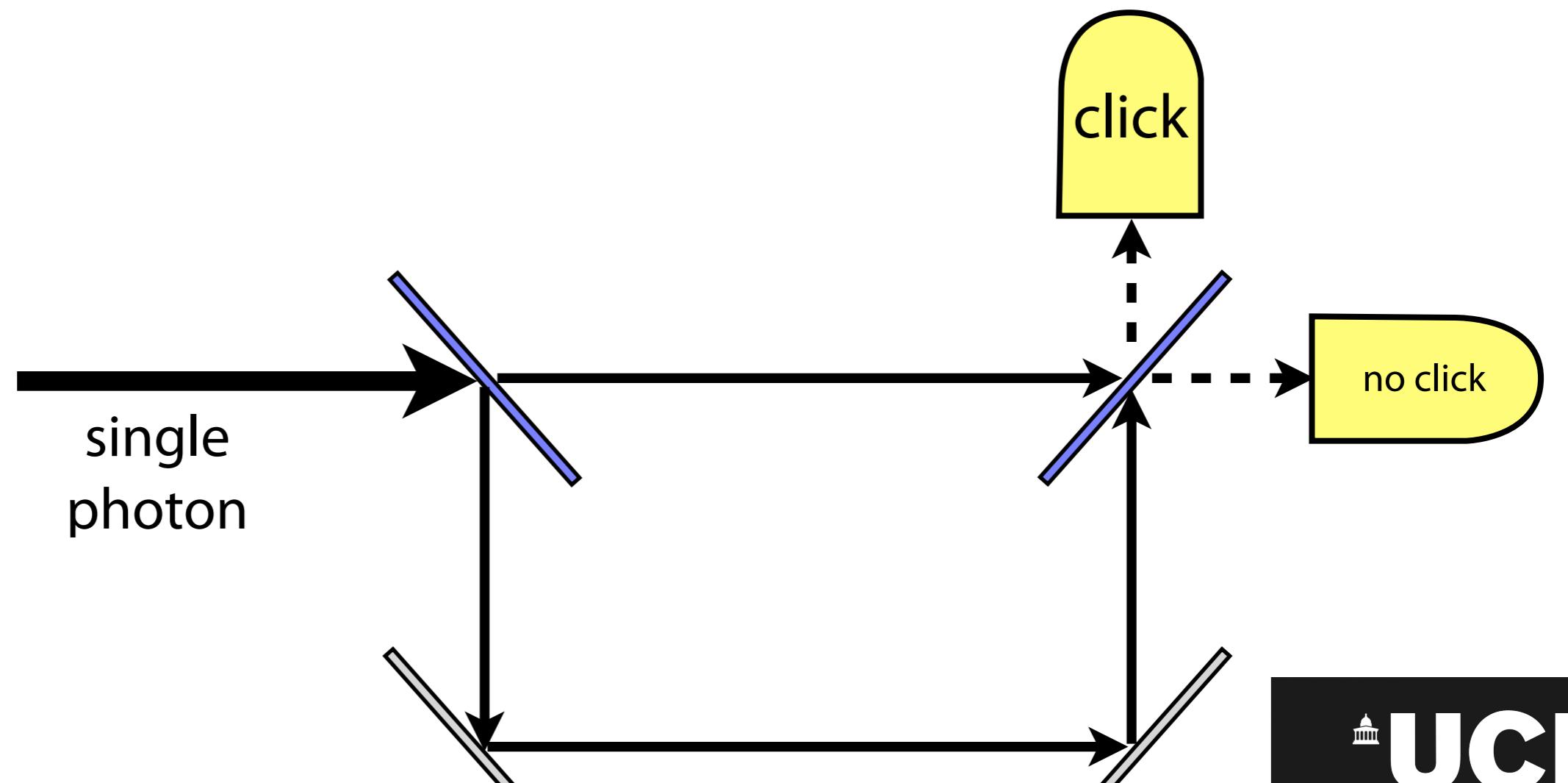
Wave-particle duality of the photon

- If no path is blocked - the photon can **never** reach detector 2.
- But if either path is blocked - the photon **can** reach detector 2.
- So - the photon must “know” about both paths.



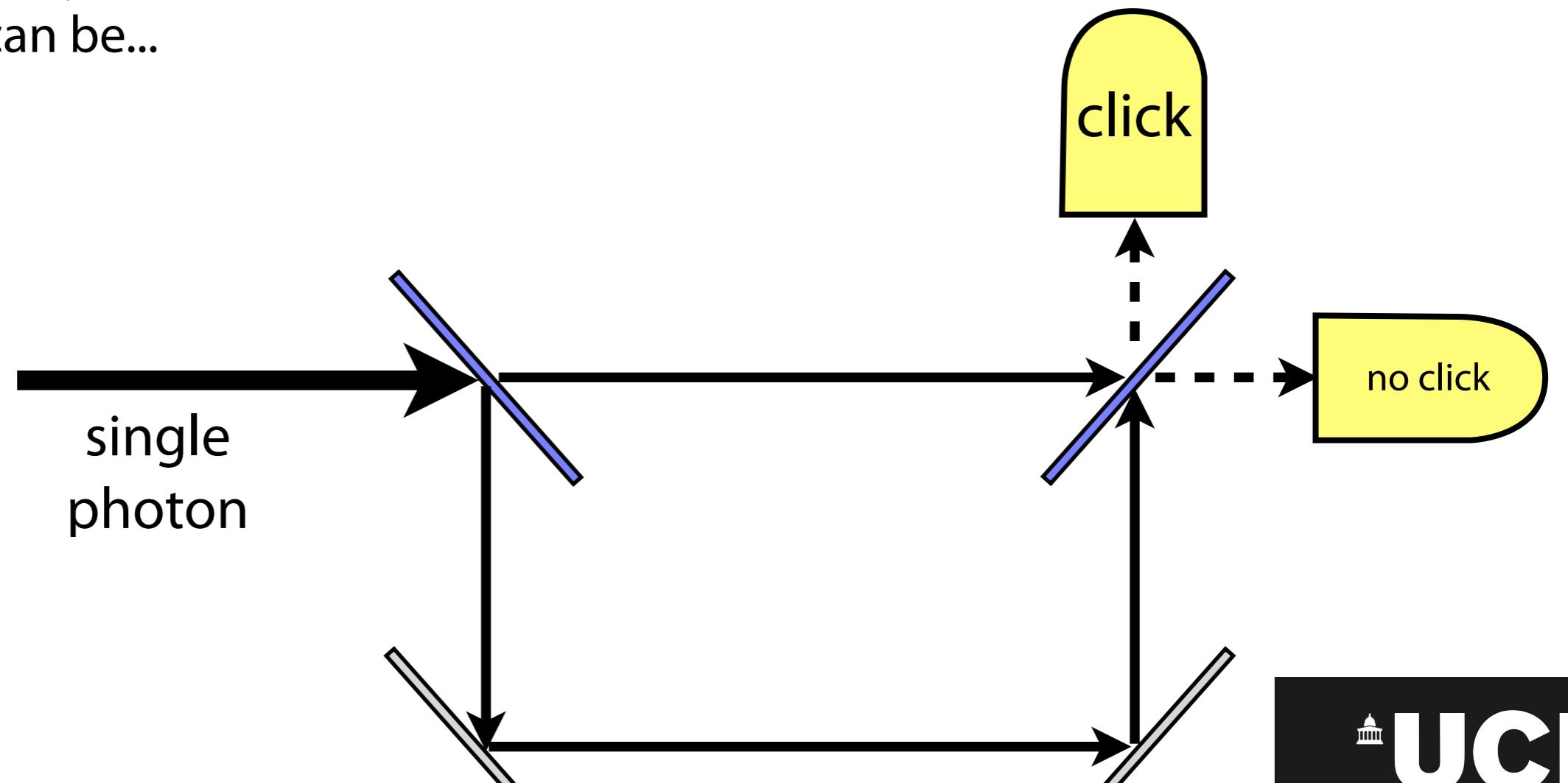
Wave-particle duality of the photon

- This cannot be explained in classical physics. The photon behaves both like a wave - and a particle.
- When **travelling**, a photon behaves like a wave (e.g. taking both arms in the interferometer).
- When **detected**, a photon behaves like a particle - it can only **click** in **one detector** (following our probability rule).



Measurement in quantum physics

- This illustrates a crucial non-classical aspect of quantum mechanics
 - **Measurement has a direct effect on the measured system.**
 - When **not observed**, the photon behaves like a **wave**.
 - When **measured**, the photon behaves like a **particle!**
- This is a hint of just how counterintuitive quantum mechanics can be...



Summary of Part 1

- We saw that the **wave** model was the correct classical model of light, as **only** this model can explain **interference** experiments.
- We saw evidence, (Black-body radiation, Photo-electric effect) that light is **absorbed** and emitted as **packets** of energy.
- Planck's energy of a **photon**: $E = hf$
- We saw that **single photon behaviour** can be **observed**. Photons are detected as “**clicks**” in a **photon detector**.
- **Classical - Quantum** correspondence in **interferometers**:
 - **probability of photon** reaching a detector is **proportional to intensity** in classical experiment.
- We saw that the behaviour of **single photons** in a **Mach-Zehnder** interferometer **cannot** be explained by **wave-like** or **particle-like** behaviour alone.