

$$1 \text{ atm} = 760 \text{ mmHg} = 1.01 \times 10^5 \text{ N m}^{-2}, R = 8.314 \text{ J mol}^{-1} \text{ K}^{-1}$$

- 1 The solution contains A and B, which the molar fractions are 0.4 and 0.6, respectively. At this molar fraction, the molar volume of A and that of B are $13 \text{ cm}^3 \text{ mol}^{-1}$ and $11.0 \text{ cm}^3 \text{ mol}^{-1}$, respectively. The molar volume of pure A and that of pure B are $15.0 \text{ cm}^3 \text{ mol}^{-1}$ and $13.0 \text{ cm}^3 \text{ mol}^{-1}$, respectively. The total mole of solution is 2.0 moles. According to these statements, fill the following table with the number and the unit.

	Number and unit	
V_A^0	15 cm^3/mol	
V_B^0	13 cm^3/mol	
x_A	0.4	$n_A = 0.8 \text{ mole}$
x_B	0.6	$n_B = 1.2 \text{ mole}$
\bar{V}_A	13 cm^3/mol	
\bar{V}_B	11 cm^3/mol	
$\Delta \bar{V}_A^M$	13 - 15 = -2 cm^3/mol	
$\Delta \bar{V}_B^M$	11 - 13 = -2 cm^3/mol	
V	$13 \cdot 0.4 + 11 \cdot 0.6 = 11.8 \text{ cm}^3/\text{mol}$	
ΔV^M	$11.8 - (0.4 \cdot 15 + 0.6 \cdot 13) = -2 \text{ cm}^3/\text{mol}$	
V'	23.6 cm^3	
$\Delta V'^M$	-4 cm^3/mol	

- 2 A solution is consisted of two liquids, A and B, which the total molar volume is followed by:

$$V(\text{cm}^3 \text{ mol}^{-1}) = 100 - 15x_A - 3x_A^2$$

- 2.1 The functions of partial molar volumes for A and B

- 2.2 Compared to the ideal solution, the function of the molar volume change of the mixed solution,

$$\Delta V^M.$$

$$2.1 \quad \bar{M}_A = M + (1-x_A) \frac{\partial M}{\partial x_A}, \quad \bar{M}_B = M - x_A \frac{\partial M}{\partial x_A}, \quad \frac{\partial V}{\partial x_A} = -15 - 6x_A$$

$$\Rightarrow \bar{V}_A = (100 - 15x_A - 3x_A^2) + (1-x_A)(-15 - 6x_A) = (100 - 15x_A - 3x_A^2) + 6x_A^2 + 9x_A - 15 \\ = \underline{3x_A^2 - 6x_A + 85} *$$

$$\Rightarrow \bar{V}_B = (100 - 15x_A - 3x_A^2) - x_A(-15 - 6x_A) = 100 - 15x_A - 3x_A^2 + 15x_A + 6x_A^2 = \underline{3x_A^2 + 100} *$$

$$2.2 \quad V_A^0 = 82, \quad V_B^0 = 100$$

$$\Delta V^M = (x_A \bar{V}_A + x_B \bar{V}_B) - (x_A V_A^0 + x_B V_B^0) = (x_A \bar{V}_A + \bar{V}_B - x_A \bar{V}_B) - (x_A V_A^0 + V_B^0 - x_A V_B^0) \\ = [x_A (\bar{V}_A - \bar{V}_B) + \bar{V}_B] - [x_A (V_A^0 - V_B^0) + V_B^0] \\ = [x_A (-6x_A - 15) + 3x_A^2 + 100] - [-18x_A + 100] \\ = -3x_A^2 - 15x_A + 100 + 18x_A - 100 = \underline{-3x_A^2 + 3x_A} *$$