		(	GAS.			
Sulphuretted hydrogen				47 c.c.	·0731 grammes.	
Carbonic acid	•••	•••		61 ,,	·1200	"
Nitrogen	•••	•••	•••	10 ,,	.0126	>>
				118	2057	

There are two other sulphur springs, less warm and containing much more salt, but they are not being used at present. The sulphur waters resemble those of Aix-les-Bains and Enghien (Paris).

2. There are two Iron springs, the water of which is pleasant to drink and is taken gratis by large numbers of people. The bicarbonate of iron in the water comes from the protoxide of iron in the sand. The sand, which in the spring is black, takes a yellowish-grey colour when dried in the air, the protoxide being turned into sesquioxide of iron. The water is clear, colourless, odourless, and tastes a little salt and bitter. Temperature, 77° Fahr.; specific gravity, 1.0445. It does not leave a deposit of carbonate of iron when exposed to the air.

## ANALYSIS OF ONE LITRE.

Chloride of sodium			37.2671	grammes.
Chloride of magnesium		***	10.6020	59
Bicarbonate of lime		•••	5.9422	"
Sulphate of magnesium			2.3507	22
Chloride of calcium		•••	1.5250	31
Sulphate of lime	•••		1.0820	**