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Title

Linking geological and infrastructural requirements for large-scale underground hydrogen storage in Germany

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Content

Hydrogen storage might be key to the success of the hydrogen economy, and hence the energy transition in Germany. One option for cost-effective storage of large quantities of hydrogen is the geological subsurface. However, previous experience with underground hydrogen storage is restricted to salt caverns, which are limited in size and space. In contrast, pore storage facilities in aquifers -and/or depleted hydrocarbon reservoirs- could play a vital role in meeting base load needs due to their wide availability and large storage capacity, but experiences are limited to past operations with hydrogen-bearing town gas. To overcome this barrier, here we investigate hydrogen storage in porous storage systems in a two-step process: 1) First, we investigate positive and cautionary indicators for safe operations of hydrogen storage in pore storage systems. 2) Second, we estimate hydrogen storage capacities of pore storage systems in (current and decommissioned) underground natural gas storage systems and saline aquifers. Our systematic review highlights that optimal storage conditions in terms of energy content and hydrogen quality are found in sandstone reservoirs in absence of carbonate and iron bearing accessory minerals at a depth of approx. 1,100 m and a temperature of at least 40°C. Porosity and permeability of the reservoir formation should be at least 20% and $5 \times 10^{-13} \text{ m}^2$ (~500 mD), respectively. In addition, the pH of the brine should fall below 6 and the salinity should exceed 100 mg/L. Based on these estimates, the total hydrogen storage capacity in underground natural gas storages is estimated to be up to 8 billion cubic meters or (0.72 Mt at STP) corresponding to 29 TWh of energy equivalent of hydrogen. Saline aquifers may offer additional storage capacities of 81.6–691.8 Mt of hydrogen, which amounts to 3.2 to 27.3 PWh of energy equivalent of hydrogen, the majority of which is located in the North German basin. Pore storage systems could therefore become a crucial element of the future German hydrogen infrastructure, especially in regions with large industrial hydrogen (storage) demand and likely hydrogen imports via pipelines and ships.

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KEYWORDS

keywords

underground hydrogen storage, UHS, porous media, energy transition, UHS criteria, UHS storage potential, saline aquifers

Headline
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1 Introduction

Global environmental changes and future energy needs are among the most fundamental challenges facing humanity in the near future. Green hydrogen produced by renewable energies has the capacity to overcome these obstacles by decarbonizing the industry, electricity generation, transport and heat. With its National Hydrogen Strategy (NWS), the German government has given a strong boost to the hydrogen economy in all sectors of the energy system (BMW, 2020). But renewable energy sources needed for the production of green hydrogen are subjected to weather-related and seasonal fluctuations. Hence, measures are needed to stabilize and secure future energy supplies. Hydrogen storage could therefore be the decisive factor in the energy transition. Only then the divergent hydrogen demand and production will be balanced.

In addition to the storage of hydrogen in tanks or pipes, the geological subsurface also offers possibilities for the long-term storage of hydrogen. The hydrocarbon industry has been using the underground for decades in Germany and around the world to store large quantities of natural gas in the pore space of rocks or in salt caverns (e.g., American Gas Association, 1997; Sedlacek, 1999a; Sedlacek, 2002; Evans and Chadwick, 2009). More than 23 billion cubic meters of natural gas are stored underground in Germany (EEK Redaktion, 2021). This gives Germany the fourth-largest storage capacity in the world after the United States, Russia and the Ukraine (EEK Redaktion, 2015). About 36% (~8.6 billion cubic meters) of the current technically usable working gas volume is stored in pore storage systems for seasonal base load coverage and 74% of the installed capacity are stored in salt caverns for peak load coverage with more efficient injection and withdrawal cycles (EEK Redaktion, 2021). Nevertheless, current research into underground hydrogen storage potential in Germany and Europe has mainly focused on salt caverns only (Iordache et al., 2014; Simón et al., 2015; Michalski et al., 2017; Tarkowski and Czapowski, 2018; Caglayan et al., 2020; Lankof, and Tarkowski, 2020; Williams et al., 2022). The total energy storage potential in salt caverns is estimated to be as high as 84.8 PWh in Europe and 35.7 PWh in Germany alone (Caglayan et al., 2020). It is therefore not surprising that first flagship projects for underground storage of hydrogen in salt caverns at the field scale have already been launched in Germany at the Bad Lauchstädt Energy Park (HYPOS), in Rüdersdorf (HyCavMobil) and in Epe (GetH2). However, suitable salt deposits are locally limited and only 9% of the total worldwide working gas capacity is attributed to salt caverns (Sedlacek, 1999b; Cedigaz, 2021). Pore storage facilities such as depleted gas fields and saline aquifers could therefore play an important role in the energy transition due to their wide spread and broad availability across Europe (e.g., Foh et al., 1979; Panfilov, 2016; Tarkowski, 2019; Heinemann et al., 2021a; Zivar et al., 2021; Hematpur et al., 2023).

Here, we investigate safety requirements and storage potentials in the light of future hydrogen needs. The aim is to formulate recommendations for future underground hydrogen storage in porous media. Our study starts with an overview of the past, present, and future of underground hydrogen storage (UHS) in Germany, discussing the experience with UHS and the prospects for hydrogen network structures. It highlights the progress that has been made and the challenges that need to be overcome for widespread

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deployment. In particular, there are critical factors that need to be considered in the design and operation of UHS facilities to ensure their safe and efficient operation. These critical factors or indicators, such as biotic and abiotic reactions, sealing capacity, cap rock integrity, cushion gas requirements, etc., are discussed before estimates of UHS storage capacity in Germany are made. Our estimates are made with respect to the discussion of technical hydrogen storage capacity in former and current underground gas storage facilities and the theoretical hydrogen storage capacity in saline aquifers. This section provides a quantitative assessment of the potential of UHS to support the energy transition in Germany. The fifth and final section discusses storage capacity and demand in Europe and assesses the costs and purity requirements of UHS. It builds on the findings of the previous sections and highlights the economic and technical challenges associated with UHS and the potential role of UHS in supporting the development of a sustainable hydrogen economy in Europe.

heading
2 Past, present, and future of underground hydrogen storage in Germany

Headline

2.1 Experiences with underground hydrogen storage

Content

Before the introduction of natural gas, town gas with a hydrogen content of more than 50% hydrogen was produced, transported and stored in Germany and Europe. Town gas was derived from coal, and its production still persists in China and Asia. When storage tanks could no longer meet the requirements of the gas industry, underground gas storage facilities were built in the porous geological subsurface starting from the 1950s. Pore storage systems were launched in Beynes (France), Lobodice (Czech Republic), Engelbostel (Hannover), Reitbrook (Hamburg), Hähnlein (Darmstadt), Kirchheiligen (Mühlhausen/Tübingen), Eschenfelden (Fürth/Oberpfalz), Ketzin (Brandenburg) even before the first salt caverns were put into operations in Bad Lauchstädt and Kiel in the 1970s (e.g., Jones and Machsen, 1963; Šmigaj et al., 1990; Sedlacek, 2002; Panfilov, 2016; Marcogaz, 2017). Today, these underground storage facilities have been converted to natural gas storage facilities or have since been abandoned. Currently, pure hydrogen (>95% H₂) is solely stored in salt caverns at a few sites, namely, Teesside (United Kingdom) and Clemens Dome (United States), Spindletop (United States), and Moss Bloss (United States) (e.g., Panfilov, 2016; Zivar et al., 2021). Given this sparse and limited experience with hydrogen in pore storage systems, it is necessary to formulate recommendations for the safe handling of hydrogen in the subsurface.

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2.2 Prospects for hydrogen network structures

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The design of future hydrogen infrastructure and storage requirements in Germany remains uncertain and complex (Lux et al., 2022; Neuwirth et al., 2022). It is challenging to analyze

underground storage requirements and infrastructures in such a highly complex and swiftly evolving environment. While the German National Hydrogen Strategy was published in 2020 to promote the use of hydrogen in all sectors of the hydrogen economy, it didn't specify the design of the hydrogen transport and storage infrastructure (BMW, 2020). To examine the potential use of underground hydrogen storage in porous media, three inter-dependent variables that could impact future storage requirements in Germany have to be considered: 1) spatial distribution of hydrogen production via electrolysis, 2) hydrogen demand centres, and 3) hydrogen pipelines and import terminals.

Lux et al. (2022) investigated the positioning of electrolyzers, underground storage and transport infrastructure in a greenhouse gas-neutral German energy system using five scenarios. Regardless of the scenario, their calculations indicate that 71% of the total hydrogen production in Germany will be located in the North and Baltic Seas, with a combined hydrogen production of 83–129 TWh by 2050 and total electrolyzer capacities of 37–55 GWel (ibid.). The concentration in these regions is due to the high potential for wind power, and it does not conflict with the alternative of expanding the electricity grid. Sens et al. (2022) report similar findings, as they calculate the second-lowest hydrogen supply cost at the North Sea, while Husarek et al. (2021) also support the idea of regional concentration, which contrasts with the results of Gils et al. (2021). Lux et al. (2022) demonstrate that hydrogen can serve as a seasonal and long-term storage medium, with a working gas volume of 42–104 TWh of hydrogen in 2050. In contrast, Gils et al. (2021) report a required storage capacity of 53 TWh in the same year. In the modelled energy systems, hydrogen storage levels decrease during winter when renewable electricity generation is lower and increase in summer with higher renewable electricity feed-in (Lux et al., 2022). This seasonal profile for hydrogen storage has been identified in previous studies by Gils et al. (2021) and Welder et al. (2018) for Germany, and Cárdenas et al. (2021), Samsatli and Samsatli (2019), Gabrielli et al. (2020), and Hernandez and Gençer (2021) for different regions or countries. Large-scale storage directly on site or at the end of a transport pipeline at the demand location can reduce hydrogen costs by decreasing excess electricity (Welder et al., 2018; Gils et al., 2021; Sens et al., 2022).

In order to establish a cost-effective hydrogen economy and a greenhouse-gas neutral energy system in the long run, a German-wide transport network is necessary (Gils et al., 2021). However, the design of such a network and the routing of its pipelines is still up for debate. Although detailed expansion plans differ, there is consensus in the literature on the need to connect production hubs and import terminals in the North with demand centers in the South of Germany (ibid.; Lux et al., 2022; Husarek et al., 2021). The European gas network operators have proposed a European-wide hydrogen pipeline vision (hydrogen backbone), which focuses on the repurposing of existing natural gas infrastructures (Wang et al., 2020; Wang et al., 2021). For example, in Germany, a North-South pipeline from Rostock to Lake Constance is planned to be established from 2035 onwards, repurposing the existing natural gas infrastructure, which is considered more cost effective than building new pipelines. The repurposing of pipelines and port

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infrastructure is also a part of the IPCEI Hydrogen initiative (important projects of common European interests). The association of the supra-regional gas transmission companies in Germany (FNB Gas) develops a joint network development plan every 2 years, which is then submitted to the German Federal Network Agency. The 2022 plan (FNB Gas, 2022) models a national hydrogen network in Germany, based on a survey of transportation needs. However, due to the current regulatory framework, this network is not legally binding. The 2020 plan (FNB Gas, 2020) only contained a rough outline of a visionary hydrogen network, in contrast to the more detailed model presented in the 2022 plan.

Assessing hydrogen storage requirements is a multi-faceted endeavour that includes an evaluation of potential future hydrogen demand and its regional distribution. Future hydrogen demand is still uncertain, according to Neuwirth et al. (2022), while Lux et al. (2022) indicate a range of hydrogen demand scenarios between 34 and 667 TWh (final energy demand in 2050), attributable to different usage categories such as industry, transport, and heating. To derive a more comprehensive estimate, Neuwirth et al. (2022) conducted a site- and process-specific bottom-up analysis of hydrogen demand from the energy-intensive sectors alone, revealing a potential hydrogen demand of 326 TWh/a. The spatial distribution of this demand is heterogeneous, with concentration in a few regions. Consequently, we posit that underground hydrogen storage is more likely in these regions, given that they are probable endpoints for pipelines, and industry clusters require an economic and steady hydrogen supply, which can be secured through storage.

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3 Indicators for safe operations of hydrogen in pore storage systems

Content

There are recommendations and evaluation schemes for the geotechnical and energetic use of the subsurface on a German and European level based on the operation of hydrocarbon reservoirs, underground storage facilities for both natural gas and CO₂, the search for radioactive repositories and the utilization of geothermal resources (e.g., Sedlacek, 1999a; Sedlacek, 1999b; Chadwick et al., 2008; Reinhold et al., 2011; Alfarra et al., 2020; Stober and Bucher, 2021). Here we report positive and cautionary indicators for safe operations with hydrogen in underground pore storage systems based on a literature survey. The results are summarized in Table 1.

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3.1 Biotic and abiotic reactions

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Depleted fields and aquifers are composed of either sandstones or limestones. Sandstones, consist primarily of stable, non-reactive silicate minerals, namely, quartz and feldspar that will not react with hydrogen at reservoir conditions (Flesch, et al., 2018; Yekta et al., 2018). Major components in limestone reservoirs and mineral accessories in sandstones or in the caprock, however, may form reactions with hydrogen forming toxic gases such as hydrogen sulphide. Possible reactants include carbonate [CaCO₃, MgCO₃,

imageDescription

TABLE 1 Listing of requirements and positive and negative CCS and H₂ indicators for underground gas storage in aquifers and depleted gas fields.

Requirement	figure Natural gas in aquifers ^a	figure Natural gas in depleted fields ^a	figure CCS positive indicator	figure CCS negative indicator	figure H ₂ positive indicator	figure H ₂ cautionary indicator	figure Comment
small Depth	small 800–1,000 m	small 1,000–1,200 m	small 1,000–2,500 m	small <800 m	small 1,100 m	small <800 m	figure H ₂ mass peak at 1,100 m
small Thickness	small 15–40 m	small 20–50 m	small >50 m	small >2,500 m ^b	small >3,700 m ^c	small >3,700 m ^c	small Efficiency
small Lithology	small Sandstone	small Sandstone, limestone			small Sandstone	small Aquifer minerals, e.g. pyrite, carbonate	figure Many trigger redox reactions
small Porosity	small 18–25%	small 18–25%	small >20%	small <10%	small >20%	small <20%	figure Efficiency/Capacity
figure Permeability	small >600 mD	small >100 mD	small >500 mD	small <200 mD	small >500 mD	small <100 mD	figure Efficiency
small Salinity			figure >100 g/L ^d	figure <30 g/L	small >100 g/L ^e	small <100 g/L	small Biotic growth
small Temperature					small >40°C	small <40°C	small Biotic growth peak: 0–40°C
small pH					small <6	small >8	small Biotic growth peak: pH 6–7.5
small References	small Sedlacek (1999b)	figure Sedlacek (1999b)	small Chadwick et al. (2008)	small Chadwick et al. (2008)	small Chadwick et al. (2008)	small Chadwick et al. (2008)	small Small reduction favoured at pH > 8
					small Panfilov (2016)	small Panfilov (2016)	
					small Heinemann et al. (2021c)	small Heinemann et al. (2021c)	
					small Truche et al. (2013)	small Truche et al. (2013)	
					figure Iglauder (2022)		

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^aNatural gas: operational experience.
^bCO₂: upper limit set by porosity at depth.
^cH₂: upper limit set by water-rock-contact angle.
^dCO₂: to avoid spoiling potable water resources.
^eH₂: to inhibit biotic reactions and to avoid spoiling potable water resources.

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(Ca, Mg) CO₃], sulfate (CuSO₄, CuSO₄H₂O), sulphide (S, FeS, FeS₂, PbS, HgS, ZnS, Cu₂S, CuFeS₂, CS₂), and oxide (Fe₂O₃, Fe₃O₄, FeO, MgO) minerals and reservoir gases such as CO, CO₂, and hydrocarbons (Foh et al., 1979). However most of these reactions are insignificant at reservoir conditions, because hydrogen redox reactivity is kinetically limited due to its high bonding energy and thus most of the possible redox reactions remain insignificant at low temperature (Truche et al., 2013; Panfilov, 2016). However, experimental data by Truche et al. (2013) demonstrates that hydrogen induced reduction of pyrite into pyrrhotite is likely to produce sulphide at reservoir conditions and that the pH of the media is a critical parameter controlling the extent of the reaction at low temperature. This is consistent with field observations in the town gas storage field in Beynes, France, where pyrite reduction is believed to be responsible for the observed rise in hydrogen sulphide within the reservoir (Bourgeois et al., 1979). Although this represents a potential risk to underground hydrogen storage, the study by Truche et al. (2013) also shows that pyrite reduction could be prevented at acidic reservoir conditions.

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In contrast to abiotic reactions, the extent and impact of which is discussed in the literature (e.g., Foh et al., 1979; Truche et al., 2010; 2013; Yekta et al., 2018), it is widely agreed that biotic processes may endanger hydrogen quality and/or lead to hydrogen consumption (Šmigáň et al., 1990; Buzek, et al., 1994; Reitenbach et al., 1995; Panfilov, 2016; Heinemann et al., 2021c; Zivar et al., 2021). Several classes of microorganisms present at reservoir conditions are known for consuming hydrogen, which are methanogens, hydrogenotrophic sulfate reducers, homoacetogens and hydrogenotrophic iron (III)-reducing bacteria (Reitenbach et al., 1995; Panfilov, 2016; Heinemann et al., 2021c). Possible implications might include the production of methane (CH₄), acetic acid (CH₃COOH), and hydrogen sulphide (H₂S) in addition to hydrogen loss (Reitenbach et al., 1995; Panfilov, 2016; Heinemann et al., 2021c). A well-known example for such biotic hydrogen loss is the Czech underground storage facility Lobodice in which methanogens were shown to transform hydrogen and carbon dioxide to methane, resulting in significant hydrogen reduction from 54 vol% (input) to 37 vol% (output) over a time period of 7 months (Šmigáň et al., 1990). Optimum conditions

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where growth peaks and critical conditions beyond which no growth is possible were summarized for several classes of microorganisms by Heinemann et al. (2021c) showing that most bacteria favour moderate temperatures (20–40°C), salinity (<100 g/L) and pH (6–7.5) [Heinemann et al. (2021c) and references therein].

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3.2 Petrophysical properties of prospective pore storage systems

Content

The petrophysical properties of prospective pore storage systems are important factors to consider when selecting a site for underground gas storage (e.g., Hematpur et al., 2023). It is noteworthy that the optimal porosity and intrinsic permeability values for rock formations utilized in gas storage may vary depending on the specific storage application and the type of gas being stored. For instance, high-pressure storage applications may necessitate higher permeability than low-pressure storage. Other factors such as effective stress state, temperature, and deformation state must also be considered when evaluating the suitability of a rock formation for underground storage.

Technically in UHS operations, a rock with high porosity and permeability is more desirable for hydrogen storage because it allows for higher storage volumes as well as higher injection and withdrawal rates representing the unloading and loading cycles, respectively. Higher energy efficiencies tend to be expected for high intrinsic permeability rock types when hydrogen is injected into them (Pan et al., 2021a). Furthermore, the hydraulic properties of the reservoir rock are a key factor influencing the interaction between fluid pressure and mechanical properties of the reservoir, known as hydro-mechanical coupling (e.g., Rutqvist and Stephansson, 2003; Bai and Tahmasebi, 2002). The permeability controls the fluid flow and thus the temporal and spatial distribution of fluid pressure, which in turn affects the effective stress state of the reservoir and its mechanical properties. This interaction must be considered and studied in the design and operation of UHS systems, as it can change the security and efficiency of the storage process to a significant degree. Zones of high permeability can reduce the risk of high fluid-pressure transients in the reservoir and around the wells, and thus the risk of inelastic rock deformation. Local high-pressure zones can form within low-permeability rocks, leading to aftershocks and induced seismicity in worst-case scenarios (e.g., Pawar et al., 2015; White and Foxall, 2016). In addition, low-pressure zones in the reservoir and around the wells can lead to a decrease in the production rate.

Although hydrogen storage has been extensively researched, to the authors' knowledge, there is currently no recommendation as to which rock type, porosity, and permeability is best suited for UHS. Therefore, requirements for the effective porosity (i.e., connected porosity), intrinsic permeability and reservoir thickness for UHS operations are derived from the experience of the hydrocarbon and geothermal industry (Sedlacek, 1999b; May et al., 2004; Chadwick et al., 2008; Heidug, 2013; Stober and Bucher, 2021) as well as conceptual studies on potential international hydrogen storage in depleted reservoir settings: Chadwick et al. (2008) summarized that porosities and permeabilities of at least 0.20 and 500 mD (approx. $5 \times 10^{-15} \text{ m}^2$) are considered ideal geologic storage conditions. Porosities of less than 0.1 and permeabilities of less than 200 mD (approx. $2 \times 10^{-13} \text{ m}^2$) are considered cautionary indicators by these authors. Sedlacek (1999b) propounds that -from a geological and

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reservoir perspective-porosities and permeabilities within the range of 18 to 15% and greater than 100 mD are suitable for gas reservoirs, respectively. All these indicative limits are consistent with the study by Pfeiffer and Bauer (2015) for subsurface porous media hydrogen storage, which used on-site and off-site porosities and permeabilities of the Middle Rhaetian (main sandstone) of Northern Germany of 0.33 and 572 mD, respectively. Similar porosities (0.1–0.3) and permeabilities in the range of 20–2,500 mD are also used by other studies on UHS (Amid et al., 2016; Heinemann et al., 2018; Hemme and Berk, 2018; Zivar et al., 2021; Lysy et al., 2021). On the basis of these literature values, we suggest permeabilities and porosities greater than 500 mD and 0.2, respectively, for optimal and safe operation of UHS systems (Table 1). However, the range of porosity and permeability depends on the specific properties of the rock formation, so a thorough petrophysical characterization of the rock is required in advance to determine its suitability for hydrogen storage. In addition to the consideration of intrinsic permeability, the characterization of relative permeability in wet porous rocks is crucial for optimizing storage capacity and recovery efficiency. The determination of the relative permeability of hydrogen-brine rock systems is very complex as it depends on the number of hydrogen injection and reproduction cycles as well as the capillary forces responsible for residual hydrogen trapping. Accordingly, these parameters should be determined based on experimental measurements to determine the intrinsic permeability.

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3.3 Storage depth, sealing capacity, and integrity of caprock

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The seal capacity of a reservoir refers to the ability of the rock layers above the reservoir (seals) to prevent the upward migration of fluids or gases out of the reservoir. The seal capacity is therefore critical for the effective trapping and storage of hydrogen in a reservoir, allowing for the efficient extraction and utilization of the resource. The dominant trapping mechanism is the capillary properties of the cap rock that will trap a hydrogen column until the net buoyancy (i.e., difference between hydrogen and brine) exceeds the capillary displacement pressure of the seal (Watts, 1987). The properties that define the capillary entry pressure are the size of the largest interconnected pore throat of the cap rock, the hydrogen-brine interfacial tension (IFT) and wetting behavior of the cap rock with respect to hydrogen and brine, expressed by the brine-rock-hydrogen contact angle θ (see detailed review by Aslannezhad et al., 2023). The height of the trapped hydrogen column can thus be expressed by balancing the buoyancy and capillary entry pressure:

$$h = \frac{2 \times \text{IFT} \times \cos \theta}{r \times g \times \Delta \rho} \quad (1)$$

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The hydrogen-water and hydrogen-brine interfacial tension was measured experimentally by Chow et al. (2018) and Hosseini et al. (2022) and was shown to decline with depth, but increase with salinity. However, under storage conditions (1,000 m and 50°C), the IFT remains relatively high ($\sim 70 \text{ mN/m}^2$) and hydrogen and brine remain immiscible (Chow et al., 2018; Hosseini et al., 2022; Iglaue, 2022). Hydrogen wettability of selected minerals were also measured and calculated in a few studies. Hydrogen wettability of quartz (pure and aged with stearic acid) were reported to increase with organic acid surface concentration, pressure and temperature (Iglaue et al.,

TABLE 2 Comparison of geological conditions and volumetric characteristics of existing active and inactive underground storage fields in Germany (see also Figure 3).

Active fields	German state	Storage type	Depth (m)	Period	Series	Lithology	Withdrawal (mio. m ³ /d)	Volume (mio. m ³)	WV CH ₄ (mio. m ³)	HHV CH ₄ (GWh)	WV H ₂ (mio. m ³)	HHV H ₂ (GWh)	Town gas	Data publ.
Allmenhausen	TH	Depleted gas field	350	Triassic	Buntsandstein	Sandstone	62	18	62	685	5	11	No	December 2020
Bad Lauchstädt	ST	Depleted gas field	800	Permian	Rotliegend	Sandstone	238	670	440	4,864	349	1,231	Yes	December 2020
Bierwang	BY	Depleted gas field	1,560	Palaeogene	Oligocene (Chatt)	Sandstone	1,200	3,600	1,000	11,057	80	2,840	No	December 2020
Breitbrunn-Eggstätt	BY	Depleted gas field	1,900	Palaeogene	Oligocene (Chatt)	Sandstone	520	2,075	992	10,967	827	2,917	No	December 2020
Eschenfelden	BY	Aquifer	600	Triassic	Keuper, Muschelkalk	Clastic	95	168	72	792	50	18	Yes	December 2020
Frankenthal	RP	Aquifer	600, 1,000	Neogene	—	Sandstone	130	300	90	995	23	254	No	December 2020
Fronhofen-Illmensee	BW	Depleted gas field	1,750–2,000	Triassic	Upper Muschelkalk	Dolomite	30	15	10	111	8	3	No	December 2020
Hähnlein	HE	Aquifer	500	Neogene	Pliocene	Sandstone	100	200	30	844	6	6	Yes	December 2020
Inzenham	BY	Depleted gas field	680–880	Neogene	Miocene (Aquitainian)	Arenaceous marl	300	880	425	4,699	339	1,196	No	December 2020
Rheden	NI	Depleted gas field	1,900–2,250	Permian	Zechstein	Dolomite	2,400	6,780	900	43,117	5,341	11,620	No	December 2020
Sandhausen	BW	Aquifer	600	Tertiary	—	Clastic (sand)	45	68	30	332	25	88	No	December 2020
Schmidhausen	BY	Depleted gas field	1,015	Neogene	Miocene (Aquitainian)	Clastic (sand)	150	3	54	1,703	12	7	No	December 2020
Stockstadt	HE	Depleted gas field	500	Neogene	Pliocene	Clastic	45	94	45	498	38	134	No	December 2020
Stockstadt	HE	Aquifer	450	Neogene	Pliocene	Sandstone	90	200	90	995	70	68	No	December 2020
Uelsen	NI	Depleted gas field	1,470–1,525	Triassic	Buntsandstein	Sandstone	430	1,579	860	9,508	689	2,431	No	December 2020
Wolfersberg	BY	Depleted gas field	2,930–3,500	Oligocene	Rupel (Lithothamnien)	Limestone	240	583	365	4,035	330	1,164	No	December 2020
							Total volume (mio. m ³)		615	95,234	7,234	2,234		
Inactive fields	German state	Storage type	Depth (m)	Period	Series	Lithology	Withdrawal (mio. m ³ /d)	Volume (mio. m ³)	WV CH ₄ (mio. m ³)	HHV CH ₄ (GWh)	WV H ₂ (mio. m ³)	HHV H ₂ (GWh)	Town gas	Data publ.
Kirchheiligen	TH	Depleted gas field	900	Permian	Zechstein	Dolomite	25	25	90	2,100	10	6	Yes	December 2020
Berlin	BE	Aquifer	750–1,000	Triassic	Buntsandstein	Sandstone	225	573	143	1,581	113	399	No	December 2016

Continued on following page

TABLE 2 (Continued) Comparison of geological conditions and volumetric characteristics of existing active and inactive underground storage fields in Germany (see also Figure 3).

Inactive fields	German state	Storage type	Depth (m)	Period	Series	Lithology	Withdrawal (mio. m ³ /d)	Volume (mio. m ³)	WV CH ₄ (mio. m ³)	HHV CH ₄ (GW/h)	WV H ₂ (mio. m ³)	HHV H ₂ (GW/h)	Town gas	Data publ.
Buchholz	BB	Aquifer	570–610	Triassic	Buntsandstein	Sandstone	80	199	140	1,548	115	406	No	December 2015
Kalle	NI	Aquifer	2,100	Triassic	Buntsandstein	Sandstone	450	630	2	387	82	642	Yes	December 2015
Reitbrook	HH	Depleted oil field	640–725	Cretaceous-Paleogene	Upper Cretaceous-Oligocene	Limestone, dolomite, clastic sands	350	493	350	3,869	284	1,002	Yes	December 2013
Lehrte	NI	Depleted oil field	1,000–1,150	Jurassic	Dogger (Cornbrash)	Clastic	20	120	35	387	12	95	No	December 2012
Dötlingen	NI	Aquifer	570–610	Triassic	Buntsandstein	Sandstone	80	234	175	1,935	143	504	No	December 2010
Ketzin	BB	Aquifer	230	Jurassic	Lower Jurassic	Sandstone	79	271	135	493	23	434	Yes	December 1999
Total volume (mio. m ³)									1,383	15,290	1,136	4,008		

Besides the withdrawal, working volume (WV) and higher heating value (HHV) of methane (CH₄) and hydrogen (H₂) are given for the respective fields. The data have been taken from Sedlacek (1999b), EEK Redaktion, (2011), EEK Redaktion, (2013), EEK Redaktion, (2014), EEK Redaktion, (2016), EEK Redaktion, (2017), EEK Redaktion, (2021). The literature used always contains the data status of the previous year [e.g., EEK Redaktion, (2011) shows data as of December 2019].

Content

2021). While pure quartz remained always strongly water-wet (brine contact angles $<50^\circ$) for all conditions tested, aged quartz samples were weakly water-wet (brine contact angles $50\text{--}70^\circ$) for most scenarios examined (Iglauer et al., 2021). Al-Yaseri et al. (2021) and Al-Yaseri et al. (2022) reported contact angles for the clay brine and shale brine system based on empirical correlations as a function of pressure and temperature. All clays showed water wetting ($\theta < 40^\circ$) under all experimental conditions and were shown to be conducive to capillary and/or residual trapping of the gas (Al-Yaseri and Jha, 2021). The shale's H_2 sealing capacity was shown to decrease with increasing depth and TOC values, but remained at strongly water-wet conditions with contact angles of not more than 17° at highest pressure and TOC (Al-Yaseri et al., 2022). The above measured and estimated data show that rocks are highly water-wet (contact angle < 50) in the presence of H_2 at most storage conditions. However, based on the hydrogen wetting behavior of aged quartz and aged mica, Iglauer (2022) and Ali et al. (2022) demonstrated that θ may increase to 90° or more at greater depth (3,700 m for quartz; Iglauer et al., 2021), which sets a limit to hydrogen storage capacity below which hydrogen will percolate through the cap rock. In practice however, limitations on the depth are questions of economics and depths greater than 1,200 m are uncommon in underground gas storage facilities (Foh et al., 1979; Sedlacek, 1999b). The optimum storage depth for hydrogen can be computed based on the sealing capacity of the cap rock and the density of hydrogen gas that was estimated to be at 1,100 m depth assuming average geothermal and hydrostatic gradients (Iglauer, 2022). In line with CO_2 storage, a minimum depth of 800 m is recommended in order to avoid spoiling potable water resources (Sedlacek, 1999b; Chadwick et al., 2008).

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3.4 Cushion gas requirements

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The capacity of any gas reservoir consists of two components: cushion gas (also known as base gas), which is used to maintain the minimum reservoir pressure and hence desired production levels, and working gas, which refers to the portion of gas that can be withdrawn and used during a storage cycle. In underground gas storage the composition of the working and cushion gas is generally the same. The total amount of cushion gas required depends mainly on the type of storage, with pore storages requiring higher cushion gas shares than salt caverns. However, no general recommendations can be made because the proportion of cushion gas in pore storage facilities depends on many reservoir parameters, such as storage volume, permeability and porosity, reservoir geometry, depth and the resulting pressure and temperature conditions. It also depends on technical configurations, such as well distributions, shut-in periods between injection and production, and the respective injection and production rates. In Germany, for example, the cushion gas share in natural gas pore storage facilities active and inactive ranges from 21 to 93% (Table 2).

Unfortunately, there are no direct experiences with pure hydrogen in pore storages we can draw on, but numerical reservoir simulations demonstrate that aquifers and depleted hydrocarbon fields can be operated with reasonable hydrogen recovery rates of 78–93% (Pfeiffer and Bauer, 2015; Feldmann et al., 2016; Sáinz-García et al., 2017; Heinemann et al., 2021b; Kanaani et al., 2022). Simulations of saline aquifers show that the injected cushion gas can efficiently displace

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formation water from injection and production wells, creating the conditions for subsequent hydrogen injection and withdrawal with significant improvements in UHS performance (Sáinz-García et al., 2017; Heinemann et al., 2021b; Kanaani et al., 2022). This also results in a lower demand for cushion gas in deeper structures and reservoirs with higher permeabilities (Heinemann et al., 2021b). Since hydrogen in particular involves considerable upfront costs, alternative cushion gas compositions such as nitrogen, CO_2 or natural gas have been discussed in the literature to cut costs and/or CO_2 emissions (Foh et al., 1979; Pfeiffer and Bauer, 2015; Feldmann et al., 2016; Kanaani et al., 2022; Rhouma et al., 2022). Hence, the possibility of mixing with an inert cushion gas that is present in the reservoir must be considered (Foh et al., 1979). Gas mixing depends on viscosity and density contrasts (see Figures 1, 2), and is governed by diffusion and dispersion (Tek, 1989; Feldmann et al., 2016). A process that is particularly important to consider when converting depleted hydrocarbon fields with high gas saturation and low reservoir pressures (Feldmann et al., 2016). However numerical simulations of depleted gas reservoirs indicate that undesired physical phenomena such as viscous fingering caused by mobility and density contrasts are of minor importance in gas saturated reservoirs (Feldmann et al., 2016). Numerical simulations in saline reservoirs with hydrogen and nitrogen as working and cushion gas, respectively, demonstrate likewise the overall feasibility, but stress the importance of optimized injections schemes as the volume-weighted average hydrogen fraction of the produced gas increased from 52% during the first cycle to 85% in the fourth cycle (Pfeiffer and Bauer, 2015). In a most recent study of aquifer storage and withdrawal capacity, Kanaani et al. (2022) demonstrated likewise the operational feasibility and indicated that methane outperforms CO_2 and nitrogen in terms of hydrogen recovery. The general feasibility has also been demonstrated in converted town gas storages across Europe (Foh et al., 1979; Foh, 1991). At Beynes, for example, natural gas was being injected on one side of the reservoir from 1972 while town gas had been withdrawn from the other side of the reservoir, replacing all of the working and 40% of the cushion gas with natural gas (Foh et al., 1979; Foh, 1991). Despite the fact that 60% of town gas resided in the reservoir, only minimal mixing of gases occurred during conversion, and less than 1% of the withdrawn gas consisted of the original town gas after conversion was completed in 1973 (Foh, 1991). It follows that detailed reservoir models are required to draw accurate conclusions about the demand and cushion gas requirements for any pore storage system, despite its general feasibility.

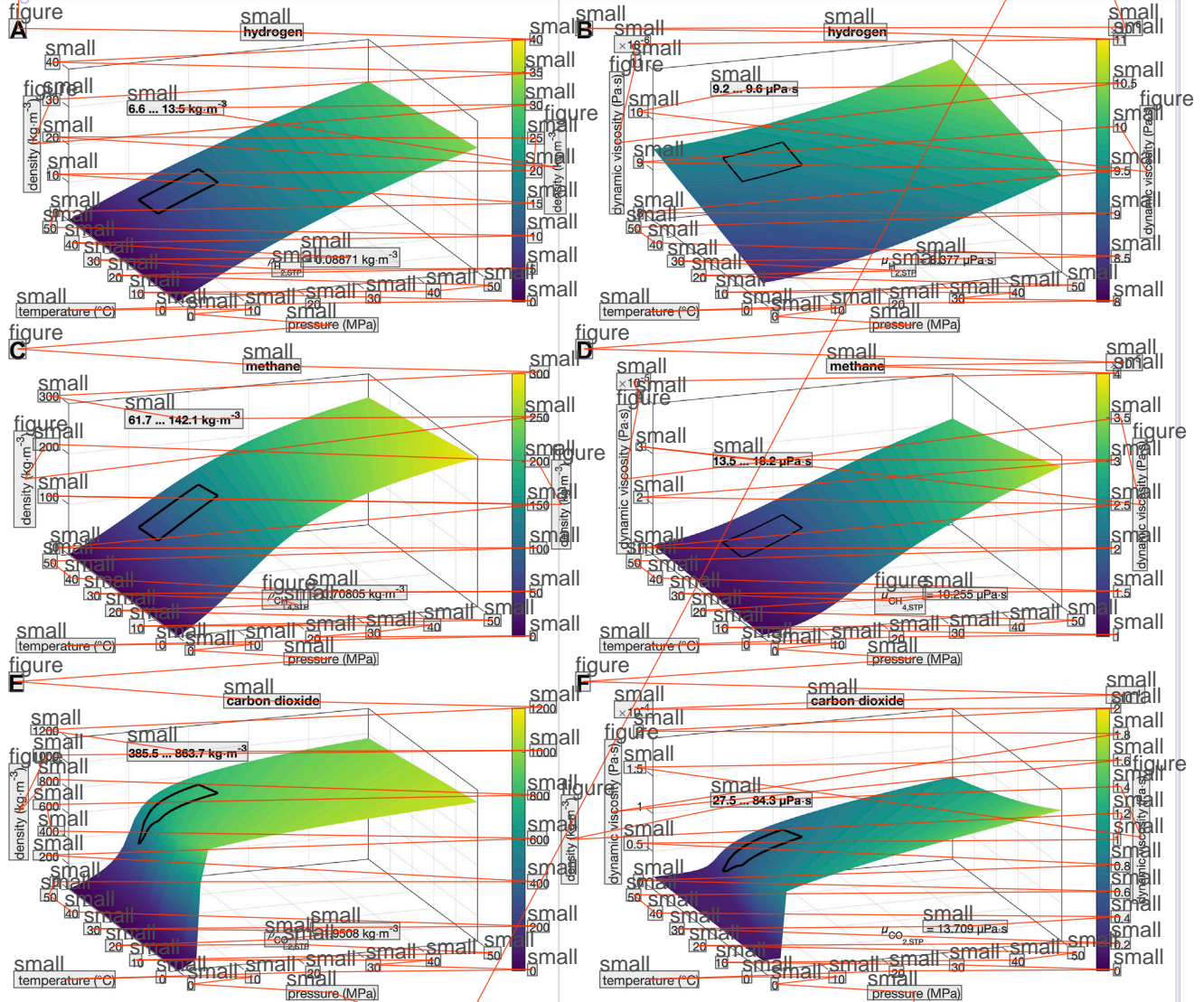
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3.5 Hydrogen solubility and losses

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The solubility of hydrogen in brine water depends on a number of factors, including temperature, pressure, and salinity, but it is generally lower compared to other gases such as CO_2 or CH_4 (e.g., Zivar et al., 2021). Hence, less loss due to dissolution is expected compared to other gases. Experimental and modelled data suggest that hydrogen solubility increases with increasing pressure, but decreases with increasing salinity resulting in less than 0.001 of mole fraction of hydrogen at storage conditions (Wiebe and Gaddy, 1934; Li et al., 2018; Chabab et al., 2020). This is important, because dissolved hydrogen may diffuse and leak out of the reservoir resulting in hydrogen loss. The rate of diffusion will likewise depend on a number of factors, including the temperature and pressure. Diffusion rates of hydrogen in brine water can be estimated

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FIGURE 1

Pressure and temperature dependence of (A,C,E) density and (B,D,F) dynamic viscosity of (A,B) hydrogen, (C,D) methane, and (E,F) carbon dioxide under conditions for UHS systems. All density and dynamic viscosity values shown are from the NIST Chemistry web book database (Lemmon et al., 2023) and were calculated according to Leachman et al. (2009) (hydrogen), Setzmann and Wagner (1991) (methane), and Span and Wagner (1996) (carbon dioxide). The areas marked by black rectangles indicate the ranges of density and dynamic viscosity expected at the proposed optimal reservoir depths (see text). The values indicated with the STP indices refer to the corresponding values at standard pressure and temperature conditions.

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using theoretical and empirical relationships (Einstein, 1906; Wilke and Chang, 1955), but have also been examined experimentally (Akgerman and Gainer, 1972), showing that at storage conditions, the rate of hydrogen diffusion is approximately twice as high when compared to methane and three times as high when compared to carbon dioxide (Akgerman and Gainer, 1972).

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3.6 Thermophysical parameters of hydrogen at underground storage conditions

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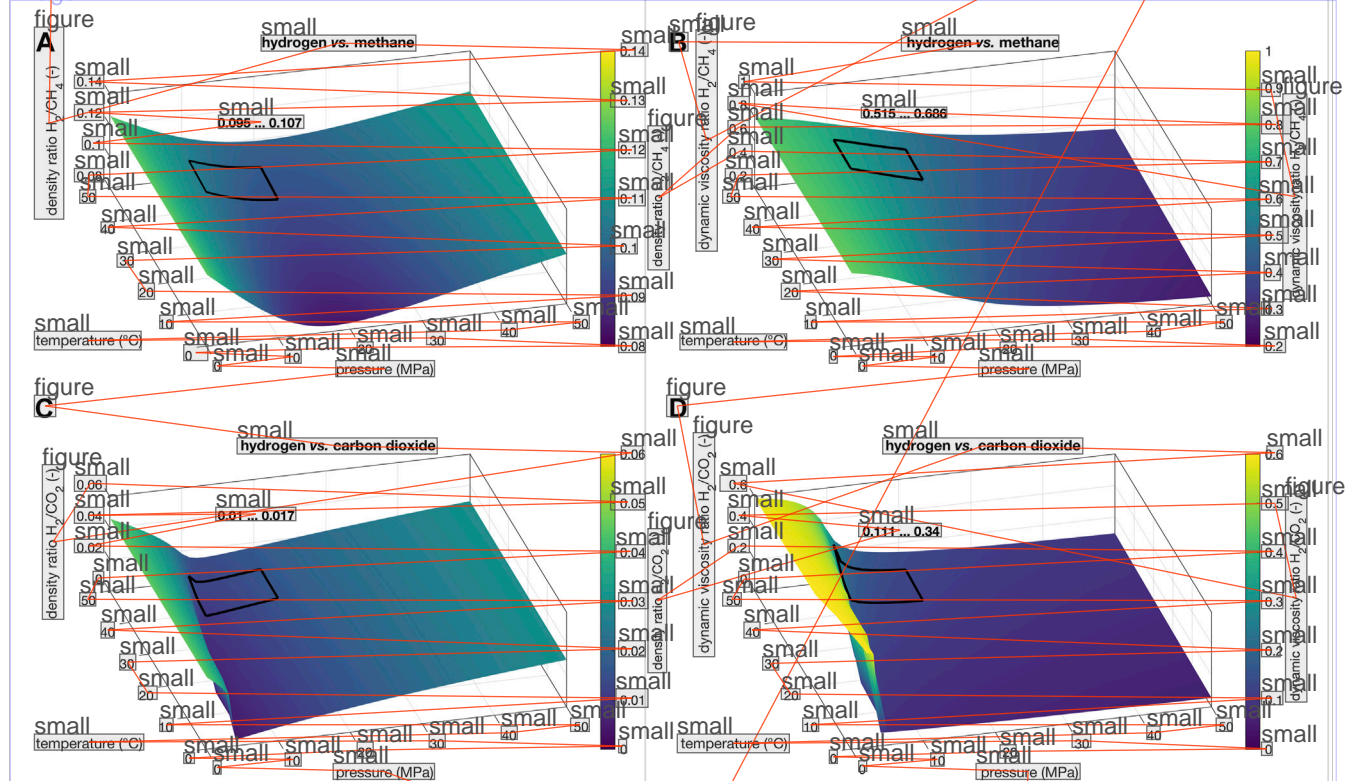
Comprehension of the thermophysical properties of hydrogen is essential for a detailed consideration of its storage potential in UHS systems. In particular, with regard to the storage of hydrogen in aquifers and depleted hydrocarbon reservoirs, a detailed consideration of its two

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thermophysical properties, density and viscosity, is essential under pressure and temperature conditions prevailing in the storage system. For this application, it is also useful to compare these properties of hydrogen with those of methane (CH₄) and carbon dioxide (CO₂) (Figures 1, 2). All thermophysical properties of H₂, CH₄, and CO₂ presented here were obtained from the NIST chemistry web book database (Lemmon et al., 2023) and are intended to serve as a reference for upcoming UHS systems.

One of the most notable thermophysical characteristics of hydrogen is its low density, which dictates that it must be compressed or liquefied for any practical application (Osman et al., 2021). At standard temperature (0°C or 273.15 K) and pressure (0.1 MPa or 1 bar), the density of hydrogen is 0.08871 kg/m³ according to the equation of state for normal hydrogen (Leachman et al., 2009). Due to the low critical temperature (−240.005°C or 33.145 K) and low critical pressure

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FIGURE 2

Pressure and temperature dependence of (A,C) density ratio and (B,D) dynamic viscosity ratio of hydrogen relative to (A,B) methane and (C,D) carbon dioxide under conditions for UHS systems. All density ratios and dynamic viscosity ratios shown are from the NIST Chemistry web book database (Lemmon et al., 2023) and were calculated according to Leachman et al. (2009) (hydrogen), Setzmann and Wagner (1991) (methane), and Span and Wagner (1996) (carbon dioxide). The areas marked by black rectangles indicate the ranges of density ratios and dynamic viscosity ratios expected at the proposed optimal reservoir depths (see text).

Content

(1.2964 MPa or 12.964 bar) of hydrogen (Leachman et al., 2009), it would be stored in the gaseous phase in porous storage systems (Züttel, 2004). Hydrogen is a highly compressible gas, and its density increases significantly with increasing pressure and barely with increasing temperature (Figure 1A). Consequently, hydrogen storage efficiency increases with increasing reservoir depth. But, because CH_4 (Setzmann and Wagner, 1991) and CO_2 (Span and Wagner, 1996) is much denser than hydrogen (Figures 2A, C), storing hydrogen gas of the same mass requires higher pressure (Tarkowski et al., 2021) and less hydrogen can be stored in the same reservoir volume compared to CH_4 and CO_2 (e.g., Lanz et al., 2001; Lankof and Tarkowski, 2020; Zivar et al., 2021; Epelle et al., 2022). This underlines the importance of storage capacity in hydrogen storage. In this context, depleted hydrocarbon storage facilities represent a more interesting storage option compared to salt caverns and aquifers due to their large storage space and high availability (e.g., Reitenbach et al., 2015). Moreover, the large density difference between H_2 and dense formation water and cushion gas (e.g., CH_4) can cause a strong gravity segregation effect (Ide et al., 2007; Jamshidnezhad et al., 2010; Rossen et al., 2010; Han et al., 2016; Rabinovich and Cheng, 2020), resulting in the accumulation of a hydrogen cap beneath the cap rock (Ruith and Meiburg, 2000; Heinemann et al., 2021c; Muhammed et al., 2022), which could ensure that H_2 will remain in the reservoir (Iglauer, 2022). Conversely, Osman et al. (2021) emphasized that an H_2 plume is susceptible to formidable buoyancy forces, with the magnitude of said

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forces directly correlating to the potential for hydrogen leakage. Literature by Heinemann et al. (2021c) and Sáinz-García et al. (2017) suggests that water may ascend towards the production well under these conditions, making seasonal storage of hydrogen in porous storage systems challenging.

While the hydrogen density and thus the compressibility in the depth range of UHS systems mainly increases with increasing pressure (Figure 1A), the dynamic viscosity increases significantly with increasing temperature (Figure 1B). In the range of standard conditions up to pressures of 50 MPa and temperatures of 50°C, the viscosity shows a comparable temperature and pressure dependence. For example, hydrogen viscosity increases by about 15% from standard temperature and pressure conditions to an elevated temperature of 50°C (at standard pressure; Leachman et al., 2009). At standard temperature, viscosity increases by about 18% in the pressure range from 0.1 to 50 MPa (Leachman et al., 2009). In direct comparison, the viscosities of CH_4 and CO_2 are larger than those of H_2 , and both show a dominant pressure dependence that far exceeds the behavior of H_2 viscosity (Figures 2B, D). Hence, hydrogen has a much higher mobility than CH_4 and CO_2 . The viscosities of H_2 and CH_4 are still most similar here in comparison and are in the same order of magnitude. As elaborated by Muhammed et al. (2023), underground hydrogen storages could thus be used as a direct substitute for CH_4 storage. Due to their viscosities, according to Edlmann et al. (2016) and Heinemann et al. (2018), H_2 and CH_4 can be stored in a wide range

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of underground reservoirs because they are not limited by reservoir conditions or depth. Rather, reservoirs are determined based on the geologic characteristics of the reservoirs, as appropriate containment features are required to seal against these mobile gases (Muhammed et al., 2023). But, Paterson (1983) and numerous subsequent studies have shown that the low density, viscosity, and molecular size of H_2 result in more viscous fingering and poorer conformation during injection, faster migration toward structurally high sites, and some amount of unrecoverable hydrogen (e.g., Hagemann et al., 2014; Feldmann et al., 2016). For this reason, Epelle et al. (2022), for example, recommend adjusting the flow rate of hydrogen and the size of the wellbore (Sørensen, 2007; Zivar et al., 2021) compared to operation with natural gas to minimize diffusion in porous media.

As previously described by us and other studies (e.g., Iglauer, 2022), the optimal depth, d , for UHS systems is in the range of 800–1,100 m depth. Assuming a mean surface temperature of 8.2°C and an average geothermal gradient of 32°C/km (Agemar et al., 2012), a temperature range of 33.8–43.4°C results for the target depth. For an average density, ρ_{lith} , of the overlying rock of 2,300–2,600 kg/m³ and a lithostatic factor, F_{lith} , of 0.5–0.7 (Foh et al., 1979), the pressure at which hydrogen could be stored in UHS systems can be determined using the following simple relationship:

$$p_{H_2} = F_{\text{lith}} \times d \times 9.81 \text{ m} \cdot \text{s}^{-2} \times \rho_{\text{lith}} = 9.03 \dots 19.64 \text{ MPa}. \quad (2)$$

Content

This temperature and pressure range corresponds to the optimal conditions for the geological storage of hydrogen in porous storage systems (black rectangles within Figures 1, 2). The so-called lithostatic factor was introduced here to represent the ratio of working pressure to lithostatic pressure in the reservoir. For this typical geological conditions for porous media storage, hydrogen has a low density and viscosity in the narrow ranges of 6.6–13.5 kg/m³ and 9.2–9.6 $\mu\text{Pa} \cdot \text{s}$, respectively (Figures 1A, B). The influence of temperature and pressure variations on the density and viscosity of highly compressible hydrogen is thus almost negligible at this reservoir depth. In contrast, the thermophysical properties of CO_2 and CH_4 behave in a much more pressure- and temperature-dependent manner, which is reflected in the relative density and viscosity ratios (Figure 2). Within the proposed reservoir range, the density and viscosity ratios of H_2 and CH_4 do not vary too much (max. $\pm 25\%$; Figures 2A, B), whereas the corresponding ratios of H_2 to CO_2 can vary by up to a factor of about 2–3 (Figures 2C, D). This has to be considered when re-using former natural gas reservoirs with H_2 .

Heading

4 Capacity estimates for UHS in Germany

Headline

4.1 Technical hydrogen storage capacity in former and current UGS systems

Content

Underground gas storage facilities can be adapted for hydrogen storage and some of the currently operating underground storage facilities have already stored town gas with a hydrogen content of about 50–60% in the past. Thus, in a first step, we focus on the installed capacity of underground pore storages in Germany. The hydrogen storage capacity is calculated based on the recoverable volume of natural gas as reported by the State Authority for Mining,

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Energy and Geology in Germany (Landesamt für Bergbau, Energie und Geologie, LBEG). These data are based on empirical values, which are derived from technical and economic considerations for natural gas and, thus, provide sound estimates of the dynamic recovery of hydrogen. The amount of energy stored as hydrogen in the working gas, which describes the portion of gas that is cycled in and out of the reservoir, E_H , is calculated according to the modified equation after Mouli-Castillo et al. (2021), that writes

$$E_{H_2} = HHV_{H_2} \times WV_{CH_4} \times \frac{\rho_{CH_4,STP}}{\rho_{CH_4,res}} \times \frac{\rho_{H_2,res}}{\rho_{H_2,STP}}, \quad (3)$$

Content

where HHV_{H_2} , WV_{CH_4} , $\rho_{H_2,STP}$, and $\rho_{CH_4,STP}$ denote the higher heating value of hydrogen (3.54 kWh/m³), the working gas volume of natural gas in the storage system at STP, and the density of hydrogen and methane at STP, respectively. $\rho_{H_2,res}$ and $\rho_{CH_4,res}$ refer to the hydrogen and methane density at pressure and temperature when the reservoir is full (P_{max}). However, estimates on the operating gas pressure $P_{\text{min/max}}$ are not publicly available and we, therefore, estimate the maximum gas operation pressure to be 70% of the overburden pressure in accordance with Foh et al. (1979). In Eq. 3, the product of the density ratio is utilized for the conversion of reported methane working gas volumes at STP to hydrogen gas volumes at STP. However, our data suggest that a constant conversion parameter of 0.85 can be used over a range of temperatures and pressures relevant to underground gas storage, as shown in Table 2. The results are summarized in Table 2 and Figure 3. At present, there are 16 underground pore storage facilities installed in Germany that have total installed capacity of around 8 bcm of natural gas with a thermal energy content of 95.244 TWh, estimated using a higher heating value of 11.07 kWh/m³ for methane. This represents about one-third of the total underground storage capacity of natural gas in Germany. The remaining natural gas is stored in 272 salt caverns. In contrast, the corresponding hydrogen storage capacity is lower, amounting to around 7 bcm of hydrogen gas with a thermal energy content of 25 TWh, given its lower energy content per volume (3.54 kWh/m³). The distribution of these pore storages is shown in Figure 3. Hence, more volume is needed to store the same amount of energy compared to natural gas and other common fuels. Furthermore, decommissioned pore storage facilities could potentially provide an additional storage potential of 1 bcm of hydrogen or 4 TWh of hydrogen energy equivalent. Hence a total of 8 bcm of hydrogen, which amounts to 29 TWh of hydrogen energy equivalent could be stored in current and former UGS pore storage systems. Figure 3 shows the distribution of all pore storage (active and inactive) and compares it to the planned network expansion initiative IPCEI.

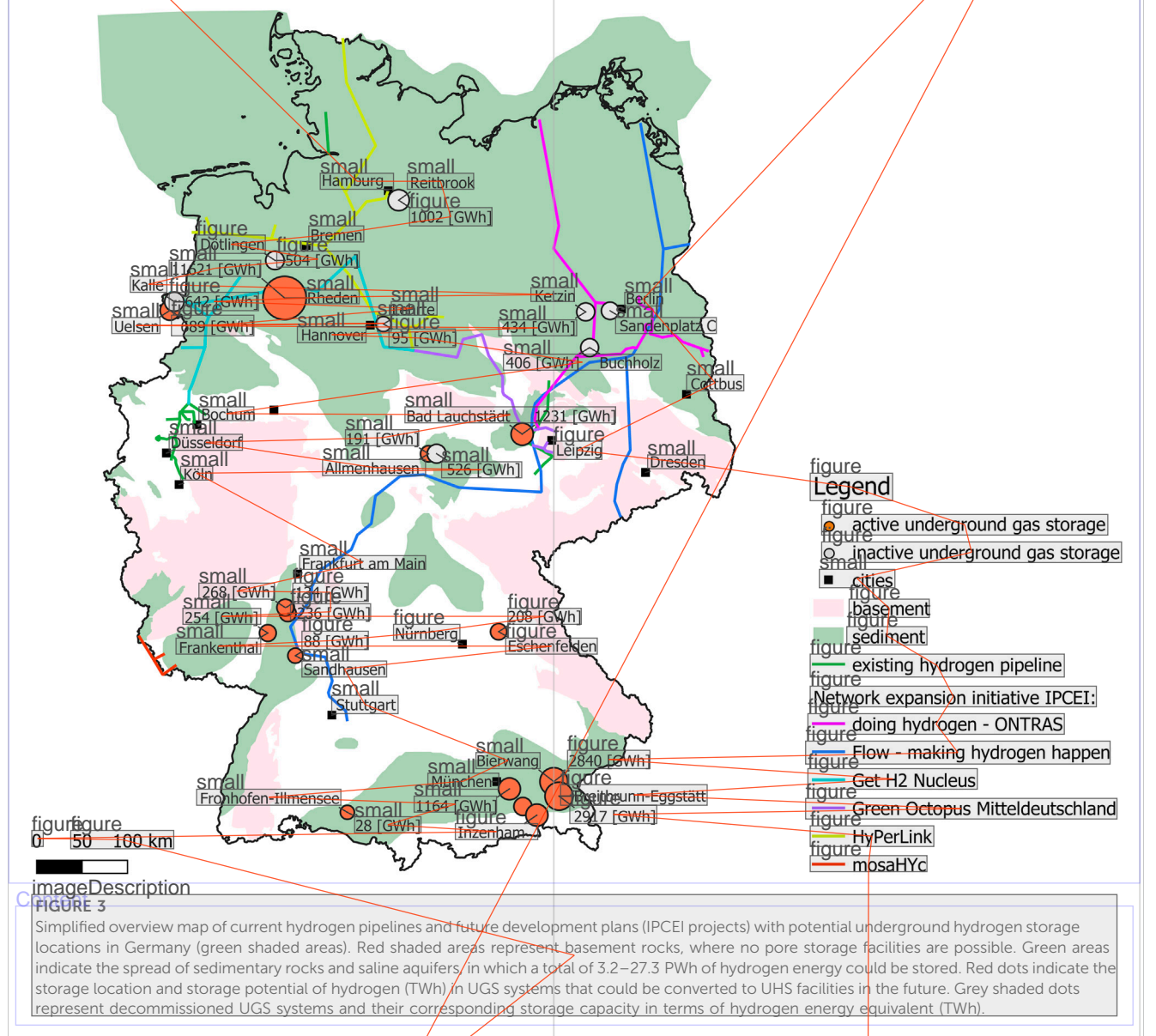
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4.2 Theoretical hydrogen storage capacity in saline aquifers

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Additional underground storage capacities may become available in saline aquifers that hold the highest storage potential due to their widespread occurrence. In order to assess the hydrogen storage capacity in saline aquifers, we employ the volumetric assessment of CO_2 storage capacities in saline aquifers, as previously conducted by Knopf and May (2017). This probabilistic approach yields a proven, median, and

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possible CO₂ storage capacity of 20.4 Gt (P10), 49.1 Gt (P50), and 115.3 Gt (P90) for all evaluated rock units (onshore and offshore; green shaded areas in Figure 3), respectively. Most of these resources are located in the North German Basin with estimated CO₂ storage capacities ranging from 19.3 Gt (P10)–108.6 Gt (P90) (Knopf and May, 2017). By applying this data, we can estimate the thermal hydrogen energy content using

$$E_{H_2} = M_{CO_2} \times HHV_{H_2} \times RG \times \frac{\rho_{H_2}}{\rho_{CO_2}} \quad (4)$$

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where E_{H_2} is the energy content of hydrogen (in PWh), M_{CO_2} is the mass of stored CO₂ (in Gt), HHV_{H_2} is the higher heating value of hydrogen (39.4 kWh/kg), RG is the recoverable working gas volume (i.e., portion of the working gas volume: 0.4–0.6), and the ratio $\frac{\rho_{H_2}}{\rho_{CO_2}}$ is used to convert the mass of stored CO₂ to the mass of stored hydrogen. Knopf and May (2017) set an upper vertical limit of 800 m (Reinhold et al., 2011) in their analysis, at which

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CO₂ is typically subjected to temperatures and pressures that preserve CO₂ in a supercritical state. At these conditions the resulting CO₂ densities are two orders of magnitude greater compared to gaseous H₂ (Figure 1). The CO₂ storage capacity can therefore be converted to the mass of stored hydrogen, M_{H_2} (in Gt), based on a conversion factor of 0.01 (i.e., $\frac{\rho_{H_2}}{\rho_{CO_2}} \approx 0.01$; Figure 2C). Hence, saline aquifers may yield an additional capacity of 81.6 (Mt)–691.8 (Mt) working gas volume of hydrogen resulting in 3.2–27.3 PWh of hydrogen energy equivalent, but almost all of these storage capacities are restricted to the North German Basin, which may hold 3.0–25.7 PWh of hydrogen energy equivalent. Nevertheless, one should keep in mind that these estimates represent theoretical capacities, and the extent to which these reserves can actually be used should be verified based on the storage criteria proposed here, but will also depend on technical, economic, environmental, and social aspects and requires further analysis.

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5 DiscussionHeadline
5.1 Storage capacities and requirements in Europe

Content

Existing and decommissioned underground pore storage facilities may account for up to 30 TWh of energy equivalent of hydrogen, but are unlikely to meet future storage requirements of 34–667 TWh (Lux et al., 2022), not least because an overall conversion of all storage facilities from natural gas to hydrogen is unlikely. To meet these future storage requirements, additional capacities may be needed in the form of depleted hydrocarbon fields and saline aquifers. Approximately 3.2–27.3 PWh of hydrogen energy equivalent could be stored in saline aquifers in Germany, which is comparable to the technical storage potential in salt caverns that has been estimated at 35.7 PWh (Caglayan et al., 2020) and similar to the hydrogen storage potential in saline aquifers in the UK with an estimated capacity of 2.1 PWh of energy equivalent of hydrogen (Scafidi et al., 2021). In practice, many of the storage sites may, however, prove unsuitable, given the technical, economic and social constraints or not least because of the increasing competition between different geotechnical applications, be it geothermal energy or storage of natural gas, CO₂, or hydrogen (Suchi et al., 2014).

Other additional storage options that have been discussed in the literature include depleted gas and oil fields, which are considered appropriate for hydrogen underground storage because of their proven containment of natural gas, well explored subsurface, and developed capacity and infrastructure (e.g., see reviews by Raza et al., 2022; Zivar et al., 2021; Heinemann et al., 2021c). Capacity estimates in the United Kingdom and the Netherlands show large potentials. Scafidi et al. (2021) identified 95 suitable depleted gas fields on the UK continental shelf with a total working gas capacity of 6.9 PWh energy equivalent of hydrogen. This estimate is slightly higher than the total hydrocarbon storage capacity in gas fields reported by Mouli-Castillo et al. (2021), which is based on the initial gas-in-place capacity of the respective hydrocarbon fields resulting in 2.7 PWh energy equivalent of hydrogen. Large storage potentials in depleted gas fields were also predicted for the Netherlands, one of the largest producers of natural gas in Europe. Estimates based on the original-gas-in-place analysis show that onshore capacities of 277 TWh and offshore capacities of 179 TWh could add up to a total of 456 TWh energy equivalent of hydrogen in the Netherlands, which is an order of magnitude higher than the estimated hydrogen storage potential in salt caverns of 43 TWh. In Germany, too, depleted gas and oil fields could offer large storage capacities, but estimates on the original gas in place capacity are not publicly available, only cumulative production numbers (LBEG, 2021). Cumulative gas production from all fields at STP is $1.058512 \times 10^{12} \text{ m}^3$ (current as of 1 January 2021; LBEG, 2021). Of this, $114.719 \times 10^9 \text{ m}^3$ of natural gas was produced from gas fields that are depleted today, most of it in Lower Saxony (LBEG, 2021).

However, more important than uncertain volume estimates, is the question what proportion of this total can be utilized for underground hydrogen storage under the parameters presented here. For example, as fields in the Weser-Ems and Elbe-Weser sector located in the Rotliegend formation may reach depths of 2,500–5,000 m (LBEG, 2021). From an economic point of view,

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storing gas at these depths is not profitable due to the high compression costs (e.g., Foh et al., 1979). Typical depths of natural gas storage facilities are 1,000–1,200 m (Sedlacek, 1999a). Physical constraints limit the storage of hydrogen to depth of 3,700 m below which hydrogen is thought to percolate through the cap rock and, hence, confinement loss (Iglauer et al., 2021). Details on the specific gas production, horizon and depth are, however, not listed. Former gas deposits in carbonates of the Stassfurt formation (Ca₂), common in the eastern part of the North German Basin and the Thuringian Basin, may prove unsuitable as they contain significant amounts of hydrogen sulphur. However, former gas deposits in the Tertiary sandstones in the Upper Rhine Graben could again show suitable potential and supplement storage capacities in saline aquifers and the UGS facilities Stockstadt, Hähnlein, and Frankenthal. According to forecasts, this could be the end point of the North-South hydrogen pipeline. However, the nearby industrial clusters would depend on an economical and steady supply of hydrogen, which could be ensured by underground hydrogen storage facilities. For the Munich area, on the other hand, there are no pipeline plans according to the Gas Network Development Plan 2022–2032 or the IPCEI project plan. The question, therefore, arises as to how the supply will be ensured, and here, too, underground hydrogen storage could be of significance. Future research should therefore focus on the technical storage potential matching it with planned hydrogen infrastructures and industrial demands. This is of paramount importance in the South of Germany where storage potential is lower than in the North, making targeted planning necessary.

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5.2 Assessment of underground hydrogen storage costs

Content

Early economic analyses have shown that the cost of service for hydrogen is approximately equal to the costs of storing natural gas, but have stressed that the costs are very site specific and that a range of costs is possible for each type of underground storage (Foh et al., 1979). In a more recent life cycle cost analysis, Lord et al. (2011) conclude on a cost per kg of hydrogen basis that depleted hydrocarbon reservoirs and aquifers would be the economically attractive options, considering costs for pipelines and wells, compressors, cavern mining, and base gas. The overall costs for UHS amount to 1.6 \$/kg for salt caverns, 0.04 \$/kg for depleted hydrocarbon fields, 0.08 \$/kg for aquifers, and 2.75 \$/kg for hard rock caverns (Lord et al., 2011). This methodology has been adapted with most recent cost estimates as of 2019 in a review by Raza et al. (2022) for UHS to be 0.14–14.03 \$/kg (0.004–0.421 \$/kWh) for salt caverns, 1.42 \$/kg (0.043 \$/kg) for depleted hydrocarbon fields, 1.49 \$/kg (0.045 \$/kWh) for aquifers, and 0.36–3.20 \$/kg (0.011–0.096 \$/kWh) for hard rock caverns. However, a comparison of natural gas storage costs in Europe and the United States shows that storage costs in Europe could be twice as high, not least because of higher regulatory requirements (Sedlacek, 1999a). But unlike in the United States, where these figures are sometimes published, cost estimates in Europe are rather poor (Sedlacek, 1999a). Something that all cost estimates for hydrogen pore storages have in common is the substantial

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influence of cushion gas on the ultimate costs of service, similar to natural gas storage facilities (Foh et al., 1979; Lord et al., 2011). At the same time it is considered more economic to leave some gas in the reservoir than to invest in the additional pumps and compressors to empty it completely (Foh et al., 1979). Therefore, one way to reduce these costs is to use lower cost cushion gas, but further research is needed to investigate the possibility of mixing, viscous fingering and gravity override that needs to be considered (see Section 3.4; Feldmann et al., 2016). The economics will therefore depend on the quantity of the cushion gas, the quality of the recovered hydrogen gas, and costs of gas separation if required (Foh, 1991).

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5.3 Hydrogen purity requirements

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Probably the most important parameters influencing the economic development of underground hydrogen storage are biotic and abiotic reactions. Potential contaminations may result from biotic reactions, such as for instance in the town gas storage Lobodice (Czech Republic) in which methanogens were shown to transform hydrogen and carbon dioxide to methane (Šmigáň et al., 1990). Or abiotic reactions like in the in the town gas storage Beynes (France) where pyrite reduction is believed to be responsible for the observed rise in hydrogen sulphide within the reservoir (Bourgeois et al., 1979). There are several process engineering options for removing hydrogen sulphide from natural gas and the hydrocarbon industry has decades of experience. In Germany, too, a large proportion of all natural gas reserves contain varying amounts of hydrogen sulphur, which have been processed at the Großenkneten desulfurization plant (Lower Saxony) since the 1970s. Here the hydrogen sulphur is removed from the gas and converted to elemental Sulphur used as feedstock by the chemical industry. However, this is a costly process and the purity requirements of hydrogen extracted from the storage facility will ultimately depend on the end user. Fuel cells, for example, require hydrogen of 99.97% purity according to Beuth Verlag GmbH, (2019) because contaminants such as carbon monoxide and hydrogen sulfide can impede electrode charging, while ammonia causes irreversible effects on fuel cell performance. In contrast, fuel quality specification for applications other than PEM fuel cell, i.e., combustion processes in hydrogen boilers will tolerate higher concentrations of impurities of about 98% (ISO, 2019). In contrast, industrial combustion applications (e.g., for sintering processes, heating of furnaces) and domestic appliances (boilers, cookers, etc.) are more tolerant with respect to the hydrogen content and impurity levels resulting in fuel-types similar to “town gas” with hydrogen contents of around 50%. Generally, these higher impurity levels could facilitate hydrogen utilization by increasing the heating value but replacement by other components, e.g., oxygen or carbon monoxide have to be considered carefully with regards to safety, decarbonization and cost-effective level of purity for end user applications and quality measurements.

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6 Summary

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Hydrogen has the capacity to meet future emission targets by decarbonizing the industry, mobility, electricity, and heat generation. But, hydrogen utilization and the implementation of a hydrogen economy at large scale requires significant storage capacities. Underground hydrogen storage in porous geological structures has the potential to meet these storage requirements. Positive and cautionary indicators for safe hydrogen storage in porous structures were identified, and a comparison and linkage with existing and planned aboveground hydrogen infrastructures was established. We have identified porous sandstone formations located at depths ranging from 800 to 1,100 m and exhibiting temperatures exceeding 40°C as highly promising reservoirs for hydrogen storage. Specifically, porosities exceeding 20% and permeabilities greater than 500 mD were identified as positive indicators for adequate storage capacity. Brine pH values below 6 and salinities above 100 mg/L are desirable for effective hydrogen storage that will minimize biotic and abiotic reactions. Based on our estimates, the total hydrogen storage capacity in underground natural gas facilities is projected to be as high as 8 billion cubic meters, or equivalently, 0.72 Mt at standard temperature and pressure. This capacity translates into a potential energy equivalent of 29 TWh of hydrogen. Additionally, saline aquifers could accommodate 81.6–691.8 Mt of hydrogen, equivalent to an energy potential of 3.2–27.3 PWh of hydrogen that would meet predicted storage requirements many times over. Most of these sites are located in the North German Basin. We conclude that pore storage systems are likely to play a vital role in the future German hydrogen infrastructure, particularly in regions with significant demand for industrial hydrogen (storage) and potential imports via pipelines and ships. We recommend that future research focus on assessing the technical storage potential of these sites and their compatibility with planned hydrogen infrastructures and industrial demand.

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Data availability statement

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The original contributions presented in the study are included in the article/Supplementary Material, further inquiries can be directed to the corresponding author.

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Author contributions

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KA designed the study. KA and BA organized the database. BA performed the thermophysical analysis. KA wrote the first draft of the manuscript. KA, BA, MG, and MN wrote sections of the manuscript. All authors contributed to manuscript revision, read, and approved the submitted. All authors listed have made a substantial, direct, and intellectual contribution to the work and approved it for publication.

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Conflict of interest

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The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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