Electric Current

I = electric current, # of coulombs of charge that passes a given point

$$I = \frac{dq}{dt}$$

Units: C/s = 1 ampere = 1 A

 v_e = conduction velocity; average velocity between collisions $\approx 10^5~\text{m/s}$

 v_d = drift velocity; average velocity on the macroscopic scale $\approx 10^{-3}$ m/s

Carrier Density

let n = charge carrier density, # of free charges per cubic meter $(1/m^3)$

pure metal
$$n = \frac{\rho N_A}{M}$$

$$N_A = 6.022 \times 10^{23} \ \mathrm{mol^{-1}}$$

 ρ = volume density (kg/m³)

M = molar mass (kg/mol)

$$I = neAv_d$$

e = charge electron

let \vec{J} = current density (A/m²)

$$|ec{J}| = rac{I}{A}$$

$$|\vec{J}| = nev_d$$

Ohm's Law

$$\vec{J} = \sigma \vec{E}$$

 σ = electric conductivity, a measure of how well the material conducts charge

let ρ = electric resistivity, a measure of how well the material resists current flow

Units of ρ : Ω m

$$\rho = \frac{1}{\sigma}$$

$$\Delta V = \left(\frac{\rho l}{A}\right) I$$

let R = resistance

$$R = \frac{\rho l}{A}$$

Units of R: 1 $\frac{V}{A}$ = 1 ohm = 1 Ω

Resistivity

 ρ is not constant, it varies with temperature

$$\rho = \rho_0 (1 + \alpha (T - T_0))$$

let α = temperature coefficient of resistivity (1/°C)

$$R=R_0 ig(1+lpha ig(T_f-T_0ig)ig)$$
 , if l and A are constant

Power Supplies

example: battery (electro chemical cell)

• two dissimilar metals and an electrolyte

let ε = electromotive force, voltage generated/created by chemical reaction inside battery, (fixed) quantity doesn't change

 V_t = terminal voltage, voltage that's available to some external circuit (varies)

r = internal resistance (varies)

$$V_t = \varepsilon - Ir$$