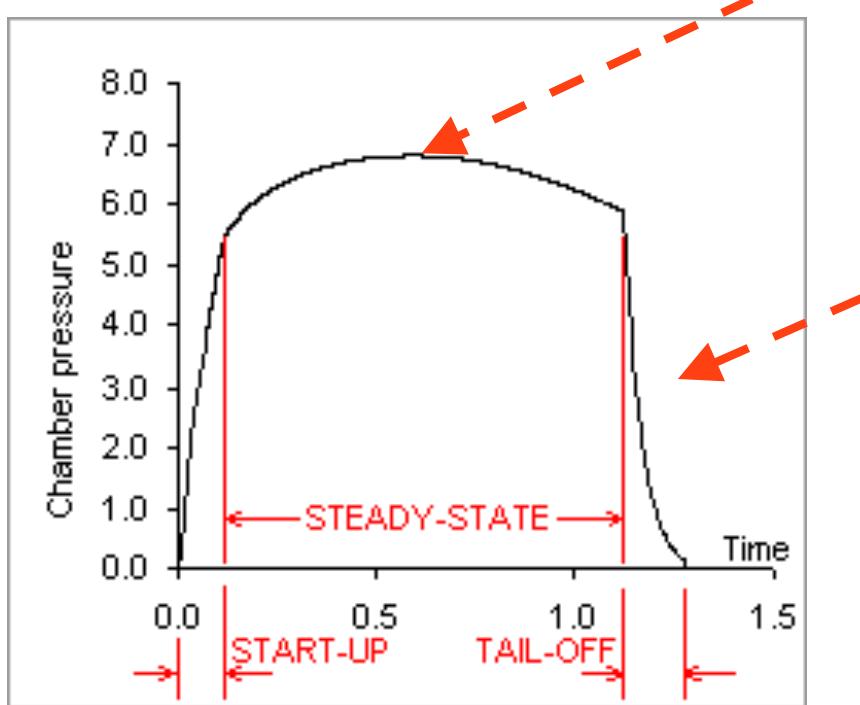


# Modeling Transient Rocket Operation

## (Lecture 6.1: Liquid Rockets)



- .. The primary goal of man is survival ... food, shelter ... basic necessities ...
- *A second aim of man is to build things that run very HOT and very LOUD and move really, really FAST ...*

Material Taken from

1. Sutton and Biblarz: Section 6.1, Section 8.1, Chapter 11, Chapter 15, Appendix 4
2. Humble and Henry, " Space Propulsion Analysis and Design"
3. Richard Nakka Web Page: [http://members.aol.com/ricnakk/th\\_pres.html](http://members.aol.com/ricnakk/th_pres.html)

## Transient Pressure Model

- Combustion Produces High temperature gaseous By-products
- Gases Escape Through Nozzle Throat
- Nozzle Throat Chokes (maximum mass flow)
- Since Gases cannot escape as fast as they are produced
  - ... Pressure builds up
- As Pressure Builds .. Choking mass flow grows
- Eventually Steady State Condition is reached

# Choking Massflow per Unit Area

- maximum Massflow/area Occurs when When  $M=1$

- Effect known as *Choking* in a Duct or Nozzle
- i.e. nozzle will Have a mach 1 throat

$$\left( \frac{\dot{m}}{A_c} \frac{\sqrt{T_0}}{p_0} \sqrt{R_g} \right)_{\max} = \left( \frac{\dot{m}}{A^*} \frac{\sqrt{T_0}}{p_0} \sqrt{R_g} \right) =$$

$$\frac{\sqrt{\gamma}}{\left[ 1 + \frac{(\gamma - 1)}{2} \right]^{\frac{\gamma + 1}{2(\gamma - 1)}}} = \sqrt{\gamma} \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma + 1}{(\gamma - 1)}} \rightarrow$$

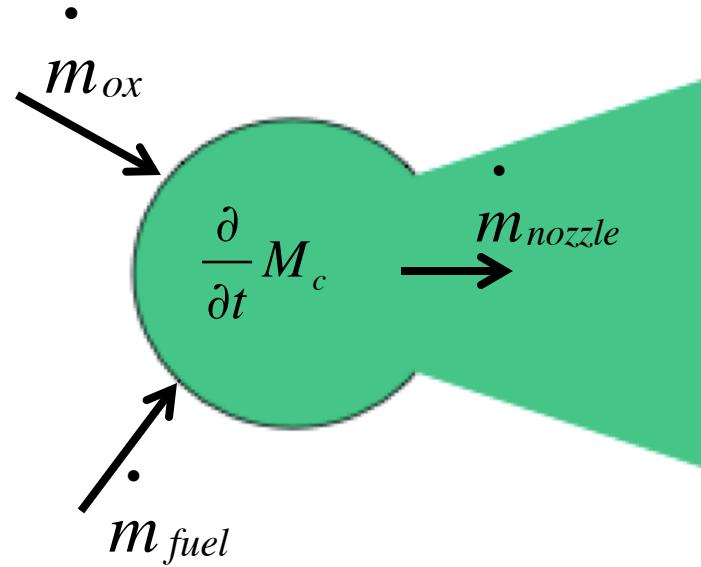
$$\frac{\dot{m}}{A^*} = \sqrt{\frac{\gamma}{R_g}} \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma + 1}{(\gamma - 1)}} \frac{p_0}{\sqrt{T_0}}$$

# Chamber Pressure Model

- Gaseous Mass Trapped in Chamber

$$\frac{\partial}{\partial t} M_c = \left[ \dot{m}_{fuel} + \dot{m}_{ox} \right] - \dot{m}_{nozzle}$$

$$\frac{\partial}{\partial t} M_c = \frac{\partial}{\partial t} [\rho_c V_c] = \frac{\partial}{\partial t} [\rho_c] V_c + \rho_c \frac{\partial}{\partial t} [V_c]$$



- Assuming nozzle chokes immediately

$$\frac{\partial}{\partial t} [\rho_c] V_c + \rho_c \frac{\partial}{\partial t} [V_c] = \left[ \dot{m}_{fuel} + \dot{m}_{ox} \right] - A^* \sqrt{\frac{\gamma}{R_g}} \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{(\gamma-1)}} \frac{P_0}{\sqrt{T_0}}$$

# Chamber Pressure Model (cont'd)

- Using ideal gas law, *Assuming constant flame temperature*

$$\rho_c = \frac{P_0}{R_g T_0} \rightarrow \frac{\partial}{\partial t} [\rho_c] \approx \frac{1}{R_g T_0} \frac{\partial}{\partial t} [P_0]$$

- Subbing into mass flow equation

$$\frac{\partial P_0}{\partial t} \frac{V_c}{R_g T_0} + \frac{P_0}{R_g T_0} \frac{\partial V_c}{\partial t} = \left[ \dot{m}_{fuel} + \dot{m}_{ox} \right] - \frac{R_g T_0}{V_c} A^* \sqrt{\frac{\gamma}{R_g}} \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{(\gamma-1)}} \frac{P_0}{\sqrt{T_0}}$$

$$\frac{\partial P_0}{\partial t} + P_0 \frac{1}{V_c} \frac{\partial V_c}{\partial t} + \frac{R_g T_0}{V_c} A^* \sqrt{\frac{\gamma}{R_g}} \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{(\gamma-1)}} \frac{P_0}{\sqrt{T_0}} = \frac{R_g T_0}{V_c} \left[ \dot{m}_{fuel} + \dot{m}_{ox} \right]$$

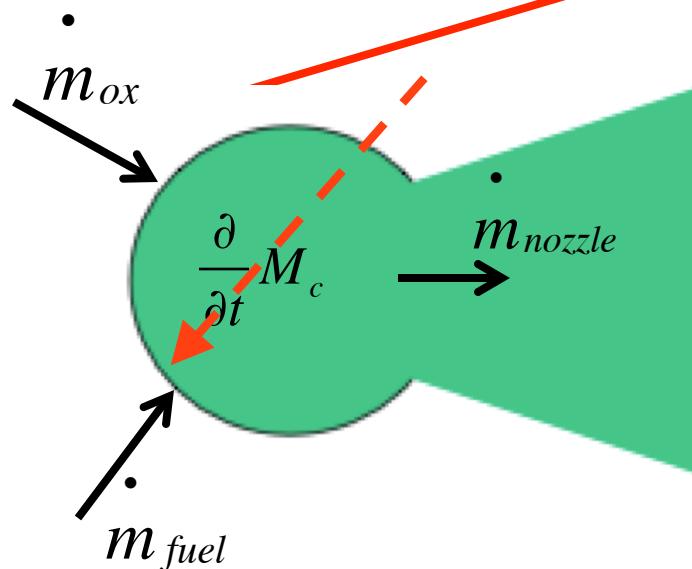
$$\frac{\partial P_0}{\partial t} + P_0 \left[ \frac{1}{V_c} \frac{\partial V_c}{\partial t} + \frac{A^*}{V_c} \sqrt{\gamma R_g T_0} \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{(\gamma-1)}} \right] = \frac{R_g T_0}{V_c} \left[ \dot{m}_{fuel} + \dot{m}_{ox} \right]$$

# Chamber Pressure Model (cont'd)

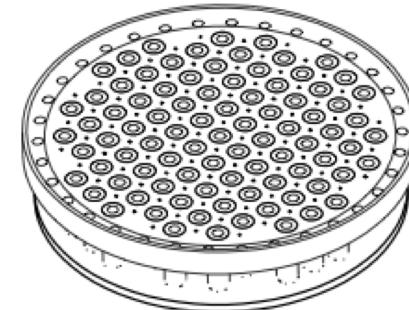
- Chamber pressure equation

$$\frac{\partial P_0}{\partial t} + P_0 \left[ \frac{1}{V_c} \frac{\partial V_c}{\partial t} + \frac{A^*}{V_c} \sqrt{\gamma R_g T \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{\gamma-1}}} \right] = \frac{R_g T_0}{V_c} \left[ \dot{m}_{fuel} + \dot{m}_{ox} \right]$$

- How do we estimate mass flows?



Injector design

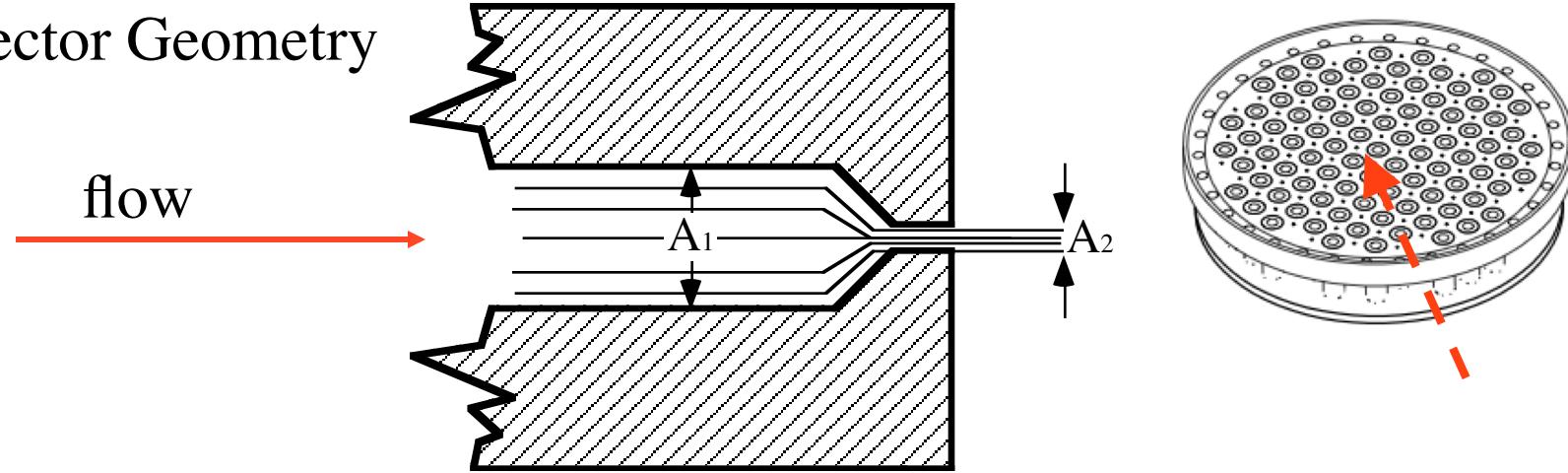


# Incompressible Injector Equation

- Many Liquid Propellants are Essentially Incompressible Fluids
- Incompressible Fuel Examples:  
*Kerosene (RP-1, RP-4), Ethanol, Methanol, UDMH (Unsymmetrical Dimethyl Hydrazine), MMH (Mono Methyl Hydrazine), Ammonia, Hydrazine, ~Liquid Hydrogen, etc.*
- Incompressible Oxidizer Examples:  
*Hydrogen Peroxide, Liquid Flourine, Nitrogen Tetraoxide, Nitric Acid, ~Liquid Oxygen, etc.*
- Incompressible Assumption Allows Simplified Form of Injector Equations

## Incompressible Injector Eq. (2)

- Injector Geometry



- Assume Liquid Propellants are incompressible ( $\rho=const$ )

- Momentum  $p_1 + \frac{1}{2} \rho V_1^2 = p_2 + \frac{1}{2} \rho V_2^2$

$$\longrightarrow p_1 - p_2 = \frac{1}{2} \rho V_2^2 \left[ 1 - \left( \frac{A_2}{A_1} \right)^2 \right]$$

- Continuity  $\rho A_1 V_1 = \rho A_2 V_2$

## Incompressible Injector Eq. (3)

- Solve for  $V_2$

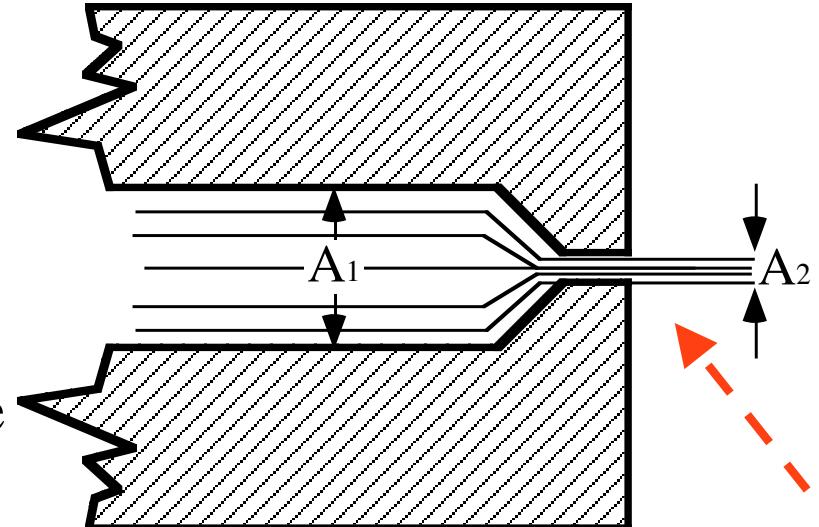
$$V_2 = \frac{1}{\left[1 - \left(\frac{A_2}{A_1}\right)^2\right]^{\frac{1}{2}}} \sqrt{2 \left( \frac{p_1 - p_2}{\rho} \right)}$$

- Friction effects in orifice will cause

$$V_{2_{actual}} < V_{2_{ideal}} \rightarrow V_{2_{actual}} \equiv C_v V_{2_{ideal}} \rightarrow$$

$$V_{2_{actual}} \equiv \frac{C_v}{\left[1 - \left(\frac{A_2}{A_1}\right)^2\right]^{\frac{1}{2}}} \sqrt{2 \left( \frac{p_1 - p_2}{\rho} \right)}$$

**C<sub>v</sub> -->  
“velocity coefficient”**

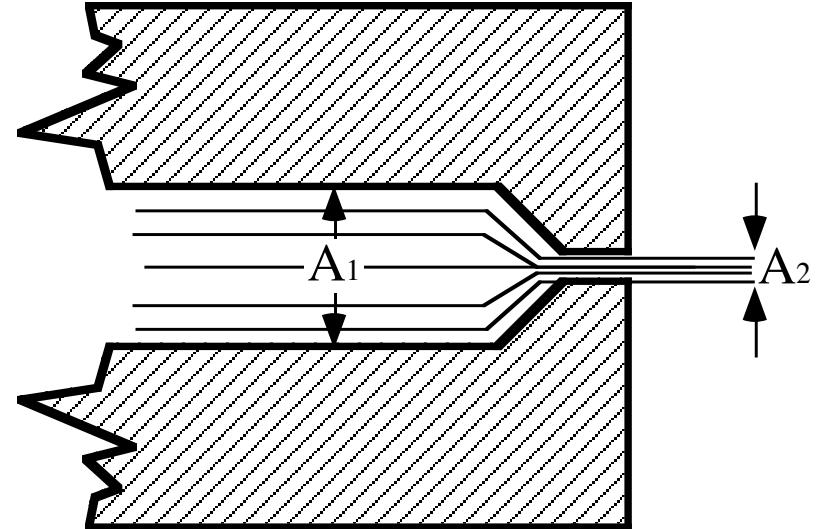


- Define “Discharge Coefficient”

$$C_d \equiv \frac{C_v}{\left[1 - \left(\frac{A_2}{A_1}\right)^2\right]^{\frac{1}{2}}}$$

# Incompressible Injector Eq. (4)

$$V_{2_{actual}} = C_d \sqrt{2 \left( \frac{p_1 - p_2}{\rho} \right)}$$

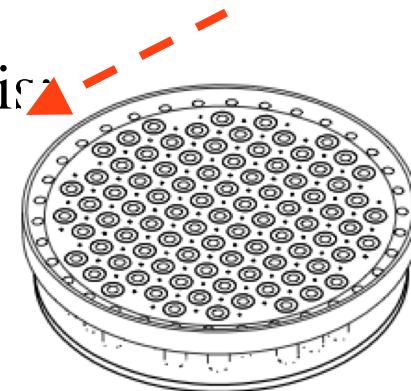


- Define Volumetric Flow as

$$Q_v = A_2 V_{2_{actual}} = A_2 C_d \sqrt{2 \left( \frac{p_1 - p_2}{\rho} \right)}$$

- Finally Incompressible Massflow Equation is

$$\dot{m} = \rho Q_v = A_2 C_d \sqrt{2 \rho (p_1 - p_2)}$$



# Compressible Injector Equation

- Some Common Propellants are in Gaseous Form
- Compressible Fuel/Oxidizer Examples:

*Gaseous Hydrogen, Methane, Ethane, Gaseous oxygen (GOX)*

*Accurate Injector Model requires Modeling of Flow Compressibility Effects*

$$\frac{\dot{m}}{A} = \sqrt{\frac{\gamma}{R_g}} \frac{p_0}{\sqrt{T_0}} \frac{M}{\left[1 + \frac{(\gamma - 1)}{2} M^2\right]^{\frac{\gamma+1}{2(\gamma-1)}}}$$

**General Massflow Equation  
for Compressible Flow**

$$\frac{P_{out}}{P_{in}} = \frac{1}{\left[1 + \frac{\gamma - 1}{2} M^2\right]^{\frac{\gamma}{\gamma-1}}} \rightarrow M = \sqrt{\frac{2}{\gamma - 1} \left[ \left( \frac{P_{in}}{P_{out}} \right)^{\frac{\gamma-1}{\gamma}} - 1 \right]}$$

**General  
Pressure Ratio  
Equation for  
Compressible  
Flow**

- Substitute and collect terms

## Compressible Injector Equation (2)

$$\dot{m} = A \sqrt{\frac{\gamma}{R_g T_0}} \cdot \frac{P_{in}}{\left(\frac{P_{in}}{P_{out}}\right)^{\frac{\gamma+1}{2\gamma}}} \sqrt{\frac{2}{\gamma-1} \left[ \left( \frac{P_{in}}{P_{out}} \right)^{\frac{\gamma-1}{\gamma}} - 1 \right]} = AP_{in} \sqrt{\frac{\gamma}{R_g T_0} \frac{2}{\gamma-1} \left[ \left( \frac{P_{in}}{P_{out}} \right)^{\frac{\gamma-1}{\gamma}} \left( \frac{P_{in}}{P_{out}} \right)^{-\frac{\gamma+1}{\gamma}} - \left( \frac{P_{in}}{P_{out}} \right)^{-\frac{\gamma+1}{\gamma}} \right]}$$

*Simplify*

$$\dot{m} = A \sqrt{\frac{P_{in}}{R_g T_0} \frac{2\gamma}{\gamma-1}} P_{in} \left[ \left( \frac{P_{in}}{P_{out}} \right)^{\frac{-2}{\gamma}} - \left( \frac{P_{out}}{P_{in}} \right)^{\frac{\gamma+1}{\gamma}} \right] = A \sqrt{\frac{2\gamma}{\gamma-1} \rho_{in} P_{in}} \left[ \left( \frac{P_{out}}{P_{in}} \right)^{\frac{2}{\gamma}} - \left( \frac{P_{out}}{P_{in}} \right)^{\frac{\gamma+1}{\gamma}} \right]$$

*Allow for Non-isentropic pressure losses ( $C_d$ )*

$$\dot{m} = C_d A \sqrt{\frac{2\gamma}{\gamma-1} \rho_{in} P_{in}} \left[ \left( \frac{P_{out}}{P_{in}} \right)^{\frac{2}{\gamma}} - \left( \frac{P_{out}}{P_{in}} \right)^{\frac{\gamma+1}{\gamma}} \right]$$

# 1-D “Lossy” Mass Flow Equations, Collected

*Unchoked Flow*   $\left( \frac{P_{in}}{P_{out}} \right)_{critical} < \left( \frac{\gamma+1}{2} \right)^{\frac{\gamma}{\gamma-1}} \Rightarrow \begin{cases} 1.8929 \\ \gamma = 1.4 \end{cases}$

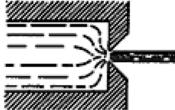
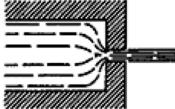
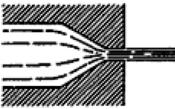
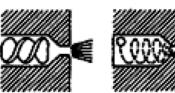
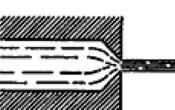
$$\dot{m} = C_d A \sqrt{\frac{2\gamma}{\gamma-1} \rho_{in} P_{in} \left[ \left( \frac{P_{out}}{P_{in}} \right)^{\frac{2}{\gamma}} - \left( \frac{P_{out}}{P_{in}} \right)^{\frac{\gamma+1}{\gamma}} \right]}$$

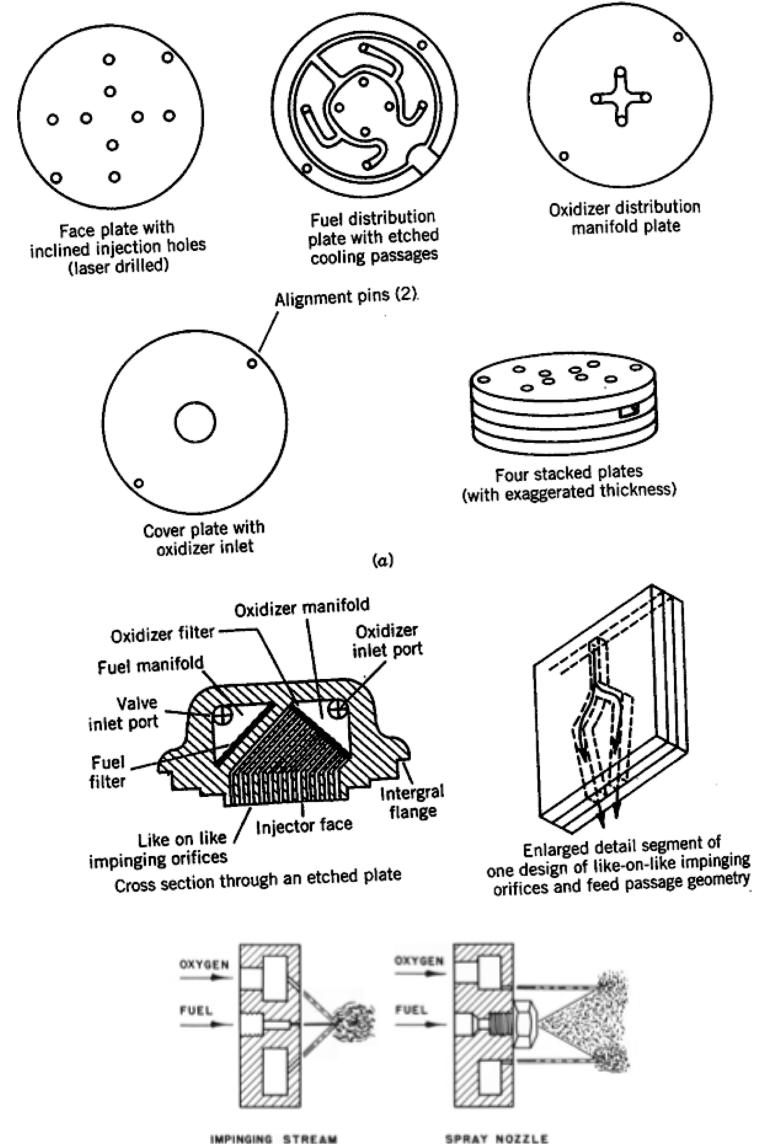
*Choked Flow:*   $\left( \frac{P_{in}}{P_{out}} \right)_{critical} \geq \left( \frac{\gamma+1}{2} \right)^{\frac{\gamma}{\gamma-1}} \Rightarrow \begin{cases} 1.8929 \\ \gamma = 1.4 \end{cases}$

$$\dot{m} = C_d A_{in}^* \cdot \sqrt{\gamma P_{in} \cdot \rho_{in} \left( \frac{2}{\gamma+1} \right)^{\frac{\gamma+1}{\gamma-1}}} \quad \text{dashed red arrow points to } \left( \frac{2}{\gamma+1} \right)^{\frac{\gamma+1}{\gamma-1}}$$

# Injector Port Design Options

TABLE 8-2. Injector Discharge Coefficients

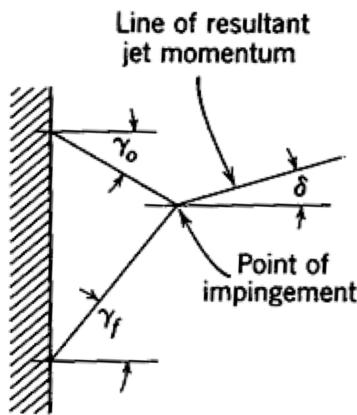
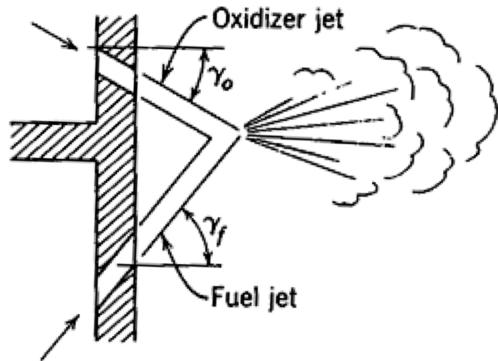
Orifice Type	Diagram	Diameter (mm)	Discharge Coefficient
Sharp-edged orifice		Above 2.5 Below 2.5	0.61 0.65 approx.
Short-tube with rounded entrance $L/D > 3.0$		1.00 1.57 1.00 (with $L/D \sim 1.0$ )	0.88 0.90 0.70
Short tube with conical entrance		0.50 1.00 1.57 2.54 3.18	0.7 0.82 0.76 0.84–0.80 0.84–0.78
Short tube with spiral effect		1.0–6.4	0.2–0.55
Sharp-edged cone		1.00 1.57	0.70–0.69 0.72



# Injector Design (cont'd))

- Mixing patterns

$$\dot{m}_{fuel} + \dot{m}_{ox} = A_{fuel} C_{d_f} \sqrt{2\rho_f (p_f - P_0)} + A_{ox} C_{d_{ox}} \sqrt{2\rho_{ox} (p_{ox} - P_0)}$$

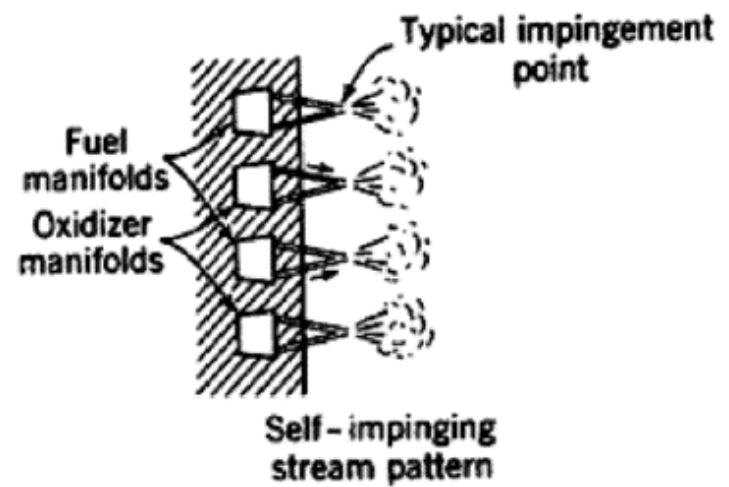
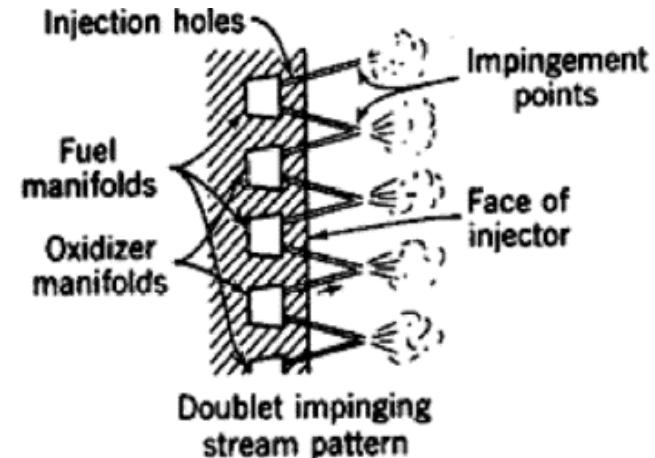


Angular relation of doublet impinging-stream injection pattern.

- Resultant momentum stream

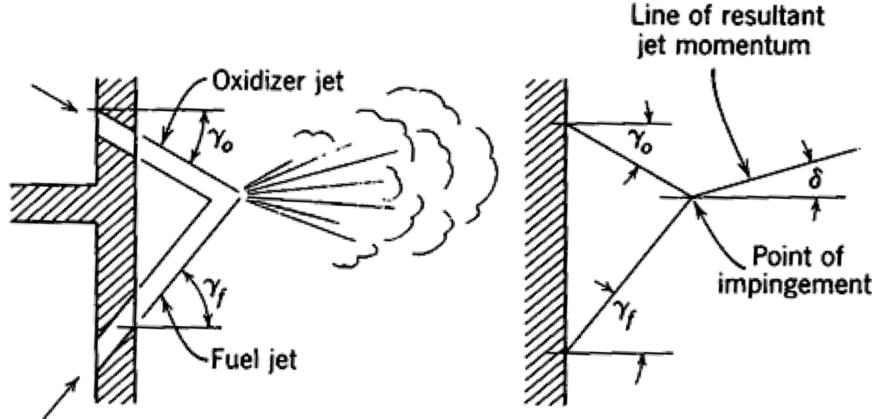
$$(\dot{m}_{ox} + \dot{m}_f) V_{ox+f} \cos(\delta) = \dot{m}_{ox} V_{ox} \cos(\gamma_{ox}) + \dot{m}_f V_f \cos(\gamma_f)$$

$$(\dot{m}_{ox} + \dot{m}_f) V_{ox+f} \sin(\delta) = \dot{m}_{ox} V_{ox} \sin(\gamma_{ox}) + \dot{m}_f V_f \sin(\gamma_f)$$



# Injector Design (cont'd)

- Mixing patterns



$$\dot{m}_{fuel} + \dot{m}_{ox} = A_{fuel} C_{d_f} \sqrt{2\rho_f (p_f - P_0)} + A_{ox} C_{d_{ox}} \sqrt{2\rho_{ox} (p_{ox} - P_0)}$$

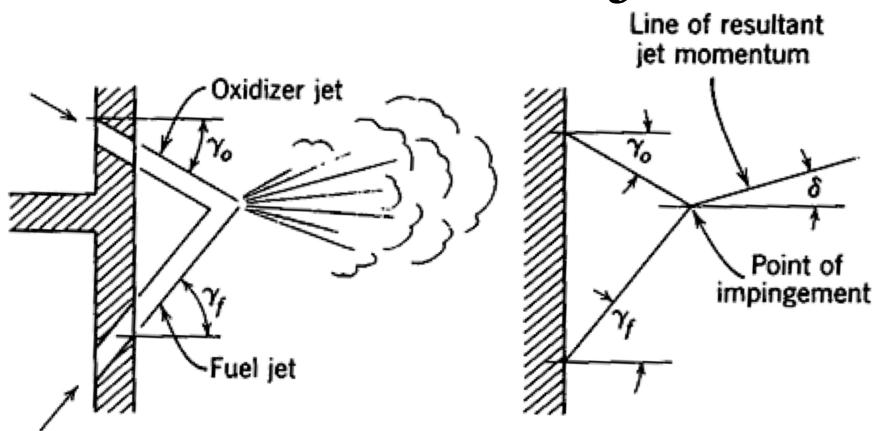
Angular relation of doublet impinging-stream injection pattern.

- Desirable for ...  $\delta \sim 0$  (*resultant momentum is directed axially*)

$$\rightarrow \delta \sim 0 \quad \dot{m}_{ox} V_{ox} \sin(\gamma_{ox}) = \dot{m}_f V_f \sin(\gamma_f)$$

# Injector Design (concluded)

- Mixing patterns



Angular relation of doublet impinging-stream injection pattern.

Condition for  $\delta \sim 0$

$$\dot{m}_{ox} V_{ox} \sin(\gamma_{ox}) = \dot{m}_f V_f \sin(\gamma_f)$$

- But ....

$$V_{actual} = C_d \sqrt{2 \left( \frac{p_1 - p_2}{\rho} \right)}$$

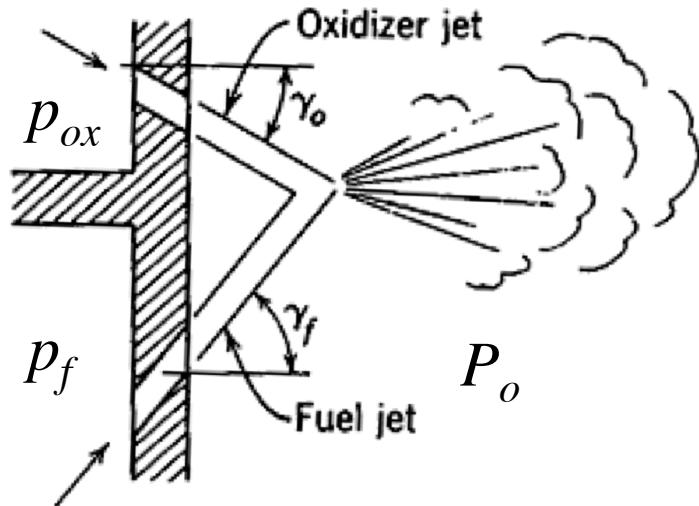
$$\dot{m} = A C_d \sqrt{2 \rho (p_1 - p_2)}$$

# Injector Design (concluded)

- Subbing in

$$A_{ox} C_{d_{ox}} \sqrt{2\rho_{ox}(p_{ox} - P_0)} \left[ C_{d_{ox}} \sqrt{2 \left( \frac{p_{ox} - P_0}{\rho_{ox}} \right)} \right] \sin(\gamma_{ox}) =$$

$$A_{fuel} C_{d_f} \sqrt{2\rho_f(p_f - P_0)} \left[ C_{d_f} \sqrt{2 \left( \frac{p_f - P_0}{\rho_f} \right)} \right] \sin(\gamma_f)$$



- Collecting terms

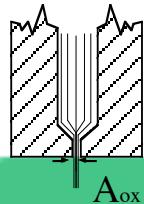
$$\rightarrow A_{ox} C_{d_{ox}}^2 (p_{ox} - P_0) \sin(\gamma_{ox}) = A_{fuel} C_{d_f}^2 (p_f - P_0) \sin(\gamma_f)$$

$$\frac{\sin(\gamma_{ox})}{\sin(\gamma_f)} = \frac{A_{fuel} C_{d_f}^2 (p_f - P_0)}{A_{ox} C_{d_{ox}}^2 (p_{ox} - P_0)}$$

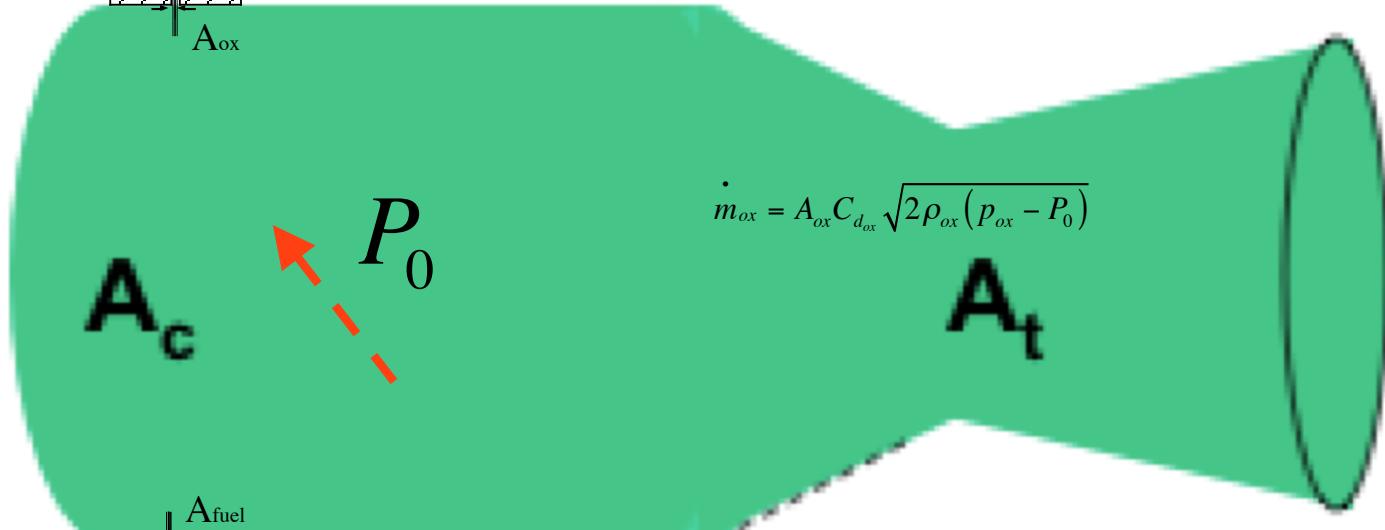
- Design criterion for injection angle

## Return to Liquid Rocket Example (cont'd)

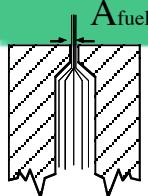
- Total Massflow for Incompressible propellants



$$\dot{m}_{fuel} + \dot{m}_{ox} = A_{fuel} C_{d_f} \sqrt{2\rho_f (p_f - P_0)} + A_{ox} C_{d_{ox}} \sqrt{2\rho_{ox} (p_{ox} - P_0)}$$



$$\dot{m}_{ox} = A_{ox} C_{d_{ox}} \sqrt{2\rho_{ox} (p_{ox} - P_0)}$$



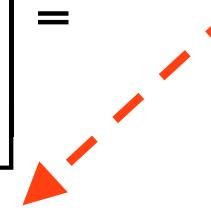
$$\left[ \begin{array}{l} A_{fuel} = \text{total area of fuel injector ports} \\ A_{ox} = \text{total area of oxydizer injector ports} \end{array} \right]$$

## Liquid Rocket Example (cont'd)

- Substitute into ODE for combustor pressure

$$\frac{\partial P_0}{\partial t} + P_0 \left[ \frac{A^*}{V_c} \sqrt{\gamma R_g T_0 \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{(\gamma-1)}}} \right] = \frac{R_g T_0}{V_c} \left[ A_{fuel} C_{d_f} \sqrt{2\rho_f (p_f - P_0)} + A_{ox} C_{d_{ox}} \sqrt{2\rho_{ox} (p_{ox} - P_0)} \right]$$

$$\frac{\partial P_0}{\partial t} + P_0 \left[ \frac{A^*}{V_c} \sqrt{\gamma R_g T_0 \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{(\gamma-1)}}} \right] =$$



$$\frac{R_g T_0}{V_c} \left[ A_{fuel} C_{d_f} \sqrt{2\rho_f (p_f - P_0)} + A_{ox} C_{d_{ox}} \sqrt{2\rho_{ox} (p_{ox} - P_0)} \right]$$

Incompressible Fluids

## Liquid Rocket Example (cont'd)

- Define **O/F** or “*mixture ratio*” of the propellants

$$\frac{\text{O/F}}{M_R} = \frac{\dot{m}_{ox}}{\dot{m}_{fuel}} = \frac{A_{ox} C_{d_{ox}} \sqrt{2\rho_{ox}(p_{ox} - P_0)}}{A_{fuel} C_{d_f} \sqrt{2\rho_f(p_f - P_0)}} \rightarrow \dot{m}_{ox} = M_R \dot{m}_{fuel}$$

- Liquid Rocket Model O.D.E.

$$\frac{\partial P_0}{\partial t} = \frac{R_g T_0}{V_c} [1 + M_R] A_{fuel} C_{d_f} \sqrt{2\rho_f(p_f - P_0)} - P_0 \left[ \frac{A^*}{V_c} \sqrt{\gamma R_g T_0 \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma + 1}{\gamma - 1}}} \right]$$

$$\frac{\partial P_0}{\partial t} = F \left[ P_0, \left\{ R_g, T_0, \gamma \right\}, \left\{ V_c, A^* \right\}, \left\{ A_{fuel}, C_{d_f}, \rho_f, p_f \right\}, \left\{ A_{ox}, C_{d_{ox}}, \rho_{ox}, p_{ox} \right\} \right]$$

↑                      ↑                      ↑

combustion products    chamber geometry    fuel injector geometry/pressure    oxydizer injector geometry/pressure

## Liquid Rocket Example (cont'd)

- Steady State behavior  $\frac{\partial P_0}{\partial t} = 0$

$$\frac{R_g T_0}{V_c} [1 + M_R] A_{fuel} C_{d_f} \sqrt{2 \rho_f (p_f - P_0)} = P_0 \left[ \frac{A^*}{V_c} \sqrt{\gamma R_g T_0 \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{(\gamma-1)}}} \right]$$

- Squaring both sides and Collecting terms

$$P_0^2 \left[ \left( \frac{A^*}{[1 + M_R] A_{fuel} C_{d_f}} \right)^2 \left( \frac{\gamma}{2 \rho_f R_g T_0} \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{(\gamma-1)}} \right) \right] + P_0 - p_f = 0$$


## Liquid Rocket Example (cont'd)

- Solve for  $P_0$

$$P_{0_{ss}} = \frac{-1 + \sqrt{1 + 4 p_f \left( \frac{A^*}{[1 + M_R] A_{fuel} C_{d_f}} \right)^2 \left( \frac{\gamma}{2 \rho_f R_g T_0} \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{(\gamma-1)}} \right)}}{2 \left( \frac{A^*}{[1 + M_R] A_{fuel} C_{d_f}} \right)^2 \left( \frac{\gamma}{2 \rho_f R_g T_0} \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{(\gamma-1)}} \right)}$$

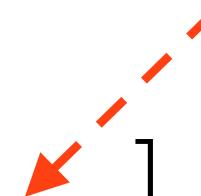

$$M_R = \frac{\dot{m}_{ox}}{\dot{m}_{fuel}} = \frac{A_{ox} C_{d_{ox}} \sqrt{2 \rho_{ox} (p_{ox} - P_0)}}{A_{fuel} C_{d_f} \sqrt{2 \rho_f (p_f - P_0)}} \rightarrow \dot{m}_{ox} = M_R \dot{m}_{fuel}$$

## Liquid Rocket Example (cont'd)

- “Tail-Off” Pressure

$$\left[ \dot{m}_{fuel} + \dot{m}_{ox} \right] = 0$$

$$\frac{\partial P_0}{\partial t} + P_0 \left[ \frac{A^*}{V_c} \sqrt{\gamma R_g T_0 \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{(\gamma-1)}}} \right] = 0$$

$$P_0(t)_{tail\ off} = P_{0\ ss} \left[ e^{-\frac{A^*}{V_c} \sqrt{\gamma R_g T_0 \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{(\gamma-1)}}} (t - t_{burnout})} \right]$$


## Liquid Rocket Example (cont'd)

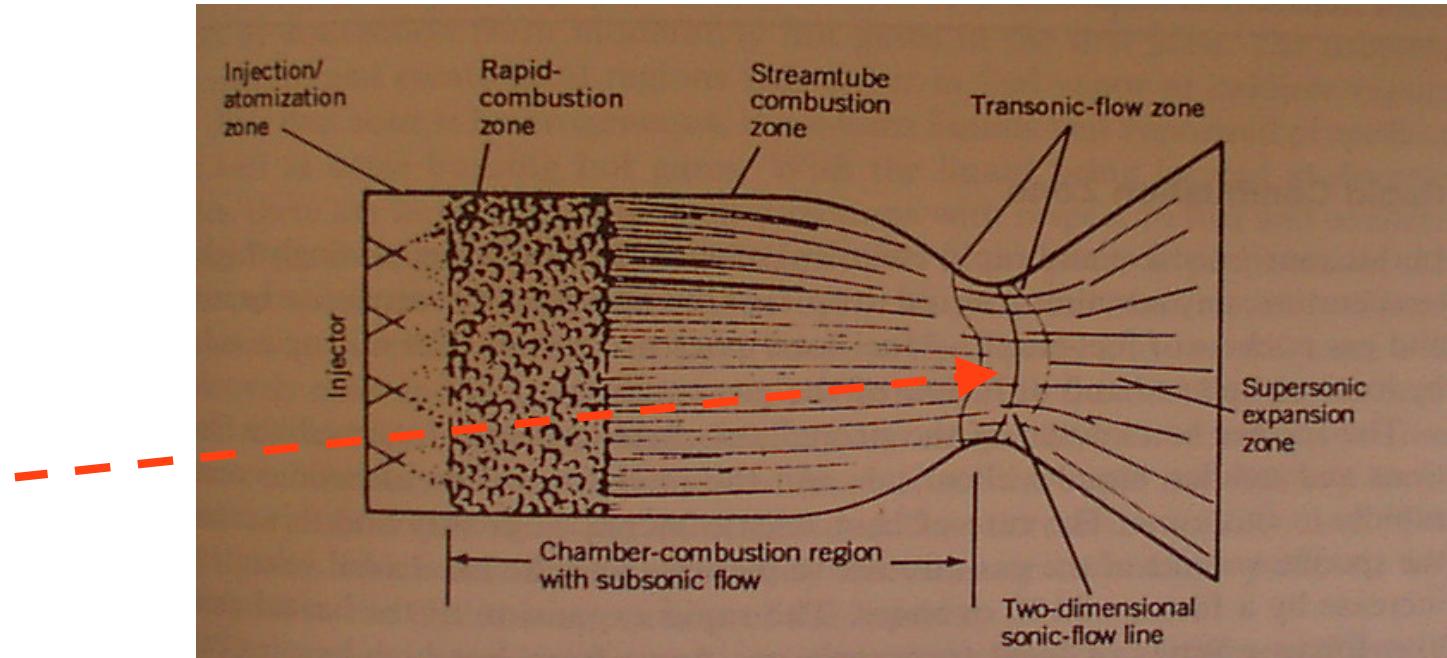
- Combustor “Response Time”

$$P_0(t)_{\substack{\text{tail} \\ \text{off}}} = P_{0_{ss}} \cdot e^{-\frac{A^*}{V_c} \cdot \left( \sqrt{\left( \gamma R_g T_0 \right) \cdot \left( \frac{2}{\gamma+1} \right)^{\frac{\gamma+1}{\gamma-1}}} \right) (t - t_{burnout})} = P_{0_{ss}} \cdot e^{-\left( \frac{t - t_{burnout}}{\tau_{combustor}} \right)}$$

$$\tau_{combustor} = \frac{V_c}{A^*} \cdot \frac{1}{\sqrt{\left( \gamma R_g T_0 \right) \cdot \left( \frac{2}{\gamma+1} \right)^{\frac{\gamma+1}{\gamma-1}}}}$$

"combustor  
response  
time"

# What happens in combustion chamber?



- The total combustion process, from injection of the reactants until completion of the chemical reactions and conversion of the products into hot gases, requires finite amounts of time and volume.
- How do we distribute the chamber volume length, diameter, convergence section to insure “good combustion?”

# What happens in combustion chamber? <sup>(2)</sup>

- Combustion chamber serves as an envelope to retain the propellants for a sufficient period to ensure complete mixing and combustion.
- Combustion “residence time” ... MUST MATCH propellant reaction rates ....

$$\tau_{combustor} = \frac{V_c}{A^*} \cdot \frac{1}{\sqrt{(\gamma R_g T_0) \cdot \left(\frac{2}{\gamma + 1}\right)^{\frac{\gamma+1}{\gamma-1}}}}$$

.... Match  $V_c/A^* = L^*$  (*Lstar*) to propellant reaction rates to  
Get best combustion characteristics

## What happens in combustion chamber? <sup>(3)</sup>

- Conventional method of establishing the  $L^*$  of a new thrust chamber design largely relies on past experience with similar propellants and engine size.
- Under a given set of operating conditions, value of the minimum required  $L^*$  evaluated by actual firings of thrust chambers.
- With throat area and minimum required  $L^*$  established, the appropriate chamber volume distribution
- $L^*$  is significantly greater than the linear length between injector face and throat plane.

# Characteristic Length parameter, $L^*$

## Characteristic length, $L^*$

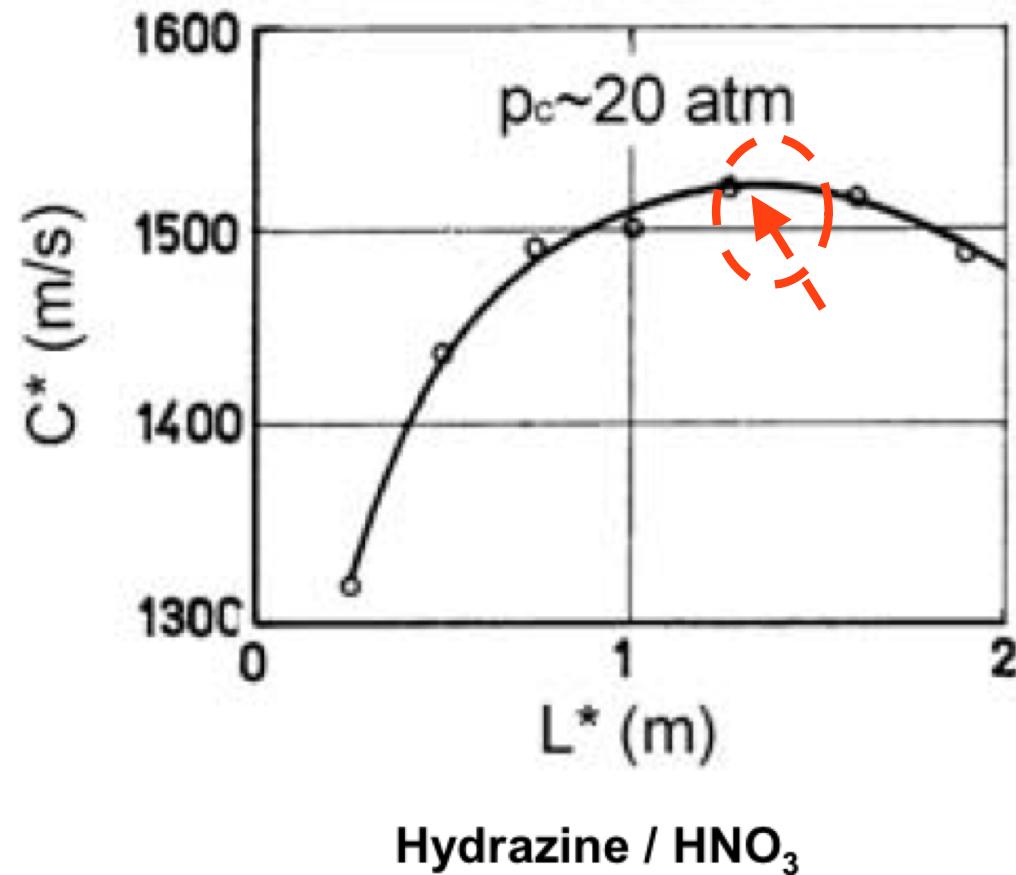
- Optimum  $L_c$  must be determined experimentally
- How to scale  $L_c$  from one engine size to another?
- Characteristic length,  $L^*$
- Value of  $L^*$  tabulated for different propellants

$$L^* = \frac{V_c}{A_t}$$

<i>Propellant</i>	$L^* \text{ (m)}$
LOX-kerosene	1.5-2.5
LOX-ethanol	2.5-3
HNO <sub>3</sub> -UDMH	1.5-2

Key parameter for combustion stability

$L^*$  has big effect on performance  
**Characteristic length,  $L^*$**



# Characteristic Length $L^*$ for various propellants

Ranges of Combustor Characteristic Length

Propellants	Characteristic Length $L^*$	
	Low(m)	High(m)
Liquid fluorine / hydrazine	0.61	0.71
Liquid fluorine / gaseous H <sub>2</sub>	0.56	0.66
Liquid fluorine / liquid H <sub>2</sub>	0.64	0.76
Nitric acid/hydrazine	0.76	0.89
N <sub>2</sub> O <sub>4</sub> / hydrazine	0.60	0.89
Liquid O <sub>2</sub> / ammonia	0.76	1.02
Liquid O <sub>2</sub> / gaseous H <sub>2</sub>	0.56	0.71
Liquid O <sub>2</sub> / liquid H <sub>2</sub>	0.76	1.02
Liquid O <sub>2</sub> / RP-1	1.02	1.27
H <sub>2</sub> O <sub>2</sub> / RP-1 (incl. catalyst)	1.52	1.78

- \*Space Propulsion Analysis
- and Design [Ronald
- W. Humble,
- Gary N. Henry, and
- Wiley J. Larson -
- McGraw-Hill, 1995]

$L^*$  is experimentally determined for given propellants and includes Effects of mixture ratio, massflow, etc. “High/Low” ranges

# Characteristic Length L\* for various propellants (2)

**Table 1:** Chamber Characteristic Length, L\*

Propellant Combination	L*, cm
Nitric acid/hydrazine-base fuel	76-89
Nitrogen tetroxide/hydrazine-base fuel	76-89
Hydrogen peroxide/RP-1 (including catalyst bed)	152-178
Liquid oxygen/RP-1	102-127
Liquid oxygen/ammonia	76-102
Liquid oxygen/liquid hydrogen (GH <sub>2</sub> injection)	56-71
Liquid oxygen/liquid hydrogen (LH <sub>2</sub> injection)	76-102
Liquid fluorine/liquid hydrogen (GH <sub>2</sub> injection)	56-66
Liquid fluorine/liquid hydrogen (LH <sub>2</sub> injection)	64-76
Liquid fluorine/hydrazine	61-71
Chlorine trifluoride/hydrazine-base fuel	51-89

# Combustion chamber scaling

- The *contraction ratio* is defined as the major cross-sectional area of the combustor divided by the throat area.
- Typically, large engines are constructed with a low contraction ratio and a comparatively long length
- Smaller chambers use a large contraction ratio with a shorter length, ... still providing sufficient  $L^*$  for adequate vaporization and combustion dwell-time.
- As a good place to start, the process of sizing a new combustion Chamber examines the dimensions of previously successful designs in the same size class and plotting such data in a rational manner.

## Combustion chamber scaling (2)

- The throat size of a new engine can be generated with a fair degree of confidence, so it makes sense to plot the data from historical sources in relation to throat diameter.
- Three geometrical shapes have been used in combustion chamber design – spherical, near-spherical, and cylindrical - with the cylindrical chamber being used most frequently in the United States.
- Compared to a cylindrical chamber of the same volume, a spherical or near-spherical chamber offers the advantage of less cooling surface and weight; however, the spherical chamber is more difficult to manufacture and has provided poorer performance in other respects.

# Combustion chamber scaling (3)

For a cylindrical chamber with a conical contraction section

The volume can be computed directly using geometric

Relationships . . . e.g for conical combustor

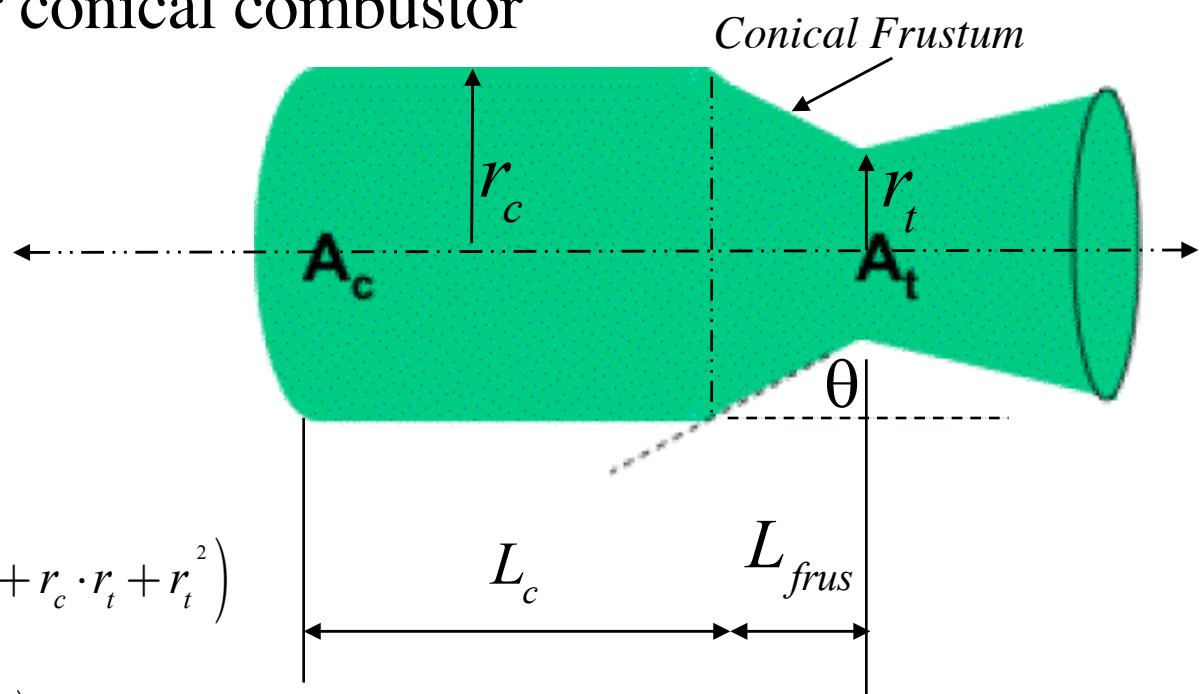
$$\varepsilon_c = \frac{A_c}{A_t} = \left( \frac{r_c}{r_t} \right)^2$$

$$L_{frus} = \frac{r_c - r_t}{\tan \theta}$$

$$V_c = V_{cyl} + V_{frus} = A_c \cdot L_c + \frac{\pi}{3} \cdot L_{frus} \cdot (r_c^2 + r_c \cdot r_t + r_t^2)$$

$$= A_t \cdot \varepsilon_c \cdot L_c + \frac{\pi}{3} \cdot \left( \frac{r_c - r_t}{\tan \theta} \right) \cdot (r_c^2 + r_c \cdot r_t + r_t^2) =$$

$$A_t \cdot \varepsilon_c \cdot L_c + \frac{1}{3} \frac{\pi \cdot r_t^2 \cdot r_t}{\tan \theta} \cdot \left( \left( \frac{r_c}{r_t} \right)^3 - 1 \right) = A_t \cdot \left[ \varepsilon_c \cdot L_c + \frac{1}{3} \cdot \sqrt{\frac{A_t}{\pi}} \cdot \frac{(\varepsilon_c \cdot \sqrt{\varepsilon_c} - 1)}{\tan \theta} \right]$$

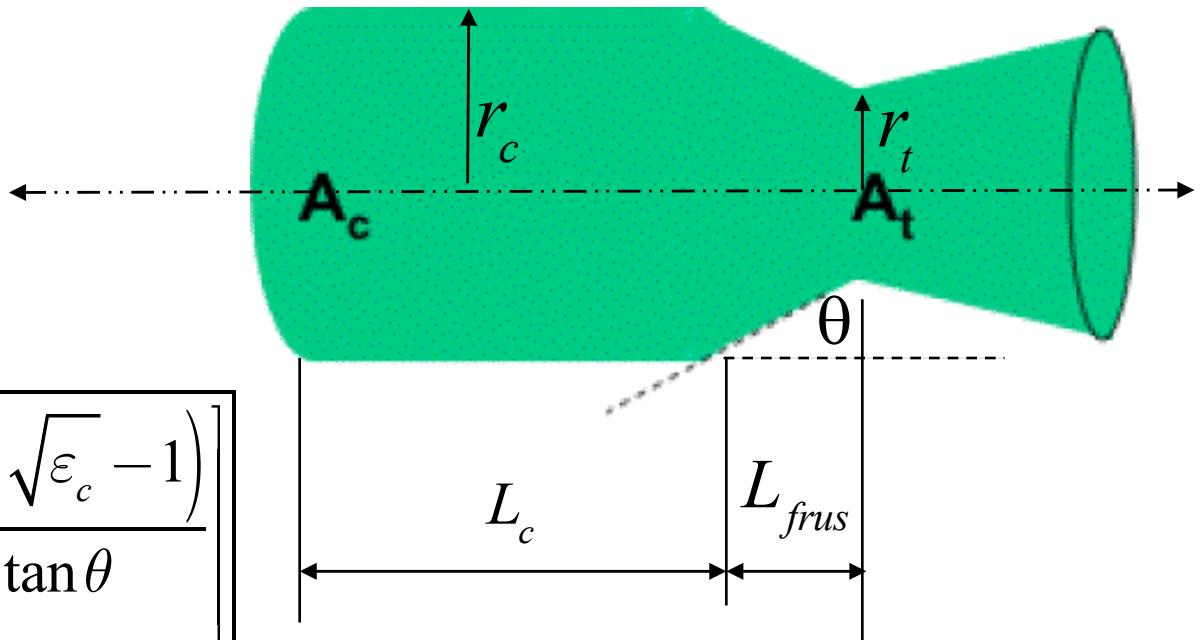


# Combustion chamber scaling (4)

For a cylindrical chamber with a conical contraction section  
The volume can be computed directly using geometric  
relationships

$$\varepsilon_c = \frac{A_c}{A_t}$$

$$V_c = A_t \cdot \left[ \varepsilon_c \cdot L_c + \frac{1}{3} \cdot \sqrt{\frac{A_t}{\pi}} \cdot \frac{(\varepsilon_c \cdot \sqrt{\varepsilon_c} - 1)}{\tan \theta} \right]$$



- What is the scaling relationship between the chamber length and the throat contraction ratio,  $\varepsilon_c$  ?

$L^*$  ... used to scale combustor dimensions

- Given throat geometry and propellants, calculate  $V_c, L_c, \varepsilon_c$
- Select  $L^*$  mid range from previous table
- Use empirical formula for contraction ratio based on throat diameter

$$\varepsilon_c = \frac{A_c}{A_t} \approx \frac{8.0}{D_t^{3/5}} + 1.25$$

**“rule of thumb”**

..... **D<sub>t</sub> is the throat diameter in centimeters.**  
**valid for**  $1/5 < A_c/A_t < 13.5$  ....  
**and**  $0.7\text{cm} < D_t < 200\text{cm}$

- The effective chamber length is derived from

$$V_c = A_t \cdot \left[ \varepsilon_c \cdot L_c + \frac{1}{3} \cdot \sqrt{\frac{A_t}{\pi}} \cdot \frac{(\varepsilon_c \cdot \sqrt{\varepsilon_c} - 1)}{\tan \theta} \right]$$

# SSME Combustor Example Revisited

- *Rules of Thumb*\*

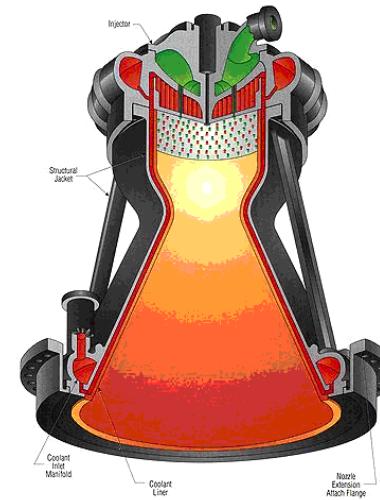
“Characteristic Length”

$$L^* = \frac{V_c}{A_t}$$

Combustor Cross sectional area

$$\varepsilon_c = \frac{A_c}{A_t} \approx \frac{8.0}{D_t^{3/5}} + 1.25 \quad D_t \sim \text{cm}$$

$$A_c = (8(27.1^{-0.6}) + 1.25) 0.05768 = 0.1358 \text{ m}^2$$



- SSME Throat Diameter, 27.1 cm

- Compute Throat Area

$$\frac{\left(\frac{27.1}{100}\right)^2 \pi}{4} = 0.05768 \text{ m}^2$$

\* Space Propulsion Analysis and Design [Ronald W. Humble, Gary N. Henry, and Wiley J. Larson - McGraw-Hill, 1995]

# SSME Combustor Example Revisited (cont'd)

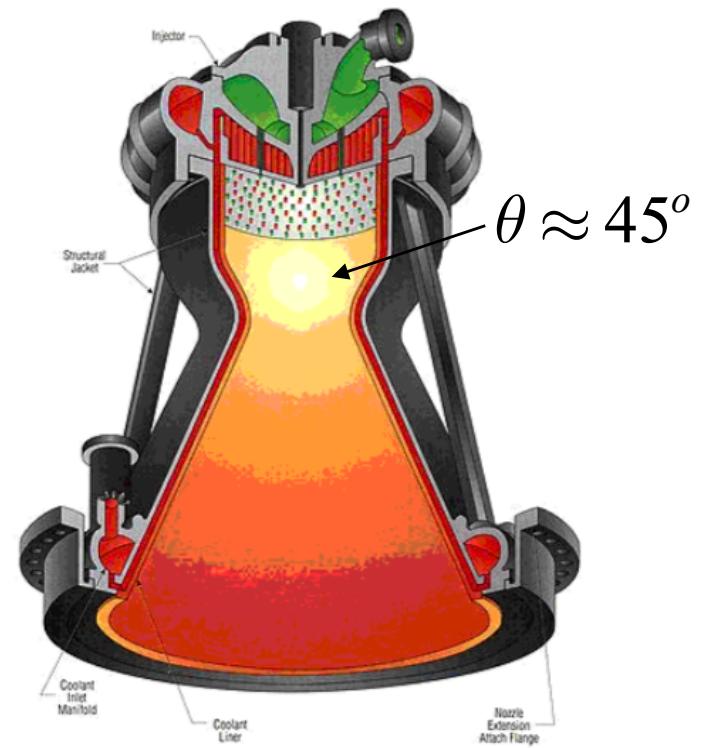
$$A_c = (8(27.1^{-0.6}) + 1.25) 0.05768 = 0.1358 \text{ m}^2$$

- Shuttle Combustor Diameter,

$$D_c = \left( \frac{0.1358 \cdot 4}{\pi} \right)^{0.5} 100 = 41.58 \text{ cm}$$

- Contraction ratio

$$\varepsilon_c = \frac{A_c}{A_t} = \frac{0.1358 \text{ m}^2}{0.05768 \text{ m}^2} = 2.354$$

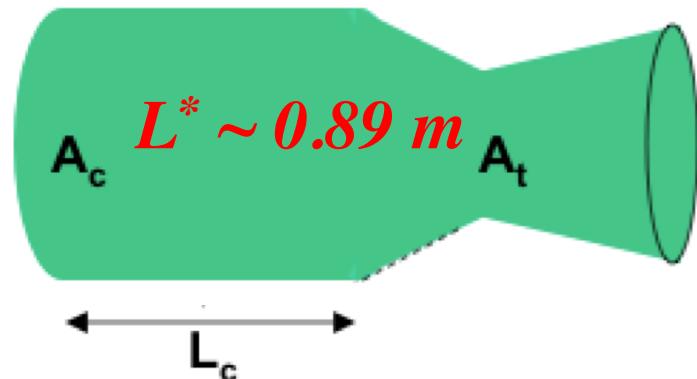


\* Space Propulsion Analysis and Design [Ronald W. Humble, Gary N. Henry, and Wiley J. Larson - McGraw-Hill, 1995]

# SSME Combustor Example Revisited (cont'd)

$$L^* = \frac{V_c}{A_t}$$

- Compute chamber volume based on “mid range” value for  $L^*$



Ranges of Combustor Characteristic Length \*

Propellants	Characteristic Length $L^*$	
	Low(m)	High(m)
Liquid O <sub>2</sub> / ammonia	0.76	1.02
Liquid O <sub>2</sub> / gaseous H <sub>2</sub>	0.56	0.71
Liquid O <sub>2</sub> / liquid H <sub>2</sub>	0.76	1.02
Liquid O <sub>2</sub> / RP-1	1.02	1.27
H <sub>2</sub> O <sub>2</sub> / RP-1 (incl. catalyst)	1.52	1.78

$$\begin{aligned} V_c &\simeq \\ &\left( \left( \frac{0.76 + 1.02}{2} \right) 0.05768 \right) \\ &= 0.05134 \text{ m}^3 \end{aligned}$$

\* Space Propulsion Analysis and Design [Ronald W. Humble, Gary N. Henry, and Wiley J. Larson - McGraw-Hill, 1995]

$L^*$  used to scale combustor dimensions (cont'd)

$$V_c = A_t \cdot \left[ \varepsilon_c \cdot L_c + \frac{1}{3} \cdot \sqrt{\frac{A_t}{\pi}} \cdot \frac{(\varepsilon_c \cdot \sqrt{\varepsilon_c} - 1)}{\tan \theta} \right]$$

$$\textcolor{red}{L^* \sim 0.89_m} \quad \theta \approx 45^\circ$$

$$\varepsilon_c = 2.345 \quad A_t = 576.8 \text{ cm}^2$$

Solve for  $L_c$

$$L_c = \frac{1}{\varepsilon_c} \left[ \frac{V_c}{A_t} - \frac{1}{3} \cdot \sqrt{\frac{A_t}{\pi}} \cdot \frac{(\varepsilon_c \cdot \sqrt{\varepsilon_c} - 1)}{\tan \theta} \right] \approx \frac{1}{\varepsilon_c} \left[ L^* - \frac{1}{3} \cdot \sqrt{\frac{A_t}{\pi}} \cdot \frac{(\varepsilon_c \cdot \sqrt{\varepsilon_c} - 1)}{\tan \theta} \right]$$

$$\frac{1}{2.354} \left\{ 0.89 \cdot 100 - \frac{1}{3} \left( \frac{576.8}{\pi} \right)^{0.5} \frac{(2.345 (2.345^{0.5}) - 1)}{\tan \left( \frac{\pi}{180} 45 \right)} \right\} = 32.84 \text{ cm}$$

# SSME Combustor Example Revisited (cont'd)

... Compare to “Effective” combustor length dimension..

$$V_c/A_c = L_{eff} = \frac{0.05134}{0.1358} 100 = 37.81 \text{ cm}$$

What is burner surface area?

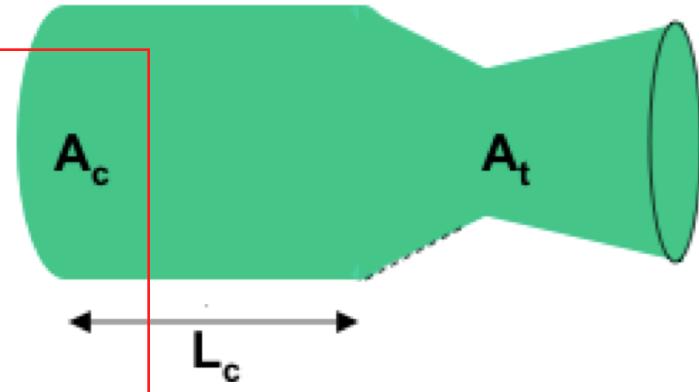
$$A_{surf} = 2 \cdot \pi \cdot r_c \cdot L_c + \pi \cdot (r_c + r_t) \cdot \sqrt{(r_c - r_t)^2 + L_{frus}^2} = \\ 2\pi 20.79 \cdot 32.84 + \pi (20.79 + 13.55) \left( (20.79 - 13.55)^2 + 7.24^2 \right)^{0.5} \\ = 5394.4 \text{ cm}^2$$

$$r_c = 20.79 \text{ cm}$$

$$r_t = 13.55 \text{ cm}$$

$$L_{frus} = \frac{r_c - r_t}{\tan \theta} = \frac{20.79 - 13.55}{1} = 7.24 \text{ cm}$$

$$L_c = 32.84 \text{ cm}$$



**Published number**

**0.54884 m<sup>2</sup>**

Pretty good estimate

$$L_c + L_{frus} = 32.84 \text{ cm} + 7.24 \text{ cm} = 40.08 \text{ cm}$$

# SSME Example

- Compute Nominal Steady State Operating Combustor Pressure



Properties of Propellant Products 2

Effective gamma	1.196
Effective MW	13.6
Idealized Flame Temperature, deg. K	3615

MAE 5540 - Propuls

$$P_{0_{ss}} = \frac{-1 + \sqrt{1 + 4p_f \left( \frac{A^*}{[1 + M_R] A_{fuel} C_{df}} \right)^2 \left( \frac{\gamma}{2\rho_f R_g T_0} \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{\gamma-1}} \right)}}{2 \left( \frac{A^*}{[1 + M_R] A_{fuel} C_{df}} \right)^2 \left( \frac{\gamma}{2\rho_f R_g T_0} \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{\gamma-1}} \right)}$$

Combustor Output data      Fuel Injector Properties      Oxydizer Injector Properties

Rg, J/kg Deg-K <b>611.3539</b>	Ox, Mdot, kg/sec <b>67.9644</b>	Injector Area, m^2 <b>0.002877</b>
Cp, J/kg Deg-K 2 <b>3730.51</b>	Fuel, Mdot, kg/sec <b>409.952</b>	Discharge Coefficient <b>0.81</b>
Cstar, Ideal, m/sec <b>2295.04</b>	Mixture, ratio <b>6.03186</b>	Fuel density, kg/M^3 <b>71</b>
Chamber /Nozzle geometry	Mdot, total kg/sec <b>477.916</b>	Injector Pressure, kPa <b>24950</b>
Volume, M^3 <b>0.05119</b>	Total Fuel mass, kg <b>35477.5</b>	Total Oxydizer mass, kg <b>213995</b>
Throat Area, M^2 <b>0.05785</b>	A/A*	
A/A* <b>77.5</b>		

## SSME Example (cont'd)

- Compute Nominal Steady State Operating Combustor Pressure

$$P_{0_{ss}} = \frac{-1 + \sqrt{1 + 4 p_f \left( \frac{A^*}{[1 + M_R] A_{fuel} C_{d_f}} \right)^2 \left( \frac{\gamma}{2 \rho_f R_g T_0} \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{(\gamma-1)}} \right)}}{2 \left( \frac{A^*}{[1 + M_R] A_{fuel} C_{d_f}} \right)^2 \left( \frac{\gamma}{2 \rho_f R_g T_0} \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{(\gamma-1)}} \right)} =$$

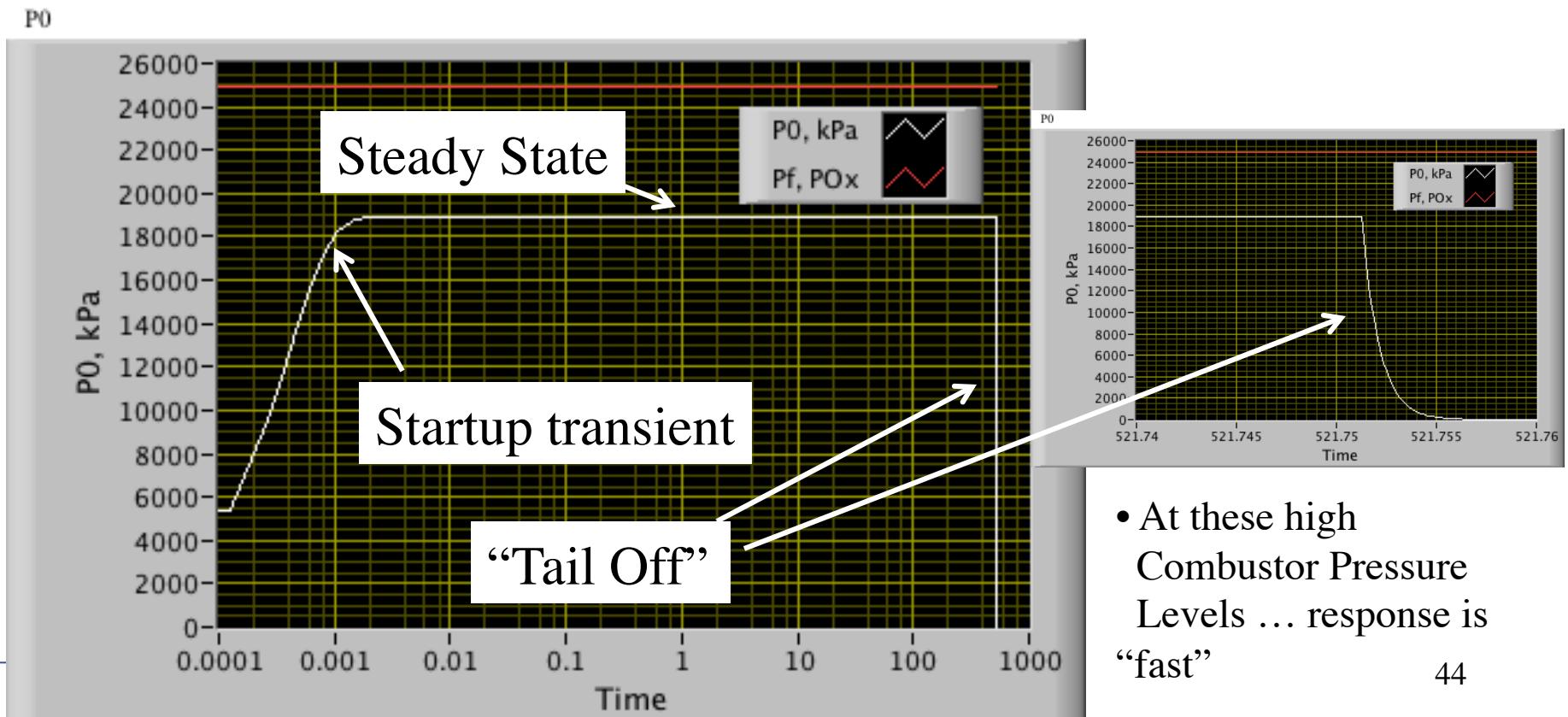
$$\frac{\left( 1 + \left( 4 \cdot 24950 \cdot 10^3 \left( \left( \frac{0.05785}{(1 + 6.03186)(0.002877 \cdot 0.81)} \right)^2 \left( \frac{1.196}{2 \cdot 71 \cdot 611.3539 \cdot 3615} \right) \left( \frac{2}{1.196 + 1} \right)^{\frac{(1.196 + 1)}{(1.196 - 1)}} \right) \right)^{0.5} - 1 \right)}{2 \left( \left( \frac{0.05785}{(1 + 6.03186)(0.002877 \cdot 0.81)} \right)^2 \left( \frac{1.196}{2 \cdot 71 \cdot 611.3539 \cdot 3615} \right) \left( \frac{2}{1.196 + 1} \right)^{\frac{(1.196 + 1)}{(1.196 - 1)}} \right) 1000}$$

$$= 18960.03 \text{ kPa} \dots \dots \dots !$$

... and we know from  
earlier (5.2) this is “correct”  
answer

## SSME: Look at Transient Response

$$\frac{\partial P_0}{\partial t} = \frac{R_g T_0}{V_c} [1 + M_R] A_{fuel} C_{df} \sqrt{2\rho_f (p_f - P_0)} - P_0 \left[ \frac{A^*}{V_c} \sqrt{\gamma R_g T_0} \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{\gamma-1}} \right]$$

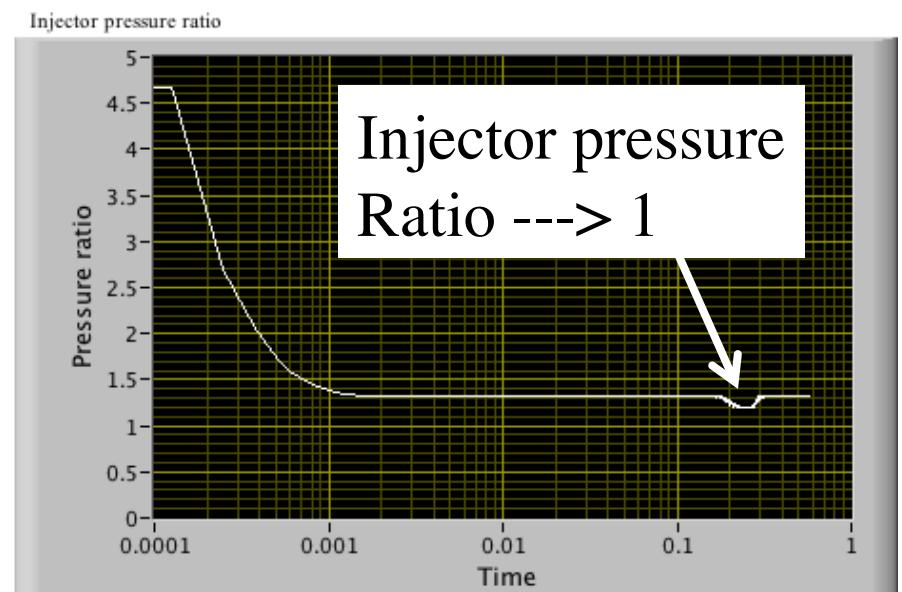
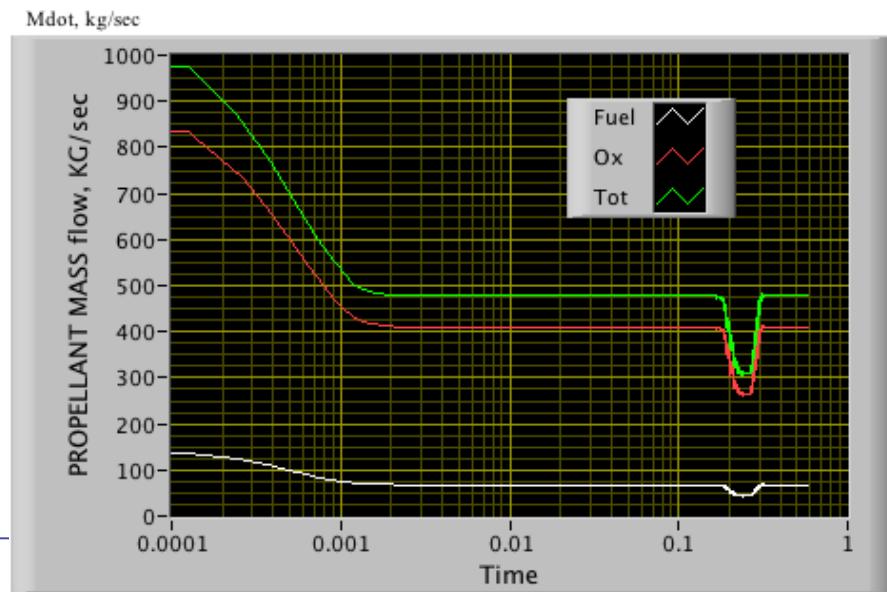
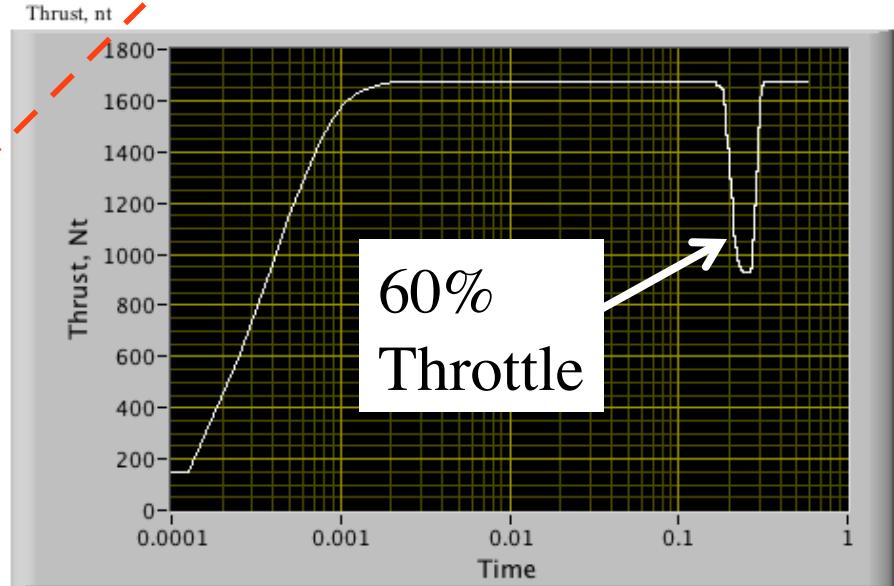
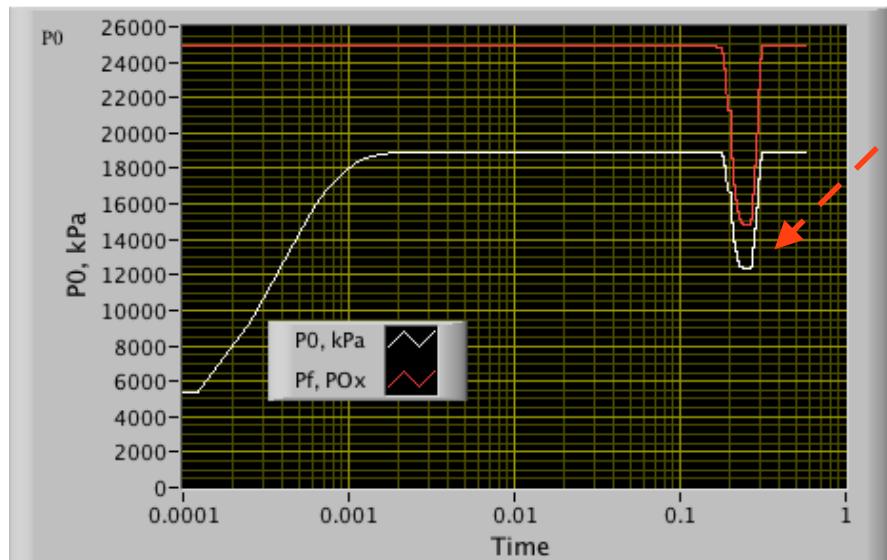


# A Mechanism for Throttling

$$\frac{\partial P_0}{\partial t} = -P_0 \left[ \frac{A^*}{V_c} \sqrt{\gamma R_g T_0 \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma+1}{\gamma-1}}} \right] + \\ \frac{R_g T_0}{V_c} \left[ A_{fuel} C_{d_f} \sqrt{2\rho_f \left( \left\{ \frac{T_h \%}{100} \times p_f \right\} - P_0 \right)} + A_{ox} C_{d_{ox}} \sqrt{2\rho_{ox} \left( \left\{ \frac{T_h \%}{100} \times p_{ox} \right\} - P_0 \right)} \right]$$

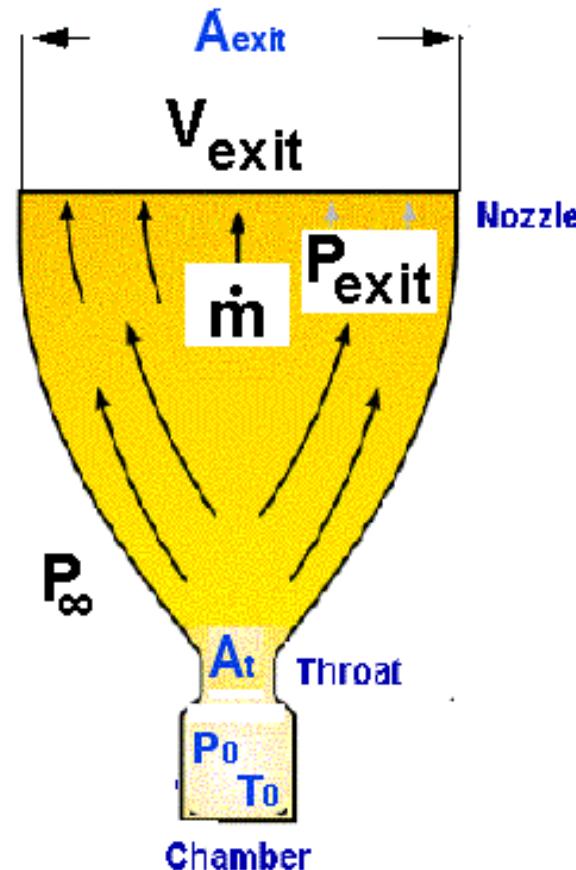
- By modulating the injector pressures, we can throttle the rocket engine

# A Mechanism for Throttling (cont'd)



# Deep-Throttle Rocket Engines

- In theory rocket engine motor can be throttled back until the throat is no longer sonic by reducing propellant Flow rate (*injector pressure*)
- Difficult problem in practice.



$$T_{\text{thrust}} = \dot{m}V_{\text{exit}} + A_{\text{exit}}(P_{\text{exit}} - P_{\infty})$$

# Deep-Throttle Rocket Engines

(cont'd)

- Essential for pressure drop across injector  
**> 25% of chamber pressure**

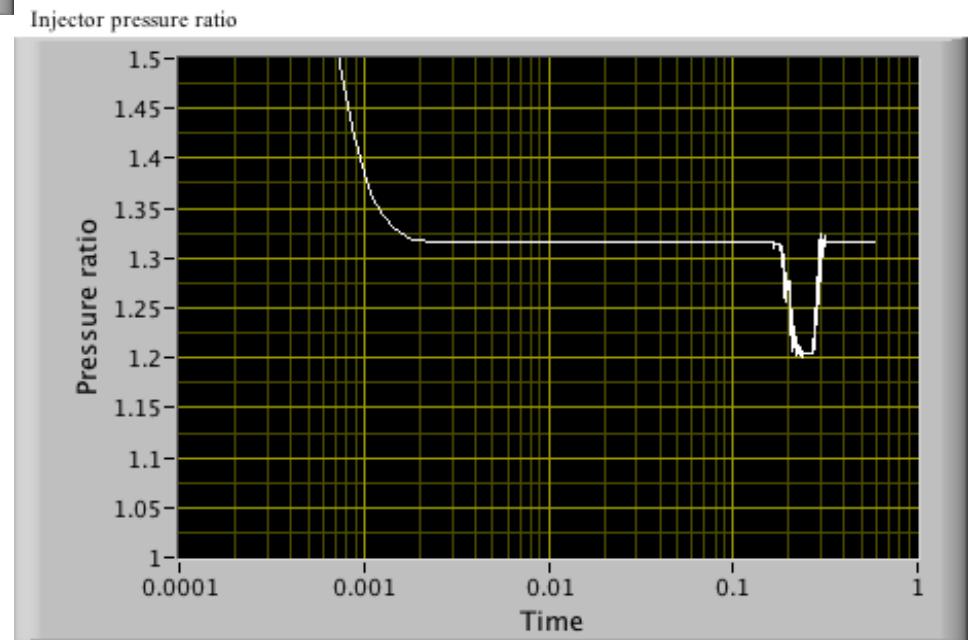
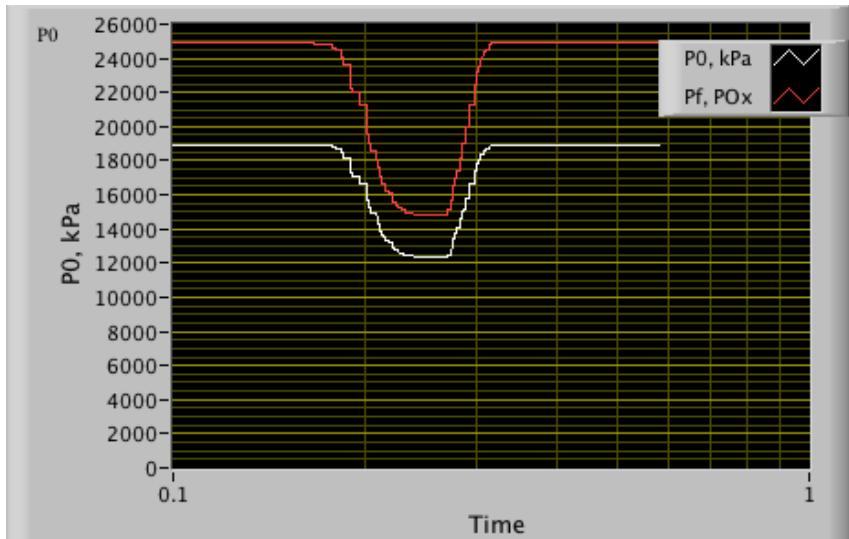
-- Pressure ratio insures propellant flow rates  
are independent of fluctuations in chamber pressure.

- Fixed geometry injectors

Reduction of Propellant flow rates causes injector pressure  
to drop faster than the chamber pressure  
... until injector pressure becomes so low that coupling between  
chamber and propellant feed system occurs  
... causing combustor instability (a.k.a explosion or flameout)

# Deep-Throttle Rocket Engines

(cont'd)



# Deep-Throttle Rocket Engines

(cont'd)

- Typically rocket motor with fixed injector geometry can be throttled down to 60-70% of nominal thrust without serious stability problems (SSME)
- **Highest Operational “turndown ratio” Engine**
  - Lunar Module descent engine, 10:1 turndown ratio
  - Even with a variable-geometry injector there were stability Problems, Thrust levels between 100% and 65% were never used because mixture ratio was so hard to properly control.
- **Obviously this is a big Challenge**
  - Current State-of-The-Art
  - “Pintle-injectors”

# Deep-Throttle Rocket Engines

(cont'd)

- **Northrop Grumman (TRW) TR-106**
  - Pintle-injection (similar to LEM descent engine)
- **Low Cost Pintle Engine (LCPE)**

The key element of the LCPE's design is its single element coaxial pintle injector, used to introduce propellants into the combustion chamber.

- Moveable pintle injector attributes include deep throttle capability, 10:1 turndown ratio

TRW has tested more than 50 different pintle injector engines, using more than 25 different propellant combinations with complete combustion stability

Range in size from the 100-pound thrust liquid apogee engine used on NASA's Chandra X-ray Observatory to the 10,0-pound thrust Delta and LMDE engines.

- Small Scale Test Version?



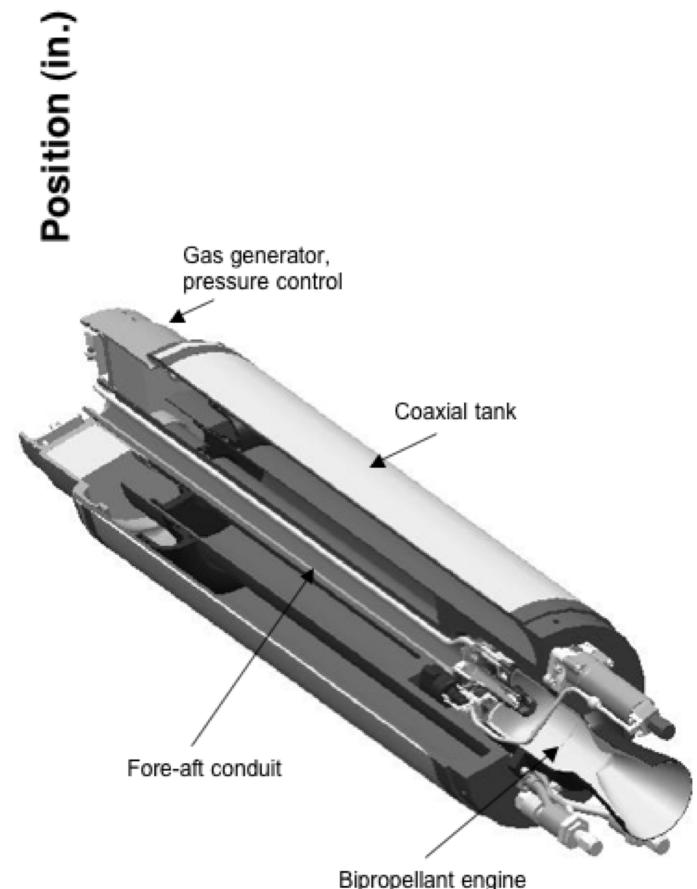
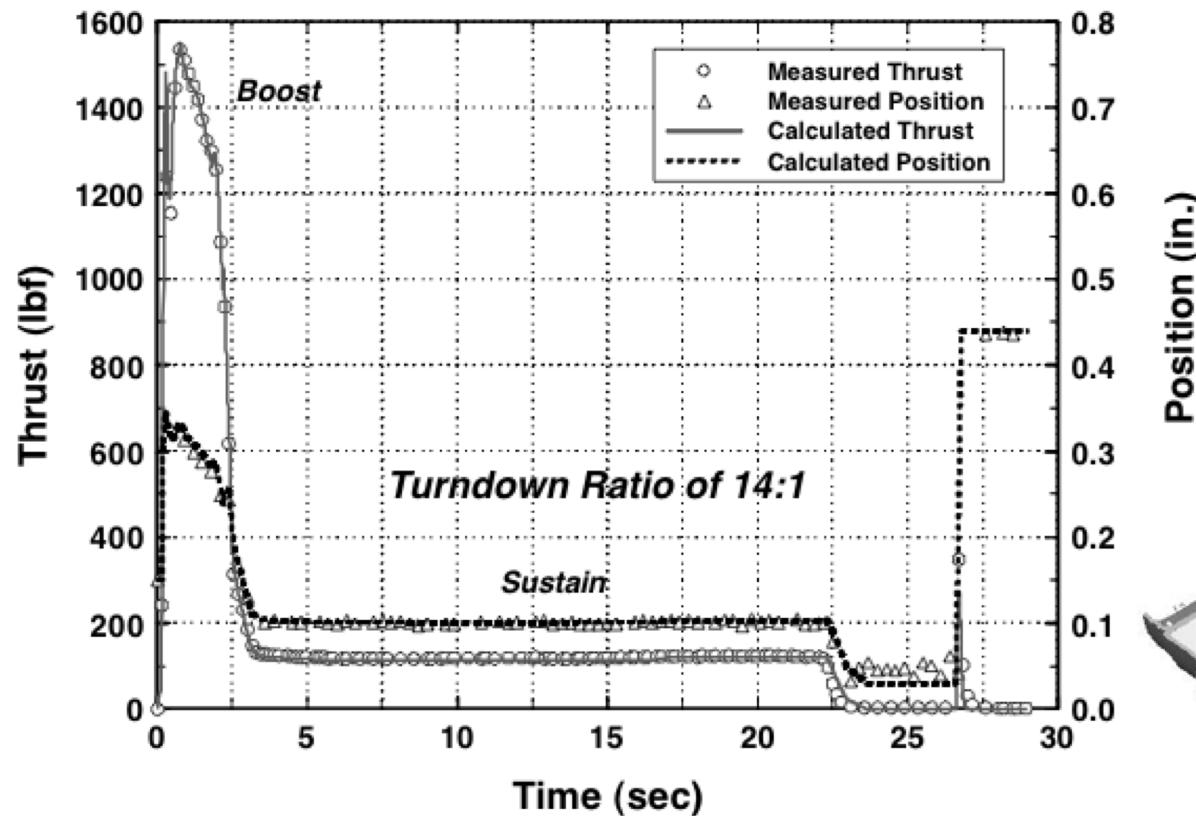
(LCPE)



# Deep-Throttle Rocket Engines

(cont'd)

- Small Scale Test Version?



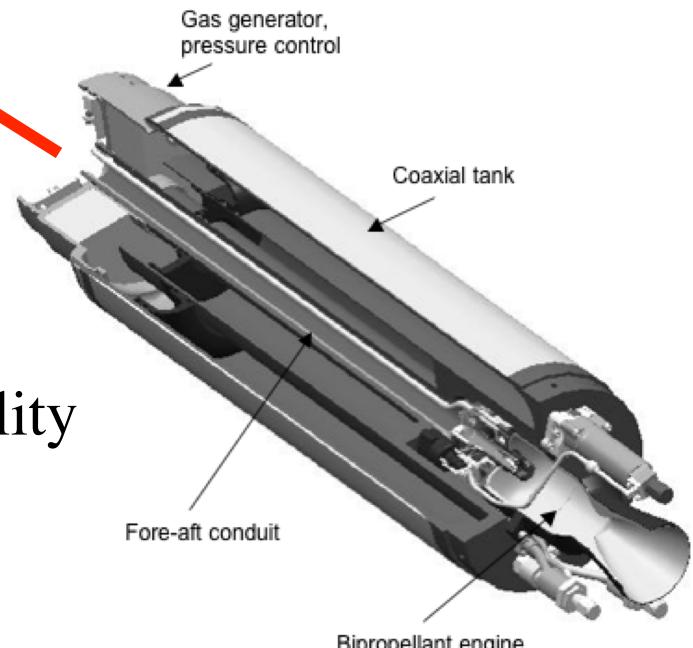
# Deep-Throttle Rocket Engines

(cont'd)

- Pintle Injectors

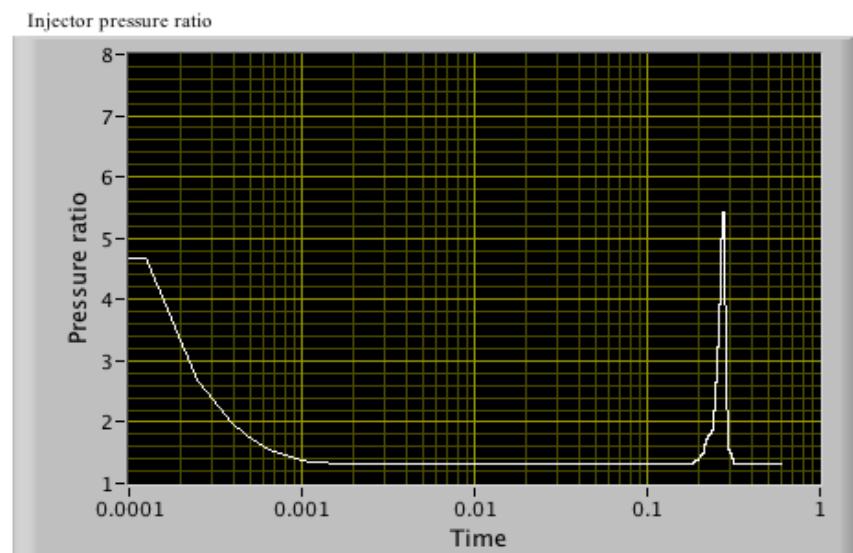
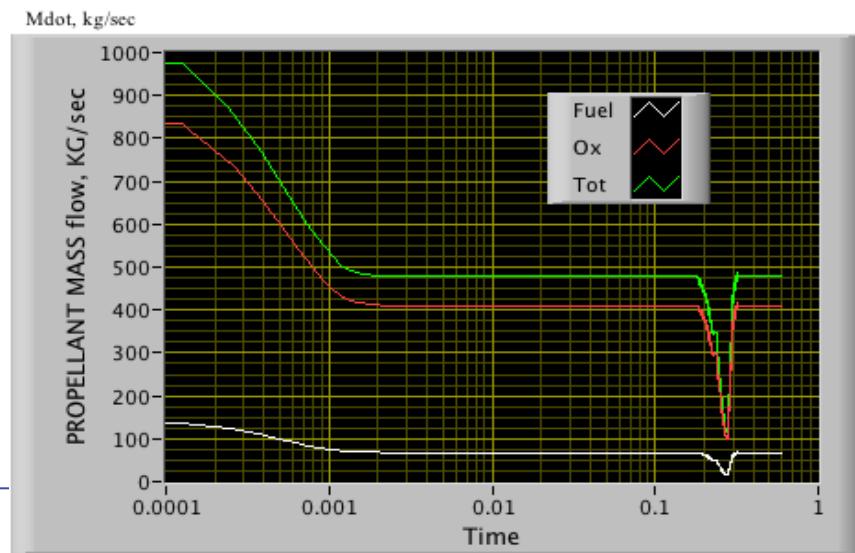
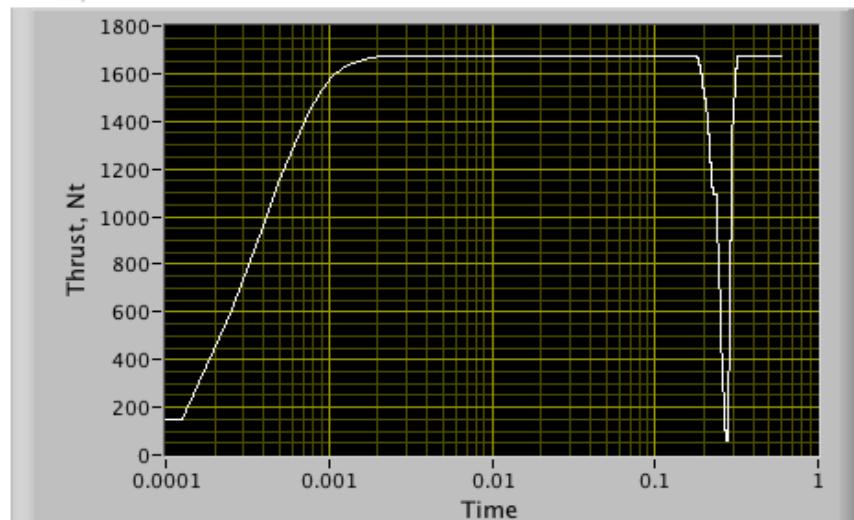
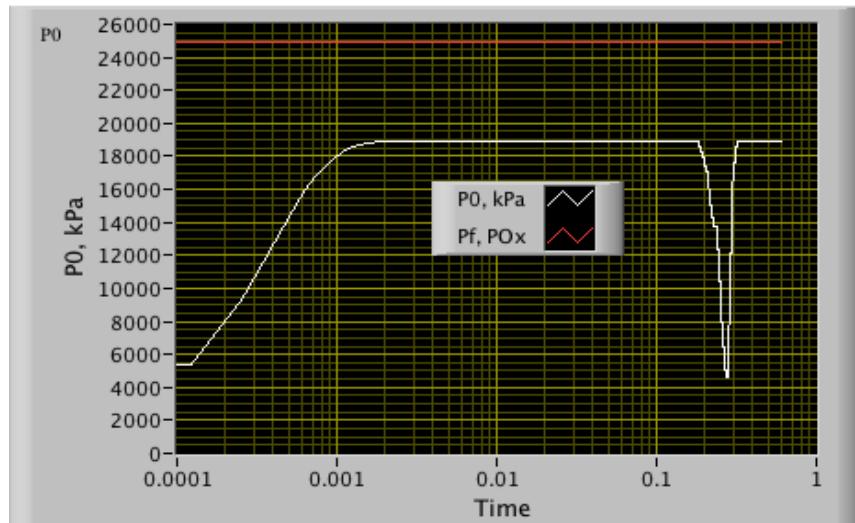
$$\frac{\partial P_0}{\partial t} = \frac{R_g T_0}{V_c} [1 + M_R] \underline{A_{fuel} C_{d_f}} \sqrt{2 \rho_f (p_f - P_0)} - P_0 \left[ \frac{A^*}{V_c} \sqrt{\gamma R_g T_0} \left( \frac{2}{\gamma + 1} \right)^{\frac{\gamma + 1}{\gamma - 1}} \right]$$

- By moving the pintle shaft  
Propellant mass flow is modulated  
by changing the effective injector area
- Injection pressure remains relatively  
constant ... reducing combustion instability  
problem



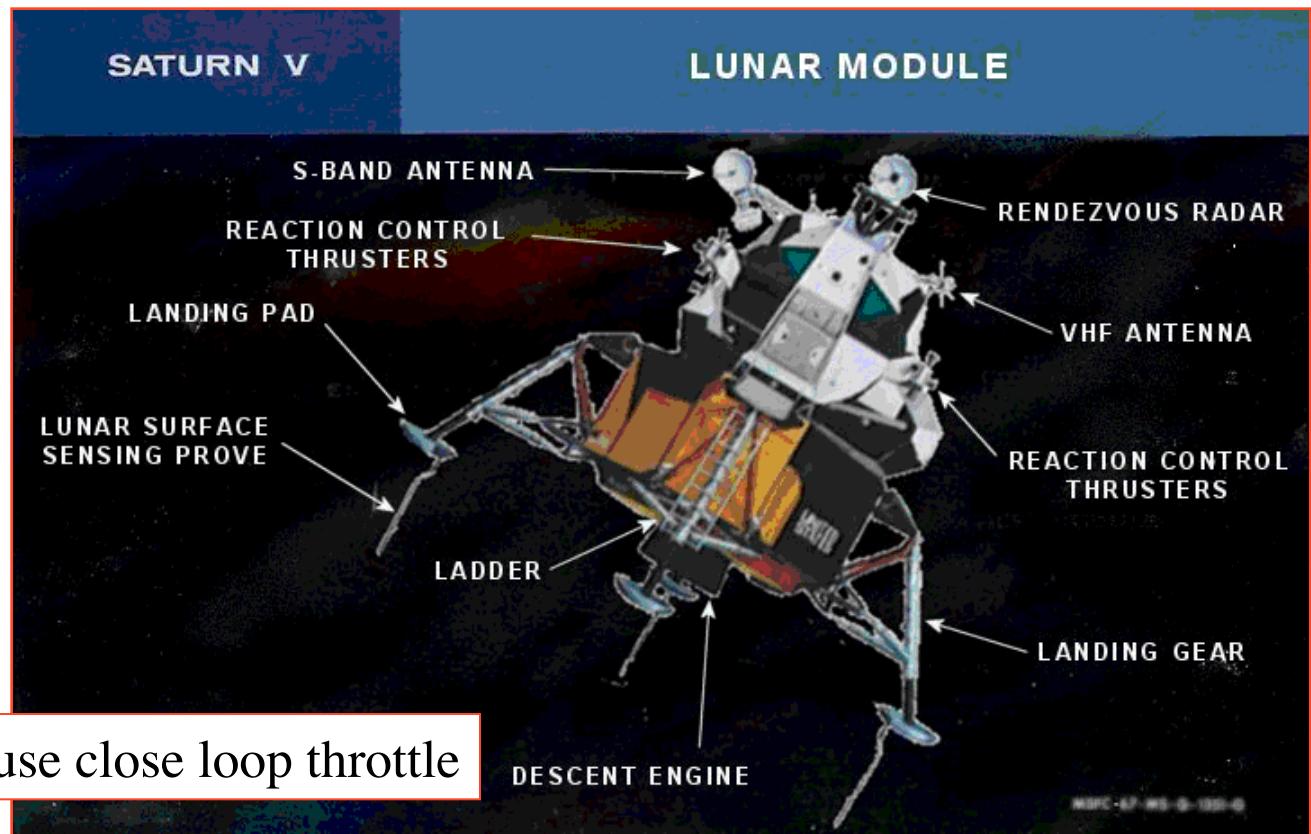
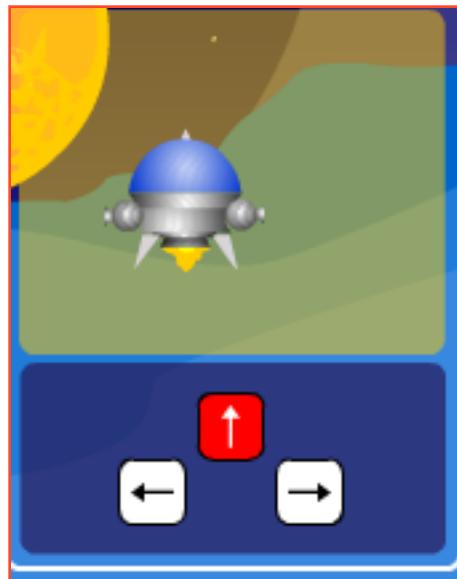
# Deep-Throttle Rocket Engines

- SSME Throttle Simulation using Pintle Injector (Area Modulation)



# Applications of Deep Throttling (I)

- Lunar Lander ... Throttling key element in ability to perform precision landing ... Apollo --> Open loop throttle under control of mission commander



New LM will likely use close loop throttle

Homework **5**, Assigned **Monday March 15** , Due Monday  
**Wednesday March 24** 8 pts Total

- 1) 1 points, Propellant Performance
- 2) 2 points, Combustor Cooling
- 3) 1 points, Combustor Geometry
- 4) 2 points, Combustor Pressure, Injector Design
- 5) 2 points, Optimal Performance

## 1) Propellant Performance (1 Point)

- A Bi-Propellant Rocket Engine Burns LOX/LH<sub>2</sub> ... with an O/F ratio of 5.3333

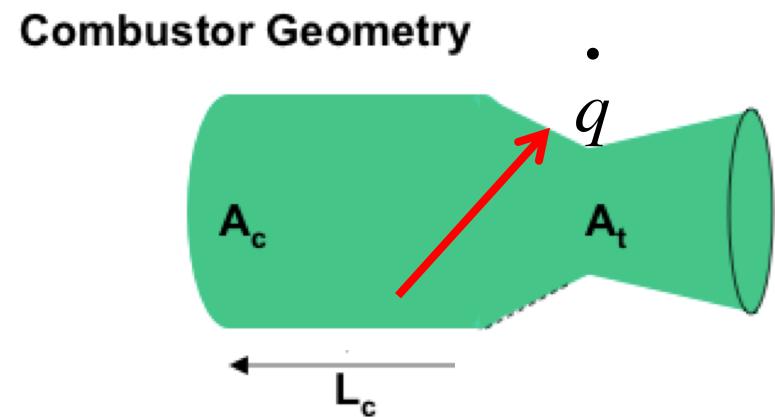
*... compute the Ideal characteristic velocity, C\*  
and ideal specific impulse (infinite nozzle)*

## 2) Non-Ideal Combustion Temperature <sup>2</sup> (<sup>1</sup> point)

- For the same Rocket in (1) and (2) ... Assume that there is regenerative cooling in the combustor .....

When operating in steady state mode  
if the regenerative cooling system removes

700 kWatts / kg/sec of propellant mass flow  
entering the combustor



**Calculate the actual stagnation temperature at the throat of the combustor and the efficiency of the combustion-->**

$$\eta^* = \frac{1}{C_{ideal}^*} \cdot \left( \frac{P_0 \cdot A^*}{\dot{m}} \right) = \frac{C_{actual}^*}{C_{ideal}^*}$$

Assume that heat transfer has no effect on combustion products and that Flow is “frozen” with species mix defined by O/F = 5.333

### 3) Combustor Geometry (1 point)

- For the same Rocket in (1) and (2) ...

Ranges of Combustor Characteristic Length

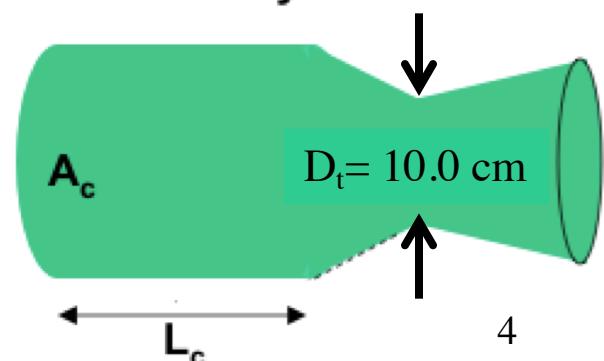
Propellants	Characteristic Length $L^*$	
	Low(m)	High(m)
Liquid fluorine / hydrazine	0.61	0.71
Liquid fluorine / gaseous H <sub>2</sub>	0.56	0.66
Liquid fluorine / liquid H <sub>2</sub>	0.64	0.76
Nitric acid/hydrazine	0.76	0.89
N <sub>2</sub> O <sub>4</sub> / hydrazine	0.60	0.89
Liquid O <sub>2</sub> / ammonia	0.76	1.02
Liquid O <sub>2</sub> / gaseous H <sub>2</sub>	0.56	0.71
Liquid O <sub>2</sub> / liquid H <sub>2</sub>	0.76	1.02
Liquid O <sub>2</sub> / RP-1	1.02	1.27
H <sub>2</sub> O <sub>2</sub> / RP-1 (incl. catalyst)	1.52	1.78

- Given the following Characteristic length Values,

*Calculate a mid range value for The combustor Volume based on*

$$L_{mean}^* = \frac{L_{high}^* + L_{low}^*}{2}$$

#### Combustor Geometry



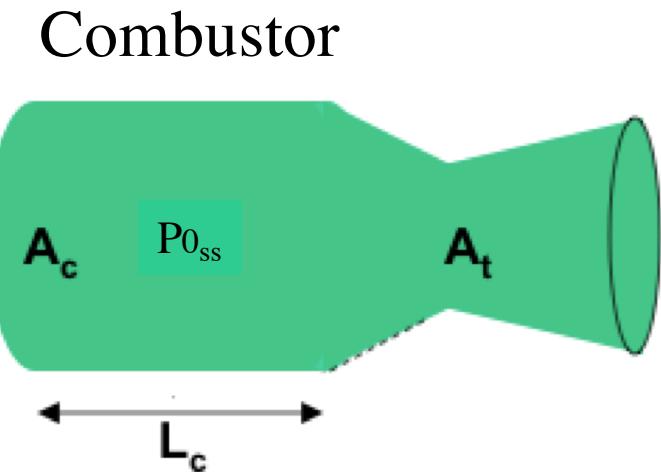
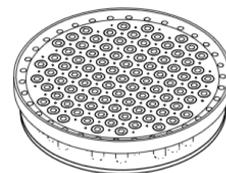
## 4) Steady Combustor Pressure (2 points)

- For the same Rocket in (1) and (2), and the combustor  $L^*$  geometry from (3) ... Use the true stagnation temperature calculated in (2), and assume that the combustion products are frozen as in (1) ... for the following injector properties .. ***Calculate the steady state combustor pressure (assume incompressible propellants)***
- What LOX Injector feed pressure is required to give the O/F ratio calculated in Part 1?***
- LOX Injector

Port diameter: 0.1841 cm  
100 injector ports  
LOX density: 1.140 g/cm<sup>3</sup>  
Cd (discharge coeff.): 0.75

- LH<sub>2</sub> Injector

Port diameter: 0.1596 cm  
100 injector ports  
LH<sub>2</sub> density: 71 g/cm<sup>3</sup>  
Injector pressure 3700 kPa  
Cd (discharge coeff.): 0.75



***Base calculations on actual combustor temperature***

## 5) Design Specific impulse (2 points)

- Given the results of problem (1), (2), (3) and (4)

*calculate the Optimal (design) Specific Impulse when the nozzle has A design altitude of 20 km ( $p^\infty = 5.4748 \text{ kPa}$ )*

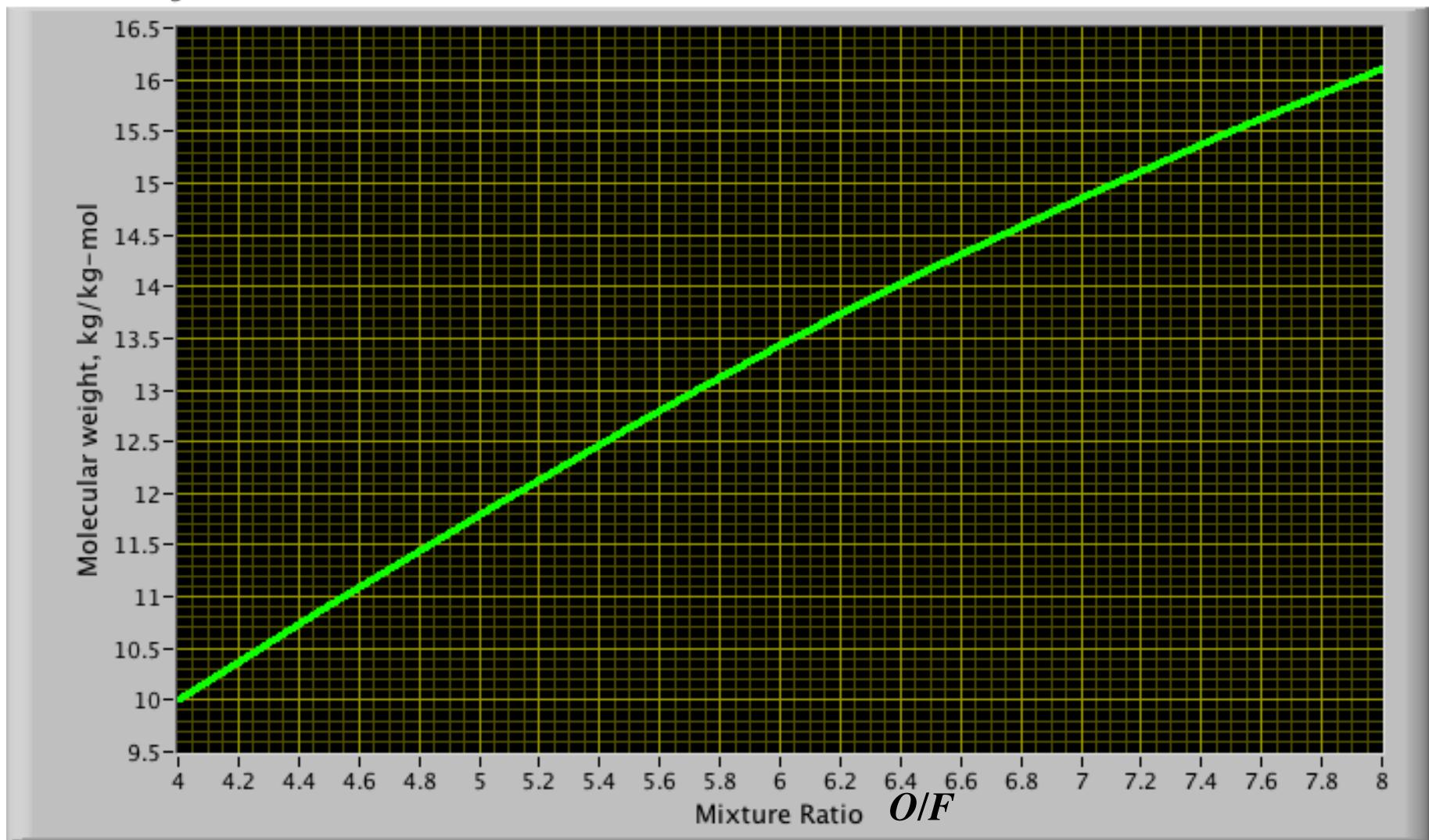
.... Assume nozzle exit divergence angle is zero, *be sure to include effects of combustor heat loss*

*Calculate the expansion ratio of this nozzle*

*What Thrust and Isp Penalty would you pay if you used a conical nozzle of same expansion ratio; but with an exit angle of 20 degrees*

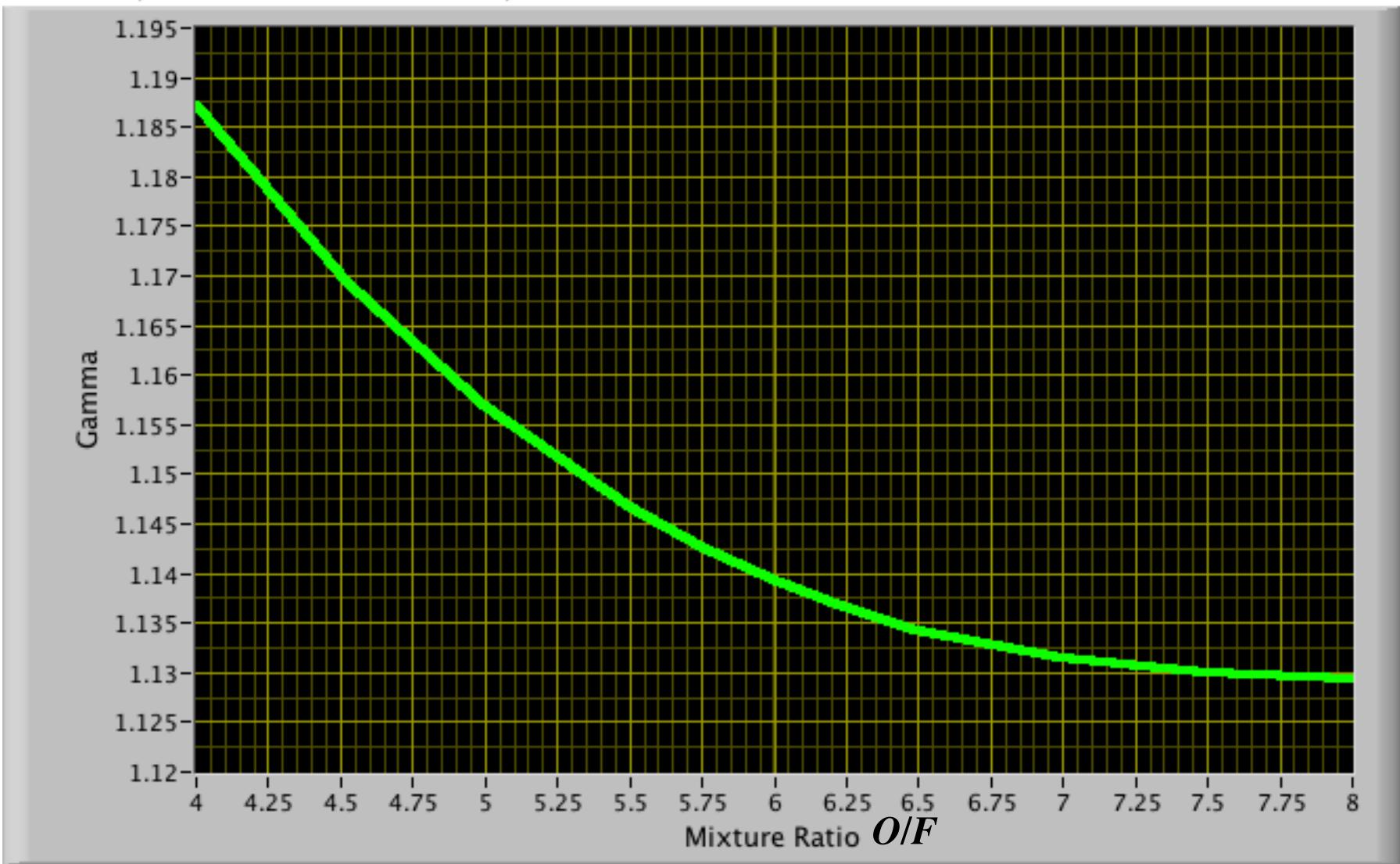
# Propellant Combustion Properties

Molecular Weight of Idealized Combustion Products



# Propellant Combustion Properties

Idealized Specific heat ratio of Combustion products



# Propellant Combustion Properties

Adiabatic Flame (Ideal Combustor) temperature

