(Classical) Molecular Dynamics for Dummies

Computational Techniques for Molecular Modeling 2020-2021

Bibliography

will be extend as course continues

- Griebel, Knapek, Zumbusch, Numerical Simulation in Molecular Dynamics, (Springer 2007)
- Frenkel, Smit, Understanding Molecular Simulation: from Algorithms to Applications, (Academic Press, 2002)
- Rapaport, The Art of Molecular Dynamics Simulation, (Cambridge University Press, 2004)
- Tuckerman, Statistical Mechanics: Theory and Molecular Simulation, (Oxford University Press, 2010)
- Allen, Tildesley, Computer Simulation of Liquids, (Clarendon Press, 1989)
- Leimkuhlerm, Reich, Simulating Hamiltonian Dynamics (Cambridge University Press, 2004)
- Israelachvili, Intermolecular and Surface Forces, (Academic Press, 2011)
- Kuriyan, Konforti, Wemmer, The MOLECULES of LIFE: Physical and Chemical Principles, (Garland, 2013)
- Hairer, Lubich, Wanner, Geometric Numerical Integration: Structure-Preserving Algorithms for Ordinary Differential Equations, (Springer, 2006)

Interactions Among Molecules

Molecular dynamics: numerical, step-by-step, solution of the classical equations of motion

$$m_i \dot{r}_i = f_i \text{ or, equivalently} \begin{cases} \dot{r}_i = \frac{p_i}{m_i} \\ \dot{p}_i = f_i \end{cases}$$

the forces f_i are assumed to be gradients of a potential energy $U, f_i = -\nabla_i U$

 r_i are the **coordinates** of the i-th atom

Non-bonded Interactions Molecules

The non-bonded potential U_{nb} can be split as

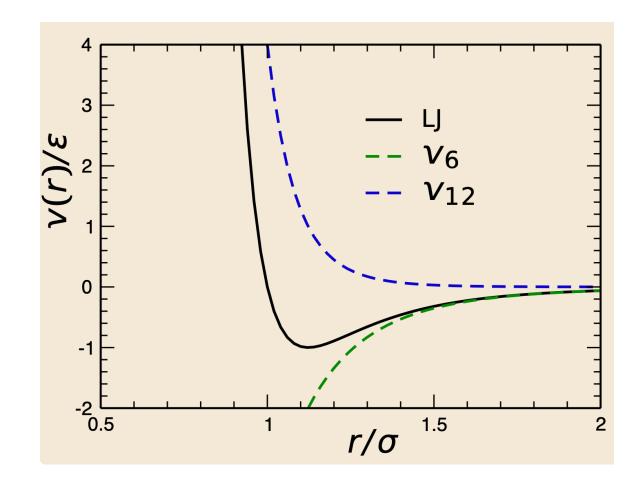
$$U_{nb}(r) = \sum_{i} u(r_i) + \sum_{i,j>i} v(r_i, r_j) + \dots$$

- The external potential u is often neglected for bulk systems
- Pair-wise interactions v
- Higher order interactions usually neglected

There is an extensive literature on how these potentials are computed (e.g. Israelachvili 20013)

Lennard-Jones Potential

- Sometimes even simplest models provide interesting insight into physics
- The Lennard-Jones potential is very often used



$$v^{LJ}(r_i, r_j) = 4\varepsilon \left(\left(\frac{\sigma}{|r_i - r_j|} \right)^{12} - \left(\frac{\sigma}{|r_i - r_j|} \right)^6 \right)$$

- σ diameter, ε potential barrier height
- models empirically dipole-dipole and dipole-induced dipole interactions

Electrostatics

• Electrically charged ions interact via Coulomb potential (long range)

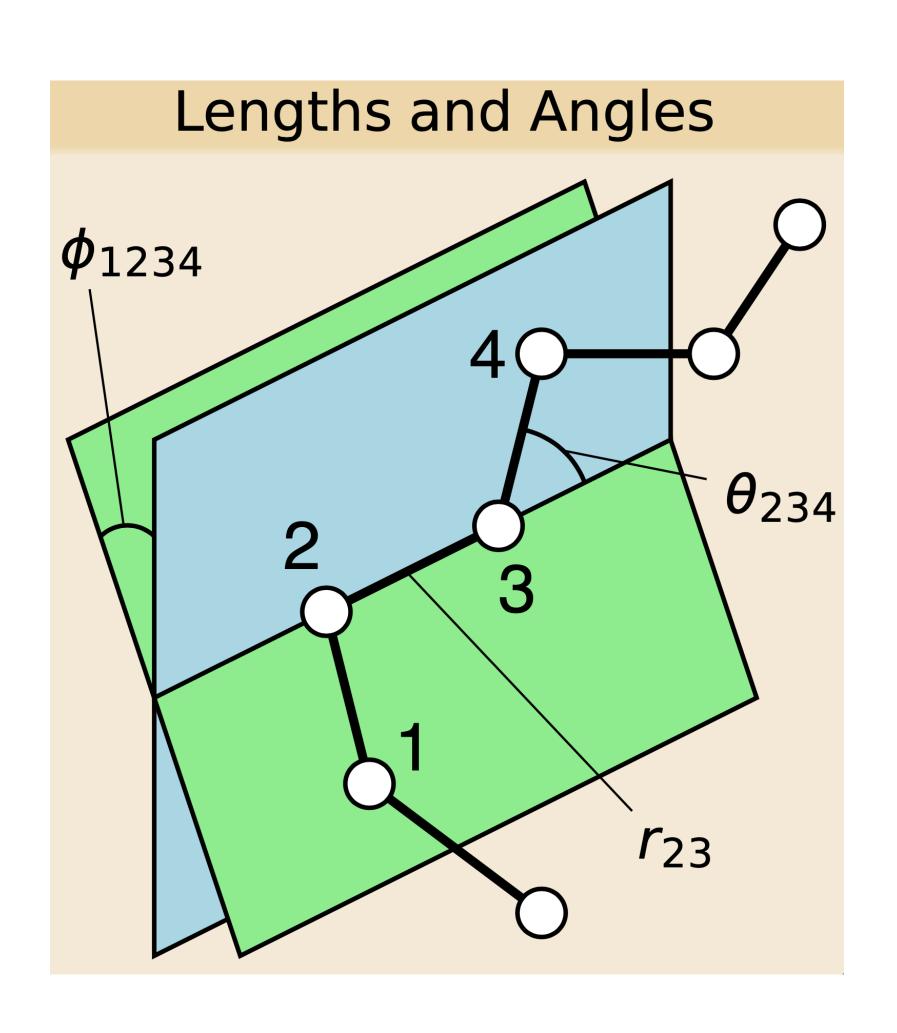
$$v^{qq}(r_i, r_j) = \frac{q_i q_j}{4\pi \varepsilon_0 |r_i - r_j|}$$

- q_i , q_i charges, ε_0 permittivity of free space
- correctly handling long range interactions in MD algorithms is a complex task
- dipole and multipole interactions

Atoms and Molecules

- For molecular systems, we simply build the molecules out of Lennard-Jones site-site potentials, or similar
- Typically, a single-molecule quantum-chemical calculation may be used to estimate the electron density throughout the molecule
- This may then be modelled by a distribution of partial charges
- or more accurately by a distribution of electrostatic multipoles

Bonding Interactions



- For molecules we must also consider the intramolecular bonding interactions.
 Consider this geometry of an alkyl chain (just showing the carbons).
- interatomic distance r_{23}
- bend angle θ_{234}
- torsion angle Φ_{1234}

Example

$$U_{int} = \frac{1}{2} \sum_{ij} k_{ij}^{r} (|r_i - r_j| - r_0)^2 + \frac{1}{2} \sum_{ijk} k_{ijk}^{\theta} (\theta_{ijk} - \theta_0)^2 + \frac{1}{2} \sum_{ijk} \sum_{ijkl} k_{ijkl}^{\Phi,m} \left[1 + \cos(m\Phi_{ijkl} - \gamma_m) \right]$$

Bond Stretching

$$\frac{1}{2} \sum_{ij} k_{ij}^{r} (|r_i - r_j| - r_0)^2$$

- Distance between atoms
- Harmonic oscillator with specified equilibrium separation
- Trivial to compute forces
- High vibration frequencies
 - C-H bond implies about 10fs period
 - Explicit simulation would require time steps in the order of fs

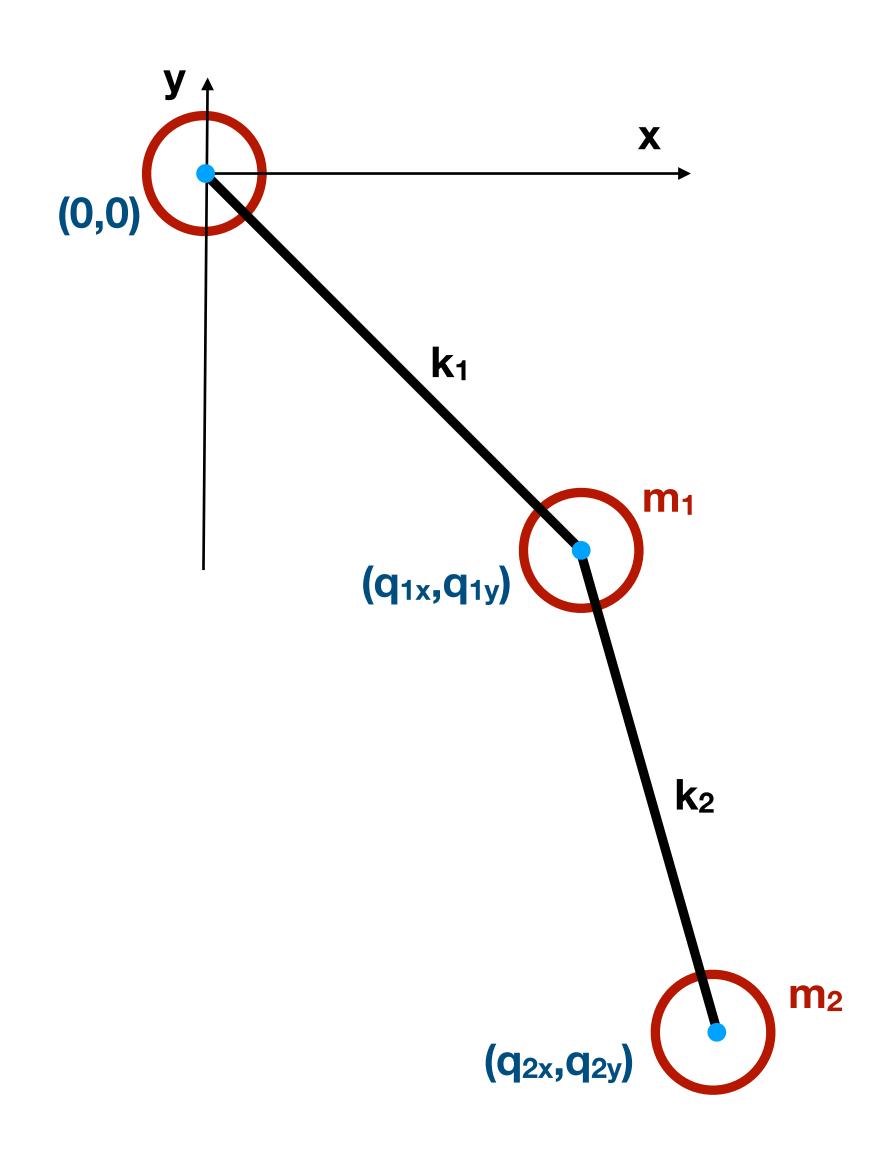
Bond Bending

$$\frac{1}{2} \sum_{ijk} k_{ijk}^{\theta} (\theta_{ijk} - \theta_0)^2 + \frac{1}{2} \sum_{m} \sum_{ijkl} k_{ijkl}^{\Phi,m} \left[1 + \cos(m\Phi_{ijkl} - \gamma_m) \right]$$

- Angles involve at least 2 bonds (3 atoms)
- Forces affect all involved atoms
- Computation involved but straightforward
- Timescales
 - in H_2O about 20 fs
 - Explicit simulation would require time steps in the order of fs

MD Algorithm

- Calculating forces is expensive, typically a pairwise sum over atoms, so we need to perform this as infrequently as possible.
 - Wish to make the timestep as large as possible
 - Hence, simulation algorithms tend to be low order (i.e. do not use high derivatives of r)
 - This allows the time step to be increased as much as possible without jeopardizing energy conservation.
- Cannot accurately follow true trajectory for very long simulation times.
 - trajectories diverge from each other exponentially
 - however long-term energy conservation is possible

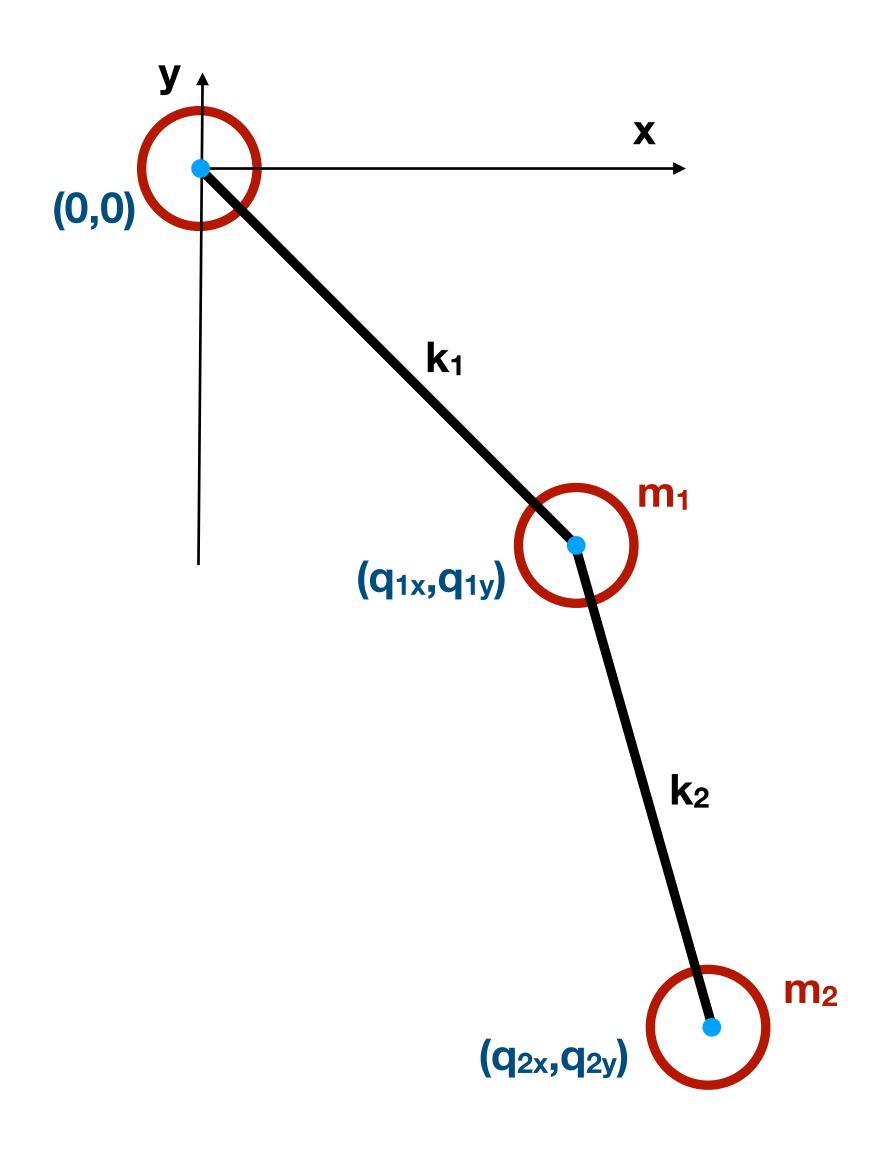


$$\mathbf{p}_i := m_i \dot{\mathbf{q}}_i$$

$$E = K(\mathbf{p}_1, \mathbf{p}_2) + T(\mathbf{q}_1, \mathbf{q}_2)$$

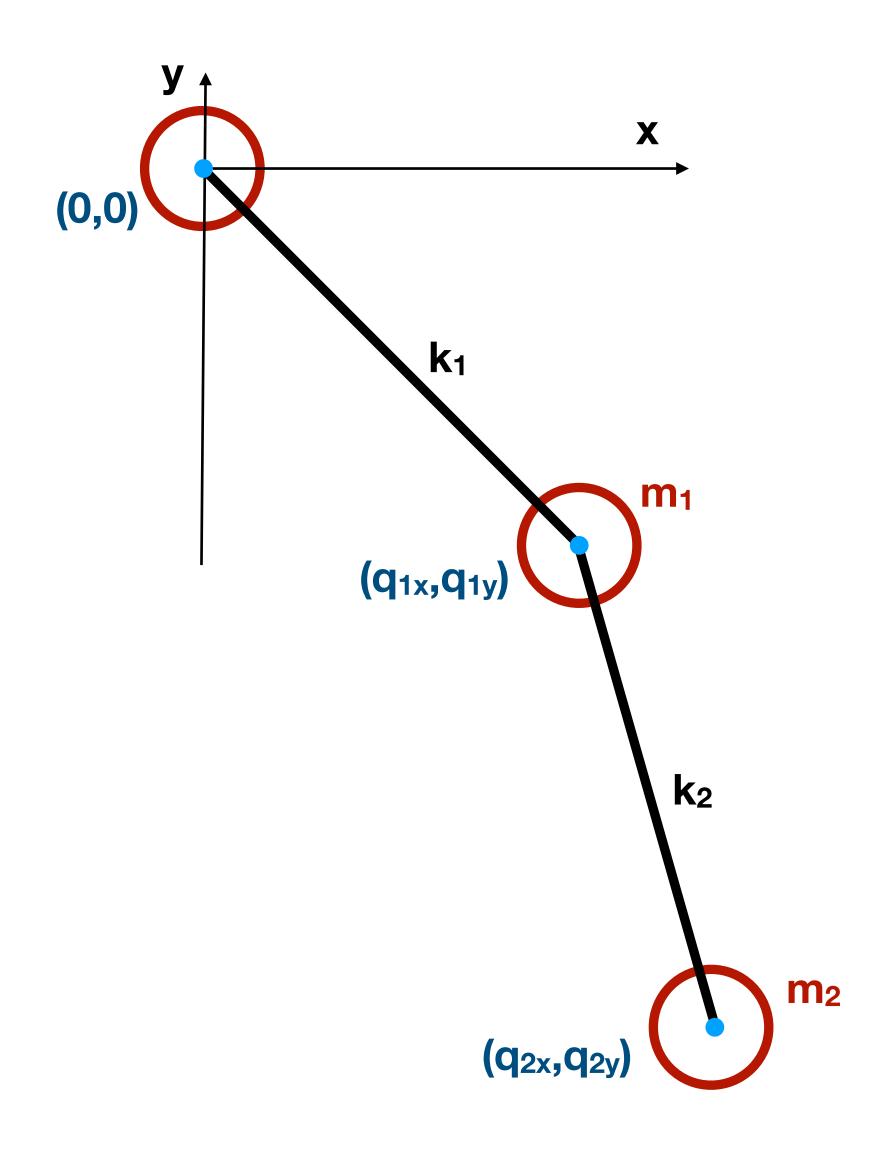
$$K(\mathbf{p}_1, \mathbf{p}_2) = \frac{|\mathbf{p}_1|^2}{2m_1} + \frac{|\mathbf{p}_2|^2}{2m_2}$$

$$T(\mathbf{q}_1, \mathbf{q}_2) = -m_1 \mathbf{g} \cdot \mathbf{q}_1 - m_2 \mathbf{g} \cdot \mathbf{q}_2 + \frac{k_1}{2} \left(\left| \mathbf{q}_1 \right| - l_1 \right)^2 + \frac{k_2}{2} \left(\left| \mathbf{q}_1 - \mathbf{q}_2 \right| - l_2 \right)^2$$



$$\dot{\mathbf{q}}_i = \frac{\partial K}{\partial \mathbf{p}_i}$$

$$\dot{\mathbf{p}}_i = -\frac{\partial T}{\partial \mathbf{q}_i}$$



```
m1 = 1;

m2 = .2;

k1 = 100;

k2 = 100;

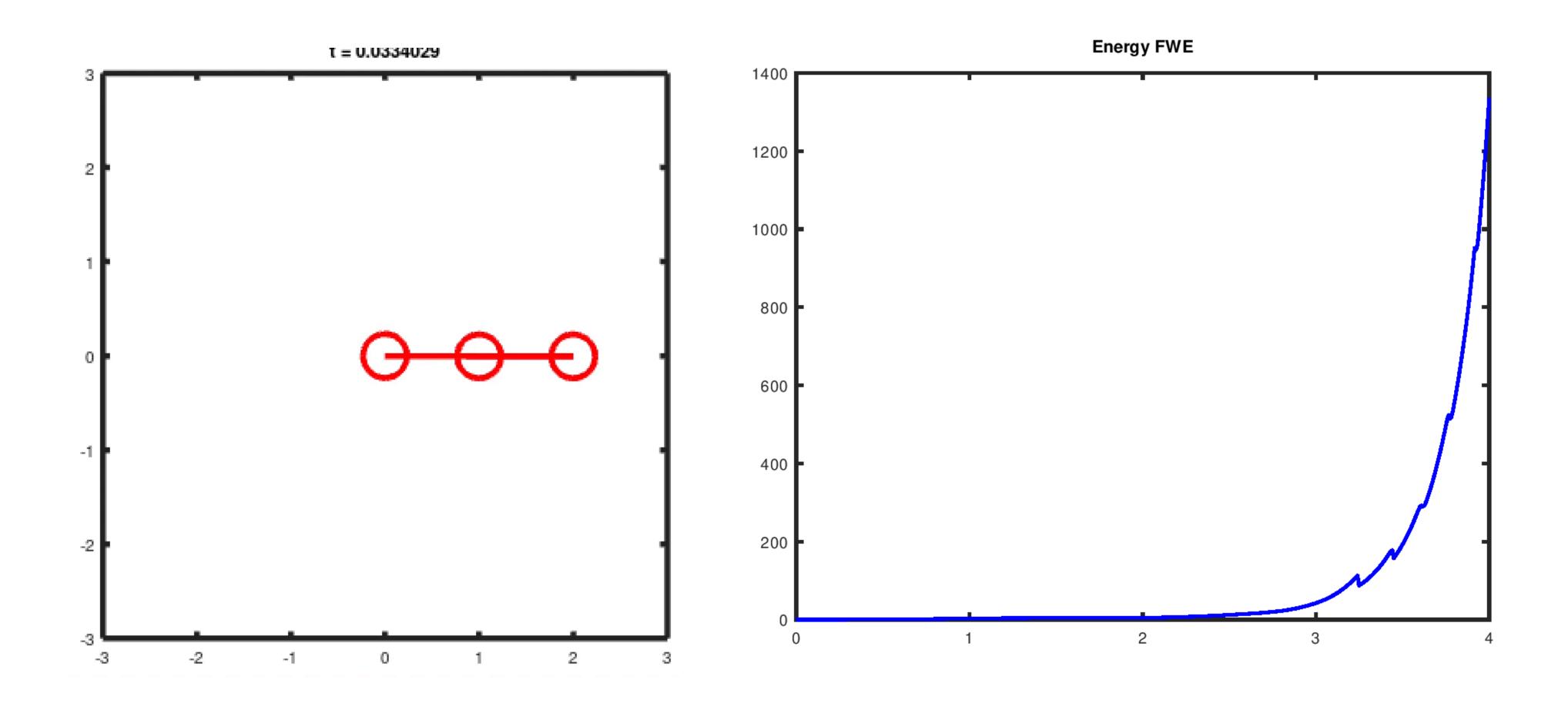
l1 = 1;

l2 = 1;

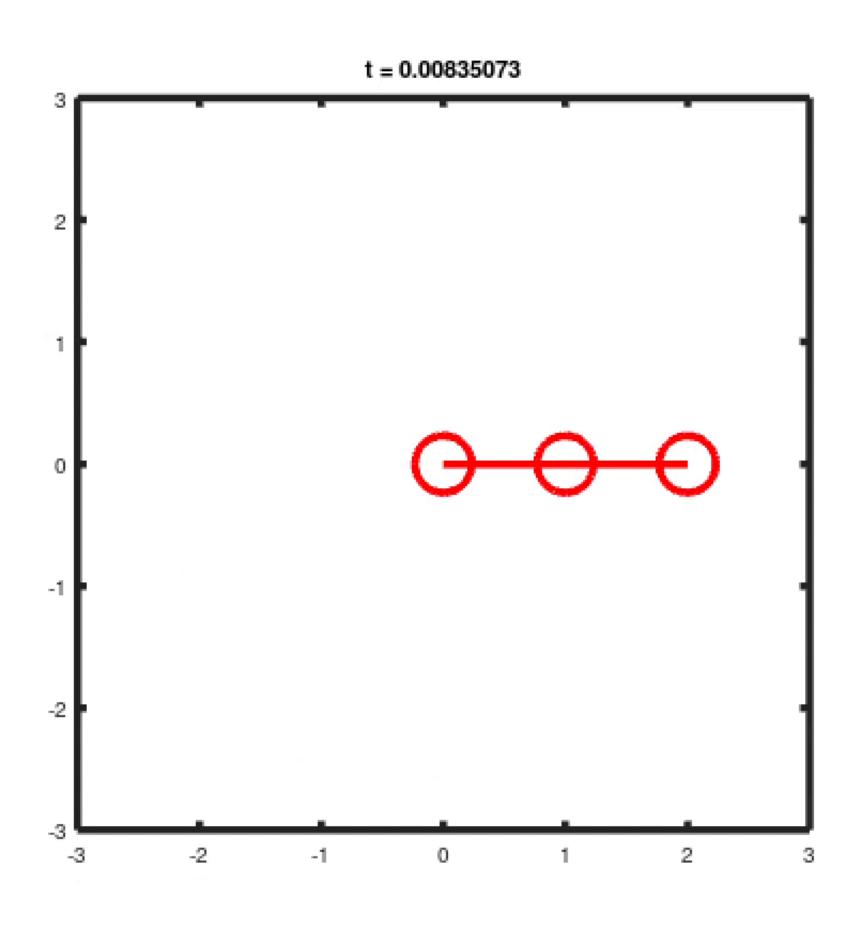
gx = 0;

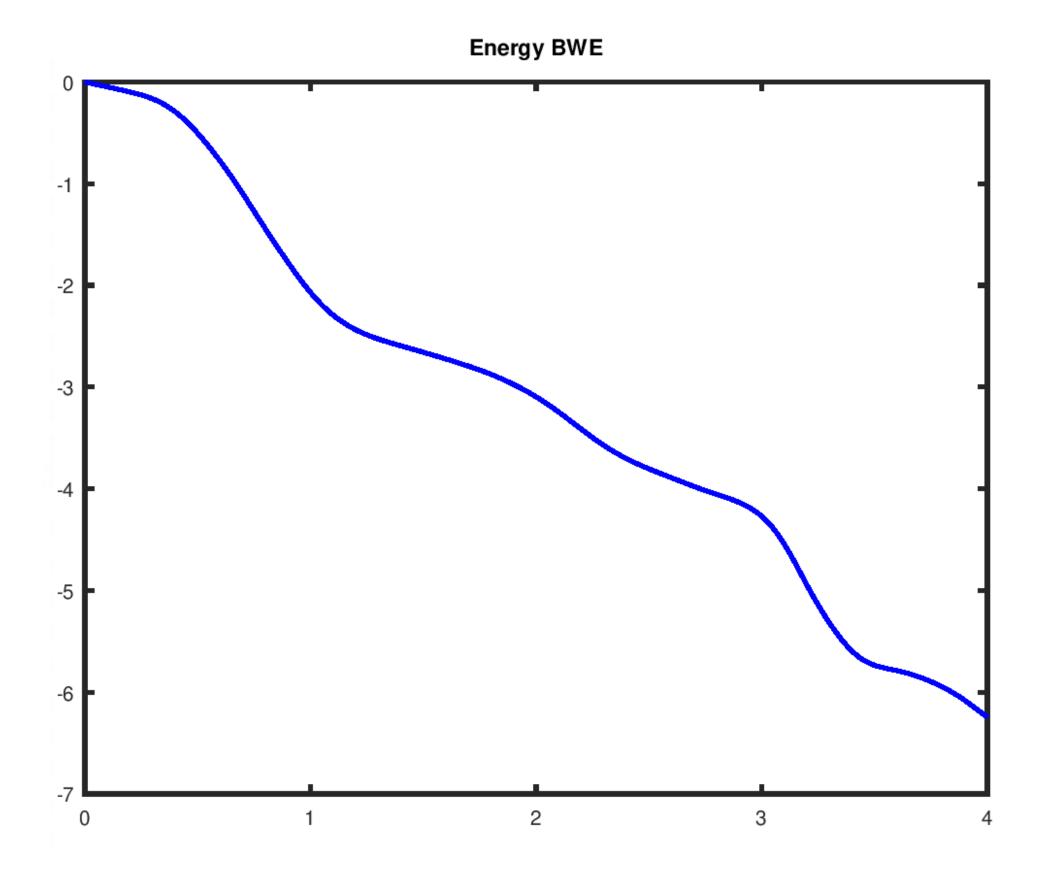
gy = -9.81;
```

Forward Euler

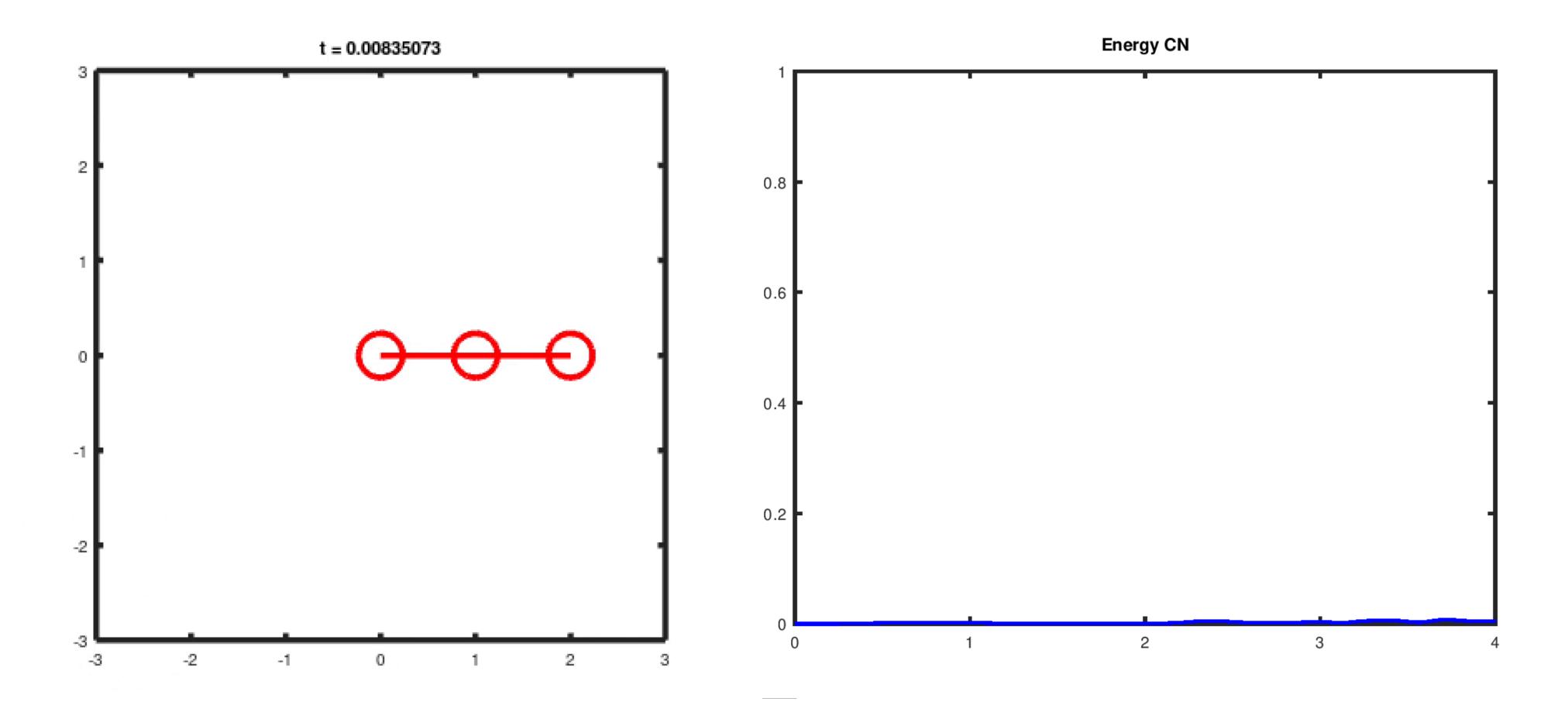


Backward Euler

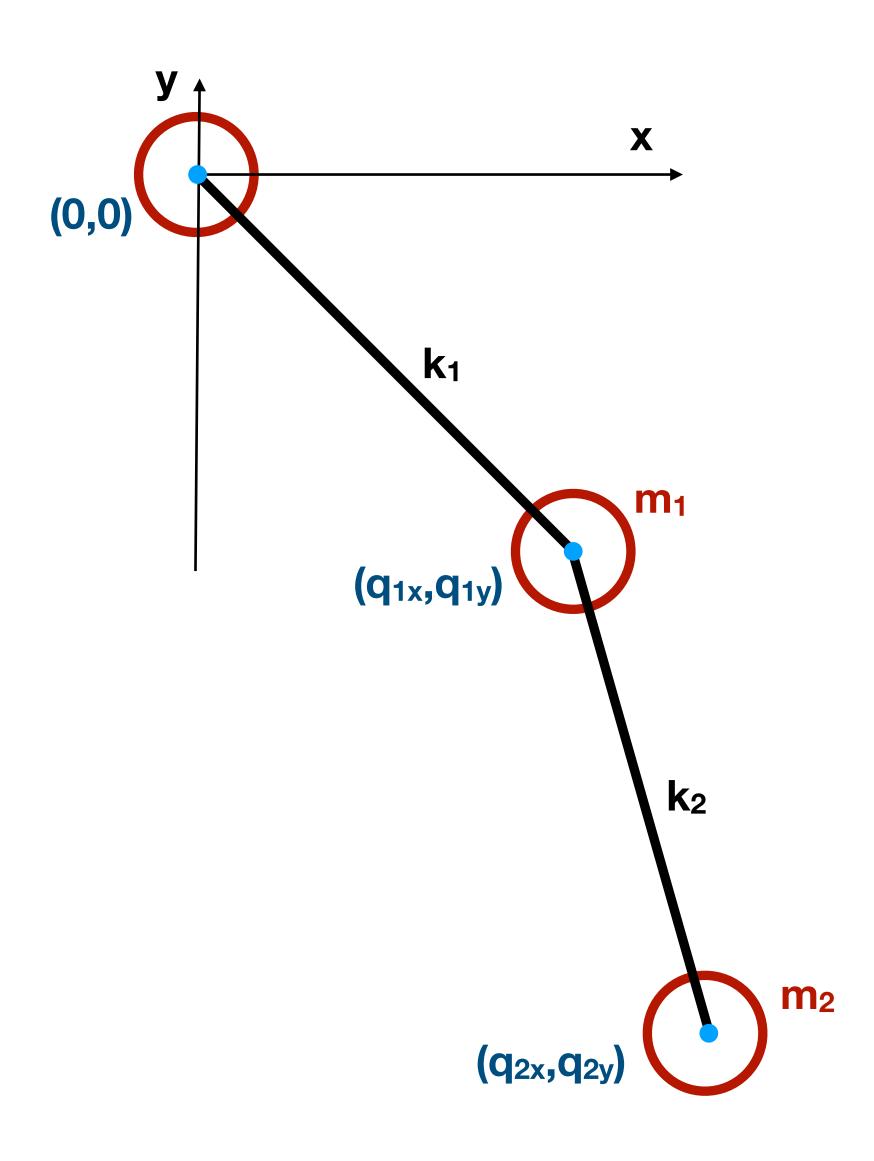




Trapezoidal Rule



Double Elastic Pendulum with Friction

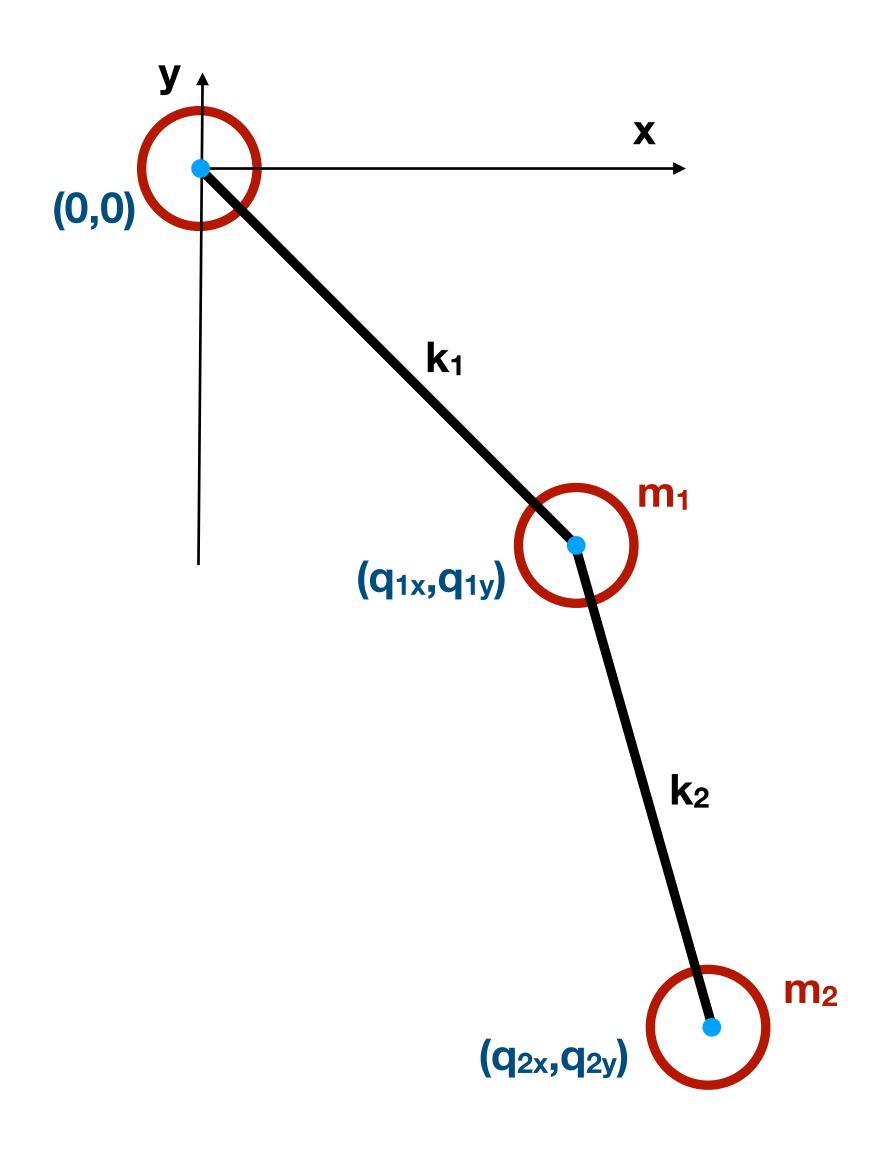


$$\mathbf{p}_i := m_i \dot{\mathbf{q}}_i$$

$$E = K(\mathbf{p}_1, \mathbf{p}_2) + T(\mathbf{q}_1, \mathbf{q}_2) + \theta S$$

$$K(\mathbf{p}_1, \mathbf{p}_2) = \frac{|\mathbf{p}_1|^2}{2m_1} + \frac{|\mathbf{p}_2|^2}{2m_2}$$

$$T(\mathbf{q}_1, \mathbf{q}_2) = -m_1 \mathbf{g} \cdot \mathbf{q}_1 - m_2 \mathbf{g} \cdot \mathbf{q}_2 + \frac{k_1}{2} \left(\left| \mathbf{q}_1 \right| - l_1 \right)^2 + \frac{k_2}{2} \left(\left| \mathbf{q}_1 - \mathbf{q}_2 \right| - l_2 \right)^2$$

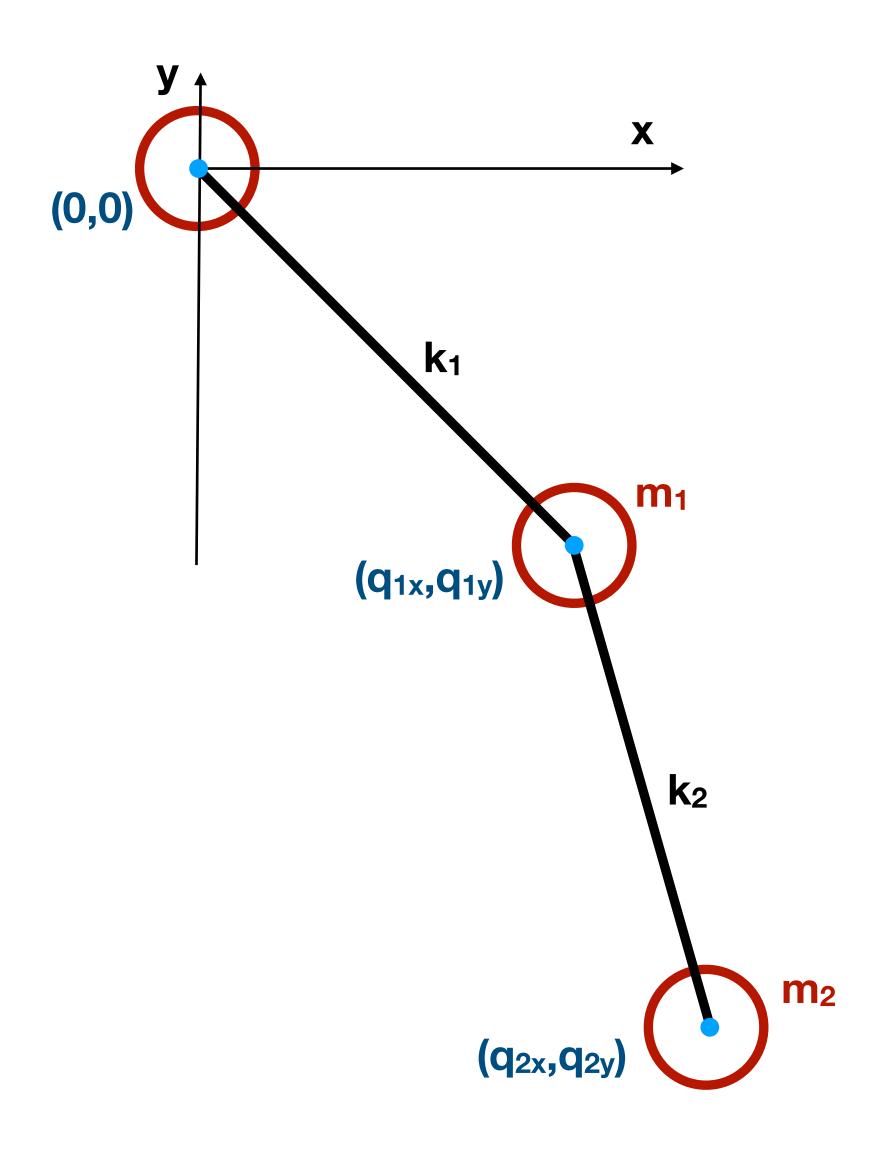


$$\dot{\mathbf{q}}_i = \frac{\partial K}{\partial \mathbf{p}_i}$$

$$\dot{\mathbf{p}}_{i} = -\frac{\partial T}{\partial \mathbf{q}_{i}} \frac{\lambda}{m_{i}} \mathbf{p}_{i}$$

$$\dot{s} = \frac{\lambda}{\theta m_{i}^{2}} \left| \mathbf{p}_{i} \right|^{2}$$

$$\dot{s} = \frac{\lambda}{\theta \ m_i^2} \left| \mathbf{p}_i \right|^2$$



```
m1 = 1;

m2 = .2;

k1 = 100;

k2 = 100;

l1 = 1;

l2 = 1;

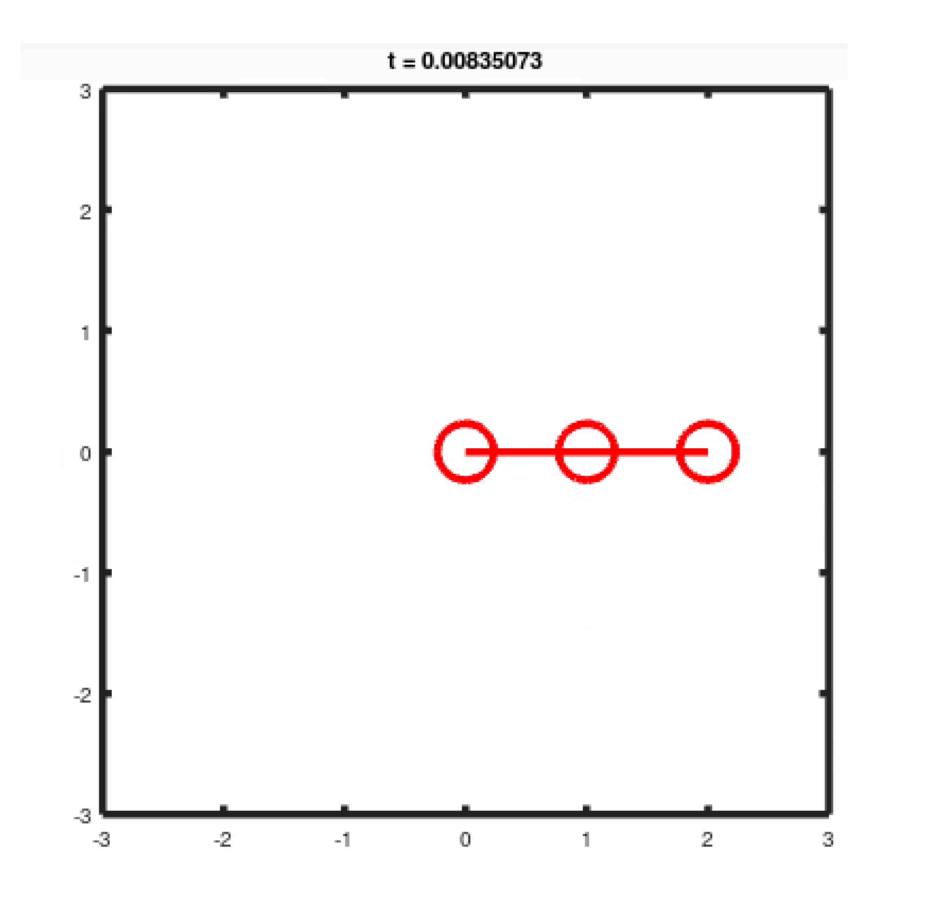
gx = 0;

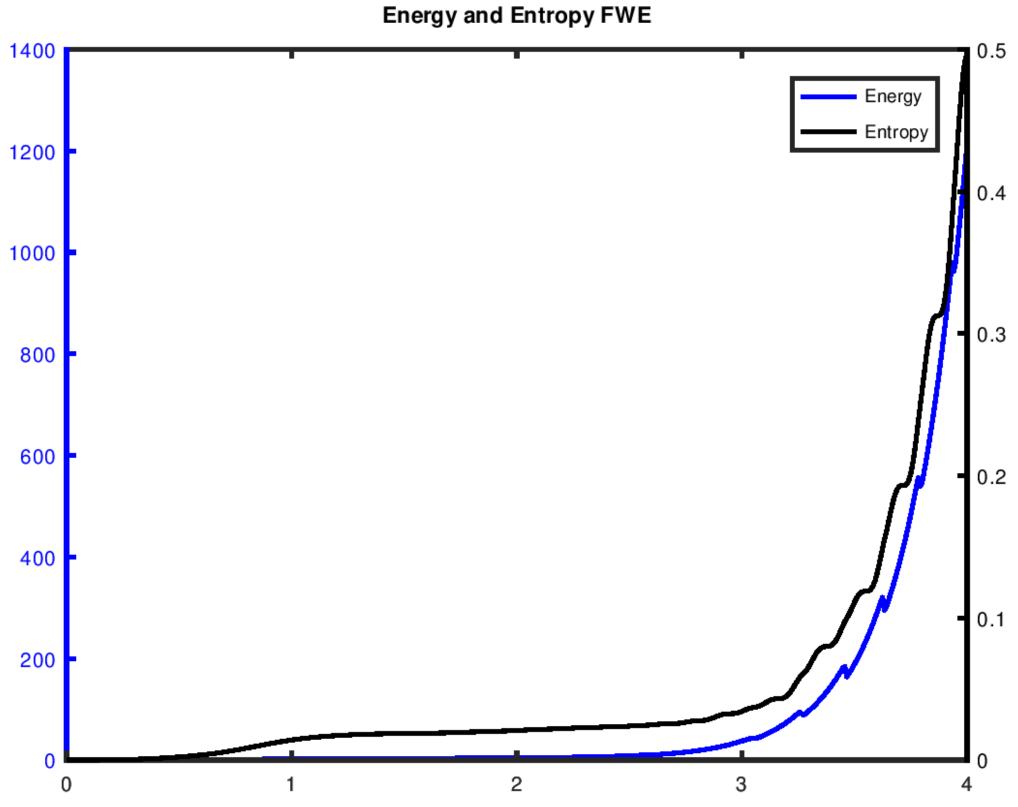
gy = -9.81;

fr = .1;

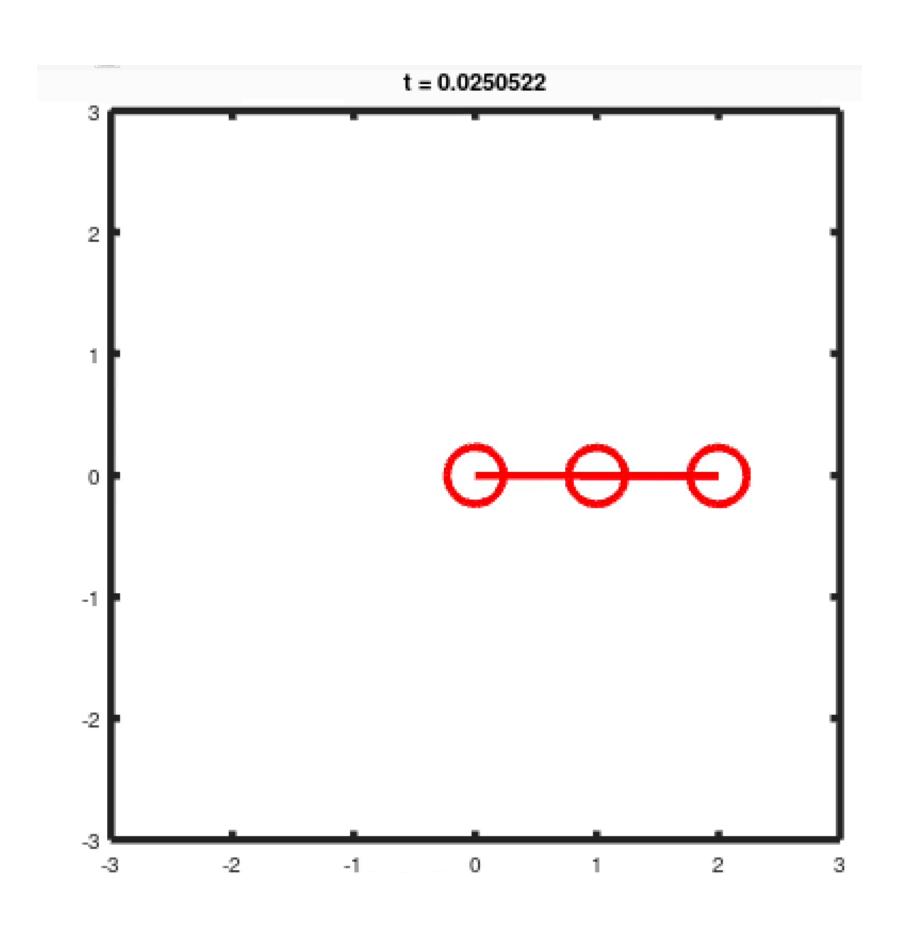
th = 300;
```

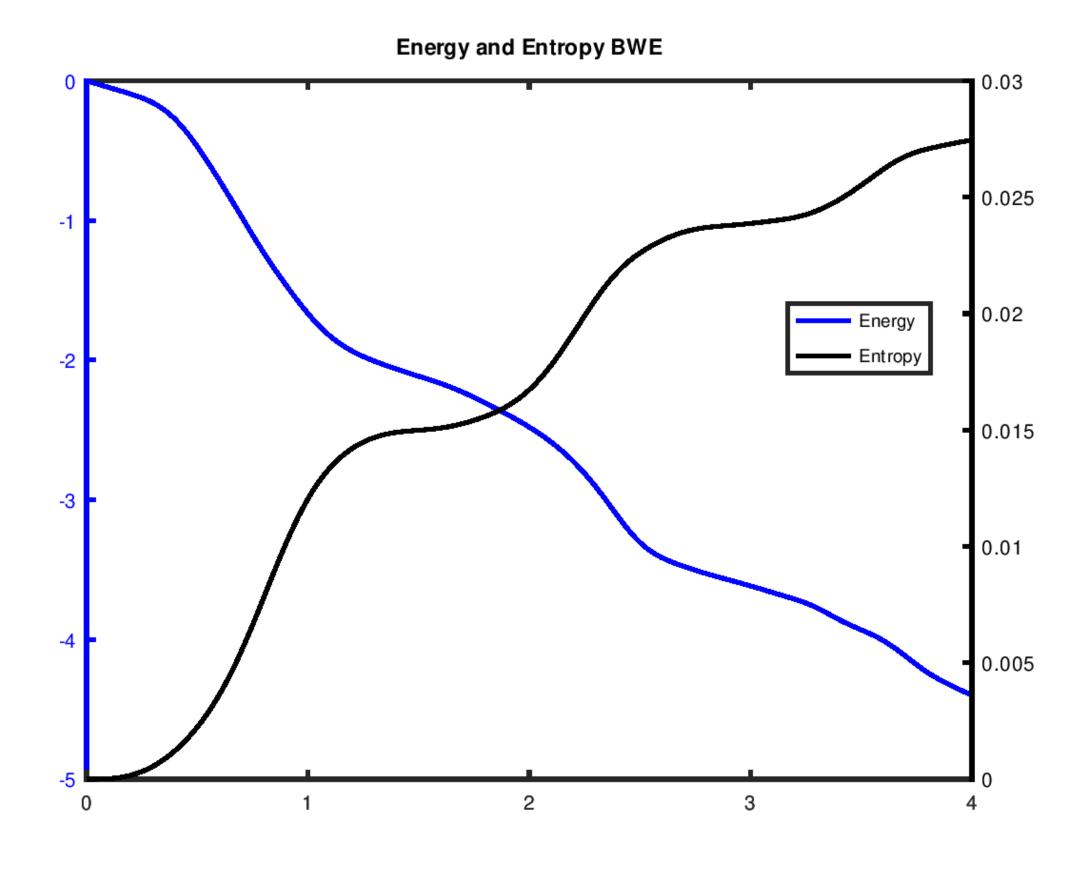
Forward Euler



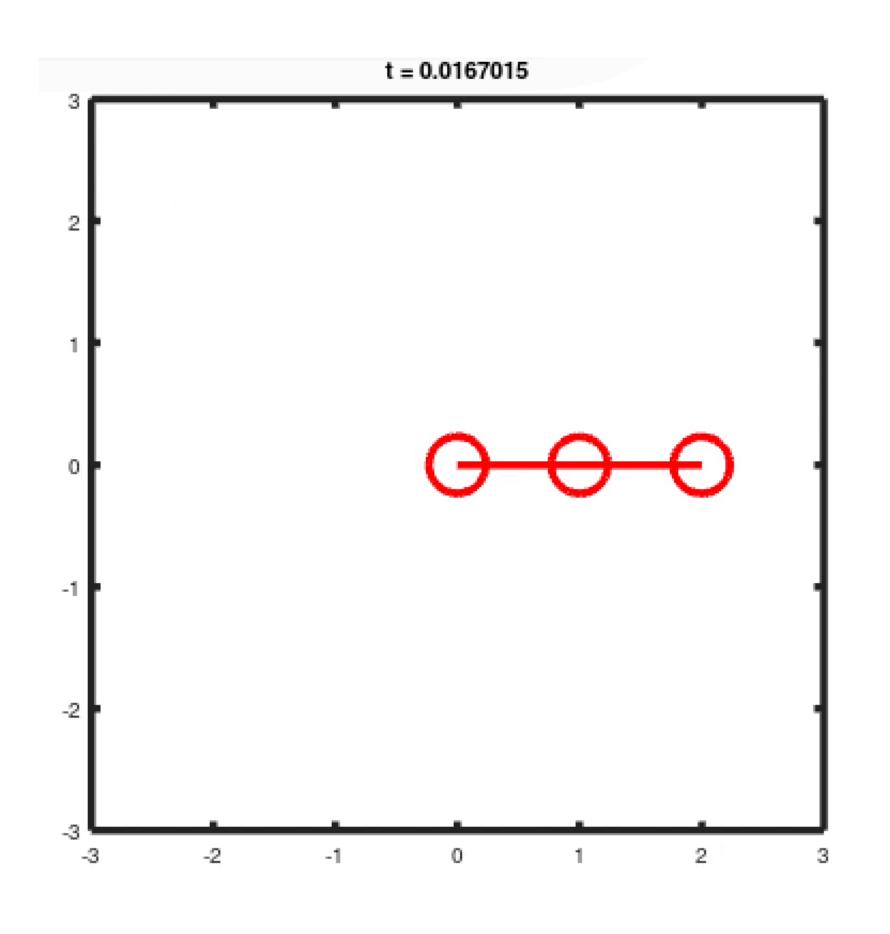


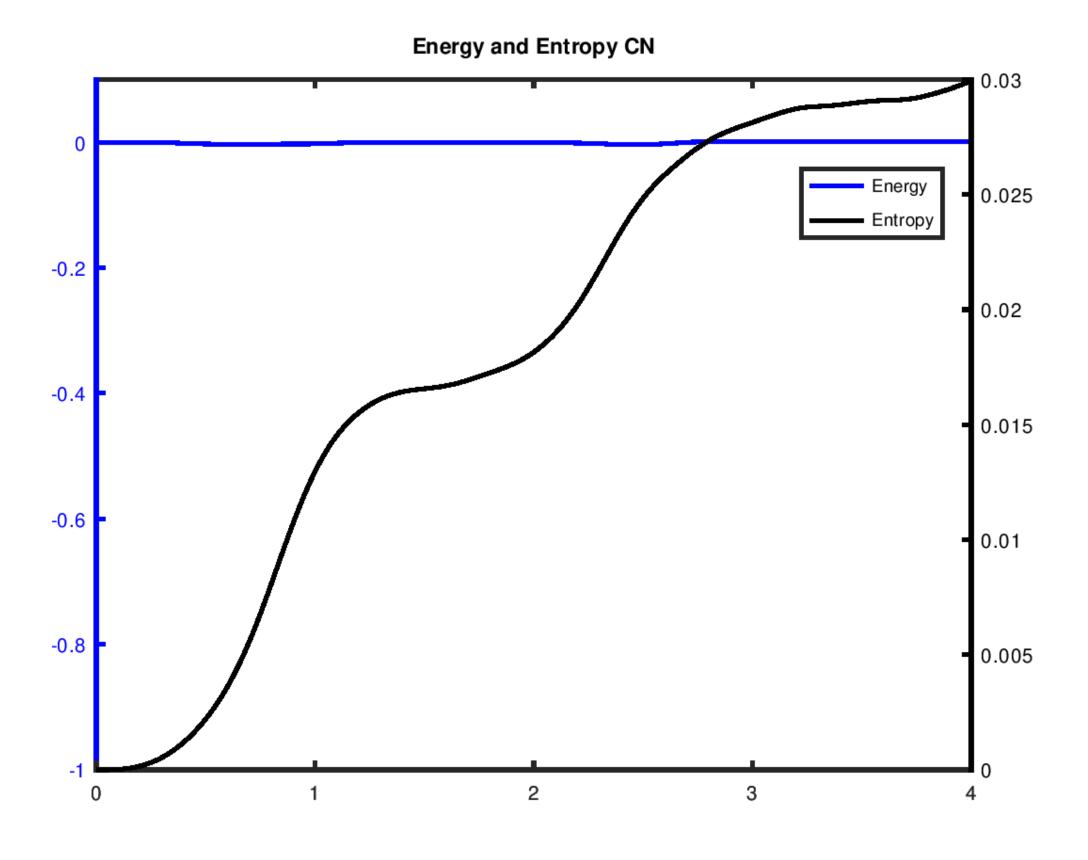
Backward Euler





Trapezoidal Rule





Symplectic (Geometric) Integrators

- Methods especially tailored for simulating Hamiltonian systems
- Very popular in MD is the Velocity-Verlet algorithm

$$p_{n+1/2} = p_n + h/2 f_n$$

$$r_{n+1} = r_n + h p_{n+1/2}/m$$

$$p_{n+1} = p_{n+1/2} + h/2 f_{n+1}$$

- Explicit, 1-step, 2-stage method
- h time step

Verlet Algorithm

- Important features of the Verlet algorithm are:
 - it is exactly time reversible
 - it is symplectic
 - it is 2nd order in time, permitting long timesteps
 - it requires just one force evaluation per step
 - it is easy to program

Verlet Algorithm

- The trajectories generated by the above scheme are approximate, and will not conserve the true energy H.
- Nonetheless, they do exactly conserve a "pseudo-hamiltonian" or "shadow hamiltonian" H[‡]
- H and H[‡] differ from each other by a small amount, H[‡] =H + $O(\Delta t2)$.
- This means that the system will remain on a hypersurface in phase space which is "close" to the true constant-energy hypersurface.
- Such a stability property is extremely useful in MD, since we wish to sample constant-energy states.