A LINE SURVEY OF ORION-KL FROM 607 TO 725 GHz

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ABSTRACT

With the Caltech Submillimeter Telescope, we have performed an unbiased line survey of Orion-KL in the frequency range 607–725 GHz. We were able to identify lines down to a threshold of 1–2 K in main beam brightness temperature units, and we found 1064 spectral features consisting of 2032 lines, partially blended. Apart from the abundant diatomic rotors such as CO, CS, SO, the spectrum is dominated by CH₃OH and SO₂, both in terms of numbers of lines and integrated flux. The number of unidentified lines is 155 or 14%. We also report the tentative identification of the new molecule SiH in our data. For some complex organic molecules, we find rotation temperatures and column densities which are much higher than those found in earlier studies. It is likely the cause of this is a significant contribution from a very compact, hot component of the Orion molecular cloud core which was not visible in lower frequency surveys.

Subject headings: ISM: individual (Orion Kleinmann-Low) — ISM: molecules — line: identification —

radio lines: ISM — surveys

On-line material: machine-readable tables

1. INTRODUCTION

It is now well known that the interstellar medium contains regions of high density $[n(H_2) = 10^3 - 10^8 \text{ cm}^{-3}]$, known as molecular clouds, in which star formation takes place. In these regions, a complex chemistry gives rise to a set of abundant molecular species. Each molecule generates a set of spectral lines, resulting in a "line forest." As the sensitivity of telescopes and receivers has improved, it has become possible to reach the line confusion limit toward some of these regions, which are particularly prolific line sources. To determine the presence and measure the abundance of any given molecule requires the observations of many lines. The best way to do this is by means of a line survey. In fact, unbiased spectral line surveys are necessary to obtain an unbiased view of the chemistry of a specific region. Line surveys so far have concentrated on two objects: IRC +10216 as a representative of an evolved star, and Orion-KL as a representative of star-forming hot cores (Groesbeck, Phillips, & Blake 1994; Schilke et al. 1997 and references therein). A few surveys have been performed for other objects of these two classes (e.g., Sutton et al. 1991 in Sgr B2, Bell, Avery, & Watson 1993 in W51, Macdonald et al. 1996 in G34.3, and Helmich & van Dishoeck 1997 in W3) and some work for dark clouds (Kaifu et al. 1987), but nothing approaching the extent of data toward these two sources. The data presented so far cover only the spectral range up to 360 GHz, with the exception of Harris et al. (1995), who surveyed a small frequency range around from 685–692 GHz in Orion-KL. Some spectral regions are inaccessible to ground based observations because of atmospheric absorption, but two large windows in the submillimeter, at 450 μm (corresponding to ≈ 600 – 725 GHz) and at 350 μ m (corresponding to \approx 780– 960 GHz), remain largely unobserved.

In this paper we complete the study of the 450 μ m window for Orion-KL, while a survey in the 350 μ m range has been started. It might be thought that observations of these bands would produce only redundant information, because most molecules have been covered well by lower frequency surveys. This is not so, because for almost all species the transitions emitting in the submillimeter region originate from higher energy levels than at lower frequencies, thereby tracing gas with higher densities and temperatures. This trend is further enhanced by the smaller beam sizes at higher frequencies (for a given telescope) which allow a more efficient coupling to very small, highdensity, hot sources. It also is interesting to investigate the interaction of gas with dust continuum radiation, which starts to have significant optical depths in these frequency ranges. Once the database on Orion is complete, one can begin to compare other sources with Orion from a chemical point of view. The basic collected data set of Orion spectra, as observed from ground-based telescopes, will be useful to many current observers and particularly to spectroscopic studies carried out with future instruments such as the SMA, ALMA, and FIRST.

2. OBSERVATIONS AND DATA PROCESSING

Using the Caltech Submillimeter Observatory (CSO), we performed a line survey of the Orion-KL region in the frequency range 607–725 GHz (Fig. 1). This spectral range covers almost the entire 450 μ m atmospheric window bracketed by very broad, strong atmospheric H₂O absorption lines at 556.9 and 752.0 GHz. Two gaps are present in the band: one at 620.7 GHz, caused by atmospheric H₂O, and one, less deep, at 715.4 GHz, caused by atmospheric O₂. The line at 620.7 GHz prohibits observations between 620 and 623 GHz. Additionally, O₃ lines around 656 GHz worsen the transmission in this range. Any spectral lines close to these frequencies have a much higher calibration uncertainty. The part of the atmospheric window between 600 and 607 GHz could not be observed due to local oscillator power constraints.

The data were taken between 1994 November and 1995 January, using the 650 GHz facility SIS receiver (Kooi et al.

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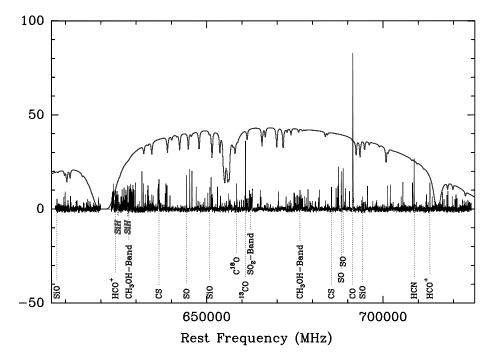


Fig. 1.—Spectrum of Orion-KL from 607 to 725 GHz. The strongest lines and frequencies of the possible newly detected molecule SiH are marked. Overlaid is a plot of the zenith atmospheric transmission for 1 mm of precipitable water vapor at Mauna Kea. The ordinate is valid for the spectrum in T_A^* units, and for the atmospheric transmission in percent.

1994). The two facility CSO 500 MHz AOS with 1024 channels each were used in a serial mode to cover approximately 1 GHz at each tuning. The observations consisted of two complete coverages of the band: the first coverage was done with a spacing of 800 MHz to ensure overlap, the second coverage was done with the same spacing, but offset by 400 MHz to the first coverage. Since the weather conditions were excellent, eight nights were sufficient to cover the

whole spectrum twice. The single sideband system temperatures were in the range 1500-3000 K, and the spectral resolution obtained is 1 MHz. The beam efficiency was measured to be 0.55 on Jupiter, and the line intensities (T_a^*) were corrected by that factor to give T_R^* -values. In the case of Orion, observing the KL region with the present resolution (10''-12'') depending on the frequency breaks it up into distinct subsources (see, e.g., Sutton et al. 1994),

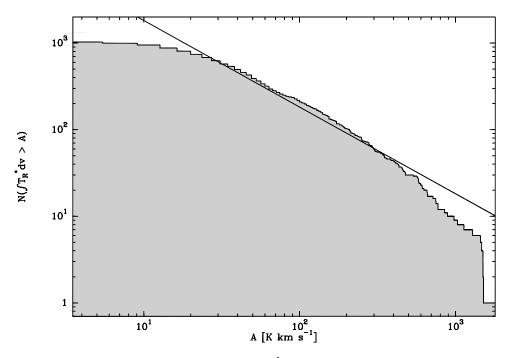


Fig. 2.—Number of lines with integrated main-beam brightness temperatures, $\int T_R^* dv > A$, larger than a value A. The power-law fit obviously does not provide a good fit to the very intense or very weak lines.

TABLE 1

LTE ROTATION TEMPERATURES AND BEAM-AVERAGED COLUMN DENSITIES, DETERMINED USING A BOLTZMANN PLOT

Line	T _{rot} (K)	σ (K)	N (cm ⁻²)	σ (cm ⁻²)	Notes
CO	100	0	6.0(19)		From $C^{18}O$, T_{rot} fixed
CS	127		1.0(15)		Two lines only
SiO	110	50	5.4(14)	6.5(14)	
HC1	150	0	1.0(14)		$T_{\rm rot}$ fixed
SiH	100	0	9.1(15)		$T_{\rm rot}$ fixed
SO	64	5	2.3(17)	0.7(17)	Corrected for optical depth
NO	90	14	1.5(17)	0.6(17)	
HCN	100	0	2.6(16)		From HC ¹⁵ N, T_{rot} fixed
SO ₂	187	4	6.0(16)	0.3(16)	
³⁴ SO ₂	192	12	8.2(15)	1.1(15)	
H_2S	129	10	1.2(16)	0.3(16)	
NH_2D	75	11	9.7(15)	3.8(15)	
CH ₃ OH	303	6	5.0(16)	0.2(16)	Optically thick?
¹³ CH ₃ OH	229	14	9.6(15)	1.1(15)	
H ₂ CO	190	9	3.2(15)	0.5(15)	
HNCO	240	13	3.1(15)	0.5(15)	
HCOOCH ₃	316	9	1.3(17)	0.1(17)	Optically thick?
CH ₃ OCH ₃	360	26	2.8(16)	0.4(16)	Optically thick?
CH ₃ CH ₂ CN	239	4	3.1(16)	0.2(16)	Optically thick?

Notes.—The column densities are determined using the main beam brightness line temperature scale. For species where only one line was available, a rotation temperature of 100 K was used, except in the case of HCl, where a temperature of 150 K was used for consistency with Schilke et al. 1995.

notably the hot core (near IRc2), where the sulphur-bearing molecules and the continuum peak, and the compact ridge, 10" southeast of it, where organic oxygen-bearing molecules have their maximum. We always were pointed toward the hot core position $\alpha_{1950} = 5^{\text{h}}32^{\text{m}}47^{\text{s}}0$, $\delta_{1950} = -5^{\circ}24'24''$. The pointing was checked by observing planets and by observing lines of molecules which are known to peak in the vicinity of the hot core. We estimate an overall accuracy of 3". The calibration was affected by occasional slight sideband imbalances (typically in the 20% range, occasionally reaching 50%). However, since each line was observed four times with different tunings, this was not a serious problem. We estimate the overall calibration accuracy to be 30%, except close to the deep atmospheric absorptions, where it is worse. The observations were performed using symmetric position switching to positions 5' away in azimuth. Since

the OMC-1 ridge runs north-south and the data were not taken at very low elevations, this setup excludes off-source contamination for all species with the possible exception of ¹²CO(6–5).

A zero order baseline (i.e., a constant offset) was subtracted from the DSB data, the offset being defined by the line free part of the spectrum, or, in case of spectra without obvious line free parts, by the lowest points of the spectrum. In very crowded spectra these lowest points were defined by a few channels only. The double sideband data were subsequently deconvolved using a maximum entropy deconvolution method (similar to the one used and described by Sutton et al. 1994), to produce a single sideband spectrum. One advantage compared to CLEAN type deconvolution methods, which had been used for earlier Caltech surveys (Sutton et al. 1985; Blake et al. 1986; Schilke et al. 1997) is

TABLE 2
INTEGRATED LINE INTENSITIES FOR EACH SPECIES

Line	$\int_{K} T_R^* dv$ (K km s ⁻¹)	Line	$\int_{K} T_R^* dv$ (K km s ⁻¹)	Line	$\int_{K} T_R^* dv$ (K km s ⁻¹)
CO	11476.7	so	10055.6	SO ₂	27889.0
CS	1321.4	NO	611.8	CH ₃ OH	27463.6
SiO	913.7	NS	306.2	H ₂ CO	3934.3
SiH	319.9	CN	85.4	H ₂ CS	234.2
HCl	785.8	ССН	88.8	HNCO	1196.5
HCN	2730.3	CH ₃ CN	2830.2	HCOOCH ₃	2466.0
HNC	319.2	H ₂ O	82.7	CH ₃ OCH ₃	3466.0
HCO+	839.4	H ₂ S	1556.4	CH ₃ CH ₂ CN	2939.3
HCS ⁺	8.3	NH_2D	78.2	U	7130.1

NOTE.—The numbers are integrals over all transitions (including torsionally or vibrationally excited ones) and all isotopomers.

that it permits fitting the sideband gain ratios, while these ratios had to be provided for CLEAN. The achieved sensitivity is sufficiently high to securely identify 1 K lines over most of the band.

The main difficulty facing all sideband deconvolution methods is that pointing errors result in line shape and strength changes for the separate measurements of the same line (due to the structured source), which may result in spurious features in the synthesized SSB spectrum. The MEM method seems to produce fewer of these than the CLEAN deconvolution. Using a model spectrum helps suppressing the spurious features. In questionable cases, inspections of the original DSB spectra were performed. Some of the unidentified lines could be spurious features and if so, they are marked in the tables as questionable assignments.

Line blending also makes secure identifications difficult in some cases, this time independent of the observing method. Here line strength predictions based on other lines of the same species provide one way to disentangle features and make reliable assignments. Obviously, some iteration is required to derive a self-consistent result. Still, since one expects the presence of many weak lines just at the detection threshold, the likelihood of significant contamination by these lines is very high for weak lines. This renders their intensities rather uncertain, but not in an easily quantifiable way. Another problem, connected to blending, is the difficulty in determining a reliable baseline, which particularly affects the estimation of the line parameters of weak lines.

3. RESULTS

The intention of this paper is to present the data and some basic results. The full analysis of both physical and chemical parameters will be presented elsewhere (P. Schilke & T. G. Phillips 2001, in preparation), as discussed further below. The analysis of the survey reveals 1064 spectral features consisting of 2032 lines, partially blended (see Fig. 1). The number of U lines is 155 or about 14%. This number is presumably due to the sparseness of laboratory data in this wavelength range and will hopefully be reduced in the future. The spectrum is dominated by CO, CS, SO, SiO, HCN, HCO⁺, H₂CO, SO₂, and CH₃OH, with many lines from very highly excited states. The sensitivity to these highly excited lines is partly due to the small beam size (about 12"), which is well matched to the hot core. Contrary to the findings of Harris et al. (1995), we have no lack of weak or intermediate strength features in our spectrum (see also Fig. 2).

The line identification was mainly performed on the basis of the JPL line catalog⁴ (Pickett et al. 1998) and by the help of various papers by E. Herbst and collaborators (e.g., Anderson, Herbst, & De Lucia 1992, 1993; Pearson et al. 1994, 1995). For SiO, the frequencies from Mollaaghabada et al. (1991) were used. The paper by Xu & Lovas (1997) on methanol allowed the assignment of 23 previously unidentified lines to methanol. For C³³S, the frequencies were calculated using the constants in Bogey, Demuynck, & Destombes (1981).

Classically, one distinguishes four velocity components in Orion spectra (Blake et al. 1987): extended ($v_{\rm LSR}\approx 9$ km s $^{-1}$, $\Delta v\approx 4$ km s $^{-1}$) and compact ridge ($v_{\rm LSR}\approx 8$

km s $^{-1}$, $\Delta v \approx 3$ km s $^{-1}$), plateau $(v_{\rm LSR} \approx 6-10$ km s $^{-1}$, $\Delta v \gtrsim 20$ km s $^{-1}$), and hot core $(v_{\rm LSR} \approx 3-6$ km s $^{-1}$, $\Delta v \approx 5-10$ km s $^{-1}$). These components are emitted from regions with differing physical and chemical conditions, which are described by (Blake et al. 1987). Here we have made no attempt to make multicomponent fits to disentangle the contributions of the various components, but we note if one or several components dominate the spectra of the various species.

It has been predicted that the number of lines per frequency interval should drop as one goes toward higher frequencies, mostly because of the high critical densities required to excite the levels. Indeed, some molecules familiar from lower frequency work are not seen any more, notably the heavier linear rotors like HC₃N, OCS, and symmetric tops like CH₃CCH. Even CH₃CN gets weaker, although many lines are still detectable. As predicted, the detectable molecules include the simple di- and triatomic rotors, but also asymmetric molecules such as H₂CO, CH₃OH, and SO₂. However, even some heavy asymmetric rotors such as CH₃CH₂CN, HCOOCH₃, and CH₃OCH₃ are still seen. The complex spectra of these species results in a line density per frequency interval which is not much different from that at lower frequencies. New molecules, which do not contribute significantly to lower frequency surveys are the light hydrides, such as SiH, HCl, H₂CO, and H₂S, which begin to become prominent.

One goal in analyzing the data is to determine molecular parameters, such as rotation temperatures, column densities etc. Traditionally, the approach here has been by-hand fitting of lines, possibly with multiple velocity components, and the production of rotation diagrams. For surveys with a large number of lines (such as the present one or future FIRST surveys), by-hand fitting is impractical. Furthermore, one often suffers from line blending problems, which makes determining line parameters difficult. A serious drawback of rotation diagram methods is that they may give erroneous results if the lines fitted happen to be optically thick. This can be amended by using isotopomeric lines to determine optical depths and correcting for that, but again this is a time-consuming process, impractical for large amounts of data. Here we give only results from the traditional rotation-diagram technique (Table 1). No optical depth correction has been attempted, except in the case of SO. High optical depths can falsify both column density and rotation temperature. In most cases, high optical depth tends to equalize the brightness temperatures of all lines of a molecule, which results in a flattening of the Boltzmann-plot and to apparent rotation temperatures which are too high. In some cases, notably for SO, where the low-excitation lines have low line strengths, high optical depths can lead to a steepening of the Boltzmann-plot and thus to a lower rotation temperature (Serabyn & Weisstein 1995). In the discussion of the individual molecules we will point out suspected high optical depth effects.

In the future it will be better to approach the problem of uniform analysis in a novel way. Using a molecular line catalog, the complete spectrum of a line survey can be fitted at once, yielding directly the molecular parameters one is interested in. The application of this technique to the present data and the data from Schilke et al. (1997) will be described in Schilke & Phillips (2001, in preparation).

The CO molecule, as expected, is the most abundant species with a total column density of 6×10^{19} cm⁻².

⁴ http://spec.jpl.nasa.gov.

 $\begin{array}{c} \text{TABLE 3} \\ \text{Table of All Observed Features} \end{array}$

y O. (III.)	36.1 1	y O.G.	36.1 1	V O CIII)	N. 1. 1	y O.G.	36.1 1
(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule
607175.1	$H^{13}CO^{+}$	611264.9	CCH	613678.0	CH ₃ OCH ₃	615288.6	CH_3OCHO E
607215.3	$CH_3OH\ E$	611265.4	CCH	613678.0	CH ₃ OCH ₃	615628.7	SO_2
607608.0	SiO	611327.2	CCH	613849.9	CH ₃ OCH ₃	615985.5	$^{33}SO_2$
608021.3	C_2H_5CN	611327.6	CCH	613849.9	CH ₃ OCH ₃	616226.3	U
608021.3	C_2H_5CN	611441.6	H_2S	613852.6	CH ₃ OCH ₃	616322.2	$CH_3OH\ E$
608093.5	CH ₃ OCH ₃	611552.5	SO	613852.7	CH ₃ OCH ₃	616347.3	$SO_2(v_2)$
608096.1	CH ₃ OCH ₃	611580.0	CH_3OH A	613855.4	CH ₃ OCH ₃	616472.4	SO_2
608097.5	CH ₃ OCH ₃	612125.1	¹³ CH ₃ OH <i>E</i>	613855.5	CH ₃ OCH ₃	616603.6	$oldsymbol{U}^-$
608097.5	CH ₃ OCH ₃	612227.4	C_2H_5CN	613855.5	CH ₃ OCH ₃	616637.3	$H_2^{13}CO$
608098.7	CH ₃ OCH ₃	612280.7	CH ₃ OCH ₃	613855.6	CH ₃ OCH ₃	616980.0	$CH_3OH E$
608098.7	CH ₃ OCH ₃	612280.8	CH ₃ OCH ₃	613904.8	¹³ CH ₃ OH <i>E</i>	617097.0	$oldsymbol{U}$
608100.1	CH ₃ OCH ₃	612280.8	CH ₃ OCH ₃	613959.0	CH ₃ OCH ₃	617149.4	U
608101.5	CH ₃ OCH ₃	612280.8	CH ₃ OCH ₃	613962.3	CH ₃ OCH ₃	617180.3	C_2H_5CN
608196.0	CH ₃ CN	612281.2	CH ₃ OCH ₃	613962.4	CH ₃ OCH ₃	617180.5	C_2H_5CN
608196.0	CH ₃ CN	612281.3	CH ₃ OCH ₃	613965.6	CH ₃ OCH ₃	617234.3	C_2H_5CN
608196.0	CH ₃ CN	612281.3	CH ₃ OCH ₃	613965.7	CH ₃ OCH ₃	617234.4	C_2H_5CN
608269.3	U	612281.8	CH ₃ OCH ₃	613965.7	CH ₃ OCH ₃	617346.6	HNCO
608287.2	CH ₃ CN	612442.7	$SO_2(v_2)$	613965.8	CH ₃ OCH ₃	617617.7	C_2H_5CN
608287.2	CH ₃ CN	612479.8	¹³ CH ₃ OH <i>A</i>	613970.7	CH ₃ OCH ₃	617618.8	C_2H_5CN
608287.2	CH ₃ CN	612867.6	HNCO	613970.7	CH ₃ OCH ₃	617627.4	CH ₃ OH E
608389.1	CH ₃ CN	612891.5	$^{13}\mathrm{CH_3OH}~A$	613974.1	CH ₃ OCH ₃	617919.0	C ₂ H ₅ CN
608389.1	CH ₃ CN	612891.5	¹³ CH ₃ OH <i>A</i>	613974.2	CH ₃ OCH ₃	618152.0	SO_2
608389.1	CH ₃ CN	612929.9	CH ₃ OCH ₃	613977.5	CH ₃ OCH ₃	618592.2	CH ₃ OCHO E
608980.0	CH ₃ CN	612929.9	CH ₃ OCH ₃	613977.6	CH ₃ OCH ₃	619156.9	U
608980.0	CH ₃ CN	612930.8	CH ₃ OCH ₃	613977.6	CH ₃ OCH ₃	619251.1	$SO_2(v_2)$
608980.0	CH ₃ CN	612931.0	CH ₃ OCH ₃	613977.8	CH ₃ OCH ₃	619318.7	U
609131.2	U	612931.6	CH ₃ OCH ₃	613978.0	CH ₃ OCH ₃	619340.5	C_2H_5CN
609163.9	$SO_2(v_2)$	612931.9	CH ₃ OCH ₃	613978.0	CH ₃ OCH ₃	619365.1	U
609164.8	C_2H_5CN	612931.9	CH ₃ OCH ₃	613990.0	CH ₃ OH E	623071.7	CH ₃ OH A
609164.8	C_2H_5CN	612932.1	CH ₃ OCH ₃	614019.0	CH_3OCHO A	623145.7	$CH_3OH E$
609506.2	HN ¹³ C	612934.1	¹³ CH ₃ OH <i>A</i>	614090.0	¹³ CH ₃ OH E	623193.2	$CH_3OH E$
609558.4	SO ₂	613076.2	SO ₂	614113.6	SO_2	623262.1	CH ₃ OH E
609703.6	CH ₃ OCH ₃	613303.9	¹³ CH ₃ OH <i>A</i>	614303.1	CH ₃ OCH ₃	623296.4	U
609704.8	CH ₃ OCH ₃	613334.0	¹³ CH ₃ OH <i>A</i>	614303.1	CH ₃ OCH ₃	623316.9	U
609706.0	CH ₃ OCH ₃	613335.0	¹³ CH ₃ OH <i>A</i>	614303.1	CH ₃ OCH ₃	623340.9	CH₃OH E
609706.0	CH ₃ OCH ₃	613338.1	$^{33}SO_2$	614303.1	CH ₃ OCH ₃	623363.4	$HCN(v_2)$
609707.8	CH ₃ OCH ₃	613350.9	¹³ CH ₃ OH <i>A</i>	614360.6	¹³ CH ₃ OH E	623487.1	U
609707.8	CH ₃ OCH ₃	613522.6	¹³ CH ₃ OH E	614628.4	CH ₃ CHO E	623516.7	$SO_2(v_2)$
609709.1	CH ₃ OCH ₃	613590.3	CH ₃ OCH ₃	614741.5	CH ₃ CHO E	623570.6	U
609712.1	CH ₃ OCH ₃	613590.3	CH ₃ OCH ₃	614741.6	$CH_3CHO\ E$	623613.5	SO ₂
609959.6	SO	613592.3	CH ₃ OCH ₃	614756.6	$CH_3CHO\ E$	623644.1	U
610692.6	¹³ CH ₃ OH <i>A</i>	613592.3	CH ₃ OCH ₃	614758.3	$CH_3CHO\ E$	623693.1	$^{33}SO_2$
610692.6	¹³ CH ₃ OH <i>A</i>	613594.5	CH ₃ OCH ₃	614780.0	HNCO	623737.7	$CH_3OH E$
610721.8	¹³ CH ₃ OH <i>E</i>	613594.5	CH ₃ OCH ₃	614780.2	HNCO	623779.8	C ₂ H ₅ CN
610768.0	¹³ CH ₃ OH <i>E</i>	613594.5	CH ₃ OCH ₃	614976.3	HNCO	623784.7	C_2H_5CN
610844.3	CH ₃ OCH ₃	613594.5	CH ₃ OCH ₃	615052.0	HNCO	623848.6	U
610844.3	CH ₃ OCH ₃	613612.0	¹³ CH ₃ OH <i>E</i>	615098.9	HNCO	623964.8	CH ₃ CN
610847.6	CH ₃ OCH ₃	613676.4	SO ₂	615248.6	C ₂ H ₅ CN	623964.8	CH ₃ CN
610850.8	CH ₃ OCH ₃	613678.0	CH ₃ OCH ₃	615249.3	C_2H_5CN C_2H_5CN	623964.8	CH ₃ CN CH ₃ CN
610890.4	¹³ CH ₃ OH <i>E</i>	613678.0	CH ₃ OCH ₃	615276.7	U_2 U_5 U	624023.9	CH_3CN $CH_3OCHO\ E$
010070.7	C113O11 E	0130/0.0	C113OC113	013270.7	<u> </u>	027023.3	CII3OCIIO E

TABLE 3
Continued

ν		ν		ν		ν	-
(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule
-							
624031.4	CH₃OCHO A	624901.3	CH ₃ OCH ₃	626007.7	¹³ CH ₃ OH E	626989.0	CH ₃ OCH ₃
624031.5	CH₃OCHO A	624905.4	CH ₃ OCH ₃	626041.1	CH₃OH A	627004.6	CH₃OH A
624072.8	U G II GN	624905.6	CH ₃ OCH ₃	626041.2	CH ₃ OH A	627013.0	CH₃OH E
624102.0	C ₂ H ₅ CN	624907.8	CH ₃ OCH ₃	626043.5	³³ SO ₂	627015.6	CH ₃ OCH ₃
624103.1	C_2H_5CN	624908.8	CH ₃ OCH ₃	626073.6	CH₃OH A	627018.1	CH ₃ OCH ₃
624108.1	SO ₂	624909.0	CH ₃ OCH ₃	626073.6	CH ₃ OH A	627019.4	CH ₃ OCH ₃
624166.1	CH ₃ CN	624914.5	CH ₃ CN	626075.4	CH₃OH E	627019.5	CH ₃ OCH ₃
624166.2	CH ₃ CN	624914.5	CH ₃ CN	626087.3	SO_2	627020.6	CH ₃ OCH ₃
624166.2	CH ₃ CN	624914.5	CH ₃ CN	626103.1	CH_3OH A	627020.6	CH ₃ OCH ₃
624208.7	HCO ⁺	624914.5	CH ₃ CN	626111.2	$CH_3OH E$	627022.0	CH ₃ OCH ₃
624235.1	$CH_3OH A$	624914.5	CH ₃ CN	626112.2	$CH_3OH E$	627023.3	CH ₃ OCH ₃
624254.5	$^{33}SO_2$	624920.1	SiH	626137.5	$CH_3OH E$	627057.4	NS
624263.6	U	624924.6	SiH	626156.8	$CH_3OH E$	627057.8	CH ₃ CN
624344.1	CH ₃ CN	624926.4	CH ₃ CN	626185.8	$^{33}SO_2$	627057.8	CH ₃ CN
624344.1	CH ₃ CN	624926.5	CH ₃ CN	626191.9	$^{33}SO_2$	627057.8	CH ₃ CN
624344.1	CH ₃ CN	624926.4	CH ₃ CN	626196.5	$CH_3OH\ E$	627058.1	NS
624344.4	SO_2	624932.0	CH ₃ OCH ₃	626306.2	CH ₃ CN	627058.4	NS
624365.4	$CH_3OH\ E$	624932.3	CH ₃ OCH ₃	626306.2	CH ₃ CN	627058.6	NS
624460.9	U	624935.5	CH ₃ OCH ₃	626306.2	CH ₃ CN	627059.0	NS
624498.4	CH ₃ CN	624935.7	CH ₃ OCH ₃	626349.2	$C^{34}S$	627059.6	NS
624498.4	CH ₃ CN	624935.8	SiH	626374.8	$CH_3OH\ E$	627082.5	$^{33}SO_2$
624498.4	CH ₃ CN	624936.4	CH ₃ OCH ₃	626376.4	$CH_3OH E$	627138.6	$^{13}\text{CH}_{3}\text{OH }A$
624551.6	CH_3OCHO A	624964.3	H ³⁷ Cl	626397.8	$CH_3OH\ E$	627170.6	$CH_3OH\ E$
624551.9	CH_3OCHO A	624977.8	H ³⁷ Cl	626451.9	$CH_3OH\ E$	627209.3	CH_3OH A
624552.1	CH_3OCHO A	624988.3	$H^{37}C1$	626474.6	H_2S	627331.1	SO_2
624552.3	CH₃OCHO A	625024.6	C_2H_5CN	626476.7	CH_3OH A	627335.1	$^{33}SO_2$
624582.7	$CH_3OCHO\ E$	625155.7	$CH_3OH E$	626476.7	CH_3OH A	627398.1	CH ₃ CN
624583.0	$CH_3OCHO\ E$	625207.6	$CH_3OH\ E$	626482.6	H_2CS	627398.1	CH ₃ CN
624583.1	$CH_3OCHO\ E$	625063.3	$oldsymbol{U}$	626489.9	$CH_3OH\ E$	627398.1	CH ₃ CN
624583.3	CH_3OCHO E	625072.7	CH_3OH A	626511.9	CH_3OH A	627444.1	CH_3OH A
624626.0	¹³ CH ₃ OH <i>E</i>	625072.7	CH_3OHA	626511.9	CH_3OHA	627444.1	CH_3OHA
624629.1	CH ₃ CN	625294.0	CH_3OCHO E	626555.0	$CH_3OH E$	627476.0	U
624629.1	CH ₃ CN	625294.0	CH_3OCHO E	626555.9	$CH_3OH E$	627528.6	CH_3OH E
624629.1	CH ₃ CN	625294.0	CH_3OCHO E	626608.7	$CH_3OH E$	627558.5	CH_3OH A
624670.3	CH_3OCHO E	625294.0	CH_3OCHO E	626609.0	CH₃OH A	627618.1	CH_3OCHO A
624680.3	CH_3OCHO A	625335.0	U	626626.3	CH_3OHA	627618.1	CH_3OCHO A
624680.3	CH_3OCHO A	625352.5	CH_3OH A	626640.0	$CH_3OH E$	627650.6	$CH_3OH E$
624736.1	CH ₃ CN	625352.5	CH ₃ OH A	626654.1	$CH_3OH E$	627672.7	SiH
624736.1	CH ₃ CN	625383.1	$CH_3OH A$	626673.8	$CH_3OH A$	627689.9	SiH
624736.1	CH ₃ CN	625383.1	$CH_3OH A$	626724.7	$oldsymbol{U}$	627707.6	SiH
624744.3	$^{13}\mathrm{CH_3OH}~A$	625434.0	$CH_3OH E$	626803.8	$CH_3OH E$	627713.0	CH_3OH A
624778.0	U	625510.2	CH₃OH A	626809.5	SO_2	627713.0	CH_3OH A
624819.3	CH ₃ CN	625510.2	CH ₃ OH A	626864.9	CH ₃ OH A	627715.3	SO_2
624819.3	CH ₃ CN	625668.1	U	626865.8	CH ₃ OH A	627750.8	CH ₃ OH E
624819.3	CH ₃ CN	625749.5	CH ₃ OH E	626880.7	$CH_3OH E$	627807.8	$CH_3OH E$
624838.0	CH ₃ OH A	625781.3	$CH_3OH E$	626930.5	CH ₃ OH E	627816.6	CH ₃ OH E
624878.8	CH ₃ CN	625901.6	HCl	626945.2	$CH_3OH E$	627908.1	CH ₃ OH A
624878.8	CH ₃ CN	625918.7	HCl	626989.0	CH ₃ OCH ₃	627908.1	CH ₃ OH A
624878.8	CH ₃ CN	625932.0	HCl	626989.0	CH ₃ OCH ₃	627921.8	$CH_3OH E$
624887.4	SO ₂	625971.5	CH₃OH A	626989.0	CH ₃ OCH ₃	627971.4	CH ₃ OH A
	J U Z	3203.110	311301111	320,0,1011111	2113 2 2113	32.7.11	211,011 11

TABLE 3
Continued

		1		1		T	
ν		ν		ν		ν	
(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule
627971.4	CH ₃ OH A	630952.4	CH ₃ OH A	634766.6	C ₂ H ₅ CN	636274.1	CH ₃ OH A
628021.4	$CH_3OH E$	630952.4	CH ₃ OH A	634782.4	SO ₂	636279.3	CH ₃ OH A
628039.0	CH ₃ OH A	631280.5	$C^{33}S$	634878.8	U	636280.5	CH ₃ OH A
628039.0	$CH_3OH A$	631705.7	H ₂ CO	634898.4	SO,	636291.1	CH ₃ OH A
628051.8	$CH_3OH A$	631751.8	³⁴ SO	635027.5	CH ₃ OCHO A	636299.3	CH ₃ OH A
628093.3	U	632193.3	SO_2	635027.7	CH ₃ OCHO A	636303.3	CH ₃ OH A
628114.1	CH ₃ OH E	632286.6	CH ₃ OCH ₃	635027.8	CH ₃ OCHO A	636311.6	CH ₃ OH A
628115.4	$CH_3OH E$	632286.6	CH ₃ OCH ₃	635027.9	CH ₃ OCHO A	636333.5	CH ₃ OH A
628167.6	$CH_3OH E$	632289.5	CH ₃ OCH ₃	635051.0	$^{33}SO_2$	636337.4	CH ₃ OH A
628187.7	$CH_3OH A$	632292.4	CH ₃ OCH ₃	635085.0	U	636341.6	CH ₃ OH A
628187.7	CH ₃ OH A	632401.4	CH ₃ OCHO A	635121.0	C ₂ H ₅ CN	636363.8	CH_3OH A
628237.8	$CH_3OH E$	632474.6	$^{33}SO_2$	635144.2	$^{33}SO_2$	636365.6	CH ₃ OH A
628251.3	$CH_3OH E$	632505.4	¹³ CH ₃ OH <i>A</i>	635216.9	³⁰ SiO	636393.3	CH ₃ OH A
628318.2	$CH_3OH E$	632571.6	HNCO	635295.3	U	636393.8	CH ₃ OH A
628330.0	$CH_3OH E$	632659.7	³⁴ SO	635324.8	SO ₂	636413.7	CH ₃ OH A
628338.1	$CH_3OH A$	632771.4	¹³ CH ₃ OH <i>A</i>	635389.8	$CH_3OH E$	636419.9	CH ₃ OH A
628338.1	$CH_3OH A$	632852.2	U	635414.4	C_2H_5CN	636420.0	CH ₃ OH A
628341.1	$^{33}SO_2$	633023.5	U	635414.4	C_2H_5CN	636468.7	CH ₃ OCH ₃
628408.9	CH ₃ OH E	633114.5	U	635567.3	¹³ CH ₃ OH <i>E</i>	636468.8	CH ₃ OCH ₃
628445.3	$CH_3OH E$	633147.7	³³ SO ₂	635672.6	CH_3OH A	636470.7	CH ₃ OCH ₃
628469.9	CH_3^3OHA	633156.9	CH ₃ OH E	635697.1	C_2H_5CN	636472.6	CH ₃ OCH ₃
628512.1	$CH_3OH A$	633302.5	³⁴ SO	635697.1	$C_2^2H_5^3CN$	636522.9	$CH_3OH \stackrel{\circ}{A}$
628513.3	$CH_3OH A$	633423.0	CH_3OH A	635749.3	CH ₃ OCHO E	636531.8	CS
628525.0	$CH_3OH A$	633571.9	$CH_3OH E$	635750.4	CH_3OCHO A	636677.6	CH_3OH A
628657.8	CH ₃ OCH ₃	633674.3	$oldsymbol{U}$	635750.8	CH_3OCHO A	636714.0	HNCO
628659.0	CH ₃ OCH ₃	633791.4	SO_2	635750.8	CH_3OCHO A	636714.3	HNCO
628660.1	CH ₃ OCH ₃	633802.5	U	635750.8	CH ₃ OCHO A	636887.0	CH_3OH A
628660.1	CH ₃ OCH ₃	633832.7	U	635750.8	CH₃OCHO A	636946.2	CH_3OCHO E
628662.0	CH ₃ OCH ₃	633860.1	U	635751.1	CH_3OCHO A	636954.6	CH_3OCHO A
628662.0	CH ₃ OCH ₃	633891.1	U	635754.6	CH_3OCHO E	636954.6	CH₃OCHO A
628663.1	CH ₃ OCH ₃	633898.3	U	635874.1	CH_3OH A	636999.8	HNCO
628666.1	CH ₃ OCH ₃	633906.0	$CH_3OCHO\ E$	635904.9	$oldsymbol{U}$	637003.2	C_2H_5CN
628696.5	$CH_3OH\ E$	633926.9	U	635940.9	$CH_3OH\ E$	637003.2	C_2H_5CN
628775.7	CH ₃ OCH ₃	633950.0	$CH_3OCHO\ E$	636013.8	CH_3OH A	637037.7	HNCO
628775.7	CH ₃ OCH ₃	633960.3	CH_3OCHO A	636053.0	$^{33}SO_2$	637056.6	$SO_2(v_2)$
628776.5	CH ₃ OCH ₃	633960.3	CH_3OCHO A	636073.6	CH_3OCHO A	637161.5	CH_3OH A
628777.3	CH ₃ OCH ₃	634005.9	$CH_3OCHO\ E$	636073.7	CH_3OCHO A	637307.3	$^{33}SO_2$
628816.1	$CH_3OH\ E$	634081.1	$oldsymbol{U}$	636105.5	CH_3OH A	637512.3	CH_3OH A
628869.0	CH_3OH A	634118.9	$SO_2(v_2)$	636161.0	CH_3OH A	637684.0	¹³ CH ₃ OH A
628889.6	U	634276.3	SO_2	636190.1	CH_3OH A	637685.0	$^{13}\text{CH}_3\text{OH }A$
629140.5	CH_3OH A	634317.5	$CH_3OCHO\ E$	636190.3	CH_3OH A	637723.3	C_2H_5CN
629321.5	$CH_3OH\ E$	634373.6	CH₃OCHO E	636190.9	CH_3OH A	637797.4	$^{33}SO_2$
629363.8	CH₃OH E	634454.2	³³ SO ₂	636193.4	CH₃OH A	637843.3	CH ₃ OCH ₃
629651.8	CH₃OH E	634470.0	CH ₃ OH E	636196.9	CH₃OH A	637843.6	CH ₃ OCH ₃
629695.9	$oldsymbol{U}$	634508.1	HNC	636198.6	CH ₃ OH A	637844.0	CH ₃ OCH ₃
629790.6	C_2H_5CN	634584.6	U	636202.0	CH₃OH A	637844.0	CH₃OCH₃
629825.3	$CH_3OD E$	634634.0	U	636202.9	CH ₃ OH A	637863.7	U
629921.3	CH₃OH A	634692.0	$SO_2(v_2)$	636208.6	CH₃OH A	637952.1	CH₃OH A
630375.9	U GYV GYV T	634731.6	HNCO	636225.8	CH₃OH A	638084.3	U G H GN
630583.0	CH₃OH E	634766.6	C ₂ H ₅ CN	636248.0	CH₃OH A	638119.3	C ₂ H ₅ CN

TABLE 3
Continued

		<u> </u>		<u> </u>		I	
ν		ν		ν		ν	
(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule
638220.4	U	641056.7	NH ₂ D	643159.6	CH ₃ CN	646216.9	CH ₃ OCHO A
638279.5	CH ₃ OH E	641057.6	NH_2D	643220.8	CH ₃ CN	646216.9	$CH_3OCHO A$
638494.3	$CH_3OH A$	641117.7	¹³ CH ₃ OH <i>A</i>	643220.8	CH ₃ CN	646708.8	C_2H_5CN
638523.5	$CH_3OH A$	641206.5	SO ₂	643220.8	CH ₃ CN	646708.8	C_2H_5CN
638585.5	¹³ CH ₃ OH <i>E</i>	641224.7	U	643257.5	CH ₃ CN	646742.0	U
638635.2	$^{33}SO_2$	641324.5	³³ SO ₂	643257.5	CH ₃ CN	646766.6	CH ₃ OCH ₃
638770.3	SO_2	641361.9	CH ₃ OCH ₃	643257.5	CH ₃ CN	646768.2	CH ₃ OCH ₃
638817.8	CH ₃ OH A	641362.0	CH ₃ OCH ₃	643257.5	CH ₃ CN	646769.8	CH ₃ OCH ₃
639153.6	CH_3OH A	641363.8	CH ₃ OCH ₃	643257.5	CH ₃ CN	646769.8	CH ₃ OCH ₃
639646.1	$^{33}SO_2$	641365.8	CH ₃ OCH ₃	643269.8	CH ₃ CN	646871.6	C_2H_5CN
639650.9	SO ₂	641421.4	H_2CS	643269.8	CH ₃ CN	646927.8	U
639764.7	$H_2^{13}CO$	641825.5	SO_2	643269.8	CH ₃ CN	647084.6	H ₂ CO
639795.1	$^{33}SO_2$	641914.5	SO ₂	643277.6	CH_3OCHO E	647196.2	CH ₃ OCH ₃
639945.4	CH₃OH A	641918.1	$CH_3^2OH E$	643281.8	$CH_3^{3}OCHO$ A	647196.2	CH ₃ OCH ₃
640002.2	$^{33}SO_2$	641993.9	CH ₃ OH A	643412.6	CH ₃ OCH ₃	647199.6	CH ₃ OCH ₃
640116.7	$^{33}SO_2$	642136.1	U	643413.6	C_2H_5CN	647203.0	CH ₃ OCH ₃
640148.3	$^{33}SO_2$	642231.9	SO_2	643417.8	CH ₃ OCH ₃	647319.8	U
640159.1	$^{13}\mathrm{CH_3OH}~A$	642280.6	CH ₃ CN	643423.0	CH ₃ OCH ₃	647395.6	$CH_3OH\ E$
640167.1	$^{33}SO_2$	642280.6	CH ₃ CN	643423.0	CH ₃ OCH ₃	647418.4	U
640167.3	$^{13}\text{CH}_3\text{OH }E$	642280.6	CH ₃ CN	643643.0	CH ₃ OCH ₃	647448.5	SO_2
640175.9	$^{33}SO_2$	642289.1	CH ₃ OCH ₃	643643.7	CH ₃ OCH ₃	647455.1	CH ₃ OCH ₃
640177.3	$^{33}SO_2$	642294.8	CH ₃ OCH ₃	643646.2	CH ₃ OCH ₃	647456.0	CH ₃ OCH ₃
640289.6	$^{13}\mathrm{CH_3OH}~A$	642300.6	CH ₃ OCH ₃	643646.8	CH ₃ OCH ₃	647456.0	CH_3OCH_3
640297.0	$^{13}\mathrm{CH_3OH}~A$	642300.6	CH ₃ OCH ₃	643647.6	CH ₃ OCH ₃	647456.8	CH_3OCH_3
640298.1	¹³ CH ₃ OH A	642552.6	U	643658.3	SO_2	647457.5	CH ₃ OCH ₃
640298.2	¹³ CH ₃ OH <i>A</i>	642670.8	CH ₃ CN	643690.6	CH ₃ OCH ₃	647546.6	³³ SO ₂
640298.6	¹³ CH ₃ OH A	642670.8	CH ₃ CN	643693.0	CH ₃ OCH ₃	647608.3	CH ₃ OCH ₃
640299.6	¹³ CH₃OH A	642670.8	CH₃CN	643694.4	CH ₃ OCH ₃	647609.5	CH ₃ OCH ₃
640304.4	¹³ CH ₃ OH A	642739.0	U	643696.8	CH ₃ OCH ₃	647610.6	CH ₃ OCH ₃
640376.8	¹³ CH ₃ OH A	642762.0	<i>U</i> ²⁹ SiO	643697.1	CH ₃ OCH ₃	647610.6	CH ₃ OCH ₃
640398.5	¹³ CH ₃ OH A	642799.8		643697.5	CH ₃ OCH ₃	647612.4	CH ₃ OCH ₃
640403.3	¹³ CH ₃ OH <i>A</i>	642806.2	SO ₂	643728.3	U	647612.4	CH ₃ OCH ₃
640404.9 640408.5	¹³ CH ₃ OH <i>A</i>	642829.5 642829.5	CH₃CN	644185.8 644377.8	U SO	647613.5 647616.5	CH ₃ OCH ₃
640419.4	¹³ CH₃OH <i>A</i> ¹³ CH₃OH <i>A</i>	642829.5	CH₃CN CH₃CN	645198.0	SO ₂	647717.2	CH ₃ OCH ₃
640435.0	¹³ CH ₃ OH <i>A</i>	642829.8	¹³ CH ₃ OH <i>A</i>	645254.0	SO ₂	648119.7	$U_2(v_2)$
640438.7	¹³ CH ₃ OH <i>A</i>	642837.7	¹³ CH ₃ OH <i>E</i>	645875.3	SO	648134.0	U
640442.3	¹³ CH ₃ OH A	642934.7	CH ₃ OCH ₃	645923.7	CH ₃ OCH ₃	648193.0	DCO+
640466.3	¹³ CH ₃ OH <i>A</i>	642934.7	CH ₃ OCH ₃	645926.1	CH ₃ OCH ₃	648324.2	CH₃OCHO A
640469.5	¹³ CH ₃ OH <i>A</i>	642940.3	CH ₃ OCH ₃	645927.5	CH ₃ OCH ₃	648333.3	$CH_3OCHO\ E$
640494.6	¹³ CH ₃ OH A	642946.0	CH ₃ OCH ₃	645927.5	CH ₃ OCH ₃	648381.4	SO ₂
640497.1	¹³ CH ₃ OH A	642964.0	CH ₃ CN	645928.6	CH ₃ OCH ₃	648737.6	$^{33}SO_2$
640498.4	¹³ CH ₃ OH <i>A</i>	642964.0	CH ₃ CN	645928.6	CH ₃ OCH ₃	648787.4	U
640526.1	¹³ CH ₃ OH <i>A</i>	642964.0	CH ₃ CN	645930.0	CH ₃ OCH ₃	648824.7	C_2H_5CN
640526.6	¹³ CH ₃ OH <i>A</i>	643074.0	CH ₃ CN	645931.3	CH ₃ OCH ₃	648827.3	C ₂ H ₅ CN
640552.4	¹³ CH ₃ OH <i>A</i>	643074.0	CH ₃ CN	646088.7	CH ₃ OCH ₃	648827.4	H ₂ CS
640552.5	¹³ CH ₃ OH <i>A</i>	643074.0	CH ₃ CN	646088.7	CH ₃ OCH ₃	649052.1	\tilde{SO}_2
640886.3	CH_3OH A	643089.7	CH_3 OCHO E	646090.8	CH ₃ OCH ₃	649104.5	U
640942.1	CH_3OCHO E	643159.6	CH ₃ CN	646093.0	CH ₃ OCH ₃	649168.0	$CH_3OCHO\ E$
640981.1	$oldsymbol{U}$	643159.6	CH ₃ CN	646208.5	$CH_3OCHO\ E$	649219.8	$CH_3OCHO\ E$
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TABLE 3
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ν	37.1	v 2 GY	X 1 1	v	X 1 1	v	37.1
(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule
649226.6	CH ₃ OCHO A	652940.1	CH_3OH A	656760.4	SO_2	659885.9	SO_2
649226.6	CH_3OCHO A	652940.1	CH_3OH A	656900.6	$^{33}SO_2$	659898.5	SO_2
649236.3	SO_2	653045.3	$^{13}\mathrm{CH_3OH}~E$	657167.1	C_2H_5CN	659941.8	¹³ CH ₃ OH <i>E</i>
649541.5	$CH_3OH E$	653109.7	SO ₂	657169.2	C_2H_5CN	659944.6	$^{13}\mathrm{CH_{3}^{3}OH}~A$
649912.8	¹³ CH ₃ OH <i>E</i>	653244.9	CH ₃ OCH ₃	657170.8	$ \stackrel{?}{\text{CH}_3} $ OH E	659989.2	C_2H_5CN
649915.2	NH_2D	653244.9	CH ₃ OCH ₃	657222.6	SO_2	660009.5	$CH_3OCHO E$
649916.3	NH ₂ D	653246.6	CH ₃ OCH ₃	657331.5	$^{33}SO_2$	660039.2	¹³ CH ₃ OH E
649916.6	NH ₂ D	653248.3	CH ₃ OCH ₃	657331.5	$^{33}SO_2$	660044.0	¹³ CH ₃ OH <i>E</i>
649916.7	NH_2D	653323.8	C_2H_5CN	657331.6	$^{33}SO_2$	660115.1	¹³ CH ₃ OH <i>A</i>
649917.2	NH ₂ D	653535.3	U	657331.6	$^{33}SO_2$	660115.1	¹³ CH ₃ OH <i>A</i>
649954.6	NH ₂ D	653572.1	U	657406.5	¹³ CH ₃ OH <i>E</i>	660458.7	CH ₃ OCH ₃
649955.1	NH_2D	653598.0	U	657452.7	¹³ CH ₃ OH <i>A</i>	660458.7	CH ₃ OCH ₃
649955.7	NH ₂ D	653711.0	U	657452.7	¹³ CH ₃ OH <i>A</i>	660460.4	CH ₃ OCH ₃
649956.0	NH_2D	653882.9	SO ₂	657721.8	U	660462.1	CH ₃ OCH ₃
649956.1	NH_2D	653931.4	U	657663.5	H ₂ ¹³ CO	660472.7	SO ₂
649956.6	NH_2D	653973.0	H ₂ CO	657868.9	CH_3OCHO A	660593.4	CH ₃ CN
650374.1	H_2S	654030.7	C_2H_5CN	657869.8	$CH_3OCHO A$	660593.5	
650534.0	U	654069.8	$^{33}SO_2$	657885.3		660593.5	CH₃CN CH₃CN
650569.3	C ₂ H ₅ CN	654131.0	U	657933.8	${\overset{{ m SO}}{U}}$	660673.9	¹³ CH ₃ OH <i>E</i>
650573.5	C_2H_5CN C_2H_5CN	654341.8	CH ₃ OH E	658005.2	$H_2O(v_2)$	660785.2	U
650594.6		654396.8	U	658031.5	U	660806.3	
650594.6	C ₂ H ₅ CN		¹³ СН ₃ ОН <i>А</i>	658102.2	-	660806.3	CH₃CN
650680.9	C ₂ H ₅ CN	654420.1 654437.6			CH₃OH A		CH ₃ CN
	${ m C_2H_5CN}$ ${ m ^{33}SO_2}$		SO ₂	658191.6	$CH_3OH E$ $^{33}SO_2$	660806.3	CH ₃ CN
650742.4	$^{33}SO_{2}$	654457.5	H ₂ CO	658217.3		660811.6	¹³ CH ₃ OH <i>E</i>
650742.5	33GO	654457.5	H ₂ CO	658218.5	$SO_2(v_2)$	660866.6	¹³CH₃OH A
650742.6	³³ SO ₂	654519.5	C_2H_5CN	658226.8	SO ₂	660918.3	SO ₂
650742.6	³³ SO ₂	654533.3	U	658411.0	$SO_2(v_2)$	661068.0	¹³ CO
650956.6	SiO	654836.5	H ₂ CO	658466.3	CH₃OCHO A	661157.5	CH ₃ CN
651299.8	SO ₂	654836.5	H ₂ CO	658466.3	CH₃OCHO A	661157.5	CH ₃ CN
651306.2	SO ₂	654969.0	C ₂ H ₅ CN	658541.6	SO ₂	661157.5	CH ₃ CN
651410.1	SO ₂	654969.0	C ₂ H ₅ CN	658553.6	C ¹⁸ O	661190.7	¹³ CH ₃ OH E
651432.6	NO	654993.4	U	658631.7	SO ₂	661295.6	CH ₃ CN
651433.0	NO	655212.9	H ₂ CO	658714.6	U	661295.6	CH ₃ CN
651433.5	NO	655212.9	H ₂ CO	658742.7	³³ SO ₂	661295.7	CH ₃ CN
651493.4	SO	655387.3	CH₃OCHO A	658742.7	³³ SO ₂	661314.4	¹³ CH ₃ OH E
651511.2	H ₂ CS	655388.9	CH ₃ OCHO A	658743.0	³³ SO ₂	661332.4	SO ₂
651535.9	U	655390.0	CH₃OCHO A	658743.0	³³ SO ₂	661389.9	HNCO
651570.9	DCN	655391.5	CH_3OCHO A	658749.3	$SO_2(v_2)$	661408.8	CH ₃ CN
651617.4	CH₃OH E	655444.5	HNCO	658928.3	U	661408.8	CH ₃ CN
651771.4	NO	655641.9	H ₂ CO	658945.7	HNCO	661408.8	CH ₃ CN
651772.9	NO	655645.8	H ₂ CO	658953.1	¹³ CH ₃ OH E	661496.8	CH ₃ CN
651773.1	NO	656075.3	SO_2	658973.5	HNCO	661496.8	CH₃CN
651868.0	¹³ CN	656167.4	H ₂ CO	659064.0	¹³CH₃OH A	661496.8	CH ₃ CN
651868.4	¹³ CN	656168.1	$CH_3OH E$	659188.7	U	661510.8	SO_2
651868.5	¹³ CN	656224.1	$CH_3OCHO\ E$	659338.3	SO ₂	661559.7	CH₃CN
652531.0	C_2H_5CN	656467.3	H ₂ CO	659372.7	¹³ CH ₃ OH A	661559.7	CH ₃ CN
652531.0	C_2H_5CN	656549.5	$^{33}SO_2$	659372.7	¹³ CH ₃ OH A	661559.7	CH ₃ CN
652534.2	C_2H_5CN	656593.1	HNCO	659390.3	$^{13}\mathrm{CH_3OH}~A$	661597.4	CH ₃ CN
652534.2	C_2H_5CN	656656.2	SO_2	659421.0	SO_2	661597.4	CH₃CN
652652.8	$^{33}SO_2$	656724.0	U	659495.2	U	661597.4	CH ₃ CN
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TABLE 3
Continued

ν (ΜΠ-)	M-11-	ν (ΜΠ-)	M-11-	ν (ΜΠ-)	M-11-	ν (ΜΠΤ=)	M-11-
(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule
661597.4	CH ₃ CN	663890.9	CH ₃ CN	666441.8	CH_3OCHO A	669485.2	CH ₃ OCH ₃
661597.4	CH ₃ CN	663890.9	CH ₃ CN	666518.3	C_2H_5CN	669485.6	CH ₃ OCH ₃
661610.0	CH ₃ CN	663951.1	C_2H_5CN	666518.3	C_2H_5CN	669486.8	CH ₃ OCH ₃
661610.0	CH ₃ CN	663951.1	C_2H_5CN	666553.8	CH ₃ OCH ₃	669487.2	CH ₃ OCH ₃
661610.0	CH ₃ CN	664404.6	SO ₂	666554.8	CH ₃ OCH ₃	669488.8	CH ₃ OCH ₃
661668.2	SO ₂	664445.5	CH_3^2OCHO E	666555.9	CH ₃ OCH ₃	669488.8	CH ₃ OCH ₃
661715.1	$SO_2^2(v_2)$	664449.2	CH_3 OCHO A	666555.9	CH ₃ OCH ₃	670011.2	U \tilde{J}
661760.8	¹³ CH ₃ OH <i>E</i>	664449.2	CH ₃ OCHO A	666557.7	CH ₃ OCH ₃	670073.3	NO
661961.0	$^{33}SO_2$	664475.6	$^{33}SO_2$	666557.7	CH ₃ OCH ₃	670073.8	C_2H_5CN
661961.0	$^{33}SO_2^2$	664682.8	C_2H_5CN	666558.8	CH ₃ OCH ₃	670073.8	C_2H_5CN
661961.1	$^{33}SO_2^2$	664682.8	C_2H_5CN	666561.7	CH ₃ OCH ₃	670076.3	NO 3
661961.1	$^{33}SO_{2}^{^{2}}$	664760.3	SO_2	666624.8	C_2H_5CN	670088.2	NO
661962.2	SO_2	664780.9	U	666624.8	C_2H_5CN	670088.8	CH ₃ OCH ₃
661970.5	$^{33}\text{SO}_2$	664814.9	CH ₃ OCH ₃	666706.0	$^{13}CH_{3}OH A$	670089.2	CH ₃ OCH ₃
662087.2	U	664817.2	CH ₃ OCH ₃	666821.2	U	670089.2	CH ₃ OCH ₃
662202.7	SO ₂	664818.5	CH ₃ OCH ₃	666908.7	C ₂ H ₅ CN	670089.6	CH ₃ OCH ₃
662212.0	H ₂ CO	664818.6	CH ₃ OCH ₃	666908.7	C_2H_5CN	670090.1	CH ₃ OCH ₃
662295.8	CH ₃ OH E	664819.4	CH ₃ OCH ₃	667026.8	C_2H_5CN	670090.5	CH ₃ OCH ₃
662319.1	CH ₃ OCH ₃	664819.6	CH ₃ OCH ₃	667026.8	C ₂ H ₅ CN	670091.4	CH ₃ OCH ₃
662320.6	CH ₃ OCH ₃	664820.9	CH ₃ OCH ₃	667147.5	CH ₃ OH A	670091.4	CH ₃ OCH ₃
662321.6	CH ₃ OCH ₃	664822.2	CH ₃ OCH ₃	667148.1	CH ₃ OH A	670091.7	NO
662323.6	CH ₃ OCH ₃	665203.6	SO ₂	667470.6	CH_3OCHO E	670094.6	NO
662404.2	SO_2	665246.8	SO_2	667497.6	$CH_3OH E$	670365.9	SO_2
662414.5	CH ₃ OCH ₃	665393.7	H_2S	667718.9	C ₂ H ₅ CN	670422.6	CH ₃ OH A
662416.1	CH ₃ OCH ₃	665442.4	CH ₃ OH E	667718.9	C_2H_5CN	670499.6	C_2H_5CN
662416.4	CH ₃ OCH ₃	665506.5	C_2H_5CN	667817.3	C_2H_5CN	670499.6	C_2H_5CN
662417.5	CH ₃ OCH ₃	665508.2	C ₂ H ₅ CN	667817.3	C_2H_5CN	670756.5	¹³ CH ₃ OH <i>A</i>
662566.9	SO ₂	665509.0	CH ₃ OCHO A	667949.3	C_2H_5CN	670852.2	U
662697.6	SO_2^2	665560.6	$CH_3^{3}OCHO$ E	667949.3	C_2H_5CN	670894.4	U
662799.4	SO_2^2	665568.6	CH_3 OCHO A	668073.0	CH ₃ OCH ₃	671161.3	CH ₃ OCH ₃
662876.9	SO_2^2	665618.5	$CH_3^{3}OCHO$ E	668073.0	CH ₃ OCH ₃	671161.3	CH ₃ OCH ₃
662933.5	SO_2^2	665705.8	CH ₃ OCHO E	668073.6	CH ₃ OCH ₃	671161.4	CH ₃ OCH ₃
662972.8	SO_2	665710.5	CH ₃ OCH ₃	668074.4	CH ₃ OCH ₃	671161.5	CH ₃ OCH ₃
662997.7	SO_2^2	665711.2	CH ₃ OCH ₃	668077.1	CH ₃ OCH ₃	671162.0	CH ₃ OCH ₃
663011.2	SO_2	665711.7	CH ₃ OCH ₃	668077.8	CH ₃ OCH ₃	671162.0	CH ₃ OCH ₃
663014.3	SO_2^2	665712.4	CH ₃ OCH ₃	668077.8	CH ₃ OCH ₃	671162.0	CH ₃ OCH ₃
663016.0	SO_2^2	665716.0	CH ₃ OCH ₃	668081.2	CH ₃ OCH ₃	671162.6	CH ₃ OCH ₃
663362.7	CH ₃ CN	665717.0	CH ₃ OCH ₃	668118.1	$CH_3OH E$	671191.9	$SO_2(v_2)$
663362.7	CH ₃ CN	665721.3	CH ₃ OCH ₃	668457.8	NS	671208.8	CH ₃ OCH ₃
663362.7	CH ₃ CN	665721.8	CH ₃ OCH ₃	668457.9	NS	671209.0	CH ₃ OCH ₃
663599.0	CH ₃ CN	665814.4	U	668457.9	NS	671209.3	CH ₃ OCH ₃
663599.0	CH ₃ CN	666371.3	C_2H_5CN	668850.6	NS	671209.3	CH ₃ OCH ₃
663599.0	CH ₃ CN	666371.3	C_2H_5CN	668850.7	NS	671312.7	CH ₃ OCH ₃
663639.0	U	666382.0	$^{33}SO_2$	668850.8	NS	671312.7	CH ₃ OCH ₃
663715.3	CH ₃ CN	666382.0	$^{33}SO_2^2$	668854.9	$CH_3OH E$	671313.0	CH ₃ OCH ₃
663715.3	CH ₃ CN	666382.0	$^{33}SO_{2}^{^{2}}$	669407.5	CH ₃ OCH ₃	671313.3	CH ₃ OCH ₃
663715.3	CH ₃ CN	666382.0	$^{33}SO_2^2$	669407.5	CH ₃ OCH ₃	671313.7	CH ₃ OCH ₃
663715.3	CH ₃ CN	666418.3	$C_2H_5^2CN$	669409.0	CH ₃ OCH ₃	671314.0	CH ₃ OCH ₃
663842.8	C ₂ H ₅ CN	666418.3	C_2H_5CN	669410.5	CH ₃ OCH ₃	671314.0	CH ₃ OCH ₃
663890.9	CH₃CN	666441.8	$ \stackrel{?}{\text{CH}_3} $ OCHO A	669485.2	CH ₃ OCH ₃	671314.6	CH ₃ OCH ₃
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TABLE 3
Continued

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ν		ν		ν		ν	
(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule
671408.3	U	674597.5	CH ₃ OH A	676205.5	CH ₃ OH A	678357.9	U
671480.7	CH₃OH A	674618.4	CH ₃ OH A	676205.5	$CH_3OH A$	678417.7	U
671480.7	$CH_3OH A$	674618.4	CH ₃ OH A	676215.0	CH ₃ OH A	678451.9	HNCO
671716.8	U	674658.9	CH ₃ OH E	676250.6	$CH_3OH E$	678546.9	U
671912.5	CH ₃ OH E	674667.8	³³ SO ₂	676269.5	CH ₃ OH A	678675.6	CH ₃ OH E
672184.4	¹³ CH ₃ OH E	674710.4	CH ₃ OH A	676269.5	$CH_3OH A$	678710.3	U
672360.8	CH ₃ OH A	674716.7	CH ₃ OH E	676349.3	CH ₃ OH E	678785.4	CH ₃ OH A
672447.3	U	674743.0	$CH_3OH E$	676362.3	CH ₃ OH E	679121.6	CH ₃ CN
672459.4	CH₃OCHO A	674762.2	CH ₃ OH E	676408.5	CH ₃ OCHO A	679121.6	CH ₃ CN
672564.5	SO_2	674774.0	³⁴ SO	676416.1	$CH_3OH E$	679121.6	CH ₃ CN
672903.4	CH ₃ OCH ₃	674791.5	CH ₃ OH A	676425.5	CH ₃ OH A	679314.7	CH ₃ CN
672903.4	CH_3OCH_3	674812.7	H_2CO	676425.5	CH_3OH A	679314.7	CH ₃ CN
672903.6	$CH_3OH E$	674990.4	$CH_3OH A$	676484.3	SO_2	679314.7	CH ₃ CN
672905.5	CH ₃ OCH ₃	675034.5	$CH_3^{3}OHA$	676494.6	$CH_3^2OH E$	679342.4	CH₃OH A
672907.5	CH ₃ OCH ₃	675035.7	$CH_3^{3}OHA$	676496.8	$CH_3OH E$	679342.4	$CH_3OH A$
672999.0	$SO_2(v_2)$	675038.9	H ₂ CS	676503.2	CH₃OH A	679393.0	C_2H_5CN
673072.1	U	675044.0	$CH_3OH E$	676504.3	$CH_3OH E$	679393.0	C ₂ H ₅ CN
673101.5	$^{33}SO_2$	675047.8	$CH_3OH E$	676585.4	$CH_3OH E$	679482.3	CH₃CN
673229.9	$SO_2(v_2)$	675097.2	$CH_3OH E$	676591.4	$CH_3OH E$	679482.3	CH ₃ CN
673416.1	$CH_3OH E$	675134.8	$CH_3OH E$	676604.2	CH_3OH A	679482.3	CH ₃ CN
673559.8	$oldsymbol{U}$	675135.1	$^{13}\mathrm{CH_3OH}~E$	676604.2	CH ₃ OH A	679554.5	$^{13}\mathrm{CH_3OH}~A$
673657.8	$SO_2(v_2)$	675144.1	$CH_3OH E$	676677.7	$CH_3OH E$	679504.0	CH ₃ CN
673671.1	C_2H_5CN	675176.7	CH_3OH A	676712.5	$CH_3OH\ E$	679504.0	CH ₃ CN
673671.1	C_2H_5CN	675231.6	$CH_3OH\ E$	676749.5	CH_3OH A	679504.0	CH ₃ CN
673675.0	$CH_3OH\ E$	675324.6	³⁴ SO	676823.5	CH_3OH A	679653.1	$CH_3OCHO\ E$
673676.3	CH_3OH A	675347.7	CH_3OH A	676829.5	CH_3OH A	679682.0	$CH_3OCHO\ E$
673676.3	CH_3OH A	675555.3	$^{13}\mathrm{CH_3OH}~A$	676926.5	SO_2	679683.6	CH_3OCHO A
673708.6	$CH_3OCHO\ E$	675566.1	CH_3OCHO A	677013.0	$CH_3OH E$	679683.6	CH_3OCHO A
673745.9	CH_3OH A	675566.1	CH_3OCHO A	677058.0	$SO_2(v_2)$	679740.3	CH ₃ CN
673969.5	$CH_3OH E$	675612.8	CH_3OH A	677112.4	U	679740.3	CH ₃ CN
673991.7	$^{33}SO_2$	675654.3	CH_3OCHO A	677190.6	$CH_3OH\ E$	679740.3	CH ₃ CN
673997.0	$^{33}SO_2$	675678.1	$CH_3OCHO\ E$	677233.3	CH_3OH A	679760.7	U_{-}
673998.0	³⁴ SO	675710.6	$CH_3OCHO\ E$	677273.0	U	679781.0	$C^{33}S$
674009.2	$C^{17}O$	675773.6	$CH_3OH E$	677296.5	$SO_2(v_2)$	679791.2	CH ₃ OCH ₃
674017.3	CH₃OH A	675777.8	CH₃OH A	677380.9	SO ₂	679791.2	CH ₃ OCH ₃
674129.1	CH ₃ OH E	675777.8	CH ₃ OH A	677417.5	CH ₃ OH E	679792.7	CH ₃ OCH ₃
674129.2	CH₃OH A	675887.8	CH₃OH E	677508.3	³⁰ SiO	679794.2	CH ₃ OCH ₃
674129.2	CH ₃ OH A	675963.5	CH₃OH E	677567.5	U	679830.6	CH ₃ CN
674147.8	CH₃OH E	675983.9	U GYL O GYLO	677656.8	$SO_2(v_2)$	679830.6	CH ₃ CN
674162.1	CH ₃ OH E	675997.1	CH ₃ OCHO A	677709.6	CH ₃ OH E	679830.6	CH ₃ CN
674162.4	CH₃OH A	675997.1	CH ₃ OCHO A	677857.1	¹³ CH ₃ OH E	679895.3	CH₃CN
674162.4	CH₃OH A	676010.8	³³ SO ₂	677885.8	$SO_2(v_2)$	679895.3	CH₃CN
674196.5	CH₃OH A	676013.7	$CH_3OH E$	677919.9	U SO (n.)	679895.3	CH₃CN
674201.7	CH₃OH E	676021.4	$SO_2(v_2)$	677961.7	$SO_2(v_2)$	679934.0	CH ₃ CN
674254.3 674284.5	CH₃OH E	676034.6	CH ₃ OH E	677984.8 678005.7	SO ₂	679934.0 679934.0	CH ₃ CN
674447.7	$U \\ \mathrm{SO}_2(v_2)$	676111.8 676111.8	$CH_3OCHO A$ $CH_3OCHO A$	678054.6	$CH_3OCHO A$	679934.0	CH ₃ CN
674469.8	$\mathrm{C^{34}S}$	676130.6	CH₃OCHO A CH₃OH A	678128.2	$SO_2(v_2)$	679934.0	CH ₃ CN
674512.5	CH₃OH E	676130.6	CH ₃ OH A CH ₃ OH A	678237.4	SO_2 HNCO	679947.0	CH₃CN CH₃CN
674597.5	CH ₃ OH <i>L</i> CH ₃ OH <i>A</i>	676138.6	CH ₃ OH E	678252.6	CH ₃ OH E	679947.0	CH ₃ CN CH ₃ CN
014371.3	CII3OII A	070136.0	C113O11 E	070232.0	C113O11 E	017771.0	C113C1V

TABLE 3
Continued

Molecule CH ₃ CN SO ₂ (v ₂) ³ CH ₃ OH A CN	(MHz) 681631.8 681631.8	Molecule	(MHz)	Molecule	(MHz)	Molecule
$^3\mathrm{CO}_2(v_2)$ $^3\mathrm{CH}_3\mathrm{OH}~A$		CII CNI				
3 CH $_3$ OH A	681631.8	CH ₃ CN	684257.3	$CH_3OH\ E$	687696.6	$oldsymbol{U}$
		CH ₃ CN	684296.0	$oldsymbol{U}$	687718.2	U
CN	681631.8	CH ₃ CN	684325.2	CH ₃ OCH ₃	687871.1	$CH_3OCHO\ E$
	681631.8	CH ₃ CN	684327.6	CH ₃ OCH ₃	688113.0	$CH_3OCHO\ E$
CH ₃ OCH ₃	681668.6	CH ₃ CN	684330.0	CH ₃ OCH ₃	688203.5	SO
CH ₃ OCH ₃	681668.6	CH ₃ CN	684330.0	CH ₃ OCH ₃	688275.7	$HC^{13}N$
CH ₃ OCH ₃	681668.6	CH ₃ CN	684381.5	SO_2	688611.6	$^{13}\mathrm{CH_3OH}~E$
CH ₃ OCH ₃	681674.0	SO_2	684429.9			SO
CH ₃ OCH ₃	681789.6	$CH_3OH E$	684675.6	$H_2^{13}CO$		$CH_3OD E$
CH ₃ OCH ₃	681913.1	C_2H_5CN	684840.8	CH ₃ OCH ₃		CH_3OCHO A
CH ₃ OCH ₃	681913.1	C_2H_5CN	684840.8	CH ₃ OCH ₃	688995.3	CH_3OCHO A
CH ₃ OCH ₃	681946.1	C_2H_5CN	684842.1	CH ₃ OCH ₃	688997.2	$CH_3OCHO\ E$
CN	681989.8	CH_3OH A	684843.3	CH ₃ OCH ₃	688997.4	$CH_3OCHO\ E$
CN	682092.2	U	684929.0	SO_2	688999.1	CH₃OCHO A
CN	682271.9	$CH_3OH E$	685336.5	CH ₃ OCH ₃	688999.1	CH_3OCHO A
C ₂ H ₅ CN	682303.3	CH ₃ CN	685336.5	CH ₃ OCH ₃	689006.5	CH₃OCHO A
C ₂ H ₅ CN	682303.3	CH ₃ CN	685340.1	CH ₃ OCH ₃	689007.6	C_2H_5CN
CH₃OCHO A	682303.3	CH ₃ CN	685343.8	CH ₃ OCH ₃	689070.1	CH₃OCHO E
IJ	682370.9	¹³ CH ₃ OH E	685434.7	CS	689120.1	H_2S
CN	682440.6	HCS+	685492.5	CH ₃ OCH ₃	689233.7	$\overline{SO}_2(v_2)$
CN	682583.7	C_2H_5CN	685493.5	CH ₃ OCH ₃	689289.5	U
CN	682660.0		685494.5	CH ₃ OCH ₃	689438.7	SO_2
C ₂ H ₅ CN	682660.0	C_2H_5CN	685494.5	CH ₃ OCH ₃	689522.6	SO_2
C ₂ H ₅ CN	682899.3	CH ₃ OCHO E	685496.3	CH ₃ OCH ₃	690294.3	$CH_3OH E$
INCO	682935.6	CH_3OCHO E	685496.3	CH ₃ OCH ₃	690465.2	SO_2
$SO_2(v_2)$	682938.3	CH_3OCHO A	685497.4	CH ₃ OCH ₃	690551.4	$H^{13}CN$
$\mathrm{SO}_2(v_2)$	682938.3	CH₃OCHO A	685500.2	CH ₃ OCH ₃	690629.5	$^{13}\text{CH}_3\text{OH }E$
INCO	683170.7	$oldsymbol{U}$	685505.1	$CH_3OH\ E$	690672.9	$oldsymbol{U}$
CH_3OH A	683407.5	HNCO	685594.4	²⁹ SiO	691472.9	CO
C ₂ H ₅ CN	683425.2	CH_3OH A	685611.1	C_2H_5CN	691649.4	C_2H_5CN
INCO	683425.2	CH_3OH A	685971.8	CH_3OCHO E	691649.4	C_2H_5CN
INCO	683476.7	U	686016.9	CH_3OCHO E	691815.7	CH_3OCHO E
CH ₃ OCH ₃	683510.6	$oldsymbol{U}$	686019.4	CH_3OCHO A	691841.9	CH ₃ OCHO A
CH ₃ OCH ₃	683571.6	CH₃OCHO A	686019.4	CH_3OCHO A	691841.9	CH₃OCHO A
CH ₃ OCH ₃	683685.7	CH ₃ OCH ₃	686373.6	CH_3OCHO A	691842.6	CH_3OCHO E
CH ₃ OCH ₃	683687.9	CH ₃ OCH ₃	686678.9	$^{33}SO_2$	691991.8	$^{33}SO_2$
CH ₃ OCHO A	683689.0	CH ₃ OCH ₃	686731.3	CH_3OH A	692079.1	$H_{2}^{18}O$
$^{3}SO_{2}$	683689.5	CH ₃ OCH ₃	687030.5	CH ₃ OCH ₃	692200.9	C_2H_5CN
$^{3}SO_{2}$	683689.8	CH ₃ OCH ₃	687030.5	CH ₃ OCH ₃	692200.9	C_2H_5CN
$^{3}SO_{2}$	683690.4	CH ₃ OCH ₃	687031.2	CH ₃ OCH ₃	692204.0	CH ₃ OCHO A
$^{3}SO_{2}$	683691.5	CH ₃ OCH ₃	687032.0	CH ₃ OCH ₃	692204.0	CH ₃ OCHO A
CH ₃ OCH ₃	683692.7	CH ₃ OCH ₃	687034.5	CH ₃ OCH ₃		U
CH ₃ OCH ₃	683749.8	CH₃OH Ĕ	687035.3	CH ₃ OCH ₃	692726.2	$oldsymbol{U}$
CH ₃ OCH ₃	683751.0	¹³ CH ₃ OH A	687035.3	CH ₃ OCH ₃	693269.4	SO
CH ₃ OCH ₃	683762.5	CH_3OCHO E	687038.6	CH ₃ OCH ₃	693420.5	$SO_2(v_2)$
CH ₃ CN	683770.5	$CH_3^{\circ}OCHO$ A	687224.5	CH_3OH A	693469.0	$^{33}SO_2$
CH ₃ CN	683773.4	$CH_3^{\circ}OCHO$ A	687303.4	H_2S	693790.2	U
CH ₃ CN	683780.8	$CH_3^{3}OCHO$ E	687456.3	so	693876.8	$\mathrm{H}^{13}\mathrm{CO}^{+}$
CH ₃ CN	683960.3	-	687544.7	$oldsymbol{U}$	694138.2	SO_2
CH ₃ CN	683960.3		687580.7	$oldsymbol{U}$	694294.5	SiO
CONDITION OF THE CONTRACT OF T	H ₃ OCH ₃ N N N N N N N N N N N N N N N N N N N	H ₃ OCH ₃	H ₃ OCH ₃ 681674.0 SO ₂ H ₃ OCH ₃ 681789.6 CH ₃ OH E H ₃ OCH ₃ 681913.1 C ₂ H ₅ CN H ₃ OCH ₃ 681913.1 C ₂ H ₅ CN H ₃ OCH ₃ 681946.1 C ₂ H ₅ CN N 681989.8 CH ₃ OH A N 682092.2 U N 682271.9 CH ₃ OH E 2H ₅ CN 682303.3 CH ₃ CN 2H ₅ CN 682303.3 CH ₃ CN 4 682370.9 13CH ₃ OH E N 682440.6 HCS ⁺ N 682370.9 13CH ₃ OH E N 682440.6 HCS ⁺ N 682583.7 C ₂ H ₅ CN N 682660.0 C ₂ H ₅ CN N 682660.0 C ₂ H ₅ CN NCO 682938.3 CH ₃ OCHO E NCO 682938.3 CH ₃ OCHO	H3OCH3 681674.0 SO2 684429.9 H3OCH3 681789.6 CH3OH E 684675.6 H3OCH3 681913.1 C2H3CN 684840.8 H3OCH3 681913.1 C2H3CN 684840.8 H3OCH3 681946.1 C2H3CN 684842.1 N 681989.8 CH3OH A 684843.3 N 682092.2 U 684929.0 N 682271.9 CH3OH E 685336.5 2H3CN 682303.3 CH3CN 685340.1 H3OCH A 682303.3 CH3CN 685340.1 H3OCH A 682303.3 CH3CN 685340.1 H3OCH A 682303.3 CH3CN 685340.1 N 682440.6 HCS+ 685492.5 N 682600.0 C2H3CN 685494.5 2H3CN 68260.0 C2H3CN 685494.5 2H3CN 68293.3 CH3CN 685494.5 2H3CN 68260.0 C2H3CN 685494.5 2H3CN 68260.0 C3H3CN 685494.5 2H3CN 682938.3 CH3CCHO A 685496.3 NCO 682935.6 CH3CCHO A 685496.3 NCO 682938.3 CH3CCHO A 685500.2 NCO 683170.7 U 685500.2 H3OH A 683407.5 HNCO 685594.4 2H3CN 68345.2 CH3OCHO A 68591.8 NCO 683476.7 U 68601.9 H3OCH3 68345.2 CH3OCHO A 68591.8 NCO 68345.2 CH3OCHO A 68601.9 H3OCH3 68368.7 CH3OCH3 686731.3 6802 683689.8 CH3OCHO A 68601.9 H3OCH3 683689.0 CH3OCHO A 68601.9 H3OCH3 683689.0 CH3OCH3 686731.3 6802 683689.8 CH3OCH3 687030.5 6802 68369.8 CH3OCH3 687030.5 6802 683689.8 CH3OCH3 687030.5 6802 68369.8 CH3OCH0 A 687030.5 6802 6802 6802 680369.8 CH3OCH0 A 687030.5 6802 6802 6802 6802 6802 6802 6802 6802	H ₃ OCH ₃ 681674.0 SO ₂ 684429.9 U H ₃ OCH ₃ 681789.6 CH ₃ OH E 684675.6 H ₂ ³CO H ₃ OCH ₃ 681913.1 C ₂ H ₂ CN 684840.8 CH ₃ OCH ₃ H ₃ OCH ₃ 681913.1 C ₂ H ₂ CN 684840.8 CH ₃ OCH ₃ H ₃ OCH ₃ 681989.8 CH ₃ OH A 6848421.1 CH ₃ OCH ₃ N 682092.2 U 684929.0 SO ₂ N 682271.9 CH ₃ OH E 685336.5 CH ₃ OCH ₃ 2H ₃ CN 682303.3 CH ₃ CN 685336.5 CH ₃ OCH ₃ 2H ₃ CN 682303.3 CH ₃ CN 685336.5 CH ₃ OCH ₃ 4H ₃ OCHO A 682303.3 CH ₃ CN 685343.7 CS N 682440.6 HCS* 685493.5 CH ₃ OCH ₃ N 682440.6 HCS* 685492.5 CH ₃ OCH ₃ N 682660.0 C ₂ H ₃ CN 685493.5 CH ₃ OCH ₃ 2H ₃ CN 682660.0 C ₂ H ₃ CN 6854945.5 CH ₃	H ₂ OCH ₃

TABLE 3
Continued

		T		1			
ν		ν		ν		ν	
(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule
694494.0	SO ₂	698161.1	CH ₃ CN	701375.9	H ₂ CO	705726.9	¹³ CH ₃ OH <i>E</i>
694726.2	U	698227.4	CH ₃ CN	702072.6	¹³ CH ₃ OH <i>E</i>	705985.8	CH ₃ OCH ₃
695067.0	U	698227.4	CH ₃ CN	702103.7	SO ₂	705985.8	CH ₃ OCH ₃
695119.3	³³ SO ₂	698227.4	CH ₃ CN	702293.3	SO_2	705986.5	CH ₃ OCH ₃
695122.8	$CH_3OCHO E$	698267.2	CH ₃ CN	702417.3	CH ₃ OH E	705987.3	CH ₃ OCH ₃
695156.8	$CH_3OCHO\ E$	698267.2	CH ₃ CN	702479.5	NH ₂ D	705989.8	CH ₃ OCH ₃
695157.7	$CH_3OCHO\ A$	698267.2	CH ₃ CN	702479.9	NH ₂ D	705990.5	CH ₃ OCH ₃
695157.7	CH ₃ OCHO A	698267.2	CH ₃ CN	702480.3	NH ₂ D	705990.5	CH ₃ OCH ₃
695432.5	CH ₃ OCH ₃	698267.2	CH ₃ CN	702480.7	NH_2D	705993.7	CH ₃ OCH ₃
695435.0	CH ₃ OCH ₃	698280.5	CH ₃ CN	702481.2	NH ₂ D	706082.9	CH_3OCH_3 $CH_3OH\ E$
695437.5	CH ₃ OCH ₃	698280.5	CH ₃ CN	702501.7	HNCO	706234.8	SO ₂
695437.5	CH ₃ OCH ₃	698280.5	CH ₃ CN	702502.1	³³ SO ₂	706256.8	¹³ CH ₃ OH <i>A</i>
695525.3	U	698347.5	CH ₃ OCH ₃	702532.5	CH ₃ OCH ₃	706411.6	SO ₂
695632.5	SO_2	698354.6	CH ₃ OCH ₃	702534.5	CH ₃ OCH ₃	706631.7	U
695773.7	U	698494.8	$CH_3OH E$	702535.2	CH ₃ OCH ₃	706924.5	¹³ CH ₃ OH <i>A</i>
695821.7	¹³ CH ₃ CN	698541.5	CCH	702536.0	CH ₃ OCH ₃	707113.2	¹³ CH ₃ OH <i>E</i>
695821.7	¹³ CH ₃ CN	698541.8	CCH	702536.5	CH ₃ OCH ₃	707167.2	¹³ CH ₃ OH <i>E</i>
695821.7	¹³ CH ₃ CN	698603.8	ССН	702537.2	CH ₃ OCH ₃	707173.4	¹³ CH ₃ OH <i>A</i>
696257.5	$CH_3OH E$	698604.2	ССН	702538.0	CH ₃ OCH ₃	707173.4	¹³ CH ₃ OH <i>A</i>
696527.0	SO ₂	698787.6	C_2H_5CN	702539.3	CH ₃ OCH ₃	707244.2	¹³ CH ₃ OH <i>E</i>
696532.2	$HN^{13}C$	698787.6	C_2H_5CN	702718.5	HNCO	707335.1	CH ₃ OCH ₃
696958.7	CH ₃ CN	698930.9	CH ₃ OH A	702831.4	HNCO	707337.6	CH ₃ OCH ₃
696958.8	CH ₃ CN	699071.6	U	702835.4	HNCO	707340.0	CH_3OCH_3
696958.8	CH ₃ CN	699432.6	CH ₃ OCH ₃	702895.7	SO_2	707340.0	CH ₃ OCH ₃
697061.3	U	699432.6	CH ₃ OCH ₃	703891.9	¹³ CH ₃ OH <i>E</i>	707342.2	¹³ CH ₃ OH E
697147.8	CH_3OH E	699433.5	CH ₃ OCH ₃	704192.6	CH_3OCHO E	707368.0	¹³ CH ₃ OH <i>A</i>
697209.2	CH ₃ CN	699434.5	CH ₃ OCH ₃	704193.5	CH_3OCHO A	707368.0	¹³ CH ₃ OH <i>A</i>
697209.2	CH ₃ CN	699437.3	CH ₃ OCH ₃	704193.5	CH_3OCHO A	707442.9	¹³ CH ₃ OH <i>E</i>
697209.2	CH ₃ CN	699437.6	CH ₃ OCH ₃	704195.2	¹³ CH ₃ OH <i>A</i>	707477.5	¹³ CH ₃ OH <i>E</i>
697297.4	U	699438.7	CH ₃ OCH ₃	704270.3	SO_2	707518.2	$^{13}CH_3OH$ A
697433.5	CH ₃ CN	699440.0	CH ₃ OCH ₃	704289.2	CH_3OH A	707606.3	¹³ CH ₃ OH <i>A</i>
697433.5	CH ₃ CN	699453.1	CH ₃ OCH ₃	704422.8	CH ₃ OCH ₃	707609.0	¹³ CH ₃ OH <i>A</i>
697433.6	CH ₃ CN	699453.1	CH ₃ OCH ₃	704423.8	CH ₃ OCH ₃	707614.7	$^{13}\mathrm{CH_3OH}~A$
697500.5	$oldsymbol{U}$	699454.4	CH ₃ OCH ₃	704424.7	CH ₃ OCH ₃	707727.9	C_2H_5CN
697631.7	CH ₃ CN	699455.7	CH ₃ OCH ₃	704424.8	CH ₃ OCH ₃	707727.9	C_2H_5CN
697631.7	CH ₃ CN	699719.8	CH ₃ OCH ₃	704426.6	CH ₃ OCH ₃	707812.7	$^{13}\mathrm{CH_3OH}~E$
697631.7	CH ₃ CN	699720.2	CH ₃ OCH ₃	704426.6	CH ₃ OCH ₃	708010.5	¹³ CH ₃ OH E
697660.6	$oldsymbol{U}$	699721.1	CH ₃ OCH ₃	704427.5	CH ₃ OCH ₃	708055.2	$^{13}\mathrm{CH_3OH}~A$
697761.4	$oldsymbol{U}$	699828.9	$SO_2(v_2)$	704430.3	CH ₃ OCH ₃	708216.1	$U_{\perp \perp}$
697803.6	CH ₃ CN	699873.9	$H_2^{13}CO$	704638.0	$^{33}SO_2$	708266.5	$H_2^{13}CO$
697803.6	CH ₃ CN	699941.5	CH ₃ CN	704638.0	³³ SO ₂	708392.5	SO_2
697803.6	CH ₃ CN	699941.5	CH ₃ CN	704638.1	$^{33}SO_2$	708470.4	H ₂ S
697949.2	CH ₃ CN	699941.5	CH ₃ CN	704638.1	³³ SO ₂	708546.1	¹³ CH ₃ OH <i>E</i>
697949.2	CH ₃ CN	700308.1	HNCO	704919.6	$^{13}\mathrm{CH_3OH}~E$	708654.2	U
697949.2	CH ₃ CN	700312.9	SO ₂	705181.3	CH ₃ OH E	708785.6	$HCN(v_2)$
698068.4	CH ₃ CN	700639.0	C ₂ H ₅ CN	705422.3	HNCO	708813.1	CH₃OH E
698068.4	CH ₃ CN	700639.0	C ₂ H ₅ CN	705457.7	C ₂ H ₅ CN	708877.2	HCN
698068.4	CH ₃ CN	700950.3	HNCO	705464.4	¹³ CH ₃ OH A	708951.1	C ₂ H ₅ CN
698161.1	CH₃CN	701022.3	¹³ CH ₃ OH A	705467.5	CH₃OH E	708951.1	C ₂ H ₅ CN
698161.1	CH ₃ CN	701366.7	$CH_3OH E$	705672.1	SO_2	708979.3	SO_2

TABLE 3
Continued

				l			
ν		ν		ν		ν	
(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule	(MHz)	Molecule
700006.2	7.7	71 47 47 0	NII D	71.60.42.0	II CO	701252.0	CH OCH
709006.3	<i>U</i> ¹³ СН ₃ ОН <i>E</i>	714747.0	NH ₂ D	716943.8	H ₂ CO CH ₃ OCHO <i>A</i>	721353.8 721355.7	CH ₃ OCH ₃
709200.6	-	714747.2	NH ₂ D	717124.3	$CH_3OCHO A$ $CH_3OCHO A$		CH ₃ OCH ₃
709308.2	$egin{array}{c} U \ U \end{array}$		NH ₂ D	717128.9		721356.3	CH ₃ OCH ₃
		714749.1	NH ₂ D	717168.5	CH₃OCHO E	721356.8	CH ₃ OCH ₃
709510.6	SO ₂	714779.5	SO ₂	717177.9	CH₃OCHO E	721357.9 721519.8	CH ₃ OCH ₃
709628.6	C ₂ H ₅ CN	714897.5	NS	717308.2	<i>U</i> ³⁴ SO		CH ₃ OH E
709628.6	C ₂ H ₅ CN	714897.6	NS NG	717348.8		721542.6	CH₃OH A
709817.8	C ₂ H ₅ CN	714897.6	NS	717466.9	CH₃OD A	721542.6	CH₃OH A
710177.6	$SO_2(v_2)$	714971.8	U	717818.9	C ₂ H ₅ CN	721723.7	CH₃OH E
710184.8	C ₂ H ₅ CN	715195.9	U	717818.9	C ₂ H ₅ CN	721793.0	CH₃OH A
710184.8	C ₂ H ₅ CN	714983.9	U CH OH F	717837.8	CH ₃ OCH ₃	721998.1	CH ₃ OH E
710385.8	¹³ CH ₃ OH <i>A</i>	715237.9	CH₃OH E	717837.9	CH ₃ OCH ₃	722039.5	CH₃OH A
710386.5	H ₂ ¹³ CO	715314.6	CH₃OCHO A	717838.9	CH ₃ OCH ₃	722075.4	CH ₃ OH E
710393.4	H ₂ ¹³ CO	715407.1	U	717840.0	CH ₃ OCH ₃	722161.5	HNCO
710517.6	CH ₃ OCHO A	715446.0	U	718158.8	CH ₃ OH A	722316.5	¹³ CH ₃ OH E
710517.6	CH₃OCHO A	715767.2	U GYL GYL	718209.3	¹³ CH ₃ OH E	722544.6	CH₃OH E
710517.8	CH ₃ OCHO E	716121.4	CH₃CN	718305.4	CH ₃ OCH ₃	722575.1	CH ₃ OCH ₃
710572.6	³³ SO ₂	716121.4	CH₃CN	718305.5	CH ₃ OCH ₃	722575.3	CH ₃ OCH ₃
710729.4	H ¹³ NC	716121.4	CH₃CN	718306.4	CH ₃ OCH ₃	722575.4	CH ₃ OCH ₃
710918.7	SO ₂	716167.4	<i>U</i>	718307.4	CH ₃ OCH ₃	722575.4	CH ₃ OCH ₃
711021.0	³³ SO ₂	716208.9	³⁴ SO	718436.1	CH₃OH E	722576.7	C ³⁴ S
711304.5	C ₂ H ₅ CN	716270.8	CH ₃ CN	718474.6	CH ₃ CN	722592.0	CH₃OH E
711304.5	C_2H_5CN	716270.8	CH ₃ CN	718474.6	CH₃CN	722615.8	SO_2
711316.9	<i>U</i>	716270.8	CH ₃ CN	718474.6	CH₃CN	722704.8	SO ₂
711415.1	H ₂ ¹³ CO	716393.1	CH₃CN	718477.3	CH₃CN	722705.7	CH₃OH A
712010.5	SO ₂	716393.1	CH₃CN	718477.3	CH₃CN	722705.7	CH₃OH A
712371.8	$HCN(v_2)$	716393.1	CH₃CN	718477.3	CH₃CN	722742.3	CH ₃ OH E
712527.5	C ₂ H ₅ CN	716488.1	CH₃CN	718618.0	C ₂ H ₅ CN	722789.8	CH₃OH A
712527.5	C ₂ H ₅ CN	716488.1	CH₃CN	718618.0	C_2H_5CN	722804.1	CH ₃ OH E
712673.8	C ₂ H ₅ CN	716488.1	CH₃CN	718765.6	$^{33}SO_{2}$ $^{33}SO_{2}$	722823.7	CH₃OH E
712673.8	C ₂ H ₅ CN	716556.1	CH ₃ CN	718765.9	SO ₂	722849.0	CH ₃ OH E
712723.8	CH ₃ OCH ₃	716556.1	CH₃CN	718771.4	$SO_2(v_2)$	722889.6	CH₃OH A
712723.8	CH ₃ OCH ₃	716556.1	CH₃CN	718830.5	³³ SO ₂	723040.8	CH₃OH E
712746.8	C ₂ H ₅ CN	716596.9	CH₃CN	719178.6	C ₂ H ₅ CN	723192.6	CH₃OH A
712746.8	C ₂ H ₅ CN	716596.9	CH₃CN	719178.6	C ₂ H ₅ CN	723194.4	CH₃OH A
	¹³ CH ₃ OH A	716596.9	CH ₃ CN	719462.5 719557.6	SO ₂	723202.7	CH ₃ OH E
712825.6	SO ₂	716596.9	CH₃CN	719557.6	CH ₃ OCH ₃	723250.8	CH ₃ OH E
	HCO ⁺	716596.9	CH₃CN		CH ₃ OCH ₃		CH ₃ OH E
713409.5	U SO (n.)		CH₃CN	719558.6 719559.7	CH ₃ OCH ₃	723336.7	CH₃OH E
713588.3	$SO_2(v_2)$	716610.1	CH₃CN		CH ₃ OCH ₃	723342.8	CH ₃ OCH ₃
713982.4	CH ₃ OH	716610.1	CH ₃ CN	719664.8	CH ₃ OH A	723343.7	CH ₃ OCH ₃
714223.6	$^{33}SO_2$	716662.4	C ₂ H ₅ CN	719788.7	³⁰ SiO	723344.6	CH ₃ OCH ₃
714375.2	$egin{array}{c} U \ U \end{array}$	716662.4	C ₂ H ₅ CN	719947.9	U	723344.6	CH ₃ OCH ₃
714455.9		716859.9	NH ₂ D	720069.2	CH ₃ OH E	723346.4	CH ₃ OCH ₃
714505.5	NS NC	716860.6	NH ₂ D	720441.3	CH₃OH A	723346.4	CH ₃ OCH ₃
714505.5	NS NC	716861.3	NH ₂ D	720723.6	SO ₂	723350.0	CH ₃ OCH ₃
714505.5	NS	716861.6	NH ₂ D	720812.0	CH₃OH A	723582.7	HNCO
714617.4	U NH D	716861.8	NH ₂ D	721011.1	CH₃OH E	723619.6	CH ₃ OH A
714746.2	NH ₂ D	716862.8	NH ₂ D ³⁴ SO	721352.7	CH ₃ OCH ₃	723761.6	$^{33}SO_2$ $^{33}SO_2$
714746.6	NH ₂ D	716876.1	ა0	721353.1	CH ₃ OCH ₃	723761.6	302

TABLE 3
Continued

v (MHz)	Molecule	v (MHz)	Molecule	v (MHz)	Molecule	v (MHz)	Molecule
723761.7 723761.7 723866.6 724121.5 724345.4 724153.9 724482.1	33SO ₂ 33SO ₂ U CH ₃ OH E CH ₃ OH A CH ₃ OH A CH ₃ OH A CH ₃ OH A	724506.2 724565.0 724595.5 724644.6 724647.3 724648.5	SO ₂ CH ₃ OH E CH ₃ OH E CH ₃ OH A CH ₃ OH A HNCO CH ₃ OH E CH ₃ OH E	724740.8 724761.4 724823.6 724851.5 724855.1 724855.2 724937.8	CH ₃ OH E HNCO CH ₃ OH E CH ₃ OH E CH ₃ OH A CH ₃ OCH ₃ CH ₃ OCH ₃	724938.6 724939.3 724941.7 724942.5 724942.5 724945.6 725013.1	CH ₃ OCH ₃ CH ₃ OH E

Note.—This table is available in machine-readable form in the electronic edition of The Astrophysical Journal.

Unexpectedly, the molecule with the next highest column density is SO, with 2.3×10^{17} cm⁻². The strongest lines of this molecule are very optically thick and it is only through the isotopomer ³⁴SO and by weak transitions of the main isotopomer that the high column density is established. Other surprises are the very high apparent column density of HCOOCH₃ and very high rotation temperatures of HCOOCH₃, CH₃OCH₃, and CH₃CH₂CN. While apparent high rotation temperatures could be faked by high optical depths, high optical depths would lower the column densities, while we determine higher, or in the case of HCOOCH₃ much higher column densities. This together with the fact that we do indeed observe very highly excited lines of all three species (up to 1000 K for HCOOCH₃, 800 K for CH₃OCH₃, and 1600 K for CH₃CH₂CN) point at the interpretation that we do see a probably compact and hot component which has been missed by lower frequency surveys.

The interpretation of the rotation diagrams is not easy, because in particular the very high excitation lines are rather weak and consequently suffer very much from blending. The assignment of each individual feature to the highly excited transitions of HCOOCH₃, CH₃OCH₃, or CH₃CH₂CN would be questionable. Only the large number of detected lines gives confidence in the assignments and the resulting high column densities and temperatures.

As in the 345 GHz survey (Schilke et al. 1997), CO is *not* the dominant coolant of the gas (see Table 2). The combined integrated line intensities in SO₂ and CH₃OH are each more than 2.5 times that of CO, which is very comparable to the value of SO. Other molecules, such as HCN, CH₃CN, H₂CO, HCOOCH₃, CH₃OCH₃, and CH₃CH₂CN also contribute significantly to the cooling. One important coolant, H₂O in the vibrational ground state, is of course unobserved, but it is evident that any models of the thermal balance of Orion-KL have to take as many species as possible into account.

Plotting the number of molecules with $\int T_R^* dv > A$ (Fig. 2) shows that the number of lines with integrated intensities between ≈ 20 and 1000 K km s⁻¹ can be fitted roughly by a power law with a slope of -1. Using this power law in a calculation of the integrated flux would result in an infinite contribution from weak lines. Clearly, the power law is not a good approximation for low flux values and cannot be used to estimate the flux contribution from weak, unresolved lines.

Using the continuum offset of the chopped observations in the line free parts of the DSB spectra, we obtain an offset of 7.3 K = 237 Jy at the CSO, taking the DSB calibration

for the continuum offset into account. The average line emission over the range of the survey is 1.3 K or 41 Jy, giving a total flux of 278 Jy. Wright et al. (1992) measure a peak flux of 375 Jy with the JCMT. Considering the different beam sizes, these two values are comparable to each other for a source size of about 10". Hence, the integrated line emission contributes about 15% to the total emission at this frequency. This value is in agreement with a prediction by Groesbeck (1994), who finds that the line contribution to the total flux in Orion drops from 50% at 800 μ m to 10% at 450 μ m. This estimate takes only the identifiable features into account, including the U lines (see Table 3 for all observed features). A forest of low-level lines and line wings could contribute a pseudo-continuum of unknown strength. Models using Gaussian line shapes and the information available in the JPL catalog (Groesbeck 1994) predict that the contribution of this pseudo-continuum is low, but the existence of non-Gaussian wings could alter the contribution significantly. However, it seems implausible that the relative contribution of the pseudo-continuum to the identifiable line emission changes very much with frequency. Thus, although the absolute values of the line contribution to the apparent continuum will be a lower limit, it seems safe to conclude that the relative importance of line contribution to the continuum drops with frequency between 350 and 650 GHz.

4. COMMENTS ON INDIVIDUAL SPECIES

4.1. Diatomic Molecules without Electronic Angular Momentum (Table 4)

CO.—The carbon monoxide (6–5) lines of 12 CO, 13 CO, 13 CO, and 12 CO(6–5) range from ≈ -80 km s $^{-1}$ to 100 km s $^{-1}$. The peak temperature of the 12 CO(6–5) line is about 150 K, although calibration using the beam efficiency on Jupiter may not be appropriate for the extended component of the emission. The line shows a little dip at the very top, 5 most likely due to self-absorption, although contamination from emission at the OFF-position, while improbable, cannot be ruled out completely. 13 CO(6–5) is also very strong and shows line wings, but not quite as wide. The 18 O(6–5) line is partially blended with an SO₂ line and other lines, so that the line wing contribution is difficult to determine. It is noteworthy that neither the 18 O nor the 13 CO line shows

⁵ The CO line is plotted with a reduced intensity in Fig. 4 to show this lip.

TABLE 4
TABLE OF DIATOMIC MOLECULES

	ν		T_R^{*a}	$\int T_R^* dv^a$	
Molecule	(MHz)	$J_{(F)}$	(K)	$(K \text{ km s}^{-1})$	Notes
со	691473.0	6–5	153.8	9811.6	
¹³ CO	661068.1	6–5	70.1	1445.9	
C ¹⁸ O	658553.6	6–5	25.0	409.7	1
C ¹⁷ O	674009.3	6–5	6.6	109.5	2
CS	636531.8	13–12	29.9	598.3	3
	685434.8	14–13	23.1	535.7	
C ³⁴ S	626349.3	13–12	8.8	108.2	
	674469.9	14–13	3.9	30.7	
	722576.8	15–14	8.1	48.5	4
SiO	607608.0	14–13	10.6	325.6	5
	650956.7	15–14	21.8	609.8	
	694294.6	16–15	13.0	424.4	
²⁹ SiO	642799.8	15–14	2.9	28.5	6
	685594.4	16–15	2.5	19.6	
³⁰ SiO	635217.0	15–14	3.5	66.6	
	677508.3	16–15	2.1	12.5	
	719788.8	17–16	2.5	29.5	
SiH	624920.1	$(3/2)_2$ - $(1/2)_1$	16.9	278.6	7
	624924.6	$(3/2)_1$ - $(1/2)_1$	_	_	
	624935.8	$(3/2)_1$ - $(1/2)_0$	_	_	
	627672.7	$(3/2)_1$ - $(1/2)_1$	5.3	37.2	
	627689.9	$(3/2)_2$ – $(1/2)_1$	_	_	
	627707.6	$(3/2)_1$ - $(1/2)_0$	2.4	14.1	8
HCl	625901.6	$1_2 - 0_2$	15.2	612.7	
	625918.8	$1_3 - 0_2$	15.2	612.7	
	625932.0	$1_{1}-0_{2}$	15.2	612.7	
$H^{37}C1$	624964.4	$1_{2}-0_{2}$	9.6	173.1	
	624977.8	$1_3 - 0_2$	_	_	
	624988.3	1,-02	_	_	

Notes.—(1) Blend with SO_2 at 658541.6. (2) Blend with $^{34}SO_2$ at 673997.1, ^{34}SO at 673998.1 and CH_3OH at 674017.3. (3) Blend with CH_3OH at 636522.9. (4) Blend with CH_3OCH_3 at 722576 and CH_3OH at 722566.7. (5) Uncertain calibration. (6) Blend with SO_2 at 642806.2. (7) Blend with CH_3CN at 624914.6 and 624926.4 and CH_3OCH_3 between 624932.0 and 624936.4. (8) Blend with CH_3OH at 627713.1 and SO_2 at 627715.3. This table is available in machine-readable form in the electronic edition of *The Astrophysical Journal*.

^a Dashes indicate that the line is part of a blend, and intensity and area are given by the values preceding it.

any contribution from the hot core velocity component. The C¹⁷O line is blended with ³⁴SO₂ and CH₃OH lines and cannot be used to determine column densities.

CS.—Carbon monosulfide is detected in the C³²S(13–12) and (14-13) lines, in C³⁴S(13-12), (14-13) and (15-14) and in C³³S(13-12) and (14-13). ¹³CS lines could not be found, but one of the two possible lines is buried under a H₂CO line while the other is in a noisy part of the spectrum. The main isotopomer lines display wings, while lines from the ³⁴S and ³³S isotopomers show mostly a contribution from the hot core. In spite of that, the rotation temperature determined by C³⁴S is anomalously low, only 32 K. However, it is based on only three lines, two of which are situated at the edges of the band, where the atmosphere has a low transmission and the calibration is not reliable. The line ratios of the $C^{32}S/C^{34}S$ (13–12) and (14–13) can be used to determine optical depths of the main isotope. Assuming a ³²S/³⁴S ratio of 23 and equal excitation temperatures, one determines C32S optical depths of 5 and 0.5, respectively. The latter value is more trustworthy, since the calibration of the $C^{34}S(14-13)$ is better than for (13-12). This means that $C^{32}S$ has an optical depth close to unity and that the rotation

temperature and column density determined from the two $C^{32}S$ lines, $T_{\text{rot}} = 127 \text{ K}$ and $N(CS) = 1 \times 10^{15} \text{ cm}^{-2}$, have physical meanings.

SiO.—Silicon monosulfide is found in the (14–13), (15–14), and (16–15) lines of the main isotopomer ²⁸SiO, in the (15–14) and (16–15) lines of the ²⁹SiO and in the (15–14), (16–15), and (17–16) lines of the ³⁰SiO isotopomer. The lines are broad, and all of them, including the isotopomeric lines, only show the plateau component. The isotopomeric ratios suggest that the ²⁸SiO lines also have $\tau \approx 1$. The SiO rotation temperature and column density, determined by the main isotopomer, are $T_{\rm rot} = 110 \pm 50$ K and $N({\rm SiO}) = (5.4 \pm 6.5) \times 10^{14}$ cm⁻², respectively, which is consistent with the isotopomeric results.

SiS.—SiS is not found in this survey. It has previously been identified by Ziurys (1991) in a couple of sources, including Orion. Based on her results, the lines in our frequency range would be too weak to be detectable.

HCl.—The ground state lines of HCl and H³⁷Cl are found. The HCl line has previously been investigated by Schilke, Phillips, & Wang (1995), and H³⁷Cl toward a different position in Orion has been observed by Salez et al. (1996). The line ratio of H³⁷Cl/H³⁵Cl in this survey supports the conclusion that the H³⁵Cl emission is optically thin, as determined by Schilke et al. by analysis of the hyperfine components. A weak feature is present at the frequency of the DCl(2-1) line. Using the working hypothesis that it is indeed DCl and using the same parameters as for HCl ($T_{\text{rot}} = 150 \text{ K}$), one determines a column density of $1 \times 10^{14} \text{ cm}^{-2}$, corresponding to a DCl/HCl ratio of 0.2, which is outside the range of deuterium enhancement usually observed. A lower rotation temperature could only lower this ratio by a factor of 2, which is still much too high. The most probable conclusion is that the feature we observe is not DCl.

SiH.—SiH has two groups of three hyperfine transitions from the ${}^{2}\Pi_{1/2}$ state within our band: one at 624.9 and the other at 627.7 GHz (Brown, Curl, & Evenson 1985). We see evidence for both of them (Fig. 3). At 624.9 GHz, the SiH line is blended with the $CH_3CN(34_0-33_0)$ and (34_1-33_1) lines, and with CH₃OCH₃(16–15) lines. The combined estimated line intensity of these blends accounts for most of the line intensity but does not exclude a contribution of SiH with a strength similar to the 627.7 GHz line. The latter is blended in one hyperfine component with two CH₃OH lines of a line strength similar to the adjacent CH₃OH lines (at $v_{LSR} = -30$ and 30 in Fig. 3), and with a highly excited SO₂ line. The resulting line strength is *lower* than each of the CH₃OH lines individually, which is a puzzle. But the strongest hyperfine component is not blended with any known line, and a trace of the weakest hyperfine component is present. Due to these uncertainties, we regard the identification as only tentative. The only hope of confirming or rejecting the detection in this source are interferometric observations, in case the SiH molecule shows a different spatial distribution than the others. The further discussion adopts the working hypothesis that at least the main hyperfine component of the 627.7 GHz transition is indeed due to SiH and derives the column density and abundances.

The line velocity is 4.7 km s⁻¹, which points to the hot core as the origin. Based on the integrated intensity of the $(J' - J''; F' - F'' = 3/2 - 1/2; 2^+ - 1^-)$ transition at 627.6869 GHz and using a dipole moment of 0.124 D (Lewerenz et al. 1983), one derives a column density in the

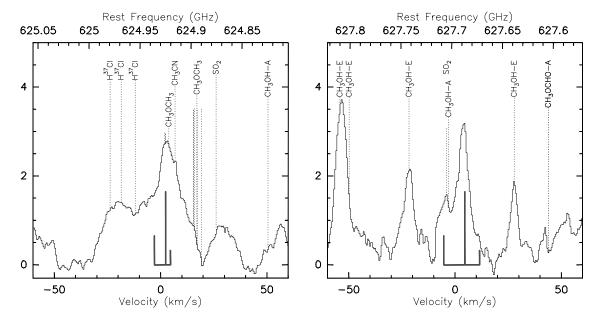


Fig. 3.—Spectra of SiH at 624.9 and 627.7 GHz. The positions and strengths of the hyperfine components are indicated.

upper level of 4.2×10^{15} cm⁻², which translates to a total column density of 9.1×10^{15} cm⁻² for SiH. Here we assumed an excitation temperature of 100 K and we neglected the population in the ${}^2\Pi_{3/2}$ state (200 K above ground). Using a H₂ column density of 3×10^{24} cm⁻² for the hot core (Schilke et al. 1992), the resulting abundance of SiH in the hot core is 3×10^{-9} . In a study of silicon chemistry in hot cores, MacKay (1995) assumes that silicon comes off grains in the form of SiH₄, which is then converted by hydrogen abstraction reactions with atomic hydrogen to SiH₃, SiH₂, SiH, and finally Si. Both Si and SiH can react with molecular oxygen to form SiO. He finds that the relative abundances of the silicon hydrides and SiO depend much on the initial abundance of molecular oxygen in the gas phase. If molecular oxygen is absent, he predicts peak SiH abundances of a few times 10^{-10} , not too far from what we observe. In this model, SiO is absent in the hot core. Indeed, recent interferometer observations of SiO in Orion-KL (Chandler & De Pree 1995; Wright et al. 1995) as well as the 345 GHz survey of Schilke et al. (1997) and the present paper find SiO exclusively in the plateau or outflow component, but not in the hot core.

DF.—We find a feature at the frequency of the DF(1–0) transition. In earlier conference reports of this survey (Schilke 1996) we suggested this as a very tentative detection of the molecule. However, recent measurements failed to reproduce it (D. M. Mehringer & D. C. Lis 1998, private communication).

4.2. Polyatomic Linear Molecules and Ions without Electronic Angular Momentum (Table 5)

HCN and HNC.—The (8–7) lines of HCN, $\rm H^{13}CN$ and $\rm HC^{15}N$ are detected, along with DCN(9–8) and three lines of vibrationally excited HCN in the $v_2=1$ state. The $v_2=2$ transitions have been searched for, but not conclusively found. Although the HCN(7–6) line is in the range of this survey, it unfortunately is placed at a frequency where absorption by the 620.7 GHz $\rm H_2O$ line causes the atmosphere to be completely opaque. The HNC(7–6) and $\rm HN^{13}C(7–6)$ and (8–7) lines are also observed.

 ${
m HC_3N}$ and OCS.—No lines of cyanoacetylene or carbonyl sulfide are detected. This is not surprising, since in this frequency range the levels would have J quantum numbers in the 70s or 50s range, respectively, and level energies of ≈ 1000 K.

HCO⁺.—The (7–6) and (8–7) lines are found in the main isotopomer, and also, tentatively, in the ¹³C isotopomers. The calibration of all four lines is not very reliable, since the main isotopomeric lines are situated close to the two aforementioned gaps in the band, the H¹³CO⁺(7–6) lines at the very edge of our observing band and the (8–7) line in a region of the spectrum with line contaminations from both sidebands. Therefore, not much can be said about optical depths or excitation of the lines.

TABLE 5
TABLE OF POLYATOMIC MOLECULES

Molecule	v (MHz)	$J_{(F)}$	T_R^* (K)	$\int T_R^* dv$ (K km s ⁻¹)	Notes
HCN	708877.2	8–7	48.7	1501.8	
$H^{13}CN$	690551.5	8–7	17.5	456.3	
HC ¹⁵ N	688275.8	8–7	11.5	166.9	
$HCN-v_2 = 1$	623363.5	$7_{1}-6_{1}$	15.7	218.1	
-	708785.7	$8_{1}^{1}-7_{1}^{1}$	13.4	146.6	1
	712371.8	$8_{1}-7_{1}$	11.6	170.3	
DCN	651570.9	9–8	5.5	70.3	
HNC	634508.2	7–6	14.8	248.6	
HN ¹³ C	609506.3	7–6	3.6	17.2	2
	696532.3	8–7	2.8	53.3	3
HCO ⁺	624208.7	7–6	14.3	142.8	
	713341.8	8–7	24.7	577.2	
H ¹³ CO ⁺	607175.1	7–6	2.9	19.8	4
	693876.9	8–7	14.0	99.6	
HCS ⁺	682440.7	16–15	0.9	8.3	5

Notes.—(1) Blend with HCN at 708877.2. (2) Questionable assignment. (3) Blend with SO_2 at 696527.1. (4) Uncertain calibration. (5) Questionable assignment. This table is available in machine-readable form in the electronic edition of *The Astrophysical Journal*.

TABLE 6
TABLE OF SO MOLECULES

	ν		T_R^*	$\int T_R^* dv$	
Molecule	(MHz)	N_J	(K)	$(K km s^{-1})$	Notes
so	609959.7	1010-910	17.2	476.4	
	611552.6	$5_4 - 2_3$	3.7	67.2	
	644377.8	15 ₁₄ -14 ₁₃	33.9	980.7	
	645254.1	15 ₁₅ -14 ₁₄	39.6	1289.7	1
	645875.4	15 ₁₆ -14 ₁₅	39.0	1134.0	
	651493.5	11 ₁₁ -10 ₁₁	4.7	49.5	2
	693269.4	$12_{12} - 11_{12}$	3.2	54.8	
³⁴ SO	631751.8	1514-1413	14.8	189.9	3
	632659.8	15 ₁₅ -14 ₁₄	15.0	305.7	4
	633302.6	15 ₁₆ -14 ₁₅	17.1	412.7	
	675324.6	16 ₁₇ -15 ₁₆	6.6	136.2	
	716209.0	17 ₁₆ -16 ₁₅	4.7	75.5	
	717348.9	17 ₁₈ -16 ₁₇	9.7	145.5	

Notes.—(1) Blend with SO_2 at 645198.0. (2) Blend with H_2CS at 651511.3. (3) Large frequency error. (4) Large frequency error. This table is available in machine-readable form in the electronic edition of *The Astrophysical Journal*.

4.3. Linear Molecules with Electronic Angular Momentum

SO.—Nine lines of SO and nine of ³⁴SO were found in this survey (see Table 6). The lines with high transition strengths from the main isotopomer are very broad and, judged from the isotopomeric ratio, are optically thick. The ³⁴SO lines give a rotation temperature of 39 ± 4 K and a column density for ^{32}SO (assuming a $^{32}S/^{34}S$ ratio of 23) of $(1.4 \pm 1.1) \times 10^{18}$ cm $^{-2}$. This column density seems excessively high. Combining the ³⁴SO lines with the main isotopomer lines and iteratively correcting those for high optical depths lead to much lower column densities and higher rotation temperatures, 64 ± 5 K and (2.3 ± 0.5) \times 10¹⁷ cm⁻². These values are similar to earlier surveys (Serabyn & Weisstein 1995; Sutton et al. 1994). In the rotation plot it is evident that the ³⁴SO line group at 631-633 GHz has line intensities which are too high and are the cause of the low temperature and high column density of the ³⁴SO data. This frequency range is at the edge of the deep 620.7 GHz O₂ absorption line, which may be the cause of a calibration problem.

NO.—Four hyperfine triplets of NO lines are found in this survey (see Table 7). The triplets are blended, so that they cannot be used to determine optical depths based on hyperfine ratios reliably. A rotation diagram gives a rotation temperature of 122 \pm 25 K and a column density of $(1.5 \pm 0.6) \times 10^{17}$ cm⁻².

NS.—Four hyperfine multiplets of NS are found, from J = (27/2)-(25/2) to J = (31/2)-(29/2), partly blended with other lines (see Table 7). The line velocities and widths point to a hot core origin. Due to the blends, the data cannot be used for rotation diagram analyses. This molecule has been extensively studied by McGonagle & Irvine (1997) in Orion and other sources, it also was detected in Orion by Schilke et al. (1997) at 345 GHz.

CN.—Three hyperfine groups of the CN(6–5) line are detected, although the lines are weak (see Table 8). Line widths and velocities are consistent with an extended ridge origin, so the excitation temperature is probably too low to produce strong emission in this frequency range.

CCH.—The (7–6) and (8–7) lines of CCH are found but are very weak (see Table 8). The narrow line width and the line velocity point to an extended ridge origin, similar to CN. The weakness of the lines prohibits a useful rotation diagram analysis.

4.4. Symmetric Rotors

 ${\rm CH_3CN}$.—Methyl cyanide ${\rm CH_3CN}$ is found in a plethora of lines, ranging from the (J=34-33) lines to the (J=39-38) lines in the ground vibrational state and the $v_8=1$ and $v_8=2$ lines (see Table 9). The level energies reach from 500 to 1500 K. No reliable lines of the $^{13}{\rm C}$ isotopomers were found. The rotation diagram shows a large scatter, possibly because of blending or misidentification of some lines. Another possibility is that this molecule cannot be described properly by a single temperature LTE distribution, particularly if one takes the vibrationally excited lines into account. At 345 GHz the ${\rm CH_3CN}$ lines were found to be optically thick.

CH₃CCH.—Lines from methyl acetylene are not detected in this survey. This molecule has been found in earlier surveys to trace the extended ridge, so that it is not surprising that the lines in this band, which have lower levels of more than 650 K above ground, are not seen.

4.5. Inorganic Asymmetric Rotors

 SO_2 .—Sulphur dioxide is observed in the ^{32}S , ^{34}S , and ^{33}S isotopomers, including some vibrationally excited levels (see Table 10). Even ground state levels reaching up to 1400 K above ground are observed. Including all levels, the rotation temperature is 187 ± 4 K with a column density of $(6.0 \pm 0.3) \times 10^{16}$ cm⁻². The $^{34}SO_2$ data give a rotation temperature of 192 ± 12 K with a column density of

TABLE 7
TABLE OF NO AND NS MOLECULES

Molecule	v (MHz)	N_J	T*a (K)	$\int_{0}^{\infty} T_{R}^{*} dv^{a}$ (K km s ⁻¹)	Notes
NO	651432.7	13/2, 13/2–11/2, 11/2e	15.5	236.7	
	651433.0	13/2, 15/2–11/2, 13/2 <i>e</i>	_	_	
	651433.6	13/2, 11/2–11/2, 9/2e	_	_	
NS	714505.5	31/2, 31/2-29/2, 29/2e	2.8	23.5	1
	714505.6	31/2, 29/2–29/2, 25/2e	_	_	
	714505.6	31/2, 33/2-29/2, 31/2e	_	_	
	714897.7	31/2, 31/2–29/2, 29/2f	_	_	

Note.—(1) Questionable assignment. This table is available in machine-readable form in the electronic edition of *The Astrophysical Journal*.

^a Dashes indicate that the line is part of a blend, and intensity and area are given by the values preceding it.

TABLE 8 TABLE OF CN AND C_2H Molecules

Molecule	v (MHz)	N, J, F	T*a (K)	$\int_{K} T_R^* dv^a$ (K km s ⁻¹)	Notes
CN	680027.2	6, 11/2, 11/2-5, 9/2, 11/2	4.1	19.6	
	680264.5	6, 13/2, 13/2–5, 11/2, 11/2	4.1	34.8	
	680264.5	6, 13/2, 13/2-5, 11/2, 11/2	_	_	
	680264.8	6, 13/2, 13/2-5, 11/2, 11/2	_	_	
CCH	611265.0	7, 15/2, 15/2–6, 13/2, 13/2	2.2	29.0	
	611265.4	7, 15/2, 15/2–6, 13/2, 13/2	_	_	
	611327.2	7, 13/2, 13/2–6, 11/2, 11/2	3.8	10.6	
	611327.7	7, 13/2, 13/2–6, 11/2, 11/2	_	_	
	698541.6	8, 17/2, 17/2–7, 15/2, 15/2	2.2	22.1	
	698541.9	8, 17/2, 17/2–7, 15/2, 15/2	_	_	
	698603.9	8, 15/2, 15/2–7, 13/2, 13/2	3.1	27.1	
	698604.2	8, 15/2, 15/2–7, 13/2, 13/2	_		

NOTE.—This table is available in machine-readable form in the electronic edition of *The Astrophysical Journal*.

 $(8.2 \pm 1.1) \times 10^{15}$ cm⁻². The line ratio between $^{32}SO_2$ and $^{34}SO_2$ indicates that the strongest lines of the main isotopomer are moderately optically thick.

 H_2O .—The $(1_{1,0}-1_{0,1})$ line of vibrationally excited water in the $v_2 = 1$ state, 2330 K above ground, is found (see Table 11). This is the vibrationally excited equivalent of the ortho-H₂O ground state line at 556.9 GHz and has been found to show strong maser action in various oxygen-rich red giant and supergiant stars (Menten & Young 1995). A calculation shows that the line strength is consistent with thermal excitation at 200 K, if a water abundance of 10⁻⁵ in the hot core (Gensheimer et al. 1996) is assumed, i.e., a column density of 3×10^{19} cm⁻². A weak line at the frequency of the highly excited H₂¹⁸O(5_{3,2}-4_{4,1}) line (the H₂¹⁸O equivalent of the atmospheric H₂O line at 620.7 GHz) is found. It is weaker than predicted by the above parameters, but it is close to the CO line and has an SO₂ line in the other sideband, so the calibration may be somewhat uncertain. The corresponding $H_2^{17}O$ line is buried under $C^{18}O$, so nothing can be said about its strength. No strong HDO lines are expected in our band.

 $\rm H_2S$.—Six lines of $\rm H_2S$ are found in the survey. The line shapes indicate a plateau component (see Table 11). From the rotation diagram we derive a rotation temperature of $129 \pm 10~\rm K$ and a column density of $(1.2 \pm 0.3) \times 10^{16}~\rm cm^{-2}$. No frequencies for $\rm H_2^{34}S$ lines in this frequency range are known, but the stronger ones should be detectable. No HDS lines are found in our band.

 ${
m NH_2D.--}$ Five hyperfine groups of ${
m NH_2D}$ were found (see Table 11). The rotation diagram gives a rotation temperature of $75\pm11~{
m K}$ and a column density of $(9.7\pm3.8)\times10^{15}~{
m cm}^{-2}$. The rotation temperature agrees, within the error bars, with the one determined by Walmsley et al. (1987), but our column density is much larger, even taking different beam filling factors into account. However, most of the lines we observe are blended, so that we could have easily overestimated the integrated area.

4.6. Organic Asymmetric Rotors

CH₃OH.—Methanol is one of the strongest contributors to the spectrum, displaying 351 lines, including lines in the $v_t = 1$ and $v_t = 2$ torsionally excited states; additionally one observes 126 lines of 13 CH₃OH also including lines in the

 $v_t = 1$ and $v_t = 2$ torsionally excited states (see Table 12). Using all CH₃OH data, a rotation temperature of 303 ± 6 K and a column density of $(5.0 \pm 0.2) \times 10^{16}$ cm⁻² are derived. Using 13 CH₃OH, one determines a rotation temperature of 229 ± 14 K and a column density of $(9.6 \pm 1.1) \times 10^{15}$ cm⁻². This indicates that many 12 CH₃OH lines are optically thick and that the rotation diagram results may need corrections. Some lines at frequencies of deuterated methanol (CH₃OD) have been found, but the lines could not be fitted by a rotation diagram, hence the assignment is questionable.

 $\rm C_2H_5OH$.—Ethanol has been found in Orion-KL by Sutton et al. (1994), Ohishi et al. (1995), and Schilke et al. (1997). Since currently published frequencies only reach up to 600 GHz, nothing can be said about the existence of $\rm C_2H_5OH$ in this frequency range, but it could well take account of a few unidentified lines.

 ${\rm H_2CO.--}$ Formaldehyde has 14 lines in the main isotopomer and seven in the ${\rm H_2^{13}CO}$ isotopomer (see Table 13). The main isotopomer gives a rotation temperature of 190 ± 9 K and a column density of $(3.2 \pm 0.5) \times 10^{15}$ cm⁻². The isotopomeric lines are too weak to give a useful rotation diagram, but the line ratios suggest that the stronger ${\rm H_2^{12}CO}$ lines are optically thick.

H₂CS.—Thioformaldehyde is tentatively found with five transitions. The lines are very weak, which in this frequency range is to be expected on the basis of earlier surveys, so they cannot be used to determine physical parameters of this molecule.

HNCO.—Isocyanic acid is found in 29 line groups consisting of 79 lines, mostly a-type transitions ($\Delta K_a = 0$) of J = 28-27 to J = 33-32 with $K_a = 0...3$, but also a few b-type ($\Delta K_a = 1$) transitions of lower level energies and lower strengths (see Table 14). The rotation diagram yields a rotation temperature of 240 ± 13 K and a column density of $(3.1 \pm 0.5) \times 10^{15}$ cm⁻². As already discussed by Churchwell et al. (1986) and Schilke et al. (1997), the $K_a > 0$ ladders are populated by FIR pumping, so that the rotation temperature is a measure of the dust temperature.

HCOOH.—We find six features at frequencies of formic acid lines, but the detections are rather tentative and the lines are weak, so we do not consider this molecule detected in this survey and produce no table.

a Dashes indicate that the line is part of a blend, and intensity and area are given by the values preceding it.

TABLE 9
TABLE OF CH₃CN MOLECULES

		3			
	ν		T_R^{*a}	$\int T_R^* dv^a$	
Molecule	(MHz)	J_{K}	(K)	$(K \text{ km s}^{-1})$	Notes
	(IVIIIZ)	- K	(11)	(It km 5)	110105
CH ₃ CN	623964.8	$34_9 - 33_9$	4.0	30.9	
3	624166.2	34 ₈ -33 ₈	8.4	72.9	1
	624344.1	$34_{7}-33_{7}$	16.4	203.8	2
	624498.4			154.6	2
		34 ₆ –33 ₆	9.7		2
	624629.1	34 ₅ -33 ₅	16.7	186.7	3
	624736.1	34 ₄ -33 ₄	9.7	120.2	4
	624819.3	$34_{3} - 33_{3}$	6.7	98.1	5
	624878.9	$34_2 - 33_2$	6.7	96.6	
	624914.6	$34_{1} - 33_{1}$	_	_	
	624914.6	34 ₁ -33 ₁	16.9	278.6	
	624926.4	$34_0 - 33_0$			
	642280.7	$35_9 - 34_9$	2.0	26.3	7
	642670.9	35 ₇ -34 ₇	3.7	18.5	,
					0
	642829.6	35 ₆ -34 ₆	2.6	24.7	8
	642964.0	35 ₅ -34 ₅	2.0	16.9	
	643074.1	35 ₄ -34 ₄	2.4	30.4	9
	643159.7	$35_{3} - 34_{3}$	1.9	18.1	
	643220.9	$35_2 - 34_2$	1.8	20.6	
	643257.6	35 ₁ -34 ₁	4.3	59.8	10
	643269.9	$35_0 - 34_0$	4.3	59.8	
	660593.4	36,-35,	0.9	3.2	
	660806.4	36 ₈ -35 ₈	2.2	26.5	11
	661157.6	$36_{6} - 35_{6}$	4.7	35.4	
			6.2		
	661295.7	36 ₅ -35 ₅		51.0	
	661408.8	36 ₄ -35 ₄	2.8	22.4	4.0
	661496.8	$36_3 - 35_3$	6.4	132.6	12
	661559.8	$36_2 - 35_2$	5.7	68.1	
	661597.4	36 ₁ -35 ₁	7.3	128.3	
	661610.1	$36_0 - 35_0$	_	_	
	679121.6	$37_{8} - 36_{8}$	2.7	16.7	
	679314.8	37 ₇ -36 ₇	2.0	8.9	
	679482.4	$37_{6}^{\prime} - 36_{6}^{\prime}$	3.6	21.4	
	679740.3	37 ₄ –36 ₄	4.5	49.5	
	679830.7	$37_{3} - 36_{3}$	4.7	60.9	
	679895.3	37 ₂ -36 ₂	5.7	77.1	
	679934.1	37 ₁ -36 ₁	7.7	140.1	
	679947.0	$37_0 - 36_0$	_	_	
	696958.8	$38_{10} - 37_{10}$	2.1	22.9	
	697209.2	38 ₉ –37 ₉	1.5	11.7	
	697433.6	$38_{8} - 37_{8}$	1.5	9.7	
	697631.8	38 ₇ -37 ₇	2.5	18.2	
	697803.7	$38_6 - 37_6$	5.6	58.8	
	697949.2	38 ₅ -37 ₅	2.9	20.2	
	698068.4	38 ₄ -37 ₄	3.7	26.2	
	698161.2	38 ₃ -37 ₃	6.6	59.8	
	698227.4		5.7	45.4	
	698267.2	38 ₂ -37 ₂	7.5	57.9	
		38 ₁ -37 ₁			
	698280.5	$38_0 - 37_0$	6.0	42.6	
	716121.4	39 ₆ –38 ₆	3.7	19.4	
	716270.9	$39_{5} - 38_{5}$	4.0	35.3	
	716393.1	$39_{4} - 38_{4}$	3.1	7.5	
	716488.2	$39_3 - 38_3$	5.2	22.0	
	716556.2	$39_2 - 38_2$	1.7	11.3	
	716596.9	39,-38,	4.4	33.8	
	716610.2	$39_0^1 - 38_0^1$	3.6	27.7	
$CH_3CN(v_8 = 1)$	608196.1	332-322	3.3	28.5	
3 (-8 *)***	626306.2	$34_{5} - 33_{5}$	3.9	44.1	
	663362.8	$36_{1}-35_{1}$	2.9	32.9	
			2.9		
	681275.8	37 ₆ –36 ₆		35.1	
	681631.8	$37_0 - 36_0$	2.8	33.0	12
	681668.6	37 ₃ -36 ₃	7.8	209.5	13
	699941.6	38 ₄ -37 ₄	2.0	17.5	
	718474.6	$39_2 - 38_2$	4.2	42.8	
	718477.3	$39_{1} - 38_{1}$	_	_	

TABLE 9—Continued

Molecule	v (MHz)	$J_{\it K}$	T*a (K)	$\int_{K} T_R^* dv^a$ (K km s ⁻¹)	Notes
$CH_3CN(v_8 = 2)$	608287.2	33 ₅ -32 ₅	5.9	76.5	
	608389.2	33 ₄ -32 ₄	3.3	30.4	
	627057.8	34 ₂ -33 ₂	5.7	47.3	14
	627398.1	34 ₁ -33 ₄	2.7	17.9	
	663599.1	36 ₄ -35 ₄	1.8	19.2	
	663715.4	36 ₃ -35 ₃	1.4	8.6	
	663715.4	36,-35,	1.4	8.6	
	663890.9	36 ₂ -35 ₂	1.5	14.2	
	679504.0	37 ₁₅ -36 ₁₅	2.9	18.9	
	682303.3	37 ₂ -36 ₂	3.7	51.2	

Notes.—(1) Blend with HCO⁺ at 624208.7. (2) Blend with SO₂ at 624344.5. and CH₃OH at 624365.4. (3) Blend with ¹³CH₃OH at 624626.1. (4) Blend with ¹³CH₃OH at 624744.3. 5. Blend with CH₃OH at 624838.0. (6) Blend with SiH at 624920.1, 624926.4 and 624935.8; and CH₃OCH₃ between 624929.9 and 624936.4. (7) Blend with CH₃OCH₃ at 642289.2. (8) Blend with ¹³CH₃OH at 642829.9. (9) Blend with HCOOCH₃ at 643089.8. (10) Blend with HCOOCH₃ at 643277.7. (11) Blend with ¹³CH₃OH at 660811.6. (12) Blend with SO₂ at 661510.8. (13) Blend with SO₂ at 681674.3. (14) Blend with NS at 627057.4. This table is available in machine-readable form in the electronic edition of *The Astrophysical Journal*.

HCOOCH₃.—Methyl formate is found in this survey with 75 line groups (see Table 15). In contrast to the 345 GHz survey (Schilke et al. 1997), where this species was the dominating molecule in terms of number of lines, it shows far fewer lines in this survey than, e.g., CH₃OH. The excitation energies are clustered in two groups, one ranging from 300–500 K, the other from 800–1500 K. The identifications of lines in the latter cluster is difficult. This is both because the errors of the catalog frequencies are large, and because the high excitation lines are rather weak, so that contamination by other weak lines is probable. This would raise the intensity of the high excitation lines, making the Boltzmann-Plot flatter and producing a rotation temperature which is too high.

Using all data, one gets a rotation temperature of 316 ± 9 K and a column density of $(1.3 \pm 0.1) \times 10^{17}$ cm⁻². Both values are much higher than found in earlier surveys (e.g., Schilke et al. 1997 find a rotation temperature of 89 ± 3 K and column density of $(1.5 \pm 0.1) \times 10^{16}$ cm⁻²). That many lines with excitation energies around 1000 K are found suggests the existence of a second component, a compact source either with high temperatures or with very high optical depths. Because of the above mentioned frequency uncertainty, the velocity of this component cannot be determined. Note that chemically related molecules like CH₃OH or CH₃OCH₃ also display rather high rotation temperatures. For CH₃OH, the isotopomer ¹³CH₃OH gives a lower temperature, indicating that optical depth effects are the cause of the high rotation temperatures. For HCOOCH₃ and CH₃OCH₃, there are no isotopomeric frequencies known and the isotopomeric lines would be too weak anyway. Hence, the question cannot be settled with the current data set.

 CH_3OCH_3 .—Dimethyl ether, like $HCOOCH_3$, has fewer lines than in the 345 GHz survey (see Table 16). A rotation temperature of 360 + 26 K and a column density of

^a Dashes indicate that the line is part of a blend, and intensity and area are given by the values preceding it.

 $\begin{array}{c} \text{TABLE 10} \\ \text{Table of SO}_2 \text{ Molecules} \end{array}$

	Table of SO ₂ Molecules				ν		T*a	$\int T_R^* dv^a$			
Molecule	v (MHz)	I	T*a (K)	$\int T_R^* dv^a$ (K km s ⁻¹)	Notes	Molecule	(MHz)	${J}_{K_p,K_m}$	(K)	(K km s ⁻¹)	Notes
		J_{K_p,K_m}			Notes	SO ₂	663011.3 663014.4	$9_{7,3}-9_{6,4}$	_	_	
SO_2	609558.4	$33_{2,32} - 32_{1,31}$	11.7	215.0			663016.0	$7_{7,1}$ $-7_{6,2}$ $8_{7,1}$ $-8_{6,2}$		_	
	613076.2	$8_{5,3} - 7_{4,4}$	7.8	283.3	1		664404.6	$20_{3,17}$ $-20_{0,20}$	4.8	71.5	
	613676.5 614113.7	$41_{3,39} - 41_{2,40}$	2.6 2.9	27.6 30.0	1		664760.4	44 _{2,42} -44 _{1,43}	2.6	20.4	
	615628.7	$42_{7,35} - 42_{6,36} \\ 12_{9,3} - 13_{8,6}$	1.7	19.9			665203.7	$25_{12,14}^{2,12} - 26_{11,15}^{1,15}$	20.4	747.2	
	616472.4	$29_{3,27}^{-13} - 28_{2,26}^{-13}$	5.6	138.3			665246.9	$16_{4,12}$ $-15_{3,13}$	_	_	
	618152.1	$34_{2,32} - 33_{3,31}$	2.4	42.6			670365.9	$11_{5,7}$ – $10_{4,6}$	17.8	569.3	
	623613.5	$22_{11,11} - 23_{10,14}$	4.6	32.6			672564.5	$17_{4,14} - 16_{3,13}$	13.5	436.0	24
	624108.1	38 _{1,37} -38 _{0,38}	3.9	28.9	2		676484.4	$6_{6,0}$ $-5_{5,1}$	20.0	578.6	21
	624344.5	$35_{1,35} - 34_{0,34}$	16.4	203.8	3		676926.6	$38_{0,38} - 37_{1,37}$	7.7	203.4	
	624887.5	$42_{4,38} - 41_{5,37}$	6.7	89.1	4		677381.0 677984.9	$14_{10,4} - 15_{9,7}$	3.3 3.4	25.7 52.0	22
	626087.3	$14_{4,10} - 13_{3,11}$	21.1	716.6	5		678128.3	$41_{2,40}$ $-41_{1,41}$ $37_{2,36}$ $-36_{1,35}$	6.7	172.1	22
	626809.5	27 _{12,16} -28 _{11,17}	5.9	99.8	6		681674.1	$37_{2,36} - 30_{1,35}$ $35_{3,33} - 34_{2,32}$	7.8	209.5	23
	627331.1	$16_{2,14} - 15_{1,15}$	14.7	248.6	7		684381.5	24 _{12,12} -25 _{11,15}	2.8	46.5	23
	627715.3	$42_{2,40} - 42_{1,41}$	2.4	14.1	8		684929.1	45 _{3,43} -45 _{2,44}	3.7	49.4	
	632193.4	9 _{5,5} -8 _{4,4}	25.6	774.4			689438.7	12 _{5,7} -11 _{4,8}	19.3	665.0	
	633791.5 634276.4	$47_{7,41}$ $-47_{6,42}$ $45_{7,39}$ $-45_{6,40}$	3.2 2.6	25.3 18.9			689522.6	$11_{3,9}^{3,7}-10_{0,10}^{3,0}$	2.7	41.6	
	634782.4	$11_{9,3}^{-13_{6,40}}$	3.3	44.5			690465.2	$52_{8,44} - 52_{7,45}$	1.7	13.5	
	634898.5	$31_{3,29} - 30_{2,28}$	12.9	210.4			694138.2	$38_{1,37} - 37_{2,36}$	5.3	87.3	
	635324.8	$38_{7,31} - 38_{6,32}$	1.4	11.9			694494.0	$39_{1,39} - 38_{0,38}$	6.8	111.3	
	638770.4	$41_{7,35}$ $-41_{6,36}$	2.7	32.5			695632.6	$7_{6,2}-6_{5,1}$	22.8	771.3	
	639651.0	$15_{4,12}^{7,33} - 14_{3,11}^{6,36}$	13.8	362.2	9		696527.1	$13_{10,4}$ $-14_{9,5}$	2.8	53.3	24
	641206.6	$9_{3,7}^{-12} - 8_{0,8}^{-11}$	1.6	16.8			700313.0	$18_{11,7} - 19_{10,10}$	3.1	27.5	25
	641825.5	$36_{0,36} - 35_{1,35}$	4.5	75.5			702103.8	$19_{4,16} - 18_{3,15}$	23.7	659.0	26
	641914.5	$39_{7,33} - 39_{6,34}$	1.6	19.3	10		702293.4 702895.7	$38_{2,36} - 37_{3,35}$	5.2	63.4 75.6	
	642232.0	$36_{7,29} - 36_{6,30}$	1.7	18.7			704270.4	$18_{4,14} - 18_{1,17} \\ 16_{4,12} - 16_{1,15}$	5.8 3.5	28.6	
	642806.2	$21_{11,11}$ – $22_{10,12}$	2.9	28.5	11		705672.1	$18_{4,14} - 17_{3,15}$	25.3	628.1	
	643658.3	$35_{2,34} - 34_{1,33}$	4.0	45.1	10		706234.8	28 _{13,15} -29 _{12,18}	1.7	11.4	
	645198.0	$37_{7,31} - 37_{6,32}$	39.6	1289.7	12		706411.7	$40_{3,37} - 39_{4,36}$	3.2	25.4	
	647448.5 648381.5	$34_{7,27} - 34_{6,28}$	5.8 4.7	73.1 48.7	13		708392.5	$13_{5,9}^{3,37} - 12_{4,8}^{4,36}$	26.6	851.5	
	649052.1	$35_{7,29} - 35_{6,30} $ $38_{3,35} - 37_{4,34}$	1.7	3.1			708979.4	$20_{4,16}$ $-20_{1,19}$	4.5	47.7	
	649236.4	$43_{3,41} - 43_{2,42}$	2.7	14.4			709510.7	$37_{3,35} - 36_{2,34}$	5.7	68.2	
	651299.9	$10_{5,5} - 9_{4,6}$	32.0	888.1			710918.7	$14_{4,10} - 14_{1,13}$	3.4	27.9	
	651306.3	33 _{7,27} –33 _{6,28}	_	_			712010.6	$40_{0,40}$ – $39_{1,39}$	7.2	90.5	
	651410.1	$32_{7,25}^{7,27} - 32_{6,26}^{6,28}$	11.3	118.5			712825.7	$39_{2,38} - 38_{1,37}$	6.3	64.4	
	653109.7	$18_{3,15}^{7,23} - 17_{2,16}^{7,23}$	20.7	475.6			714779.6	$8_{6,2}$ $-7_{5,3}$	7.9	102.6	
	653882.9	$31_{7,25} - 31_{6,26}$	5.5	63.5			719462.6	$17_{11,7} - 18_{10,8}$	1.7	7.2	
	654437.6	$30_{7,23} - 30_{6,24}$	6.6	64.0			720723.7	$20_{3,17} - 19_{2,18}$	13.2	462.8	
	656075.4	$29_{7,23}$ – $29_{6,24}$	4.0	55.4			722615.9 722704.9	$18_{2,16}$ $-17_{1,17}$	9.4 2.7	139.2 11.9	27
	656656.3	$33_{3,31} - 32_{2,30}$	9.9	144.1			724506.2	$22_{12,10} - 23_{11,13} \\ 22_{4,18} - 22_{1,21}$	4.5	29.7	21
	656760.5	$28_{7,21}$ $-28_{6,22}$	7.2	124.8		$^{34}\mathrm{SO}_2\dots$	613338.1	$14_{4,10} - 13_{3,11}$	5.1	38.1	28
	657222.7	$48_{3,45} - 48_{2,46}$	3.8	69.3	14	50 ₂	615985.5	9 _{5,5} -8 _{4,4}	4.3	104.2	
	657885.4	$27_{7,21}$ $-27_{6,22}$	5.4	49.5	15		623693.1	$46_{3,43} - 46_{2,44}$	10.3	173.7	29
	658226.8 658541.6	$15_{10,6} - 16_{9,7} \\ 26_{7,19} - 26_{6,20}$	5.2 25.0	76.2 409.7	15 16		624254.5	$35_{7,29} - 35_{6,30}$	19.7	234.5	30
	658631.7	$36_{1,35} - 35_{2,34}$	10.2	115.5	10		626043.6	$15_{4,12}$ $-14_{3,11}$	21.1	716.6	31
	659338.3	$25_{7,19}^{1,35}$ $25_{2,34}^{2,34}$	8.5	115.5			626185.8	$25_{12,14} - 26_{11,15}$	5.4	36.8	32
	659421.0	$37_{1,37} - 36_{0,36}$	9.9	154.3			626191.9	$20_{11,9}$ – $21_{10,12}$	_	_	
	659886.0	$40_{1,39}^{1,37} - 40_{0,40}^{1,39}$	10.1	227.8			627082.6	$32_{7,25} - 32_{6,26}$	1.4	9.6	
	659898.6	$24_{7,17}^{1,33} - 24_{6,18}^{6,48}$	_	_			627335.2	$33_{7,27} - 33_{6,28}$	14.7	248.6	33
	660472.7	$23_{7,17}$ $-23_{6,18}$	8.6	87.6			628341.2	$16_{2,14} - 15_{1,15}$	17.1	293.3	34
	660918.3	$22_{7,15}$ $-22_{6,16}$	14.0	334.2			632474.6	29 _{7,23} –29 _{6,24}	3.0	34.5	
	661332.5	$21_{7,15}$ – $21_{6,16}$	11.5	181.0			633147.8 634454.2	$28_{7,21}$ $-28_{6,22}$	4.1 5.4	29.3 64.5	
	661510.8	$36_{2,34} - 35_{3,33}$	6.4	132.6	17		635051.1	$27_{7,21}$ $-27_{6,22}$	5.5	70.3	
	661668.3	$20_{7,13}$ $-20_{6,14}$	11.9	294.8	4.0		635144.2	$10_{5,5} - 9_{4,6} \\ 26_{7,19} - 26_{6,20}$	1.9	17.9	
	661962.2	$19_{7,13} - 19_{6,14}$	11.2	287.9	18		636053.0	$25_{7,19}$ – $25_{6,20}$ $25_{7,19}$ – $25_{6,20}$	1.1	5.1	
	662202.7	$18_{7,11} - 18_{6,12}$	20.1	431.3	19		637307.3	$23_{7,17}$ $23_{6,20}$ $23_{7,17}$ $-23_{6,18}$	1.1	9.8	
	662404.3	17 _{7,11} –17 _{6,12}	15.7	459.7 470.3	20		637797.4	$22_{7,15}^{7,17}$ $23_{6,18}^{6,18}$ $22_{7,15}^{7,15}$ $-22_{6,16}^{7,15}$	2.3	44.9	
	662566.9 662697.6	$16_{7,9}$ $-16_{6,10}$	17.0	479.3 354.5			638635.2	$20_{7,13}^{-13}$ $-20_{6,14}^{-13}$	1.2	7.4	
	662799.5	$15_{7,9} - 15_{6,10}$	12.7 12.5	300.0			639646.1	$16_{7,9}^{7,13} - 16_{6,10}^{6,14}$	13.8	362.2	35
	662877.0	$14_{7,7}-14_{6,8}$ $13_{7,7}-13_{6,8}$	14.1	346.2			639795.1	$15_{7,9}^{7,9} - 15_{6,10}^{6,10}$	1.4	12.0	
	662933.6	$13_{7,7} - 13_{6,8}$ $12_{7,5} - 12_{6,6}$	17.2	328.4			640002.2	$13_{7,7}$ $-13_{6,8}$	1.1	6.4	
	662972.8	$11_{7,5}^{-12_{6,6}}$ $11_{7,5}^{-11_{6,6}}$	20.8	313.1			640116.8	$11_{7,5}$ $-11_{6,6}$	2.5	22.3	
	662997.7	$10_{7,3}^{-1}$ $10_{6,4}^{-6}$	20.9	454.1			640148.4	$10_{7,3}$ – $10_{6,4}$	2.1	47.0	36
		/,36,4									

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TABLE 10-Continued

	ν		T_R^{*a}	$\int T_{R}^{st} dv^{ m a}$	
Molecule	(MHz)	${J}_{K_p,K_m}$	(K)	$(K \text{ km s}^{-1})$	Notes
³⁴ SO ₂	640167.1	9 _{7,3} –9 _{6,4}	_	_	
_	640175.9	8 _{7.1} -8 _{6.2}		_	
	640177.3	$7_{7,1}^{-7} - 7_{6,2}^{-7}$		_	
	641324.6	$39_{2,38} - 39_{1,39}$	1.6	28.7	
	647546.7	$33_{3,31} - 32_{2,30}$	2.7	37.9	
	648737.6	$18_{3,15} - 17_{2,16}$	4.8	89.5	
	652652.9	$16_{4,12} - 15_{3,13}$	6.0	91.4	
	654069.9	$11_{5,7}^{-10}$	4.9	55.4	
	656549.5	$37_{1,37} - 36_{0,36}$	1.5	9.7	
	656900.7	$6_{6,0}$ $-5_{5,1}$	4.2	49.2	
	658217.4	$17_{4,14} - 16_{3,13}$	5.2	76.2	37
	661970.5	$38_{3,35} - 37_{4,34}$	11.2	287.9	38
	664475.7	$18_{11,7} - 19_{10,10}$	3.4	22.3	
	673101.5	$12_{5,7}$ $-11_{4,8}$	5.0	104.8	
	673991.8	$38_{0,38} - 37_{1,37}$	6.6	109.5	39
	673997.1	$35_{3,33} - 34_{2,32}$	_		
	674667.9	$11_{3,9} - 10_{0,10}$	3.8	50.0	40
	676010.8	$7_{6,2}-6_{5,1}$	5.5	47.8	41
	686678.9	$19_{4,16} - 18_{3,15}$	3.6	43.9	
	691991.8	$13_{5,9} - 12_{4,8}$	2.3	37.5	
	693469.0	$18_{4,14} - 17_{3,15}$	5.6	135.3	
	695119.3	$8_{6,2} - 7_{5,3}$	7.1	111.1	42
	702502.1	$46_{2,44} - 46_{1,45}$	2.6	26.6	43
	710572.6	$21_{4,18} - 20_{3,17}$	3.8	57.7	
	711021.1	$14_{5.9} - 13_{4.10}$	5.2	57.4	
	714223.7	$9_{6,4} - 8_{5,3}$	2.7	27.0	
	718830.6	$20_{3,17}$ – $19_{2,18}$	1.9	21.8	
$^{33}SO_2 \dots$	650742.4	$16_{7,9}$ $-16_{6,10}$	4.0	71.4	
	657331.6	$36_{1,35} - 35_{2,34}$	3.4	46.1	44
	658742.8	$16_{4,12} - 15_{3,13}$	6.9	92.5	45
	661961.0	$11_{5,7}$ $-10_{4,6}$	11.2	287.9	46
	666382.0	$6_{6,0}$ $-5_{5,1}$	5.3	173.7	47
	681012.8	$12_{5,7}$ – $11_{4,8}$	2.8	24.9	
	704638.1	$8_{6,2}$ $-7_{5,3}$	1.9	29.9	
	718765.7	$21_{4,18}$ $-20_{3,17}$	3.0	35.0	48
	723761.6	$9_{6,4} - 8_{5,3}$	3.8	54.8	

Notes.—(1) Blend with CH₃OCH₃ at 613678.0. (2) Blend with CH₃CH₂CN at 624102.1. (3) Blend with CH₃CN at 624344.1 and CH₃OH at 624365.4. (4) Blend with CH $_3$ CN at 624878.9 and CH $_3$ OCH $_3$ between 624901.3 and 624909.1. (5) Blend with CH $_3$ OH at 626041.2 and 34 SO $_2$ at 626043.6. (6) Blend with CH_3OH at 626803.9. (7) Blend with $^{34}SO_2$ at 627335.2. (8) Blend with SiH at 627689.9 and CH₃OH at 627713.1. (9) Blend with $^{34}SO_2$ at 639646.1. (10) Blend with CH₃OH at 641918.1. (11) Blend with ²⁹SiO at 642799.8. (12) Blend with SO at 645254.1. (13) Blend with CH_3OCH_3 at 647455.2. (14) Too strong, possible blend with U line. (15) Blend with $^{34}SO_2$ at 658217.4. (16) Blend with ^{C18}O at 658553.6. (17) Blend with CH₃CN at 661496.8. (18) Blend with SO₂ at 661970.5 and ³³SO₂ at 662962.0. (19) Blend with H₂CO at 662212.1. (20) Blend with CH₃OCH₃ at 662414.6. (21) Blend with CH₃OH at 676494.7. (22) Blend with $SO_2-v_2 = 1$ at 677961.8. (23) Blend with CH₃CN at 681668.6. (24) Blend with HN¹³C at 696532.3. (25) Blend with HNCO at 700308.1. (26) Blend with ¹³CH₃OH at 702072.6. (27) Blend with CH₃OH at 722705.8. (28) Blend with ¹³CH₃OH at 613334.1 and 613335.0. (29) Questionable assignment. (30) Blend with CH₃OH at 624235.2. (31) Blend with CH₃OH at 626041.2 and SO₂ at 626087.3. (32) Blend with CH₃OH at 626196.5 (33) Blend with SO₂ at 627331.1. (34) Blend with CH₃OH at 628338.2. (35) Blend with SO₂ at 649651.0. (36) Blend with ¹³CH₃OH at 640159.1 and 640167.3. (37) Blend with SO₂ at 658226.8. (38) Blend with ³⁴SO₂ at 661962.2 and 33SO₂ at 662962.0. (39) Blend with 34SO at 673998.1. (40) Blend with CH₃OH at 647658.9. (41) Blend with CH₃OH at 676013.8 and $SO_2-v_2 = 1$ at 676021.4. (42) Blend with HCOOCH₃ at 695122.9. (43) Blend with HNCO at 705502.2. (44) Too strong, possible blend with U line. (45) Blend with SO_2 - v_2 = 1 at 658749.4. (46) Blend with SO_2 at 661970.5 and $^{34}SO_2$ at 661962.2. (47) Blend with CH_3CH_2CN at 666371.4. (48) Blend with $SO_2^2 - v_2 = 2$ at 718771.4. This table is available in machinereadable form in the electronic edition of The Astrophysical Journal.

^a Dashes indicate that the line is part of a blend, and intensity and area are given by the values preceding it.

 $(2.8\pm0.4)\times10^{16}~{\rm cm^{-2}}$ are found. Like HCOOCH₃, it also shows a much higher rotation temperature than found in previous surveys, but unlike HCOOCH₃, the column density derived here is consistent with earlier work [e.g., Schilke et al. 1997 find a rotation temperature of $89\pm5~{\rm K}$ and column density of $(1.8\pm0.2)\times10^{16}~{\rm cm^{-2}}$]. See the discussion for HCOOCH₃ for the possible causes of these high rotation temperatures.

CH₂CHCN.—Vinyl cyanide is not observed in this survey. Extrapolations using the results of Schilke et al. (1997) predict maximum line intensities of about 1–2 K in the present band, which are at our detection limit.

CH₃CH₂CN.—Ethyl cyanide shows 81 line groups up to almost 1600 K above ground (see Table 17). The rotation diagram fit gives a rotation temperature of 239 \pm 4 K and a column density of $(3.1 \pm 0.2) \times 10^{16}$ cm⁻². As for the complex organics, the rotation temperature is significantly higher than in previous surveys, while the column density is only slightly higher [as compared to Schilke et al. 1997, who find a rotation temperature of 99 \pm 3 K and column density of $(1.3 \pm 0.2) \times 10^{16}$ cm⁻²].

4.7. Unidentified Lines

We find 155 unidentified lines in the survey, some of them questionable (see Table 18). Those are marked as "Questionable assignment" in the table. The strongest reliable lines reach 10–14 K; most are in the 2–8 K range.

4.8. Conclusions

In this line survey of the Orion hot core we find 1064 spectral features consisting of 2032 lines, partially blended (see Figs. 1 and 4). The number of U lines is 155 or about 14%. This number is due to the sparseness of laboratory data in this wavelength range and will hopefully be reduced in the future. The spectrum is dominated by CO, CS, SO, SiO, HCN, HCO⁺, H₂CO, SO₂, and CH₃OH, with many lines from very highly excited states. The sensitivity to these highly excited lines is partly due to the small beam size (about 12"), well matched to the hot core. Contrary to the findings of Harris et al. (1995), we have no lack of weak or intermediate strength features in our spectrum (see also Fig. 2). For some molecules, a rotation diagram analysis yields very high rotation temperatures, above 300 K. It seems possible that these are not real but are due to optically thick emission from hot (150 to 200 K) compact regions rich in complex molecules. This hot component does not show up in lower frequency surveys, due to its compactness. The dominant coolants in this frequency range are SO₂ and CH₃OH, which outperform CO by a large factor. Finally, the line contribution to the apparent continuum, as measured in a broadband experiment, is only about 15% as compared to about 50% in lower frequency surveys.

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 $\label{eq:table 11} TABLE\ 11$ Table of $H_2O,\,H_2S$ and NH_2D Molecules

	v (MHz)	$J_{K_p,K_m}(,v_F)$	T* (K)	$\int_{K} T_R^* dv$ (K km s ⁻¹)	Notes
$H_2O-v_2 = 1$	658005.3	$1_{1.0} - 1_{0.1}$	8.9	150.4	
H ¹⁸ ₂ O	692079.1	$5_{3,2}-4_{4,1}$	1.3	6.0	
H_2S	611441.6	5 _{3,2} -5 _{2,3}	13.2	235.0	
	626474.6	$7_{6,1} - 7_{5,2}$	13.1	97.4	1
	650374.2	$4_{4,1} - 4_{3,2}$	15.9	343.9	
	665393.8	$4_{2,2} - 4_{1,3}$	8.6	178.6	
	687303.4	$2_{0,2}-1_{1,1}$	20.3	581.2	
	689120.2	$9_{7,2} - 9_{6,3}$	5.9	107.6	
	708470.4	$3_{1,2} - 3_{0,3}$	23.9	656.7	
NH_2D	641056.7	$3_{3,0}, 1_3 - 3_{2,2}, 0_3$	1.3	12.2	
	649915.2	$2_{0,2},1_1-1_{0,1},1_1$	2.8	51.0	2
	702479.6	$4_{2,3},1_3-3_{3,1},0_2$	1.7	10.0	
	714746.2	$3_{0,3}, 0_2 - 2_{1,1}, 1_1$	3.1	40.4	
	716859.9	$2_{1,1}^{0,0},1_1^{0}-1_{1,0}^{0,1},1_0$	3.5	28.5	3

Notes.—(1) Blend with CH₃OH at 626476.8. (2) Blend with ¹³CH₃OH at 649912.8. (3) Blend with ³⁴SO at 716876.1. This table is available in machine-readable form in the electronic edition of *The Astrophysical Journal*.

 $\begin{array}{c} \text{TABLE 12} \\ \text{Table of CH_3OH Molecules} \end{array}$

	ν		T_R^{*a}	$\int T_R^* dv^a$	
Molecule	(MHz)	J_{K}	(K)	$(K km s^{-1})$	Notesa
CH ₃ OH	607215.3	$12_{2}E-11_{1}E$	3.5	43.9	1
3	611580.1	$18_{2}\overset{?}{A}^{-}-17_{3}\overset{?}{A}^{-}$	4.2	35.1	
	613990.1	$17_{-7}E-18_{-6}E$	2.8	15.8	
	616980.0	$4_{-2}E-3_{-1}E$	5.7	81.4	
	623071.8	$18_{2}\tilde{A}^{+}-17_{3}\tilde{A}^{+}$	5.9	62.2	
	624235.2	$21_{1}A^{-}-20_{2}A^{-}$	19.7	234.5	2
	624365.4	$24_{2}E-23_{3}E$	16.4	203.8	3
	625352.5	$24_{10}A^{+}-25_{0}A^{+}$	4.6	45.5	
	625352.5	$24_{10}A^{-}-25_{0}A^{-}$	_	_	
	625749.6	13_0E-12_0E	18.1	179.0	
	626489.9	$21_{-1}E-20_{2}E$	8.5	57.3	
	626626.3	$3_2A^2_1A^-$	26.2	280.1	
	626640.0	$13_0 E - 12_0 E$	_	_	
	627170.7	$13_{-1}E-12_{-1}E$	19.7	206.2	
	627558.6	$13_0A^+ - 12_0A^+$	23.4	224.2	
	627650.7	$13_{10}E-12_{10}E$	3.0	12.2	
	627713.1	$13_{10}A^{-}-12_{10}A^{-}$	2.4	14.1	4
	627713.1	$13_{10}A^{+}-12_{10}A^{+}$	_	_	
	627750.9	$13_{-10}E-12_{-10}E$	3.5	22.4	
	627807.9	13_9E-12_9E	6.3	41.7	
	627816.6	$13_{-9}E-12_{-9}E$	_	_	
	627908.2	$13_9A^+ - 12_9A^+$	4.7	30.6	
	627908.2	$13_9A^12_9A^-$	_	_	
	627921.9	$13_{-8}E-12_{-8}E$	6.1	40.5	
	627971.4	$13_8A^+ - 12_8A^+$	6.7	48.4	
	627971.4	$13_8A^12_8A^-$	_	_	
	628021.4	13_8E-12_8E	_	_	
	628039.1	$13_7A^ 12_7A^-$	7.5	52.3	
	628039.1	$13_{7}A^{+}-12_{7}A^{+}$			
	628051.9	$13_2A^12_2A^-$	22.0	279.4	
	628114.1	$13_7 E - 12_7 E$	7.8	52.9	
	628115.4	13 ₋₇ E-12 ₋₇ E			
	628167.7	13_6E-12_6E	10.5	79.1	
	628187.8	$13_6A^ 12_6A^-$	12.2	103.8	
	628187.8	$13_6A^+ - 12_6A^+$		_	
	628237.8	$13_{-6}E - 12_{-6}E$	11.7	84.4	
	628251.4	13_5E-12_5E	13.0	111.7	-
	628318.2	$13_{-5}E-12_{-5}E$	17.1	293.3	5

TABLE 12—Continued

	ν		T_R^{*a}	$\int T_R^* dv^a$	
Molecule	(MHz)	$J_{\it K}$	(K)	$(K \text{ km s}^{-1})$	Notes ^a
CH OH	(20220.0		17.1	202.2	
CH₃OH	628330.0	$13_{-4}E - 12_{-4}E$	17.1	293.3	5
	628338.2	$13_5A^+ - 12_5A^+$	_	_	_
	628338.2 628408.9	$13_{5}A^{-}-12_{5}A^{-}$ $13_{4}E-12_{4}E$	16.3	153.6	_
	628445.4	$13_{-3}E-12_{-3}E$ $13_{-3}E-12_{-3}E$	17.3	149.0	
	628469.9	$13_{-3}E^{-1}2_{-3}E$ $13_{3}A^{+}-12_{3}A^{+}$	19.9	212.1	
	628512.2	$13_{4}A^{-}-12_{4}A^{-}$	25.1	384.1	
	628513.4	$13_4A^+ - 12_4A^+$			
	628525.0	$13_3A^ 12_3A^-$	_	_	
	628696.6	13_1E-12_1E	18.8	192.5	
	628816.1	$13_{3}^{1}E-12_{3}^{1}E$	16.3	154.4	
	628869.1	$13_{2}A^{+}-12_{2}A^{+}$	16.3	156.1	
	629140.5	$3_{2}A^{+}-2_{1}A^{+}$	21.1	155.4	
	629321.6	13_2E-12_2E	19.2	182.1	
	629651.9	$13_{-2}E-12_{-2}E$	17.4	159.1	
	629921.3	$7_1A^+ - 6_0A^+$	25.7	212.9	
	630952.4	$10_6A^+ - 11_5A^+$	8.4	57.0	
	630952.4	$10_6 A^ 11_5 A^-$	_	_	
	633423.1	$13_{1}A^{-}-12_{1}A^{-}$	23.2	247.5	
	634470.1	$21_{-9}E-22_{-8}E$	6.4	42.9	
	635389.9	26_4E-26_3E	2.5	35.5	
	635672.7	$25_4A^+-25_3A^+$	5.1	58.0	
	635874.1	$24_4A^+-24_3A^+$	3.8	36.2	
	635940.9 636013.9	$24_{-3}E-23_{-4}E$ $23_{4}A^{+}-23_{3}A^{+}$	1.3 3.5	11.0 21.5	
	636105.6	$23_4A^{+}-23_3A^{+}$ $22_4A^{+}-22_3A^{+}$	5.6	56.3	
	636161.0	$21_4A^+ - 21_3A^+$	7.6	60.7	
	636190.2	$15_4A^+ - 15_3A^+$	16.5	193.3	
	636190.3	$20_4A^+ - 20_3A^+$	_	_	
	636190.9	$14_4A^+ - 14_3A^+$	_		
	636193.4	$16_4A^+-16_3A^+$	_	_	
	636196.9	$13_4A^+-13_3A^+$	_	_	
	636198.7	$17_4A^+-17_3A^+$	_	_	
	636202.0	$19_4A^+ - 19_3A^+$	_	_	
	636202.9	$18_4A^+ - 18_3A^+$	_	_	
	636208.6	$12_4A^+ - 12_3A^+$	17.2	109.9	
	636225.9	$11_4A^+ - 11_3A^+$	16.2	141.4	
	636248.0	$10_4A^+ - 10_3A^+$	15.8	140.2	
	636274.2	$9_4A^+-9_3A^+$ $10_4A^10_3A^-$	21.2	192.5	
	636279.3 636280.5	$10_4A^ 10_3A^-$ $11_4A^ 11_3A^-$	_		
	636299.4	$11_4A^{-}-11_3A^{-}$ $12_4A^{-}-12_3A^{-}$	20.7	113.7	
	636303.4	$8_4A^+ - 8_3A^+$		—	
	636311.7	$8_4A^ 8_3A^-$	19.8	130.5	
	636333.5	$7_4A^+ - 7_3A^+$	21.3	235.3	
	636337.4	$7_{4}^{4}A^{-}-7_{3}^{3}A^{-}$	_	_	
	636341.6	$13_4A^13_3A^-$	_		
	636363.8	$6_4A^+-6_3A^+$	20.6	172.7	
	636365.7	$6_4A^6_3A^-$	_	_	
	636393.3	$5_4A^+ - 5_3A^+$	20.6	171.6	
	636393.8	$5_4A^ 5_3A^-$	_	_	
	636413.8	$14_4A^ 14_3A^-$	19.8	214.6	
	636419.9	$4_4A^+-4_3A^+$	_	_	
	636420.1	$4_4A^4_3A^-$	20.0	509.2	
	636522.9	$15_4A^ 15_3A^-$	29.9 10.2	598.3	6
	636677.6 636887.0	$16_4A^ 16_3A^ 17_4A^ 17_3A^-$	10.2 6.7	102.9 61.2	
	637161.5	$17_4A - 17_3A$ $18_4A - 18_3A -$	5.6	49.8	
	637512.4	$19_4A^ 19_3A^-$	5.1	55.6	
	637952.2	$20_4A^20_3A^-$	4.1	38.6	
	638279.6	$10_0 E - 9_{-1} E$	16.1	150.2	
	638494.4	$21_{4}A^{-}-21_{3}A^{-}$	2.5	17.3	
	638523.5	$8_3A^+ - 7_2A^+$	29.0	189.0	
	638817.8	$8_{3}A^{-}-7_{2}A^{-}$	25.9	180.5	
		. <u>.</u>			

TABLE 12—Continued

	ν		$T_R^{*\mathrm{a}}$	$\int T_R^* dv^a$	
Molecule	(MHz)	${J}_{K}$	(K)	$(K \text{ km s}^{-1})$	Notes
СН ₃ ОН	639153.6	22 ₄ A ⁻ -22 ₃ A -	2.9	26.1	
· ·	639945.4	$23_4A^23_3A^-$	1.3	12.4	
	640886.4	$24_4A^24_3A^-$	0.9	4.1	
	641993.9	$25_4A^25_3A^-$	1.4	15.4	
	647395.7	$12_{-8}E-13_{-7}E$	3.4	23.7	
	649541.5	14_1E-13_2E	10.6	129.1	
	651617.4	$10_{1}E-9_{0}E$	12.0	122.9	7
	656168.1 658102.2	13_2E-12_1E $19_2A^18_3A^-$	16.6 3.7	298.5 39.9	/
	658191.6	$16_{7}E-17_{6}E$	2.9	40.2	
	662295.8	$16_{-7}E - 17_{-6}E$	2.2	16.0	
	665442.4	$5_{-2}E-4_{-1}E$	19.6	204.2	
	667147.5	$10_{5}\overset{2}{A}^{+}-11_{4}\overset{1}{A}^{+}$	7.5	95.1	
	667148.1	$10_{5}A^{-}-11_{4}A^{-}$		_	
	667497.7	$6_4E - 7_3E$	6.7	67.8	
	668118.2	$18_0 E - 17_1 E$	4.8	49.0	
	668854.9	$25_{2}E-24_{3}E$	4.1	56.5	8
	670422.7	$14_1A^+ - 13_1A^+$	9.7	124.6	
	671480.8	$18_8A^+ - 19_7A^+$	4.1	46.4	
	671480.8 672360.9	$18_{8}A^{-}-19_{7}A^{-}$ $19_{2}A^{+}-18_{3}A^{+}$	2.9	33.8	
	672903.7	$19_{2}A - 18_{3}A$ $17_{-1}E - 16_{0}E$	6.2	62.3	9
	673416.2	$17_{-1}E - 10_{0}E$ $14_{0}E - 13_{0}E$	8.8	85.4	,
	673745.9	$4_2A^3_1A^-$	9.9	94.2	
	674990.4	$8_{1}A^{+}-7_{0}A^{+}$	17.1	172.3	
	675134.8	$14_{-1}E-13_{-1}E$	12.9	140.7	10
	675144.2	$14_{-4}E - 13_{-4}E$	_	_	_
	675612.9	$14_0A^+ - 13_0A^+$	14.1	151.3	
	675773.7	3_3E-2_2E	15.9	209.5	
	675777.9	$14_4A^13_4A^-$	_	_	
	675777.9	$14_4A^+ - 13_4A^+$			
	675963.6	$14_{-10}E - 13_{-10}E$	3.7	30.1	11
	676013.8	14_9E-13_9E	5.5	47.8	11
	676034.6 676130.7	$14_{-9}E-13_{-9}E$ $14_{9}A^{-}-13_{9}A^{-}$	4.8 3.1	46.6 13.2	
	676130.7	$14_9A^+ - 13_9A^+$		——————————————————————————————————————	
	676138.7	$14_{-8}E - 13_{-8}E$	4.5	39.6	
	676205.6	$14_8A^13_8A^-$	14.5	197.4	
	676205.6	$14_{8}A^{+}-13_{8}A^{+}$	_	_	
	676215.1	$14_{2}A^{-}-13_{2}A^{-}$	_	_	
	676250.7	$14_8E - 13_8E$	4.7	36.6	
	676269.5	$14_{7}A^{+}-13_{7}A^{+}$	7.3	69.9	
	676269.5	$14_{7}A^{-}-13_{7}A^{-}$		_	
	676349.4	$14_{-7}E - 13_{-7}E$	6.6	53.6	
	676362.4	$14_{7}E-13_{7}E$	6.9	54.1	
	676416.1 676425.6	14_6E-13_6E $14_6A^+-13_6A^+$	10.6	155.9	
	676425.6	$14_{6}A^{-}-13_{6}A^{-}$ $14_{6}A^{-}-13_{6}A^{-}$		_	
	676494.7	$14_{5}E - 13_{5}E$	20.0	578.6	12
	676496.8	$14_{-6}E - 13_{-6}E$			_
	676503.2	$19_0A^+ - 18_1A^+$	_	_	_
	676504.3	$13_{-6}E-14_{-5}E$	_	_	_
	676585.4	$14_{-4}E-13_{-4}E$	11.1	225.6	
	676591.4	$14_{-5}E-13_{-5}E$	_	_	
	676604.2	$14_5A^+ - 13_5A^+$		_	
	676604.2	$14_5A^ 13_5A^-$	_		
	676677.8	14_4E-13_4E	9.1	110.7	
	676712.5	$14_{-3}E-13_{-3}E$	10.0	139.0	
	676749.5	$14_3A^+ - 13_3A^+$	13.2	195.1	
	676823.6 676829.6	$14_4A^+ - 13_4A^+ 14_3A^ 13_3A^-$	14.9	259.8	
	677013.1	$14_{3}A - 13_{3}A$ $14_{1}E - 13_{1}E$	12.2	138.3	
		$14_{1}E-13_{1}E$ $14_{3}E-13_{3}E$	12.2		
	677190.6	14°E-13°E	171	136.3	

TABLE 12—Continued

Molecule	v (MHz)	I	T_R^{*a} (K)	$\int_{R} T_{R}^{*} dv^{a}$ (K km s ⁻¹)	Notesa
Molecule	(IVITIZ)	J_{K}	(K)	(K KIII 5)	INOTES
CH ₃ OH	677709.7	$14_{2}E-13_{2}E$	14.1	212.8	
	678252.7	$14_{-2}E-13_{-2}E$	12.5	152.9	13
	678785.4	$4_2A^+ - 3_1A^+$ $9_6A^+ - 10_5A^+$	21.8	248.8	
	679342.4 679342.4	$9_{6}A^{-1}0_{5}A^{-1}$ $9_{6}A^{-1}0_{5}A^{-1}$	9.5	93.9	
	680804.4	$22_{1}A^{-}-21_{2}A^{-}$	8.1	131.7	14
	681989.9	$14_{1}A^{-}-13_{1}A^{-}$	20.2	213.3	1.
	683749.8	$12_{-2}^{1}E-11_{1}^{1}E$	6.2	50.9	15
	684257.3	$25_{-3}E-24_{-4}E$	6.1	49.4	
	685505.1	$11_0E-10_{-1}E$	19.8	250.0	16
	686731.3	$9_3A^+ - 8_2A^+$	21.9	208.4	
	687224.6	$9_3A^8_2A^-$	23.9	253.6	
	697147.9	$15_{1}E-14_{2}E$	9.5	127.4	17
	701366.8 704289.2	$11_{1}E-10_{0}E$	22.6 5.2	478.2 39.0	17
	704289.2	$20_{2}A^{-}-19_{3}A^{-}$ $14_{2}E-13_{1}E$	12.4	101.5	
	705467.5	$25_{-3}E - 25_{-2}E$			
	706083.0	15_7E-16_6E	2.4	10.6	
	708813.2	$19_{0}E-18_{1}E$	16.3	119.8	18
	718158.8	$15_1 A^+ - 14_1 A^+$	16.3	251.1	
	718436.2	$4_{-4}E-3_{-3}E$	17.5	170.6	
	719664.9	$9_1A^+ - 8_0A^+$	12.7	117.4	
	720441.3	$5_2A^4_1A^-$	6.5	70.8	
	721011.1	15_0E-14_0E	11.0	97.7	
	721793.0	$20_2A^+-19_3A^+$	2.6	16.7	10
	721998.2 723040.8	$23_{-3}E-23_{-2}E$	1.6 8.0	11.0 84.1	19
	723281.4	$15_{-1}E-14_{-1}E$ $18_{-1}E-17_{0}E$	5.3	33.7	
	723619.6	$15_{0}A^{+}-14_{0}A^{+}$	7.5	63.3	
	724121.6	4_3E-3_2E	8.9	85.9	
	724345.4	$15_2 A^ 14_2 A^-$	2.6	20.6	
	724153.9	$15_{-10}E-14_{-10}E$	1.9	12.6	
	724482.2	$15_{7}A^{+}-14_{7}A^{+}$	4.7	35.7	
	724482.2	$15_7A^14_7A^-$	_	_ .	
	724565.0	$15_{-7}E-14_{-7}E$	7.0	53.1	
	724595.5	$15_{7}E-14_{7}E$	4.5	29.5	20
	724644.6	$15_6A^ 14_6A^- 15_6A^+ - 14_6A^+$	6.5	52.2	20
	724644.6 724648.6	$15_{6}A - 14_{6}A$ $15_{6}E - 14_{6}E$		_	
	724719.4	$15_{5}E-14_{5}E$	4.6	37.7	
	724740.9	$15_{-6}E-14_{-6}E$	3.9	28.2	
	724823.7	$15_{-4}E-14_{-4}E$	4.9	39.4	
	724851.5	$15_{-5}E-14_{-5}E$	5.7	50.0	
	724855.2	$15_5A^+-14_5A^+$	_	_	
	724855.2	$15_5A^14_5A^-$	_	_	
	724963.0	$15_{-3}E-14_{-3}E$	5.8	58.7	
CII OII(n. 1)	725013.2	$15_3A^+-14_3A^+$	6.2	46.0	
$CH_3OH(v_t-1)$	616322.2 623145.8	$20_{2}E-19_{3}E$ $10_{-7}E-10_{-6}E$	9.7 6.3	160.5 64.4	
	623193.3	$10_{-7}E - 10_{-6}E$ $14_{-7}E - 14_{-6}E$	9.2	117.4	
	623262.2	$9_{-7}E_{-9}E_{-6}E$	4.8	59.4	21
	623340.9	$8_{-7}E - 8_{-6}E$	17.3	126.7	
	623737.8	$9_{0}E - 8_{1}E$	6.6	58.5	
	625510.2	$13_8 A^+ - 12_8 A^+$	5.7	53.2	
	625781.4	13_3E-12_3E	6.1	35.3	
	625971.6	$13_{1}A^{+}-12_{1}A^{+}$	5.5	39.3	
	626196.5	$13_7 E - 12_7 E$	3.1	16.7	22
	626374.8	$13_{-8}E-12_{-8}E$	2.5	17.3	
	626376.4	$13_{-9}E-12_{-9}E$	42	25.4	
	626397.8	$13_{-6}E - 12_{-6}E$	4.2 6.8	25.4 58.1	
	626451.9 626476.8	$13_{-2}E-12_{-2}E$ $13_{7}A^{+}-12_{7}A^{+}$	13.1	97.4	23

TABLE 12—Continued

	ν	_	T_R^{*a}	$\int T_R^* dv^a$	
Molecule	(MHz)	J_{K}	(K)	(K km s ⁻¹)	Note
$CH_3OH(v_t = 1)$	626555.1	13_4E-12_4E	13.8	123.4	
	626555.9	17_0E-16_1E		_	
	626608.8	$13_{-3}E-12_{-3}E$	14.2	100.2	
	626609.0	$13_2A^+ - 12_2A^+$	_	_	
	626654.1	$13_{1}E-12_{1}E$	18.2	143.7	
	626673.9	$13_2A^ 12_2A^-$	15.3	169.4	24
	626803.9 626864.9	13_8E-12_8E $13_3A^+-12_3A^+$	5.9 8.6	99.8 58.5	24
	626880.8	$13_{3}A - 12_{3}A$ $13_{2}E - 12_{2}E$	9.0	76.8	
	626930.5	$13_{2}E-12_{2}E$ $13_{-1}E-12_{-1}E$	9.8	68.5	
	626945.2	$13_{-4}E - 12_{-4}E$	12.0	92.5	
	627004.6	$13_0A^+ - 12_0A^+$	8.8	52.9	
	627013.0	$13_{-5}E-12_{-5}E$	14.6	107.7	25
	627209.4	$13_{1}A^{-}-12_{1}A^{-}$	8.7	68.3	
	627444.1	$13_4A^+ - 12_4A^+$	7.4	62.7	
	627528.7	13_5E-12_5E	6.5	47.3	
	630583.0	$7_{-1}E - 8_{-2}E$	4.0	20.5	
	633156.9	$4_{-2}E - 5_{-3}E$	3.9	32.4	
	633572.0	$4_{-2}E-5_{-3}E$	12.9	99.7	_
	641918.1	9_3E-10_4E	1.6	19.3	26
	652940.1	$13_9A^+ - 12_8A^+$	3.4	59.5	
	654341.9	6_3E-5_2E	2.1	18.5	27
	657170.8	$8_{-5}E - 9_{-6}E$	6.0	84.2	27
	671912.5 673675.0	$10_0 E - 9_1 E$ $14_3 E - 13_3 E$	2.2 4.3	11.8 54.6	20
	673676.4	$14_{3}E-13_{3}E$ $14_{6}A^{-}-13_{6}A^{-}$	4.3	J4.0 —	28
	674017.3	$14_{6}A^{+}-13_{6}A^{+}$ $14_{1}A^{+}-13_{1}A^{+}$	6.6	109.5	29
	674129.2	$14_{-7}E - 13_{-7}E$	2.1	16.9	
	674129.2	$14_3A^13_3A^-$	_	_	
	674147.9	$14_{6}E-13_{6}E$	1.8	13.0	
	674512.5	$14_{-2}E-13_{-2}E$	3.2	29.0	
	674597.6	$14_{7}A^{+}-13_{7}A^{+}$	1.4	6.9	
	674618.4	$14_5A^13_5A^-$	2.5	22.6	
	674658.9	$14_{4}E-13_{4}E$	3.8	50.0	30
	674710.4	$14_2A^+ - 13_2A^+$	4.0	39.6	
	674716.8	$14_{-3}E-13_{-3}E$	_	_	
	674743.0	$14_{0}E-13_{0}E$	6.4	66.2	21
	674762.2	$14_{1}E-13_{1}E$	7.5	105.8	31
	674791.5 675034.5	$14_{2}A^{-}-13_{2}A^{-}$ $14_{3}A^{+}-13_{3}A^{+}$	6.6 3.5	41.8 18.5	32 33
	675044.0	$14_{3}A - 13_{3}A$ $14_{8}E - 13_{8}E$	3.7	43.0	34
	675047.9	$14_{2}E - 13_{2}E$			_
	675097.2	$14_{-1}E - 13_{-1}E$	4.5	34.4	
	675176.8	$14_0A^+ - 13_0A^+$	4.3	33.6	
	675231.6	$14_{-5}E-13_{-5}E$	2.9	24.6	
	675347.8	$14_{1}A^{-}-13_{1}A^{-}$	4.6	47.1	
	675887.9	14_5E-13_5E	5.6	71.0	
	678675.6	8 ₈ E-7 ₇ E	6.1	47.6	
	681789.7	$3_{-2}E-4_{-3}E$	14.2	161.0	
	682271.9	$16_{11}E-15_{10}E$	2.4	22.4	
	683425.2	$11_4A^12_5A^-$	3.6	34.6	25
	690294.3	8_3E-9_4E	2.9	27.3	35
	696257.5 698494.9	$13_{5}E-14_{4}E$ $13_{5}E-14_{4}E$	1.4 1.9	11.7 14.3	
	698930.9	$3_{1}A^{-}-2_{2}A^{-}$	2.6	11.1	
	702417.4	$7_{1}A - 2_{2}A$ $7_{3}E - 6_{2}E$	3.3	25.0	
	715237.9	$9_{-1}E - 8_{2}E$	5.2	40.7	
	720069.2	11_0E-10_1E	2.5	11.4	
	721519.8	15_3E-14_3E	1.2	3.4	
	721542.7	$15_6A^14_6A^-$	2.5	15.7	
	721723.8	$15_{-10}E-14_{-10}E$	2.7	20.9	
	500000	$15_{1}A^{+}-14_{1}A^{+}$	1.5	7.8	36
	722039.6				
	722039.6 722544.7	$15_{-2}E-14_{-2}E$	2.7	25.3	
					37

TABLE 12—Continued

	173	BLE 12—Continuea			
	v		T_R^{*a}	$\int T_R^* dv^a$	
Molecule	(MHz)	${J}_{K}$	(K)	$(K \text{ km s}^{-1})$	Notesa
$CH_3OH(v_t = 1)$	722742.3	15 ₄ E-14 ₄ E	1.7	4.5	
$CH_3OH(v_t=1)$	722789.8	$15_{4}E-14_{4}E$ $15_{2}A^{+}-14_{2}A^{+}$	2.3	9.7	
	722804.1	$15_{-3}E-14_{-3}E$	3.8	16.5	
	722823.8	15_0E-14_0E	2.7	15.3	
	722849.1	$15_{1}E-14_{1}E$	3.2	10.1	
	722889.6	$15_2A^14_2A^-$	2.1	6.9	
	723192.7	$15_3A^+-14_3A^+$	5.0	45.2	
	723202.8	15_2E-14_2E			
	723250.8	$15_{-1}E-14_{-1}E$	1.9	4.3	20
CII OII(2)	723336.8	$15_{-4}E - 14_{-4}E$	6.1	52.7	38
$CH_3OH(v_t=2)$	617627.4 624838.0	$20_{-7}E-20_{-6}E$ $13_{1}A^{+}-12_{1}A^{+}$	3.3 6.7	10.4 98.1	39 40
	625155.8	$13_{1}A - 12_{1}A$ $13_{11}E - 12_{11}E$	8.1	60.3	40
	625207.6	$13_{-10}E - 12_{-10}E$	5.7	66.4	
	625072.8	$13_9A^+ - 12_9A^+$	6.0	36.7	
	625383.1	$13_{10}A^{+}-12_{10}A^{+}$	4.6	35.6	
	625434.0	$13_{10}E - 12_{10}E$	2.2	6.8	
	626041.2	$13_3A^12_3A^-$	21.1	716.6	41
	626073.7	$13_4A^+ - 12_4A^+$	_	_	_
	626075.4	$13_{-2}E-12_{-2}E$	_		_
	626103.2	$13_0A^+ - 12_0A^+$	_	_	_
	626111.2	13_3E-12_3E	_	_	_
	626112.2	$13_{-7}E - 12_{-7}E$	_	_	_
	626137.5 626156.8	$13_{2}E-12_{2}E$ $13_{-1}E-12_{-1}E$	6.2	31.8	_
	629363.8	$17_{1}E-12_{-1}E$ $17_{1}E-16_{0}E$	5.3	37.9	
	673969.6	$14_{-4}E - 13_{-4}E$	3.8	28.7	42
	674162.1	$14_{-2}E-13_{-2}E$	1.7	17.6	
	674162.4	$14_4 \bar{A}^ 13_4 \bar{A}^-$	_	_	
	674196.5	$14_0A^+-13_0A^+$	1.4	12.1	
	674201.8	14_3E-13_3E		_	
	674254.4	$14_{-1}E-13_{-1}E$	1.2	6.2	
	677417.5	$18_{1}E-17_{0}E$	3.5	30.0	
	720812.0 722075.4	$15_{1}A^{+}-14_{1}A^{+}$ $15_{6}E-14_{6}E$	2.1 0.9	17.8 5.9	43
¹³ CH ₃ OH	612125.2	$13_{6}E-14_{6}E$ $13_{-1}E-12_{-1}E$	2.5	32.6	43
C113O11	612479.8	$13_{0}A^{+}-12_{0}A^{+}$	1.9	8.8	
	612891.6	$13_{7}A^{+}-12_{7}A^{+}$	1.7	12.6	
	612891.6	$13_{7}A^{-}-12_{7}A^{-}$	_	_	
	612934.2	$13_2A^12_2A^-$	1.8	21.8	44
	613303.9	$13_3A^+-12_3A^+$	7.5	127.3	
	613334.1	$13_4A^ 12_4A^-$	5.1	38.1	45
	613335.0	$13_4A^+ - 12_4A^+$	_	_	
	613350.9	$13_3A^ 12_3A^-$	4.1	23.2	
	613522.6 613612.1	$13_{1}E-12_{1}E$ $13_{3}E-12_{3}E$	1.9 2.4	13.4 14.3	
	613904.9	$4_{-2}E-3_{-1}E$	1.5	9.2	
	614090.1	$13_{2}E-12_{2}E$	2.7	24.4	
	614360.7	$13_{-2}E-12_{-2}E$	3.1	40.4	
	624626.1	$13_{-8}E-14_{-7}E$	16.7	186.7	46
	624744.3	$3_2A^2_1A^-$	9.7	120.2	47
	626007.8	$10_0 E - 9_{-1} E$	5.0	30.7	
	627138.7	$3_2A^+-2_1A^+$	6.9	44.9	
	632505.4	$8_3A^+ - 7_2A^+$	5.2	31.5	
	632771.4	$8_3A^ 7_2A^-$	6.7 1.5	61.4	
	637684.0 637685.1	$11_{5}A^{+}-12_{4}A^{+} 11_{5}A^{-}-12_{4}A^{-}$	1.3	18.1	
	638585.6	$17_{5}A - 12_{4}A$ $17_{7}E - 18_{6}E$	1.2	3.3	48
	640159.1	$23_4A^+-23_3A^+$	2.1	47.0	49
	640167.3	$10_{1}E - 9_{0}E$	_	_	_
	640289.7	$20_4A^+-20_3A^+$	3.8	64.2	
	640297.1	$16_4A^+ - 16_3A^+$	_	_	
	640298.1	$17_4A^+ - 17_3A^+$	_	_	
	640298.2	$19_4A^+ - 19_3A^+$		_	

TABLE 12—Continued

	ν		T_R^{*a}	$\int T_R^* dv^a$	
Molecule	(MHz)	$J_{\it K}$	(K)	(K km s ⁻¹)	Notes
¹³ CH ₃ OH	640298.7	$15_4A^+ - 15_3A^+$	_	_	
	640299.6	$18_4A^+ - 18_3A^+$	_	_	
	640304.4	$14_4A^+ - 14_3A^+$	_		
	640376.8 640398.5	$10_4A^+ - 10_3A^+ 11_4A^ 11_3A^-$	1.1 2.0	4.9 14.4	
	640403.4	$10_4A^ 10_3A^-$		—	
	640404.9	$9_4A^+ - 9_3A^+$		_	
	640408.6	$12_4A^12_3A^-$			
	640419.4	$9_4A^9_3A^-$	2.1	13.9	
	640435.0	$8_4A^+ - 8_3A^+ & 2.6$	2.6	20.8	
	640438.8 640442.4	$13_4A^ 13_3A^- 8_4A^ 8_3A^-$			
	640466.3	$7_4A^+ - 7_3A^+$	2.4	24.9	
	640469.6	$7_4A^{-} - 7_3A^{-}$	_	_	
	640494.6	$14_{4}^{4}A^{-}-14_{3}A^{-}$	3.4	23.7	
	640497.1	$6_4A^+-6_3A^+$	_	_	
	640498.4	$6_4A^6_3A^-$	_	_	
	640526.2	$5_4A^+ - 5_3A^+$	2.6	14.9	
	640526.6 640552.4	$5_4A^ 5_3A^- $ $4_4A^+ - 4_3A^+$	2.0	12.2	
	640552.6	$4_{4}A^{-}-4_{3}A^{-}$		12.2 —	
	641117.8	$18_4A^ 18_3A^-$	1.9	49.7	50
	642829.9	$22_{4}^{4}A^{-}-22_{3}^{3}A^{-}$	2.6	24.7	51
	642837.8	13_2E-12_1E	_	_	_
	649912.8	$18_0 E - 17_1 E$	2.8	51.0	52
	653045.4	$17_{-1}E-16_{0}E$	2.4	18.4	
	654420.2 657406.5	$14_{1}A^{+}-13_{1}A^{+}$ $14_{0}E-13_{0}E$	3.7 1.8	29.8 12.0	
	658953.1	$14_{-1}E - 13_{-1}E$	1.7	10.5	53
	659372.8	$14_{11}A^{+}-13_{11}A^{+}$	3.3	25.1	
	659372.8	$14_{11}A^{-}-13_{11}A^{-}$	_	_	
	659390.4	$14_0A^+ - 13_0A^+$	5.4	38.1	
	659941.9	14_8E-13_8E	2.8	15.3	
	659944.6	$14_2A^ 13_2A^-$	2.5	41.4	5.1
	660039.2 660044.0	$14_{-7}E - 13_{-7}E$ $14_{7}E - 13_{7}E$	3.5	41.4	54
	660115.2	$14_{6}A^{-}-13_{6}A^{-}$	3.0	27.4	
	660115.2	$14_{6}A^{+}-13_{6}A^{+}$	_	_	
	660673.9	$14_{1}E-13_{1}E$	3.2	31.1	
	660811.6	14_3E-13_3E	2.2	26.5	55
	660866.7	$14_2A^+ - 13_2A^+$	2.7	46.2	
	661190.8	$5_{-2}E-4_{-1}E$	2.8	10.2	56
	661314.4 661760.9	$14_{2}E-13_{2}E$ $14_{-2}E-13_{-2}E$	7.1 2.8	32.2 29.5	30
	666706.1	$8_1A^+ - 7_0A^+$	4.7	55.7	
	670756.6	$4_{2}A^{-}-3_{1}A^{-}$	3.1	25.0	
	672184.4	$11_0E-10_{-1}E$	1.7	14.4	
	675135.1	3_3E-2_2E	12.9	140.7	57
	675555.3	$4_2A^+ - 3_1A^+$	4.6	36.4	58
	677857.1 679554.5	15_1E-14_2E $9_3A^+-8_2A^+$	1.7 6.6	8.6 45.0	
	680000.2	$9_3A^8_2A^-$ $9_3A^8_2A^-$	9.2	111.4	59
	682370.9	$26_{2}E-25_{3}E$	4.4	117.5	37
	688611.6	$11_{1}^{2}E-10_{0}^{3}E$	2.5	20.1	
	690629.6	$14_{2}E-13_{1}E$	3.7	33.3	
	701022.4	$15_1A^+-14_1A^+$	1.5	5.9	
	702072.6	$18_{-1}E - 17_{0}E$	23.7	659.0	60
	703891.9	15_0E-14_0E	1.9	10.0	<i>L</i> 1
	704195.2 705726.9	$20_0A^+-19_1A^+$ $15_{-1}E-14_{-1}E$	1.9 2.9	12.7 10.5	61
	706256.9	$15_{-1}E^{-1}4_{-1}E$ $15_{0}A^{+}-14_{0}A^{+}$	4.3	37.5	
	706924.5	$15_{2}A^{-}-14_{2}A^{-}$	2.6	17.2	
	707113.2	$15_{7}E-14_{7}E$	1.2	4.2	
	707113.2	/ /			

TABLE 12—Continued

Molecule	v (MHz)	J_{κ}	T*a (K)	$\int T_R^* dv^a$ (K km s ⁻¹)	Notes ^a
-	(IVIIIE)		(11)	(II kiii 5)	110105
¹³ CH ₃ OH	707173.4	$15_6A^14_6A^-$	_	_	
	707173.4	$15_6A^+-14_6A^+$	_	_	
	707244.2	$15_{5}E-14_{5}E$	1.3	3.3	
	707342.2	$15_{-4}E-14_{-4}E$	1.9	11.2	62
	707368.0	$15_5A^+-14_5A^+$	3.6	31.7	
	707368.0	$15_5A^14_5A^-$	_	_	
	707442.9	$15_{4}E-14_{4}E$	1.7	7.9	
	707477.5	$15_{-3}E-14_{-3}E$	1.9	10.3	
	707518.2	$15_3A^+-14_3A^+$	1.9	10.9	
	707606.4	$15_4A^14_4A^-$	3.8	29.1	
	707609.0	$15_4A^+-14_4A^+$	_	_	
	707614.8	$15_3A^14_3A^-$	_	_	
	707812.8	$15_{1}E-14_{1}E$	2.6	14.7	
	708010.6	15_3E-14_3E	2.4	16.8	
	708055.2	$15_2A^+-14_2A^+$	2.3	15.3	
	708546.2	$6_{-2}E - 5_{-1}E$	4.1	23.7	
	709200.7	$15_{-2}E-14_{-2}E$	1.9	6.1	
	710385.8	$9_1A^+ - 8_0A^+$	4.5	23.7	63
	712808.4	$15_1A^14_1A^-$	2.9	20.8	
	718209.4	$4_{-4}E-3_{-3}E$	2.7	10.6	
	722316.5	$4_{3}E-3_{2}E$	2.7	11.2	
	610692.7	$13_6A^+ - 12_6A^+$	8.0	33.2	
	610721.9	$13_{-12}E-12_{-12}E$	8.8	76.1	
	610768.1	$13_{-10}E - 12_{-10}E$	6.9	59.3	
$^{13}CH_{3}OH(v_{t}-1)$	610890.4	$13_{-11}E-12_{-11}E$	4.3	44.7	
	635567.3	$4_{-2}E - 5_{-3}E$	1.4	5.9	
	657452.8	$14_6A^+ - 13_6A^+$	2.1	17.3	
	659064.0	$14_{1}A^{-}-13_{1}A^{-}$	1.5	11.2	
	683751.0	$26_2A^+-25_1A^+$	6.2	50.9	64
	704919.7	$15_{7}E-14_{7}E$	2.7	22.0	
	705464.4	$15_2A^14_2A^-$	_	_	
13 CH $_3$ OH $(v_t$ -2)	657452.8	$14_{2}A^{-}-13_{2}A^{-}$	1.7	11.7	
CH ₃ OD	629825.3	5_3E-4_2E	1.5	2.2	
	688887.2	$4_{-3}E-3_{-2}E$	0.7	1.5	65
	717466.9	$7_3A^+ - 6_2A^+$	1.1	7.1	66

Notes.—(1) Uncertain calibration. (2) Blend with ³⁴SO₂ at 624254.5. (3) Blend with CH₃CN at 624344.1 and SO₂ at 624344.5. (4) Blend with SiH at 627689.9 and SO₂ at 627715.3. (5) Blend with $^{34}SO_2$ at 628341.2. (6) Blend with CS at 636431.8. (7) Blend with 12 Co at 656167.4. (8) Blend with NS at 668859.8. (9) Blend with CH₃OCH₃ at 672903.4. (10) Blend with 13 CH₃OH at 675135.1. (11) Blend with 34 SO₂ at 676010.8 and SO₂- 12 C = 1 at 676021.4. (12) Blend with SO₂ at 676484.4. (13) Blend with HNCO at 678237.6. (14) Blend with HNCO at 680787.0. (15) Blend with 13 CH₃OH at 683751.1. (16) Blend with CH₃OCH₃ at 685500.3. (17) Blend with H₂CO at 701376.0. (18) Blend with HCN at 708877.2. (19) Questionable assignment. (20) Blend with HNCO at 724647.4. (21) Large frequency error. (22) Blend with ³⁴SO₂ at 626191.9. (23) Blend with H₂S at 626474.5. (24) Blend with SO₂ at 626809.5. (25) Blend with CH₃OCH₃ at 627015.6. (26) Blend with SO₂ at 641914.5. (27) Blend, large error in frequency. (28) Blend with CH₃CH₂CN at 674671.2. (29) Blend with $C^{17}O$ at 674009.3. (30) Blend with $^{34}SO_2$ at 674667.9. (31) Blend with ^{34}SO at 674774.1. (32) Blend with H₂CO at 674791.5. (33) Blend with H₂CS at 675039.0. (34) Blend with H₂CS at 675039.0. (35) Large frequency error, found at 690357.7. (36) Questionable assignment. (37) Blend with SO_2 at 722704.9. (38) Blend with CH_3OCH_3 at 723336.8. (39) Large frequency error. (40) Blend with CH_3CN at 624819.3. (41) Blend with $^{34}SO_2$ at 626043.6 and SO_2 at 626087.3. (42) Blend with $^{34}SO_2$ at 673991.8. (43) Questionable assignment. (44) Blend with $^{CH_3}OCH_3$ group at 612931.9. (45) Blend with $^{34}SO_2$ at 613338.1. (46) Blend with $^{CH_3}OCH_3$ group at 612931.9. (45) Blend with $^{34}SO_2$ at 613338.1. (46) Blend with ^{CH_3}CN at 624629.1. (47) Blend with CH₃CN at 624736.1. (48) Large frequency error. (49) Blend with $^{34}\mathrm{SO}_2$ at 640148.4 and 640167.1. (50) Too strong, possible blend with U line. (51) Blend with CH₃CN at 642829.6. (52) Blend with NH₂D at 639915.2. (53) Blend with HNCO at 658945.8. (54) Blend with ¹³CH₃OH at 660043.9. (55) Blend with CH₃CN at 660806.4. (56) Blend with SO₂ at 661332.5. (57) Blend with CH₃OH at 675134.8. (58) Blend with HCOOCH₃ at 675566.2. (59) Blend with SO_2 - $v_2 = 1$. (60) Blend with SO_2 at 702104.8. (61) Blend with $HCOOCH_3$ at 704193.5. (62) Blend with CH_3OCH_3 at 707340.1. (63) Blend with $H_2^{13}CO$ at 710386.6. (64) Blend with CH₃OH at 683749.8. (65) Questionable assignment. (66) Questionable assignment. This table is available in machine-readable form in the electronic edition of The Astrophysical Journal.

^a Dashes indicate that the line is part of a blend, and intensity and area are given by the values preceding it.

 $\label{eq:table 13} TABLE~13$ Table of $\rm H_2CO~and~H_2CS~Molecules$

	ν		T*a	$\int T_R^* dv^a$	
Molecule	(MHz)	J_{K_p,K_m}	(K)	$(K \text{ km s}^{-1})$	Notes ^a
H ₂ CO	631705.8	9 _{1.9} -8 _{1.8}	37.1	749.4	
-	647084.7	$9_{0,9}^{1,5} - 8_{0,8}^{1,5}$	21.9	339.2	
	653973.1	$9_{2,8} - 8_{2,7}$	13.5	213.5	
	654457.5	9 _{7.3} -8 _{7.2}	3.7	30.0	
	654457.5	9 _{7,2} -8 _{7,1}	_	_	
	654836.6	$9_{6,4} - 8_{6,3}$	3.2	33.4	
	654836.6	$9_{6,3} - 8_{6,2}$	_	_	
	655212.9	$9_{5,5} - 8_{5,4}$	7.0	115.1	
	655212.9	$9_{5,4} - 8_{5,3}$	_	_	
	655641.9	$9_{4,6} - 8_{4,5}$	7.8	107.4	
	655645.8	$9_{4,5} - 8_{4,4}$	_	_	
	656167.4	$9_{3,7} - 8_{3,6}$	16.6	298.5	1
	656467.3	$9_{3,6} - 8_{3,5}$	8.5	170.8	
	662212.1	$9_{2,7} - 8_{2,6}$	20.1	431.3	2
	674812.7	$9_{1,8} - 8_{1,7}$	16.2	242.0	
	701376.0	$10_{1,10} - 9_{1,9}$	22.6	478.2	3
	716943.9	$10_{0,10} - 9_{0,9}$	18.1	400.9	
H ¹³ ₂ CO	616637.3	$9_{1,9} - 8_{1,8}$	2.7	38.6	
	639764.8	$9_{3,7} - 8_{3,6}$	1.6	15.2	
	657663.6	$9_{1,8} - 8_{1,7}$	3.0	23.5	
	684675.7	$10_{1,10} - 9_{1,9}$	4.9	94.5	
	699873.9	$10_{0,10} - 9_{0,9}$	3.4	63.4	
	708266.5	$10_{2,9} - 9_{2,8}$	4.1	24.0	
	710386.6	$10_{4,7} - 9_{4,6}$	4.5	23.7	4
	710393.4	$10_{4,6} - 9_{4,5}$	_	_	_
	711415.2	$10_{3,7} - 9_{3,6}$	3.6	64.2	
H_2CS	626482.7	$18_{1,17} - 17_{1,16}$	13.1	97.4	5
	641421.4	$19_{1,19}$ $-18_{1,18}$	2.1	19.1	6
	648827.4	$19_{0,19} - 18_{0,18}$	2.1	27.2	
	651511.3	$19_{2,18}$ $-18_{2,17}$	4.7	49.5	7
	675039.0	$20_{1,20}$ – $19_{1,19}$	3.5	18.5	8

Notes.—(1) Blend with CH₃OH at 656168.1. (2) Blend with SO₂ at 662202.7. (3) Blend with CH₃OH at 701366.8. (4) Blend with ¹³CH₃OH at 710385.8. (5) Blend with H₂S at 626474.5 and CH₃OH at 626476.8. (6) Large frequency error. (7) Blend with SO at 651493.5. (8) Blend with CH₃OH at 675035.8 and CH₃OH at 675044.0. This table is available in machine-readable form in the electronic edition of *The Astrophysical Journal*.

TABLE 14
TABLE OF HNCO MOLECULES

Molecule	v (MHz)	J_{K_p,K_m}	T*a (K)	$\int_{K} T_R^* dv^a$ (K km s ⁻¹)	Notes
HNCO	612867.6	28 _{1,28} -27 _{1,27}	2.6	28.0	
	614780.0	$28_{3,26}^{1,25} - 27_{3,25}^{1,27}$	2.9	28.8	
	614780.2	28 _{3,25} -27 _{3,24}	_	_	
	614976.4	28 _{2,27} -27 _{2,26}	1.7	7.1	
	615052.1	$28_{2,26}$ $-27_{2,25}$	2.5	17.5	
	615098.9	$28_{0,28} - 27_{0,27}$	4.5	42.3	
	617346.7	$28_{1,27}$ $-27_{1,26}$	3.3	23.0	
	632571.7	$11_{1,11} - 12_{0,12}$	6.2	77.8	
	634731.7	$29_{1,29} - 28_{1,28}$	4.8	40.1	
	636714.1	$29_{3,27} - 28_{3,26}$	2.2	23.9	
	636714.3	$29_{3,26} - 28_{3,25}$	_	_	
	636999.9	$29_{2,27} - 28_{2,26}$	2.3	14.2	1
	637037.8	$29_{0,29} - 28_{0,28}$	7.7	74.8	
	655444.5	$10_{1,10}$ $-11_{0,11}$	3.9	31.4	
	656593.1	$30_{1,30}$ – $29_{1,29}$	2.5	18.8	
	658945.8	$30_{2,28}$ – $29_{2,27}$	1.7	10.5	2
	658973.5	$30_{0,30} - 29_{0,29}$	2.2	10.7	
	661390.0	$30_{1,29} - 29_{1,28}$	2.3	18.5	
	678237.5	$9_{1,9}$ $-10_{0,10}$	12.5	152.9	3
	678452.0	$31_{1,31} - 30_{1,30}$	1.7	11.2	
	680575.0	$31_{3,29} - 30_{3,28}$	1.0	9.6	
	680787.0	$31_{2,30} - 30_{2,29}$	8.1	131.7	4
	680889.6	$31_{2,29} - 30_{2,28}$	4.2	35.8	
	680906.1	$31_{0,31} - 30_{0,30}$	7.9	62.2	
	683407.6	$31_{1,30} - 30_{1,29}$	3.6	34.6	
	700308.1	$32_{1,32} - 31_{1,31}$	3.1	27.5	5
	700950.3	$8_{1,8} - 9_{0,9}$	2.4	7.9	
	702501.8	$32_{3,30} - 31_{3,29}$	2.6	26.6	6
	702718.6	$32_{2,31} - 31_{2,30}$	3.0	18.5	
	702831.5	$32_{2,30} - 31_{2,29}$	4.9	65.0	
	702835.5	$32_{0,32} - 31_{0,31}$	_	_	
	705422.4	$32_{1,31} - 31_{1,30}$	2.4	19.3	
	722161.5	$33_{1,33} - 32_{1,32}$	3.4	35.8	
	723582.7	$7_{1,7} - 8_{0,8}$	1.7	16.0	
	724647.4	$33_{2,32} - 32_{2,31}$	6.5	52.2	7
	724761.4	$33_{0,33} - 32_{0,32}$	4.3	22.3	
-					

Notes.—(1) Blend with CH_3CH_2CN at 637003.5. (2) Blend with $^{13}CH_3OH$ at 658953.1. (3) Blend with CH_3OH at 678252.7. (4) Blend with CH_3OH at 680804.4. (5) Blend with SO_2 at 700312.9. (6) Blend with $^{34}SO_2$ at 702502.1. (7) Blend with CH_3OH at 724644.6. This table is available in machine-readable form in the electronic edition of *The Astrophysical Journal*

^a Dashes indicate that the line is part of a blend, and intensity and area are given by the values preceding it.

^a Dashes indicate that the line is part of a blend, and intensity and area are given by the values preceding it.

 $\begin{array}{c} \text{TABLE 15} \\ \text{Table of HCOOCH}_{3} \text{ Molecules} \end{array}$

	I ABL	E OF HCOOCH ₃ MOLI	ECULES		
Molecule	v (MHz)	J_{K_p,K_m}	T*a (K)	$\int T_R^* dv^a$ (K km s ⁻¹)	Notes ^a
HCOOCH ₃	614019.0	$34_{9,26}A - 33_{8,25}A$	4.0	28.4	
110000113	615288.6	$50_{14,37}E-49_{0,49}E$	6.1	96.7	
	618592.2	$50_{20,31}E-49_{20,30}E$	2.1	33.0	1
	624023.9	$25_{12,14}^{20,31}E-24_{11,14}^{20,30}E$	2.4	22.6	
	624031.4	$25_{12,14}A - 24_{11,13}A$	_	_	
	624031.5	$25_{12,13}A-24_{11,14}A$	_	_	
	624551.7	$54_{5,49}A - 53_{6,48}A$	4.5	63.0	
	624551.9	$54_{6,49}A - 53_{6,48}A$	_	_	
	624552.1	$54_{5,49}A - 53_{5,48}A$	_	_	
	624552.4	$54_{6,49}A - 53_{5,48}A$	_	_	
	624582.8	$54_{11,44}E - 53_{2,51}E$	6.5	68.8	
	624583.0	$54_{12,43}E - 53_{2,51}E$	_	_	
	624583.1	$54_{11,44}E - 53_{11,43}E$	_	_	
	624583.4	$54_{12,43}E - 53_{11,43}E$	_	_	
	624670.4	$16_{16,1}E-15_{15,1}E$	4.1	24.1	
	624680.4	$16_{16,0}A - 15_{15,1}A$	3.1	20.9	
	624680.4	$16_{16,1}A - 15_{15,0}A$	_	_	
	625294.0	$55_{10,46}E - 54_{9,45}E$	7.3	101.8	
	625294.0	$55_{9,46}E-54_{9,45}E$	_	_	
	625294.1	$55_{10,46}E - 54_{10,45}E$	_	_	
	625294.1	$55_{9,46}E-54_{10,45}E$	_	_	
	627618.2	$23_{13,11}A - 22_{12,10}A$	1.1	2.3	
	627618.2	$23_{13,10}A - 22_{12,11}A$	_	_	_
	632401.4	$51_{17,35}A - 50_{17,34}A$	3.6	39.0	2
	633906.0	$19_{15,4}E-18_{14,4}E$	5.2	23.4	
	633950.0	$19_{15,5}E-18_{14,5}E$	7.9	80.1	
	633960.4	$19_{15,4}A - 18_{14,5}A$	_	_	
	633960.4	$19_{15,5}A-18_{14,4}A$	2.0	41.2	2
	634005.9	$51_{14,37}E - 50_{14,36}E$	3.9	41.3	3
	634317.6	$53_{12,41}E - 52_{12,40}E$	3.9	31.4	4
	634373.7	$53_{13,41}E - 52_{13,40}E$	4.1	48.1	5
	635027.6 635027.8	$55_{5,50}A - 54_{6,49}A$	3.6	42.0	
	635027.8	55 _{6,50} A-54 _{6,49} A 55 _{5,50} A-54 _{5,49} A		_	
	635027.9	$55_{6,50}A - 54_{5,49}A$ $55_{6,50}A - 54_{5,49}A$			
	635749.3	$51_{17,35}E-50_{17,34}E$	1.5	12.3	
	635750.4	$31_{10,22}A - 30_{9,21}A$			
	635750.8	$56_{4,52}A - 55_{5,51}A$	_	_	
	635750.8	$56_{5,52}A - 55_{5,51}A$	_	_	
	635750.8	$56_{4,52}A - 55_{4,51}A$	_	_	
	635750.8	$56_{5,52}A - 55_{4,51}A$		_	
	635751.2	$51_{15,37}A - 50_{15,36}A$	_	_	
	635754.6	$31_{10,22}E-30_{7,23}E$	_	_	
	636073.6	$26_{12,15}A - 25_{11,14}A$	1.2	2.8	
	636073.8	$26_{12,14}^{12,13}A-25_{11,15}^{11,14}A$	_	_	
	636946.2	$17_{16,2}E-16_{15,2}E$	2.7	19.3	
	636954.7	$17_{16,1}^{16,2}A - 16_{15,2}^{16,2}A$			
	636954.7	$17_{16,2}A-16_{15,1}A$	_	_	
	640942.1	$51_{15,37}E-50_{15,36}E$	1.1	3.7	
	643089.8	$22_{14,9}E-21_{13,9}E$	2.4	30.4	6
	643277.7	$29_{11,18}E-28_{10,18}E$	4.3	59.8	7
	643281.8	$29_{11,19}A - 28_{10,18}A$	_	_	_
	646208.6	$20_{15,6}E-19_{14,6}E$	2.3	19.3	
	646216.9	$20_{15,5}A - 19_{14,6}A$	_		
	646216.9	$20_{15,6}A-19_{14,5}A$	_		
	648324.2	$32_{10,22}A - 31_{9,23}A$	5.8	91.0	
	648333.4	$32_{8,24}E-31_{7,25}E$	_	_	
	649168.1	$18_{16,2}E-17_{15,2}E$	3.6	51.9	
	649219.8	$18_{16,3}E-17_{15,3}E$	3.2	15.5	
	649226.7	$18_{16,2}A-17_{15,3}A$	_	_	
	649226.7	$18_{16,3}A - 17_{15,2}A$	_	_	
	655387.4	$56_{6,50}A - 55_{7,49}A$	2.7	30.1	
	655388.9	$56_{7,50}A - 55_{7,49}A$	_	_	
	655390.0	$56_{6,50}A - 55_{6,49}A$		_	

TABLE 15—Continued

	v		T_R^{*a}	$\int T_R^* dv^a$	
Molecule	(MHz)	${J}_{K_p,K_m}$	(K)	$(K km s^{-1})$	Notes
HCOOCH ₃	655391.6	56 _{7,50} <i>A</i> –55 _{6,49} <i>A</i>	_		
, and the second	656224.1	$33_{8,26}E-32_{7,25}E$	1.7	11.7	
	657868.9	$53_{17,37}A - 52_{17,36}A$	3.7	33.1	
	657869.8	$53_{17,36}A - 52_{17,35}A$	3.7	33.1	
	658466.4	$21_{15,6}A-20_{14,7}A$	3.8	23.8	
	658466.4	$21_{15,7}A-20_{14,6}A$	_	_	
	660009.6	$28_{12,16}E-27_{11,16}E$	3.1	17.6	
	664445.5	$17_{17,1}E-16_{16,1}E$	6.5	77.1	
	664449.2	$17_{17,0}A-16_{16,1}A$	_	_	
	664449.2	$17_{17,1}A-16_{16,0}A$	_	_	
	665509.0	56 _{7,49} A-55 _{7,48} A	2.3	28.0	8
	665560.7	$56_{13,43}E - 55_{13,42}E$	3.0	26.7	
	665568.7	55 _{9,47} A-54 _{9,46} A			
	665618.6	$55_{3,53}E-54_{3,52}E$	1.9	20.2	
	665705.8	$54_{20,34}E - 53_{20,33}E$	3.3	36.5	9
	666441.8	$58_{6,53}A - 57_{6,52}A$	5.5	94.0	
	666441.9	$58_{5,53}A - 57_{5,52}A$	4.0	05.7	
	667470.7	$24_{14,10}E - 23_{13,10}E$	4.8	95.7	
	672459.4	$54_{16,39}A - 53_{16,38}A$	1.5	8.4	
	673708.7	$20_{16,4}E-19_{15,4}E$	2.7	20.2	10
	675566.2	$55_{29,26}A - 54_{29,25}A$	4.6	36.4	10
	675566.2 675654.3	$55_{29,27}A - 54_{29,26}A$	2.3	 14.9	_
	675678.1	$56_{8,48}A - 55_{9,47}A$ $55_{29,26}E - 54_{29,25}E$	0.9	6.9	11
	675710.7	$56_{14,43}E - 55_{3,53}E$	0.7	2.1	11
	675997.2	$55_{27,28}A - 54_{27,27}A$	3.6	16.6	
	675997.2	$55_{27,29}A - 54_{27,27}A$ $55_{27,29}A - 54_{27,28}A$	J.0	10.0	
	676111.8	$27_{13,15}A - 26_{12,14}A$	4.2	38.8	
	676111.8	$27_{13,15}A - 26_{12,15}A$			
	676408.5	$52_{11,41}A - 51_{11,40}A$	5.4	52.9	
	678005.8	$32_{11,21}A - 31_{10,22}A$	2.9	18.0	
	679653.1	$25_{14,11}E - 24_{13,11}E$	1.1	5.0	
	679682.0	$25_{14,11}E - 24_{13,11}E$ $25_{14,12}E - 24_{13,12}E$	2.2	10.6	
	679683.6	$25_{14,11}A - 24_{13,12}A$		_	
	679683.6	$25_{14,12}A - 24_{13,11}A$	_	_	
	680169.9	$54_{13,42}A - 53_{13,41}A$	4.3	39.7	
	680967.4	$36_{10,27}A - 35_{9,26}A$	3.5	40.7	
	682899.4	$23_{15,8}E-22_{14,8}E$	1.7	11.1	
	682935.7	$23_{15,9}E-22_{14,9}E$	2.2	23.1	
	682938.4	$23_{15,8}A - 22_{14,9}A$	_	_	
	682938.4	$23_{15,9}A-22_{14,8}A$	_	_	
	683571.6	$53_{12,41}A - 52_{12,40}A$	3.9	120.2	
	683762.5	$30_{2,28}E-29_{11,18}E$	3.9	44.9	
	683770.6	$30_{12,19}A - 29_{11,18}A$	_	_	
	683773.4	$30_{12,18}A - 29_{11,19}A$	_	_	
	685971.8	$21_{16,5}E-20_{15,5}E$	3.8	32.4	
	686016.9	$21_{16,6}E-20_{15,6}E$	3.1	19.6	
	686019.4	$21_{16,5}A - 20_{15,6}A$	_	_	
	686019.4	$21_{16,6}A - 20_{15,5}A$	_	_	
	686373.6	$57_{8,49}A - 56_{8,48}A$	3.1	31.9	
	687871.1	$56_{15,42}E - 55_{15,41}E$	2.5	36.7	
	688113.0	$56_{28,29}E - 55_{28,28}E$	2.3	21.5	
	688995.4	$56_{25,31}A - 55_{25,30}A$	1.6	13.9	12
	688995.4	$56_{25,32}A - 55_{25,31}A$	_	_	_
	688997.2	$56_{23,33}E-55_{25,31}E$	_	_	_
	688997.4	$19_{17,3}E-18_{16,3}E$	_	_	_
	688999.1	$19_{17,2}A-18_{16,3}A$	_	_	_
	688999.1	$19_{17,3}A-18_{16,2}A$	_	_	_
	689006.5	$53_{11,42}A - 52_{11,41}A$	_	— 7.5.5	_
	689070.1	$56_{25,31}E - 55_{25,30}E$	7.1	75.5	13
	691815.8	$26_{14,12}E-25_{13,12}E$	1.6	11.9	
	691841.9	$26_{14,13}A - 25_{13,12}A$	2.8	12.1	
	691841.9	$26_{14,12}A - 25_{13,13}A$	_	_	
	691841.9 691842.6 692204.0	$26_{14,12}A - 25_{13,13}A$ $26_{14,13}E - 25_{13,13}E$ $56_{20,37}A - 55_{20,36}A$	 1.1		14

TABLE 15—Continued

Molecule	v (MHz)	${J}_{K_p,K_m}$	T*a (K)	$\int T_R^* dv^a$ (K km s ⁻¹)	Notes ^a
HCOOCH ₃	692204.0	56 _{20,36} A-55 _{20,35} A	_	_	14
-	695122.9	$24_{15.9}E-23_{14.9}E$	7.1	111.1	15
	695156.9	$24_{15.10}E-23_{14.10}E$	2.4	19.9	
	695157.8	$24_{15.9}A - 23_{14.10}A$	_	_	
	695157.8	$24_{15,10}A-23_{14,9}A$	_	_	
	704192.7	$18_{18.1}E-17_{17.1}E$	1.9	12.7	16
	704193.5	$18_{18.0}A - 17_{17.1}A$	_	_	_
	704193.5	$18_{18,1}A - 17_{17,0}A$	_	_	_
	710517.7	$23_{16.7}A - 22_{15.8}A$	0.8	1.0	
	710517.7	$23_{16.8}A - 22_{15.7}A$	_	_	
	710517.9	$23_{16.8}E-22_{15.8}E$	_	_	
	715314.6	$57_{12.46}A - 56_{12.45}A$	2.9	25.4	
	717124.3	$56_{13,43}A - 55_{13,42}A$	2.3	5.0	
	717128.9	$57_{14,44}A - 56_{14,43}A$	_	_	
	717168.6	$56_{7,49}E-55_{7,48}E$	1.9	1.8	
	717177.9	$57_{10,47}E - 56_{10,46}E$	_	_	

Notes.—(1) Large frequency error. (2) Questionable assignment. (3) Large frequency error. (4) Large frequency error. (5) Large frequency error. (6) Blend with CH₃CN at 643074.1. (7) Blend with CH₃CN at 643269.9. (8) Blend with CH₃CH₂CN at 665506.6. (9) Blend with CH₃OCH₃ at 665710.6. (10) Blend with ¹³CH₃OH at 675555.3, large frequency error. (11) Large frequency error. (12) Blend with CH₃CH₂CN at 689007.7. (13) Large frequency error. (14) Blend CH₃CH₂CN at 692200.9. (15) Blend with ³⁴SO₂ at 691119.3. (16) Blend with ¹³CH₃OH at 704195.3. This table is available in machine-readable form in the electronic edition of *The Astrophysical Journal*.

^a Dashes indicate that the line is part of a blend, and intensity and area are given by the

^a Dashes indicate that the line is part of a blend, and intensity and area are given by the values preceding it.

 $\begin{array}{c} \text{TABLE 16} \\ \text{Table of CH_3OCH_3 Molecules} \end{array}$

Molecule	v (MHz)	${J}_{K_p,K_m}$	T_R^{*a} (K)	$\int T_R^* dv^a$ (K km s ⁻¹)	Notes
CH₃OCH₃	608093.5	12 _{7,5} EA-11 _{6,5} EA	5.8	116.1	
	608096.1	$12_{7,5}EE-11_{6,5}EE$	_	_	
	608097.5	$12_{7,6}AE-11_{6,5}AE$	_	_	
	608097.5	$12_{7,5}AE-11_{6,6}AE$	_	_	
	608098.8	$12_{7,6}AA-11_{6,5}AA$	_	_	
	608098.8	$12_{7,5}AA-11_{6,6}AA$	_	_	
	608100.1	12 _{7,6} EE-11 _{6,6} EE	_	_	
	608101.5	$12_{7,6}EA - 11_{6,6}EA$	1.2		
	609703.6 609704.9	$9_{8,2}EA - 8_{7,1}EA$	1.2	6.8	
	609704.9	$9_{8,2}EE-8_{7,1}EE \ 9_{8,1}AA-8_{7,2}AA$		_	
	609706.1	$9_{8,2}AA - 8_{7,1}AA$	_	_	
	609707.9	$9_{8,1}AE - 8_{7,2}AE$	_	_	
	609707.9	$9_{8,2}^{8,1}AE-8_{7,1}^{7,2}AE$	_	_	
	609709.1	$9_{8,1}^{3,2}EE-8_{7,2}^{7,1}EE$	_	_	
	609712.1	$9_{8,1}EA - 8_{7,2}EA$	_	_	
	610844.4	$20_{3,17}AE-19_{2,18}AE$	4.7	52.7	1
	610844.4	$20_{3,17}EA-19_{2,18}EA$	_	_	_
	610847.6	$20_{3,17}EE-19_{2,18}EE$	_	_	_
	610850.9	$20_{3,17}AA-19_{2,18}AA$		_	_
	612280.8	$30_{11,19}EA-30_{10,20}EA$	2.7	22.3	2
	612280.8	$30_{11,19}EE-30_{10,20}EE$	_	_	_
	612280.9 612280.9	$30_{11,19}AA - 30_{10,20}AA$ $30_{11,20}AA - 30_{10,21}AA$		_	
	612281.2	$30_{11,19}AE-30_{10,20}AE$		_	
	612281.3	$30_{11,20}AE - 30_{10,20}AE$	_	_	_
	612281.3	$30_{11,20}EE-30_{10,21}EE$	_	_	_
	612281.8	$30_{11,20}^{11,20}EA-30_{10,21}^{10,21}EA$	_	_	_
	612929.9	$27_{11,16}AA-27_{10,17}AA$	1.8	21.8	3
	612929.9	$27_{11,17}AA - 27_{10,18}AA$	_	_	_
	612930.8	$27_{11,16}EE-27_{10,17}EE$	_	_	_
	612931.1	$27_{11,17}EE-27_{10,18}EE$	_	_	_
	612931.6	$27_{11,16}EA-27_{10,17}EA$	_	_	_
	612931.9	27 _{11,16} AE-27 _{10,17} AE	_		_
	612931.9	27 _{11,17} AE-27 _{10,18} AE	_	_	_
	612932.2 613590.3	$27_{11,17}EA-27_{10,18}EA$	1.9	12.5	
	613590.3	$22_{11,12}AA - 22_{10,13}AA$ $22_{11,11}AA - 22_{10,12}AA$			
	613592.4	$22_{11,11}EE-22_{10,12}EE$	_	_	
	613592.4	$22_{11,12}EE-22_{10,13}EE$	_	_	
	613594.5	$22_{11,11}^{11,12}EA-22_{10,12}^{10,13}EA$	_	_	
	613594.5	$22_{11,12}AE-22_{10,13}AE$	_	_	
	613594.5	$22_{11,11}AE-22_{10,12}AE$	_	_	
	613594.5	$22_{11,12}EA-22_{10,13}EA$	_	_	
	613678.0	$21_{11,11}EA-21_{10,12}EA$	2.6	27.6	4
	613678.0	$21_{11,10}AE-21_{10,11}AE$		_	_
	613678.0	21 _{11,11} AE-21 _{10,12} AE	_		_
	613678.1	$21_{11,10}EA-21_{10,11}EA$	2.4	27.1	_
	613849.9	$18_{11,8}AA - 18_{10,9}AA$	2.4	37.1	
	613849.9 613852.7	$18_{11,7}AA - 18_{10,8}AA 18_{11,8}EE - 18_{10,9}EE$		_	
	613852.8	$18_{11,7}EE-18_{10,8}EE$	_	_	
	613855.4	$18_{11.8}EA - 18_{10.9}EA$	_	_	
	613855.5	$18_{11,8}AE-18_{10,9}AE$		_	
	613855.5	$18_{11,7}^{11,8}AE-18_{10,8}^{10,9}AE$	_	_	
	613855.6	$18_{11,7}EA-18_{10,8}EA$	_	_	
	613959.1	$14_{11,4}AA-14_{10,5}AA$	3.2	40.9	
	613962.3	$14_{11,4}EE-14_{10,5}EE$	_	_	
	613962.4	14 _{11,3} EE-14 _{10,4} EE	_	_	
	613965.6	$14_{11,4}EA-14_{10,5}EA$		_	
	613965.8	14 _{11,4} AE-14 _{10,5} AE	_	_	
	613965.8	$14_{11,3}AE-14_{10,4}AE$	_	_	
	613965.9	$14_{11,3}EA-14_{10,4}EA$			

TABLE 16—Continued

Molecule	v (MHz)	J_{K_p,K_m}	T_R^{*a} (K)	$\int T_R^* dv^a$ (K km s ⁻¹)	Notes ^a
CH ₃ OCH ₃	613970.8	13 _{11,3} AA-13 _{10,4} AA			
3 3	613974.1	$13_{11,3}EE-13_{10,4}EE$	_	_	
	613974.2	$13_{11,2}EE-13_{10,3}EE$	_	_	
	613977.5	$13_{11,3}EA-13_{10,4}EA$	_	_	
	613977.7	$13_{11,2}AE-13_{10,3}AE$	_	_	
	613977.7	$13_{11,3}AE-13_{10,4}AE$	_	_	
	613977.8	$13_{11,2}EA-13_{10,3}EA$	_	_	
	613978.0	$12_{11,2}AA - 12_{10,3}AA$	_	_	
	613978.0 614303.1	$12_{11,1}AA - 12_{10,2}AA 26_{4,23}AA - 25_{1,24}AA$	2.4	23.3	
	614303.2	$26_{4,23}EE-25_{1,24}EE$			
	614303.2	$26_{4,23}AE-25_{1,24}AE$	_	_	
	614303.2	$26_{4,23}^{4,23}EA-25_{1,24}^{1,24}EA$	_	_	
	624901.3	$16_{6,11}EA-15_{5,10}EA$	6.7	89.1	5
	624905.4	$16_{6,11}AE-15_{5,10}AE$	_	_	_
	624905.6	$16_{6,11}EE-15_{5,10}EE$	_	_	_
	624907.9	$16_{6,10}EA-15_{5,10}EA$	_	_	_
	624908.8	$16_{6,11}AA-15_{5,10}AA$	_	_	_
	624909.1	$16_{6,10}EE-15_{5,10}EE$	_		_
	624932.1	16 _{6,11} EE-15 _{5,11} EE	16.9	278.6	6
	624932.4	$16_{6,10}AE-15_{5,11}AE$	_	_	_
	624935.5	$16_{6,10}EE-15_{5,11}EE$		_	
	624935.8	$16_{6,10}AA-15_{5,11}AA$	_	_	_
	624936.4 626989.0	$16_{6,10}EA-15_{5,11}EA$ $35_{2,33}AE-34_{1,34}AE$	4.3	36.7	_
	626989.0	$35_{2,33}AE - 34_{1,34}AE$ $35_{2,34}EA - 34_{1,34}EA$	4.3	30.7 —	
	626989.0	$35_{2,34}EE - 34_{1,34}EE$	_	_	
	626989.1	$35_{2,33}AA-34_{1,34}AA$	_	_	
	627015.6	$13_{7,6}EA-12_{6,6}EA$	14.6	107.7	7
	627018.1	$13_{7,6}EE-12_{6,6}EE$	_	_	_
	627019.4	$13_{7,7}AE-12_{6,6}AE$	_	_	_
	627019.5	$13_{7,6}AE-12_{6,7}AE$	_	_	_
	627020.7	$13_{7,7}AA-12_{6,6}AA$	_	_	_
	627020.7	$13_{7,6}AA - 12_{6,7}AA$	_	_	_
	627022.0	13 _{7,7} EE-12 _{6,7} EE	_	_	_
	627023.4	$13_{7,7}EA-12_{6,7}EA$	12.0	963	_
	628657.9	$10_{8,3}EA-9_{7,2}EA$	12.9	86.2	
	628659.0 628660.2	$10_{8,3}EE-9_{7,2}EE \ 10_{8,3}AA-9_{7,2}AA$		_	
	628660.2	$10_{8,3}AA - 9_{7,2}AA$ $10_{8,2}AA - 9_{7,3}AA$			
	628662.0	$10_{8,3}AE - 9_{7,2}AE$	_	_	
	628662.0	$10_{8,2}AE - 9_{7,3}AE$	_	_	
	628663.2	$10_{8,2}^{8,2}EE-9_{7,3}^{7,3}EE$	_	_	
	628666.2	$10_{8,2}EA-9_{7,3}EA$	_	_	
	628775.8	$34_{4,30}AE-33_{3,31}AE$	1.9	16.4	
	628775.8	$34_{4,31}EA - 33_{3,30}EA$	_	_	
	628776.6	$34_{4,31}EE-33_{3,30}EE$	_	_	
	628777.3	$34_{4,30}AA - 33_{3,31}AA$	_	_	
	632286.6	$35_{1,34}AE-34_{4,31}AE$	2.3	9.1	
	632286.6	35 _{1,34} EA-34 _{4,30} EA	_	_	
	632289.5	35 _{1,34} EE-34 _{4,30} EE	_	_	
	632292.4 636468.8	$35_{1,34}AA - 34_{4,31}AA$ $20_{5,16}EA - 19_{4,15}EA$	6.4	42.4	
	636468.8	$20_{5,16}EA - 19_{4,15}EA$ $20_{5,16}AE - 19_{4,15}AE$		 27	
	636470.8	$20_{5,16}EE - 19_{4,15}EE$	_	_	
	636472.7	$20_{5,16}AA - 19_{4,15}AA$	_	_	
	637843.4	$34_{3,32}AA - 33_{4,29}AA$	2.9	29.8	
	637843.7	$34_{3,31}EE-33_{4,30}EE$	_	_	
	637844.0	$34_{3,31}EA-33_{4,30}EA$	_	_	
	637844.1	$34_{3,32}AE-33_{4,29}AE$	_	_	
	641361.9	$20_{5,15}AE-19_{4,16}AE$	2.2	19.4	
	641362.0	$20_{5,15}EA-19_{4,16}EA$	_	_	
	641363.9	$20_{5,15}EE-19_{4,16}EE$	_	_	
	641365.8 642289.2	$20_{5,15}AA-19_{4,16}AA 33_{6,27}AA-33_{1,32}AA$	2.0	26.3	8

TABLE 16—Continued

Molecule	v (MHz)	J_{K_p,K_m}	T_R^{*a} (K)	$\int T_R^* dv^a$ (K km s ⁻¹)	Notes
CH ₃ OCH ₃	642294.9	33 _{6,27} EE-33 _{1,32} EE	_	_	_
	642300.6	$33_{6,27}AE - 33_{1,32}AE$	_	_	_
	642300.6	$33_{6,27}EA - 33_{1,32}EA$	_	_	_
	642934.8	$30_{2,28}EA - 30_{3,28}EA$	2.5	48.0	
	642934.8 642940.4	$30_{2,28}AE-30_{3,27}AE \ 30_{2,28}EE-30_{3,28}EE$		_	
	642946.0	$30_{2,28}AA - 30_{3,27}AA$	_	_	
	643412.7	$32_{6,26}^{2,26}AA-32_{1,31}^{3,27}AA$	1.2	8.5	9
	643417.8	$32_{6,26}EE-32_{1,31}EE$	_	_	_
	643423.0	$32_{6,26}AE-32_{1,31}AE$	_	_	_
	643423.0	$32_{6,26}EA - 32_{1,31}EA$	22	— 41.5	_
	643643.1 643643.8	$17_{6,12}AE-16_{5,11}AE$ $17_{6,12}EE-16_{5,11}EE$	3.3	41.5	
	643646.2	$17_{6,12}AA - 16_{5,11}AA$	_	_	
	643646.9	$17_{6,11}^{6,12}EA-16_{5,11}^{5,11}EA$	_	_	
	643647.7	$17_{6,11}EE-16_{5,11}EE$	_	_	
	643690.6	$17_{6,12}EA-16_{5,12}EA$	3.1	24.1	
	643693.0	17 _{6,12} EE-16 _{5,12} EE	_	_	
	643694.4 643696.9	$17_{6,11}AE-16_{5,12}AE$ $17_{6,11}EE-16_{5,12}EE$	_	_	
	643697.1	$17_{6,11}EE - 10_{5,12}EE$ $17_{6,11}EA - 16_{5,12}EA$		_	
	643697.6	$17_{6,11}AA-16_{5,12}AA$	_	_	
	645923.8	$14_{7,7}EA-13_{6,7}EA$	10.5	98.0	
	645926.2	$14_{7,7}EE-13_{6,7}EE$	_	_	
	645927.5	14 _{7,8} <i>AE</i> –13 _{6,7} <i>AE</i>	_	_	
	645927.6 645928.6	$14_{7,7}AE-13_{6,8}AE$		_	
	645928.7	$14_{7,8}AA - 13_{6,7}AA \ 14_{7,7}AA - 13_{6,8}AA$	_	_	
	645930.0	14 _{7,8} EE-13 _{6,8} EE	_	_	
	645931.3	$14_{7,8}^{7,8}EA-13_{6,8}^{6,8}EA$	_	_	
	646088.8	$22_{4,18}AE-21_{3,19}AE$	2.5	18.2	
	646088.8	$22_{4,18}EA-21_{3,19}EA$	_	_	
	646090.9	$22_{4,18}EE-21_{3,19}EE$	_	_	
	646093.0 646766.7	$22_{4,18}AA - 21_{3,19}AA$ $30_{4,27}AA - 29_{1,28}AA$	1.9	12.8	
	646768.2	$30_{4,26}EE-29_{1,28}EE$		———	
	646769.8	$30_{4,26}EA-29_{1,28}EA$	_	_	
	646769.8	$30_{4,27}AE-29_{1,28}AE$	_	_	
	647196.2	$21_{3,18}EA-20_{2,19}EA$	2.1	16.4	
	647196.2	$21_{3,18}AE-20_{2,19}AE$	_	_	
	647199.7 647203.1	$21_{3,18}EE-20_{2,19}EE 21_{3,18}AA-20_{2,19}AA$	_	_	
	647455.2	$35_{6,29}AA - 35_{1,34}AA$	5.8	73.1	10
	647456.1	$35_{4,31}AE-34_{3,32}AE$	_	_	_
	647456.1	$35_{4,32}EA-34_{3,31}EA$	_	_	_
	647456.8	$35_{4,32}EE-34_{3,31}EE$	_	_	_
	647457.5	$35_{4,31}AA - 34_{3,32}AA$			_
	647608.4	$11_{8,4}EA-10_{7,3}EA$	8.7	74.5	
	647609.5 647610.6	$11_{8,4}EE-10_{7,3}EE$ $11_{8,4}AA-10_{7,3}AA$		_	
	647610.6	$11_{8,3}AA-10_{7,4}AA$	_	_	
	647612.4	$11_{8,3}AE-10_{7,4}AE$	_	_	
	647612.4	$11_{8,4}AE-10_{7,3}AE$	_	_	
	647613.6	$11_{8,3}EE-10_{7,4}EE$	_	_	
	647616.5	$11_{8,3}EA-10_{7,4}EA$	1 1	 2.1	
	653244.9 653244.9	$21_{5,17}EA-20_{4,16}EA 21_{5,17}AE-20_{4,16}AE$	1.1	2.1	
	653246.6	$21_{5,17}AE-20_{4,16}AE$ $21_{5,17}EE-20_{4,16}EE$	_	_	
	653248.4	$21_{5,17}AA - 20_{4,16}AA$		_	
	660458.8	$21_{5,16}AE-20_{4,17}AE$	3.6	30.6	11
	660458.8	$21_{5,16}EA-20_{4,17}EA$	_	_	_
	660460.4	21 _{5,16} EE-20 _{4,17} EE		_	_
	660462.2 662319.1	$21_{5,16}AA - 20_{4,17}AA$ $18_{6,13}EA - 17_{5,12}EA$			_
		13 H 4 - 1 / H 4	4.5	27.1	

TABLE 16—Continued

36.1	<i>y</i>	*	T_R^{*a}	$\int_{R} T_{R}^{*} dv^{a}$	NT :
Molecule	(MHz)	J_{K_p,K_m}	(K)	$(K \text{ km s}^{-1})$	Note
CH ₃ OCH ₃	662321.7	$18_{6,13}EE-17_{5,12}EE$		_	
	662323.6	$18_{6,13}AA - 17_{5,12}AA$			40
	662414.6	$18_{6,12}AE-17_{5,13}AE$	15.7	459.7	12
	662416.1	$18_{6,12}EA-17_{5,13}EA$	_	_	_
	662416.4 662417.5	$18_{6,12}EE-17_{5,13}EE$ $18_{6,12}AA-17_{5,13}AA$		_	
	664814.9	$15_{7,9}EA-14_{6,8}EA$	6.7	59.6	
	664817.2	$15_{7,9}EE-14_{6,8}EE$	_	_	
	664818.5	$15_{7,9}AE-14_{6,8}AE$	_	_	
	664818.7	$15_{7,8}AE-14_{6,9}AE$	_	_	
	664819.4	$15_{7,9}AA-14_{6,8}AA$	_	_	
	664819.7	$15_{7,8}AA-14_{6,9}AA$	_	_	
	664820.9	15 _{7,8} EE-14 _{6,9} EE	_	_	
	664822.2	$15_{7,8}EA-14_{6,9}EA$		 26.5	12
	665710.6	$44_{12,33}EA - 44_{11,34}EA$	3.3	36.5	13
	665711.2 665711.8	$44_{12,32}AE - 44_{11,33}AE$ $44_{12,33}AE - 44_{11,34}AE$		_	
	665712.4	$44_{12,32}EA - 44_{11,33}EA$	_	_	
	665716.1	$44_{12,33}EE-44_{11,34}EE$	_	_	_
	665717.0	$44_{12,32}EE-44_{11,33}EE$	_	_	_
	665721.4	$44_{12,32}AA-44_{11,33}AA$	_	_	_
	665721.8	$44_{12,33}AA - 44_{11,34}AA$	_	_	_
	666553.8	$12_{8,5}EA-11_{7,4}EA$	5.3	47.6	
	666554.9	$12_{8,5}EE-11_{7,4}EE$	_	_	
	666555.9	$12_{8,5}AA-11_{7,4}AA$	_	_	
	666555.9	$12_{8,4}AA-11_{7,5}AA$	_	_	
	666557.8 666557.8	$12_{8,5}AE-11_{7,4}AE \ 12_{8,4}AE-11_{7,5}AE$		_	
	666558.8	$12_{8,4}AE-11_{7,5}AE$ $12_{8,4}EE-11_{7,5}EE$	_	_	
	666561.8	$12_{8,4}EA - 11_{7,5}EA$	_	_	
	668073.0	$9_{9,0}^{8,4}AA-8_{8,1}AA$	5.8	66.0	
	668073.0	$9_{9,1}AA - 8_{8,0}AA$	_	_	
	668073.7	$9_{9,1}EE-8_{8,1}EE$	_	_	
	668074.4	$9_{9,1}EA - 8_{8,1}EA$	_	_	
	668077.1	$9_{9,0}EE - 8_{8,0}EE$	_	_	
	668077.9	$9_{9,0}AE - 8_{8,1}AE$	_		
	668077.9 668081.2	9 _{9,1} AE-8 _{8,0} AE 9 _{9,0} EA-8 _{8,0} EA	_	_	
	669407.6	$22_{5,18}EA - 21_{4,17}EA$	2.1	12.0	
	669407.6	$22_{5,18}AE - 21_{4,17}AE$			
	669409.0	$22_{5,18}EE-21_{4,17}EE$		_	
	669410.5	$22_{5,18}AA-21_{4,17}AA$		_	
	669485.2	$36_{12,24}AE-36_{11,25}AE$	2.7	26.5	
	669485.2	$36_{12,25}AE-36_{11,26}AE$	_	_	
	669485.6	$36_{12,24}EA - 36_{11,25}EA$	_	_	
	669486.9	36 _{12,25} EE-36 _{11,26} EE	_	_	
	669487.2	36 _{12,24} EE-36 _{11,25} EE		_	
	669488.9 669488.9	$36_{12,24}AA - 36_{11,25}AA$ $36_{12,25}AA - 36_{11,26}AA$		_	
	670088.8	$34_{12,23}EA - 34_{11,24}EA$	5.5	39.1	14
	670089.2	$34_{12,23}EA - 34_{11,23}AE$ $34_{12,22}AE - 34_{11,23}AE$			_
	670089.2	$34_{12,23}AE-34_{11,24}AE$	_	_	_
	670089.6	$34_{12,22}^{12,23}EA-34_{11,23}^{11,24}EA$	_	_	_
	670090.1	$34_{12,23}EE-34_{11,24}EE$	_	_	_
	670090.6	$34_{12,22}EE-34_{11,23}EE$	_	_	_
	670091.4	$34_{12,22}AA - 34_{11,23}AA$	_	_	_
	670091.4	$34_{12,23}AA - 34_{11,24}AA$		_	_
	671161.4	$29_{12,17}AA - 29_{11,18}AA$	3.2	35.4	
	671161.4 671161.4	$29_{12,18}AA - 29_{11,19}AA$		_	
	671161.4 671161.5	$29_{12,17}EE-29_{11,18}EE \ 29_{12,17}EA-29_{11,18}EA$	_	_	
	671161.3	29 _{12,18} EE-29 _{11,19} EE			
	6/1162.1	291217AE-291110AE			
	671162.1 671162.1	$29_{12,17}AE-29_{11,18}AE \ 29_{12,18}AE-29_{11,19}AE$	_	_	

TABLE 16—Continued

Molecule	v (MHz)	${J}_{K_p,K_m}$	T_R^{*a} (K)	$\int T_R^* dv^a$ (K km s ⁻¹)	Notes
CH ₃ OCH ₃	671208.9	36 _{3,34} AA-35 _{4,31} AA	1.9	13.3	
	671209.1	$36_{3,33}EE-35_{4,32}EE$	_		
	671209.3	$36_{3,33}EA-35_{4,32}EA$	_	_	
	671209.3	$36_{3,34}AE-35_{4,31}AE$	_	_	
	671312.8	$28_{12,17}AA - 28_{11,18}AA$	2.2	49.9	
	671312.8	$28_{12,16}AA - 28_{11,17}AA$	_	_	
	671313.1	28 _{12,16} EE-28 _{11,17} EE	_	_	
	671313.4	28 _{12,16} EA-28 _{11,17} EA	_	_	
	671313.8	28 _{12,17} EE-28 _{11,18} EE	_	_	
	671314.1 671314.1	$28_{12,17}AE - 28_{11,18}AE$ $28_{12,16}AE - 28_{11,17}AE$		_	
	671314.7	$28_{12,17}EA-28_{11,18}EA$	_	_	
	672903.4	$23_{4,19}AE-22_{3,20}AE$	6.2	62.3	15
	672903.4	$23_{4,19}EA-22_{3,20}EA$	_	_	_
	672905.5	$23_{4,19}^{4,19}EE-22_{3,20}^{3,20}EE$	_	_	_
	672907.5	$23_{4,19}AA-22_{3,20}AA$	_	_	
	679791.2	$22_{5,17}AE-21_{4,18}AE$	4.4	46.9	
	679791.2	$22_{5,17}EA-21_{4,18}EA$	_	_	
	679792.8	22 _{5,17} EE-21 _{4,18} EE	_	_	
	679794.2	$22_{5,17}AA - 21_{4,18}AA$			1.0
	680039.1	$38_{2,37}EA - 37_{1,37}EA$	3.8	31.0	16
	680039.1 680039.2	$38_{2,36}AE-37_{1,37}AE 38_{2,37}EE-37_{1,37}EE$		_	
	680039.2	$38_{2,37}EE - 37_{1,37}EE$ $38_{2,36}AA - 37_{1,37}AA$			
	680041.9	$38_{1,38}AE-37_{2,35}AE$	_	_	_
	680041.9	$38_{1,38}EA - 37_{2,36}EA$	_	_	_
	680042.0	$38_{1,38}EE-37_{2,36}EE$	_	_	_
	680042.0	$38_{1,38}AA - 37_{2,35}AA$	_	_	
	680923.9	$19_{6,14}EA-18_{5,13}EA$	7.2	52.4	
	680924.6	$19_{6,14}AE-18_{5,13}AE$	_	_	
	680925.8	19 _{6,14} EE-18 _{5,13} EE	_	_	
	680927.2	$19_{6,14}AA - 18_{5,13}AA$	_		
	681090.1	$19_{6,13}AE-18_{5,14}AE$	5.5	29.2	
	681090.9 681091.6	$19_{6,13}EA-18_{5,14}EA$	_	_	
	681091.0	$19_{6,13}EE-18_{5,14}EE$ $19_{6,13}AA-18_{5,14}AA$		_	
	683685.8	$16_{7,10}EA-15_{6,9}EA$	11.1	81.1	
	683687.9	$16_{7,10}EE-15_{6,9}EE$	_	_	
	683689.0	$16_{7,10}AE-15_{6,9}AE$	_	_	
	683689.6	$16_{7,9}^{7,10}AE-15_{6,10}^{6,9}AE$	_	_	
	683689.9	$16_{7,10}AA-15_{6,9}AA$	_	_	
	683690.4	$16_{7,9}AA-15_{6,10}AA$	_	_	
	683691.5	$16_{7,9}EE-15_{6,10}EE$	_		
	683692.8	$16_{7,9}EA-15_{6,10}EA$			
	684325.2	$34_{4,31}AA - 33_{1,32}AA$	4.7	48.3	
	684327.6 684330.1	$34_{4,30}EE-33_{1,32}EE$	_	_	
	684330.1	$34_{4,30}EA - 33_{1,32}EA$ $34_{4,31}AE - 33_{1,32}AE$	_	_	
	684840.9	$23_{5,19}EA-22_{4,18}EA$	4.4	27.0	
	684840.9	$23_{5,19}AE-22_{4,18}AE$	_	_	
	684842.1	23 _{5,19} EE-22 _{4,18} EE	_	_	
	684843.3	$23_{5,19}AA-22_{4,18}AA$	_	_	
	685336.6	$22_{3,19}EA-21_{2,20}EA$	1.9	16.4	
	685336.6	$22_{3,19}AE-21_{2,20}AE$	_	_	
	685340.2	$22_{3,19}EE-21_{2,20}EE$	_	_	
	685343.8	$22_{3,19}AA - 21_{2,20}AA$	10.0	250.0	4.5
	685492.6 685403.6	$13_{8,6}EA-12_{7,5}EA$	19.8	250.0	17
	685493.6 685494.6	$13_{8,6}EE-12_{7,5}EE$	_	_	_
	685494.6 685494.6	$13_{8,6}AA-12_{7,5}AA$ $13_{8,5}AA-12_{7,6}AA$	_	_	_
	685496.4	$13_{8,6}AE-12_{7,6}AE$	_	_	_
	685496.4	$13_{8,5}AE - 12_{7,6}AE$	_	_	_
	685497.4	$13_{8,5}EE-12_{7,6}EE$	_	_	_
	685500.2	$13_{8,5}^{8,5}EA-12_{7,6}^{7,0}EA$	_	_	_
		$10_{9,2}AA - 9_{8,1}AA$	12.4		

TABLE 16—Continued

	v O SIII)	_	T_R^{*a}	$\int_{R} T_{R}^{*} dv^{a}$	•
Molecule	(MHz)	J_{K_p,K_m}	(K)	(K km s ⁻¹)	Notesa
CH_3OCH_3	687030.6	$10_{9,1}AA - 9_{8,2}AA$	_	_	
	687031.2	$10_{9,2}EE-9_{8,2}EE$	_		
	687032.0 687034.6	$10_{9,2}EA - 9_{8,2}EA \ 10_{9,1}EE - 9_{8,1}EE$	_	_	
	687035.3	$10_{9,2}AE-9_{8,1}AE$	_	_	
	687035.3	$10_{9,1}^{9,2}AE-9_{8,2}^{8,1}AE$	_	_	
	687038.6	$10_{9,1}EA - 9_{8,1}EA$	_	_	
	695432.6	35 _{4,32} AA-34 _{1,33} AA	4.8	49.6	
	695435.1 695437.5	35 _{4,31} EE-34 _{1,33} EE 35 _{4,31} EA-34 _{1,33} EA	_	_	
	695437.5	$35_{4,32}AE - 34_{1,33}EA$ $35_{4,32}AE - 34_{1,33}AE$	_	_	
	698347.5	$40_{6,34}EE-39_{3,36}EE$	1.3	2.1	
	698354.6	$40_{6,34}AA - 39_{3,37}AA$	_	_	
	699432.7	$24_{5,20}EA-23_{4,19}EA$	3.3	22.0	
	699432.7	24 _{5,20} AE-23 _{4,19} AE	_	_	
	699433.6 699434.5	$24_{5,20}EE-23_{4,19}EE$ $24_{5,20}AA-23_{4,19}AA$	_	_	
	699437.3	$20_{6,15}EA-19_{5,14}EA$	_	_	
	699437.6	$20_{6,15}AE-19_{5,14}AE$	_	_	
	699438.8	$20_{6,15}EE-19_{5,14}EE$	_	_	
	699440.0	$20_{6,15}AA-19_{5,14}AA$	_	_	
	699453.1	$23_{5,18}AE-22_{4,19}AE$	2.5	16.0	
	699453.2 699454.4	$23_{5,18}EA-22_{4,19}EA$ $23_{5,18}EE-22_{4,19}EE$	_	_	
	699455.8	$23_{5,18}AA - 22_{4,19}AA$	_	_	
	699719.9	$20_{6,14}^{3,16}AE-19_{5,15}^{4,19}AE$	1.7	7.3	
	699720.2	$20_{6,14}EA-19_{5,15}EA$	_	_	
	699721.1	$20_{6,14}EE-19_{5,15}EE$	_	_	
	702532.5 702534.5	$17_{7,11}EA-16_{6,10}EA$	5.2	45.6	
	702535.2	$17_{7,11}EE-16_{6,10}EE \ 17_{7,11}AE-16_{6,10}AE$		_	
	702536.1	$17_{7,11}AA-16_{6,10}AA$	_	_	
	702536.5	$17_{7,10}AE-16_{6,11}AE$	_	_	
	702537.2	$17_{7,10}AA-16_{6,11}AA$	_	_	
	702538.1 702539.3	$17_{7,10}EE-16_{6,11}EE$	_	_	
	704422.9	$17_{7,10}EA-16_{6,11}EA $ $14_{8,7}EA-13_{7,6}EA$	5.9	50.1	
	704423.8	14 _{8,7} EE-13 _{7,6} EE	_	_	
	704424.8	$14_{8,7}AA-13_{7,6}AA$	_	_	
	704424.8	$14_{8,6}AA - 13_{7,7}AA$	_	_	
	704426.6	$14_{8,7}AE-13_{7,6}AE$	_	_	
	704426.6 704427.6	$14_{8,6}AE-13_{7,7}AE \ 14_{8,6}EE-13_{7,7}EE$		_	
	704427.0	$14_{8,6}EE-13_{7,7}EE$ $14_{8,6}EA-13_{7,7}EA$	_	_	
	705985.8	$11_{9,2}AA-10_{8,3}AA$	7.4	45.8	
	705985.8	$11_{9,3}AA-10_{8,2}AA$	_	_	
	705986.6	$11_{9,3}EE-10_{8,3}EE$	_	_	
	705987.3	$11_{9,3}EA-10_{8,3}EA \ 11_{9,2}EE-10_{8,2}EE$	_	_	
	705989.8 705990.5	$11_{9,2}EE-10_{8,2}EE$ $11_{9,2}AE-10_{8,3}AE$		_	
	705990.5	$11_{9,3}AE - 10_{8,3}AE$	_	_	
	705993.8	$11_{9,2}EA-10_{8,2}EA$	_	_	
	707335.2	$36_{4,33}AA - 35_{1,34}AA$	1.9	11.2	18
	707337.6	36 _{4,32} EE-35 _{1,34} EE	_	_	_
	707340.1 707340.1	$36_{4,32}EA - 35_{1,34}EA$	_	_	_
	712723.8	$36_{4,33}AE - 35_{1,34}AE$ $44_{1,43}AE - 43_{4,40}AE$	2.0	11.0	
	712723.8	$44_{1,43}EA - 43_{4,39}EA$		_	
	717837.8	$21_{6,16}EA - 20_{5,15}EA$	1.8	9.4	
		$21_{6,16}AE-20_{5,15}AE$	_	_	
	717837.9				
	717838.9	$21_{6,16}EE-20_{5,15}EE$	_	_	
	717838.9 717840.0	$21_{6,16}EE-20_{5,15}EE \\ 21_{6,16}AA-20_{5,15}AA$	_ _ 40	 24 8	
	717838.9	$21_{6,16}EE-20_{5,15}EE$	 4.9 	 24.8 	

TABLE 16—Continued

Molecule	v (MHz)	J_{K_p,K_m}	T*a (K)	$\int T_R^* dv^a$ (K km s ⁻¹)	Notesa
CH ₃ OCH ₃	718307.4	21 _{6,15} AA-20 _{5,16} AA	_	_	
33	719557.6	$24_{5,19}AE-23_{4,20}AE$	1.1	1.9	19
	719557.6	$24_{5,19}^{5,19}EA-23_{4,20}^{4,20}EA$	_	_	_
	719558.7	$24_{5,19}EE-23_{4,20}EE$	_	_	_
	719559.8	$24_{5,19}AA-23_{4,20}AA$	_	_	_
	721352.8	$18_{7,12}EE-17_{6,11}EE$	2.3	17.5	
	721353.2	$18_{7,12}AE-17_{6,11}AE$	_	_	
	721353.8	$18_{7,12}AA-17_{6,11}AA$	_	_	
	721355.8	$18_{7,11}AE-17_{6,12}AE$	_	_	
	721356.4	$18_{7,11}AA-17_{6,12}AA$	_	_	
	721356.9	$18_{7,11}EE-17_{6,12}EE$	_	_	
	721357.9	$18_{7,11}EA-17_{6,12}EA$	_	_	
	722575.2	$39_{5,35}AA - 38_{4,34}AA$	8.1	48.5	20
	722575.3	$39_{5,35}EE-38_{4,35}EE$	_	_	_
	722575.4	$39_{5,35}EA-38_{4,35}EA$	_	_	_
	722575.4	$39_{5,35}AE-38_{4,34}AE$	_	_	_
	723342.9	$15_{8,8}EA-14_{7,7}EA$	6.1	52.7	21
	723343.8	15 _{8.8} EE-14 _{7.7} EE	_	_	_
	723344.6	$15_{8,8}AA-14_{7,7}AA$	_	_	_
	723344.6	$15_{8,7}AA-14_{7,8}AA$	_	_	_
	723346.4	$15_{8.8}AE-14_{7.7}AE$	_	_	_
	723346.4	$15_{8,7}AE-14_{7,8}AE$	_	_	_
	723350.0	$15_{8,7}EA-14_{7,8}EA$	_	_	_
	724937.9	$12_{9,4}AA-11_{8,3}AA$	5.9	52.7	
	724937.9	$12_{9,3}AA-11_{8,4}AA$	_	_	
	724938.6	$12_{9.4}EE-11_{8.4}EE$	_	_	
	724939.4	$12_{9,4}EA-11_{8,4}EA$	_	_	
	724941.8	$12_{9,3}EE-11_{8,3}EE$	_	_	
	724942.5	$12_{9,4}AE-11_{8,3}AE$	_	_	
	724942.5	$12_{9,3}AE-11_{8,4}AE$	_	_	
	724945.7	$12_{9,3}EA-11_{8,3}EA$	_	_	

Notes.—(1) Too strong, possible blend with U line. (2) Too strong, possible Blend with U line. (3) Blend with 13 CH₃OH at 612934.2. (4) Blend with SO₂ at 613676.5. (5) Blend with CH₃CN at 624878.9 and SO₂ at 624887.5. (6) Blend with SiH at 624920.1, 624926.4 and 624935.8; and CH₃CN at 624914.6 and 624926.4. (7) Blend with CH₃OH at 627013.0. (8) Blend with CH₃CN at 642280.7. (9) Blend with CH₃CH₂CN at 643412.7. (10) Blend with SO₂ at 647455.2. (11) Blend with SO₂ at 660472.7. (12) Blend with SO₂ at 662414.6. (13) Blend with CH₃OH₃ at 665705.8. (14) Blend with NO at 670091.7. (15) Blend with CH₃OH at 672903.7. (16) Blend with CN at 680947.4. (17) Blend with CH₃OH at 685505.1. (18) Blend with 13 CH₃OH at 707342.3. (19) Questionable assignment. (20) Blend with CH₃OH at 722566.7 and 24 S at 722576.8. (21) Blend CH₃OH at 723336.8. This table is available in machine-readable form in the electronic edition of *The Astrophysical Journal*.

^a Dashes indicate that the line is part of a blend, and intensity and area are given by the values preceding it.

TABLE 17 TABLE OF C_2H_5CN Molecules

		Table of C ₂ H ₅ CN Molecules		C 5774 1 0	
Molecule	v (MHz)	${J}_{K_p,K_m}$	T_R^{*a} (K)	$\int T_R^* dv^a$ (K km s ⁻¹)	Notesa
C ₂ H ₅ CN	608021.4	$24_{9,15} - 23_{8,16}$	5.9	34.2	
C ₂ 113C1	612227.4	$35_{7,28} - 34_{6,29}$	5.1	68.7	
	615248.7	$30_{8,23} - 29_{7,22}$	2.5	20.5	
	617180.3	$72_{2,71}^{3,23} - 71_{2,70}^{7,22}$	2.9	3.6	
	617180.5	$72_{1,71}^{2,71} - 71_{1,70}^{2,70}$	_	_	
	617234.4	$69_{12,58} - 68_{12,57}$	4.9	24.9	
	617234.4	$69_{12,57} - 68_{12,56}$	_	_	
	617618.8	$69_{11,58} - 68_{11,57}$	_	_	
	617919.0	$41_{6,35} - 40_{5,36}$	6.1	108.4	1
	619340.6	$43_{6,38} - 42_{5,37}$	8.0	62.2	2
	623779.8	$72_{3,70} - 71_{3,69}$	4.8	19.0	
	625024.7	$44_{6,39} - 43_{5,38}$	1.7	5.7	
	629790.6	$71_{4,67} - 70_{4,66}$	2.0	10.1	
	634766.7	$71_{13,59} - 70_{13,58}$	2.9	19.7	
	634766.7	$71_{13,58} - 70_{13,57}$	_	_	
	635414.4	$71_{23,48} - 70_{23,47}$	1.8	11.1	
	635414.4	$71_{23,49} - 70_{23,48}$	_	_	
	635697.2	$75_{1,75} - 74_{1,74}$	3.3	39.2	
	635697.2	$75_{0,75} - 74_{0,74}$	_	_	
	637723.4	$12_{12,1} - 11_{11,0}$	1.7	9.0	
	638119.4	$38_{7,31} - 37_{6,32}$	2.1	17.4	
	646708.8	$13_{12,2} - 12_{11,1}$	_	_	
	650569.4	$34_{8,27} - 33_{7,26}$	2.1	26.9	
	650594.7	$76_{1,75} - 75_{1,74}$	_	_	
	650680.9	$42_{4,38} - 41_{3,39}$	_	_	
	652534.2	$73_{13,61} - 72_{13,60}$	_		
	653323.8	$73_{11,62} - 72_{11,61}$	1.9	19.9	
	654030.8	$73_{10,64} - 72_{10,63}$	2.0	18.2	
	654519.6	$40_{7,34} - 39_{6,33}$	2.1	20.2	
	659989.2	$56_{6,51} - 55_{5,50}$	2.7	14.4	
	663842.8	74 _{7,68} – 73 _{7,67}	1.7	12.5	
	663951.2	$20_{11,9} - 19_{10,10}$	2.5	18.7	
	663951.2	$20_{11,10} - 19_{10,9}$	2.2	20.6	
	664682.9	$15_{12,3} - 14_{11,4}$	3.3	39.6	
	664682.9	$15_{12,4} - 14_{11,3}$	_	_	
	666418.4	$20_{15,5} - 20_{14,6}$	_	_	
	666518.4	$22_{15,7} - 22_{14,8}$	_	_	
	666624.8 666908.8	$24_{15,10} - 24_{14,11}$	3.0	57.4	
	666908.8	$29_{15,14} - 29_{14,15}$	3.0	37.4	
	667026.8	$29_{15,15} - 29_{14,16}$	3.5	46.0	
	667026.8	$31_{15,17} - 31_{14,18}$	3.3	40.0	
	667718.9	$31_{15,16} - 31_{14,17}$	2.0	 19.4	
	667718.9	$43_{15,28} - 43_{14,29}$	2.0	19.4	
	667817.3	$43_{15,29} - 43_{14,30}$	2.2	38.4	
	667817.3	$45_{15,31} - 45_{14,32} $ $45_{15,30} - 45_{14,31}$			
	667949.4	$48_{15,30} - 48_{14,34}$ $48_{15,33} - 48_{14,34}$	2.7	14.8	
	667949.4	$48_{15,34} - 48_{14,35}$	2.7		
	670499.6	$31_{9,23} - 30_{8,22}$	2.6	26.1	
	670499.6	$31_{9,23} - 30_{8,22}$ $31_{9,22} - 30_{8,23}$	2.0	20.1	
	679393.0	$31_{9,22} 30_{8,23} \\ 32_{9,24} - 31_{8,23}$	3.4	26.6	
	679393.0	$32_{9,24} - 31_{8,23}$ $32_{9,23} - 31_{8,24}$	J. T		
	680094.6	$76_{11,65} - 75_{11,64} \ 32_{9,23} - 31_{8,24}$		_	
	680481.1	$78_{4,75} - 77_{4,74}$	2.9	31.9	
	680507.1	$78_{3,75} - 77_{3,74}$	2.4	26.5	
	680841.3	$43_{7,36} - 42_{6,37}$	2.4	28.9	
	681913.1	$22_{11,11} - 21_{10,12}$	1.5	2.7	
	681913.1	$22_{11,11} - 21_{10,12} 22_{11,12} - 21_{10,11}$			
	681946.1	$76_{9,68} - 75_{9,67}$	3.5	28.2	
	682583.8	$76_{8,69} - 75_{8,68}$	3.5	40.8	
	682660.1	$17_{12,5} - 16_{11,6}$	4.4	43.1	
	682660.1	$17_{12,6} - 16_{11,5}$	_	_	
	683960.4	$80_{1,79} - 79_{1,78}$	_	_	
	685611.2	$38_{8,30} - 37_{7,31}$	3.0	40.3	
		8,30/,31	2.3		

TABLE 17—Continued

Molecule	v (MHz)	${J}_{K_p,K_m}$	T*a (K)	$\int T_R^* dv^a$ (K km s ⁻¹)	Notes ^a
C ₂ H ₅ CN	691649.4	$18_{12.7} - 17_{11.6}$		_	
23	698787.6	$29_{10,19} - 28_{9,20}$	0.8	2.9	3
	698787.6	$29_{10,20} - 28_{9,19}$	_		_
	700639.0	$19_{12.7} - 18_{11.8}$	1.9	14.9	
	700639.0	$19_{12.8} - 18_{11.7}$			
	707727.9	$30_{10,20} - 29_{9,21}$	1.8	30.8	
	707727.9	$30_{10,21} - 29_{9,20}$	_	_	
	708951.2	$83_{2.82}^{10,21} - 82_{2.81}^{9,20}$	2.9	24.4	
	708951.2	$83_{1.82}^{2,82} - 82_{1.81}^{2,81}$	_	_	
	709628.6	$20_{12.8} - 19_{11.9}$	1.7	9.2	
	709628.6	$20_{12.9}^{12.8} - 19_{11.8}^{11.9}$	_	_	
	711304.5	$18_{16,2}^{12,9} - 18_{15,3}^{11,8}$	1.7	19.2	
	711304.5	$18_{16.3}^{10,2} - 18_{15.4}^{13,3}$	_		
	712527.6	$37_{16,22} - 37_{15,23}$	2.6	25.4	
	712527.6	$37_{16,21} - 37_{15,22}$		_	
	712673.9	$39_{16,24}^{16,24} - 39_{15,25}^{15,25}$	1.7	11.9	
	712673.9	$39_{16,23}^{16,23} - 39_{15,24}^{15,25}$	_	_	
	712746.9	$40_{16,25} - 40_{15,26}$	4.0	22.7	
	712746.9	$40_{16,24} - 40_{15,25}$	_	_	
	716662.4	$31_{10.21} - 30_{9.22}$	_	_	
	718618.1	$21_{12,10} - 20_{11,9}$	_	_	
	719178.6	$16_{13.3} - 15_{12.4}$	2.3	5.5	
	719178.6	$16_{13,4} - 15_{12,3}$	_	_	

Notes.—(1) Too strong, possible blend with U line. (2) Too strong, possible blend with U line. (3) Questionable assignment. This table is available in machine-readable form in the electronic edition of *The Astrophysical Journal*.

^a Dashes indicate that the line is part of a blend, and intensity and area are given by the values

preceding it.

	TABL	T_R^*	$\int T_R^* dv$		Molecule	v (MHz)	T* _R (K)	$\int_{K} T_R^* dv$ (K km s ⁻¹)	Notes
Molecule	(MHz)	(K)	$(K \text{ km s}^{-1})$	Notes	<i>U</i>	649104.5	7.1	134.7	
<i>U</i>	608269.3	5.2	72.3			650534.1	3.0	22.6	
·	609131.2	2.1	9.7	1		651535.9	5.7	73.0	
	615276.7	6.1	96.7			653535.4	1.3	12.2	
	616226.3	3.7	57.7			653572.1	1.2	12.7	
	616603.6	3.2	34.1			653598.0 653711.0	0.9 2.7	6.3 39.3	
	617097.1	6.8	61.3			653931.4	3.2	35.6	
	617149.5	7.0	87.5			654131.1	2.3	37.1	
	619157.0 619318.7	5.5 11.7	58.3 158.6			654396.9	1.9	22.5	
	619365.2	9.8	87.6			654533.4	2.0	16.9	
	623296.5	11.7	67.1			654993.4	3.1	17.2	
	623316.9	9.6	77.7			656724.0	4.1	35.1	
	623487.1	10.4	114.5			657721.9	2.9	41.1	
	623570.7	19.9	388.4			657933.8	6.4	61.1	
	623644.1	13.4	162.4			658031.5 658714.7	3.2 6.7	20.0 100.4	
	623848.6	5.8	99.6			658928.3	2.6	8.3	
	624072.8	4.7	30.2			659188.8	3.2	39.8	
	624263.6 624460.9	4.9 8.5	34.9 147.2			659495.2	9.9	198.8	
	624778.0	5.9	61.2			660785.2	2.8	26.4	
	625063.4	4.8	44.3			662087.2	1.6	11.8	
	625335.0	4.6	49.0			663639.1	2.2	33.8	
	625668.1	4.9	22.7			664780.9	3.5	34.4	
	626724.8	3.7	26.2			665814.5	3.4	39.5	
	627476.1	6.1	38.8			666821.3	2.5	16.9	
	628093.4	2.7	15.1			670011.3	2.4	20.2	
	628889.7	5.4	35.2			670852.2 670894.5	5.3 3.3	43.6 39.4	
	629696.0	2.3	7.3			671408.4	3.1	124.0	
	630376.0 632852.2	3.4	24.3 17.7			671716.8	1.7	12.6	
	633023.5	1.9 3.7	37.0			672447.3	1.9	11.2	
	633114.6	3.4	18.0			673072.1	4.0	56.9	
	633674.4	3.9	55.9			673559.9	2.5	72.1	
	633802.5	3.2	24.0			674284.5	2.0	13.0	
	633832.8	7.2	61.3			675984.0	3.7	24.6	
	633860.1	5.0	50.7			677112.4 677273.0	2.8	37.8	
	633891.1	4.6	27.4			677567.6	3.6 3.6	28.9 42.9	
	633898.3	4.6	27.6			677919.9	2.9	16.8	
	633926.9	5.2 5.4	25.9 51.3			678358.0	2.0	20.4	
	634081.2 634584.6	6.7	0.0			678417.7	4.3	55.5	
	634634.0	2.8	24.2			678546.9	3.6	37.0	
	634878.9	9.1	107.6			678710.4	3.3	21.7	
	635085.0	4.2	36.5			679760.8	3.7	18.1	
	635295.3	1.3	3.2			680247.8	4.4	52.5	
	635904.9	2.5	32.7			682092.3	3.0	28.5	
	637863.7	4.1	44.4			683170.8 683476.8	2.8 4.4	39.7 25.3	
	638084.3	2.6	46.7			683510.6	4.4 4.6	63.7	
	638220.4	2.0	27.5			684296.1	3.9	44.3	
	640981.1 641224.8	1.6	1.7			684430.0	6.6	52.8	
	642136.2	1.4 2.0	16.3 37.7			687544.7	5.1	51.6	2
	642552.7	1.5	11.9			687580.7	4.8	98.7	3
	642739.0	6.6	107.1			687696.7	3.0	34.9	
	642762.1	2.2	14.0			687718.3	3.1	16.8	
	643728.4	2.3	34.3			689289.6	2.3	15.0	_
	644185.9	3.5	18.3			690672.9	4.3	35.5	4
	646742.0	1.9	30.1			692674.4	2.5	14.9	
	646927.8	1.8	23.6			692726.2 693790.3	4.0 3.1	59.1 43.8	
	647319.8	1.9	19.8			693790.3	6.8	43.8 42.1	
	647418.5	2.8	23.9			695067.1	4.7	87.2	
	648119.8 648134.0	6.4 5.2	72.8 41.6			695525.3	2.1	11.5	
			71.0						

TABLE 18—Continued

Molecule	v (MHz)	T* (K)	$\int_{K} T_R^* dv$ (K km s ⁻¹)	Notes						
<i>U</i>	695773.7	2.1	18.5							
	697061.3	3.5	41.4							
	697297.5	1.9	11.2							
	697500.6	2.1	21.9							
	697660.7	1.7	12.6							
	697761.5	2.6	22.0							
	699071.6	3.5	37.7							
	706631.8	5.7	133.8							
	708216.1	2.8	21.9							
	708654.2	3.7	28.3							
	709006.3	4.1	35.7							
	709308.3	6.7	195.9							
	709466.2	5.1	53.4							
	711316.9	1.7	19.2							
	713409.6	3.6	30.7							
	714375.3	2.0	16.2							
	714455.9	3.2	40.8	5						
	714617.5	2.8	41.5							
	714971.8	4.4	28.6							
	715195.9	3.9	60.8							
	714984.0	5.2	41.3							
	715407.1	2.9	25.4							
	715446.0	3.7	47.1							
	715767.2	3.3	64.7							
	716167.5	5.1	69.5							
	717308.3	3.1	25.6							
	719948.0	2.7	22.9							
	723866.6	1.5	3.9							

Notes.—(1) Questionable assignment. (2) Questionable assignment. (3) Questionable assignment. (4) Questionable assignment. (5) Questionable assignment. This table is available in machine-readable form in the electronic edition of *The Astrophysical Journal*.

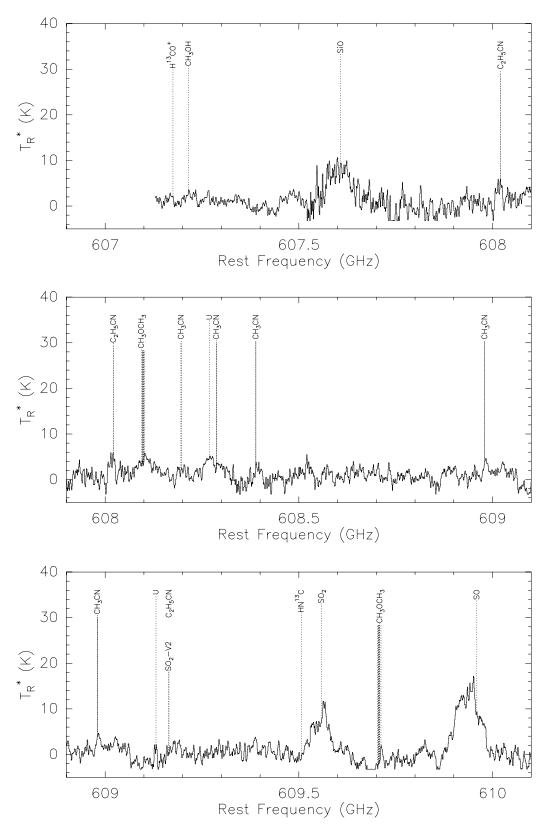


Fig. 4.—Line survey from 607–725 GHz. All observed lines are marked. The adopted source velocity is 9 km s $^{-1}$, and the spectra are plotted with 1 MHz resolution. Notice the gap between 620 and 623 GHz.

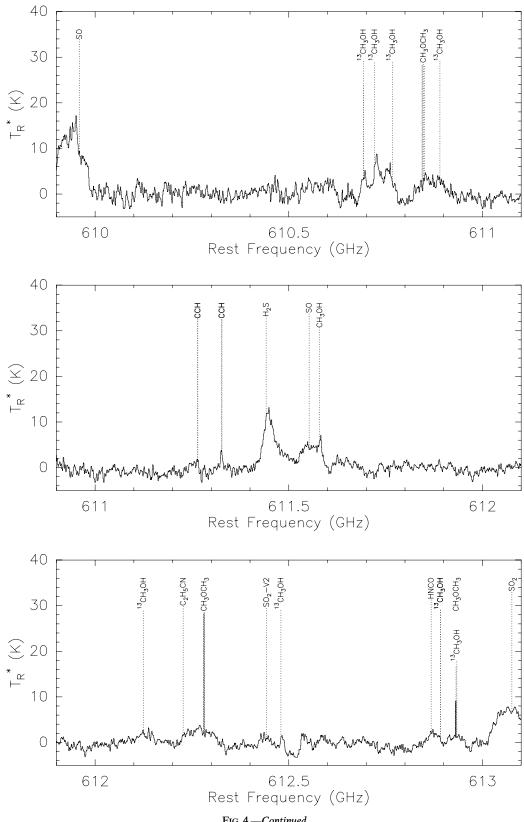


Fig. 4.—Continued

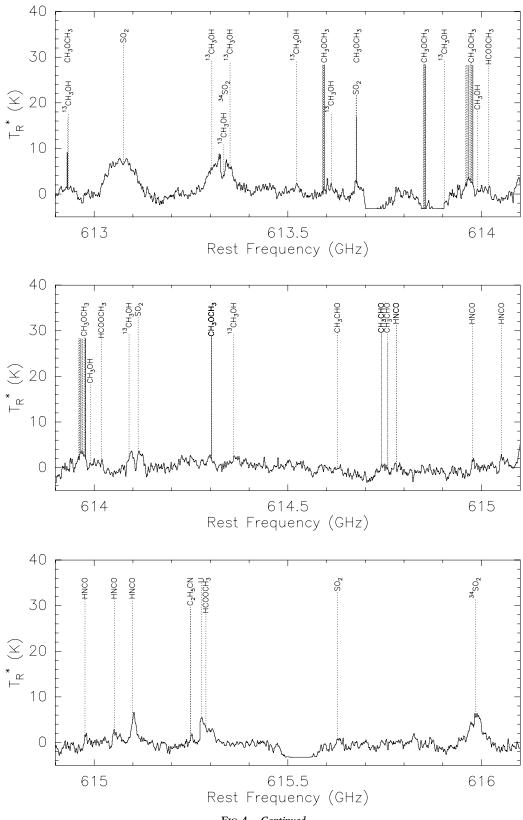


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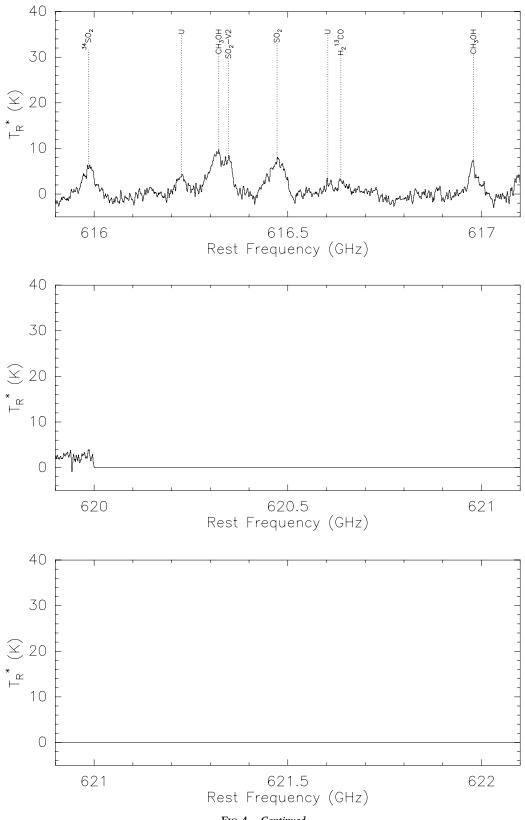
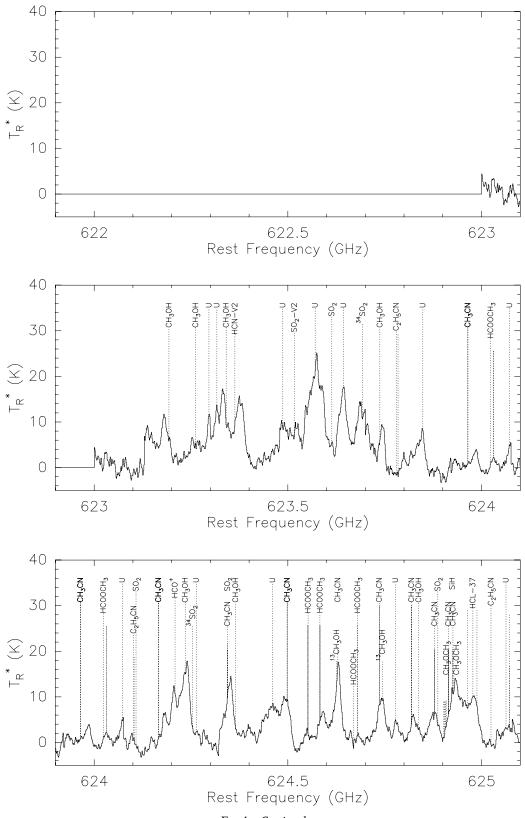


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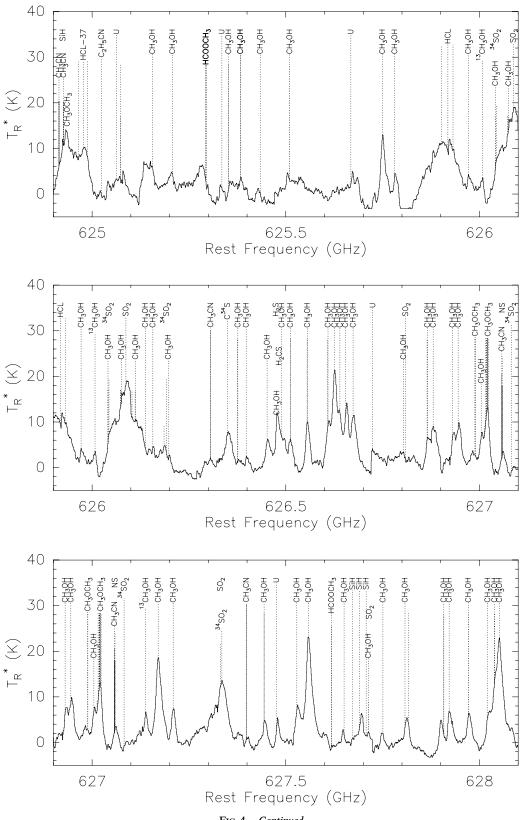


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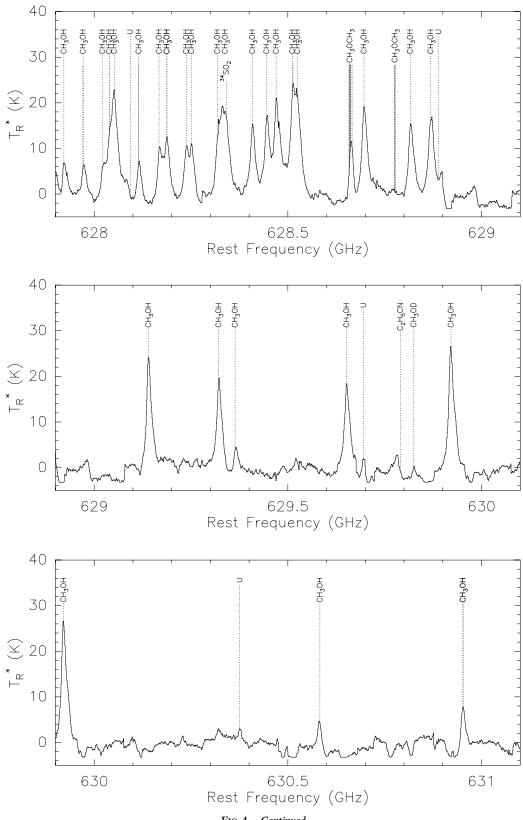


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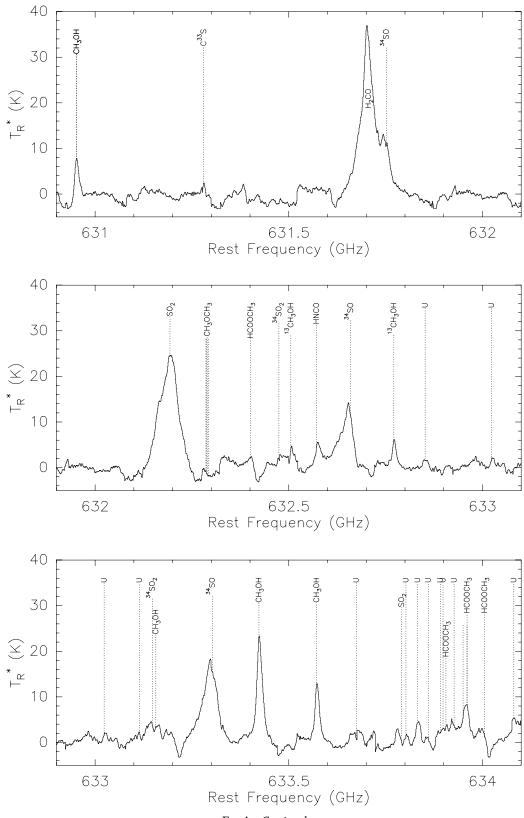


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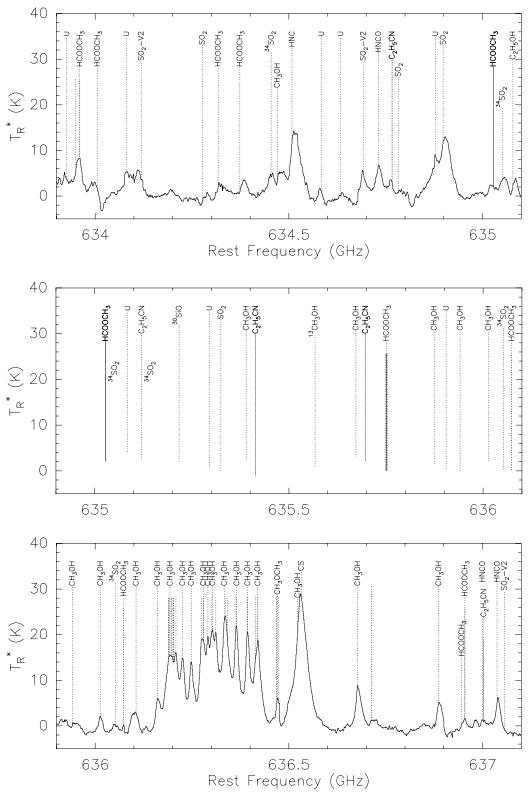


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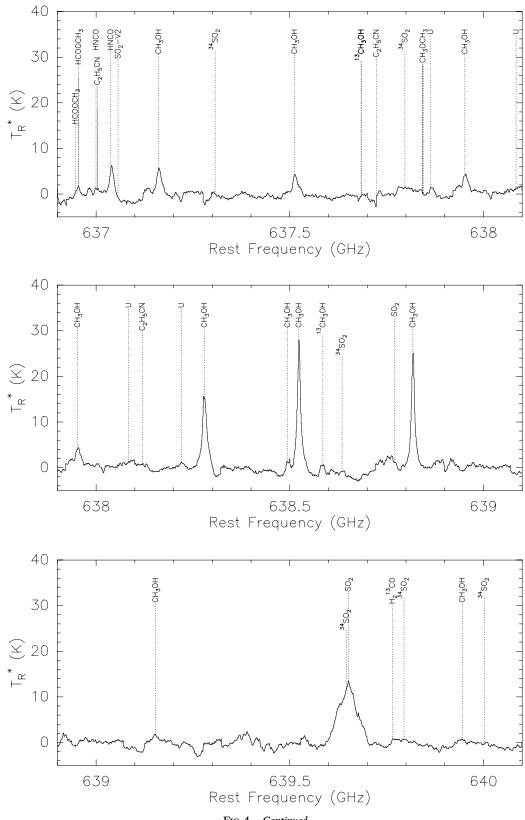


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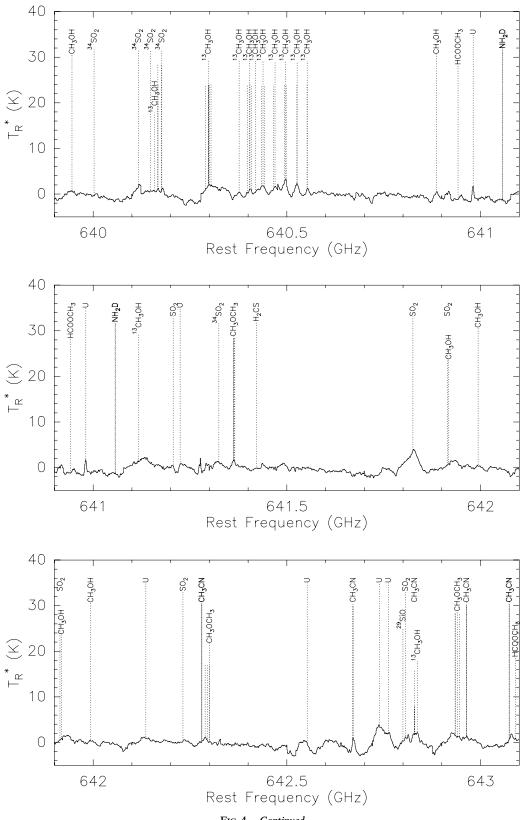


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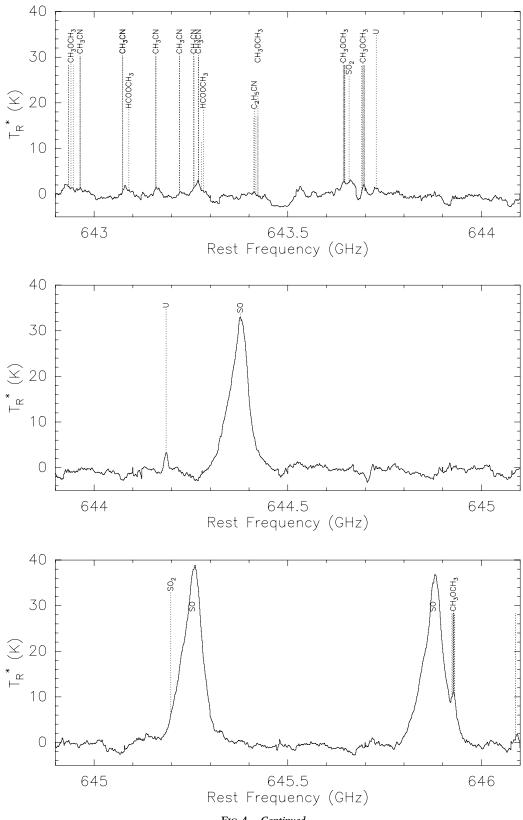


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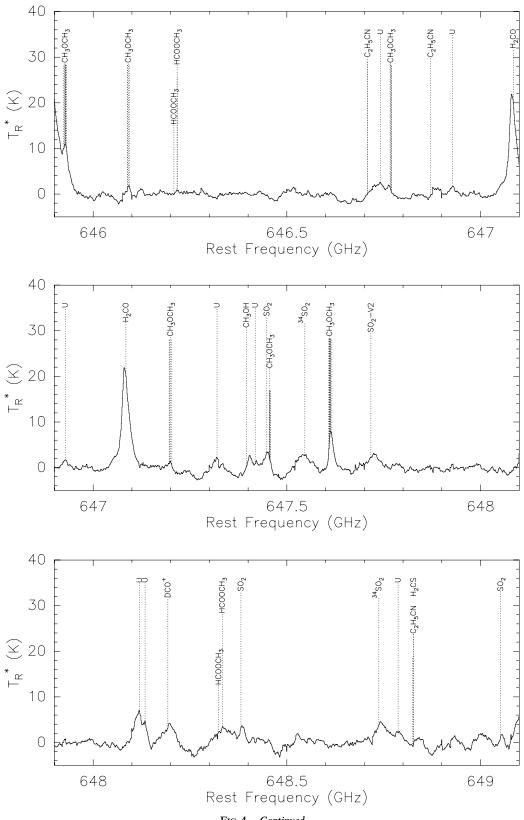


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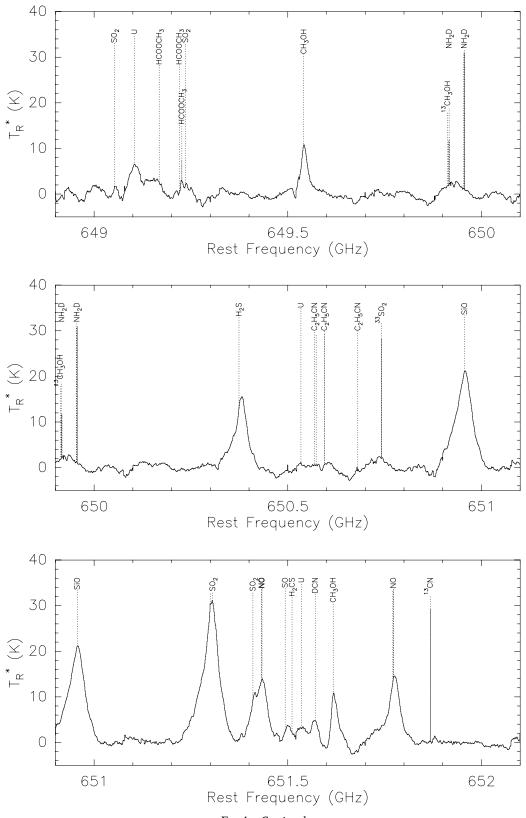


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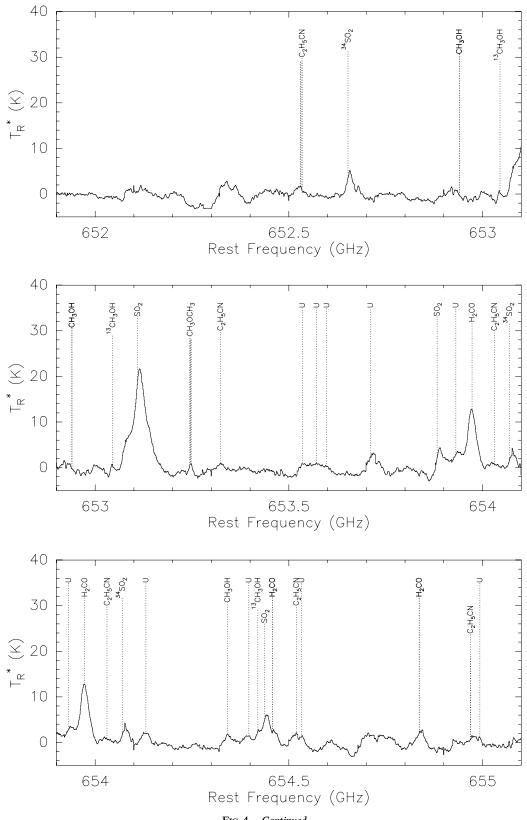


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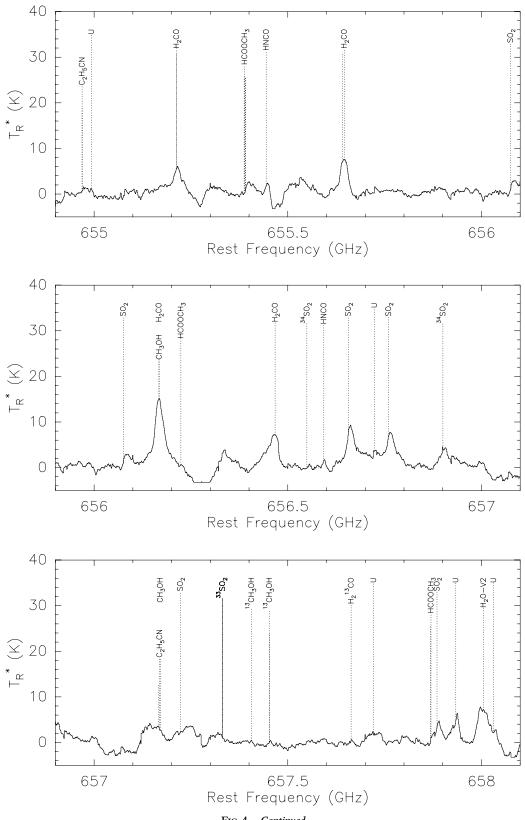


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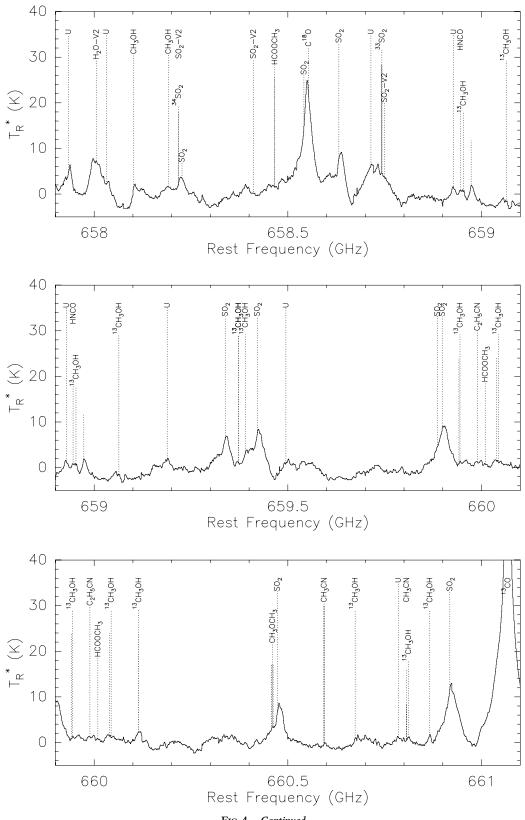


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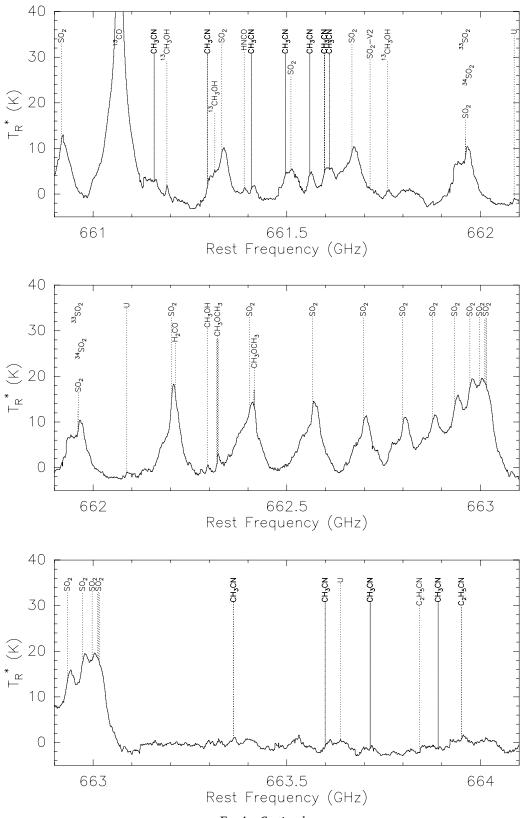


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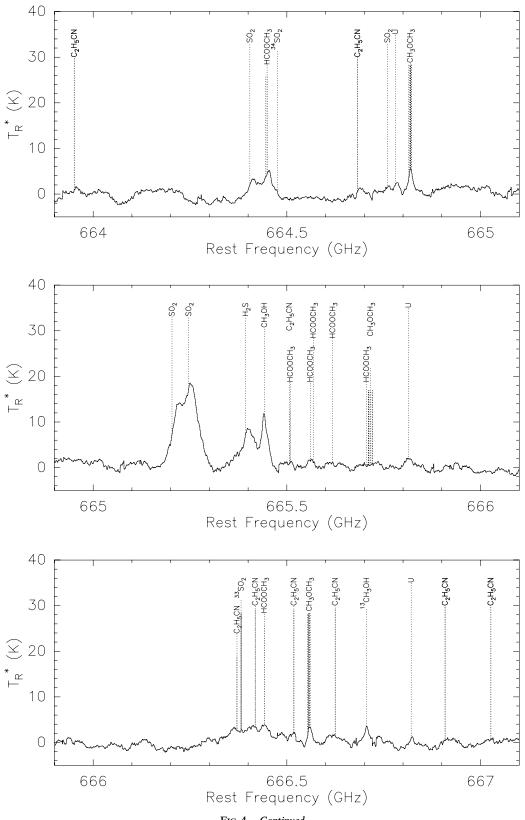


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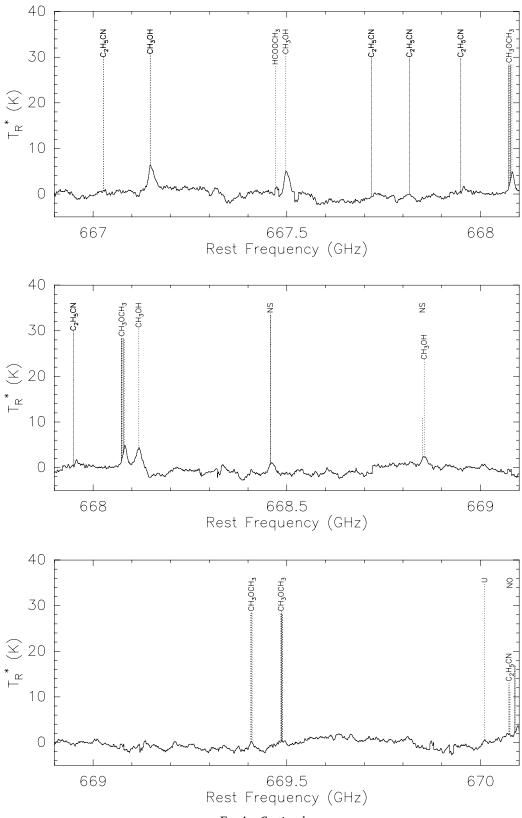


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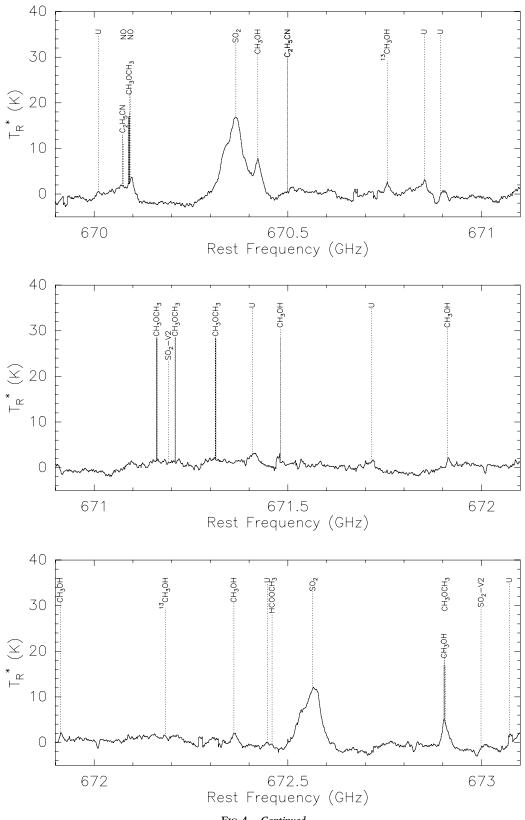


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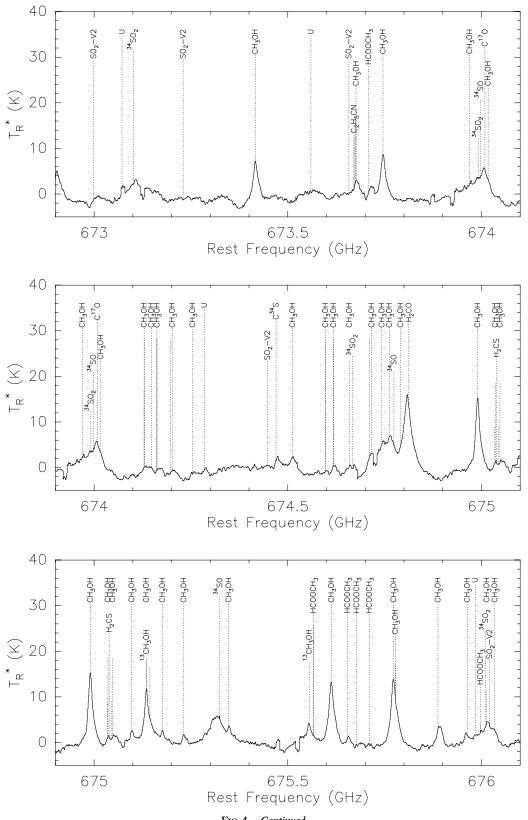


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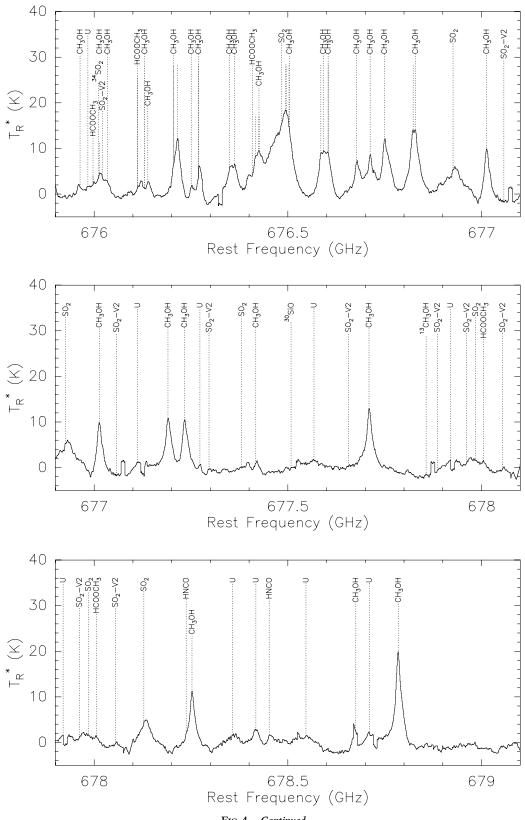


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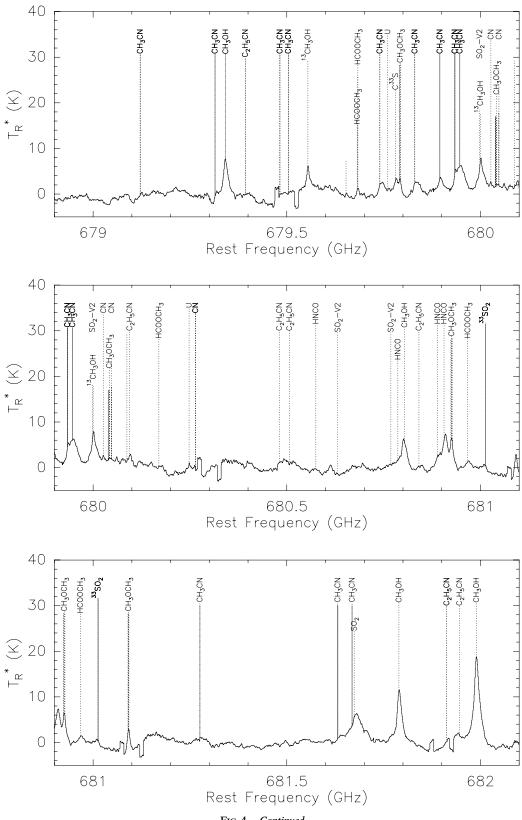


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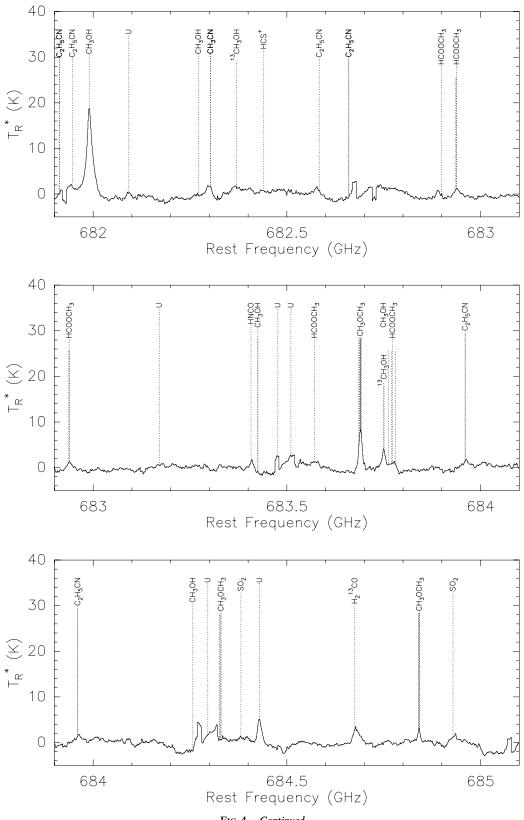


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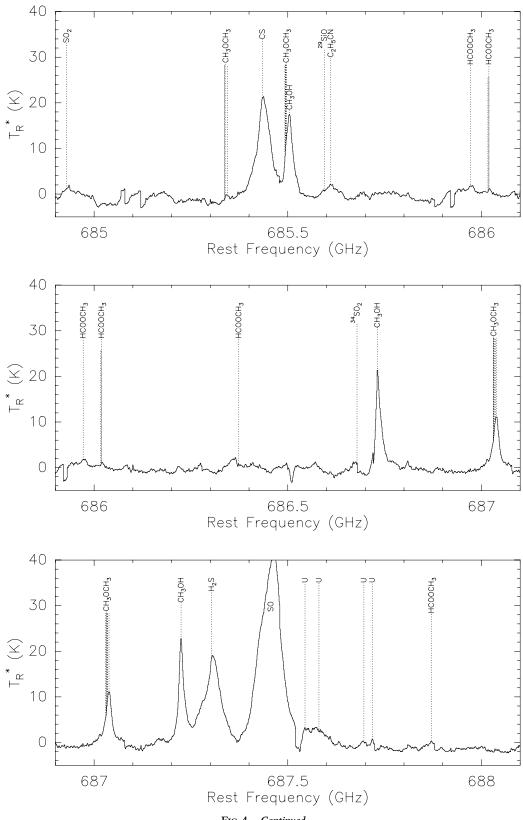


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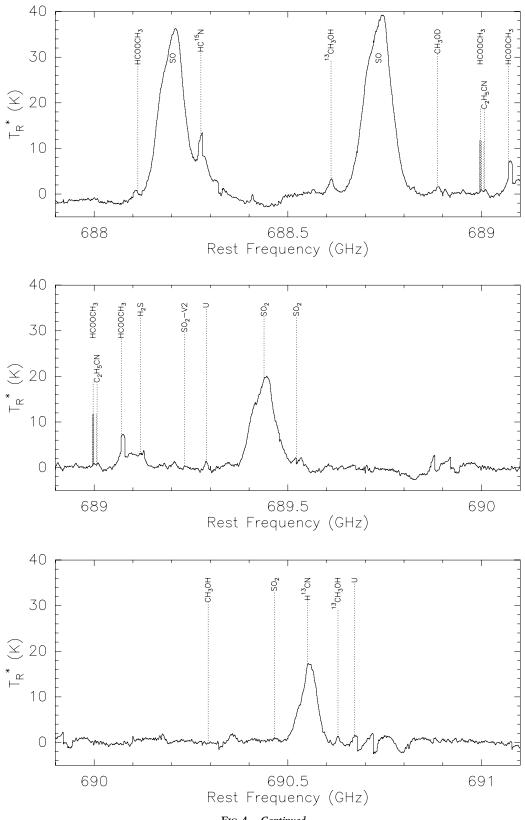


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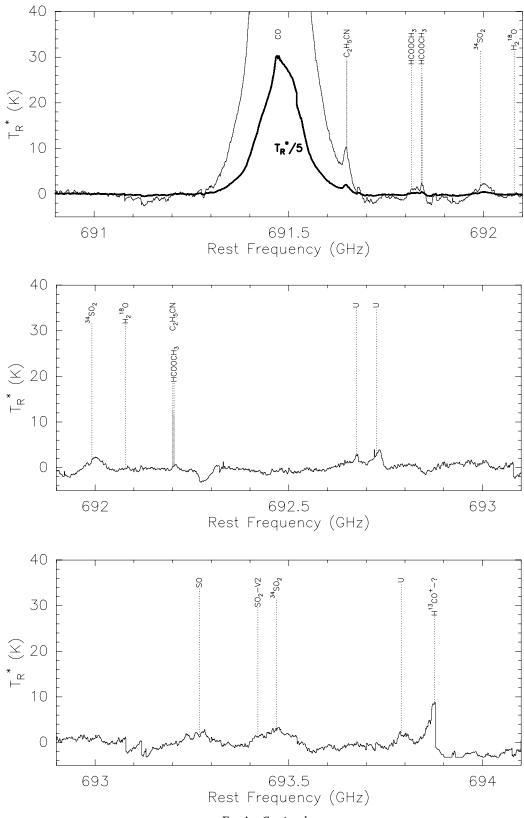


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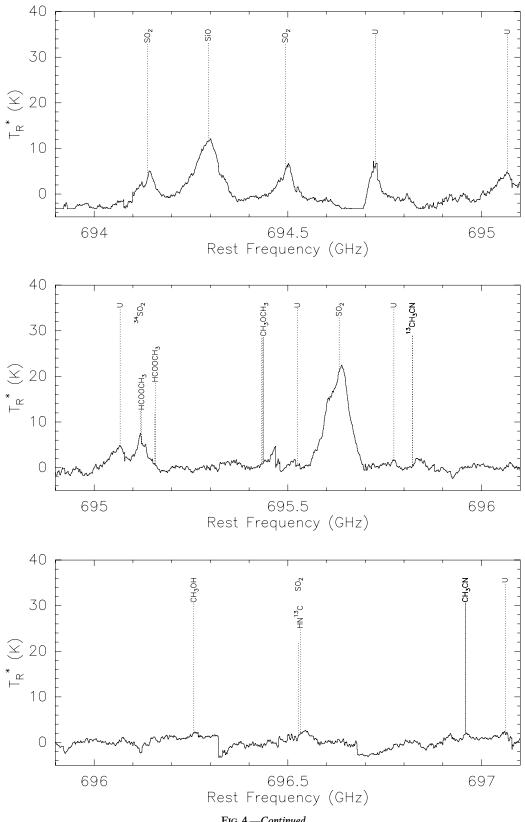


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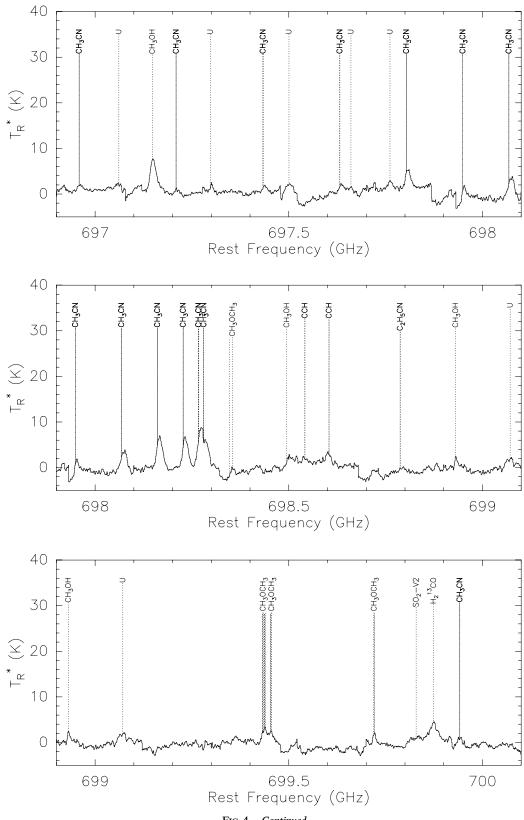


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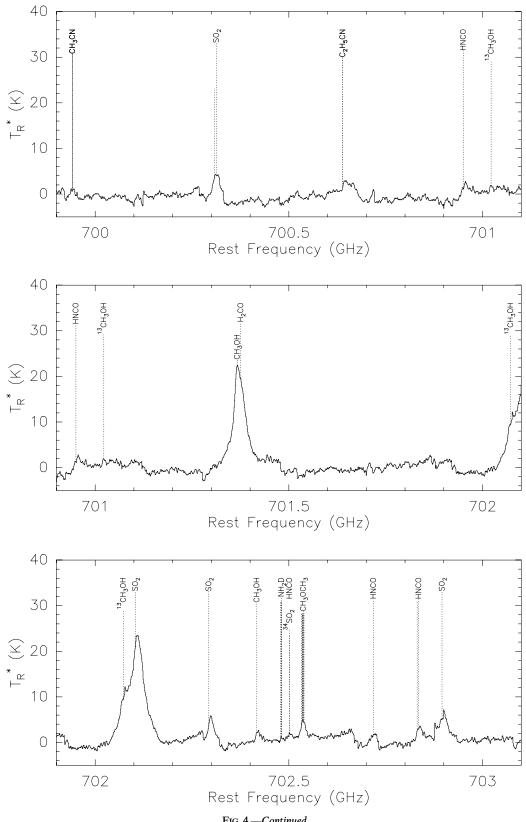


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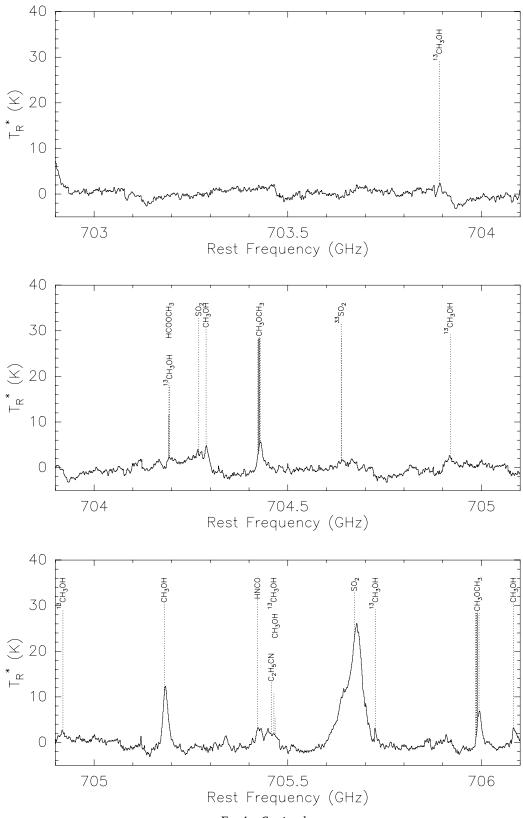


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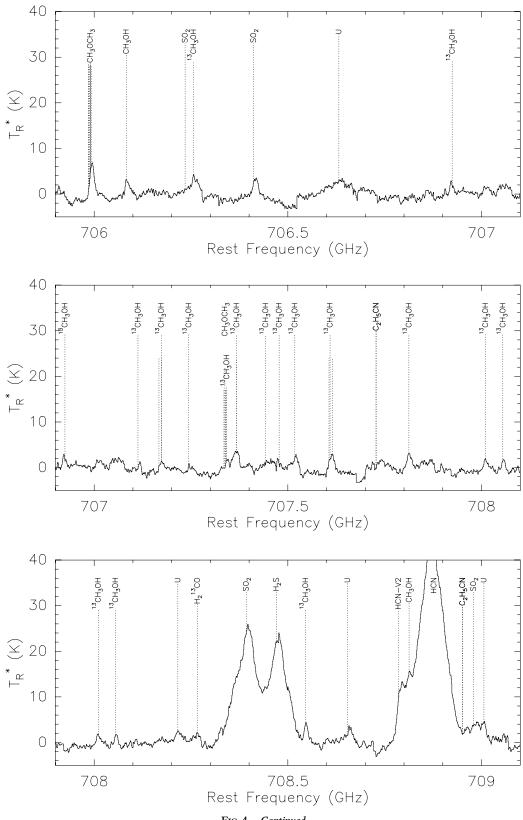


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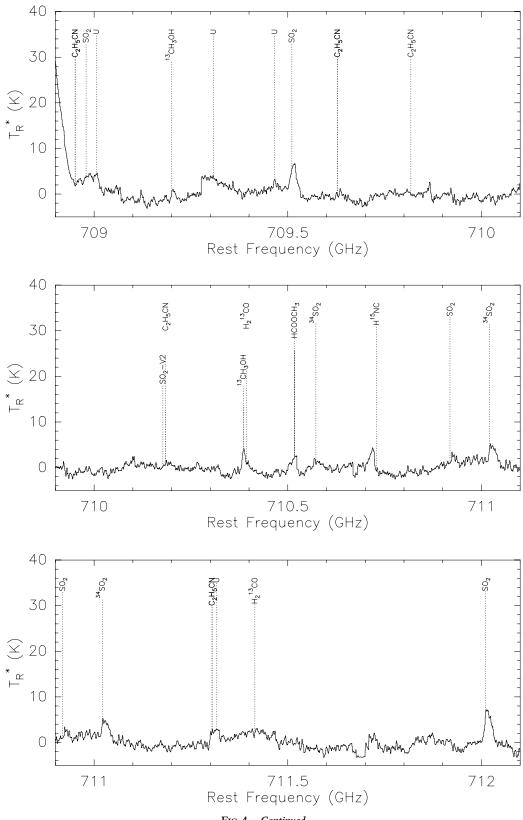


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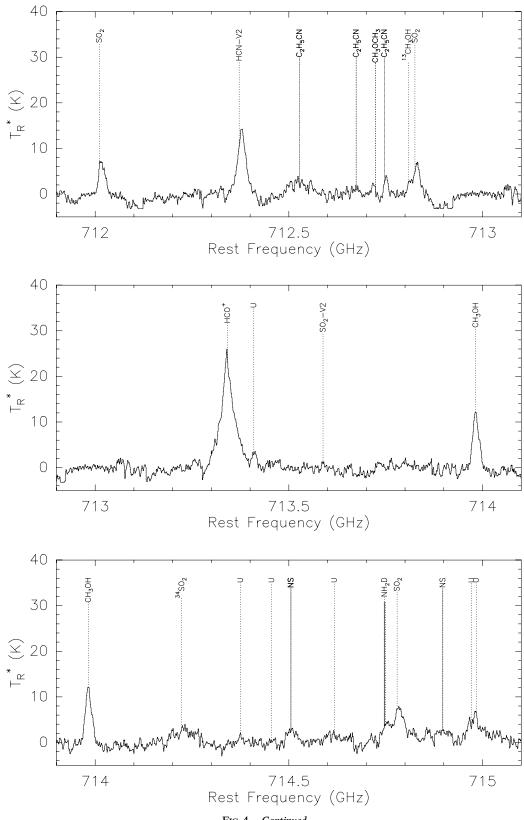


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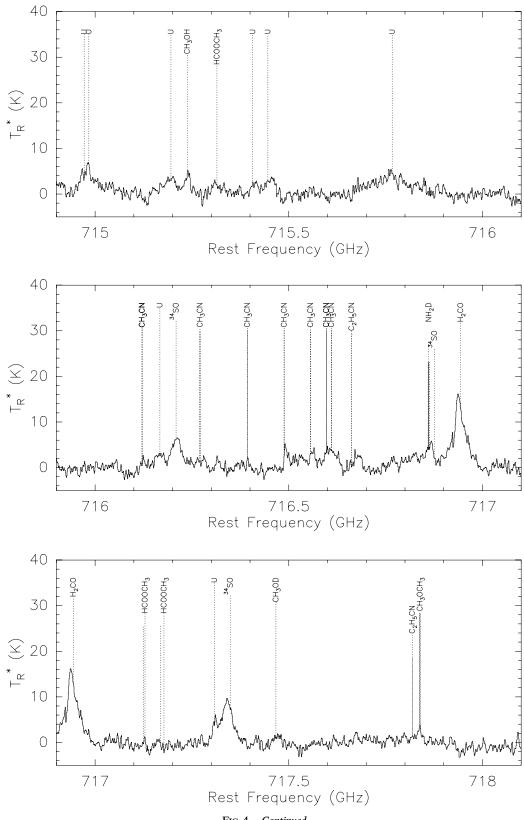


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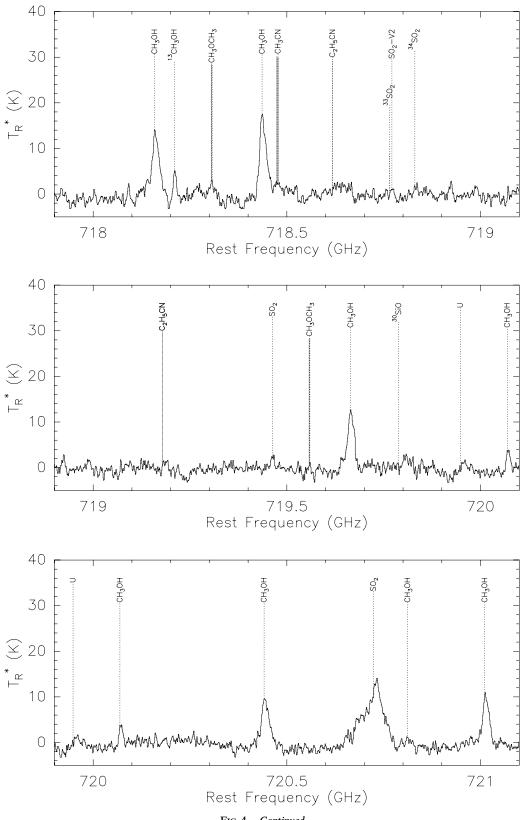


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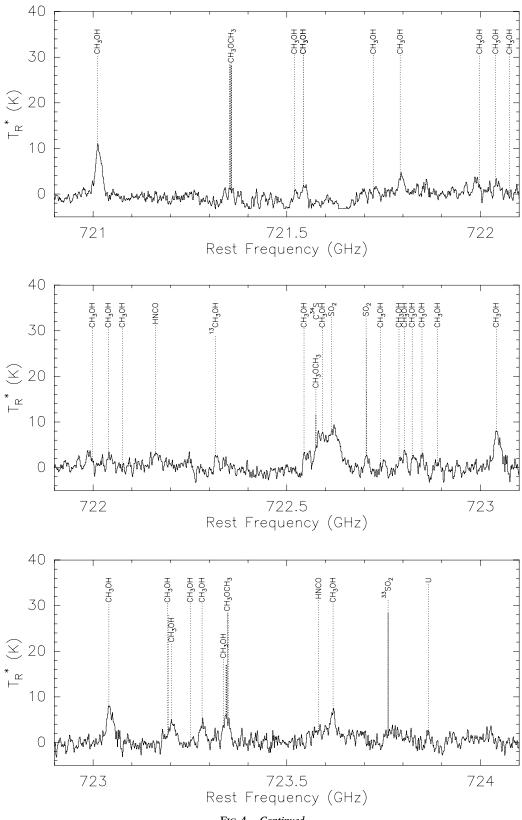


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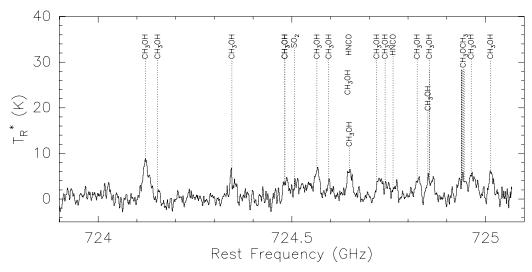


Fig. 4.—Continued

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