

Responsive Functional Materials

Ass.-Prof. Dr. Heidi A. Schwartz

Photoactive Hybrid Materials

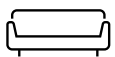
Universität Innsbruck



<https://www.uibk.ac.at/en/aatc/ag-schwartz/>



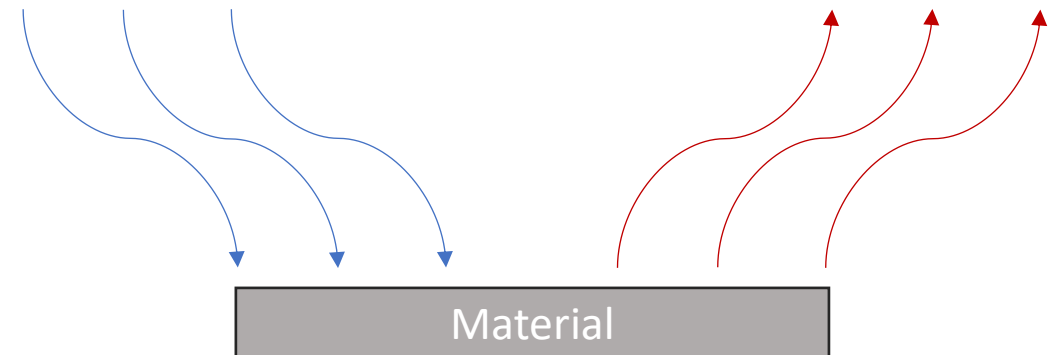
heidi.schwartz@uibk.ac.at



L01.063

External stimuli

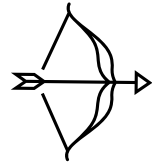
Functionality



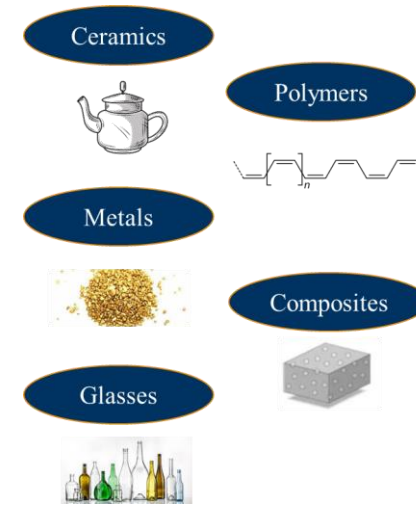
Outline for today's lecture

1. Short conclusion of last lecture

2. Learning objectives



- Analyzation tools and characterization methods I
- Next time: Analyzation tools and characterization methods II

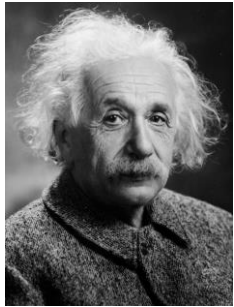


Photoelectric & photovoltaic effect

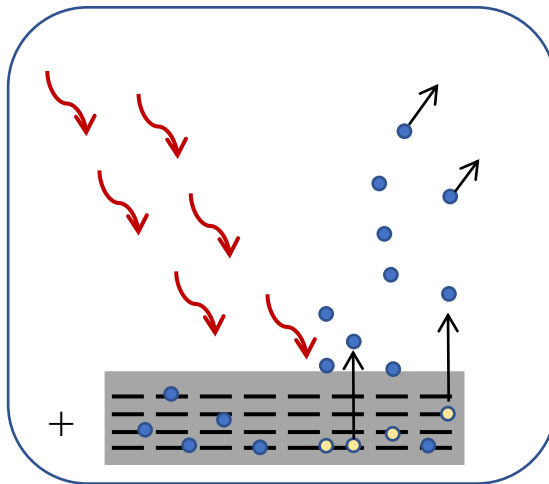
“excitation of electrons to a higher-energy state, when a material is exposed to electromagnetic radiation”



Note: An external stimulus is changing the material and therewith its properties!



A. Einstein



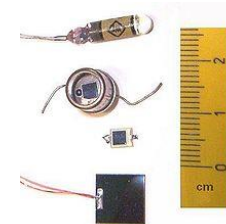
Electrons leave the material.



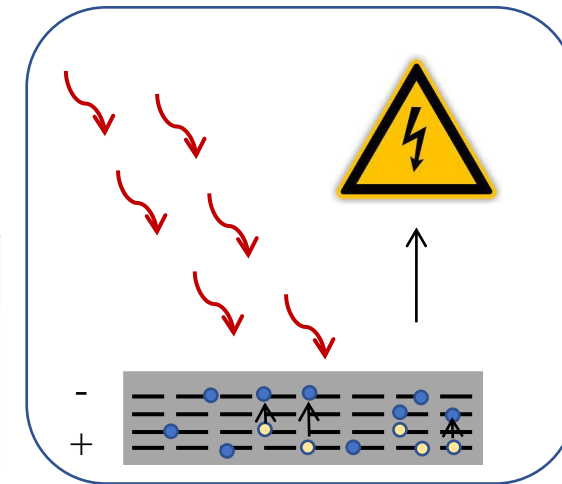
XPS.



Solar cell.



Photodiode.



Electrons stay within the material.

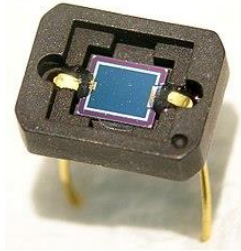


E. Becquerel

Photovoltaic effect is also called the inner photoelectric effect.

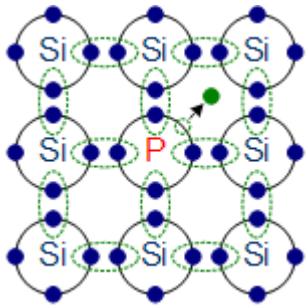
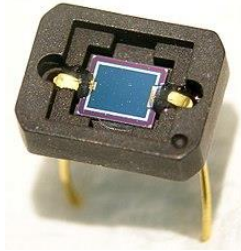
Photodiodes – what we need for detection or TV

Photodiodes are typically
built up from Si, Ge.



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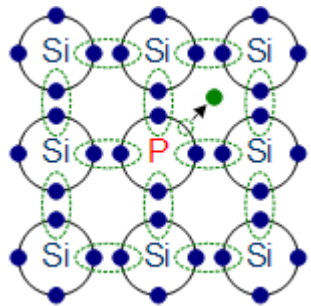
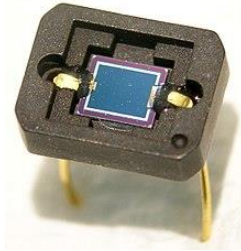


Doped semiconductors, whose conductivity originates from free **negative** electrons.

n-doping

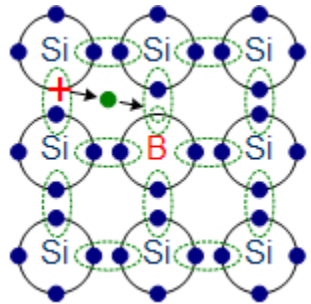
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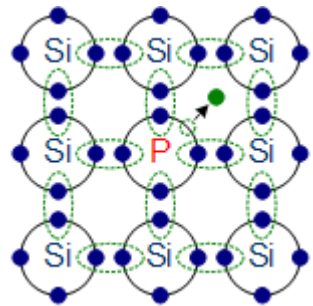
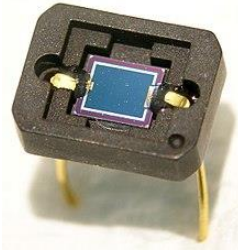


Doped semiconductors, whose conductivity originates from the presence of **p**ositive holes.

p-doping

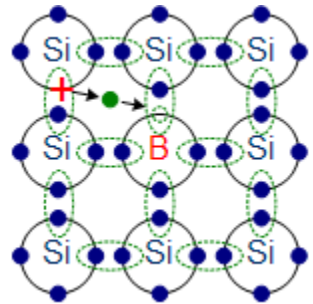
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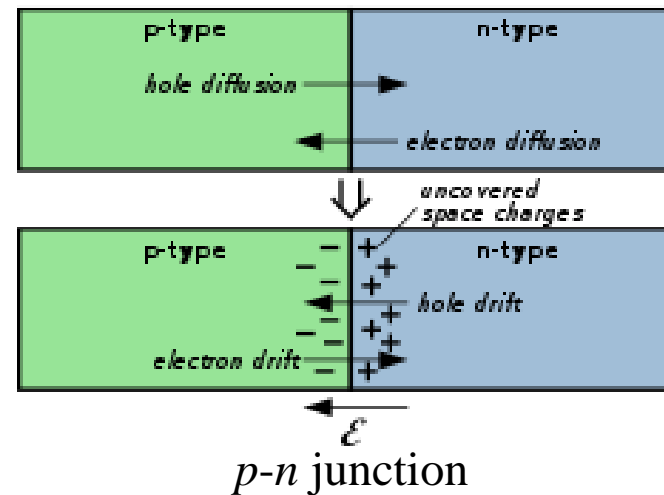
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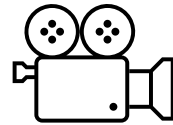
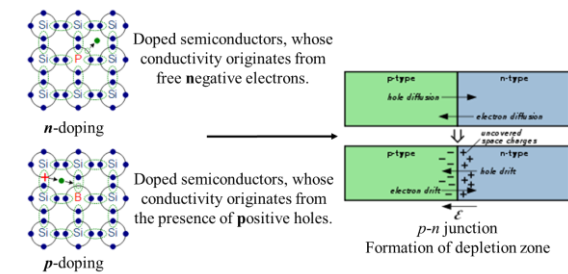
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Doped semiconductors, whose conductivity originates from the presence of **positive** holes.



Formation of depletion zone

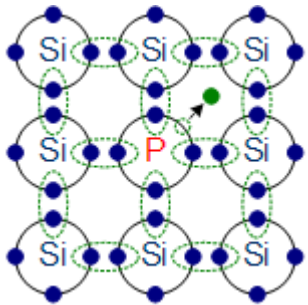
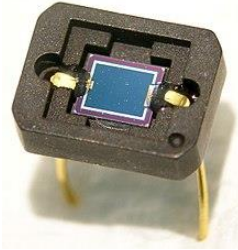
Photodiodes – what we need for detection or TV



[p-n-Übergang - Halbleiterdiode | LEIFIphysik](#)

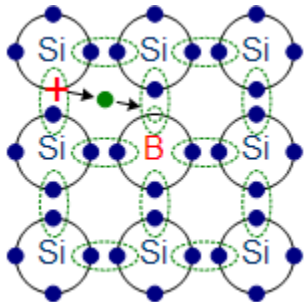
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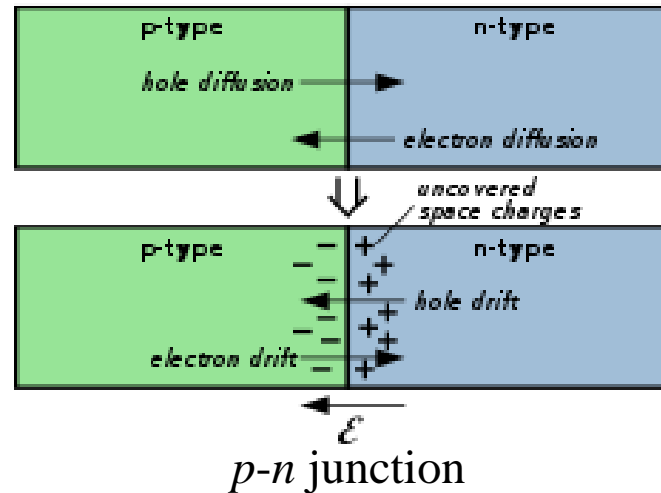


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Doped semiconductors, whose conductivity originates from the presence of **positive** holes.

Formation of local charges in *n*- and *p*-doped areas, which is called built-in voltage.

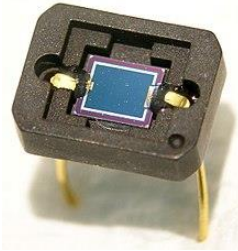
Diodes: Applying a voltage **in** forward direction creates an electric field: electric current!



Formation of depletion zone

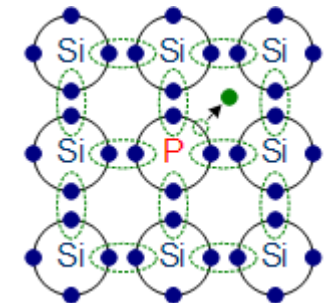
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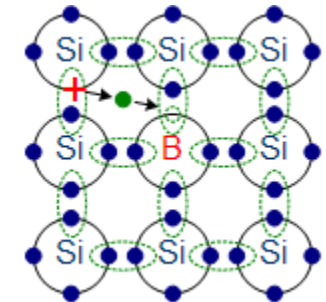
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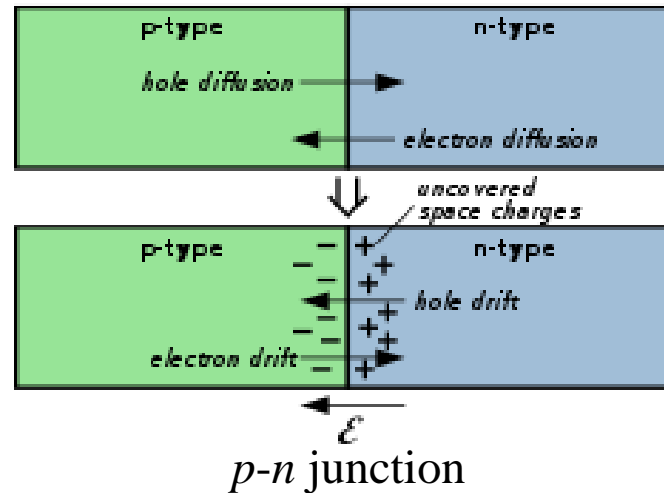
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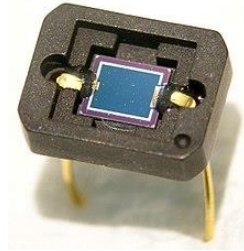
Formation of depletion zone

Photodiodes:

Voltage applied **against** forward direction!

Photons create electron-hole pairs, which are separated and measured as photo-current!

Photodiodes – what we need for detection or TV

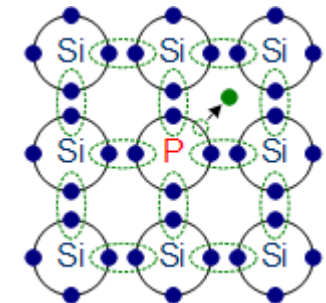


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Diodes: Applying a voltage **in** forward direction creates an electric field: electric current!

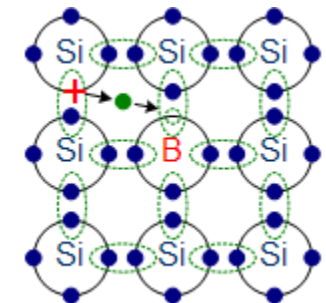
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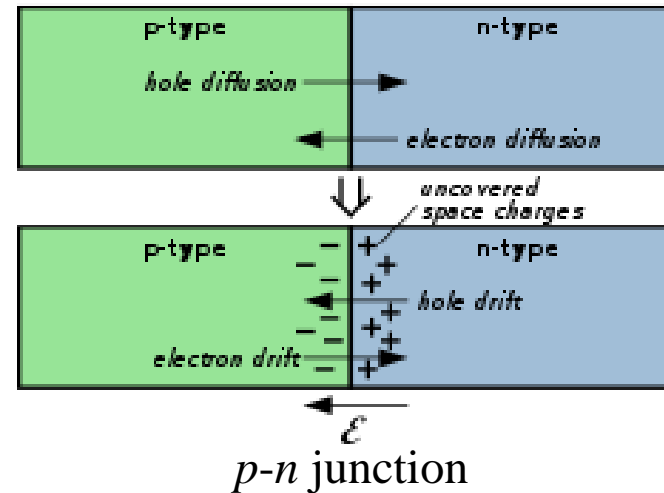
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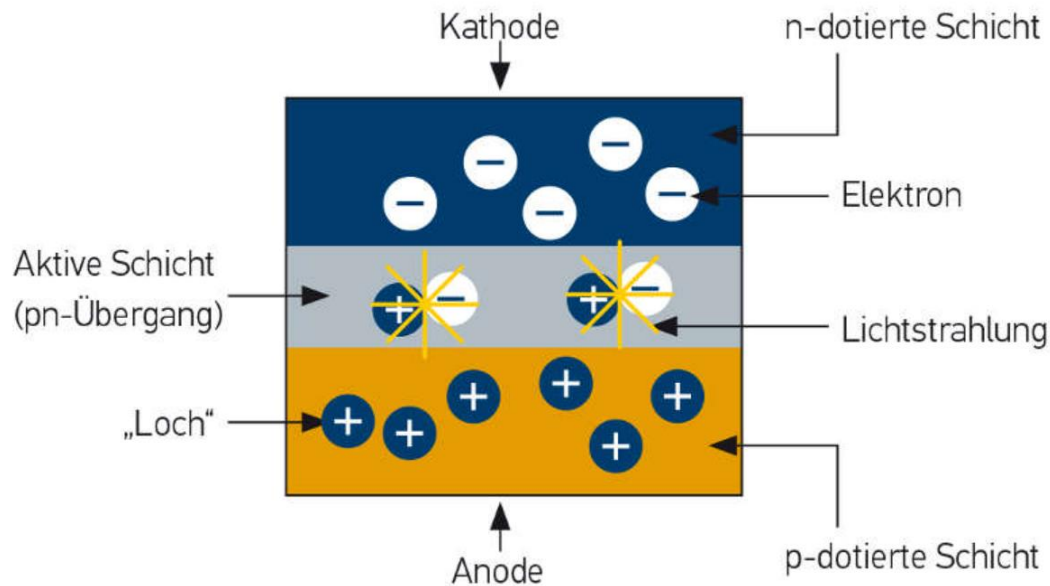
Working principle of a LED

Difference to “normal” non-emitting diode:

- built up from direct semi-conductor
- Recombination in diode occurs radiation-less



Indium and gallium.



Applying a voltage **in** forward direction:

- Charge carriers move towards each other; e^- from n to p - n junction
- Recombination in depletion zone under light emission

Solar cells – special photodiodes



Solar panels of the ISS.

Solar cells are built up like photodiodes, except that they are not operated as a radiation detector, but as power source.

White Board

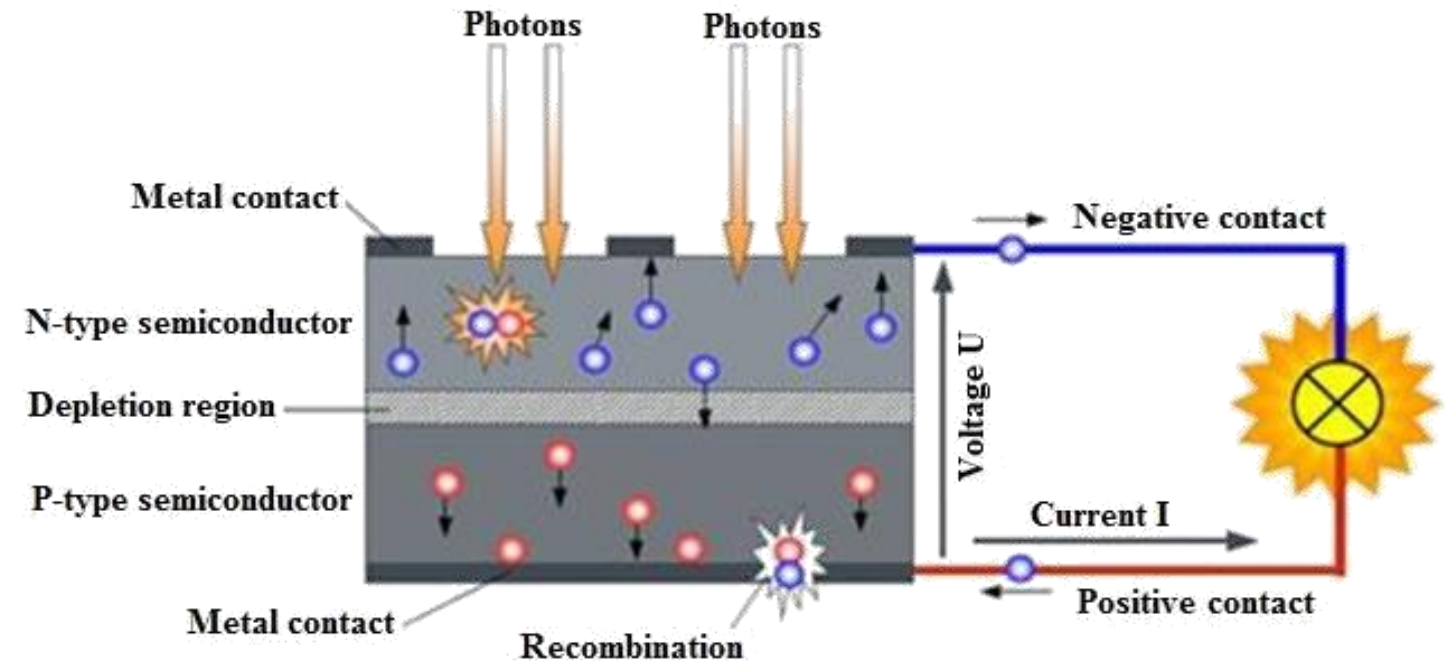
Kavaz, A & Hodžić, S & Hubana, Tarik & Curevac, Semra & Dozić, N & Merzic, Hamza & Tanković, H & Dervišević, K & Alihodžić, E & Sikira, E & Rahić, D & Kavazovic, Nihad & Tanković, Faris & Šestan, B. (2014). Solar Tree Project.

Solar cells – special photodiodes



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White Board

● Hole = positive charge
● Electron = negative charge

Solar cells – special photodiodes



Solar panels of the ISS.

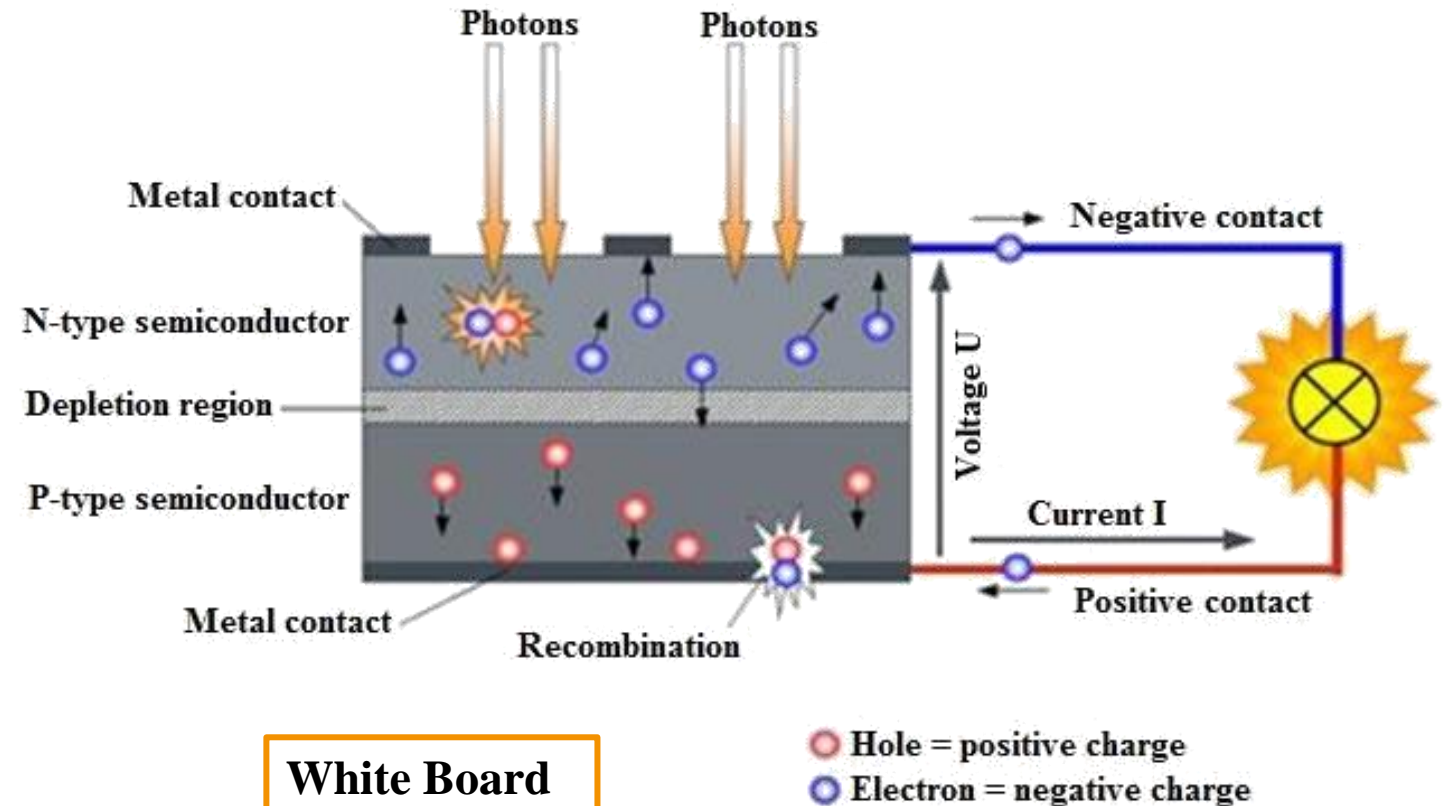
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Photons separate electron-hole pairs
in the depletion zone

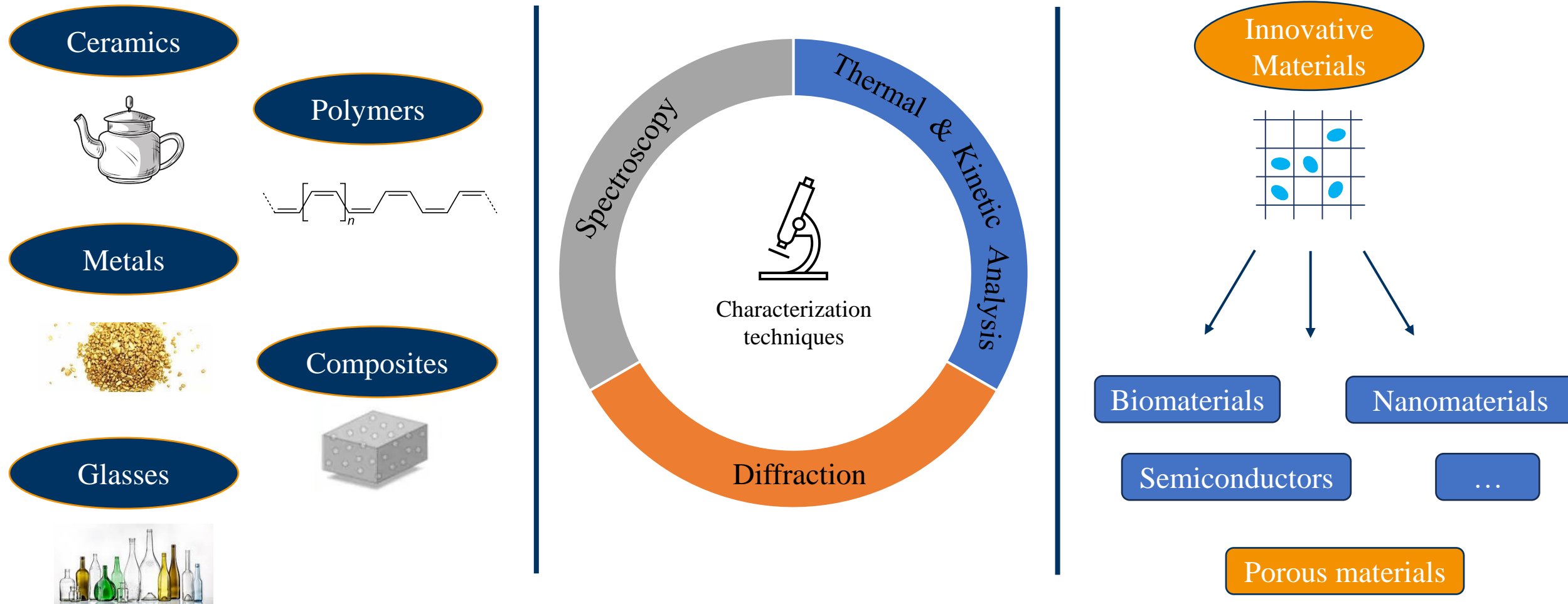
e^- travel to *n*-layer, holes to *p*-layer

Travelling electrons

Recombination in depletion zone



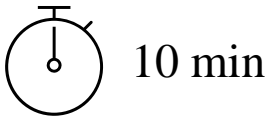
A functional material could be defined as being prepared from a “target-motivated” approach



Group-/Tandem-Work

What can we learn from different analyzation techniques?

Shortly sum up, which methods you know and what kind of information you can extract!



Characterization methods – there are so many...

There are various methods to determine the average and the local structure: diffraction and spectroscopic methods.

Diffraction / no change in electronic structure



- X-ray diffraction
- Neutron diffraction
- Pair Distribution Function (PDF)

Characterization methods – there are so many...

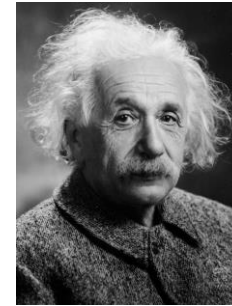
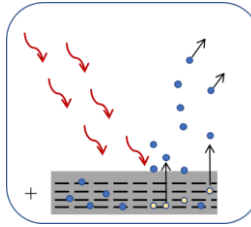
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Photoelectric effect



- XAS, XPS (AES) & UPS

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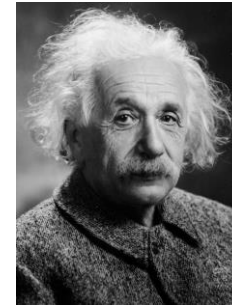
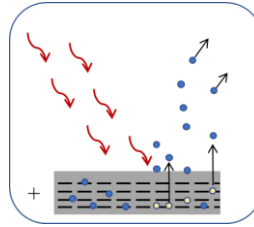
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Vibrations and Rotations

- IR and Raman

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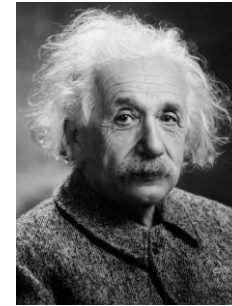
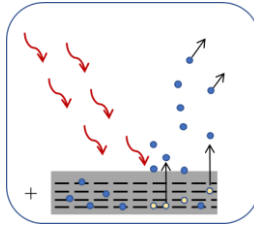
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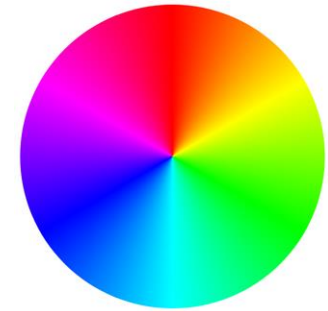


- XAS, XPS (AES) & UPS

Vibrations and Rotations

- IR and Raman

Electronic, nuclear transitions



- UV/Vis spectroscopy
- NMR spectroscopy

Characterization methods – there are so many...

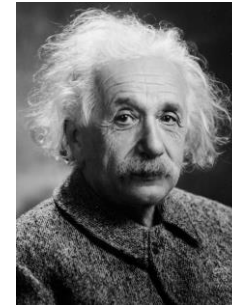
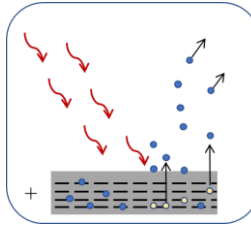
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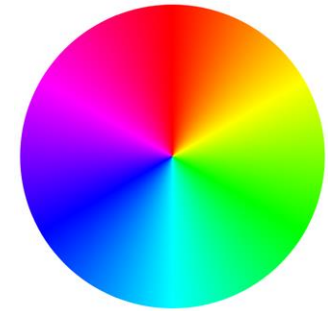


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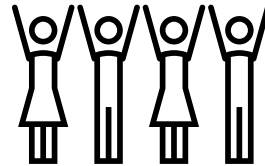
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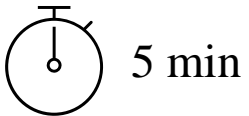
- UV/Vis spectroscopy
- NMR spectroscopy

Further methods: EDX, mass spectrometry, DSC-TGA, BET.



Group-/Tandem-Work

For this week, set up a table with method, sample requirement, information gained and limitations.



Fundamentals of diffractive methods

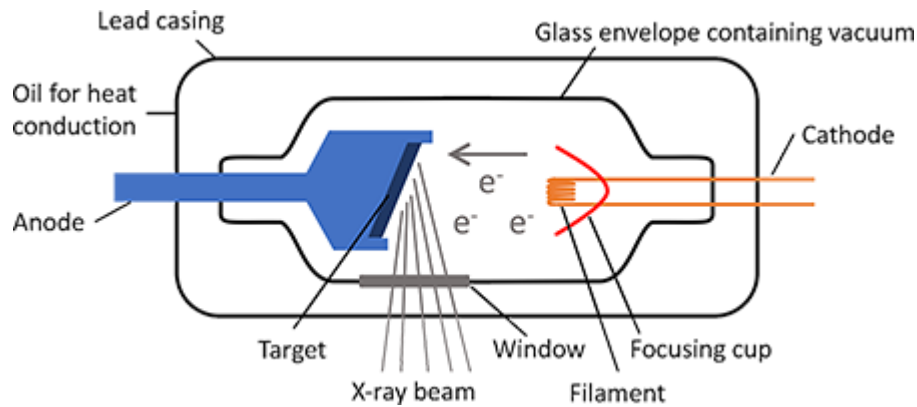
X-ray diffraction – scattering at the electrons

X-rays were discovered by Wilhelm Conrad Röntgen in 1895, for which he was awarded with the Nobel Prize in Physics in 1901.



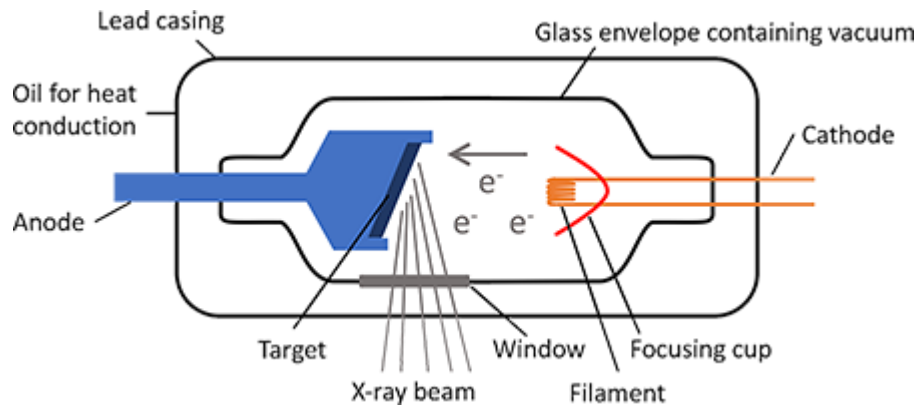
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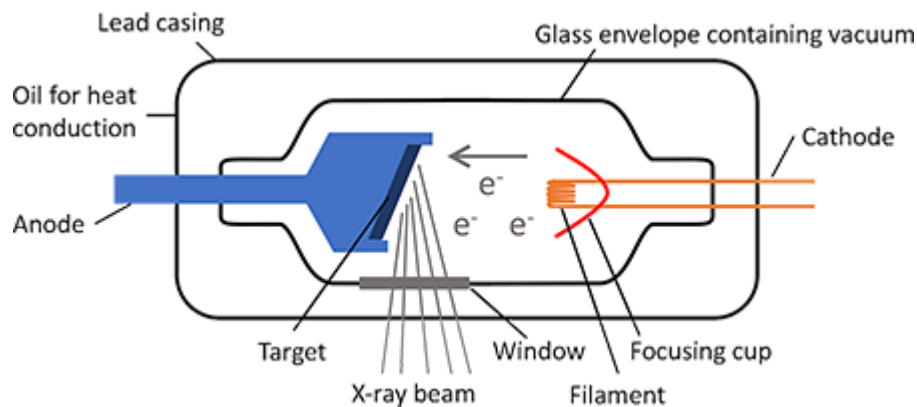
Electrons are accelerated onto anode material in a high vacuum

$\text{Cu } K_{\alpha 1}: 1.54 \text{ \AA}$

$\text{Mo } K_{\alpha 1}: 0.70 \text{ \AA}$

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X-ray diffraction provides **electron density maps**

http://ruby.chemie.uni-freiburg.de/Vorlesung/methoden_II_3.xhtml (11th June 2018)

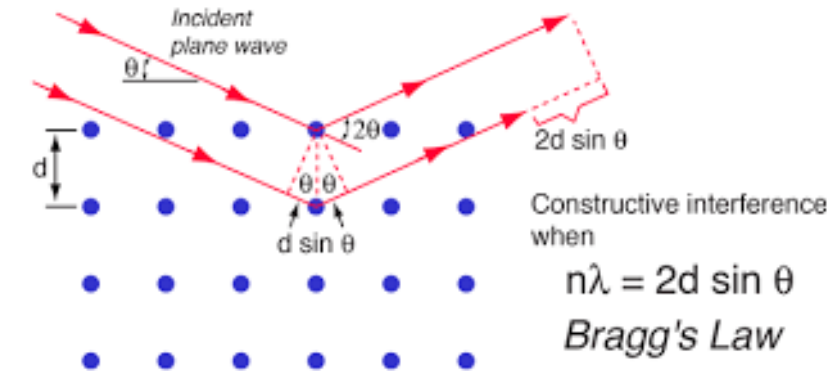
W. Massa, Kristallstrukturbestimmung, Teubner GmbH, 2002.

[Production of X-rays - Radiology Café](#) (27th March 2023)

X-ray diffraction – scattering at the electrons

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Sample requirement: long-range order.



Scattering at net planes –
electron shell

X-ray diffraction provides **electron density maps**

X-ray diffraction – scattering at the electrons

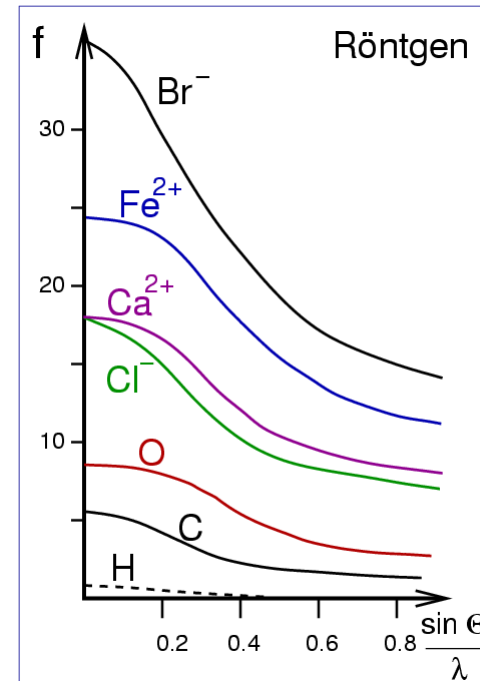
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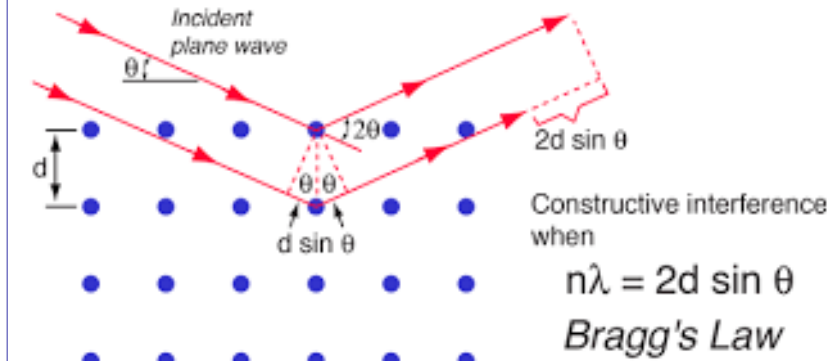
Challenges in X-ray diffraction

- Determination of proton positions difficult
- Differentiation of atoms with the same or similar number of electrons hardly possible

X-ray diffraction provides **electron density maps**



Intensity is dependent of atomic form factor



Scattering at net planes –
electron shell

Neutron diffraction – scattering at the nucleus

Neutron radiation is generated in two ways: Either a nuclear reaction (^{235}U) or spallation sources are used.

Sample requirement: long-range order.



ILL Grenoble

- Scattering of neutrons by the point-like nucleus
- Absorber: Cadmium, Boron, Hafnium, Europium



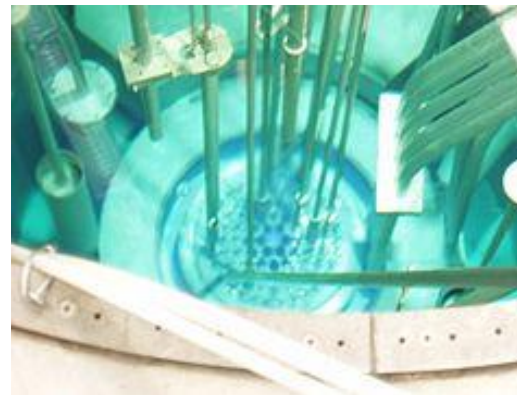
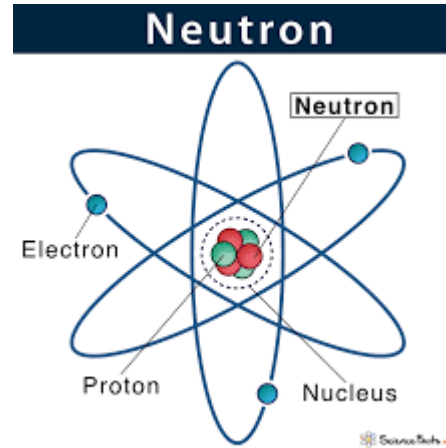
ORNL Oak Ridge

Neutron diffraction – scattering at the nucleus

Core reactions

- Neutrons are continuously produced from the fission of uranium: very fast (high-energy) neutrons
- Can be adjusted by moderators:
 - I. thermic neutrons, 1.5 \AA , heavy water
 - II. hot neutrons: 0.5 \AA , graphite at $2500 \text{ }^{\circ}\text{C}$
 - III. cold neutrons, 5 \AA , liquid H_2

Neutrons are guided through waveguides (Ni mirrored glass tubes) / monochromators also applied

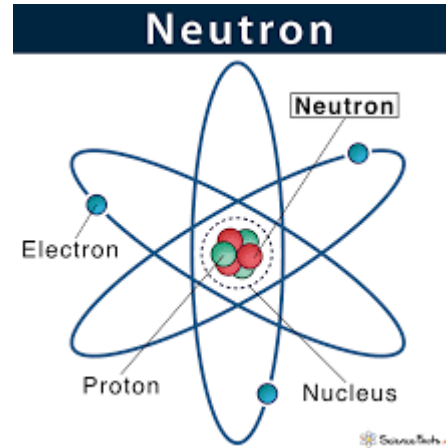


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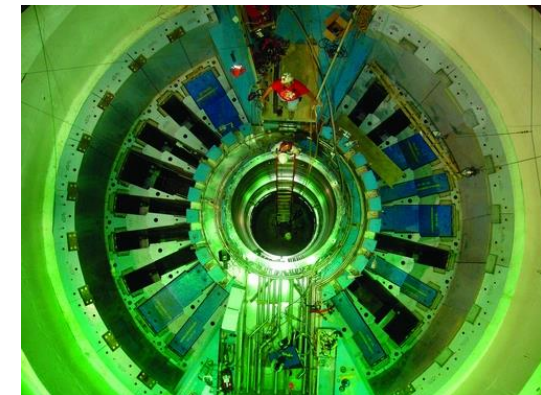
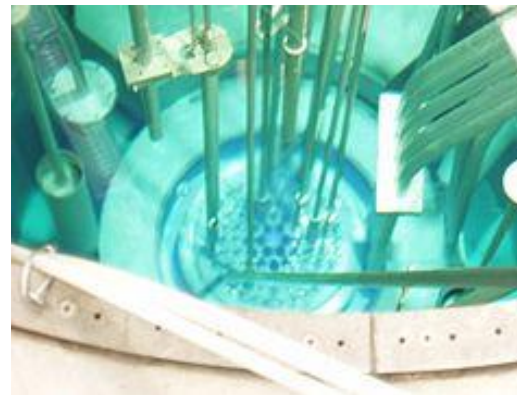
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Spallation sources

- I. Pulsed proton beam (50 Hz) from synchrotron hits heavy metal target (U, Pb, Hg, Ta) and ejects around 50 neutrons per proton
- II. Velocity is known and related to E and λ : no monochromator required!



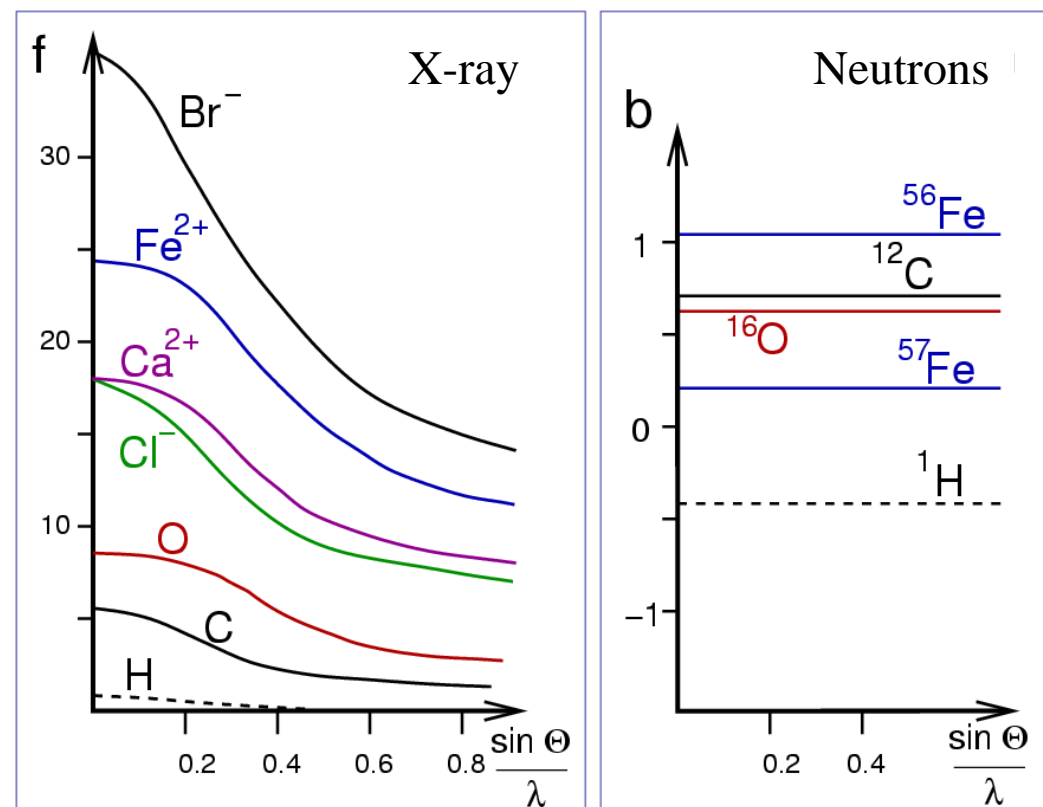
Neutron diffraction – scattering at the nucleus

Neutron scattering lengths hardly decrease with increasing diffraction angle.

Sample requirement: long-range order.

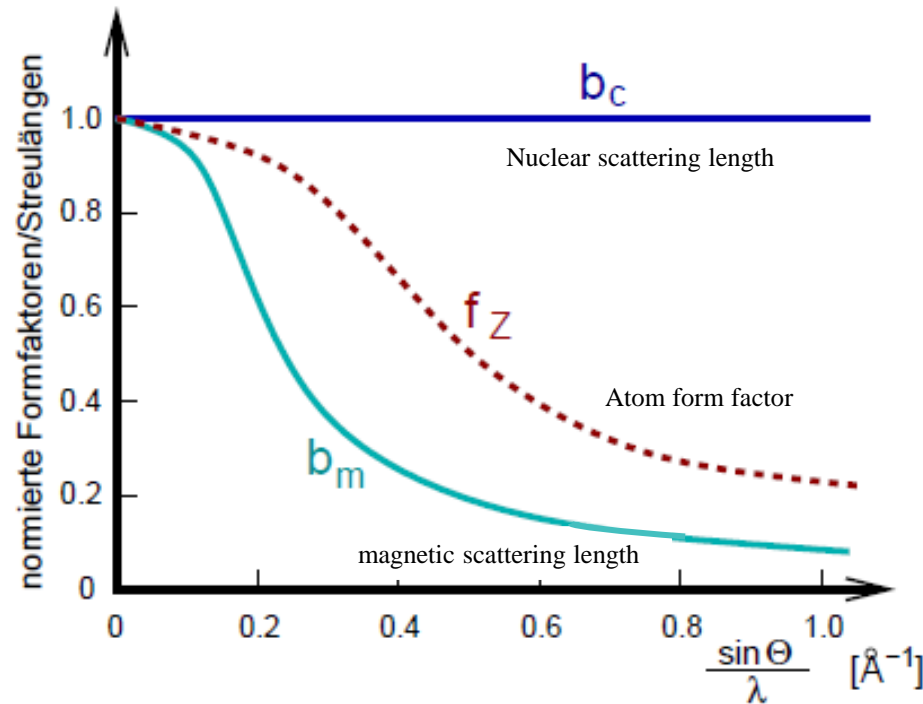
- light and heavy elements have comparable scattering force
- different b -lengths of elements with similar atomic number
- Slow decrease of $b \rightarrow$ measurements at high resolution

Neutron diffraction provides **core density maps**



Neutron diffraction – it's much more!

Not only core density maps are yielded, but also information about the magnetic structure is obtained.



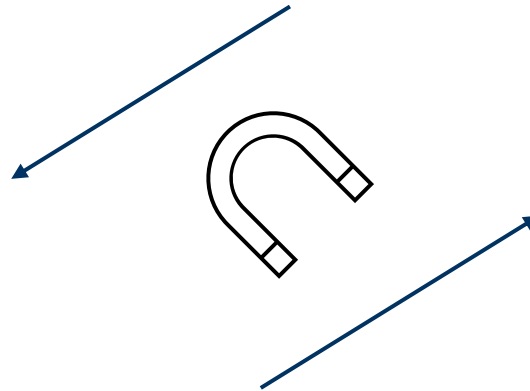
- Interaction with magnetic field (spin and orbital moments of electrons of the atomic shell)
- Electron shell is not point like:
 Θ dependence as known from X-ray diffraction

Magnetic scattering: Neutron has magnetic moment of $\mu = 1/2$

Neutron diffraction – it's much more!

Differentiation between ferro- and antiferromagnetism due to additional peaks!

Ferromagnetic material

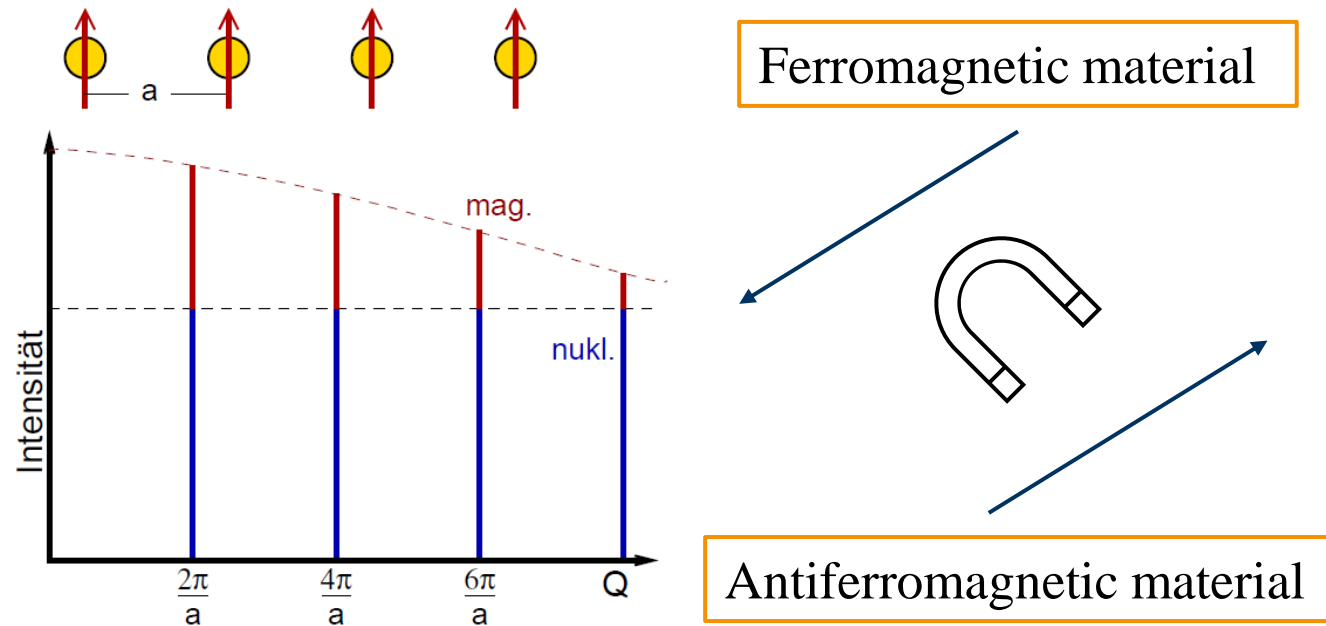


Antiferromagnetic material

Magnetic scattering: Neutron has magnetic moment of $\mu = \frac{1}{2}$

Neutron diffraction – it's much more!

Differentiation between ferro- and antiferromagnetism due to additional peaks!

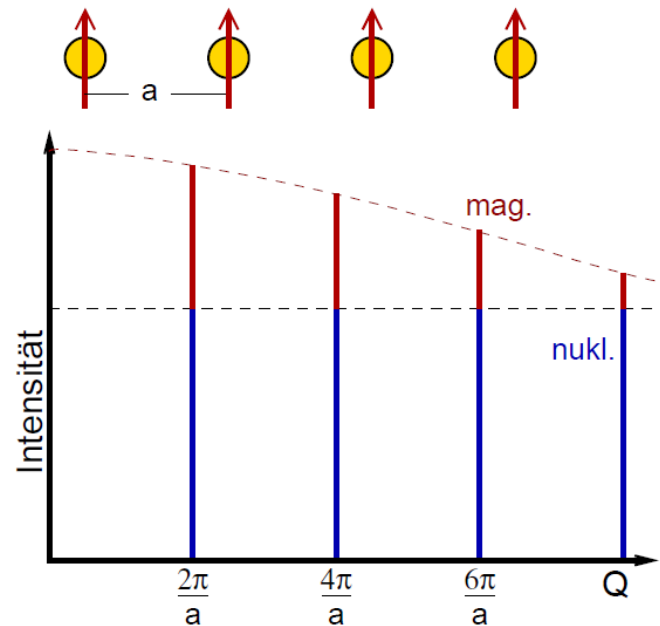


- magnetic Bragg peaks at the same positions as nuclear peaks
- θ dependence of magnetic structure

Magnetic scattering: Neutron has magnetic moment of $\mu = \frac{1}{2}$

Neutron diffraction – it's much more!

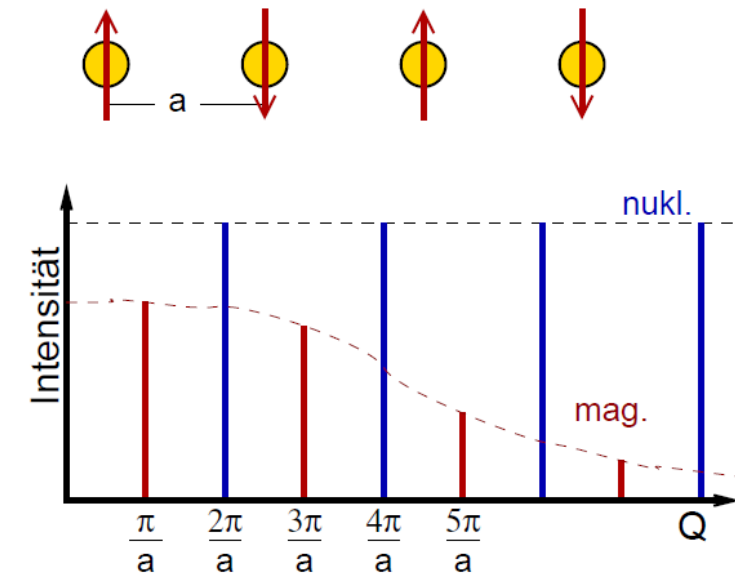
Differentiation between ferro- and antiferromagnetism due to additional peaks!



Ferromagnetic material



Antiferromagnetic material



- magnetic Bragg peaks at the same positions as nuclear peaks
- θ dependence of magnetic structure

- magnetic Bragg peaks between nuclear peaks
- formation of magnetic superstructure

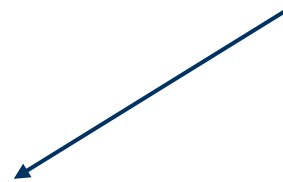
Magnetic scattering: Neutron has magnetic moment of $\mu = 1/2$

Neutron diffraction vs. X-ray diffraction

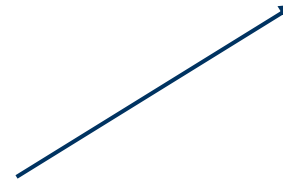
Highly penetrating due to 0 charge.
Probing internal structure of sample.

X-rays are carrying no charge, provide
data on the surface structure mainly.

Neutron diffraction



X-ray diffraction



Neutron diffraction vs. X-ray diffraction

Highly penetrating due to 0 charge.
Probing internal structure of sample.

Interacts with nuclei: scatters strongly
from both light and heavy atoms and
can differentiate between isotops.

Neutron diffraction

X-rays are carrying no charge, provide
data on the surface structure mainly.

Interacts with electron clouds: scatters
strongly from heavier elements with
bigger electron clouds.

X-ray diffraction

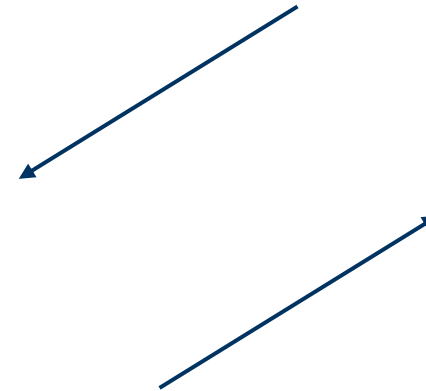
Neutron diffraction vs. X-ray diffraction

Highly penetrating due to 0 charge.
Probing internal structure of sample.

Interacts with nuclei: scatters strongly
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can differentiate between isotops.

Neutrons are spin $\frac{1}{2}$ and carry magnetic
moment: study of magnetism over short
ranges.

Neutron diffraction



X-ray diffraction

X-rays are carrying no charge, provide
data on the surface structure mainly.

Interacts with electron clouds: scatters
strongly from heavier elements with
bigger electron clouds.

X-rays can be utilized to investigate
magnetism but are less accurate.

Neutron diffraction vs. X-ray diffraction

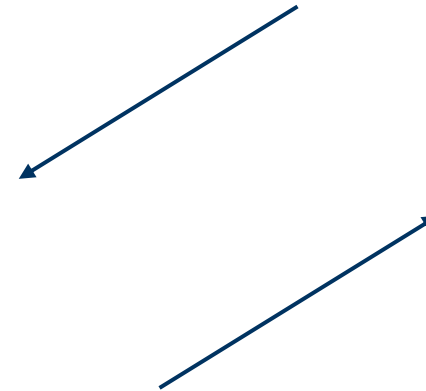
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Expensive and lot of sample amount is
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Neutron diffraction



X-ray diffraction

X-rays are carrying no charge, provide
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Interacts with electron clouds: scatters
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bigger electron clouds.

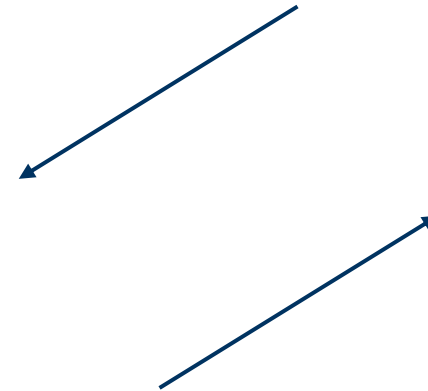
X-rays can be utilized to investigate
magnetism but are less accurate.

Less expensive and small sample
amounts are possible.

Neutron diffraction vs. X-ray diffraction

Highly penetrating due to 0 charge. Probing internal structure of sample.
Interacts with nuclei: scatters strongly from both light and heavy atoms and can differentiate between isotops.
Neutrons are spin $\frac{1}{2}$ and carry magnetic moment: study of magnetism over short ranges.
Expensive and lot of sample amount is needed.
Requires the use of high energy synchrotrons.

Neutron diffraction



X-ray diffraction

X-rays are carrying no charge, provide data on the surface structure mainly.
Interacts with electron clouds: scatters strongly from heavier elements with bigger electron clouds.
X-rays can be utilized to investigate magnetism but are less accurate.
Less expensive and small sample amounts are possible.
Can also be performed in house – no need for synchrotrons.

What diffraction methods cannot serve for...



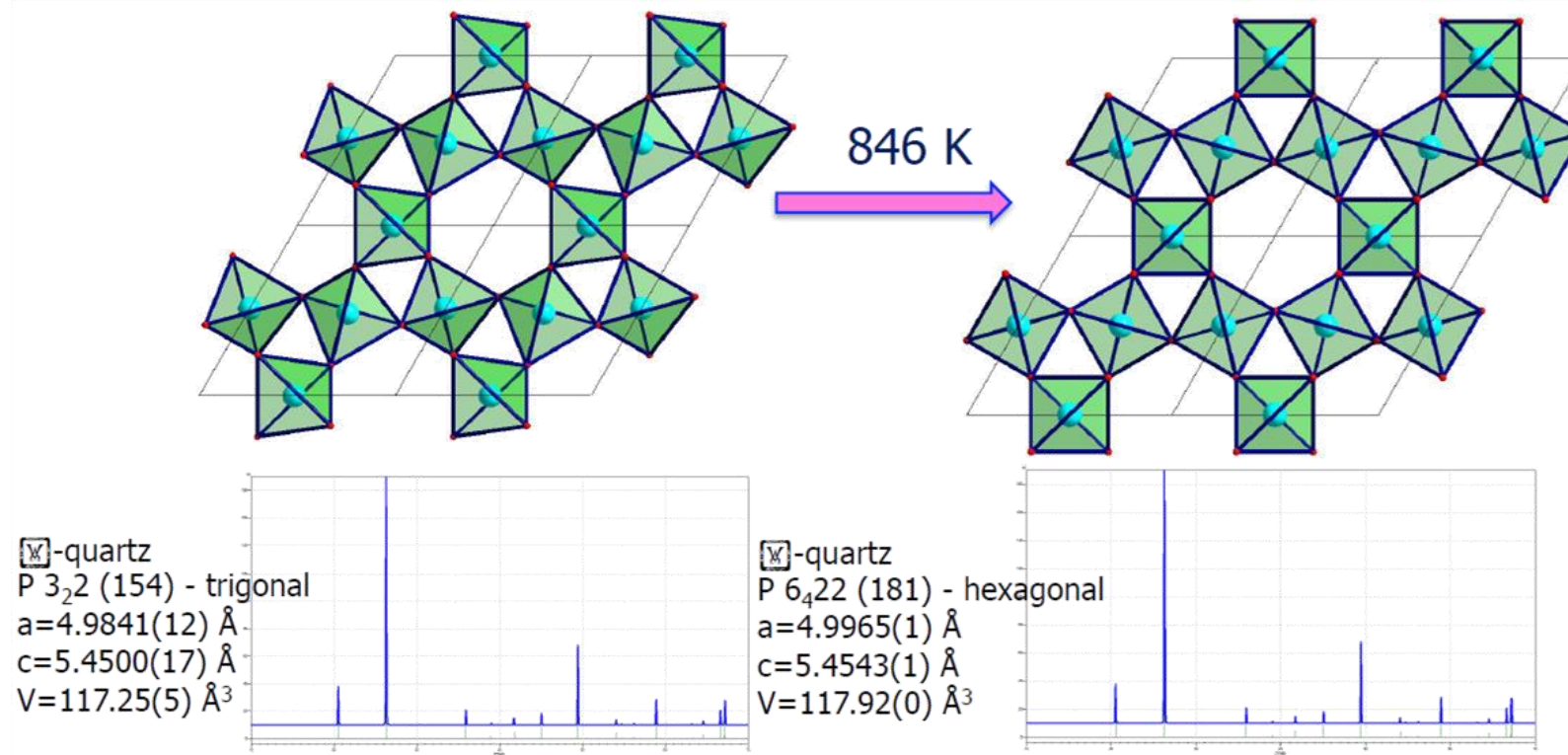
- Average crystal structure is obtained – no statements on the local structure
- No structural solution of non-crystalline materials
- No structure solution of disordered materials
- Limited structure solution of nano-crystalline substances (poorly defined Bragg peaks)



Methods for exact determination of the local structure – short range order.

The problem with the local structure....

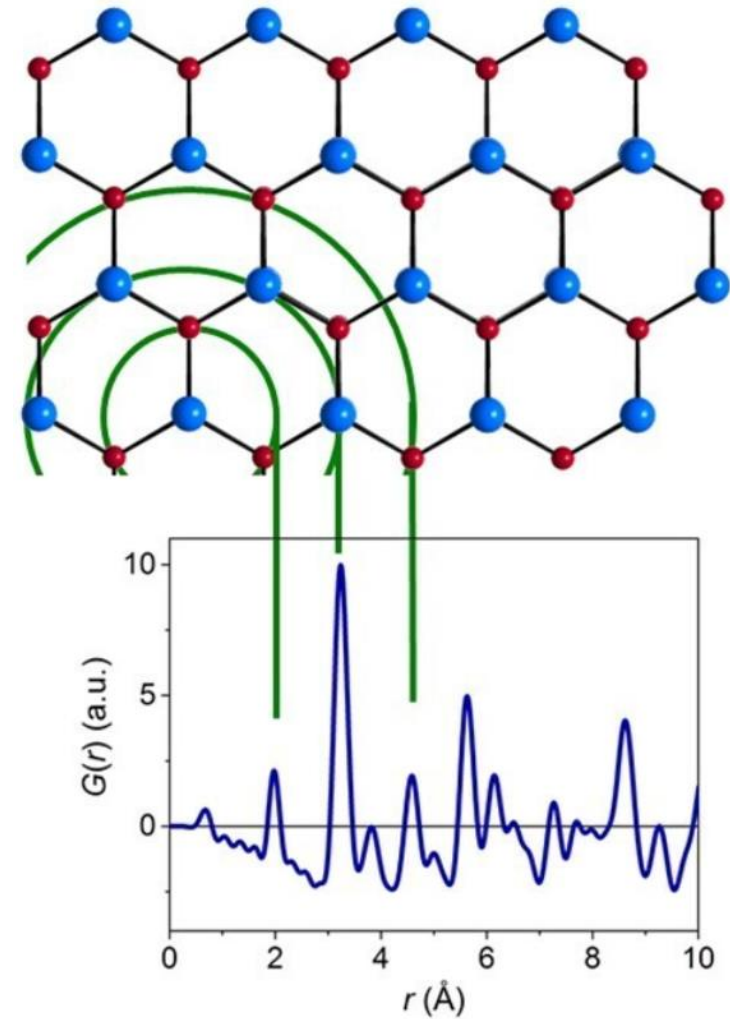
Average crystallographic structure is obtained *via* X-ray or neutron scattering – no statements on the local structure can be made.



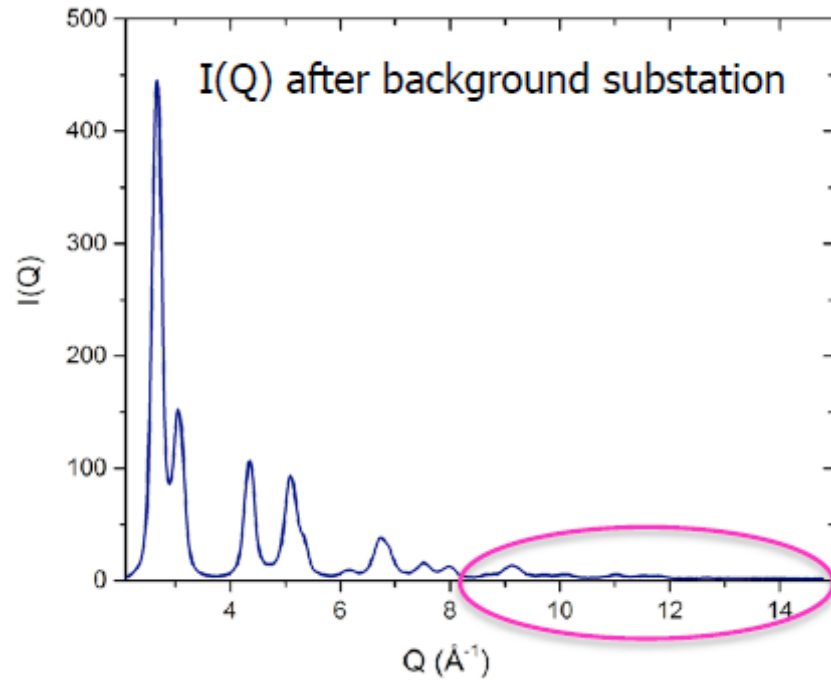
When we want to look closer - PDF

Both Bragg scattering and the underlying diffuse scattering are determined. The arrangement of atoms in the long range can be inferred from the Bragg peaks in an X-ray diffraction pattern. The short-range arrangement, i.e., the local atomic structure, is visible in the broad, less well-defined features of the diffraction pattern. This local structure is quantitatively described by the atomic pair distribution function.

- The pair distribution function (PDF) corresponds to a histogram of interatomic distances
- Each atom in the sample serves as an arbitrary centre from which all interatomic distances to all other atoms are measured
- All interatomic distances are summed and projected over the radius r

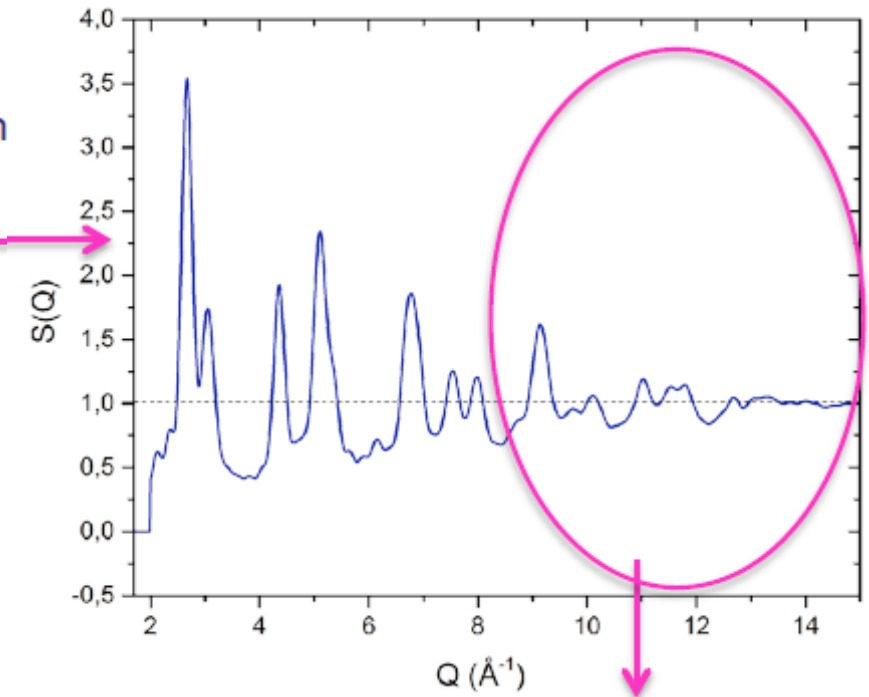


PDF - Pair Distribution Function



Total scattering structure function

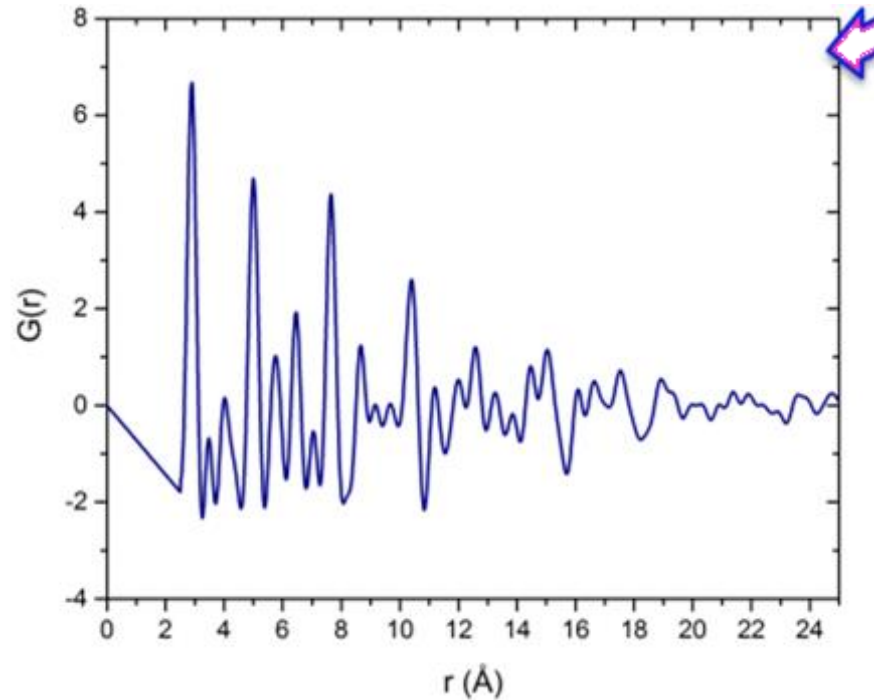
$$S(Q) = 1 + \frac{I(Q) - \langle f^2(Q) \rangle_{\text{composition}}}{\langle f(Q) \rangle_{\text{composition}}^2}$$



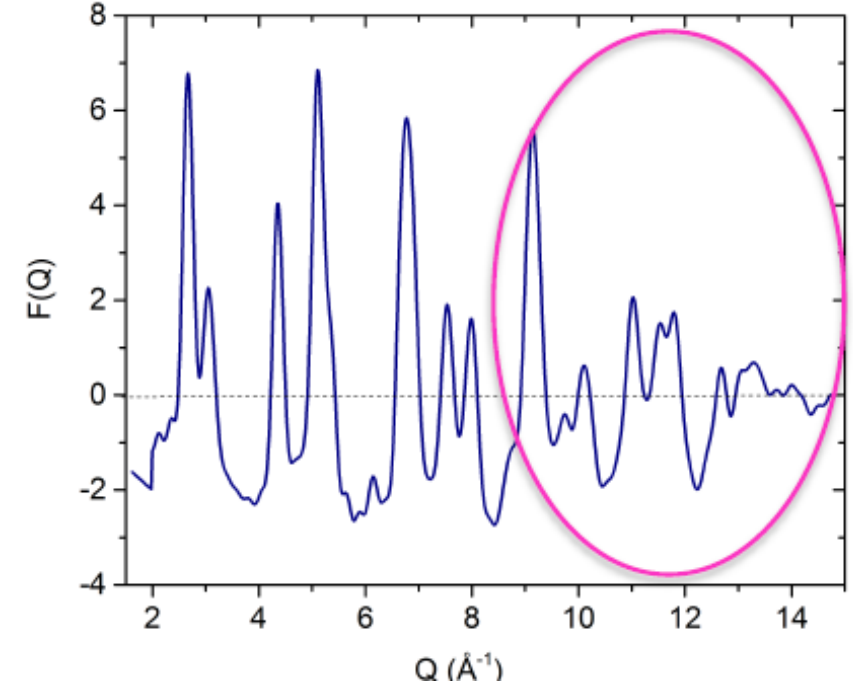
Reduced total scattering structure function

$$F(Q) = Q[S(Q) - 1]$$

PDF - Pair Distribution Function

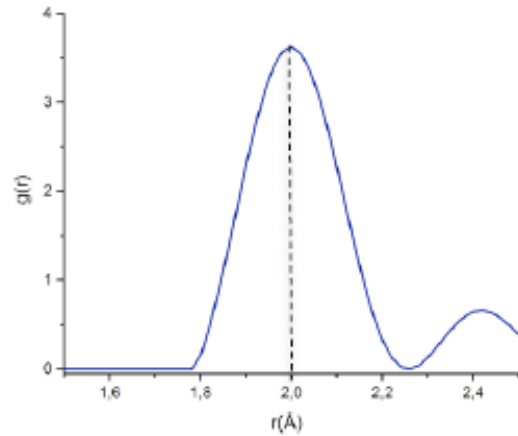


$$G(r) = \frac{2}{\pi} \int_0^{\infty} F(Q) \sin(Qr) dQ$$



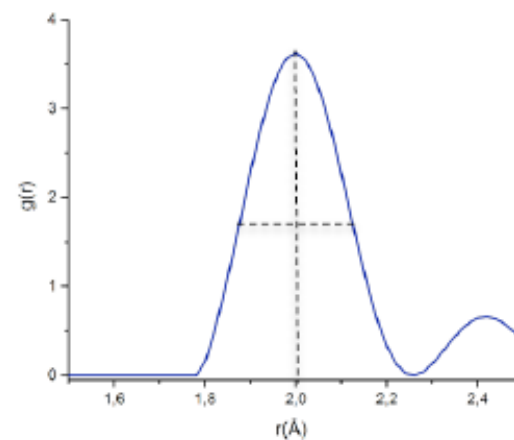
Fourier-Transformation

PDF - Pair Distribution Function



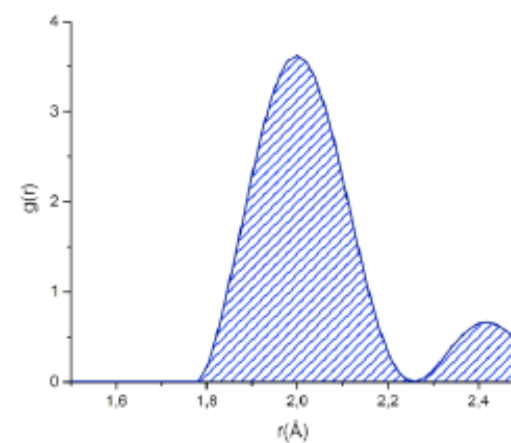
Peak positions

Average interatomic
distances



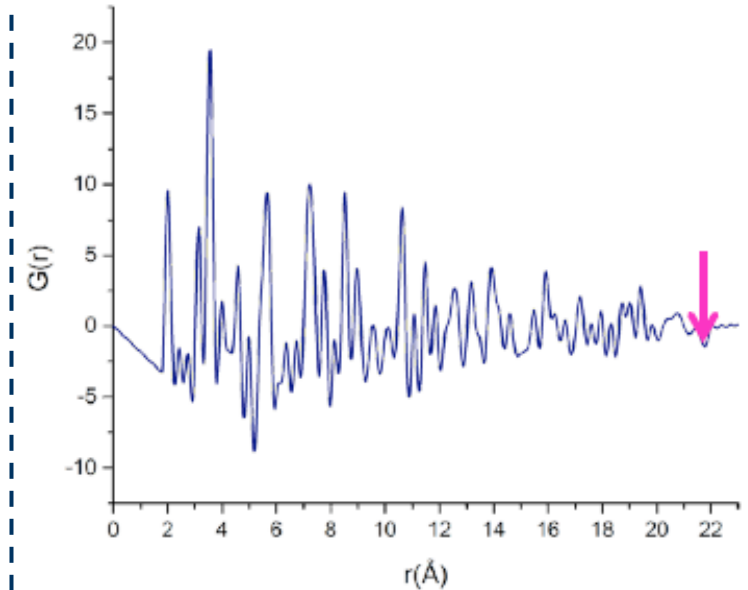
Peak Width

Structural disorders



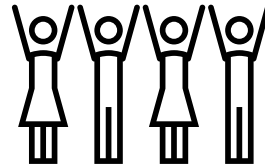
Integrals

Average coordination
properties



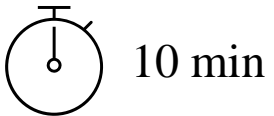
Cut-Off PDF

Particle size effect



Plenum

Methods-Table: Comparison results



Questions?

