

MCB 150

Proteins, Part 1

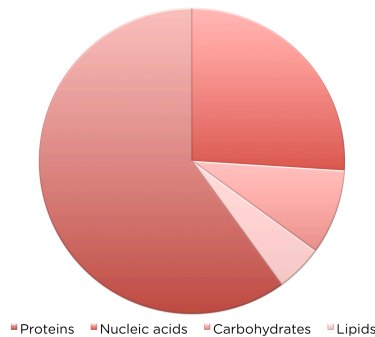
Today's Learning Catalytics Session ID is:
65931785

Announcements:

- Exam I is Thursday, February 8, from 7:00–9:00 PM
 - Check Canvas or your TA for room assignments
 - Additional practice questions should be available by tomorrow
 - Fill out Conflict Exam Request Form by Monday at 5:00 PM

Our Final Macromolecule: Proteins

- We are the product of our proteins and protein activity
 - The study of proteins and protein activity: Proteomics
- Proteins account for most of the dry weight in the cell



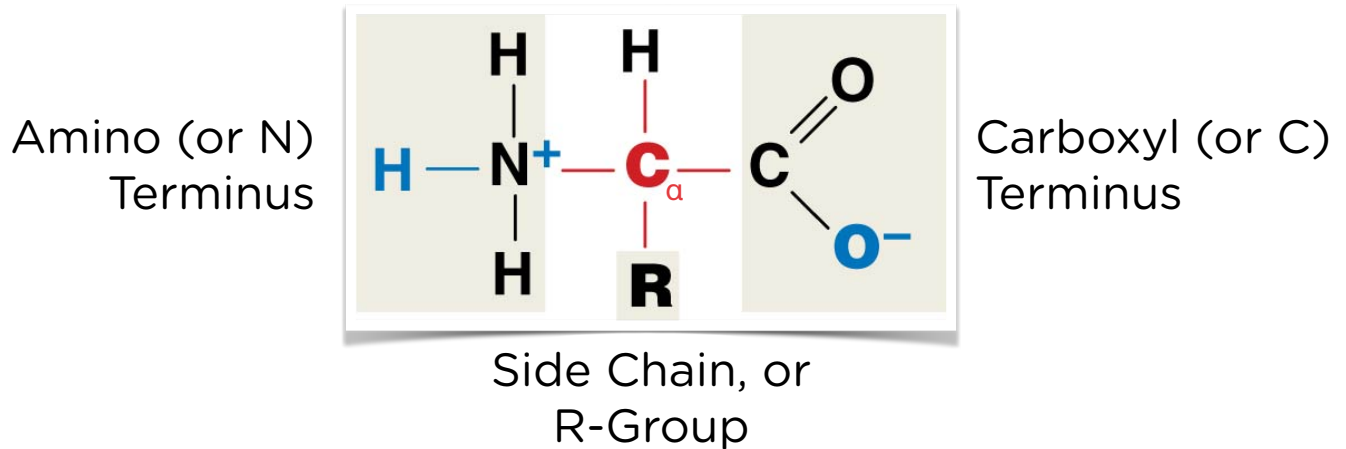
Proteins are involved in nearly all categories of cellular function:

- Movement (Actin/Myosin)
- Defense (Antibodies)
- Structure (Keratin)
- Transport (Hemoglobin)
- Signaling (Glucagon)
- Catalysis/Regulation/Metabolism

Most of our (useful) genetic information instructs the cell how to build proteins or regulates that process

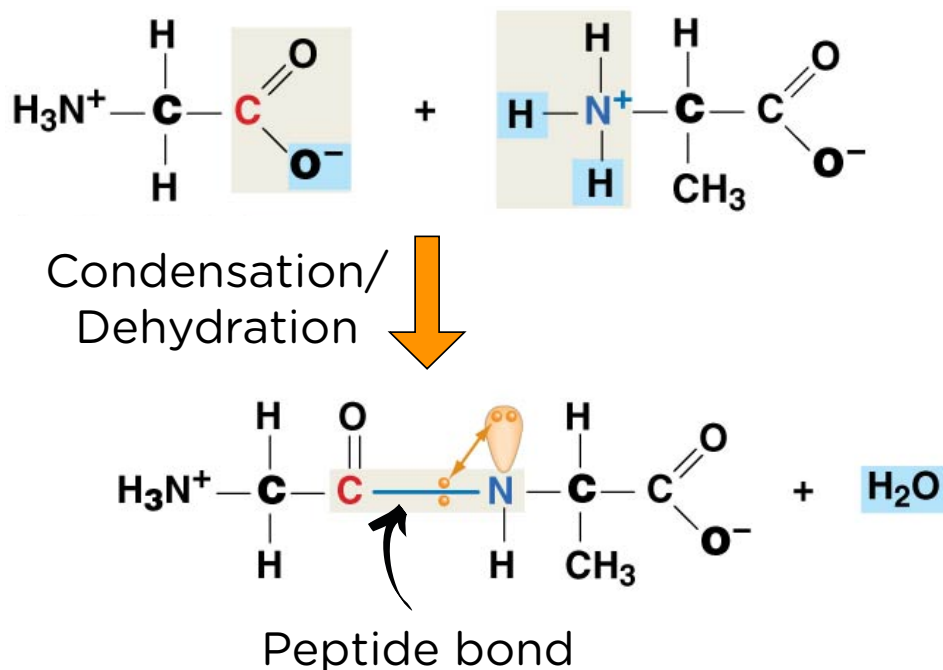
Amino Acids are the monomers of Proteins

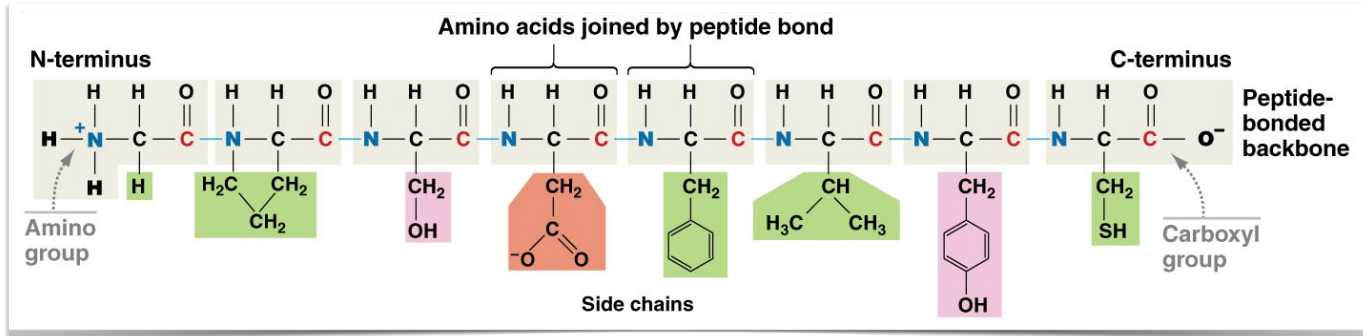
- Basic structure of an amino acid (ionized form):



The R group—the only part that differs—is what makes one amino acid different from another

Peptide bond formation:





During protein synthesis, ribosomes link amino acids by constructing covalent **PEPTIDE BONDS** that join the NH_2 (or NH_3^+) group of the incoming amino acid to the COOH (or COO^-) group of what was already there, in the $\text{N} \rightarrow \text{C}$ direction.

Some common (and familiar) terminology:

- Two amino acids = **DIPEPTIDE**
- A few amino acids = **OLIGOPEPTIDE**
- A long chain of amino acids = **POLYPEPTIDE**
- A polypeptide with a purpose = **PROTEIN**

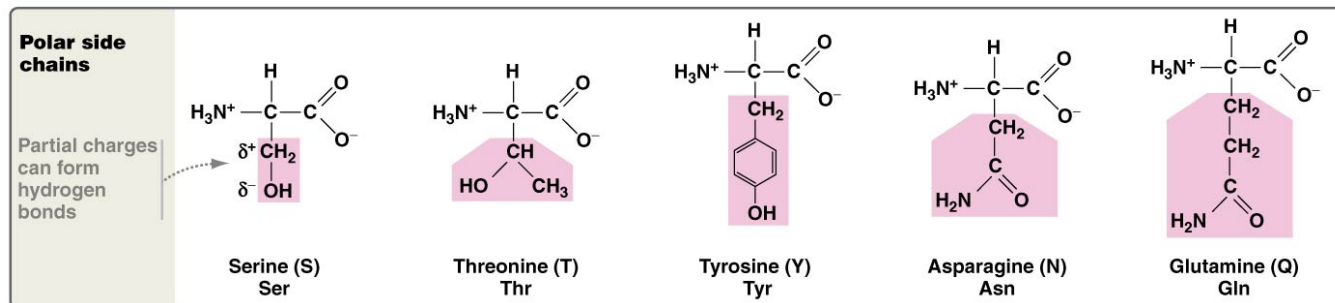
20 different amino acids commonly found in proteins

- differ only in R groups, which confer distinct properties to that amino acid
- large number of amino acids makes possible a huge number of different amino acid sequences
 - 20^2 (=400) possibilities for dipeptides
 - 20^3 (=8,000) possibilities for tripeptides
 - 20^5 (=3,200,000) possibilities for pentapeptides
 - most proteins are >100 amino acids!!

Amino acid R-groups (4 classes based on charge):

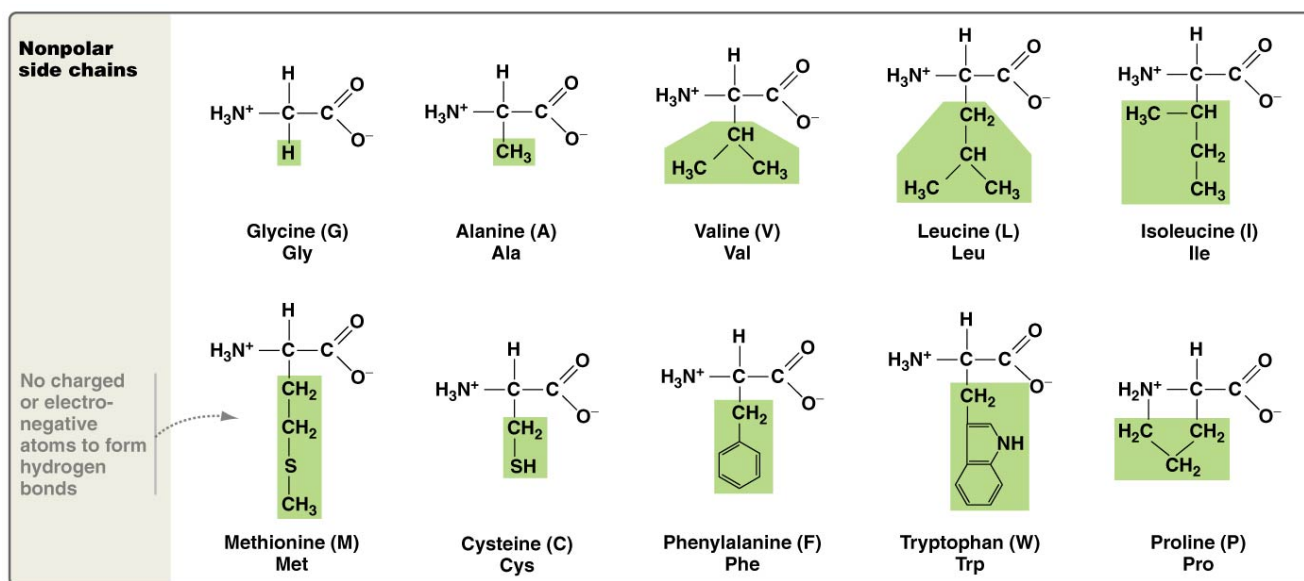
- Uncharged, but polar
- Uncharged and non-polar (hydrophobic)
- Positively-charged (basic)
- Negatively-charged (acidic)

Polar amino acids:



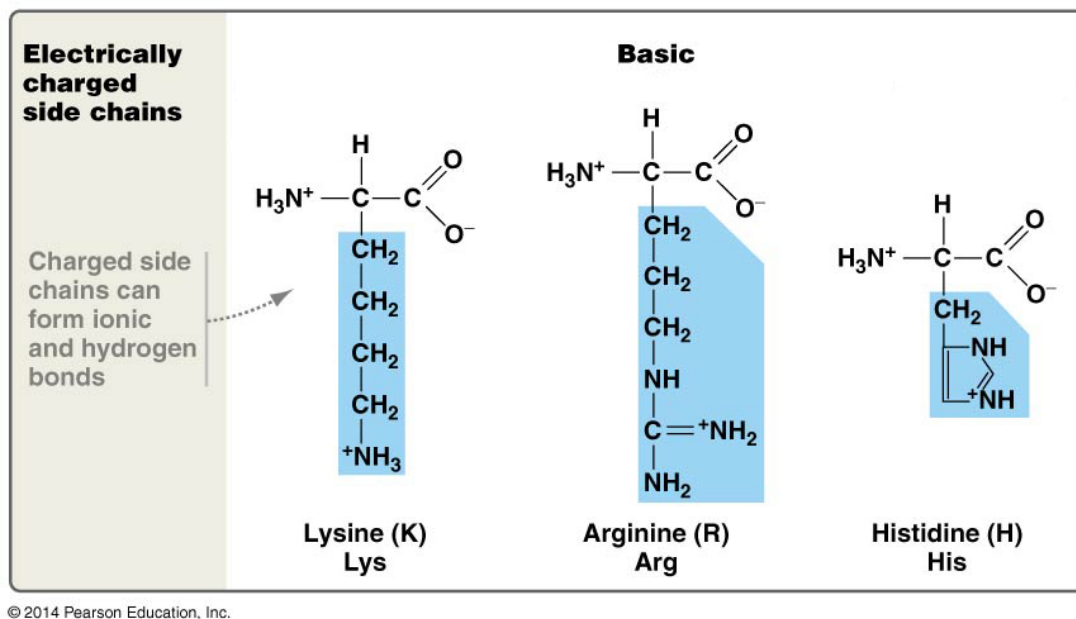
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Nonpolar (hydrophobic) amino acids:

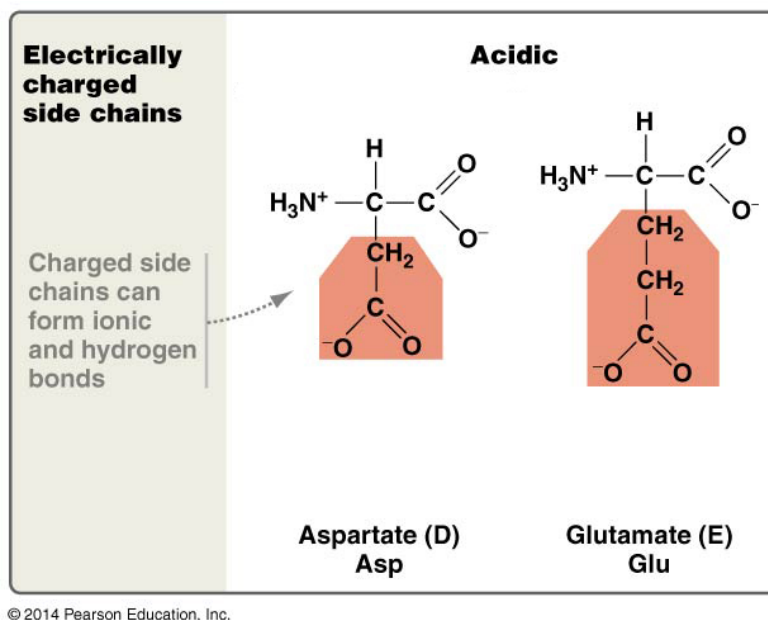


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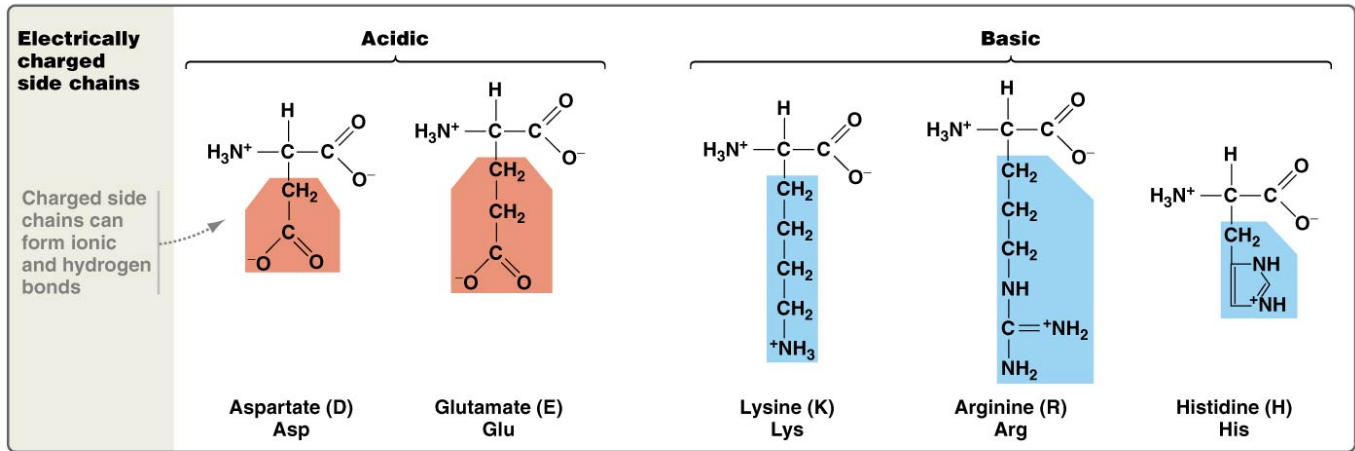
Basic (positively-charged) amino acids:



Acidic (negatively-charged) amino acids:



Charged amino acids:



Proteins exist in a virtually infinite number of 3-dimensional conformations

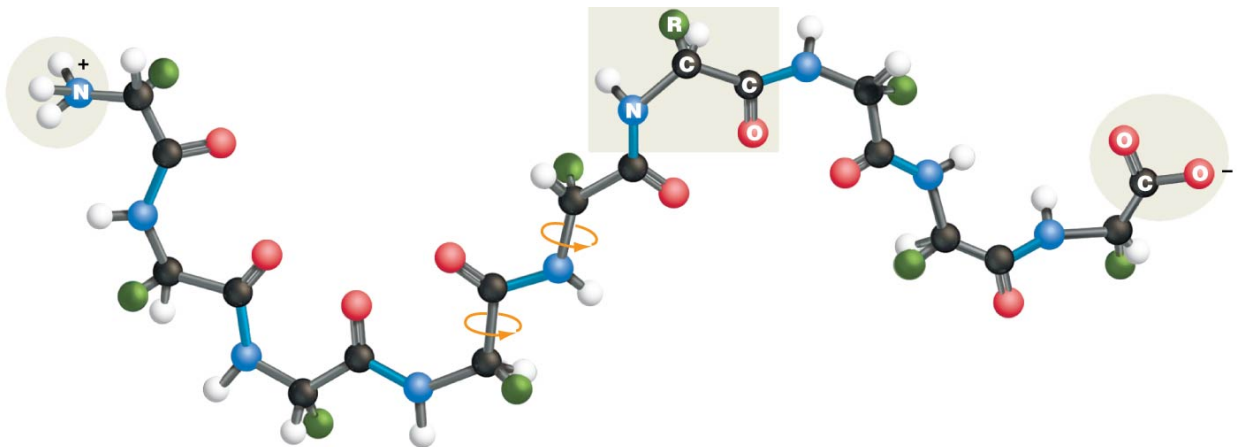
- That conformation is critical to the functioning of each protein
- The consequence of folding improperly is usually very significant
 - Alzheimer's, CF, Parkinson's, Mad Cow -- all caused by errors in protein folding → accumulation of toxic insoluble "gunk" (e.g. "plaques" in Alzheimer's)

To describe how linear protein chains fold into their 3-D conformations, protein structure is organized into 4 different categories:

- 1° (pronounced 'primary')
- 2° (pronounced 'secondary')
- 3° (pronounced 'tertiary')
- 4° (pronounced 'quaternary')

Primary structure (1°, or primary sequence):

- Linear sequence of amino acids from N → C ("beads on a string")
- All proteins have a UNIQUE primary structure

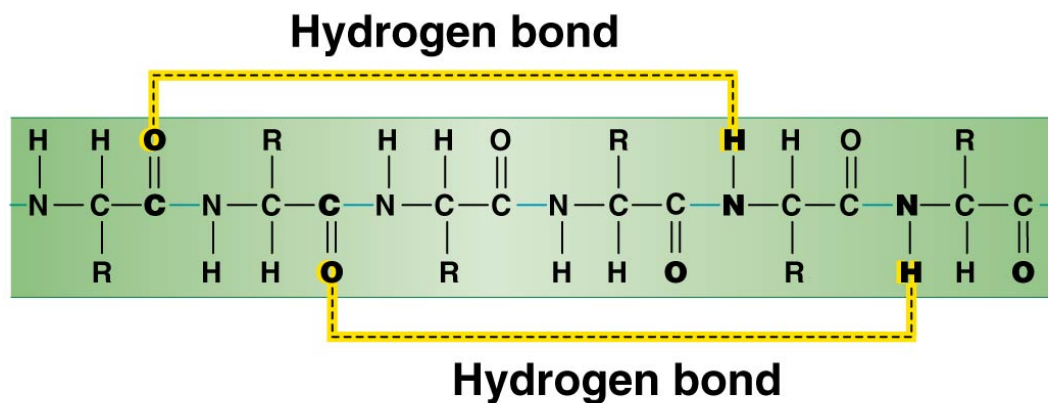


Secondary Structure (2°):

- First level of folding
- Stabilized by (relatively weak) hydrogen bonds between peptide linkages
 - Peptide backbone is polar (N-H is partially +, C=O is partially -)
- Independent of R groups, so found in most proteins
- α -helix and β -pleated sheet are 2 major types

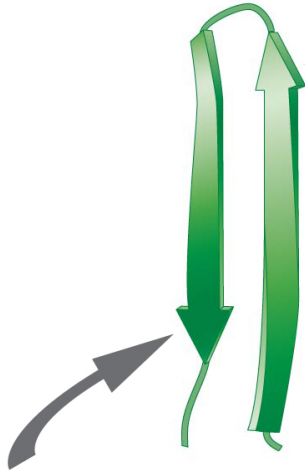
Secondary Structure (2°):

(a) Hydrogen bonds can form between nearby amino and carbonyl groups on the same polypeptide chain.

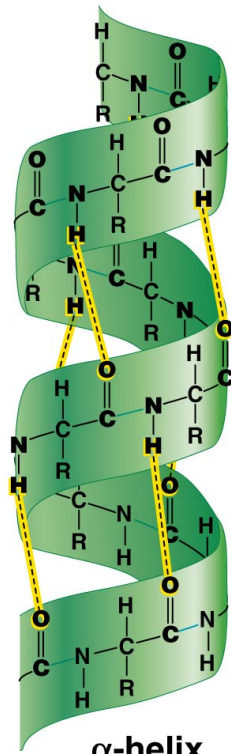


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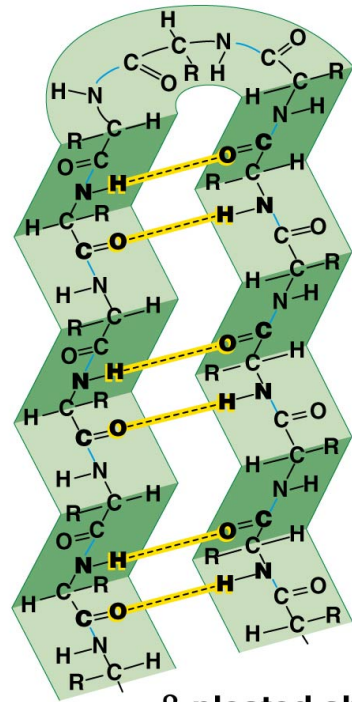
Secondary Structure (2°):



Arrowheads point toward the C-terminus of the primary structure



α -helix



β -pleated sheet