## Quiz Chapter 8 & 9 - Question Two

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$$10(.708) = 7.08[g]$$

$$\frac{7.08}{82} = .0863[mol]$$

$$P = \frac{nRT}{V}$$

$$\frac{.0863 \cdot .0821 \cdot 298}{.25} = 8.45[ATM]$$

$$8.45 - .7066 = 7.7434[ATM_{liquid}]$$

$$\frac{7.7434}{.7066} \cdot .0863 = .946[mol]$$

$$.946 \cdot 82 = 77.572[g]$$

$$(1)$$