

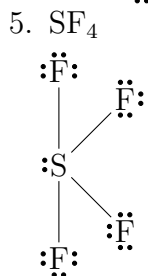
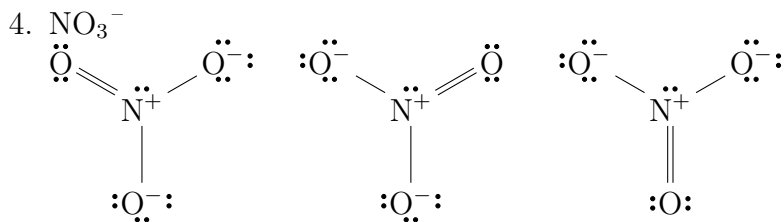
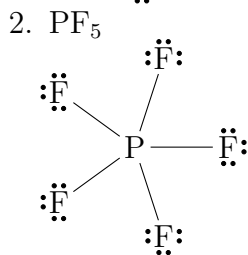
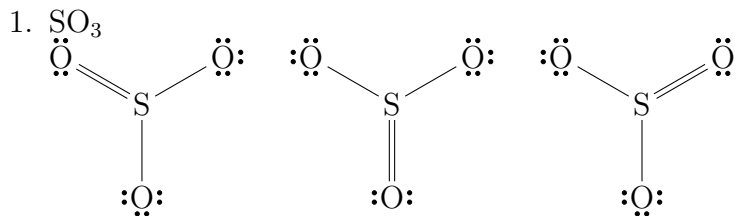
Problem Solving – Chapter 7

Michael Brodskiy

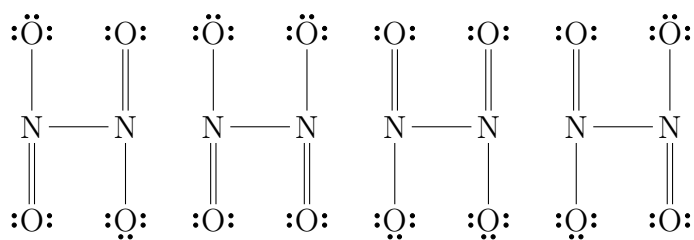
Instructor: Mr. Morgan

November 19, 2020

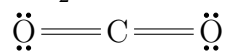
- Draw the Lewis Structure and all resonance forms for each:



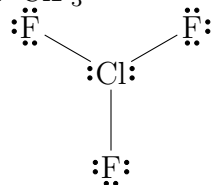
6. N_2O_4



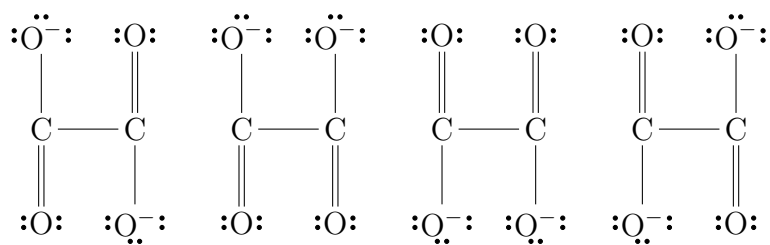
7. CO_2



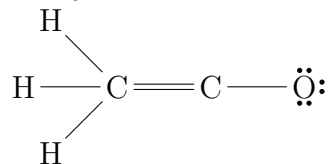
8. ClF_3



9. $\text{C}_2\text{O}_4^{2-}$



10. $\text{C}_2\text{H}_3\text{O}^-$



11. O_2^{2-}

12. NH_2OH

13. XeF_2

14. HSO_4^-

15. PO_4^{3-}

16. HSO_4^-

17. H_2Se

18. ClF_5

19. SeO_2

20. C_2H_4

21. SF_2

22. O_2

- 23. XeF_4
- 24. CHCl_3
- 25. SO_3^{2-}
- 26. CN^-
- 27. C_3H_8
- 28. ClF_2^+