## Quiz Chapter 5 — Question Two

Michael Brodskiy

Instructor: Mr. Morgan

November 5, 2020

$$\begin{split} \mathrm{mol_N} &= \frac{1 \cdot 2}{298 \cdot .0821} \\ &= .0817 [\mathrm{mol}] \\ \mathrm{mol_O} &= \frac{3 \cdot 2}{298 \cdot .0821} \\ &= .245 [\mathrm{mol}] \\ P &= \frac{(.245 + .0817) \cdot .0821 \cdot 298}{(3+2)} \\ &= 1.6 [atm] \end{split} \tag{1}$$