

Homework 6

Michael Brodskiy

Professor: A. Stepanyants

November 4, 2023

2. (a) We can find the density of states $d\varepsilon$ as:

$$D(\varepsilon) d\varepsilon = \frac{V \varepsilon^2}{\pi^2 \hbar^3 c^3} d\varepsilon$$

We can then find:

$$N = \int_0^{\varepsilon_F} \frac{V \varepsilon^2}{\pi^2 \hbar^3 c^3} d\varepsilon$$

$$N = \frac{V \varepsilon_F^3}{3\pi^2 \hbar^3 c^3}$$

$$N(3\pi^2 \hbar^3 c^3) = V \varepsilon_F^3$$

$$\varepsilon_F^3 = \frac{N}{V} (3\pi^2 \hbar^3 c^3)$$

$$\varepsilon_F^3 = 3n(\pi^2 \hbar^3 c^3)$$

$$\varepsilon_F^3 = \frac{3n}{\pi} (\pi \hbar c)^3$$

And finally:

$$\varepsilon_F = \left(\frac{3n}{\pi} \right)^{\frac{1}{3}} (\pi \hbar c)$$

- (b) The energy can be defined in a similar manner:

$$U = \int_0^{\varepsilon_F} \varepsilon D(\varepsilon) d\varepsilon$$

Which gives us:

$$U = \int_0^{\varepsilon_F} \frac{V \varepsilon^3}{\pi^2 \hbar^3 c^3} d\varepsilon$$

$$U = \frac{V \varepsilon_F^4}{4\pi^2 \hbar^3 c^3}$$

$$U = \left(\frac{V \varepsilon_F^3}{3\pi^2 \hbar^3 c^3} \right) \frac{3}{4} \varepsilon_F$$

And finally:

$$\boxed{U_o = \frac{3}{4} N \varepsilon_F}$$

3. (a) We know that the ground state energy may be expressed as:

$$U_o = \frac{3N\hbar^2}{10M} \left(\frac{3\pi^2 N}{V} \right)^{\frac{2}{3}}$$

We also know the relation:

$$P = -\frac{\partial U}{\partial V}$$

This gives us:

$$P = - \left(-\frac{2}{3} \frac{3\hbar^2 (3\pi^2)^{\frac{2}{3}}}{10M} \left(\frac{N}{V} \right)^{\frac{5}{3}} \right)$$

$$P = \frac{\hbar^2 (3\pi^2)^{\frac{2}{3}}}{5M} \left(\frac{N}{V} \right)^{\frac{5}{3}}$$

$$\boxed{P = \frac{(3\pi^2)^{\frac{2}{3}}}{5} \frac{\hbar^2}{M} \left(\frac{N}{V} \right)^{\frac{5}{3}}}$$

(b) We can define:

$$\delta\sigma(\tau) = \int \frac{1}{\tau} dU$$

We know the heat capacity of an electron gas as:

$$C_V = \frac{1}{2} \pi^2 N \left(\frac{\tau}{\tau_F} \right)$$

We know the heat capacity is equal to $\frac{\partial U}{\partial \tau}$, which allows us to write:

$$\sigma(\tau) = \int_0^\tau \frac{1}{\tau} C_v d\tau$$

$$\sigma(\tau) = \int_0^\tau \frac{1}{2} \pi^2 N \frac{1}{\tau_F} d\tau$$

$$\sigma(\tau) = \frac{1}{2}\pi^2 N \frac{\tau}{\tau_F}$$

Notice: $\sigma = C_V$, as is correct for a degenerate gas

5. (a) We can describe $n = \frac{N}{V}$ as:

$$\begin{aligned} n &= \frac{\rho}{M} = \frac{.081}{3 \cdot 1.67 \cdot 10^{-24}} \\ n &= 1.617 \cdot 10^{22} [\text{cm}^{-3}] \\ n &= 1.617 \cdot 10^{28} [\text{m}^{-3}] \end{aligned}$$

We can then use the formula:

$$\begin{aligned} \varepsilon_F &= \frac{\hbar^2}{2M} (3\pi^2 n)^{\frac{2}{3}} \\ \varepsilon_F &= \frac{(1.055 \cdot 10^{-34})^2}{2 \cdot 3 \cdot 1.67 \cdot 10^{-27}} (3\pi^2 (1.617 \cdot 10^{28}))^{\frac{2}{3}} \\ \varepsilon_F &= 6.79 \cdot 10^{-23} [\text{J}] \\ \varepsilon_F &= 4.24 \cdot 10^{-4} [\text{eV}] \end{aligned}$$

Now, we can find v_F as:

$$\begin{aligned} \varepsilon_F &= \frac{1}{2} M v_F^2 \\ v_F &= \sqrt{\frac{2\varepsilon_F}{M}} \\ v_F &= \sqrt{\frac{2(6.79 \cdot 10^{-23})}{3 \cdot 1.67 \cdot 10^{-27}}} \\ v_F &= 164.638 \left[\frac{\text{m}}{\text{s}} \right] \end{aligned}$$

And, finally, T_F :

$$\begin{aligned} T_F &= \frac{\tau_F}{k_B} = \frac{\varepsilon_F}{k_B} \\ T_F &= \frac{6.79 \cdot 10^{-23}}{1.381 \cdot 10^{-23}} \\ T_F &= \frac{6.79}{1.381} \\ T_F &= 4.92 [\text{K}] \end{aligned}$$

(b) We may begin by using:

$$C_V = \frac{\pi^2 N}{2} \frac{\tau}{\tau_F}$$

$$C_V = \frac{\pi^2}{2\tau_F} (N\tau)$$

$$C_V = \frac{2(\pi^2)}{(4.92)} (Nk_B T)$$

$$\boxed{C_V = 1.003 Nk_B T}$$

The coefficient $1.003 < 2.89$, the experimental value.

6. (a) The energy may be defined as:

$$U = -G \int_0^R \frac{\rho^2 (4\pi r^2) \left(\frac{4}{3}\pi r^3\right)}{r} dr$$

$$U = -\rho^2 G \int_0^R \frac{(4\pi)^2 r^5}{3r} dr$$

$$U = -\frac{16\pi^2 \rho^2 G}{3} \int_0^R r^4 dr$$

$$U = -\frac{16\pi^2 \rho^2 G}{3} \frac{R^5}{5}$$

$$U = -\frac{16\pi^2 \rho^2 G R^5}{15}$$

We know the density becomes $\rightarrow \frac{3M}{4\pi R^3}$, which gives:

$$\boxed{U = -\frac{3M^2 G}{5R}}$$

Thus, we can see this is on the order of $\frac{GM^2}{R}$

(b) The Fermi energy may be defined as:

$$\varepsilon_F = \frac{\hbar^2}{2M} (3\pi^2 n)^{\frac{2}{3}}$$

And the volume is:

$$V = \frac{4}{3}\pi R^3$$

We can write the total energy as:

$$NK_e = N\varepsilon_F$$

$$K_{tot} = N \frac{\hbar^2}{2m} \left(3\pi^2 \frac{N}{V} \right)^{\frac{2}{3}}$$

$$K_{tot} = \frac{\hbar^2}{2m} \left(\frac{9\pi^2}{4\pi R^3} \right)^{\frac{2}{3}} N^{\frac{5}{3}}$$

$$K_{tot} = \frac{\hbar^2}{2mR^2} \left(\frac{9}{4}\pi \right)^{\frac{2}{3}} N^{\frac{5}{3}}$$

$$K_{tot} = \frac{\hbar^2}{mR^2} \left(\frac{81}{128}\pi^2 \right)^{\frac{2}{3}} N^{\frac{5}{3}}$$

This can then be written:

$$K_{tot} = c \frac{\hbar^2}{mR^2} N^{\frac{5}{3}}$$

with $c = \left(\frac{81}{128}\pi^2 \right)^{\frac{2}{3}}$. Now, since $M_H \ll M$, we can write $N = \frac{M}{M_H}$, which gives us:

$$K_{tot} = c \frac{\hbar^2 M^{\frac{5}{3}}}{m M_H^{\frac{5}{3}} R^2}$$

Thus, we can see the momentum is on the order of $\frac{\hbar^2 M^{\frac{5}{3}}}{m M_H^{\frac{5}{3}} R^2}$

(c) From the two results above, we may write:

$$\frac{\hbar^2 M^{\frac{5}{3}}}{m M_H^{\frac{5}{3}} R^2} \approx \frac{GM^2}{R}$$

Rearranging, we get:

$$\frac{\hbar^2}{Gm M_H^{\frac{5}{3}}} \approx M^{\frac{1}{3}} R$$

Now we can calculate:

$$M^{\frac{1}{3}} R \approx \frac{\hbar^2}{Gm M_H^{\frac{5}{3}}}$$

We will have two different values for \hbar , one in electron-volts, and one in joules to verify the correct dimensionality with respect to G and m :

$$M^{\frac{1}{3}}R \approx \frac{(6.582 \cdot 10^{-22})(1.055 \cdot 10^{-34})c^2}{(6.67 \cdot 10^{-11})[.511](1.67 \cdot 10^{-24})^{\frac{5}{3}}}$$

$$M^{\frac{1}{3}}R \approx 7.8 \cdot 10^{11}$$

Note: this result does not have the correct units. Correcting this, we find:

$$7.8 \cdot 10^{11}(10^8) = 7.8 \cdot 10^{19}$$

Then multiplying by the constants/coefficients removed in (a) and (b), we get:

$$7.8 \cdot 10^{19} \cdot \left(\frac{81\pi^2}{128}\right)^{\frac{2}{3}} \frac{5}{3} = 4.4 \cdot 10^{20}$$

Thus, we can see that the value is on the order of

$$\boxed{M^{\frac{1}{3}}R \rightarrow 10^{20}[\text{g}^{\frac{1}{3}}\text{cm}]}$$

(d) We know:

$$\rho = \frac{3M}{4\pi R^3}$$

$$\rho = \frac{3M}{4\pi \left(\frac{10^{20}}{M}\right)}$$

$$\rho = \frac{3M^2}{4\pi (10^{60})}$$

$$\rho = \frac{3(2 \cdot 10^{66})}{4\pi (10^{60})}$$

$$\boxed{\rho = 4.77465 \cdot 10^5 \left[\frac{\text{g}}{\text{cm}^3}\right]}$$

(e) The only difference between a proton and electron gas would be the energy per c^2 , which, for neutrons, is on the order of 10^9 (giga) instead of 10^6 (mega). Thus, $M^{\frac{1}{3}}R$ would be divided by a factor of 10^3 :

$$10^{20}(10^{-3}) = 10^{17}[\text{g}^{\frac{1}{3}}\text{cm}]$$

We can then write:

$$R = \frac{10^{17}}{(2 \cdot 10^{33})^{\frac{1}{3}}} = 793701[\text{cm}]$$

$$\boxed{R = 7.937[\text{km}]}$$

7. Since we want to find the critical temperature at which $N_e < N$, we can define:

$$N = N_e = \frac{2.404V\tau^3}{\pi^2\hbar^3c^3}$$

Which can be rearranged to give:

$$\tau = \left(\frac{N\pi^2\hbar^3c^3}{2.404V} \right)^{\frac{1}{3}}$$

We can convert:

$$10^{20}[\text{cm}^{-3}] \rightarrow 10^{26}[\text{m}^{-3}]$$

We can then calculate:

$$k_B T = \left(\frac{\pi^2(1.055 \cdot 10^{-34})^3(3 \cdot 10^8)^3}{2.404} 10^{26} \right)^{\frac{1}{3}}$$

$$k_B T = 2.35 \cdot 10^{-17}$$

$$\boxed{T = 1.7 \cdot 10^6[\text{K}]}$$

Below this temperature, $N_e < N$