$$g(N,s) = \frac{N!}{\left(\frac{1}{2}N + s\right)\left(\frac{1}{2}N - s\right)} = \frac{N!}{N_{\uparrow}!N_{\downarrow}!} \qquad g(N,s) \approx \sqrt{\frac{2}{\pi N}} 2^{N} e^{-\frac{2s^{2}}{N}}$$

$$U(s) = -2smB \qquad 2s = N_{\uparrow} - N_{\downarrow}$$

$$\sigma(N,s) = \ln(g(N,s)) \qquad \qquad \frac{1}{\tau} = \left(\frac{\partial \sigma}{\partial U}\right)_{N,V}$$

$$S = k_{B}\sigma \qquad \qquad \tau = k_{B}T$$

$$\frac{\text{Accessible States}}{g(s) = \sum_{s} g_{1}(s_{1})g_{2}(s - s_{1})} \qquad \text{System/Reservoir State Probability:}$$

$$z = \sum_{s} e^{-\frac{\varepsilon_{s}}{\tau}} \qquad P = -\left(\frac{\partial U}{\partial V}\right)_{\sigma} = \tau \left(\frac{\partial \sigma}{\partial V}\right)_{U} = -\left(\frac{\partial F}{\partial V}\right)_{\tau}$$

$$F = U - \tau\sigma \quad \text{min. in eq. with const } \tau, V \qquad F = -\tau \ln(z) \quad \text{to derive } P, \sigma$$

Ideal Monatomic Gas

Given
$$N$$
 atoms: $z_N = \frac{(n_Q V)^N}{N!}$ If $n = \frac{N}{V} << n_Q$, $n_Q = \left(\frac{M\tau}{2\pi\hbar^2}\right)^{\frac{3}{2}}$
$$PV = N\tau \qquad \qquad \sigma = N \left[\ln\left(\frac{n_Q}{n}\right) + \frac{5}{2}\right] \qquad \qquad C_V = \frac{3}{2}N$$

A process is reversible if the system remains infinitesimally close to the equilibrium state at all times during the process.

Average in Mode at freq.
$$\omega$$
: $\langle s \rangle = \frac{1}{e^{\frac{\hbar \omega}{\tau}} - 1}$ Energy Density at τ : $\langle s \rangle = \frac{U}{V} = \frac{\pi^2}{15\hbar^3 c^3} \tau^4$ Radiant energy per vol: $U_\omega = \frac{\hbar}{\pi^2 c^3} \frac{\omega^3}{e^{\frac{\hbar \omega}{\tau}} - 1}$ Flux Density: $J_U = \sigma_B T^4$, $\sigma_B = \frac{\pi^2 k B^4}{60\hbar^3 c^2}$ Heat Capacity of Dielectric Solid: $C_V = \frac{12\pi^4 N k_B}{5} \left(\frac{T}{\theta}\right)^3 \to \theta = \left(\frac{\hbar \omega}{k_B}\right) \left(\frac{6\pi^2 N}{V}\right)^{\frac{1}{3}}$

$$\mu = \left(\frac{\partial F}{\partial N}\right)_{\tau,V} = \left(\frac{\partial U}{\partial N}\right)_{\sigma,V} = -\tau \left(\frac{\partial \sigma}{\partial N}\right)_{U,V}$$
 In diffusive equilibrium if: $\mu_1 = \mu_2$

$$\mu = \mu_{int} + \mu_{ext}$$

$$\mu_{int} = \tau \ln \left(\frac{n}{n_Q}\right)$$

$$\mu_{ext} = \frac{U_{ext}}{N}$$

$$3 = \sum_{N} \sum_{s} e^{\frac{N\mu - \varepsilon_s}{\tau}}$$

Gibbs Factor: $P(N, \varepsilon_s) = \frac{e^{\frac{N\mu - \varepsilon_s}{\tau}}}{3}$ Prob. chem. potential μ and temp τ has N particles in q.s. s of energy ε_s

$$\lambda = e^{\frac{\mu}{\tau}} \to 3 = \sum \lambda^N e^{-\frac{\varepsilon_s}{\tau}}$$

Therm. Average:
$$\langle N \rangle = \lambda \frac{\partial}{\partial \lambda} \ln(3)$$

Quant. Particle in Box: $\varepsilon = \frac{\hbar^2 \pi^2}{2mL^2} (n_x^2 + n_y^2 + n_z^2)$

$$U = \tau^{2} \left(\frac{\partial \ln(z)}{\partial \tau}\right)_{V}$$

$$W = PA\Delta x = P\Delta V$$

$$\int_{-\infty}^{\infty} e^{-\alpha^{2}n^{2}} dn = \frac{\sqrt{\pi}}{2\alpha}$$

$$\sigma = -\left(\frac{\partial F}{\partial \tau}\right)_{VN}$$

$$\langle N \rangle = \sum_{N} \sum_{s} N(N, \varepsilon_{s}) = \tau \left(\frac{\partial \ln(3)}{\partial \mu}\right)_{\tau, V}$$

$$\langle \varepsilon_s \rangle = \sum_N \sum_s \varepsilon_s(N, \varepsilon_s) = \tau^2 \left(\frac{\partial \ln(3)}{\partial \tau} \right)_{\mu, V} + \tau \mu \left(\frac{\partial \ln(3)}{\partial \mu} \right)_{\tau, V} \qquad f(\varepsilon_n) \text{ avg. occupancy}$$

Bose-Einstein:
$$f(\varepsilon_n) = \frac{1}{e^{\frac{\varepsilon_n - \mu}{\tau}} - 1} \qquad f($$

$$f(\varepsilon_n) = \frac{1}{e^{\frac{\varepsilon_n - \mu}{\tau}} + 1}$$

Classical Limit:
$$f(\varepsilon_n) = e^{\frac{\mu - \varepsilon_n}{\tau}}$$

	ΔU	$\Delta \sigma$	W	Q
Rev. Isothermal	0	$N \ln \left(\frac{V_2}{V_1} \right)$	$-N au \ln\left(\frac{V_2}{V_1}\right)$	$N au \ln \left(rac{V_2}{V_1} ight)$
Rev. Isentropic	$-\frac{3}{2}N\tau_1\left[1-\left(\frac{V_1}{V_2}\right)^{\frac{2}{3}}\right]$	0	$ -\frac{3}{2}N\tau_1 \left[1 - \left(\frac{V_1}{V_2}\right)^{\frac{2}{3}}\right] $	0
Irrev. Expansion	0	$N \ln \left(\frac{V_2}{V_1} \right)$	0	0

$\underline{\text{Constants}}$

$$k_B = 1.381 \cdot 10^{-23} \left[\frac{J}{K} \right]$$
 $\sigma_B = 5.67 \cdot 10^{-8} \left[\frac{J}{m^2 s K^4} \right]$ $c = 3 \cdot 10^8 \left[\frac{m}{s} \right]$