

TABLE 2-352 Saturated Water Substance—Temperature (SI units)

Temp., K	Pressure, bar ^a	Volume, m ³ /kg		Enthalpy, kJ/kg	Entropy, kJ/(kg·K)		Specific heat, C _p , kJ/(kg·K)		Viscosity, Ns/m ²		Thermal conductivity, W/(m·K)		Prandtl no.		Surface tension, N/m	Temp., K
		Condensed†	Vapor	Condensed†	Vapor	Condensed†	Vapor	Condensed†	Vapor	Condensed†	Vapor	Condensed†	Vapor	Condensed†		
150	6.30–11	1.073–3	9.55+9	–539.6	227.3	–2.187	16.54	1.155			3.73					150
160	7.72–10	1.074–3	9.62+8	–525.7	229.1	–2.106	15.49	1.233			3.52					160
170	7.99–9	1.076–3	1.08+8	–511.7	231.0	–2.026	14.57	1.311			3.34					170
180	5.38–8	1.077–3	1.55+7	–497.8	232.8	–1.947	13.76	1.389			3.18					180
190	3.23–7	1.078–3	2.72+6	–483.8	234.7	–1.868	13.03	1.467			3.04					190
200	1.62–6	1.079–3	5.69+5	–467.5	236.6	–1.789	12.38	1.545			2.91					200
210	7.01–6	1.081–3	1.39+5	–451.2	238.4	–1.711	11.79	1.623			2.79					210
220	2.65–5	1.082–3	3.83+4	–435.0	240.3	–1.633	11.20	1.701			2.69					220
230	8.91–5	1.084–3	1.18+4	–416.3	242.1	–1.555	10.79	1.779			2.59					230
240	3.72–4	1.085–3	4.07+3	–400.1	244.0	–1.478	10.35	1.857			2.50					240
250	7.59–4	1.087–3	1.52+3	–381.5	245.9	–1.400	9.954	1.935			2.42					250
255	1.23–3	1.087–3	956.4	–369.8	246.8	–1.361	9.768	1.974			2.38					255
260	1.96–3	1.088–3	612.2	–360.5	247.7	–1.323	9.590	2.013			2.35					260
265	3.06–3	1.089–3	400.4	–351.2	248.6	–1.281	9.461	2.052			2.31					265
270	4.69–3	1.090–3	265.4	–339.6	249.6	–1.236	9.255	2.091			2.27					270
273.15	6.11–3	1.091–3	206.3	–333.5	250.2	–1.221	9.158	2.116			2.26					273.15
273.15	0.00611	1.000–3	206.3	0	250.2	0.000	9.158	4.217	1.854	1.750–6	0.569	0.0182	12.99	0.815	0.0755	273.15
275	0.00697	1.000–3	181.7	7.8	250.5	0.028	9.109	4.211	1.855	1.652–6	0.574	0.0183	12.22	0.817	0.0753	275
280	0.00990	1.000–3	130.4	28.8	251.4	0.104	8.980	4.198	1.858	1.422–6	0.582	0.0186	10.26	0.825	0.0748	280
285	0.01387	1.000–3	99.4	49.8	252.3	0.178	8.857	4.189	1.861	1.225–6	0.590	0.0189	8.81	0.833	0.0743	285
290	0.01917	1.001–3	69.7	70.7	253.2	0.251	8.740	4.184	1.864	1.080–6	0.598	0.0193	7.56	0.841	0.0737	290
295	0.02617	1.002–3	51.94	91.6	254.1	0.323	8.627	4.181	1.868	959–6	0.606	0.0195	6.62	0.849	0.0727	295
300	0.03531	1.003–3	39.13	112.5	255.0	0.393	8.520	4.179	1.872	855–6	0.613	0.0196	5.83	0.857	0.0717	300
305	0.04712	1.005–3	27.90	133.4	255.9	0.462	8.417	4.178	1.877	769–6	0.620	0.0201	5.20	0.865	0.0709	305
310	0.06221	1.007–3	22.93	154.3	256.8	0.530	8.318	4.178	1.882	695–6	0.628	0.0204	4.62	0.873	0.0700	310
315	0.08132	1.009–3	17.82	175.2	257.7	0.597	8.224	4.179	1.888	631–6	0.634	0.0207	4.16	0.883	0.0692	315
320	0.1053	1.011–3	13.98	196.1	258.6	0.649	8.151	4.180	1.895	577–6	0.640	0.0210	3.77	0.894	0.0683	320
325	0.1351	1.013–3	11.06	217.0	259.5	0.727	8.046	4.182	1.903	528–6	0.645	0.0213	3.42	0.901	0.0675	325
330	0.1719	1.016–3	8.82	237.9	260.4	0.791	7.962	4.184	1.911	489–6	0.650	0.0217	3.15	0.908	0.0666	330
335	0.2167	1.018–3	7.09	258.8	261.3	0.854	7.881	4.186	1.920	453–6	0.655	0.0220	2.88	0.916	0.0658	335
340	0.2713	1.021–3	5.74	279.8	262.2	0.916	7.804	4.188	1.930	420–6	0.660	0.0223	2.66	0.925	0.0649	340
345	0.3372	1.024–3	4.683	300.7	263.0	0.977	7.729	4.191	1.941	389–6	0.665	0.0226	2.45	0.933	0.0641	345
350	0.4163	1.027–3	3.846	321.7	263.9	1.038	7.657	4.195	1.954	365–6	0.668	0.0230	2.29	0.942	0.0632	350
355	0.5100	1.030–3	3.180	342.7	264.7	1.097	7.588	4.199	1.968	343–6	0.671	0.0233	2.14	0.951	0.0623	355
360	0.6209	1.034–3	2.645	363.7	265.5	1.156	7.521	4.203	1.983	324–6	0.674	0.0237	2.02	0.960	0.0614	360
365	0.7514	1.038–3	2.212	384.7	266.3	1.214	7.456	4.209	1.999	306–6	0.677	0.0241	1.91	0.969	0.0605	365
370	0.9040	1.041–3	1.861	405.8	267.1	1.271	7.394	4.214	2.017	289–6	0.679	0.0245	1.80	0.978	0.0595	370
373.15	1.0133	1.044–3	1.679	419.1	267.6	1.307	7.356	4.217	2.029	279–6	0.680	0.0248	1.76	0.984	0.0589	373.15
375	1.0815	1.045–3	1.574	426.8	267.9	1.328	7.333	4.220	2.036	274–6	0.681	0.0249	1.70	0.987	0.0586	375
380	1.2869	1.049–3	1.337	448.0	268.7	1.384	7.275	4.226	2.057	260–6	0.683	0.0254	1.61	0.995	0.0576	380
385	1.5233	1.053–3	1.142	469.2	269.4	1.439	7.218	4.232	2.080	248–6	0.685	0.0258	1.53	1.004	0.0566	385
390	1.794	1.058–3	0.980	490.4	270.2	1.494	7.163	4.239	2.104	237–6	0.686	0.0263	1.47	1.013	0.0556	390
400	2.455	1.067–3	0.731	532.9	271.6	1.605	7.058	4.256	2.158	217–6	0.688	0.0272	1.34	1.033	0.0536	400
410	3.302	1.077–3	0.553	575.6	272.9	1.708	6.959	4.278	2.221	200–6	0.688	0.0282	1.24	1.054	0.0515	410
420	4.370	1.088–3	0.425	618.6	274.2	1.810	6.865	4.302	2.291	185–6	0.688	0.0293	1.16	1.075	0.0494	420
430	5.699	1.099–3	0.331	661.8	275.3	1.911	6.775	4.331	2.369	173–6	0.685	0.0304	1.09	1.10	0.0472	430

440	7.333	1.110-3	0.261	705.3	2764	2.011	6.689	4.36	2.46	162-6	14.50-6	0.682	0.0317	1.04	1.12	0.0451	440
450	9.319	1.123-3	0.208	749.2	2773	2.109	6.607	4.40	2.56	152-6	14.85-6	0.678	0.0331	0.99	1.14	0.0429	450
460	11.71	1.137-3	0.167	793.5	2782	2.205	6.528	4.44	2.68	143-6	15.19-6	0.673	0.0346	0.95	1.17	0.0407	460
470	14.55	1.152-3	0.136	838.2	2789	2.301	6.451	4.48	2.79	136-6	15.54-6	0.667	0.0363	0.92	1.20	0.0385	470
480	17.90	1.167-3	0.111	883.4	2795	2.395	6.377	4.53	2.94	129-6	15.88-6	0.660	0.0381	0.89	1.23	0.0362	480
490	21.83	1.184-3	0.0922	929.1	2799	2.479	6.312	4.59	3.10	124-6	16.23-6	0.651	0.0401	0.87	1.25	0.0339	490
500	26.40	1.203-3	0.0766	975.6	2801	2.581	6.233	4.66	3.27	118-6	16.59-6	0.642	0.0423	0.86	1.28	0.0316	500
510	31.66	1.222-3	0.0631	1023	2802	2.673	6.163	4.74	3.47	113-6	16.95-6	0.631	0.0447	0.85	1.31	0.0293	510
520	37.70	1.244-3	0.0525	1071	2801	2.765	6.093	4.84	3.70	108-6	17.33-6	0.621	0.0475	0.84	1.35	0.0269	520
530	44.58	1.268-3	0.0445	1119	2798	2.856	6.023	4.95	3.96	104-6	17.73-6	0.608	0.0506	0.85	1.39	0.0245	530
540	52.38	1.294-3	0.0375	1170	2792	2.948	5.953	5.08	4.27	101-6	18.1-6	0.594	0.0540	0.86	1.43	0.0221	540
550	61.19	1.323-3	0.0317	1220	2784	3.039	5.882	5.24	4.64	97-6	18.6-6	0.580	0.0583	0.87	1.47	0.0197	550
560	71.08	1.355-3	0.0269	1273	2772	3.132	5.808	5.43	5.09	94-6	19.1-6	0.563	0.0637	0.90	1.52	0.0173	560
570	82.16	1.392-3	0.0228	1328	2757	3.225	5.733	5.68	5.67	91-6	19.7-6	0.548	0.0698	0.94	1.59	0.0150	570
580	94.51	1.433-3	0.0193	1384	2737	3.321	5.654	6.00	6.40	88-6	20.4-6	0.528	0.0767	0.99	1.68	0.0128	580
590	108.3	1.482-3	0.0163	1443	2717	3.419	5.569	6.41	7.35	84-6	21.5-6	0.513	0.0841	1.05	1.84	0.0105	590
600	123.5	1.541-3	0.0137	1506	2682	3.520	5.480	7.00	8.75	81-6	22.7-6	0.497	0.0929	1.14	2.15	0.0084	600
610	137.3	1.612-3	0.0115	1573	2641	3.627	5.318	7.85	11.1	77-6	24.1-6	0.467	0.103	1.30	2.60	0.0063	610
620	159.1	1.705-3	0.0094	1647	2588	3.741	5.259	9.35	15.4	72-6	25.9-6	0.444	0.114	1.52	3.46	0.0045	620
625	169.1	1.778-3	0.0085	1697	2555	3.805	5.191	10.6	18.3	70-6	27.0-6	0.430	0.121	1.65	4.20	0.0035	625
630	179.7	1.856-3	0.0075	1734	2515	3.875	5.115	12.6	22.1	67-6	28.0-6	0.412	0.130	2.0	4.8	0.0026	630
635	190.9	1.935-3	0.0066	1783	2466	3.950	5.025	16.4	27.6	64-6	30.0-6	0.392	0.141	2.7	6.0	0.0015	635
640	202.7	2.075-3	0.0057	1841	2401	4.037	4.912	26	42	59-6	32.0-6	0.367	0.155	4.2	9.6	0.0008	640
645	215.2	2.351-3	0.0045	1931	2292	4.223	4.732	90	42	54-6	37.0-6	0.331	0.178	12	26	0.0001	645
647.3†	221.2	3.170-3	0.0032	2107	2107	4.443	4.443	∞	∞	45-6	45.0-6	0.238	0.238	∞	∞	0.0000	647.3†

*1 bar = 10⁵ N/m².

†Above the solid line, the condensed phase is solid; below it, liquid.

‡Critical temperature.

NOTE: The notations 6.30-11, 1.073-3, 9.55-9, etc. signify 6.30 × 10⁻¹¹, 1.073 × 10⁻³, 955 × 10⁹, etc.

Tables 2-351 and 2-352 are provided for general use. Tables to higher precision are available over certain ranges and for various properties. The most current internationally accepted tables are found in Haar, L., J. S. Gallagher, and G. S. Kell, *NBS/NRC Steam Tables*, Hemisphere, Washington, DC, 1984 (320 pp.). These do not tabulate certain properties at saturation states. A revised release on the IAPWS Skeleton Tables 1985 for the thermodynamic properties of ordinary water substance, Sept. 1993 (15 pp.), is apparently the latest international publication. In *J. Phys. Chem. Ref. Data* **17**, 4 (1988): 1439-1540, H. Sato, M. Uematsu, and others review existing steam tables and present the 1985 formulation of skeleton tables. Property codes and programs include Cheng, S. C. and C. Nguyen, *Modeling and Simulation on Microcomputers* 1989 (R. W. Allen, ed.), S.C.S. Intl., San Diego, 1989 (pp. 138-141); Garland, W. J. and B. J. Hand, *Nucl. Engng. & Des.*, **113**, (1989): 21-34; Dickey, D. S., *Chem. Eng.* **98**, 9 (1991): 207-8 and **98**, 11: 235-6; Munceer, T. and S. M. Scott, *Proc. Inst. Mech. Eng.*, **205**, (1991): 25-29; and *Energy Consn. Mgmt.*, **31**, 4 (1991): 315-325. Useful pictorial representations of 20 properties as a function of both temperature (to 800°C) and pressure (to 1000 bar) are given by Griggall, U., J. Bach, et al., *Warmer- u. Stoff.*, **1** (1968): 202-213. Property equations for the saturated liquid for ice from 0 K. Ice and snow properties are reviewed by Fukusako, S., *Int. J. Thermophys.*, **11**, 2 (1990): 353-372. See also Wagner, W., A. Saul, et al., *J. Phys. Chem. Ref. Data*, **23**, 3 (1994): 515-525, and Table 2-358.