

ENTHALPY OF VAPORIZATION

The molar enthalpy (heat) of vaporization $\Delta_{\text{vap}}H$, which is defined as the enthalpy change in the conversion of one mole of liquid to gas at constant temperature, is tabulated here for approximately 850 inorganic and organic compounds. Values are given, when available, both at the normal boiling point t_b , referred to a pressure of 101.325 kPa (760 mmHg), and at 25°C. Substances are listed by molecular formula in the modified Hill order (see Preface).

The values in this table were measured either by calorimetric techniques or by application of the Claperyon equation to the variation of vapor pressure with temperature. See Reference 1 for a discussion of the accuracy of different experimental techniques and for methods of estimating enthalpy of vaporization at other temperatures.

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Mol. Form.	Name	$t_b/^\circ\text{C}$	$\Delta_{\text{vap}}H(t_b)$ kJ/mol	$\Delta_{\text{vap}}H(25^\circ\text{C})$ kJ/mol
AgBr	Silver(I) bromide	1502	198	
AgCl	Silver(I) chloride	1547	199	
AgI	Silver(I) iodide	1506	143.9	
Al	Aluminum	2519	294	
AlB ₃ H ₁₂	Aluminum borohydride	44.5	30	
AlBr ₃	Aluminum tribromide	255	23.5	
AlI ₃	Aluminum triiodide	382	32.2	
Ar	Argon	-185.85	6.43	
AsBr ₃	Arsenic(III) bromide	221	41.8	
AsCl ₃	Arsenic(III) chloride	130	35.01	
AsF ₃	Arsenic(III) fluoride	57.8	29.7	
AsF ₅	Arsenic(V) fluoride	-52.8	20.8	
AsH ₃	Arsine	-62.5	16.69	
AsI ₃	Arsenic(III) iodide	424	59.3	
Au	Gold	2856	324	
B	Boron	4000	480	
BBr ₃	Boron tribromide	91	30.5	
BCl ₃	Boron trichloride	12.65	23.77	23.1
BF ₃	Boron trifluoride	-101	19.33	
BI ₃	Boron triiodide	210	40.5	
B ₂ F ₄	Tetrafluorodiborane	-34	28	
B ₂ H ₆	Diborane	-92.4	14.28	
B ₄ H ₁₀	Tetraborane	18	27.1	
B ₅ H ₁₁	Pentaborane(11)	63	31.8	
Ba	Barium	1897	140	
BeCl ₂	Beryllium chloride	482	105	
BeI ₂	Beryllium iodide	487	70.5	
Bi	Bismuth	1564	151	
BiBr ₃	Bismuth tribromide	453	75.4	
BiCl ₃	Bismuth trichloride	447	72.61	
BrF	Bromine fluoride	20	25.1	
BrF ₃	Bromine trifluoride	125.8	47.57	

ENTHALPY OF VAPORIZATION (continued)

Mol. Form.	Name	$t_b/^{\circ}\text{C}$	$\Delta_{\text{vap}}H(t_b)$ kJ/mol	$\Delta_{\text{vap}}H(25^{\circ}\text{C})$ kJ/mol
BrF ₅	Bromine pentafluoride	40.76	30.6	
BrH	Hydrogen bromide	-66.38		12.69
BrH ₃ Si	Bromosilane	1.9	24.4	
BrIn	Indium(I) bromide	656	92	
BrTl	Thallium(I) bromide	819	99.56	
Br ₂	Bromine	58.8	29.96	30.91
Br ₂ Cd	Cadmium bromide	844	115	
Br ₂ H ₂ Si	Dibromosilane	66	31	
Br ₂ Hg	Mercury(II) bromide	322	58.89	
Br ₂ Pb	Lead(II) bromide	892	133	
Br ₂ Sn	Tin(II) bromide	639	102	
Br ₂ Zn	Zinc bromide	697	118	
Br ₃ Ga	Gallium(III) bromide	279	38.9	
Br ₃ HSi	Tribromosilane	109	34.8	
Br ₃ OP	Phosphorus(V) oxybromide	191.7	38	
Br ₃ P	Phosphorus(III) bromide	172.95	38.8	
Br ₃ Sb	Antimony(III) bromide	280	59	
Br ₄ Ge	Germanium(IV) bromide	186.35	41.4	
Br ₄ Si	Tetrabromosilane	154	37.9	
Br ₄ Sn	Tin(IV) bromide	205	43.5	
Br ₄ Ti	Titanium(IV) bromide	230	44.37	
Br ₅ Ta	Tantalum(V) bromide	349	62.3	
Cd	Cadmium	767	99.87	
CdCl ₂	Cadmium chloride	960	124.3	
CdF ₂	Cadmium fluoride	1748	214	
CdI ₂	Cadmium iodide	742	115	
ClF	Chlorine fluoride	-101.1	24	
ClFO ₃	Perchloryl fluoride	-46.75	19.33	
ClF ₂ P	Phosphorus(III) chloride difluoride	-47.25	17.6	
ClF ₃	Chlorine trifluoride	11.75	27.53	
ClF ₃ Si	Chlorotrifluorosilane	-70.0	18.7	
ClH	Hydrogen chloride	-85	16.15	9.08
ClH ₃ Si	Chlorosilane	-30.4	21	
ClNO	Nitrosyl chloride	-5.5	25.78	
ClNO ₂	Nitryl chloride	-15	25.7	
ClO ₂	Chlorine dioxide	11	30	
ClTI	Thallium(I) chloride	720	102.2	
Cl ₂	Chlorine	-34.04	20.41	17.65
Cl ₂ Cr	Chromium(II) chloride	1300	197	
Cl ₂ CrO ₂	Chromyl chloride	117	35.1	
Cl ₂ FP	Phosphorus(III) dichloride fluoride	14	24.9	
Cl ₂ F ₂ Si	Dichlorodifluorosilane	-32	21.2	
Cl ₂ H ₂ Si	Dichlorosilane	8.3	25	24.2
Cl ₂ Hg	Mercury(II) chloride	304	58.9	
Cl ₂ O	Chlorine monoxide	2.2	25.9	
Cl ₂ OS	Thionyl chloride	75.6	31.7	31
Cl ₂ O ₂ S	Sulfuryl chloride	69.4	31.4	30.1
Cl ₂ Pb	Lead(II) chloride	951	127	
Cl ₂ Sn	Tin(II) chloride	623	86.8	
Cl ₂ Ti	Titanium(II) chloride	1500	232	
Cl ₂ Zn	Zinc chloride	732	126	
Cl ₃ Ga	Gallium(III) chloride	201	23.9	
Cl ₃ HSi	Trichlorosilane	33		25.7
Cl ₃ OP	Phosphorus(V) oxychloride	105.5	34.35	38.6
Cl ₃ OV	Vanadyl trichloride	127	36.78	
Cl ₃ P	Phosphorus(III) chloride	75.95	30.5	32.1
Cl ₃ Sb	Antimony(III) chloride	220.3	45.19	
Cl ₃ Ti	Titanium(III) chloride	960	124	

ENTHALPY OF VAPORIZATION (continued)

Mol. Form.	Name	$t_b/^{\circ}\text{C}$	$\Delta_{\text{vap}}H(t_b)$ kJ/mol	$\Delta_{\text{vap}}H(25^{\circ}\text{C})$ kJ/mol
Cl ₄ Ge	Germanium(IV) chloride	86.55	27.9	
Cl ₄ OW	Tungsten(VI) oxytetrachloride	227.55	67.8	
Cl ₄ Si	Tetrachlorosilane	57.65	28.7	29.7
Cl ₄ Sn	Tin(IV) chloride	114.15	34.9	
Cl ₄ Te	Tellurium tetrachloride	387	77	
Cl ₄ Th	Thorium(IV) chloride	921	146.4	
Cl ₄ Ti	Titanium(IV) chloride	136.45	36.2	
Cl ₄ V	Vanadium(IV) chloride	148	41.4	42.5
Cl ₅ Mo	Molybdenum(V) chloride	268	62.8	
Cl ₅ Nb	Niobium(V) chloride	254.0	52.7	
Cl ₅ Ta	Tantalum(V) chloride	239.35	54.8	
Cl ₆ W	Tungsten(VI) chloride	346.75	52.7	
FH ₃ Si	Fluorosilane	-98.6	18.8	
FLi	Lithium fluoride	1673	147	
FNO	Nitrosyl fluoride	-59.9	19.28	
FNO ₂	Nitryl fluoride	-72.4	18.05	
FNS	Thionitrosyl fluoride (NSF)	4.8	22.2	
F ₂	Fluorine	-188.12	6.62	
F ₂ H ₂ Si	Difluorosilane	-77.8	16.3	
F ₂ O	Fluorine monoxide	-144.75	11.09	
F ₂ OS	Thionyl fluoride	-43.8	21.8	
F ₂ O ₂	Fluorine dioxide	-57	19.1	
F ₂ Pb	Lead(II) fluoride	1293	160.4	
F ₂ Zn	Zinc fluoride	1500	190.1	
F ₃ HSi	Trifluorosilane	-95	16.2	
F ₃ N	Nitrogen trifluoride	-128.75	11.56	
F ₃ O ₂ Re	Rhenium(VII) dioxytrifluoride	185.4	65.7	
F ₃ P	Phosphorus(III) fluoride	-101.5	16.5	
F ₃ PS	Phosphorus(V) sulfide trifluoride	-52.25	19.6	
F ₄ MoO	Molybdenum(VI) oxytetrafluoride	186.0	50.6	
F ₄ N ₂	Tetrafluorohydrazine	-74	13.27	
F ₄ ORe	Rhenium(VI) oxytetrafluoride	171.7	61.0	
F ₄ OW	Tungsten(VI) oxytetrafluoride	185.9	59.5	
F ₄ S	Sulfur tetrafluoride	-40.45	26.44	
F ₄ Se	Selenium tetrafluoride	106	47.2	
F ₄ Th	Thorium(IV) fluoride	1680	258	
F ₅ I	Iodine pentafluoride	100.5	41.3	
F ₅ Mo	Molybdenum(V) fluoride	213.6	51.8	
F ₅ Nb	Niobium(V) fluoride	229	52.3	
F ₅ Os	Osmium(V) fluoride	225.9	65.6	
F ₅ P	Phosphorus(V) fluoride	-84.6	17.2	
F ₅ Re	Rhenium(V) fluoride	221.3	58.1	
F ₅ Ta	Tantalum(V) fluoride	229.2	56.9	
F ₅ V	Vanadium(V) fluoride	48.3	44.52	
F ₆ Ir	Iridium(VI) fluoride	53.6	30.9	
F ₆ Mo	Molybdenum(VI) fluoride	34.0	29.0	
F ₆ Os	Osmium(VI) fluoride	47.5	28.1	
F ₆ Re	Rhenium(VI) fluoride	33.8	28.7	
F ₆ S	Sulfur hexafluoride			8.99
F ₆ W	Tungsten(VI) fluoride	17.1	26.5	
Ga	Gallium	2204	254	
GaI ₃	Gallium(III) iodide	340	56.5	
Ge	Germanium	2833	334	
GeH ₄	Germane	-88.1	14.06	
Ge ₂ H ₆	Digermane	30.8	25.1	
Ge ₃ H ₈	Trigermane	110.5	32.2	
HI	Hydrogen iodide	-35.55	19.76	17.36
HLiO	Lithium hydroxide	1626	188	

ENTHALPY OF VAPORIZATION (continued)

Mol. Form.	Name	$t_b/^{\circ}\text{C}$	$\Delta_{\text{vap}}H(t_b)$ kJ/mol	$\Delta_{\text{vap}}H(25^{\circ}\text{C})$ kJ/mol
HNO ₃	Nitric acid	83		39.1
HN ₃	Hydrazoic acid	35.7	30.5	
HNaO	Sodium hydroxide	1388	175	
H ₂	Hydrogen	-252.87	0.90	
H ₂ O	Water	100.0	40.65	43.98
H ₂ O ₂	Hydrogen peroxide	150.2		51.6
H ₂ S	Hydrogen sulfide	-59.55	18.67	14.08
H ₂ S ₂	Hydrogen disulfide	70.7		33.78
H ₂ Se	Hydrogen selenide	-41.25	19.7	
H ₂ Te	Hydrogen telluride	-2	19.2	
H ₃ N	Ammonia	-33.33	23.33	19.86
H ₃ P	Phosphine	-87.75	14.6	
H ₃ Sb	Stibine	-17	21.3	
H ₄ N ₂	Hydrazine	113.55	41.8	44.7
H ₄ P ₂	Diphosphine	63.5	28.8	
H ₄ Si	Silane	-111.9	12.1	
H ₄ Sn	Stannane	-51.8	19.05	
H ₆ Si ₂	Disilane	-14.3	21.2	
H ₈ Si ₃	Trisilane	52.9	28.5	
He	Helium	-268.93	0.08	
Hg	Mercury	356.73	59.11	
HgI ₂	Mercury(II) iodide	354	59.2	
IIn	Indium(I) iodide	712	90.8	
ITl	Thallium(I) iodide	824	104.7	
I ₂	Iodine	184.4	41.57	
I ₂ Pb	Lead(II) iodide	872	104	
I ₂ Sn	Tin(II) iodide	714	105	
I ₃ P	Phosphorus(III) iodide	227	43.9	
I ₃ Sb	Antimony(III) iodide	401	68.6	
I ₄ Si	Tetraiodosilane	287.35	50.2	
I ₄ Sn	Tin(IV) iodide	364.35	56.9	
I ₄ Ti	Titanium(IV) iodide	377	58.4	
Kr	Krypton	-153.22	9.08	
MoO ₃	Molybdenum(VI) oxide	1155	138	
NO	Nitric oxide	-151.74	13.83	
N ₂	Nitrogen	-195.79	5.57	
N ₂ O	Nitrous oxide	-88.48	16.53	
N ₂ O ₄	Nitrogen tetroxide	21.15	38.12	
Ne	Neon	-246.08	1.71	
O ₂	Oxygen	-182.95	6.82	
O ₂ S	Sulfur dioxide	-10.05	24.94	22.92
O ₃ S	Sulfur trioxide	45	40.69	43.14
P	Phosphorus	280.5	12.4	14.2
Pb	Lead	1749	179.5	
S	Sulfur	444.60	45	
STl ₂	Thallium(I) sulfide	1367	154	
Se	Selenium	685	95.48	
Te	Tellurium	988	114.1	
Xe	Xenon	-108.11	12.57	
CClF ₃	Chlorotrifluoromethane	-81.4	15.8	
CCl ₂ F ₂	Dichlorodifluoromethane	-29.8	20.1	
CCl ₃ F	Trichlorofluoromethane	23.7	25.1	
CCl ₄	Tetrachloromethane	76.8	29.82	32.43
CHBr ₃	Tribromomethane	149.1	39.66	46.05
CHClF ₂	Chlorodifluoromethane	-40.7	20.2	
CHCl ₂ F	Dichlorofluoromethane	8.9	25.2	
CHCl ₃	Trichloromethane	61.17	29.24	31.28
CH ₂ BrCl	Bromochloromethane	68.0	30.0	

ENTHALPY OF VAPORIZATION (continued)

Mol. Form.	Name	$t_b/^\circ\text{C}$	$\Delta_{\text{vap}}H(t_b)$ kJ/mol	$\Delta_{\text{vap}}H(25^\circ\text{C})$ kJ/mol
CH ₂ Br ₂	Dibromomethane	97	32.92	36.97
CH ₂ Cl ₂	Dichloromethane	40	28.06	28.82
CH ₂ I ₂	Diiodomethane	182	42.5	
CH ₂ O ₂	Formic acid	101	22.69	20.10
CH ₃ Br	Bromomethane	3.5	23.91	22.81
CH ₃ Cl	Chloromethane	-24.09	21.40	18.92
CH ₃ I	Iodomethane	42.55	27.34	27.97
CH ₃ NO	Formamide	220		60.15
CH ₃ NO ₂	Nitromethane	101.19	33.99	38.27
CH ₄	Methane	-161.48	8.19	
CH ₄ O	Methanol	64.6	35.21	37.43
CH ₅ N	Methylamine	-6.32	25.60	23.37
CH ₆ N ₂	Methylhydrazine	87.5	36.12	40.37
CN ₄ O ₈	Tetranitromethane	126.1	40.74	49.93
CO	Carbon monoxide	-191.5	6.04	
CS ₂	Carbon disulfide	46	26.74	27.51
C ₂ Br ₂ ClF ₃	1,2-Dibromo-1-chloro-1,2,2-trifluoroethane	93	31.17	35.04
C ₂ Br ₂ F ₄	1,2-Dibromotetrafluoroethane	47.35	27.03	28.39
C ₂ ClF ₅	Chloropentafluoroethane	-37.95	19.41	
C ₂ Cl ₂ F ₄	1,2-Dichlorotetrafluoroethane	3.8	23.3	
C ₂ Cl ₃ F ₃	1,1,1-Trichlorotrifluoroethane	46.1	26.85	28.08
C ₂ Cl ₃ F ₃	1,1,2-Trichloro-1,2,2-trifluoroethane	47.7	27.04	28.40
C ₂ Cl ₄	Tetrachloroethylene	121.3	34.68	39.68
C ₂ F ₆	Hexafluoroethane	-78.1	16.15	
C ₂ HBrClF ₃	2-Bromo-2-chloro-1,1,1-trifluoroethane	50.2	28.08	29.61
C ₂ HCl ₃	Trichloroethylene	87.21	31.40	34.54
C ₂ HCl ₅	Pentachloroethane	159.8	36.9	
C ₂ HF ₃ O ₂	Trifluoroacetic acid	73	33.3	
C ₂ H ₂ Br ₄	1,1,2,2-Tetrabromoethane	243.5	48.7	
C ₂ H ₂ Cl ₂	1,1-Dichloroethylene	31.6	26.14	26.48
C ₂ H ₂ Cl ₂	<i>cis</i> -1,2-Dichloroethylene	60.1	30.2	
C ₂ H ₂ Cl ₂	<i>trans</i> -1,2-Dichloroethylene	48.7	28.9	
C ₂ H ₂ Cl ₄	1,1,2,2-Tetrachloroethane	146.5	37.64	45.71
C ₂ H ₃ Br	Bromoethylene	15.8	23.4	
C ₂ H ₃ Cl	Chloroethylene	-13.3	20.8	
C ₂ H ₃ Cl ₂ F	1,1-Dichloro-1-fluoroethane	32.0	26.06	26.48
C ₂ H ₃ Cl ₃	1,1,1-Trichloroethane	74.09	29.86	32.50
C ₂ H ₃ Cl ₃	1,1,2-Trichloroethane	113.8	34.82	40.24
C ₂ H ₃ F ₃	1,1,1-Trifluoroethane	-47.25	18.99	
C ₂ H ₃ N	Acetonitrile	81.65	29.75	32.94
C ₂ H ₄	Ethylene	-103.77	13.53	
C ₂ H ₄ Br ₂	1,2-Dibromoethane	131.6	34.77	41.73
C ₂ H ₄ Cl ₂	1,1-Dichloroethane	57.4	28.85	30.62
C ₂ H ₄ Cl ₂	1,2-Dichloroethane	83.5	31.98	35.16
C ₂ H ₄ F ₂	1,1-Difluoroethane	-24.95	21.56	19.08
C ₂ H ₄ O	Acetaldehyde	20.1	25.76	25.47
C ₂ H ₄ O	Ethylene oxide	10.6	25.54	24.75
C ₂ H ₄ O ₂	Acetic acid	117.9	23.70	23.36
C ₂ H ₄ O ₂	Methyl formate	31.7	27.92	28.35
C ₂ H ₅ Br	Bromoethane	38.5	27.04	28.03
C ₂ H ₅ Cl	Chloroethane	12.3	24.65	
C ₂ H ₅ ClO	2-Chloroethanol	128.6	41.4	
C ₂ H ₅ I	Iodoethane	72.5	29.44	31.93
C ₂ H ₅ NO	<i>N</i> -Methylformamide	199.51		56.19
C ₂ H ₅ NO ₂	Nitroethane	114.0	38.0	
C ₂ H ₆	Ethane	-88.6	14.69	5.16
C ₂ H ₆ O	Ethanol	78.29	38.56	42.32
C ₂ H ₆ O	Dimethyl ether	-24.8	21.51	18.51

ENTHALPY OF VAPORIZATION (continued)

Mol. Form.	Name	$t_b/^{\circ}\text{C}$	$\Delta_{\text{vap}}H(t_b)$ kJ/mol	$\Delta_{\text{vap}}H(25^{\circ}\text{C})$ kJ/mol
C ₂ H ₆ OS	Dimethyl sulfoxide	189	43.1	
C ₂ H ₆ O ₂	Ethylene glycol	197.3	50.5	
C ₂ H ₆ S	Ethanethiol	35.1	26.79	27.30
C ₂ H ₆ S	Dimethyl sulfide	37.33	27.0	27.65
C ₂ H ₆ S ₂	1,2-Ethanedithiol	146.1	37.93	44.68
C ₂ H ₆ S ₂	Dimethyl disulfide	109.8	33.78	37.86
C ₂ H ₇ N	Dimethylamine	6.88	26.40	25.05
C ₂ H ₇ NO	Ethanolamine	171	49.83	
C ₂ H ₈ N ₂	1,2-Ethanediamine	117	37.98	44.98
C ₂ H ₈ N ₂	1,1-Dimethylhydrazine	63.9	32.55	35.0
C ₂ N ₂	Cyanogen	-21.1	23.33	19.75
C ₃ Cl ₂ F ₆	1,2-Dichlorohexafluoropropane	34.1	26.28	26.93
C ₃ H ₃ Cl ₃ O ₂	Methyl trichloroacetate	153.8		48.33
C ₃ H ₃ N	Acrylonitrile	77.3	32.6	
C ₃ H ₄ Cl ₂ O ₂	Methyl dichloroacetate	142.9	39.28	47.72
C ₃ H ₄ O	Acrolein	52.6	28.3	
C ₃ H ₄ O ₂	2-Oxetanone	162		47.03
C ₃ H ₅ Br	3-Bromopropene	70.1	30.24	32.73
C ₃ H ₅ Cl	3-Chloropropene	45.1	29.0	
C ₃ H ₅ ClO ₂	Methyl chloroacetate	129.5	39.23	46.73
C ₃ H ₅ Cl ₃	1,2,3-Trichloropropane	157	37.1	
C ₃ H ₅ N	Propanenitrile	97.14	31.81	36.03
C ₃ H ₆	Propene	-47.69	18.42	14.24
C ₃ H ₆	Cyclopropane	-32.81	20.05	16.93
C ₃ H ₆ Br ₂	1,2-Dibromopropane	141.9	35.61	41.67
C ₃ H ₆ Br ₂	1,3-Dibromopropane	167.3		47.45
C ₃ H ₆ Cl ₂	1,3-Dichloropropane	120.9	35.18	40.75
C ₃ H ₆ O	Allyl alcohol	97.0	40.0	
C ₃ H ₆ O	Propanal	48	28.31	29.62
C ₃ H ₆ O	Acetone	56.05	29.10	30.99
C ₃ H ₆ O	Methyloxirane	35	27.35	27.89
C ₃ H ₆ O	Oxetane	47.6	28.67	29.85
C ₃ H ₆ O ₂	Propanoic acid	141.15		32.14
C ₃ H ₆ O ₂	Ethyl formate	54.4	29.91	31.96
C ₃ H ₆ O ₂	Methyl acetate	56.87	30.32	32.29
C ₃ H ₆ S	Thietane	95	32.32	35.97
C ₃ H ₇ Br	1-Bromopropane	71.1	29.84	32.01
C ₃ H ₇ Br	2-Bromopropane	59.5	28.33	30.17
C ₃ H ₇ Cl	1-Chloropropane	46.5	27.18	28.35
C ₃ H ₇ Cl	2-Chloropropane	35.7	26.30	26.90
C ₃ H ₇ I	1-Iodopropane	102.6	32.08	36.25
C ₃ H ₇ I	2-Iodopropane	89.5	30.68	34.06
C ₃ H ₇ NO	<i>N</i> -Ethylformamide	198		58.44
C ₃ H ₇ NO	<i>N,N</i> -Dimethylformamide	153		46.89
C ₃ H ₇ NO ₂	1-Nitropropane	131.1	38.5	
C ₃ H ₇ NO ₂	2-Nitropropane	120.2	36.8	
C ₃ H ₈	Propane	-42.1	19.04	14.79
C ₃ H ₈ O	1-Propanol	97.2	41.44	47.45
C ₃ H ₈ O	2-Propanol	82.3	39.85	45.39
C ₃ H ₈ O ₂	1,2-Propylene glycol	187.6	52.4	
C ₃ H ₈ O ₂	1,3-Propylene glycol	214.4	57.9	
C ₃ H ₈ O ₂	Ethylene glycol monomethyl ether	124.1	37.54	45.17
C ₃ H ₈ O ₃	Glycerol	290	61.0	
C ₃ H ₈ S	1-Propanethiol	67.8	29.54	31.89
C ₃ H ₈ S	2-Propanethiol	52.6	27.91	29.45
C ₃ H ₈ S	Ethyl methyl sulfide	66.7	29.53	31.85
C ₃ H ₈ S ₂	1,3-Propanedithiol	172.9		49.66
C ₃ H ₉ N	Propylamine	47.22	29.55	31.27

ENTHALPY OF VAPORIZATION (continued)

Mol. Form.	Name	$t_b/^{\circ}\text{C}$	$\Delta_{\text{vap}}H(t_b)$ kJ/mol	$\Delta_{\text{vap}}H(25^{\circ}\text{C})$ kJ/mol
C ₃ H ₉ N	Isopropylamine	31.76	27.83	28.36
C ₃ H ₉ N	Trimethylamine	2.87	22.94	21.66
C ₃ H ₁₀ N ₂	1,3-Propanediamine	139.8	40.85	50.16
C ₄ F ₈	Perfluorocyclobutane	-5.9	23.2	
C ₄ F ₁₀	Perfluorobutane	-1.9	22.9	
C ₄ H ₄ N ₂	Succinonitrile	266	48.5	
C ₄ H ₄ N ₂	Pyrimidine	123.8	43.09	49.79
C ₄ H ₄ N ₂	Pyridazine	208		53.47
C ₄ H ₄ O	Furan	31.5	27.10	27.45
C ₄ H ₄ O ₂	Diketene	126.1	36.80	42.89
C ₄ H ₄ S	Thiophene	84.0	31.48	34.70
C ₄ H ₅ Cl ₃ O ₂	Ethyl trichloroacetate	167.5		50.97
C ₄ H ₅ N	2-Methylacrylonitrile	90.3	31.8	
C ₄ H ₅ N	Pyrrole	129.79	38.75	45.09
C ₄ H ₅ N	Cyclopropanecarbonitrile	135.1	35.55	41.94
C ₄ H ₅ NO ₂	Methyl cyanoacetate	200.5	48.2	
C ₄ H ₅ NS	4-Methylthiazole	133.3	37.58	43.85
C ₄ H ₆	1,2-Butadiene	10.9	24.02	23.21
C ₄ H ₆	1,3-Butadiene	-4.41	22.47	20.86
C ₄ H ₆	1-Butyne	8.08	24.52	23.35
C ₄ H ₆ Cl ₂ O ₂	Ethyl dichloroacetate	155		50.60
C ₄ H ₆ O ₂	Vinyl acetate	72.5	34.6	
C ₄ H ₆ O ₂	Methyl acrylate	80.7	33.1	
C ₄ H ₆ O ₂	γ -Butyrolactone	204	52.2	
C ₄ H ₆ O ₃	Acetic anhydride	139.5	38.2	
C ₄ H ₆ S	2,3-Dihydrothiophene	112.1	33.24	37.74
C ₄ H ₆ S	2,5-Dihydrothiophene	122.4	34.83	39.95
C ₄ H ₇ ClO ₂	Ethyl chloroacetate	144.3	40.43	49.47
C ₄ H ₇ N	Butanenitrile	117.6	33.68	39.33
C ₄ H ₇ N	2-Methylpropanenitrile	103.9	32.39	37.13
C ₄ H ₈	1-Butene	-6.26	22.07	20.22
C ₄ H ₈	<i>cis</i> -2-Butene	3.71	23.34	22.16
C ₄ H ₈	<i>trans</i> -2-Butene	0.88	22.72	21.40
C ₄ H ₈	Cyclobutane	12.6	24.19	23.51
C ₄ H ₈ Br ₂	1,4-Dibromobutane	197		53.09
C ₄ H ₈ Cl ₂	1,2-Dichlorobutane	124.1	33.90	39.58
C ₄ H ₈ Cl ₂	1,4-Dichlorobutane	161		46.36
C ₄ H ₈ Cl ₂ O	Bis(2-chloroethyl) ether	178.5	45.2	
C ₄ H ₈ O	Ethyl vinyl ether	35.5	26.2	
C ₄ H ₈ O	1,2-Epoxybutane	63.4	30.3	
C ₄ H ₈ O	Butanal	74.8	31.5	
C ₄ H ₈ O	2-Butanone	79.59	31.30	34.79
C ₄ H ₈ O	Tetrahydrofuran	65	29.81	31.99
C ₄ H ₈ O ₂	Butanoic acid	163.75		40.45
C ₄ H ₈ O ₂	2-Methylpropanoic acid	154.45		35.30
C ₄ H ₈ O ₂	Propyl formate	80.9	33.61	37.53
C ₄ H ₈ O ₂	Ethyl acetate	77.11	31.94	35.60
C ₄ H ₈ O ₂	Methyl propanoate	79.8	32.24	35.85
C ₄ H ₈ O ₂	1,3-Dioxane	106.1	34.37	39.09
C ₄ H ₈ O ₂	1,4-Dioxane	101.5	34.16	38.60
C ₄ H ₈ S	Tetrahydrothiophene	121.0	34.66	39.43
C ₄ H ₉ Br	1-Bromobutane	101.6	32.51	36.64
C ₄ H ₉ Br	2-Bromobutane	91.3	30.77	34.41
C ₄ H ₉ Br	1-Bromo-2-methylpropane	91.1	31.33	34.82
C ₄ H ₉ Br	2-Bromo-2-methylpropane	73.3	29.23	31.81
C ₄ H ₉ Cl	1-Chlorobutane	78.6	30.39	33.51
C ₄ H ₉ Cl	2-Chlorobutane	68.2	29.17	31.53
C ₄ H ₉ Cl	1-Chloro-2-methylpropane	68.5	29.22	31.67

ENTHALPY OF VAPORIZATION (continued)

Mol. Form.	Name	$t_b/^{\circ}\text{C}$	$\Delta_{\text{vap}}H(t_b)$ kJ/mol	$\Delta_{\text{vap}}H(25^{\circ}\text{C})$ kJ/mol
C ₄ H ₉ Cl	2-Chloro-2-methylpropane	50.9	27.55	28.98
C ₄ H ₉ I	1-Iodobutane	130.6	34.66	40.63
C ₄ H ₉ I	2-Iodobutane	120.1	33.27	38.46
C ₄ H ₉ I	1-Iodo-2-methylpropane	121.1	33.54	38.83
C ₄ H ₉ I	2-Iodo-2-methylpropane	100.1	31.43	35.41
C ₄ H ₉ N	Pyrrolidine	86.56	33.01	37.52
C ₄ H ₉ NO	N-Ethylacetamide	205		64.89
C ₄ H ₉ NO	N,N-Dimethylacetamide	165		50.24
C ₄ H ₉ NO	Morpholine	128	37.1	
C ₄ H ₁₀	Butane	-0.5	22.44	21.02
C ₄ H ₁₀	Isobutane	-11.73	21.30	19.23
C ₄ H ₁₀ O	1-Butanol	117.73	43.29	52.35
C ₄ H ₁₀ O	2-Butanol	99.51	40.75	49.72
C ₄ H ₁₀ O	2-Methyl-1-propanol	107.89	41.82	50.82
C ₄ H ₁₀ O	2-Methyl-2-propanol	82.4	39.07	46.69
C ₄ H ₁₀ O	Diethyl ether	34.5	26.52	27.10
C ₄ H ₁₀ O	Methyl propyl ether	39.1	26.75	27.60
C ₄ H ₁₀ O	Isopropyl methyl ether	30.77	26.05	26.41
C ₄ H ₁₀ O ₂	1,2-Butanediol	190.5	52.84	71.55
C ₄ H ₁₀ O ₂	1,3-Butanediol	207.5	54.31	74.46
C ₄ H ₁₀ O ₂	Ethylene glycol monoethyl ether	135	39.22	48.21
C ₄ H ₁₀ O ₂	Ethylene glycol dimethyl ether	85	32.42	36.39
C ₄ H ₁₀ O ₃	Diethylene glycol	245.8	52.3	
C ₄ H ₁₀ S	1-Butanethiol	98.5	32.23	36.63
C ₄ H ₁₀ S	2-Butanethiol	85	30.59	33.99
C ₄ H ₁₀ S	2-Methyl-1-propanethiol	88.5	31.01	34.63
C ₄ H ₁₀ S	2-Methyl-2-propanethiol	64.3	28.45	30.78
C ₄ H ₁₀ S	Diethyl sulfide	92.1	31.77	35.80
C ₄ H ₁₀ S	Methyl propyl sulfide	95.6	32.08	36.24
C ₄ H ₁₀ S	Isopropyl methyl sulfide	84.8	30.71	34.15
C ₄ H ₁₀ S ₂	1,4-Butanedithiol	195.5		55.10
C ₄ H ₁₀ S ₂	Diethyl disulfide	154.1	37.58	45.18
C ₄ H ₁₁ N	Butylamine	77.00	31.81	35.72
C ₄ H ₁₁ N	sec-Butylamine	62.73	29.92	32.85
C ₄ H ₁₁ N	tert-Butylamine	44.04	28.27	29.64
C ₄ H ₁₁ N	Isobutylamine	67.75	30.61	33.85
C ₄ H ₁₁ N	Diethylamine	55.5	29.06	31.31
C ₄ H ₁₁ N	Isopropylmethylamine	50.4	28.71	30.69
C ₄ H ₁₁ NO	2-Amino-2-methyl-1-propanol	165.5	50.6	
C ₄ H ₁₁ NO ₂	Diethanolamine	268.8	65.2	
C ₅ H ₂ F ₆ O ₂	Hexafluoroacetylacetone	54.15	27.05	30.58
C ₅ H ₄ O ₂	Furfural	161.7	43.2	
C ₅ H ₅ N	Pyridine	115.23	35.09	40.21
C ₅ H ₆ O ₂	Furfuryl alcohol	171	53.6	
C ₅ H ₆ S	2-Methylthiophene	112.6	33.90	38.87
C ₅ H ₆ S	3-Methylthiophene	115.5	34.24	39.43
C ₅ H ₇ N	trans-3-Pentenitrile	142.6	37.09	44.77
C ₅ H ₇ N	Cyclobutanecarbonitrile	149.6	36.88	44.34
C ₅ H ₈	Spiropentane	39	26.76	27.49
C ₅ H ₈ O	Cyclopropyl methyl ketone	111.3	34.07	39.41
C ₅ H ₈ O	Cyclopentanone	130.57	36.35	42.72
C ₅ H ₈ O ₂	Methyl cyclopropanecarboxylate	114.9	35.25	41.27
C ₅ H ₈ O ₂	Allyl acetate	103.5	36.3	
C ₅ H ₈ O ₂	Ethyl acrylate	99.4	34.7	
C ₅ H ₈ O ₂	Methyl methacrylate	100.5	36.0	
C ₅ H ₈ O ₂	2,4-Pentanedione	138	34.30	41.77
C ₅ H ₉ N	Pentanenitrile	141.3	36.09	43.60
C ₅ H ₉ N	3-Methylbutanenitrile	127.5	35.10	41.64

ENTHALPY OF VAPORIZATION (continued)

Mol. Form.	Name	$t_b/^{\circ}\text{C}$	$\Delta_{\text{vap}}H(t_b)$ kJ/mol	$\Delta_{\text{vap}}H(25^{\circ}\text{C})$ kJ/mol
C ₅ H ₉ N	2,2-Dimethylpropanenitrile	106.1	32.40	37.35
C ₅ H ₁₀	1-Pentene	29.96	25.20	25.47
C ₅ H ₁₀	<i>cis</i> -2-Pentene	36.93		26.86
C ₅ H ₁₀	<i>trans</i> -2-Pentene	36.34		26.76
C ₅ H ₁₀	2-Methyl-1-butene	31.2	25.50	25.92
C ₅ H ₁₀	3-Methyl-1-butene	20.1		23.77
C ₅ H ₁₀	2-Methyl-2-butene	38.56	26.31	27.06
C ₅ H ₁₀	Cyclopentane	49.3	27.30	28.52
C ₅ H ₁₀ Cl ₂	1,2-Dichloropentane	148.3	36.45	43.89
C ₅ H ₁₀ Cl ₂	1,5-Dichloropentane	179		50.71
C ₅ H ₁₀ O	Cyclopentanol	140.42		57.05
C ₅ H ₁₀ O	2-Pentanone	102.26	33.44	38.40
C ₅ H ₁₀ O	3-Pentanone	101.96	33.45	38.52
C ₅ H ₁₀ O	3-Methyl-2-butanone	94.33	32.35	36.78
C ₅ H ₁₀ O	3,3-Dimethyloxetane	80.6	30.85	33.94
C ₅ H ₁₀ O	Tetrahydropyran	88	31.17	34.58
C ₅ H ₁₀ O ₂	Pentanoic acid	186.1	44.1	
C ₅ H ₁₀ O ₂	2-Methylbutanoic acid	177		46.91
C ₅ H ₁₀ O ₂	Butyl formate	106.1	36.58	41.11
C ₅ H ₁₀ O ₂	Isobutyl formate	98.2	33.6	
C ₅ H ₁₀ O ₂	Propyl acetate	101.54	33.92	39.72
C ₅ H ₁₀ O ₂	Isopropyl acetate	88.6	32.93	37.20
C ₅ H ₁₀ O ₂	Ethyl propanoate	99.1	33.88	39.21
C ₅ H ₁₀ O ₂	Methyl butanoate	102.8	33.79	39.28
C ₅ H ₁₀ O ₂	Methyl isobutanoate	92.5	32.61	37.32
C ₅ H ₁₀ O ₂	Tetrahydrofurfuryl alcohol	178	45.2	
C ₅ H ₁₀ O ₃	Diethyl carbonate	126		43.60
C ₅ H ₁₀ O ₃	Ethylene glycol monomethyl ether acetate	143	43.9	
C ₅ H ₁₀ S	Thiacyclohexane	141.8	35.96	42.58
C ₅ H ₁₀ S	Cyclopentanethiol	132.1	35.32	41.42
C ₅ H ₁₁ Br	1-Bromopentane	129.8	35.01	41.28
C ₅ H ₁₁ Cl	1-Chloropentane	107.8	33.15	38.24
C ₅ H ₁₁ Cl	2-Chloropentane	97.0	31.79	36.03
C ₅ H ₁₁ Cl	1-Chloro-3-methylbutane	98.9	32.02	36.24
C ₅ H ₁₁ I	1-Iodopentane	155		45.27
C ₅ H ₁₁ N	Piperidine	106.22		39.29
C ₅ H ₁₂	Pentane	36.06	25.79	26.43
C ₅ H ₁₂	Isopentane	27.88	24.69	24.85
C ₅ H ₁₂	Neopentane	9.48	22.74	21.84
C ₅ H ₁₂ O	1-Pentanol	137.98	44.36	57.02
C ₅ H ₁₂ O	2-Pentanol	119.3	41.40	54.21
C ₅ H ₁₂ O	3-Pentanol	116.25		54.0
C ₅ H ₁₂ O	2-Methyl-1-butanol	128		55.16
C ₅ H ₁₂ O	3-Methyl-1-butanol	131.1	44.07	55.61
C ₅ H ₁₂ O	2-Methyl-2-butanol	102.4	39.04	50.10
C ₅ H ₁₂ O	3-Methyl-2-butanol	112.9		53.0
C ₅ H ₁₂ O	Butyl methyl ether	70.16	29.55	32.37
C ₅ H ₁₂ O	<i>sec</i> -Butyl methyl ether	59.1	28.09	30.23
C ₅ H ₁₂ O	Methyl <i>tert</i> -butyl ether	55.2	27.94	29.82
C ₅ H ₁₂ O	Isobutyl methyl ether	58.6	28.02	30.13
C ₅ H ₁₂ O	Ethyl propyl ether	63.21	28.94	31.43
C ₅ H ₁₂ O	Ethyl isopropyl ether	54.1	28.21	30.08
C ₅ H ₁₂ O ₂	1-Ethoxy-2-methoxyethane	102.1	34.33	39.83
C ₅ H ₁₂ O ₂	1,5-Pentanediol	239	60.7	
C ₅ H ₁₂ O ₂	Ethylene glycol monopropyl ether	149.8	41.40	52.12
C ₅ H ₁₂ O ₂	Diethoxymethane	88	31.33	35.65
C ₅ H ₁₂ O ₃	Diethylene glycol monomethyl ether	193	46.6	
C ₅ H ₁₂ S	1-Pentanethiol	126.6	34.88	41.24

ENTHALPY OF VAPORIZATION (continued)

Mol. Form.	Name	$t_b/^{\circ}\text{C}$	$\Delta_{\text{vap}}H(t_b)$ kJ/mol	$\Delta_{\text{vap}}H(25^{\circ}\text{C})$ kJ/mol
C ₅ H ₁₂ S	2-Methyl-1-butanethiol	119.1	33.79	39.45
C ₅ H ₁₂ S	2-Methyl-2-butanethiol	99.1	31.37	35.67
C ₅ H ₁₂ S	Butyl methyl sulfide	123.5	34.47	40.46
C ₅ H ₁₂ S	<i>tert</i> -Butyl methyl sulfide	99	31.47	35.84
C ₅ H ₁₂ S	Ethyl propyl sulfide	118.6	34.24	39.97
C ₅ H ₁₂ S	Ethyl isopropyl sulfide	107.5	32.74	37.78
C ₅ H ₁₃ N	Pentylamine	104.3	34.01	40.08
C ₅ H ₁₃ N	Ethylisopropylamine	69.6	29.94	33.13
C ₆ ClF ₅	Chloropentafluorobenzene	117.96	34.76	41.07
C ₆ F ₆	Hexafluorobenzene	80.26	31.66	35.71
C ₆ HF ₅	Pentafluorobenzene	85.74	32.15	36.27
C ₆ H ₄ Cl ₂	<i>o</i> -Dichlorobenzene	180	39.66	50.21
C ₆ H ₄ Cl ₂	<i>m</i> -Dichlorobenzene	173	38.62	48.58
C ₆ H ₄ Cl ₂	<i>p</i> -Dichlorobenzene	174	38.79	49.0
C ₆ H ₄ F ₂	<i>o</i> -Difluorobenzene	94	32.21	36.18
C ₆ H ₄ F ₂	<i>m</i> -Difluorobenzene	82.6	31.10	34.59
C ₆ H ₄ F ₂	<i>p</i> -Difluorobenzene	89	31.77	35.54
C ₆ H ₅ Br	Bromobenzene	156.06		44.54
C ₆ H ₅ Cl	Chlorobenzene	131.72	35.19	40.97
C ₆ H ₅ F	Fluorobenzene	84.73	31.19	34.58
C ₆ H ₅ I	Iodobenzene	188.4	39.5	
C ₆ H ₅ NO ₂	Nitrobenzene	210.8		55.01
C ₆ H ₆	Benzene	80.09	30.72	33.83
C ₆ H ₆ ClN	<i>o</i> -Chloroaniline	208.8	44.4	
C ₆ H ₆ O	Phenol	181.87	45.69	57.82
C ₆ H ₆ S	Benzenethiol	169.1	39.93	47.56
C ₆ H ₇ N	Aniline	184.17	42.44	55.83
C ₆ H ₇ N	2-Methylpyridine	129.38	36.17	42.48
C ₆ H ₇ N	3-Methylpyridine	144.14	37.35	44.44
C ₆ H ₇ N	4-Methylpyridine	145.36	37.51	44.56
C ₆ H ₇ N	1-Cyclopentenecarbonitrile			44.98
C ₆ H ₉ N	Cyclopentanecarbonitrile			43.43
C ₆ H ₉ NO ₃	Triacetamide			60.41
C ₆ H ₁₀	Cyclohexene	82.98	30.46	33.47
C ₆ H ₁₀ O	Cyclohexanone	155.43		45.06
C ₆ H ₁₀ O	Mesityl oxide	130	36.1	
C ₆ H ₁₀ O ₂	Methyl cyclobutanecarboxylate	135.5	37.13	44.72
C ₆ H ₁₀ O ₃	Propanoic anhydride	170	41.7	
C ₆ H ₁₀ O ₄	Diethyl oxalate	185.7	42.0	
C ₆ H ₁₀ O ₄	Ethylene glycol diacetate	190		61.44
C ₆ H ₁₁ N	Hexanenitrile	163.65		47.91
C ₆ H ₁₂	1-Hexene	63.48		30.61
C ₆ H ₁₂	<i>cis</i> -2-Hexene	68.8		32.19
C ₆ H ₁₂	<i>trans</i> -2-Hexene	67.9		31.60
C ₆ H ₁₂	<i>cis</i> -3-Hexene	66.4		31.23
C ₆ H ₁₂	<i>trans</i> -3-Hexene	67.1		31.55
C ₆ H ₁₂	2-Methyl-1-pentene	62.1		30.48
C ₆ H ₁₂	3-Methyl-1-pentene	54.2		28.62
C ₆ H ₁₂	4-Methyl-1-pentene	53.9		28.71
C ₆ H ₁₂	2-Methyl-2-pentene	67.3		31.60
C ₆ H ₁₂	3-Methyl- <i>cis</i> -2-pentene	67.7		32.09
C ₆ H ₁₂	3-Methyl- <i>trans</i> -2-pentene	70.4		31.35
C ₆ H ₁₂	4-Methyl- <i>cis</i> -2-pentene	56.3		29.48
C ₆ H ₁₂	4-Methyl- <i>trans</i> -2-pentene	58.6		29.97
C ₆ H ₁₂	2-Ethyl-1-butene	64.7		31.13
C ₆ H ₁₂	2,3-Dimethyl-1-butene	55.6		29.18
C ₆ H ₁₂	3,3-Dimethyl-1-butene	41.2		26.61
C ₆ H ₁₂	2,3-Dimethyl-2-butene	73.3	29.64	32.51

ENTHALPY OF VAPORIZATION (continued)

Mol. Form.	Name	$t_b/^\circ\text{C}$	$\Delta_{\text{vap}}H(t_b)$ kJ/mol	$\Delta_{\text{vap}}H(25^\circ\text{C})$ kJ/mol
C ₆ H ₁₂	Cyclohexane	80.73	29.97	33.01
C ₆ H ₁₂	Methylcyclopentane	71.8	29.08	31.64
C ₆ H ₁₂	Ethylcyclobutane	70.8	28.67	31.24
C ₆ H ₁₂ Cl ₂	1,2-Dichlorohexane	173		48.16
C ₆ H ₁₂ O	Butyl vinyl ether	94	31.58	36.17
C ₆ H ₁₂ O	2-Hexanone	127.6	36.35	43.14
C ₆ H ₁₂ O	3-Hexanone	123.5	35.36	42.47
C ₆ H ₁₂ O	3-Methyl-2-pentanone	117.5	34.16	40.53
C ₆ H ₁₂ O	4-Methyl-2-pentanone	116.5	34.49	40.61
C ₆ H ₁₂ O	2-Methyl-3-pentanone	113.5	33.84	39.79
C ₆ H ₁₂ O	3,3-Dimethyl-2-butanone	106.1	33.39	37.91
C ₆ H ₁₂ O	Cyclohexanol	160.84		62.01
C ₆ H ₁₂ O ₂	Butyl acetate	126.1	36.28	43.86
C ₆ H ₁₂ O ₂	<i>tert</i> -Butyl acetate	95.1	33.07	38.03
C ₆ H ₁₂ O ₂	Isobutyl acetate	116.5	35.9	
C ₆ H ₁₂ O ₂	Propyl propanoate	122.5	35.54	43.45
C ₆ H ₁₂ O ₂	Ethyl butanoate	121.5	35.47	42.68
C ₆ H ₁₂ O ₂	Ethyl 2-methylpropanoate	110.1	33.67	39.83
C ₆ H ₁₂ O ₂	Methyl pentanoate	127.4	35.36	43.10
C ₆ H ₁₂ O ₂	Methyl 2,2-dimethylpropanoate	101.1	33.42	38.76
C ₆ H ₁₂ O ₃	Ethylene glycol monoethyl ether acetate	156.4	40.76	52.61
C ₆ H ₁₂ S	Cyclohexanethiol	158.9	37.06	44.57
C ₆ H ₁₃ Br	1-Bromohexane	155.3		45.89
C ₆ H ₁₃ Cl	1-Chlorohexane	135	35.67	42.83
C ₆ H ₁₃ I	1-Iodoheptane	181		49.75
C ₆ H ₁₃ N	Cyclohexylamine	134	36.14	43.67
C ₆ H ₁₄	Hexane	68.73	28.85	31.56
C ₆ H ₁₄	2-Methylpentane	60.26	27.79	29.89
C ₆ H ₁₄	3-Methylpentane	63.27	28.06	30.28
C ₆ H ₁₄	2,2-Dimethylbutane	49.73	26.31	27.68
C ₆ H ₁₄	2,3-Dimethylbutane	57.93	27.38	29.12
C ₆ H ₁₄ N ₂	Azopropane	114		39.88
C ₆ H ₁₄ O	1-Hexanol	157.6	44.50	61.61
C ₆ H ₁₄ O	2-Hexanol	140	41.01	58.46
C ₆ H ₁₄ O	2-Methyl-1-pentanol	149	50.2	
C ₆ H ₁₄ O	4-Methyl-1-pentanol	151.9	44.46	60.47
C ₆ H ₁₄ O	2-Methyl-2-pentanol	121.1	39.59	54.77
C ₆ H ₁₄ O	4-Methyl-2-pentanol	131.6	44.2	
C ₆ H ₁₄ O	2-Ethyl-1-butanol	147	43.2	
C ₆ H ₁₄ O	Dipropyl ether	90.08	31.31	35.69
C ₆ H ₁₄ O	Diisopropyl ether	68.51	29.10	32.12
C ₆ H ₁₄ O	Butyl ethyl ether	92.3	31.63	36.32
C ₆ H ₁₄ O	Methyl pentyl ether	99	32.02	36.85
C ₆ H ₁₄ O ₂	2-Methyl-2,4-pentanediol	197.1	57.3	
C ₆ H ₁₄ O ₂	Ethylene glycol monobutyl ether	168.4		56.59
C ₆ H ₁₄ O ₂	1,1-Diethoxyethane	102.25	36.28	43.20
C ₆ H ₁₄ O ₂	Ethylene glycol diethyl ether	119.4	36.28	43.20
C ₆ H ₁₄ O ₃	Bis(ethoxymethyl) ether	140.6	36.17	44.69
C ₆ H ₁₄ O ₃	Diethylene glycol monoethyl ether	196	47.5	
C ₆ H ₁₄ O ₃	Diethylene glycol dimethyl ether	162	36.17	44.69
C ₆ H ₁₄ O ₄	Triethylene glycol	285	71.4	
C ₆ H ₁₄ S	Dipropyl sulfide	142.9	36.60	44.21
C ₆ H ₁₄ S	Diisopropyl sulfide	120.1	33.80	39.60
C ₆ H ₁₄ S	Isopropyl propyl sulfide	132.1	35.11	41.78
C ₆ H ₁₄ S	Butyl ethyl sulfide	144.3	37.01	44.51
C ₆ H ₁₄ S	Methyl pentyl sulfide	145.1	37.41	45.24
C ₆ H ₁₅ N	Hexylamine	132.8	36.54	45.10
C ₆ H ₁₅ N	Butylethylamine	107.5	33.97	40.15

ENTHALPY OF VAPORIZATION (continued)

Mol. Form.	Name	$t_b/^{\circ}\text{C}$	$\Delta_{\text{vap}}H(t_b)$ kJ/mol	$\Delta_{\text{vap}}H(25^{\circ}\text{C})$ kJ/mol
C ₆ H ₁₅ N	Dipropylamine	109.3	33.47	40.04
C ₆ H ₁₅ N	Diisopropylamine	83.9	30.40	34.61
C ₆ H ₁₅ N	Isopropylpropylamine	96.9	32.14	37.23
C ₆ H ₁₅ N	Triethylamine	89	31.01	34.84
C ₆ MoO ₆	Molybdenum hexacarbonyl	701	72.51	
C ₇ H ₃ F ₅	2,3,4,5,6-Pentafluorotoluene	117.5	34.75	41.12
C ₇ H ₅ F ₃	(Trifluoromethyl)benzene	102.1	32.63	37.60
C ₇ H ₅ N	Benzonitrile	191.1	45.9	
C ₇ H ₆ O	Benzaldehyde	179.0	42.5	
C ₇ H ₆ O ₂	Salicylaldehyde	197	38.2	
C ₇ H ₇ Cl	<i>o</i> -Chlorotoluene	159.0	37.5	
C ₇ H ₇ Cl	<i>p</i> -Chlorotoluene	162.4	38.7	
C ₇ H ₇ F	<i>o</i> -Fluorotoluene	115	35.4	
C ₇ H ₇ F	<i>p</i> -Fluorotoluene	116.6	34.08	39.42
C ₇ H ₈	Toluene	110.63	33.18	38.01
C ₇ H ₈ O	<i>o</i> -Cresol	191.04	45.19	
C ₇ H ₈ O	<i>m</i> -Cresol	202.27	47.40	61.71
C ₇ H ₈ O	<i>p</i> -Cresol	201.98	47.45	
C ₇ H ₈ O	Benzyl alcohol	205.31	50.48	
C ₇ H ₈ O	Anisole	153.7	38.97	46.90
C ₇ H ₉ N	Benzylamine	185		60.16
C ₇ H ₉ N	<i>o</i> -Methylaniline	200.3	44.6	
C ₇ H ₉ N	<i>m</i> -Methylaniline	203.3	44.9	
C ₇ H ₉ N	<i>p</i> -Methylaniline	200.4	44.3	
C ₇ H ₉ N	1-Cyclohexenecarbonitrile			53.55
C ₇ H ₉ N	2,3-Dimethylpyridine	161.12	39.08	47.82
C ₇ H ₉ N	2,4-Dimethylpyridine	158.38	38.53	47.49
C ₇ H ₉ N	2,5-Dimethylpyridine	156.98	38.68	47.04
C ₇ H ₉ N	2,6-Dimethylpyridine	144.01	37.46	45.34
C ₇ H ₉ N	3,4-Dimethylpyridine	179.10	39.99	50.50
C ₇ H ₉ N	3,5-Dimethylpyridine	171.84	39.46	49.33
C ₇ H ₁₀ O	Dicyclopentyl ketone	161		53.70
C ₇ H ₁₁ N	Cyclohexanecarbonitrile			51.92
C ₇ H ₁₂	1-Methylbicyclo(3,1,0)hexane	93.1	31.07	34.77
C ₇ H ₁₂ O ₄	Diethyl malonate	200	54.8	
C ₇ H ₁₄	1-Heptene	93.64		35.49
C ₇ H ₁₄	<i>cis</i> -2-Heptene	98.4		36.26
C ₇ H ₁₄	<i>trans</i> -2-Heptene	98		36.27
C ₇ H ₁₄	<i>cis</i> -3-Heptene	95.8		35.81
C ₇ H ₁₄	<i>trans</i> -3-Heptene	95.7		35.84
C ₇ H ₁₄	<i>cis</i> -3-Methyl-3-hexene	95.4		36.31
C ₇ H ₁₄	<i>trans</i> -3-Methyl-3-hexene	93.5		35.70
C ₇ H ₁₄	2,4-Dimethyl-1-pentene	81.6		33.03
C ₇ H ₁₄	4,4-Dimethyl-1-pentene	72.5		31.13
C ₇ H ₁₄	2,4-Dimethyl-2-pentene	83.4		34.19
C ₇ H ₁₄	<i>cis</i> -4,4-Dimethyl-2-pentene	80.4		32.56
C ₇ H ₁₄	<i>trans</i> -4,4-Dimethyl-2-pentene	76.7		32.81
C ₇ H ₁₄	2-Ethyl-3-methyl-1-butene	89		34.35
C ₇ H ₁₄	2,3,3-Trimethyl-1-butene	77.9		32.09
C ₇ H ₁₄	Methylcyclohexane	100.93	31.27	35.36
C ₇ H ₁₄	Ethylcyclopentane	103.5	31.96	36.40
C ₇ H ₁₄	<i>cis</i> -1,3-Dimethylcyclopentane	90.8	30.40	34.20
C ₇ H ₁₄ O	2-Heptanone	151.05		47.24
C ₇ H ₁₄ O	2,2-Dimethyl-3-pentanone	125.6	36.09	42.34
C ₇ H ₁₄ O	2,4-Dimethyl-3-pentanone	125.4	34.64	41.51
C ₇ H ₁₄ O	1-Methylcyclohexanol	155	79.0	
C ₇ H ₁₄ O	<i>cis</i> -2-Methylcyclohexanol	165	48.5	
C ₇ H ₁₄ O	<i>trans</i> -2-Methylcyclohexanol	167.5	53.0	

ENTHALPY OF VAPORIZATION (continued)

Mol. Form.	Name	$t_b/^{\circ}\text{C}$	$\Delta_{\text{vap}}H(t_b)$ kJ/mol	$\Delta_{\text{vap}}H(25^{\circ}\text{C})$ kJ/mol
C ₇ H ₁₄ O ₂	Pentyl acetate	149.2	38.42	48.56
C ₇ H ₁₄ O ₂	Isopentyl acetate	142.5	37.5	
C ₇ H ₁₄ O ₂	Ethyl pentanoate	146.1	36.96	47.01
C ₇ H ₁₄ O ₂	Ethyl 3-methylbutanoate	135.0	37.0	
C ₇ H ₁₄ O ₂	Ethyl 2,2-dimethylpropanoate	118.4	34.51	41.25
C ₇ H ₁₄ O ₂	Methyl hexanoate	149.5	38.55	48.04
C ₇ H ₁₅ Br	1-Bromoheptane	179		50.60
C ₇ H ₁₅ Cl	1-Chloroheptane	159		47.66
C ₇ H ₁₆	Heptane	98.5	31.77	36.57
C ₇ H ₁₆	2-Methylhexane	90.04	30.62	34.87
C ₇ H ₁₆	3-Methylhexane	92	30.9	
C ₇ H ₁₆	3-Ethylpentane	93.5	31.12	35.22
C ₇ H ₁₆	2,2-Dimethylpentane	79.2	29.23	32.42
C ₇ H ₁₆	2,3-Dimethylpentane	89.78	30.46	34.26
C ₇ H ₁₆	2,4-Dimethylpentane	80.49	29.55	32.88
C ₇ H ₁₆	3,3-Dimethylpentane	86.06	29.62	33.03
C ₇ H ₁₆	2,2,3-Trimethylbutane	80.86	28.90	32.05
C ₇ H ₁₆ O	Hexyl methyl ether	126.1	34.93	42.07
C ₇ H ₁₆ O	1-Heptanol	176.45		66.81
C ₇ H ₁₆ O	3-Heptanol	157	42.5	
C ₇ H ₁₆ O	Butyl propyl ether	118.1	33.72	40.22
C ₇ H ₁₆ O	Ethyl pentyl ether	117.6	34.41	41.01
C ₇ H ₁₇ N	Heptylamine	156		49.96
C ₈ F ₁₈	Perfluorooctane	105.9	33.38	41.13
C ₈ H ₈	Styrene	145	38.7	
C ₈ H ₈ O	Acetophenone	202	43.98	55.40
C ₈ H ₈ O ₂	Methyl benzoate	199		55.57
C ₈ H ₈ O ₃	Methyl salicylate	222.9	46.7	
C ₈ H ₁₀	Ethylbenzene	136.19	35.57	42.24
C ₈ H ₁₀	<i>o</i> -Xylene	144.5	36.24	43.43
C ₈ H ₁₀	<i>m</i> -Xylene	139.12	35.66	42.65
C ₈ H ₁₀	<i>p</i> -Xylene	138.37	35.67	42.40
C ₈ H ₁₀ O	2,4-Xylenol	210.98		64.96
C ₈ H ₁₀ O	2,5-Xylenol	211.1	46.9	
C ₈ H ₁₀ O	2,6-Xylenol	201.07		75.31
C ₈ H ₁₀ O	3,4-Xylenol	227		85.03
C ₈ H ₁₀ O	3,5-Xylenol	221.74		82.01
C ₈ H ₁₀ O	Phenetole	169.81		51.04
C ₈ H ₁₁ N	<i>N</i> -Ethylaniline	203.0		58.3
C ₈ H ₁₁ N	<i>N,N</i> -Dimethylaniline	194.15		52.83
C ₈ H ₁₁ N	2,4-Dimethylaniline	214		61.3
C ₈ H ₁₁ N	2,5-Dimethylaniline	214		61.7
C ₈ H ₁₁ N	2,3,6-Trimethylpyridine	171.6	39.95	50.61
C ₈ H ₁₁ N	2,4,6-Trimethylpyridine	170.6	39.87	50.33
C ₈ H ₁₄	1-Octyne	126.3	35.83	42.30
C ₈ H ₁₄	2-Octyne	137.6	37.26	44.49
C ₈ H ₁₄	3-Octyne	133.1	36.94	43.92
C ₈ H ₁₄	4-Octyne	131.6	36.0	42.73
C ₈ H ₁₄ O ₃	Butanoic anhydride	200	50.0	
C ₈ H ₁₅ N	Octanenitrile	205.25		56.80
C ₈ H ₁₆	1-Octene	121.29	34.07	40.34
C ₈ H ₁₆	<i>cis</i> -2,2-Dimethyl-3-hexene	105.5		36.86
C ₈ H ₁₆	<i>trans</i> -2,2-Dimethyl-3-hexene	100.8		37.03
C ₈ H ₁₆	3-Ethyl-2-methyl-1-pentene	109.5		37.27
C ₈ H ₁₆	2,4,4-Trimethyl-1-pentene	101.4		35.59
C ₈ H ₁₆	2,4,4-Trimethyl-2-pentene	104.9		37.23
C ₈ H ₁₆	Ethylcyclohexane	131.9	34.04	40.56
C ₈ H ₁₆	1,1-Dimethylcyclohexane	119.6	32.51	37.92

ENTHALPY OF VAPORIZATION (continued)

Mol. Form.	Name	$t_b/^{\circ}\text{C}$	$\Delta_{\text{vap}}H(t_b)$ kJ/mol	$\Delta_{\text{vap}}H(25^{\circ}\text{C})$ kJ/mol
C_8H_{16}	<i>cis</i> -1,2-Dimethylcyclohexane	129.8	33.47	39.70
C_8H_{16}	<i>trans</i> -1,2-Dimethylcyclohexane	123.5	32.96	38.36
C_8H_{16}	<i>cis</i> -1,3-Dimethylcyclohexane	120.1	32.91	38.26
C_8H_{16}	<i>trans</i> -1,3-Dimethylcyclohexane	124.5	33.39	39.16
C_8H_{16}	<i>cis</i> -1,4-Dimethylcyclohexane	124.4	33.28	39.02
C_8H_{16}	<i>trans</i> -1,4-Dimethylcyclohexane	119.4	32.56	37.90
C_8H_{16}	Propylcyclopentane	131	34.70	41.08
C_8H_{16}	Isopropylcyclopentane	126.5	33.56	39.44
C_8H_{16}	1-Ethyl-1-methylcyclopentane	121.6	33.20	38.85
$\text{C}_8\text{H}_{16}\text{O}$	2,2,4-Trimethyl-3-pentanone	135.1	35.64	43.30
$\text{C}_8\text{H}_{16}\text{O}_2$	Octanoic acid	239	58.5	
$\text{C}_8\text{H}_{16}\text{O}_2$	2-Ethylhexanoic acid	228		75.60
$\text{C}_8\text{H}_{16}\text{O}_2$	Isobutyl isobutanoate	148.6	38.2	
$\text{C}_8\text{H}_{16}\text{O}_2$	Ethyl hexanoate	167		51.72
$\text{C}_8\text{H}_{16}\text{O}_2$	Methyl heptanoate	174		51.62
$\text{C}_8\text{H}_{17}\text{Br}$	1-Bromooctane	200		55.77
$\text{C}_8\text{H}_{17}\text{Cl}$	1-Chlorooctane	181.5		52.42
$\text{C}_8\text{H}_{17}\text{F}$	1-Fluorooctane	142.4	40.43	49.65
C_8H_{18}	Octane	125.67	34.41	41.49
C_8H_{18}	2-Methylheptane	117.66	33.26	39.67
C_8H_{18}	3-Methylheptane	118.9	33.66	39.83
C_8H_{18}	4-Methylheptane	117.72	33.35	39.69
C_8H_{18}	3-Ethylhexane	118.6	33.59	39.64
C_8H_{18}	2,2-Dimethylhexane	106.86	32.07	37.28
C_8H_{18}	2,3-Dimethylhexane	115.62	33.17	38.78
C_8H_{18}	2,4-Dimethylhexane	109.5	32.51	37.76
C_8H_{18}	2,5-Dimethylhexane	109.12	32.54	37.85
C_8H_{18}	3,3-Dimethylhexane	111.97	32.31	37.53
C_8H_{18}	3,4-Dimethylhexane	117.73	33.24	38.97
C_8H_{18}	3-Ethyl-2-methylpentane	115.66	32.93	38.52
C_8H_{18}	3-Ethyl-3-methylpentane	118.27	32.78	37.99
C_8H_{18}	2,2,3-Trimethylpentane	110	31.94	36.91
C_8H_{18}	2,2,4-Trimethylpentane	99.22	30.79	35.14
C_8H_{18}	2,3,3-Trimethylpentane	114.8	32.12	37.27
C_8H_{18}	2,3,4-Trimethylpentane	113.5	32.36	37.75
C_8H_{18}	2,2,3,3-Tetramethylbutane	106.45		42.90
$\text{C}_8\text{H}_{18}\text{N}_2$	Azobutane			49.31
$\text{C}_8\text{H}_{18}\text{O}$	1-Octanol	195.16		70.98
$\text{C}_8\text{H}_{18}\text{O}$	2-Octanol	180	44.4	
$\text{C}_8\text{H}_{18}\text{O}$	2-Ethyl-1-hexanol	184.6	54.2	
$\text{C}_8\text{H}_{18}\text{O}$	Dibutyl ether	140.28	36.49	44.97
$\text{C}_8\text{H}_{18}\text{O}$	Di- <i>sec</i> -butyl ether	121.1	34.06	40.84
$\text{C}_8\text{H}_{18}\text{O}$	Di- <i>tert</i> -butyl ether	107.23	32.15	37.61
$\text{C}_8\text{H}_{18}\text{O}_2$	1,2-Dipropoxyethane			50.62
$\text{C}_8\text{H}_{18}\text{O}_3$	Diethylene glycol diethyl ether	188		58.40
$\text{C}_8\text{H}_{18}\text{S}$	Dibutyl sulfide	185		52.96
$\text{C}_8\text{H}_{18}\text{S}$	Di- <i>tert</i> -butyl sulfide	149.1	33.26	43.76
$\text{C}_8\text{H}_{18}\text{S}$	Diisobutyl sulfide	171		48.71
$\text{C}_8\text{H}_{19}\text{N}$	Dibutylamine	159.6	38.44	49.45
$\text{C}_8\text{H}_{19}\text{N}$	2-Ethylhexylamine	169.2	40.0	
$\text{C}_9\text{H}_7\text{N}$	Quinoline	237.16	49.7	59.30
$\text{C}_9\text{H}_7\text{N}$	Isoquinoline	243.22	49.0	60.26
C_9H_{10}	Cyclopropylbenzene	173.6		50.22
C_9H_{10}	Indan	177.97	39.63	48.79
$\text{C}_9\text{H}_{10}\text{O}_2$	Benzyl acetate	213	49.4	
C_9H_{12}	Propylbenzene	159.24		46.22
C_9H_{12}	Isopropylbenzene	152.41		45.13
C_9H_{12}	1,2,3-Trimethylbenzene	176.12		49.05

ENTHALPY OF VAPORIZATION (continued)

Mol. Form.	Name	$t_b/^{\circ}\text{C}$	$\Delta_{\text{vap}}H(t_b)$ kJ/mol	$\Delta_{\text{vap}}H(25^{\circ}\text{C})$ kJ/mol
C ₉ H ₁₂	1,2,4-Trimethylbenzene	169.38		47.93
C ₉ H ₁₂	1,3,5-Trimethylbenzene	164.74		47.50
C ₉ H ₁₄ O ₆	Triacetin	259		85.74
C ₉ H ₁₈	Butylcyclopentane	156.6	36.16	45.89
C ₉ H ₁₈	Propylcyclohexane	156.7		45.08
C ₉ H ₁₈	Isopropylcyclohexane	154.8		44.02
C ₉ H ₁₈ O	2-Nonanone	195.3		56.44
C ₉ H ₁₈ O	5-Nonanone	188.45		53.30
C ₉ H ₁₈ O	2,6-Dimethyl-4-heptanone	169.4		50.92
C ₉ H ₁₈ O ₂	Methyl octanoate	192.9		56.41
C ₉ H ₂₀	Nonane	150.82	37.18	46.55
C ₉ H ₂₀	2,2,5-Trimethylhexane	124.09	33.65	40.16
C ₉ H ₂₀	2,3,5-Trimethylhexane	131.4	34.43	41.41
C ₉ H ₂₀	3,3-Diethylpentane	146.3	34.61	42.0
C ₉ H ₂₀	2,2,4,4-Tetramethylpentane	122.29	32.51	38.49
C ₉ H ₂₀ O	1-Nonanol	213.37		76.86
C ₁₀ H ₇ Br	1-Bromonaphthalene	281	39.3	
C ₁₀ H ₇ Cl	1-Chloronaphthalene	259	52.1	
C ₁₀ H ₈	Naphthalene	217.9	43.2	
C ₁₀ H ₉ N	2-Methylquinoline	246.5		66.1
C ₁₀ H ₉ N	4-Methylquinoline	262		67.6
C ₁₀ H ₉ N	6-Methylquinoline	258.6		67.7
C ₁₀ H ₉ N	8-Methylquinoline	247.5		65.7
C ₁₀ H ₁₂	1,2,3,4-Tetrahydronaphthalene	207.6	43.9	
C ₁₀ H ₁₄	Butylbenzene	183.31	38.87	51.36
C ₁₀ H ₁₄	<i>sec</i> -Butylbenzene	173.3		47.98
C ₁₀ H ₁₄	<i>tert</i> -Butylbenzene	169.1		47.71
C ₁₀ H ₁₄	Isobutylbenzene	172.79		47.86
C ₁₀ H ₁₄	1-Isopropyl-4-methylbenzene	177.1	38.2	
C ₁₀ H ₁₆ O	(+)-Camphor	207.4	59.5	
C ₁₀ H ₁₈	<i>cis</i> -Decahydronaphthalene	195.8	41.0	
C ₁₀ H ₁₈	<i>trans</i> -Decahydronaphthalene	187.3	40.2	
C ₁₀ H ₁₉ N	Decanenitrile	243		66.84
C ₁₀ H ₂₀	1-Decene	170.5		50.43
C ₁₀ H ₂₀	Butylcyclohexane	180.9		49.36
C ₁₀ H ₂₀ O ₂	2-Ethylhexyl acetate	199	43.5	
C ₁₀ H ₂₀ O ₂	Isopentyl isopentanoate	190.4	45.9	
C ₁₀ H ₂₂	Decane	174.15	39.58	51.42
C ₁₀ H ₂₂	2-Methylnonane	167.1	38.23	49.63
C ₁₀ H ₂₂	3-Methylnonane	167.9	38.26	49.71
C ₁₀ H ₂₂	5-Methylnonane	165.1	38.14	49.34
C ₁₀ H ₂₂	2,4-Dimethyloctane	156	36.47	47.13
C ₁₀ H ₂₂ O	1-Decanol	231.1		81.50
C ₁₀ H ₂₂ O	Diisopentyl ether	172.5	35.1	
C ₁₀ H ₂₂ S	1-Decanethiol	240.6		65.48
C ₁₁ H ₁₀	1-Methylnaphthalene	244.7	45.5	
C ₁₁ H ₂₁ N	Undecanenitrile	253		71.14
C ₁₁ H ₂₂	Pentylcyclohexane	203.7		53.88
C ₁₁ H ₂₄	Undecane	195.9	41.91	56.58
C ₁₁ H ₂₄	2-Methyldecane	189.3	40.25	54.28
C ₁₁ H ₂₄	4-Methyldecane	187	40.70	53.76
C ₁₁ H ₂₄	2,4,7-Trimethyloctane	168.1	38.22	49.91
C ₁₂ F ₂₇ N	Tris(perfluorobutyl)amine	178	46.4	
C ₁₂ H ₁₀ O	Diphenyl ether	258.0	48.2	
C ₁₂ H ₁₆	Cyclohexylbenzene	240.1		59.94
C ₁₂ H ₂₂	Cyclohexylcyclohexane	238		57.98
C ₁₂ H ₂₃ N	Dodecanenitrile	277		76.12
C ₁₂ H ₂₄	1-Dodecene	213.8		60.78

ENTHALPY OF VAPORIZATION (continued)

Mol. Form.	Name	$t_b/^\circ\text{C}$	$\Delta_{\text{vap}}H(t_b)$ kJ/mol	$\Delta_{\text{vap}}H(25^\circ\text{C})$ kJ/mol
$\text{C}_{12}\text{H}_{26}$	2,2,4,6,6-Pentamethylheptane	177.8		48.97
$\text{C}_{12}\text{H}_{26}$	Dodecane	216.32	44.09	61.52
$\text{C}_{12}\text{H}_{26}\text{O}$	1-Dodecanol	259		91.96
$\text{C}_{12}\text{H}_{27}\text{BO}_3$	Tributyl borate	234	56.1	
$\text{C}_{12}\text{H}_{27}\text{N}$	Tributylamine	216.5	46.9	
$\text{C}_{13}\text{H}_{13}\text{N}$	<i>N</i> -Benzylaniline	306.5		79.6
$\text{C}_{13}\text{H}_{26}\text{O}_2$	Methyl dodecanoate	267		77.17
$\text{C}_{13}\text{H}_{28}$	Tridecane	235.47	46.20	66.68
$\text{C}_{14}\text{H}_{10}$	Phenanthrene	340		75.50
$\text{C}_{14}\text{H}_{12}\text{O}_2$	Benzyl benzoate	323.5	53.6	
$\text{C}_{14}\text{H}_{27}\text{N}$	Tetradecanenitrile			85.29
$\text{C}_{14}\text{H}_{30}$	Tetradecane	253.58	48.16	71.73
$\text{C}_{14}\text{H}_{30}\text{O}$	1-Tetradecanol	289		102.20
$\text{C}_{15}\text{H}_{32}$	Pentadecane	270.6	50.08	76.77
$\text{C}_{16}\text{H}_{22}\text{O}_4$	Dibutyl phthalate	340	79.2	
$\text{C}_{16}\text{H}_{32}$	1-Hexadecene	284.9		80.25
$\text{C}_{16}\text{H}_{34}$	Hexadecane	286.86	51.84	81.35
$\text{C}_{17}\text{H}_{36}$	Heptadecane	302.0	53.58	86.47
$\text{C}_{18}\text{H}_{34}\text{O}_2$	Oleic acid	360	67.4	
$\text{C}_{18}\text{H}_{38}$	Octadecane	316.3	55.23	91.44
$\text{C}_{19}\text{H}_{40}$	Nonadecane	329.9	56.93	96.4
$\text{C}_{20}\text{H}_{42}$	Eicosane	343	58.49	101.81

ENTHALPY OF FUSION

This table lists the molar enthalpy (heat) of fusion, $\Delta_{\text{fus}}H$, of over 800 inorganic and organic compounds. All values refer to the enthalpy change at equilibrium between the liquid phase and the most stable solid phase at the transition temperature. Most values of $\Delta_{\text{fus}}H$ are given at the normal melting point t_m . However, a "t" following the entry in the melting point column indicate a triple-point temperature, where the solid, liquid, and gas phases are in equilibrium. Substances are listed by molecular formula in the Hill order, with substances containing carbon (except graphite) following those that do not contain carbon.

All temperatures are given on the ITS-90 scale.

A * following an entry indicates that the value includes the enthalpy of transition between crystalline phases whose transformation occurs within 1°C of the melting point.

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Molecular formula	Name	$t_m/^\circ\text{C}$	$\Delta_{\text{fus}}H/\text{kJ mol}^{-1}$
Ag	Silver	961.78	11.28
AgBr	Silver(I) bromide	432	9.12
AgCl	Silver(I) chloride	455	13.2
AgI	Silver(I) iodide	558	9.41
AgNO ₃	Silver(I) nitrate	212	11.5
Ag ₂ S	Silver(I) sulfide	825	14.1
Al	Aluminum	660.32	10.789
AlBr ₃	Aluminum bromide	97.5	11.25
AlCl ₃	Aluminum chloride	192.6	35.4
AlF ₃	Aluminum fluoride	2250 t	98
AlI ₃	Aluminum iodide	188.28	15.9
Al ₂ O ₃	Aluminum oxide	2053	111.4
Al ₂ S ₃	Aluminum sulfide	1100	55
Am	Americium	1176	14.39
Ar	Argon	-189.36 t	1.18
As	Arsenic (gray)	817 t	24.44
AsBr ₃	Arsenic(III) bromide	31.1	11.7
AsCl ₃	Arsenic(III) chloride	-16	10.1
AsF ₃	Arsenic(III) fluoride	-5.9	10.4
Au	Gold	1064.18	12.72
B	Boron	2075	50.2
BCl ₃	Boron trichloride	-107	2.10
BF ₃	Boron trifluoride	-126.8	4.20
BHO ₂	Metaboric acid (γ form)	236	14.3
BH ₃ O ₃	Boric acid (orthoboric acid)	170.9	22.3
BN	Boron nitride	2966	81

ENTHALPY OF FUSION (continued)

Molecular formula	Name	$t_m/^\circ\text{C}$	$\Delta_{\text{fus}}H/\text{kJ mol}^{-1}$
BNaO ₂	Sodium metaborate	966	36.2
B ₂ O ₃	Boron oxide	450	24.56
Ba	Barium	727	7.12
BaBr ₂	Barium bromide	857	32.2
BaCl ₂	Barium chloride	962	15.85
BaF ₂	Barium fluoride	1368	17.8
BaH ₂	Barium hydride	1200	25
BaH ₂ O ₂	Barium hydroxide	408	16
BaI ₂	Barium iodide	711	26.5
BaO	Barium oxide	1972	46
BaO ₄ S	Barium sulfate	1580	40
BaS	Barium sulfide	2229	63
Be	Beryllium	1287	7.895
BeBr ₂	Beryllium bromide	508	18
BeCl ₂	Beryllium chloride	415	8.66
BeF ₂	Beryllium fluoride	552	4.77
BeI ₂	Beryllium iodide	470	18
BeO	Beryllium oxide	2577	86
BeO ₄ S	Beryllium sulfate	1127	6
Bi	Bismuth	271.40	11.145
BiCl ₃	Bismuth trichloride	230	10.9
BrF ₅	Bromine pentafluoride	-60.5	5.67
BrH	Hydrogen bromide	-86.80	2.41
BrIn	Indium(I) bromide	290	15
BrK	Potassium bromide	734	25.5
BrLi	Lithium bromide	552	17.6
BrNa	Sodium bromide	747	26.11
BrNaO ₃	Sodium bromate	381	28.11
BrRb	Rubidium bromide	682	15.5
BrTl	Thallium(I) bromide	460	16.4
Br ₂	Bromine	-7.2	10.57
Br ₂ Ca	Calcium bromide	742	29.1
Br ₂ Cd	Cadmium bromide	568	20.9
Br ₂ Fe	Iron(II) bromide	691	50.2
Br ₂ Hg	Mercury(II) bromide	236	17.9
Br ₂ Mg	Magnesium bromide	711	39.3
Br ₂ Pb	Lead(II) bromide	371	16.44
Br ₂ Sr	Strontium bromide	657	10.1
Br ₂ Zn	Zinc bromide	394	16.7
Br ₃ Ga	Gallium(III) bromide	121.5	12.1
Br ₃ In	Indium(III) bromide	420	26
Br ₃ Pu	Plutonium(III) bromide	681	55.2
Br ₃ U	Uranium(III) bromide	727	43.9
Br ₄ Sn	Tin(IV) bromide	29.1	12.2
Br ₄ Th	Thorium(IV) bromide	679	66.9
Br ₄ Ti	Titanium(IV) bromide	39	12.9
Br ₄ U	Uranium(IV) bromide	519	55.2
Br ₅ Ta	Tantalum(V) bromide	265	45.6
C	Carbon (graphite)	4489 t	117
Ca	Calcium	842	8.54
CaCl ₂	Calcium chloride	775	28.05
CaF ₂	Calcium fluoride	1418	30
CaH ₂	Calcium hydride	1000	6.7
CaI ₂	Calcium iodide	783	41.8
CaO	Calcium oxide	2898	80
CaO ₄ S	Calcium sulfate	1460	28
CaS	Calcium sulfide	2524	70
Cd	Cadmium	321.07	6.21

ENTHALPY OF FUSION (continued)

Molecular formula	Name	$t_m/^{\circ}\text{C}$	$\Delta_{\text{fus}}H/\text{kJ mol}^{-1}$
CdCl ₂	Cadmium chloride	564	48.58
CdF ₂	Cadmium fluoride	1110	22.6
CdI ₂	Cadmium iodide	387	15.3
Ce	Cerium	798	5.46
CeCl ₃	Cerium(III) chloride	817	54.4
ClCs	Cesium chloride	645	15.9
ClCu	Copper(I) chloride	430	10.2
ClH	Hydrogen chloride	-114.17	2.00
ClI	Iodine chloride	27.39	11.6
ClIn	Indium(I) chloride	211	21.3
ClK	Potassium chloride	771	26.53
CLi	Lithium chloride	610	19.9
CLiO ₄	Lithium perchlorate	236	29
ClNa	Sodium chloride	800.7	28.16
ClNaO ₃	Sodium chlorate	248	22.1
ClRb	Rubidium chloride	715	18.4
ClTl	Thallium(I) chloride	430	15.56
Cl ₂	Chlorine	-101.5	6.40
Cl ₂ Co	Cobalt(II) chloride	740	45
Cl ₂ Cr	Chromium(II) chloride	814	32.2
Cl ₂ Cu	Copper(II) chloride	630	20.4
Cl ₂ Fe	Iron(II) chloride	677	43.01
Cl ₂ Hg	Mercury(II) chloride	276	19.41
Cl ₂ Mg	Magnesium chloride	714	43.1
Cl ₂ Mn	Manganese(II) chloride	650	30.7
Cl ₂ Ni	Nickel(II) chloride	1009	71.2
Cl ₂ Pb	Lead(II) chloride	501	21.75
Cl ₂ Sn	Tin(II) chloride	247.1	14.52
Cl ₂ Sr	Strontium chloride	874	17.5
Cl ₃ Fe	Iron(III) chloride	304	43.1
Cl ₃ Ga	Gallium(III) chloride	77.9	11.13
Cl ₃ In	Indium(III) chloride	583	27
Cl ₃ La	Lanthanum chloride	859	43.1
Cl ₃ OP	Phosphorus(V) oxychloride	1.18	13.1
Cl ₃ P	Phosphorus(III) chloride	-112	7.10
Cl ₃ Sb	Antimony(III) chloride	73.4	12.7
Cl ₄ OW	Tungsten(VI) oxytetrachloride	211	45
Cl ₄ Si	Tetrachlorosilane	-68.74	7.60
Cl ₄ Sn	Tin(IV) chloride	-34.07	9.20
Cl ₄ Th	Thorium(IV) chloride	770	40.2
Cl ₄ Ti	Titanium(IV) chloride	-24.12	9.97
Cl ₄ U	Uranium(IV) chloride	590	45
Cl ₄ V	Vanadium(IV) chloride	-25.7	2.30
Cl ₄ Zr	Zirconium(IV) chloride	437 t	50
Cl ₅ Mo	Molybdenum(V) chloride	194	19
Cl ₅ Nb	Niobium(V) chloride	204.7	38.3
Cl ₅ Ta	Tantalum(V) chloride	216	41.6
Cl ₆ W	Tungsten(VI) chloride	275	6.60
Co	Cobalt	1495	16.06
CoF ₂	Cobalt(II) fluoride	1127	59
Cr	Chromium	1907	21.0
Cr ₂ O ₃	Chromium(III) oxide	2329	130
Cs	Cesium	28.5	2.09
CsF	Cesium fluoride	703	21.7
CsHO	Cesium hydroxide	342.3	7.78
Cs ₂ O ₄ S	Cesium sulfate	1005	35.7
Cu	Copper	1084.62	12.93
CuF ₂	Copper(II) fluoride	836	55

ENTHALPY OF FUSION (continued)

Molecular formula	Name	$t_m/^{\circ}\text{C}$	$\Delta_{\text{fus}}H/\text{kJ mol}^{-1}$
CuO	Copper(II) oxide	1446	11.8
Dy	Dysprosium	1412	11.06
Er	Erbium	1529	19.9
Eu	Europium	822	9.21
FH	Hydrogen fluoride	-83.35	4.58
FK	Potassium fluoride	858	27.2
FLi	Lithium fluoride	848.2	27.09
FNa	Sodium fluoride	996	33.35
FRb	Rubidium fluoride	833	17.3
FTl	Thallium(I) fluoride	326	13.87
F ₂	Fluorine	-219.66	0.51
F ₂ Fe	Iron(II) fluoride	1100	52
F ₂ HK	Potassium hydrogen fluoride	238.9	6.62
F ₂ Mg	Magnesium fluoride	1263	58.5
F ₂ Pb	Lead(II) fluoride	830	14.7
F ₂ Sr	Strontium fluoride	1477	28.5
F ₃ In	Indium(III) fluoride	1170	64
F ₃ Pu	Plutonium(III) fluoride	1396	59.8
F ₄ Pu	Plutonium(IV) fluoride	1027	65.3
F ₄ Th	Thorium(IV) fluoride	1110	44.0
F ₄ U	Uranium(IV) fluoride	1036	42.7
F ₄ Zr	Zirconium(IV) fluoride	932 t	64.2
F ₅ Nb	Niobium(V) fluoride	80	12.2
F ₅ V	Vanadium(V) fluoride	19.5	49.96
F ₆ Ir	Iridium(VI) fluoride	44	8.40
F ₆ Mo	Molybdenum(VI) fluoride	17.5	4.33
F ₆ Pu	Plutonium(VI) fluoride	52	17.6
F ₆ S	Sulfur hexafluoride	-50.7 t	5.02
F ₆ U	Uranium(VI) fluoride	64.0 t	19.1
F ₆ W	Tungsten(VI) fluoride	2.3	4.10
Fe	Iron	1538	13.81
FeI ₂	Iron(II) iodide	587	45
FeO	Iron(II) oxide	1377	24
FeS	Iron(II) sulfide	1188	31.5
Fe ₃ O ₄	Iron(II,III) oxide	1597	138
Ga	Gallium	29.76	5.576
GaI ₃	Gallium(III) iodide	212	12.9
GaSb	Gallium antimonide	712	25.1
Ga ₂ O ₃	Gallium(III) oxide	1806	100
Gd	Gadolinium	1313	10.0
Ge	Germanium	938.25	36.94
HI	Hydrogen iodide	-50.76	2.87
HKO	Potassium hydroxide	406	7.9
HLi	Lithium hydride	688.7	22.59
HLiO	Lithium hydroxide	471.1	20.88
HNO ₃	Nitric acid	-41.6	10.5
HNaO	Sodium hydroxide	323	6.60
HORb	Rubidium hydroxide	382	8.0
H ₂	Hydrogen	-259.34	0.12
H ₂ Mg	Magnesium hydride	327	14
H ₂ O	Water	0.00	6.01
H ₂ O ₂	Hydrogen peroxide	-0.43	12.50
H ₂ O ₂ Sr	Strontium hydroxide	535	23
H ₂ O ₄ S	Sulfuric acid	10.31	10.71
H ₂ S	Hydrogen sulfide	-85.5	2.38
H ₂ Sr	Strontium hydride	1050	23
H ₃ N	Ammonia	-77.73	5.66

ENTHALPY OF FUSION (continued)

Molecular formula	Name	$t_m/^{\circ}\text{C}$	$\Delta_{\text{fus}}H/\text{kJ mol}^{-1}$
H ₃ O ₂ P	Hypophosphorous acid	26.5	9.7
H ₃ O ₃ P	Phosphorous acid	74.4	12.8
H ₃ O ₄ P	Phosphoric acid	42.4	13.4
H ₄ IN	Ammonium iodide	551	21
H ₄ N ₂	Hydrazine	1.4	12.6
H ₄ N ₂ O ₃	Ammonium nitrate	169.7	6.40
Hf	Hafnium	2233	27.2
Hg	Mercury	-38.83	2.29
HgI ₂	Mercury(II) iodide	259	18.9
Hg ₂ I ₂	Mercury(I) iodide	290	27
Ho	Holmium	1474	17.0
In	Indium(I) iodide	364.4	17.26
IK	Potassium iodide	681	24
ILi	Lithium iodide	469	14.6
INa	Sodium iodide	660	23.6
IRb	Rubidium iodide	642	12.5
ITl	Thallium(I) iodide	441.7	14.73
I ₂	Iodine	113.7	15.52
I ₂ Mg	Magnesium iodide	634	26
I ₂ Pb	Lead(II) iodide	410	23.4
I ₂ Sr	Strontium iodide	538	19.7
I ₃ In	Indium(III) iodide	207	18.48
I ₄ Si	Tetraiodosilane	120.5	19.7
I ₄ Th	Thorium(IV) iodide	570	61.4
I ₄ Ti	Titanium(IV) iodide	150	19.8
I ₄ U	Uranium(IV) iodide	506	70.7
In	Indium	156.60	3.281
InSb	Indium antimonide	525	25.5
In ₂ O ₃	Indium(III) oxide	1912	105
Ir	Iridium	2446	41.12
K	Potassium	63.5	2.33
KNO ₃	Potassium nitrate	337	10.1
K ₂ O ₄ S	Potassium sulfate	1069	36.4
K ₂ S	Potassium sulfide	948	16.15
Kr	Krypton	-157.38 t	1.64
La	Lanthanum	918	6.20
Li	Lithium	180.50	3.00
LiNO ₃	Lithium nitrate	253	24.9
Li ₂ O ₃ Si	Lithium metasilicate	1201	28
Li ₂ O ₄ S	Lithium sulfate	859	7.50
Lu	Lutetium	1663	22
Mg	Magnesium	650	8.48
MgO	Magnesium oxide	2825	77
MgO ₄ S	Magnesium sulfate	1127	14.6
MgS	Magnesium sulfide	2226	63
Mg ₂ O ₄ Si	Magnesium orthosilicate	1897	71
Mn	Manganese	1246	12.91
MnO	Manganese(II) oxide	1839	54.4
Mo	Molybdenum	2623	37.48
MoO ₃	Molybdenum(VI) oxide	801	48
NNaO ₃	Sodium nitrate	307	15
NO	Nitric oxide	-163.6	2.30
NO ₃ Rb	Rubidium nitrate	305	5.60
NO ₃ Tl	Thallium(I) nitrate	206	9.6
N ₂	Nitrogen	-210.0	0.71
N ₂ O	Nitrous oxide	-90.8	6.54
N ₂ O ₄	Nitrogen tetroxide	-9.3	14.65

ENTHALPY OF FUSION (continued)

Molecular formula	Name	$t_m/^{\circ}\text{C}$	$\Delta_{\text{fus}}H/\text{kJ mol}^{-1}$
Na	Sodium	97.80	2.60
Na ₂ O	Sodium oxide	1132	48
Na ₂ O ₃ Si	Sodium metasilicate	1089	52
Na ₂ O ₄ S	Sodium sulfate	884	23.6
Na ₂ S	Sodium sulfide	1172	19
Nb	Niobium	2477	30
NbO	Niobium(II) oxide	1936	85
NbO ₂	Niobium(IV) oxide	1901	92
Nb ₂ O ₅	Niobium(V) oxide	1512	104.3
Nd	Neodymium	1021	7.14
Ne	Neon	-248.61 t	0.328
Ni	Nickel	1455	17.04
NiS	Nickel(II) sulfide	976	30.1
Np	Neptunium	644	3.20
OSr	Strontium oxide	2531	81
OTl ₂	Thallium(I) oxide	579	30.3
OV	Vanadium(II) oxide	1789	63
OZn	Zinc oxide	1974	52.3
O ₂	Oxygen	-218.79	0.44
O ₂ Si	Silicon dioxide (cristobalite)	1722	9.6
O ₂ Zr	Zirconium(IV) oxide	2709	87
O ₃ S	Sulfur trioxide	16.8	8.60
O ₃ Tl ₂	Thallium(III) oxide	834	53
O ₃ W	Tungsten(VI) oxide	1472	73
O ₃ Y ₂	Yttrium oxide	2438	105
O ₄ Os	Osmium(VIII) oxide	41	9.8
O ₄ SSr	Strontium sulfate	1606	36
O ₄ STl ₂	Thallium(I) sulfate	632	23
O ₅ P ₂	Phosphorus(V) oxide	562	27.2
O ₅ Ta ₂	Tantalum(V) oxide	1784	120
O ₅ V ₂	Vanadium(V) oxide	670	64.5
O ₇ Re ₂	Rhenium(VII) oxide	297	64.2
Os	Osmium	3033	57.85
P	Phosphorus (white)	44.15	0.66
Pa	Protactinium	1572	12.34
Pb	Lead	327.46	4.782
PbS	Lead(II) sulfide	1113	49.4
Pd	Palladium	1554.9	16.74
Pr	Praseodymium	931	6.89
Pt	Platinum	1768.4	22.17
Pu	Plutonium	640	2.82
Rb	Rubidium	39.3	2.19
Re	Rhenium	3186	60.43
Rh	Rhodium	1964	26.59
Ru	Ruthenium	2334	38.59
S	Sulfur (monoclinic)	115.21	1.72
SSr	Strontium sulfide	2226	63
STl ₂	Thallium(I) sulfide	448	12
Sb	Antimony	630.63	19.79
Sc	Scandium	1541	14.1
Se	Selenium (gray)	220.5	6.69
Si	Silicon	1414	50.21
Sm	Samarium	1074	8.62
Sn	Tin (white)	231.93	7.173
Sr	Strontium	777	7.43
Ta	Tantalum	3017	36.57
Tb	Terbium	1356	10.15
Tc	Technetium	2157	33.29

ENTHALPY OF FUSION (continued)

Molecular formula	Name	$t_m/^{\circ}\text{C}$	$\Delta_{\text{fus}}H/\text{kJ mol}^{-1}$
Te	Tellurium	449.51	17.49
Th	Thorium	1750	13.81
Ti	Titanium	1668	14.15
Tl	Thallium	304	4.14
Tm	Thulium	1545	16.84
U	Uranium	1135	9.14
V	Vanadium	1910	21.5
W	Tungsten	3422	52.31
Xe	Xenon	-111.79 t	2.27
Y	Yttrium	1522	11.4
Yb	Ytterbium	819	7.66
Zn	Zinc	419.53	7.068
Zr	Zirconium	1855	21.00
CBaO ₃	Barium carbonate	1555	40
CBrCl ₃	Bromotrichloromethane	-5.65	2.53
CBr ₄	Tetrabromomethane	92.3	3.76
CCaO ₃	Calcium carbonate (calcite)	1330	36
CCl ₂ O	Carbonyl chloride	-127.78	5.74
CCl ₃ F	Trichlorofluoromethane	-110.44	6.89
CCl ₄	Tetrachloromethane	-22.62	2.56
CF ₄	Tetrafluoromethane	-183.60	0.704
CHBr ₃	Tribromomethane	8.69	11.05
CHClF ₂	Chlorodifluoromethane	-157.42	4.12
CHCl ₃	Trichloromethane	-63.41	9.5
CHF ₃	Trifluoromethane	-155.2	4.06
CHI ₃	Triiodomethane	121.2	16.44
CHN	Hydrogen cyanide	-13.29	8.41
CHNaO ₂	Sodium formate	257.3	17.7
CHO ₂ Tl	Thallium(I) formate	101	10.9
CH ₂ Cl ₂	Dichloromethane	-97.2	4.60
CH ₂ N ₂	Cyanamide	45.56	7.27
CH ₂ N ₄	Tetrazole	157.3	18.2
CH ₂ O ₂	Formic acid	8.3	12.68
CH ₃ Br	Bromomethane	-93.68	5.98
CH ₃ Cl	Chloromethane	-97.7	6.43
CH ₃ NO	Formamide	2.49	8.44
CH ₃ NO ₂	Nitromethane	-28.38	9.70
CH ₃ NO ₃	Methyl nitrate	-83.0	8.24
CH ₄	Methane	-182.47	0.94
CH ₄ N ₂ O	Urea	133.3	13.9
CH ₄ N ₂ S	Thiourea	178	14.0
CH ₄ O	Methanol	-97.53	3.215
CH ₄ S	Methanethiol	-123	5.91
CH ₅ N	Methylamine	-93.5	6.13
CH ₆ N ₂	Methylhydrazine	-52.36	10.42
CK ₂ O ₃	Potassium carbonate	898	27.6
CLi ₂ O ₃	Lithium carbonate	723	41
CMgO ₃	Magnesium carbonate	990	59
CNa ₂ O ₃	Sodium carbonate	858.1	29.7
CO	Carbon monoxide	-205.02	0.833
COS	Carbon oxysulfide	-138.8	4.73
CO ₂	Carbon dioxide	-56.56 t	9.02
CO ₃ Sr	Strontium carbonate	1494	40
CO ₃ Tl ₂	Thallium(I) carbonate	272	18.4
CS ₂	Carbon disulfide	-112.1	4.39
CSe ₂	Carbon diselenide	-43.7	6.36
C ₂ Br ₂ F ₄	1,2-Dibromotetrafluoroethane	-110.32	7.04
C ₂ ClF ₃	Chlorotrifluoroethene	-158.2	5.55

ENTHALPY OF FUSION (continued)

Molecular formula	Name	$t_m/^{\circ}\text{C}$	$\Delta_{\text{fus}}H/\text{kJ mol}^{-1}$
C_2ClF_5	Chloropentafluoroethane	-99.4	1.86
$\text{C}_2\text{Cl}_2\text{F}_4$	1,2-Dichloro-1,1,2,2-tetrafluoroethane	-92.53	1.51
$\text{C}_2\text{Cl}_3\text{F}_3$	1,1,2-Trichloro-1,2,2-trifluoroethane	-36.22	2.47
C_2Cl_4	Tetrachloroethene	-22.3	10.88
$\text{C}_2\text{Cl}_4\text{F}_2$	1,1,2,2-Tetrachloro-1,2-difluoroethane	24.8	3.67
C_2Cl_6	Hexachloroethane	186.8 t	9.75
C_2F_4	Tetrafluoroethene	-131.15	7.72
C_2F_6	Hexafluoroethane	-100.05	2.69
C_2HCl_3	Trichloroethylene	-84.7	8.45
$\text{C}_2\text{HCl}_3\text{O}_2$	Trichloroacetic acid	59.2	5.90
C_2HCl_5	Pentachloroethane	-28.78	11.3
$\text{C}_2\text{H}_2\text{Cl}_2$	1,1-Dichloroethene	-122.56	6.51
$\text{C}_2\text{H}_2\text{Cl}_2$	<i>cis</i> -1,2-Dichloroethene	-80.0	7.2
$\text{C}_2\text{H}_2\text{Cl}_4$	1,1,2,2-Tetrachloroethane	-42.4	9.17
$\text{C}_2\text{H}_3\text{Br}$	Bromoethene	-139.54	5.12
$\text{C}_2\text{H}_3\text{Cl}$	Chloroethene	-153.84	4.92
$\text{C}_2\text{H}_3\text{ClO}_2$	Chloroacetic acid	63	12.28
$\text{C}_2\text{H}_3\text{Cl}_3$	1,1,1-Trichloroethane	-30.01	2.35
$\text{C}_2\text{H}_3\text{Cl}_3$	1,1,2-Trichloroethane	-36.3	11.46
$\text{C}_2\text{H}_3\text{F}_3$	1,1,1-Trifluoroethane	-111.3	6.19
$\text{C}_2\text{H}_3\text{KO}_2$	Potassium acetate	309	7.65
$\text{C}_2\text{H}_3\text{N}$	Acetonitrile	-43.82	8.16
$\text{C}_2\text{H}_3\text{NaO}_2$	Sodium acetate	328.2	17.9
C_2H_4	Ethylene	-169.15	3.35
$\text{C}_2\text{H}_4\text{Br}_2$	1,2-Dibromoethane	9.84	10.89
$\text{C}_2\text{H}_4\text{Cl}_2$	1,1-Dichloroethane	-96.9	7.87
$\text{C}_2\text{H}_4\text{Cl}_2$	1,2-Dichloroethane	-35.7	8.84
$\text{C}_2\text{H}_4\text{O}$	Acetaldehyde	-123.37	2.31
$\text{C}_2\text{H}_4\text{O}$	Ethylene oxide	-112.5	5.17
$\text{C}_2\text{H}_4\text{O}_2$	Acetic acid	16.64	11.73
$\text{C}_2\text{H}_5\text{Br}$	Bromoethane	-118.6	7.47
$\text{C}_2\text{H}_5\text{Cl}$	Chloroethane	-138.4	4.45
$\text{C}_2\text{H}_5\text{NO}$	Acetamide	80.16	15.59
$\text{C}_2\text{H}_5\text{NO}_2$	Nitroethane	-89.5	9.85
C_2H_6	Ethane	-182.79	2.72*
$\text{C}_2\text{H}_6\text{N}_2\text{O}$	<i>N</i> -Methylurea	104.9	14.0
$\text{C}_2\text{H}_6\text{O}$	Ethanol	-114.14	4.931
$\text{C}_2\text{H}_6\text{O}$	Dimethyl ether	-141.5	4.94
$\text{C}_2\text{H}_6\text{OS}$	Dimethyl sulfoxide	17.89	14.37
$\text{C}_2\text{H}_6\text{O}_2$	Ethylene glycol	-12.69	9.96
$\text{C}_2\text{H}_6\text{O}_2\text{S}$	Dimethyl sulfone	108.9	18.30
$\text{C}_2\text{H}_6\text{S}$	Ethanethiol	-147.88	4.98
$\text{C}_2\text{H}_6\text{S}$	Dimethyl sulfide	-98.24	7.99
$\text{C}_2\text{H}_6\text{S}_2$	Dimethyl disulfide	-84.67	9.19
$\text{C}_2\text{H}_6\text{Zn}$	Dimethyl zinc	-43.0	6.83
$\text{C}_2\text{H}_7\text{N}$	Dimethylamine	-92.18	5.94
$\text{C}_2\text{H}_8\text{N}_2$	1,2-Ethanediamine	11.14	22.58
$\text{C}_2\text{H}_8\text{N}_2$	1,1-Dimethylhydrazine	-57.20	10.07
$\text{C}_2\text{H}_8\text{N}_2$	1,2-Dimethylhydrazine	-8.9	13.64
C_2N_2	Cyanogen	-27.83	8.11
$\text{C}_3\text{F}_6\text{O}$	Perfluoroacetone	-125.45	8.38
C_3F_8	Perfluoropropane	-147.70	0.477
$\text{C}_3\text{H}_3\text{N}$	Acrylonitrile	-83.48	6.23
$\text{C}_3\text{H}_3\text{NS}$	Thiazole	-33.62	9.57
$\text{C}_3\text{H}_3\text{N}_3$	1,3,5-Triazine	80.3	14.56
C_3H_4	Allene	-136.6	4.40
$\text{C}_3\text{H}_4\text{N}_2$	1 <i>H</i> -Pyrazole	70.7	14.0
$\text{C}_3\text{H}_4\text{N}_2$	Imidazole	89.5	12.82

ENTHALPY OF FUSION (continued)

Molecular formula	Name	$t_m/^{\circ}\text{C}$	$\Delta_{\text{fus}}H/\text{kJ mol}^{-1}$
$\text{C}_3\text{H}_4\text{O}_2$	Acrylic acid	12.5	9.51
$\text{C}_3\text{H}_5\text{N}$	Propanenitrile	-92.78	5.03
$\text{C}_3\text{H}_5\text{N}_3\text{O}_9$	Trinitroglycerol	13.5	21.87
C_3H_6	Propene	-185.24	3.003
C_3H_6	Cyclopropane	-127.58	5.44
$\text{C}_3\text{H}_6\text{Br}_2$	1,2-Dibromopropane	-55.49	8.94
$\text{C}_3\text{H}_6\text{Br}_2$	1,3-Dibromopropane	-34.5	14.6
$\text{C}_3\text{H}_6\text{Cl}_2$	1,2-Dichloropropane, (\pm)	-100.53	6.40
$\text{C}_3\text{H}_6\text{Cl}_2$	2,2-Dichloropropane	-33.9	2.30
$\text{C}_3\text{H}_6\text{O}$	Acetone	-94.7	5.77
$\text{C}_3\text{H}_6\text{O}$	Methyloxirane	-111.9	6.53
$\text{C}_3\text{H}_6\text{O}$	Oxetane	-97	6.5
$\text{C}_3\text{H}_6\text{O}_2$	Propanoic acid	-20.5	10.66
$\text{C}_3\text{H}_6\text{O}_2$	Methyl acetate	-98.25	7.49
$\text{C}_3\text{H}_6\text{O}_2$	1,3-Dioxolane	-97.22	6.57
$\text{C}_3\text{H}_6\text{O}_3$	1,3,5-Trioxane	60.29	15.11
$\text{C}_3\text{H}_6\text{S}$	Thietane	-73.24	8.25
$\text{C}_3\text{H}_7\text{Br}$	1-Bromopropane	-110.3	6.44
$\text{C}_3\text{H}_7\text{Br}$	2-Bromopropane	-89.0	6.53
$\text{C}_3\text{H}_7\text{Cl}$	1-Chloropropane	-122.9	5.54
$\text{C}_3\text{H}_7\text{Cl}$	2-Chloropropane	-117.18	7.39
$\text{C}_3\text{H}_7\text{N}$	Cyclopropylamine	-35.39	13.18
$\text{C}_3\text{H}_7\text{NO}$	<i>N,N</i> -Dimethylformamide	-60.48	7.90
C_3H_8	Propane	-187.63	3.50
$\text{C}_3\text{H}_8\text{N}_2\text{O}$	<i>N,N</i> -Dimethylurea	182.1	23.0
$\text{C}_3\text{H}_8\text{N}_2\text{O}$	<i>N,N'</i> -Dimethylurea	106.6	13.0
$\text{C}_3\text{H}_8\text{O}$	1-Propanol	-124.39	5.37
$\text{C}_3\text{H}_8\text{O}$	2-Propanol	-87.9	5.41
$\text{C}_3\text{H}_8\text{O}_2$	1,3-Propylene glycol	-27.7	7.1
$\text{C}_3\text{H}_8\text{O}_2$	Dimethoxymethane	-105.1	8.33
$\text{C}_3\text{H}_8\text{O}_3$	Glycerol	18.1	18.3
$\text{C}_3\text{H}_8\text{S}$	1-Propanethiol	-113.13	5.48
$\text{C}_3\text{H}_8\text{S}$	2-Propanethiol	-130.5	5.74
$\text{C}_3\text{H}_8\text{S}$	Ethyl methyl sulfide	-105.93	9.76
$\text{C}_3\text{H}_9\text{N}$	Propylamine	-84.75	10.97
$\text{C}_3\text{H}_9\text{N}$	Isopropylamine	-95.13	7.33
$\text{C}_3\text{H}_9\text{N}$	Trimethylamine	-117.1	7
$\text{C}_3\text{H}_9\text{NO}$	3-Amino-1-propanol	12.4	19.7
C_4F_8	Perfluorocyclobutane	-40.19	2.77
C_4F_{10}	Perfluorobutane	-129.1	7.66
$\text{C}_4\text{H}_2\text{O}_3$	Maleic anhydride	52.56	13.60
$\text{C}_4\text{H}_4\text{N}_2$	Succinonitrile	57.98	3.70
$\text{C}_4\text{H}_4\text{N}_2$	Pyrazine	51.0	12.9
$\text{C}_4\text{H}_4\text{O}$	Furan	-85.61	3.80
$\text{C}_4\text{H}_4\text{O}_3$	Succinic anhydride	119	20.4
$\text{C}_4\text{H}_4\text{S}$	Thiophene	-38.21	5.07
$\text{C}_4\text{H}_5\text{N}$	Pyrrole	-23.39	7.91
C_4H_6	1,2-Butadiene	-136.2	6.96
C_4H_6	1,3-Butadiene	-108.91	7.98
C_4H_6	1-Butyne	-125.7	6.03
C_4H_6	2-Butyne	-32.2	9.23
$\text{C}_4\text{H}_6\text{O}$	Divinyl ether	-100.6	7.9
$\text{C}_4\text{H}_6\text{O}_2$	<i>cis</i> -Crotonic acid	15	12.6
$\text{C}_4\text{H}_6\text{O}_2$	<i>trans</i> -Crotonic acid	71.5	13.0
$\text{C}_4\text{H}_6\text{O}_2$	γ -Butyrolactone	-43.61	9.57
$\text{C}_4\text{H}_6\text{O}_3$	Acetic anhydride	-74.1	10.5
$\text{C}_4\text{H}_6\text{O}_4$	Succinic acid	187.9	32.4
$\text{C}_4\text{H}_6\text{O}_4$	Dimethyl oxalate	54.8	21.1

ENTHALPY OF FUSION (continued)

Molecular formula	Name	$t_m/^{\circ}\text{C}$	$\Delta_{\text{fus}}H/\text{kJ mol}^{-1}$
C ₄ H ₈	1-Butene	-185.34	3.96
C ₄ H ₈	<i>cis</i> -2-Butene	-138.88	7.31
C ₄ H ₈	<i>trans</i> -2-Butene	-105.52	9.76
C ₄ H ₈	Isobutene	-140.7	5.92
C ₄ H ₈	Cyclobutane	-90.7	1.09
C ₄ H ₈	Methylcyclopropane	-177.6	2.8
C ₄ H ₈ O	Butanal	-96.86	10.77
C ₄ H ₈ O	2-Butanone	-86.64	8.39
C ₄ H ₈ O	Tetrahydrofuran	-108.44	8.54
C ₄ H ₈ O ₂	Butanoic acid	-5.1	11.59
C ₄ H ₈ O ₂	Ethyl acetate	-83.8	10.48
C ₄ H ₈ O ₂	1,4-Dioxane	11.85	12.84
C ₄ H ₈ S	Tetrahydrothiophene	-96.2	7.35
C ₄ H ₉ Br	1-Bromobutane	-112.6	9.23
C ₄ H ₉ Br	2-Bromobutane, (\pm)	-112.65	6.89
C ₄ H ₉ Cl	2-Chloro-2-methylpropane	-25.60	2.07
C ₄ H ₉ N	Pyrrolidine	-57.79	8.58
C ₄ H ₉ NO	Morpholine	-4.8	14.5
C ₄ H ₁₀	Butane	-138.3	4.66
C ₄ H ₁₀	Isobutane	-159.4	4.54
C ₄ H ₁₀ O	1-Butanol	-88.6	9.37
C ₄ H ₁₀ O	2-Butanol	-88.5	5.97
C ₄ H ₁₀ O	2-Methyl-1-propanol	-101.9	6.32
C ₄ H ₁₀ O	2-Methyl-2-propanol	25.69	6.70
C ₄ H ₁₀ O	Diethyl ether	-116.2	7.19
C ₄ H ₁₀ O ₂	1,4-Butanediol	20.4	18.70
C ₄ H ₁₀ O ₂	Ethylene glycol dimethyl ether	-69.20	12.6
C ₄ H ₁₀ S	1-Butanethiol	-115.7	10.46
C ₄ H ₁₀ S	Diethyl sulfide	-103.91	10.90
C ₄ H ₁₁ N	<i>tert</i> -Butylamine	-66.94	0.882
C ₄ H ₁₂ Pb	Tetramethyl lead	-30.2	10.80
C ₄ H ₁₂ Si	Tetramethylsilane	-99.06	6.87
C ₄ H ₁₂ Sn	Tetramethylstannane	-55.1	9.30
C ₅ H ₄ O ₂	Furfural	-38.1	14.37
C ₅ H ₅ N	Pyridine	-41.70	8.28
C ₅ H ₆ O	2-Methylfuran	-91.3	8.55
C ₅ H ₆ O ₂	Furfuryl alcohol	-14.6	13.13
C ₅ H ₈	<i>cis</i> -1,3-Pentadiene	-140.8	5.64
C ₅ H ₈	<i>trans</i> -1,3-Pentadiene	-87.4	7.14
C ₅ H ₈	1,4-Pentadiene	-148.2	6.12
C ₅ H ₈	2-Methyl-1,3-butadiene	-145.9	4.93
C ₅ H ₈	Cyclopentene	-135.0	3.36
C ₅ H ₈	Spiropentane	-107.0	6.43
C ₅ H ₈ O ₂	Methyl methacrylate	-47.55	14.4
C ₅ H ₈ O ₃	4-Oxopentanoic acid	33	9.22
C ₅ H ₈ O ₄	Glutaric acid	97.8	20.3
C ₅ H ₉ N	Pentanenitrile	-96.2	9
C ₅ H ₁₀	1-Pentene	-165.12	5.94
C ₅ H ₁₀	<i>cis</i> -2-Pentene	-151.36	7.11
C ₅ H ₁₀	<i>trans</i> -2-Pentene	-140.21	8.35
C ₅ H ₁₀	2-Methyl-1-butene	-137.53	7.91
C ₅ H ₁₀	3-Methyl-1-butene	-168.43	5.36
C ₅ H ₁₀	2-Methyl-2-butene	-133.72	7.60
C ₅ H ₁₀	Cyclopentane	-93.4	0.61
C ₅ H ₁₀ O	Cyclopentanol	-17.5	1.535
C ₅ H ₁₀ O	2-Pentanone	-76.8	10.63
C ₅ H ₁₀ O	3-Pentanone	-39	11.59
C ₅ H ₁₀ O	3-Methyl-2-butanone	-93.1	9.34

ENTHALPY OF FUSION (continued)

Molecular formula	Name	$t_m/^{\circ}\text{C}$	$\Delta_{\text{fus}}H/\text{kJ mol}^{-1}$
C ₅ H ₁₀ O	Tetrahydropyran	-49.1	1.8
C ₅ H ₁₀ O ₂	Pentanoic acid	-33.6	14.16
C ₅ H ₁₁ Br	1-Bromopentane	-88.0	14.37
C ₅ H ₁₁ N	Cyclopentylamine	-82.7	8.31
C ₅ H ₁₁ N	Piperidine	-11.02	14.85
C ₅ H ₁₂	Pentane	-129.67	8.40
C ₅ H ₁₂	Isopentane	-159.77	5.15
C ₅ H ₁₂	Neopentane	-16.4	3.10
C ₅ H ₁₂ O	1-Pentanol	-77.6	10.50
C ₅ H ₁₂ O	2-Methyl-2-butanol	-9.1	4.46
C ₅ H ₁₂ O	Butyl methyl ether	-115.7	10.85
C ₅ H ₁₂ O	Methyl <i>tert</i> -butyl ether	-108.6	7.60
C ₅ H ₁₂ O ₄	Pentaerythritol	258	4.8
C ₅ H ₁₂ S	1-Pentanethiol	-75.65	17.53
C ₆ Cl ₆	Hexachlorobenzene	228.83	25.2
C ₆ F ₆	Hexafluorobenzene	5.03	11.59
C ₆ F ₁₄	Perfluorohexane	-88.2	6.84
C ₆ HF ₅	Pentafluorobenzene	-47.4	10.87
C ₆ HF ₅ O	Pentafluorophenol	37.5	16.41
C ₆ H ₂ F ₄	1,2,3,5-Tetrafluorobenzene	-46.25	6.36
C ₆ H ₂ F ₄	1,2,4,5-Tetrafluorobenzene	3.88	15.05
C ₆ H ₃ Cl ₃	1,2,3-Trichlorobenzene	51.3	17.9
C ₆ H ₃ Cl ₃	1,2,4-Trichlorobenzene	16.92	16.4
C ₆ H ₃ Cl ₃	1,3,5-Trichlorobenzene	62.8	18.1
C ₆ H ₃ N ₃ O ₆	1,3,5-Trinitrobenzene	122.9	15.4
C ₆ H ₄ ClNO ₂	1-Chloro-2-nitrobenzene	32.1	17.9
C ₆ H ₄ ClNO ₂	1-Chloro-3-nitrobenzene	44.4	19.4
C ₆ H ₄ ClNO ₂	1-Chloro-4-nitrobenzene	82	14.1
C ₆ H ₄ Cl ₂	<i>o</i> -Dichlorobenzene	-17.0	12.4
C ₆ H ₄ Cl ₂	<i>m</i> -Dichlorobenzene	-24.8	12.6
C ₆ H ₄ Cl ₂	<i>p</i> -Dichlorobenzene	53.09	18.19
C ₆ H ₄ F ₂	<i>o</i> -Difluorobenzene	-47.1	11.05
C ₆ H ₄ F ₂	<i>m</i> -Difluorobenzene	-69.12	8.58
C ₆ H ₄ O ₂	<i>p</i> -Benzoquinone	115	18.5
C ₆ H ₅ Br	Bromobenzene	-30.72	10.70
C ₆ H ₅ Cl	Chlorobenzene	-45.31	9.6
C ₆ H ₅ ClO	<i>o</i> -Chlorophenol	9.4	13.0
C ₆ H ₅ ClO	<i>m</i> -Chlorophenol	32.6	14.9
C ₆ H ₅ ClO	<i>p</i> -Chlorophenol	42.8	14.1
C ₆ H ₅ F	Fluorobenzene	-42.18	11.31
C ₆ H ₅ I	Iodobenzene	-31.3	9.75
C ₆ H ₅ NO	Nitrosobenzene	67	31.0
C ₆ H ₅ NO ₂	Nitrobenzene	5.7	12.12
C ₆ H ₅ NO ₃	<i>o</i> -Nitrophenol	44.8	17.7
C ₆ H ₅ NO ₃	<i>m</i> -Nitrophenol	96.8	20.6
C ₆ H ₅ NO ₃	<i>p</i> -Nitrophenol	113.6	18.8
C ₆ H ₆	Benzene	5.49	9.87
C ₆ H ₆ CIN	<i>o</i> -Chloroaniline	-1.9	11.9
C ₆ H ₆ CIN	<i>m</i> -Chloroaniline	-10.28	10.15
C ₆ H ₆ CIN	<i>p</i> -Chloroaniline	70.5	20.0
C ₆ H ₆ N ₂ O ₂	<i>o</i> -Nitroaniline	71.0	16.1
C ₆ H ₆ N ₂ O ₂	<i>m</i> -Nitroaniline	113.4	23.6
C ₆ H ₆ N ₂ O ₂	<i>p</i> -Nitroaniline	147.5	21.2
C ₆ H ₆ O	Phenol	40.89	11.51
C ₆ H ₆ O ₂	<i>p</i> -Hydroquinone	172.4	26.8
C ₆ H ₆ O ₂	Pyrocatechol	104.6	22.8
C ₆ H ₆ O ₂	Resorcinol	109.4	20.4
C ₆ H ₆ S	Benzenethiol	-14.93	11.48

ENTHALPY OF FUSION (continued)

Molecular formula	Name	$t_m/^{\circ}\text{C}$	$\Delta_{\text{fus}}H/\text{kJ mol}^{-1}$
C ₆ H ₇ N	Aniline	-6.02	10.54
C ₆ H ₇ N	2-Methylpyridine	-66.68	9.72
C ₆ H ₇ N	3-Methylpyridine	-18.14	14.18
C ₆ H ₇ N	4-Methylpyridine	3.67	12.58
C ₆ H ₈ N ₂	<i>o</i> -Phenylenediamine	102.1	23.1
C ₆ H ₈ N ₂	<i>m</i> -Phenylenediamine	66.0	15.57
C ₆ H ₈ N ₂	<i>p</i> -Phenylenediamine	141.1	23.8
C ₆ H ₈ N ₂	Phenylhydrazine	20.6	14.05
C ₆ H ₁₀	Cyclohexene	-103.5	3.29
C ₆ H ₁₀ O	Cyclohexanone	-27.9	1.328
C ₆ H ₁₀ O ₂	2-Oxepanone	-1.0	13.83
C ₆ H ₁₀ O ₄	Adipic acid	152.5	36.3
C ₆ H ₁₁ Cl	Chlorocyclohexane	-43.81	2.043
C ₆ H ₁₂	1-Hexene	-139.76	9.35
C ₆ H ₁₂	<i>cis</i> -2-Hexene	-141.11	8.88
C ₆ H ₁₂	2,3-Dimethyl-2-butene	-74.19	6.45
C ₆ H ₁₂	Cyclohexane	6.59	2.68
C ₆ H ₁₂	Methylcyclopentane	-142.42	6.93
C ₆ H ₁₂ O	Hexanal	-56	13.3
C ₆ H ₁₂ O	2-Hexanone	-55.5	14.9
C ₆ H ₁₂ O	3-Hexanone	-55.4	13.49
C ₆ H ₁₂ O	Cyclohexanol	25.93	1.78
C ₆ H ₁₂ O ₃	Paraldehyde	12.6	13.5
C ₆ H ₁₃ Br	1-Bromohexane	-83.7	18.1
C ₆ H ₁₃ N	Cyclohexylamine	-17.8	17.5
C ₆ H ₁₄	Hexane	-95.35	13.08
C ₆ H ₁₄	2-Methylpentane	-153.6	6.27
C ₆ H ₁₄	3-Methylpentane	-162.90	5.30
C ₆ H ₁₄	2,2-Dimethylbutane	-98.8	0.58
C ₆ H ₁₄	2,3-Dimethylbutane	-128.10	0.79
C ₆ H ₁₄ O	1-Hexanol	-47.4	15.38
C ₆ H ₁₄ O	Dipropyl ether	-114.8	10.8
C ₆ H ₁₄ O	Diisopropyl ether	-85.4	12.04
C ₆ H ₁₄ O ₂	1,6-Hexanediol	41.5	22.2
C ₇ F ₈	Perfluorotoluene	-65.49	11.54
C ₇ F ₁₆	Perfluoroheptane	-51.2	6.95
C ₇ H ₃ F ₅	2,3,4,5,6-Pentafluorotoluene	-29.78	13.1
C ₇ H ₅ ClO	Benzoyl chloride	-0.4	19.2
C ₇ H ₅ ClO ₂	<i>o</i> -Chlorobenzoic acid	140.2	25.6
C ₇ H ₅ N	Benzonitrile	-13.99	9.1
C ₇ H ₅ N ₃ O ₆	2,4,6-Trinitrotoluene	80.5	22.9
C ₇ H ₆ O ₂	Benzoic acid	122.35	18.02
C ₇ H ₆ O ₃	<i>o</i> -Hydroxybenzoic acid	159.0	14.2
C ₇ H ₇ Cl	<i>o</i> -Chlorotoluene	-35.8	9.6
C ₇ H ₇ NO	Benzamide	127.3	19.5
C ₇ H ₇ NO ₂	<i>p</i> -Nitrotoluene	51.63	16.81
C ₇ H ₈	Toluene	-94.95	6.64
C ₇ H ₈ O	<i>o</i> -Cresol	31.03	15.82
C ₇ H ₈ O	<i>m</i> -Cresol	12.24	10.71
C ₇ H ₈ O	<i>p</i> -Cresol	34.77	12.71
C ₇ H ₈ O	Benzyl alcohol	-15.4	8.97
C ₇ H ₈ O	Anisole	-37.13	12.9
C ₇ H ₉ N	<i>o</i> -Methylaniline	-14.41	11.66
C ₇ H ₉ N	<i>m</i> -Methylaniline	-31.3	7.9
C ₇ H ₉ N	<i>p</i> -Methylaniline	43.6	18.9
C ₇ H ₁₄	1-Heptene	-118.9	12.41
C ₇ H ₁₄	Cycloheptane	-8.46	1.88
C ₇ H ₁₄	Methylcyclohexane	-126.6	6.75

ENTHALPY OF FUSION (continued)

Molecular formula	Name	$t_m/^{\circ}\text{C}$	$\Delta_{\text{fus}}H/\text{kJ mol}^{-1}$
C ₇ H ₁₄ O	1-Heptanal	-43.4	23.2
C ₇ H ₁₄ O	Cycloheptanol	7.2	1.60
C ₇ H ₁₄ O ₂	Heptanoic acid	-7.17	15.13
C ₇ H ₁₅ Br	1-Bromoheptane	-56.1	21.8
C ₇ H ₁₆	Heptane	-90.55	14.03
C ₇ H ₁₆	2-Methylhexane	-118.2	9.19
C ₇ H ₁₆	3-Ethylpentane	-118.55	9.55
C ₇ H ₁₆	2,2-Dimethylpentane	-123.7	5.82
C ₇ H ₁₆	2,4-Dimethylpentane	-119.2	6.85
C ₇ H ₁₆	3,3-Dimethylpentane	-134.4	6.85
C ₇ H ₁₆	2,2,3-Trimethylbutane	-24.6	2.26
C ₇ H ₁₆ O	1-Heptanol	-33.2	18.17
C ₈ H ₈	Styrene	-30.65	10.9
C ₈ H ₈ O ₂	<i>o</i> -Toluic acid	103.5	19.5
C ₈ H ₈ O ₂	<i>m</i> -Toluic acid	109.9	15.7
C ₈ H ₈ O ₂	<i>p</i> -Toluic acid	179.6	22.7
C ₈ H ₈ O ₂	Benzenecetic acid	76.5	16.3
C ₈ H ₈ O ₂	Methyl benzoate	-12.4	9.74
C ₈ H ₁₀	Ethylbenzene	-94.96	9.18
C ₈ H ₁₀	<i>o</i> -Xylene	-25.2	13.6
C ₈ H ₁₀	<i>m</i> -Xylene	-47.8	11.6
C ₈ H ₁₀	<i>p</i> -Xylene	13.25	17.12
C ₈ H ₁₀ O	2,3-Xylenol	72.5	21.0
C ₈ H ₁₀ O	2,5-Xylenol	74.8	23.4
C ₈ H ₁₀ O	2,6-Xylenol	45.8	18.9
C ₈ H ₁₀ O	3,4-Xylenol	65.1	18.1
C ₈ H ₁₀ O	3,5-Xylenol	63.4	17.4
C ₈ H ₁₆	1-Octene	-101.7	15.31
C ₈ H ₁₆	Cyclooctane	14.59	2.41
C ₈ H ₁₆	Ethylcyclohexane	-111.3	8.33
C ₈ H ₁₆	1,1-Dimethylcyclohexane	-33.3	2.07
C ₈ H ₁₆	<i>cis</i> -1,2-Dimethylcyclohexane	-49.8	1.64
C ₈ H ₁₆	<i>trans</i> -1,2-Dimethylcyclohexane	-88.15	10.49
C ₈ H ₁₆	<i>cis</i> -1,3-Dimethylcyclohexane	-75.53	10.82
C ₈ H ₁₆	<i>trans</i> -1,3-Dimethylcyclohexane	-90.07	9.87
C ₈ H ₁₆	<i>cis</i> -1,4-Dimethylcyclohexane	-87.39	9.31
C ₈ H ₁₆	<i>trans</i> -1,4-Dimethylcyclohexane	-36.93	12.33
C ₈ H ₁₆ O ₂	Octanoic acid	16.5	21.35
C ₈ H ₁₇ Br	1-Bromooctane	-55.0	24.7
C ₈ H ₁₈	Octane	-56.82	20.73
C ₈ H ₁₈	2-Methylheptane	-109.02	11.92
C ₈ H ₁₈	3-Methylheptane	-120.48	11.69
C ₈ H ₁₈	4-Methylheptane	-121.0	10.8
C ₈ H ₁₈	2,2,4-Trimethylpentane	-107.3	9.20
C ₈ H ₁₈ O	1-Octanol	-14.8	23.7
C ₉ H ₇ N	Quinoline	-14.78	10.66
C ₉ H ₇ N	Isoquinoline	26.47	13.54
C ₉ H ₈	Indene	-1.5	10.20
C ₉ H ₁₀	Indan	-51.38	8.60
C ₉ H ₁₂	Propylbenzene	-99.6	9.27
C ₉ H ₁₂	Isopropylbenzene	-96.02	7.33
C ₉ H ₁₂	<i>o</i> -Ethyltoluene	-79.83	9.96
C ₉ H ₁₂	<i>m</i> -Ethyltoluene	-95.6	7.6
C ₉ H ₁₂	<i>p</i> -Ethyltoluene	-62.35	12.7
C ₉ H ₁₂	1,2,3-Trimethylbenzene	-25.4	8.18
C ₉ H ₁₂	1,2,4-Trimethylbenzene	-43.77	13.19
C ₉ H ₁₂	1,3,5-Trimethylbenzene	-44.72	9.51
C ₉ H ₁₈	Propylcyclohexane	-94.9	10.37

ENTHALPY OF FUSION (continued)

Molecular formula	Name	$t_m/^{\circ}\text{C}$	$\Delta_{\text{fus}}H/\text{kJ mol}^{-1}$
$\text{C}_9\text{H}_{18}\text{O}$	Nonanal	-19.3	30.5
$\text{C}_9\text{H}_{18}\text{O}$	5-Nonanone	-3.8	24.93
$\text{C}_9\text{H}_{18}\text{O}_2$	Nonanoic acid	12.4	19.82
C_9H_{20}	Nonane	-53.46	15.47
C_9H_{20}	3,3-Diethylpentane	-33.1	10.09
C_9H_{20}	2,2,3,3-Tetramethylpentane	-9.75	2.33
C_9H_{20}	2,2,4,4-Tetramethylpentane	-66.54	9.74
$\text{C}_{10}\text{H}_7\text{Br}$	1-Bromonaphthalene	6.1	15.2
$\text{C}_{10}\text{H}_7\text{Br}$	2-Bromonaphthalene	55.9	14.4
$\text{C}_{10}\text{H}_7\text{Cl}$	1-Chloronaphthalene	-2.5	12.9
$\text{C}_{10}\text{H}_7\text{Cl}$	2-Chloronaphthalene	58.0	14.0
C_{10}H_8	Naphthalene	80.26	19.01
$\text{C}_{10}\text{H}_8\text{O}$	1-Naphthol	95.0	23.1
$\text{C}_{10}\text{H}_8\text{O}$	2-Naphthol	121.5	18.1
$\text{C}_{10}\text{H}_{14}$	Butylbenzene	-87.85	11.22
$\text{C}_{10}\text{H}_{14}$	1-Isopropyl-4-methylbenzene	-67.94	9.66
$\text{C}_{10}\text{H}_{14}$	1,2,4,5-Tetramethylbenzene	79.3	21
$\text{C}_{10}\text{H}_{14}\text{O}$	Thymol	49.5	21.3
$\text{C}_{10}\text{H}_{18}$	<i>cis</i> -Decahydronaphthalene	-42.9	9.49
$\text{C}_{10}\text{H}_{18}$	<i>trans</i> -Decahydronaphthalene	-30.4	14.41
$\text{C}_{10}\text{H}_{18}\text{O}_4$	Sebacic acid	130.9	40.8
$\text{C}_{10}\text{H}_{20}$	1-Decene	-66.3	13.81
$\text{C}_{10}\text{H}_{20}$	Butylcyclohexane	-74.73	14.16
$\text{C}_{10}\text{H}_{20}\text{O}$	Decanal	-4.0	34.5
$\text{C}_{10}\text{H}_{20}\text{O}_2$	Decanoic acid	31.4	27.8
$\text{C}_{10}\text{H}_{22}$	Decane	-29.6	28.72
$\text{C}_{10}\text{H}_{22}\text{O}$	1-Decanol	6.9	43
$\text{C}_{11}\text{H}_{10}$	1-Methylnaphthalene	-30.43	6.95
$\text{C}_{11}\text{H}_{10}$	2-Methylnaphthalene	34.6	12.13
$\text{C}_{11}\text{H}_{24}$	Undecane	-25.5	22.2
C_{12}H_8	Acenaphthylene	91.8	6.9
$\text{C}_{12}\text{H}_6\text{N}$	Carbazole	246.3	24.1
$\text{C}_{12}\text{H}_{10}$	Acenaphthene	93.4	21.49
$\text{C}_{12}\text{H}_{10}$	Biphenyl	68.93	18.57
$\text{C}_{12}\text{H}_{10}\text{N}_2$	Azobenzene	67.88	22.52
$\text{C}_{12}\text{H}_{10}\text{N}_2\text{O}$	<i>trans</i> -Azoxybenzene	34.6	17.9
$\text{C}_{12}\text{H}_{10}\text{O}$	Diphenyl ether	26.87	17.22
$\text{C}_{12}\text{H}_{11}\text{N}$	Diphenylamine	53.2	18.5
$\text{C}_{12}\text{H}_{16}$	Cyclohexylbenzene	7.07	15.6
$\text{C}_{12}\text{H}_{18}$	Hexamethylbenzene	165.5	20.6
$\text{C}_{12}\text{H}_{24}$	1-Dodecene	-35.2	19.9
$\text{C}_{12}\text{H}_{24}\text{O}_2$	Dodecanoic acid	43.8	36.3
$\text{C}_{12}\text{H}_{26}$	Dodecane	-9.57	36.8
$\text{C}_{12}\text{H}_{26}\text{O}$	1-Dodecanol	23.9	40.2
$\text{C}_{13}\text{H}_{10}$	9 <i>H</i> -Fluorene	114.77	19.58
$\text{C}_{13}\text{H}_{10}\text{O}$	Benzophenone	47.9	18.19
$\text{C}_{13}\text{H}_{12}$	Diphenylmethane	25.4	18.6
$\text{C}_{13}\text{H}_{28}$	Tridecane	-5.4	28.50
$\text{C}_{13}\text{H}_{28}\text{O}$	1-Tridecanol	31.7	41.4
$\text{C}_{14}\text{H}_{10}$	Anthracene	215.76	29.4
$\text{C}_{14}\text{H}_{10}$	Phenanthrene	99.24	16.46
$\text{C}_{14}\text{H}_{10}\text{O}_2$	Benzil	94.87	23.5
$\text{C}_{14}\text{H}_{12}$	<i>trans</i> -Stilbene	124.2	27.7
$\text{C}_{14}\text{H}_{12}\text{O}_2$	α -Phenylbenzeneacetic acid	147.29	31.3
$\text{C}_{14}\text{H}_{28}\text{O}_2$	Tetradecanoic acid	54.2	45.1
$\text{C}_{14}\text{H}_{30}$	Tetradecane	5.82	45.07
$\text{C}_{14}\text{H}_{30}\text{O}$	1-Tetradecanol	38.2	25.1*
$\text{C}_{15}\text{H}_{32}$	Pentadecane	9.95	34.6

ENTHALPY OF FUSION (continued)

Molecular formula	Name	$t_m/^{\circ}\text{C}$	$\Delta_{\text{fus}}H/\text{kJ mol}^{-1}$
$\text{C}_{16}\text{H}_{10}$	Fluoranthene	110.19	18.69
$\text{C}_{16}\text{H}_{10}$	Pyrene	150.62	17.36
$\text{C}_{16}\text{H}_{32}\text{O}_2$	Hexadecanoic acid	62.5	53.7
$\text{C}_{16}\text{H}_{34}$	Hexadecane	18.12	53.36
$\text{C}_{16}\text{H}_{34}\text{O}$	1-Hexadecanol	49.2	33.6
$\text{C}_{17}\text{H}_{36}$	Heptadecane	22.0	40.16
$\text{C}_{18}\text{H}_{12}$	Benz[a]anthracene	160.5	21.4
$\text{C}_{18}\text{H}_{12}$	Benzo[c]phenanthrene	68	16.3
$\text{C}_{18}\text{H}_{12}$	Chrysene	255.5	26.2
$\text{C}_{18}\text{H}_{12}$	Triphenylene	197.8	24.74
$\text{C}_{18}\text{H}_{14}$	<i>o</i> -Terphenyl	56.20	17.19
$\text{C}_{18}\text{H}_{14}$	<i>p</i> -Terphenyl	213.9	35.3
$\text{C}_{18}\text{H}_{15}\text{N}$	Triphenylamine	126.5	24.9
$\text{C}_{18}\text{H}_{36}\text{O}_2$	Stearic acid	69.3	61.2
$\text{C}_{18}\text{H}_{38}$	Octadecane	28.2	61.7
$\text{C}_{18}\text{H}_{38}\text{O}$	1-Octadecanol	57.9	45
$\text{C}_{19}\text{H}_{40}$	Nonadecane	32.0	45.8
$\text{C}_{20}\text{H}_{12}$	Perylene	277.76	31.9
$\text{C}_{20}\text{H}_{12}$	Benzo[a]pyrene	181.1	17.3
$\text{C}_{20}\text{H}_{12}$	Benzo[e]pyrene	181.4	16.6
$\text{C}_{20}\text{H}_{14}$	2,2'-Binaphthalene	187.9	38.9
$\text{C}_{20}\text{H}_{42}$	Eicosane	36.6	69.9
$\text{C}_{20}\text{H}_{42}\text{O}$	1-Eicosanol	65.4	42
$\text{C}_{24}\text{H}_{12}$	Coronene	437.4	19.2