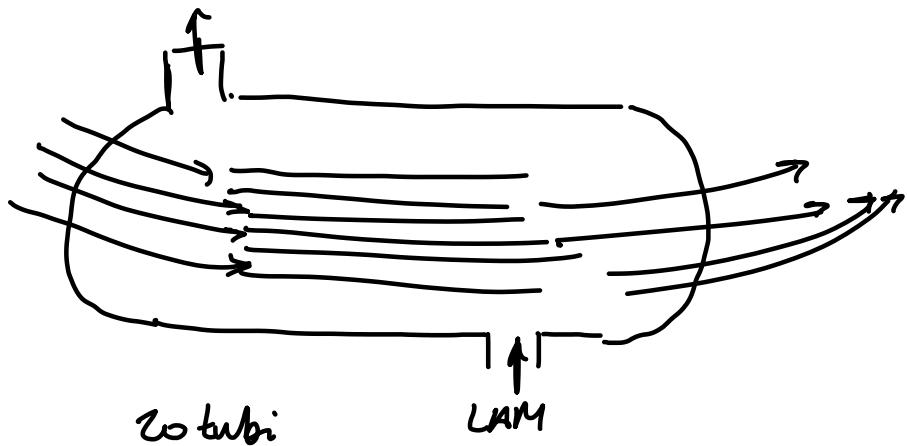


Esercitazione 14

Esercizio 3 Feb 2023



$$k = \lambda = 52 \frac{W}{mK}$$

$$\dot{M}_s = 10 \text{ kg/s}$$

$$D_1 = 10 \text{ mm}$$

$$\rho, C_p, \lambda, \mu$$

$$S = 3 \text{ mm}$$

$$T_{1in} = 120^\circ\text{C}$$

$$D_o = 12 \text{ mm}$$

$$C_{H_2O} \quad \dot{M}_a = 3 \text{ kg/s}$$

$$Nu = 3,66$$

$$Nu_D = 0,023 Re^{0,8} Pr^{0,3} \quad T_{ain} = 5^\circ\text{C} \quad T_{a,out} = 60^\circ\text{C}$$

$$Nu_D = 0,023 Re^{0,8} Pr^{0,3}$$

$$h_c = 5000 \frac{W}{m^2 K} \quad ? T_{s,out}$$

$$? h_s \quad ? V_i \rightarrow \text{Area Interna}$$

$$?\text{Lunghezza tubi}$$

$$?\Delta S$$

$$Re = \frac{\rho_i w L_c}{\mu_s} = \frac{\rho_i w D}{\mu_s} =$$

$$Pr = \frac{C_p \mu_s}{k}$$

$$\dot{Q}_{H_2O}^{\leftarrow} = \dot{m}_2 c_a (T_{a\text{out}} - T_{a\text{in}})$$

$$\dot{\vec{Q}}_1 = \dot{\vec{Q}}_a \quad T_{1\text{out}} = T_{1\text{in}} - \frac{\dot{Q}_a}{\dot{m}_2 c_1}$$

84°C

$$\dot{Q} = U A \Delta T_{m,\text{em}}$$

$$W_a = \frac{\dot{m}}{\rho A} \underbrace{n_{\text{TUB}}}_{\pi D_i^2 / 4}$$

$$Re_b = \frac{\rho_i U_i D_i}{\mu} = 127332$$

$Re > 2300$

\Rightarrow flusso turbolente

$$\text{viscous } Nu_b = 0,023 Re^{0,8} Pr^{0,3}$$

$$Pr = \frac{\mu_2 C_{Pr}}{\lambda_1} = 8,35$$

$$Nu_b = 527 = \frac{h D}{\lambda} \Rightarrow h_i$$

$$\dot{Q}_{\text{TUBO}} = [U_i A_{\text{inf}}] \Delta T_{m,\text{tg}} = \frac{\Delta T_{m,\text{em}}}{R_{\text{TOT}}}$$

$$R_{\text{TOT}} = \frac{\zeta}{h_i A_i} + \frac{\ln \frac{D_e/D_i}{2\pi k L}}{2\pi k L} + \frac{\zeta}{h_e A_e}$$

$$\dot{Q} = N_{\text{TUBI}} \cdot \dot{Q}_{\text{ITUBO}}$$

$$\frac{\Delta T_{m,lg}}{R_{\text{TOT}}} = N_{\text{TUBI}} \cdot \frac{\Delta T_{m,lg}}{R_{\text{TOT}} \cdot s_{\text{TUBO}}}$$

$$R_{\text{TOT}} = \frac{R_{\text{TOT} \cdot s_{\text{TUBO}}}}{N_{\text{TUBI}}} \rightarrow R_{\text{TOT}} = \frac{1}{h_i A_i N_{\text{TUBI}}} + \frac{\ln D_o/D_i}{2\pi L K N_{\text{TUBI}}} + \frac{1}{h_{ext} A_e N_{\text{TUBI}}}$$

↳ Con più tubi è più possibile il trasferimento di calore

$$\frac{1}{U} = R_{\text{TOT}} A_{ref}$$

↳ $A_{ref,tubi}$

$$\frac{1}{U_1} = R_{\text{TOT}} A_{ref,tubi} = \frac{1}{h_{in}} + \frac{\ln D_o/D_i \cdot r_i}{\lambda} + \frac{r_i}{h_{ext} r_e}$$

$$U_1 = \left(\frac{1}{U_1} \right)^{-1} = 2864,7 \frac{W}{m^2 K}$$

$$\Delta T_1 = 84^\circ - 5^\circ C$$

$$\Delta T_2 = 120 - 60 = 60^\circ$$

$$\Delta T_{m,lg} = \frac{\Delta T_1 - \Delta T_2}{\ln \frac{\Delta T_1}{\Delta T_2}} = 69,05 K$$

$$\dot{Q} = U A_i \Delta T_{m, \text{en}} \rightarrow \text{Area Interna}$$

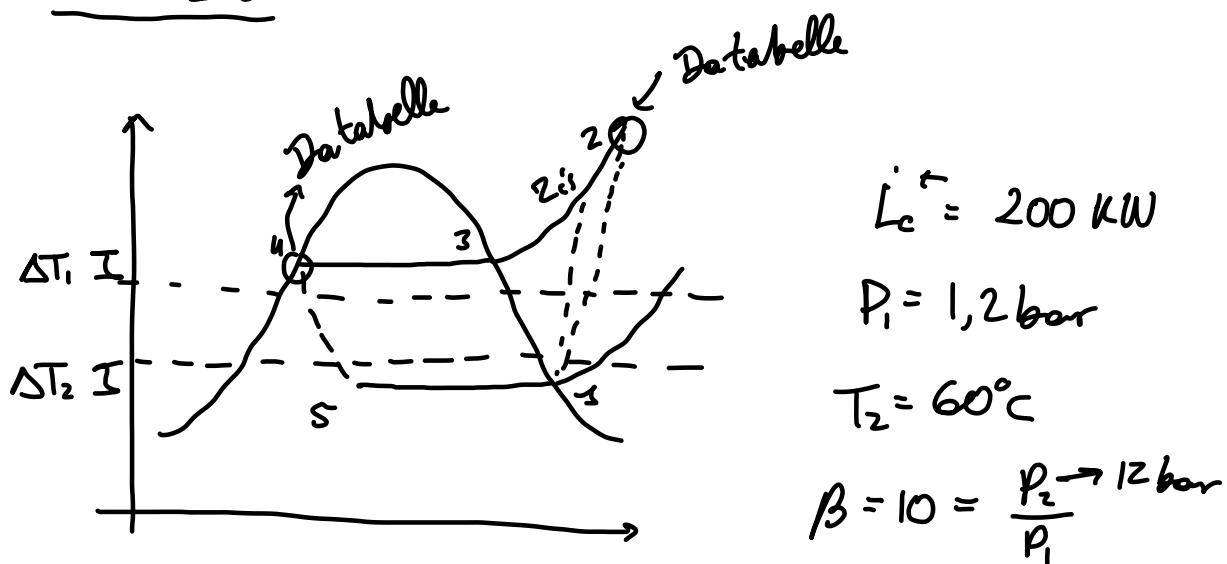
|
 $= \pi D_i L$

$$\Delta S_{\text{CALDO}} = m_1 \ln \frac{T_{1\text{OUT}}}{T_{1\text{IN}}}$$

$$\Delta S_{\text{FREDDO}} = m_2 \ln \frac{T_{2\text{OUT}}}{T_{2\text{IN}}}$$

$$\Delta S_{\text{SCAMBIO}} = \Delta S_{\text{CALDO}} - \Delta S_{\text{FREDDO}} = 42,99 \frac{W}{K}$$

Esercizio 2



? m ? S_{inr}

$$\Delta T_1 = \Delta T_2 = 10^\circ\text{C}$$

? $\eta_{\text{is,c}}$? X_s ? η_{II}

$$T_1 = -22,32^\circ\text{C} \quad V_1 = V_{vs} = 0,162 \frac{\text{m}^3}{\text{kg}}$$

$$h_1 = h_{vs} = 236,97 \text{ kJ/kg}$$

$$s_1 = 0,94775 \frac{kJ}{kgK} \quad T_{SORGENTE FREDDA} = -10,32^\circ C$$

② VAP SAT Tabella $T_2 \propto P_2$

$$h_2 = 289,64 \frac{kJ}{kg} \quad s_2 = 0,9614 \frac{kJ}{kgK}$$

$$\dot{L}_c^e = m(h_2 - h_1)$$

$$h_{2JS} = 285,18 \frac{kJ}{kg} \leftarrow \text{Databelle } P_2 \text{ e } s_{2JS} = s_2$$

$$\dot{L}_{C1S}^e = m(h_{2SS} - h_1) \quad \gamma_{ISC} = \frac{\dot{L}_{C1D}}{\dot{L}_c^e} = 0,915$$

PUNTO 4 \rightarrow LIQ SAT @ P_2

$$\bar{T}_4 = 46,29^\circ C \quad v_4 = 0,0008934 \frac{m^3}{kg} \quad T_{Calcolo} = 36,29^\circ C$$

$$h_4 = 117,77 \frac{kJ}{kg} \quad s_4 = 0,4244 \frac{kJ}{kgK}$$

PUNTO 5

$$h_5 = h_4 \quad P_5 = P_1$$

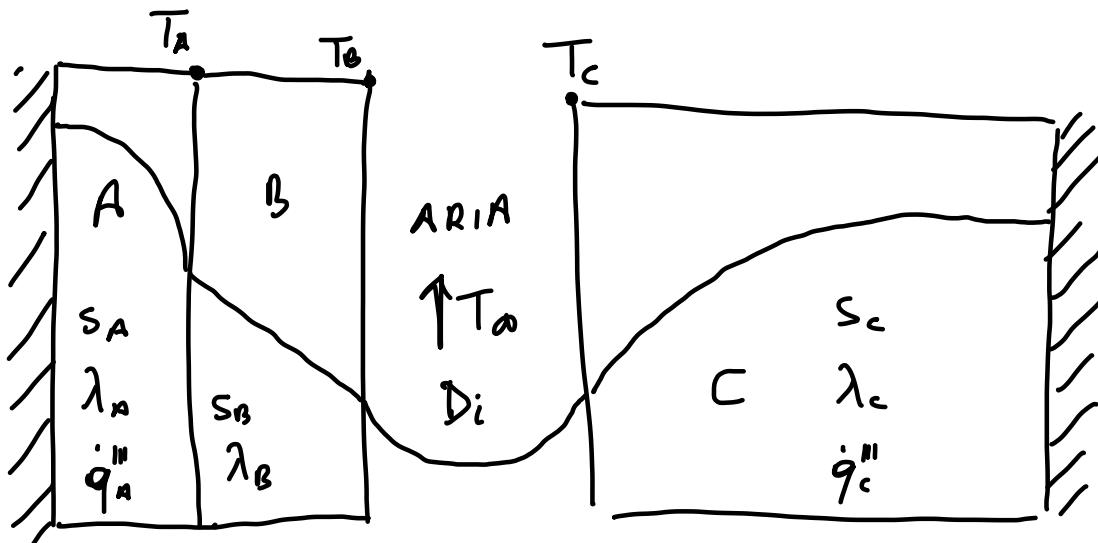
$$\chi_r = \frac{h_5 - h_{1S}}{h_{1S} - h_{2S}} = 0,44$$

$$\varepsilon_I = \frac{\dot{Q}}{\dot{L}_{C1}} = \frac{m(h_1 - h_5)}{\dot{L}_c^e} = 2,263$$

$$\Sigma_{\text{CARNOT}} = \frac{T_{\text{FREDDO}}}{T_{\text{CALDO}} - T_{\text{FREDDO}}} = 5,37 \quad \eta_{\text{II}} = \frac{\Sigma_{\text{I}}}{\Sigma_{\text{CARNOT}}} = 0,422$$

$$\Delta S_{\text{NET}} = \dot{S}_{\text{NET}} = \frac{\dot{Q}_c}{T_{\text{CALDO}}} - \frac{\dot{Q}_f}{T_{\text{FREDDO}}} \approx 0,373 \frac{W}{K}$$

Esercizio 3



In regime *No ciraggiamento*

$$?T_B \quad ?T_{\text{MAX in A}}$$

$$?T_A \quad ?T_{\text{MAX in B}}$$

$$?T_c$$

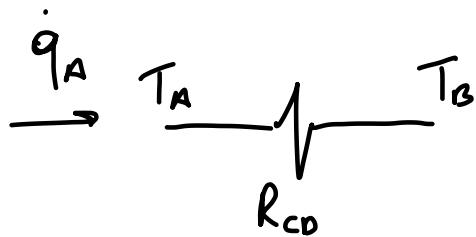
$$\dot{q}_{\text{TOT}} = \dot{q}_A + \dot{q}_C$$

$$\dot{q}_A = \dot{q}_A'' \cdot s_A = 750 \frac{W}{m^2}$$

$$\dot{q}_C = \dot{q}_C'' \cdot s_c = 2000 \frac{W}{m^2}$$

$$\dot{q}_B = h(T_B - T_\infty) \rightarrow T_B = 45,71^\circ C$$

$$\dot{q}_c = h(T_c - T_\infty) \rightarrow T_c = 63,57^\circ C$$



$$\dot{q}_A = \frac{\overline{T_A - T_B}}{\frac{s_B}{k}}$$

$$T_B = 53,21^\circ C$$

$$T_{MAXC} = \frac{\dot{q}_c'''}{2\lambda_c} \cdot s_c^2 + T_c$$

$$\frac{d^2 T}{dx^2} = -\frac{\dot{q}'''}{\lambda_c} \rightarrow \frac{dT}{dx} = -\frac{\dot{q}'''}{k} x + c_1 \rightarrow q(x) = -k \frac{dT}{dx}$$

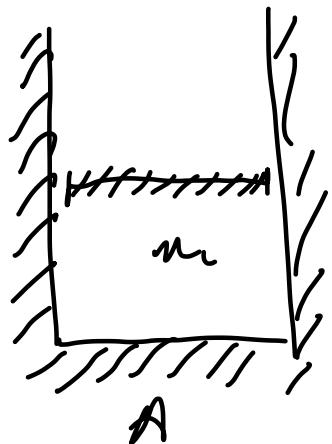
$$T(x) = -\frac{\dot{q}'''}{\lambda_c} \frac{x^2}{2} + c_1 x + c_2$$

$$T(s_c) = T_c - c_1 \cdot s_c + \frac{\dot{q}''' s_c^2}{2}$$

$$T_{MAXC}$$

Esercizio 4 → 2018

①



$$m = 10 \text{ kg}$$

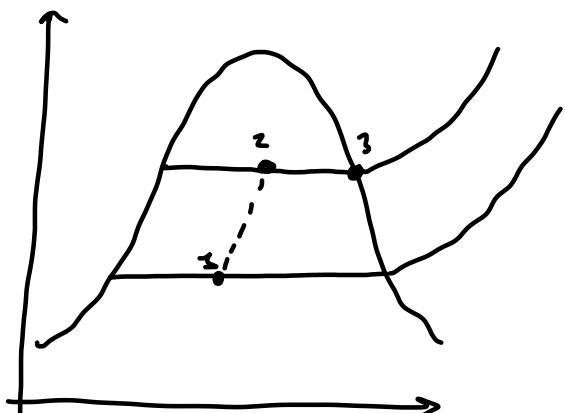
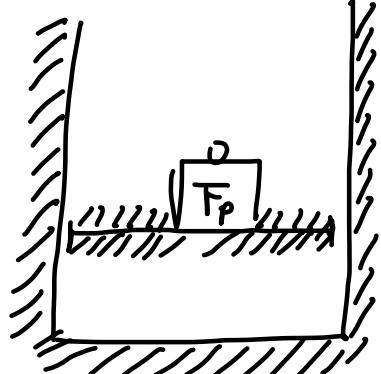
H₂O BIFASE

$$P_i = 1 \text{ bar}$$

$$V_i = 50 \text{ m}^3$$

$$F_p = 200000 \text{ N}$$

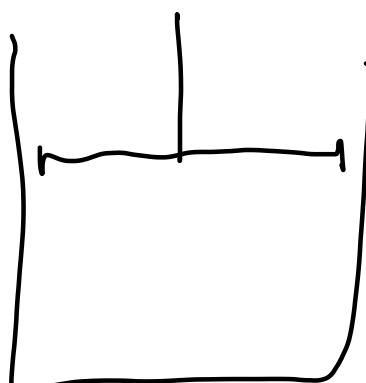
②



$$? X_2$$

$$? \Delta Y$$

$$? L^-$$



③ Pressione costante
e non isolata

Vapore saturo

$$? L^+$$

$$v_i = \frac{V_i}{m_i} = \frac{50 \text{ m}^3}{10 \text{ kg}} \rightarrow \text{Tab } @ P_i \rightarrow v_{ls} \text{ e } v_{vs}$$

$$X_i = \frac{v - v_{ls}}{v_{vs} - v_{ls}} = 0,59$$

$$\mu_i = (1-X) \mu_{ls} + X \mu_{vs} = 1643,72 \frac{\text{kg}}{\text{m}^3}$$

$$\Delta U_{12} = Q_{12}^{\leftarrow} - L_{12}^{\rightarrow} = -MP(v_2 - v_1) = M(u_2 - u_1)$$

$$X_2 = \frac{u_1 - u_{LS2} - P_2(v_{LS2} - V_1)}{u_{LS2} - u_{LS1} + P_2(V_{LS} - V_{LS})} = 0,64$$

Da questo

$$(1-X_2)u_{LS} + X_2u_{LS2} - u_1 = -P_2[(1-X_2)V_{LS} + X_2V_{LS2} - V_1]$$

$$v_2 = (1-X_2)V_{LS} + X_2v_{LS2} = V_2 = Mv_2$$

$$L_{12}^{\rightarrow} - \Delta U_{12} = -M \Delta u_{12} = -18337 \cdot 10^3 \text{ kJ}$$

$2 \rightarrow 3$

$$\Delta U_{23} = Q_{23} - L_{23} \rightarrow \Delta U_{23} = \Delta H_{23} - L_{23} \rightarrow L_{23} = \Delta H_{23} - \Delta U_{23}$$

$\hookrightarrow Q_{23} = H_3 - H_2$

↑
processo
isobaro

=

$$= M(h_3 - h_2) - M(u_3 - u_2)$$

$$= 650,1 \text{ kJ}$$