

L2 * N interacting atoms, $\mathbf{r}_i \rightarrow$ Positions

$$\left. \begin{array}{l} (1) \quad m_i \frac{\partial^2 \mathbf{r}_i}{\partial t^2} = \mathbf{F}_i, \quad i=1, \dots, N \\ (2) \quad \mathbf{F}_i = -\nabla V(\mathbf{r}_1, \dots, \mathbf{r}_N) \end{array} \right\} \Rightarrow \left. \begin{array}{l} \mathbf{r}_i(t) \\ \mathbf{v}_i(t) \end{array} \right\} \text{Trajectory}$$

* Approximations

• Born Oppenheimer

Atom dynamics \rightarrow Quantum Nature

- State defined by wave function $\Psi(\mathbf{R}_e, \mathbf{R}_N)$
 $\downarrow \quad \quad \downarrow$
electrons Nuclei

- Schrödinger eq: $i\hbar \frac{\partial}{\partial t} \Psi(\mathbf{R}_e, \mathbf{R}_N) = \hat{H} \Psi(\mathbf{R}_e, \mathbf{R}_N)$

- BO: separation of $\left\{ \begin{array}{l} \text{Fast motions of } e^- \\ \text{slow motions of nuclei.} \end{array} \right.$

$$\Psi = \underbrace{\Psi_n(\mathbf{R}_N, t)}_{\text{Nuclei}} \underbrace{\Psi_e(\mathbf{R}_e; \mathbf{R}_N)}_{\text{Electrons}}$$

- Time independent
- Move in the energy surface created by e^-
- \neq

- independent
- Instantaneously adjust to changes in \mathbf{R}_N
- Ground state
- $\mathbf{R}_N \rightarrow$ parametrically.

* Classical treatment

- Separation \gg de Broglie wave length $= \left(\frac{h^2}{2\pi m k_B T} \right)^{1/2}$

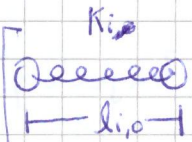
- The case for most atoms at room T

- Fails:

- Chemical reactions
- Tunneling
- Electronic excited states.

• Use of empirical interaction function, $V(r)$

"force field"



$$V(\vec{r}) = \sum_{\text{bonds}, i} \frac{K_i}{2} (l_i - l_{i,0})^2$$

$H_2O: l_0 = 0.9572 \text{ \AA}$

$K_i = 462750.4 \frac{\text{kJ mol}^{-1}}{\text{nm}^2}$

$$+ \sum_{\text{angle}, j} \frac{F_j}{2} (\theta - \theta_{j,0})^2$$

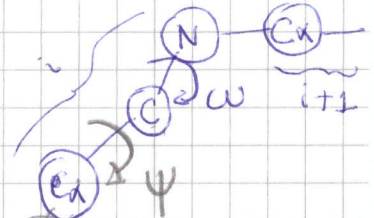
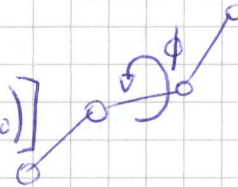


$H_2O:$

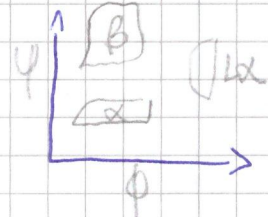
$\theta_0 = 104.520^\circ$

$F = 836.8 \frac{\text{kJ mol}^{-1}}{\text{nm}^2}$

$$+ \sum_{\text{dih}, k} \sum_n \frac{V_{k,n}}{2} [1 + \cos(n\phi - \phi_{k0})]$$



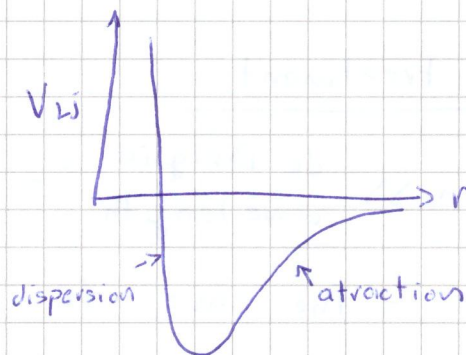
Ramachandran Plot



$$+ \sum_{ij} 4\epsilon_{ij} \left[\left(\frac{\sigma_{ij}}{r_{ij}} \right)^{12} - \left(\frac{\sigma_{ij}}{r_{ij}} \right)^6 \right]$$

Lennard-Jones Potential.

Van der Waals Int.

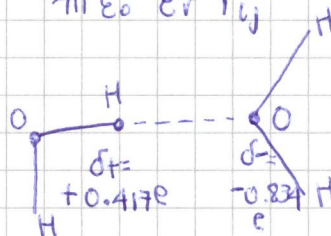


$\sigma = 3.1506 \text{ \AA}$

$\epsilon = 0.636 \text{ kJ/mol}$

$$+ \sum_{ij} \frac{q_i q_j}{4\pi \epsilon_0 \epsilon_r r_{ij}}$$

Electrostatic interactions



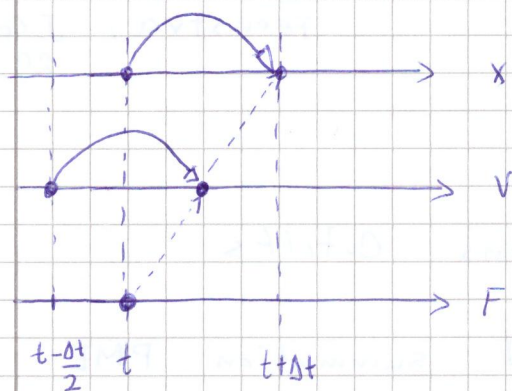
Hydrogen bond distance?

* Solving numerically Newton's eqs. of motion

Leap Frog Algorithm

$$V_i(t + \frac{\Delta t}{2}) = V_i(t - \frac{\Delta t}{2}) + \frac{F_i(t)}{m_i} \Delta t$$

$$r_i(t + \Delta t) = r_i(t) + V_i(t + \frac{\Delta t}{2}) \Delta t$$



$$\Delta t = 2 \text{ fs}$$

* Barostats and Thermostats

- We simulate a system coupled to an external bath (NPT, NVT) $\left\{ \begin{array}{l} \rightarrow \text{Temperature} \\ \rightarrow \text{Pressure} \end{array} \right\}$ Constant.

- Temperature:

$$\frac{dT}{d\gamma} = \frac{T - T_0}{\gamma} \rightarrow \text{Coupling Parameter}$$

$$V' = \lambda V(t - \frac{\Delta t}{2})$$

$$\lambda = \lambda(\Delta t, \gamma, T_0, T) \rightarrow \text{Scaling Factor}$$

Berendsen J Chem Phys ~~126: 0141~~ 81: 3684 (1984)

Bussi J Chem Phys 126: 014101 (2007)

- Pressure: similar

$$x' = \gamma x(t)$$

* Improving efficiency

- Remove fast degrees of freedom:

- Bonds (H-atoms) → Constraints
vibrations

$\Delta t = 2 \text{ fs}$

Lincs Hess, J Comp Chem
Shake 12:1463
1997

- Angle vibrations

$\Delta t = 4 \text{ fs}$

→ Vsites
Feenstra, J Comp Chem
20:786
1999

- Short range interactions: Cutoffs

- Electrostatics: Ewald summation PME

Darden JCP 98, 60089, 1993

Essmann JCP 103, 8577, 1995

N^2 problem
Direct calculation

→

$N \log N$
Cutoff
Long range

Direct space
Reciprocal Space

- Neighborlist

Verlet Buffer

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